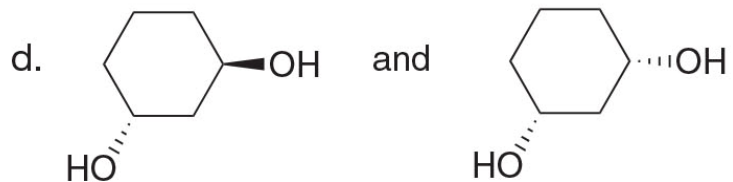
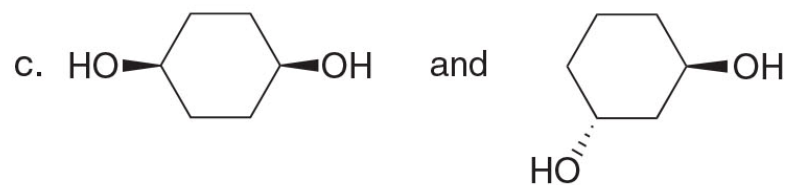
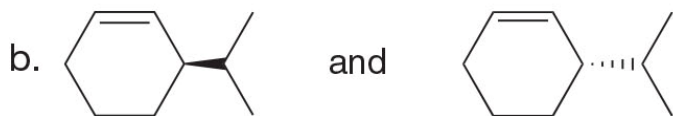
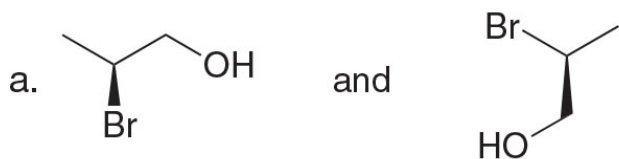
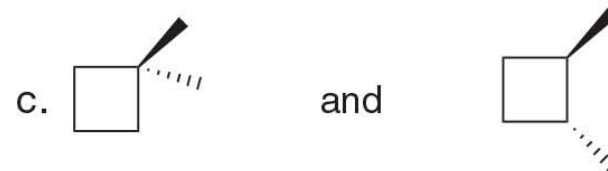
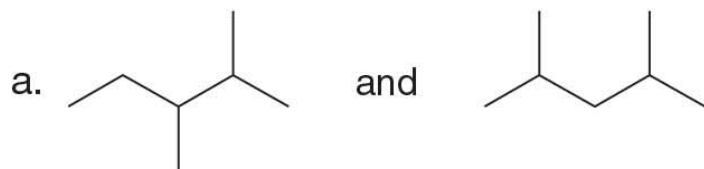
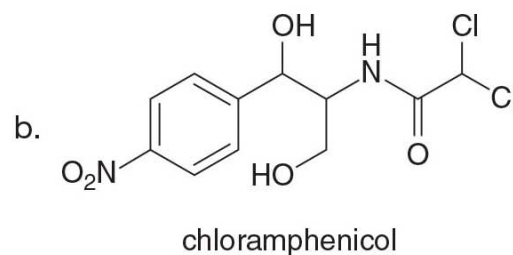
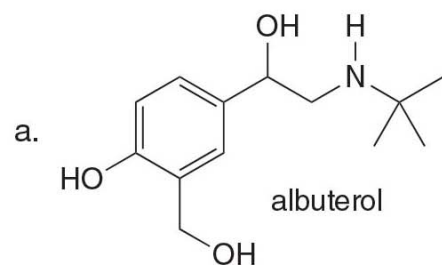


Define if constitutional isomers or stereoisomers.

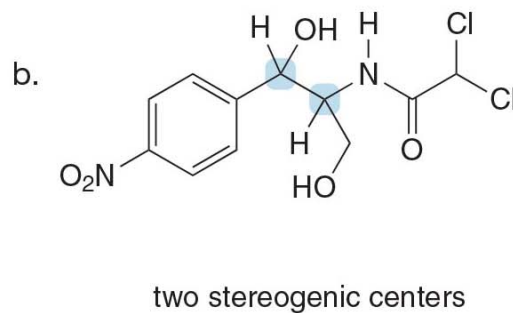
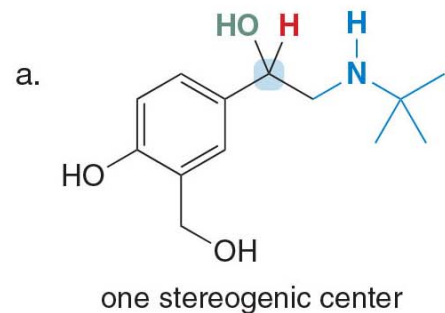


Locate the stereogenic centers in each drug. Albuterol is a bronchodilator—that is, it widens airways—so it is used to treat asthma. Chloramphenicol is an antibiotic used extensively in developing countries because of its low cost.

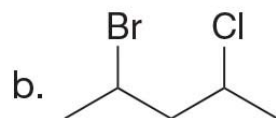
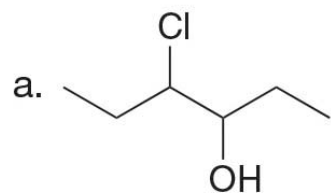
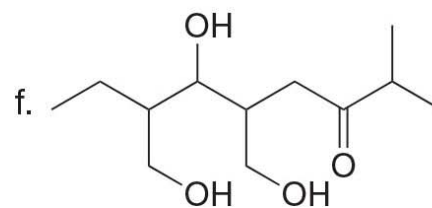
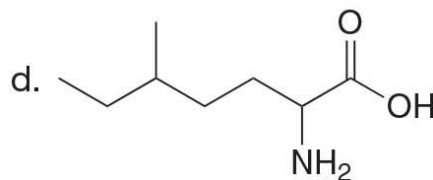
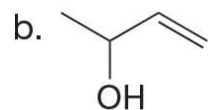
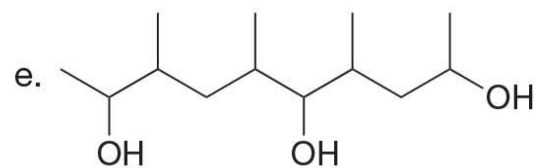
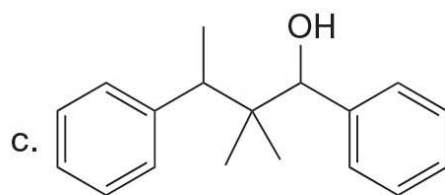
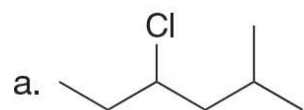


### Solution

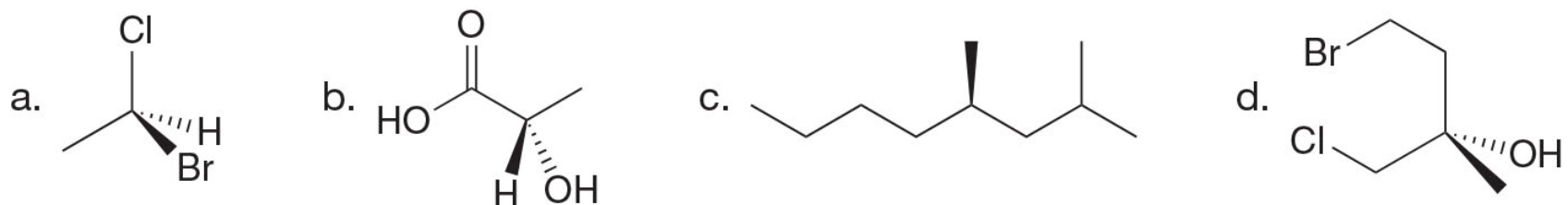
Omit all CH<sub>2</sub> and CH<sub>3</sub> groups and all doubly bonded ( $sp^2$  hybridized) C's. In albuterol, one C has three CH<sub>3</sub> groups bonded to it, so it can be eliminated as well. Draw in H atoms on tetrahedral C's in skeletal structures to more clearly see the groups. This leaves one C in albuterol and two C's in chloramphenicol surrounded by four different groups, making them stereogenic centers.



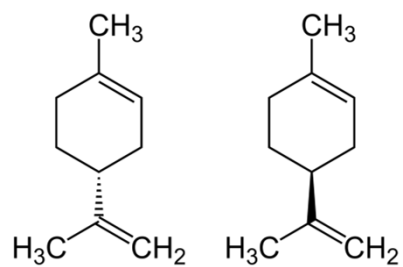
Locate the stereogenic centers in each molecule.



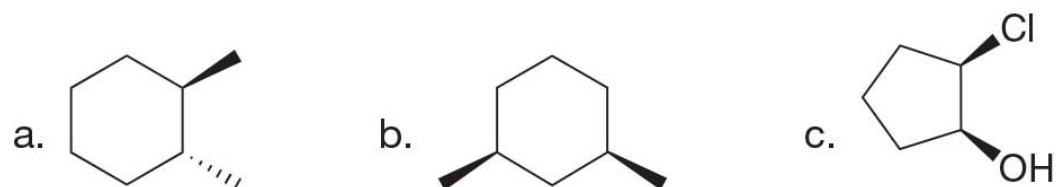
Assign R or S configuration at each stereocenter and determine if the molecule is chiral.



### Limonene



fragrance of oranges turpentine-like odor



Draw the following molecules (skeletal representation including stereochemistry).

a. (*R*)-3-methylhexane

b. (*4R,5S*)-4,5-diethyloctane

c. (*3R,5S,6R*)-5-ethyl-3,6-dimethylnonane

d. (*3S,6S*)-6-isopropyl-3-methyldecane

Assign the IUPAC name to the following molecules.

