

```
> restart;with(plots):#with(RealDomain):
```

Solution suggestion problem 300-12

We were asked to find an expression - for a situation describing two films of ZnO on top of each other, one with the isotope ^{70}Zn , and where we seek the diffusion profile of ^{70}Zn .

We will assume that we consider only diffusion times that are so short that the diffusion zone will be mostly around the interface. It means we assume $x_0 \gg \sqrt{Dt}$ where x_0 is the film thickness. Then we can ignore the Sapphire/ZnO interface and the surface of ZnO. Then the situation would be similar to two infinitely thick materials meeting at $x=x_0$.

We assume the profile is a step function initially, and that the small spread observed is the resolution (or diffusion at growth temperature)

We assume we can use the superposition principle.

We divide the ^{70}ZnO material into thin slices (with thickness dx).

...one slice at each x_1 where there is ^{70}ZnO material initially. We find the diffusion profile for each thin slice and sum them up. The thin slice contribution is similar to drive in distribution except that the slice is not on the surface. So we only get half the concentration at each depth in the case at hand. The diffusion profile from atoms at the depth x_1 is thus given by

```
> C1(x,t)=Q/sqrt(4*Pi*Dd*t)*exp(-(x-x1)^2)/4/Dd/t);
```

$$C1(x, t) = \frac{1}{2} \frac{Q e^{-\frac{1}{4} \frac{(x-x_1)^2}{Dd t}}}{\sqrt{\pi Dd t}}$$

Here Dd is the diffusivity (D is a predefined operator in Maple)

t is the diffusion time Q is defined by the initial distribution where the # atoms per area between x_1 and x_1+dx_1 is $1/2 N_{\text{ZnO}} dx$

```
> C1(x,t)*dx1=1/2*1/2*NZnO*dx1/sqrt(4*Pi*Dd*t)*exp(-(x-x1)^2)/4/Dd/t);
```

$$C1(x, t) dx_1 = \frac{1}{8} \frac{N_{\text{ZnO}} dx_1 e^{-\frac{1}{4} \frac{(x-x_1)^2}{Dd t}}}{\sqrt{\pi Dd t}}$$

You can set that equation into the diffusion equation and see that it fits, and you can see the integral is always the same as the total amount from the slab. We don't do that right now

We can now integrate up $C1(x,t)dx_1$ from $x_1 = -\infty$ to $x_1 = x_0$ to sum up all contributions

```
> C(x,t)=int(C1(x1,x,t),x1=-infinity..x0);
```

$$C(x, t) = \int_{-\infty}^{x_0} C1(x_1, x, t) dx_1$$

```
> C(x,t)='int((1/8)*NZnO*exp(-(1/4)*(x-x1)^2/(Dd*t))/sqrt(Pi*Dd*t), x1=-infinity..x0)';
```

$$C(x, t) = \int_{-\infty}^{x_0} \frac{1}{8} \frac{N_{\text{ZnO}} e^{-\frac{1}{4} \frac{(x-x_1)^2}{Dd t}}}{\sqrt{\pi Dd t}} dx_1$$

We will get

> $C(x,t) = 1/4 * N_{ZnO} * \text{erfc}((x-x_0)/2/\sqrt{Dd*t});$

$$C(x,t) = \frac{1}{4} N_{ZnO} \text{erfc}\left(\frac{1}{2} \frac{x-x_0}{\sqrt{Dd t}}\right)$$

We can check that this obeys the diffusion equation, We define it as function fC

> $fC := (x,t) \rightarrow 1/4 * N_{ZnO} * \text{erfc}((x-x_0)/2/\sqrt{Dd*t});$

$$fC := (x,t) \rightarrow \frac{1}{4} N_{ZnO} \text{erfc}\left(\frac{1}{2} \frac{x-x_0}{\sqrt{Dd t}}\right)$$

> $\text{diffeq} := \text{diff}(C(x,t),t) = Dd * \text{diff}(\text{diff}(C(x,t),x),x);$

$$\text{diffeq} := \frac{\partial}{\partial t} C(x,t) = Dd \left(\frac{\partial^2}{\partial x^2} C(x,t) \right)$$

> $\text{eval}(\text{subs}(C(x,t)=fC(x,t),\text{diffeq}));$

$$\frac{1}{8} \frac{N_{ZnO} e^{-\frac{1}{4} \frac{(x-x_0)^2}{Dd t}} (x-x_0) Dd}{\sqrt{\pi} (Dd t)^{3/2}} = \frac{1}{8} \frac{N_{ZnO} (x-x_0) e^{-\frac{1}{4} \frac{(x-x_0)^2}{Dd t}}}{\sqrt{\pi} t \sqrt{Dd t}}$$

> $\text{simplify}(\text{lhs}(\%) - \text{rhs}(\%));$

0

So left hand side is equal to the right hand side, and the expression is a solution of the diffusion equation.

Lets check that the initial condition is OK by checking fC for $(0 < x < x_0$ and $x > x_0$

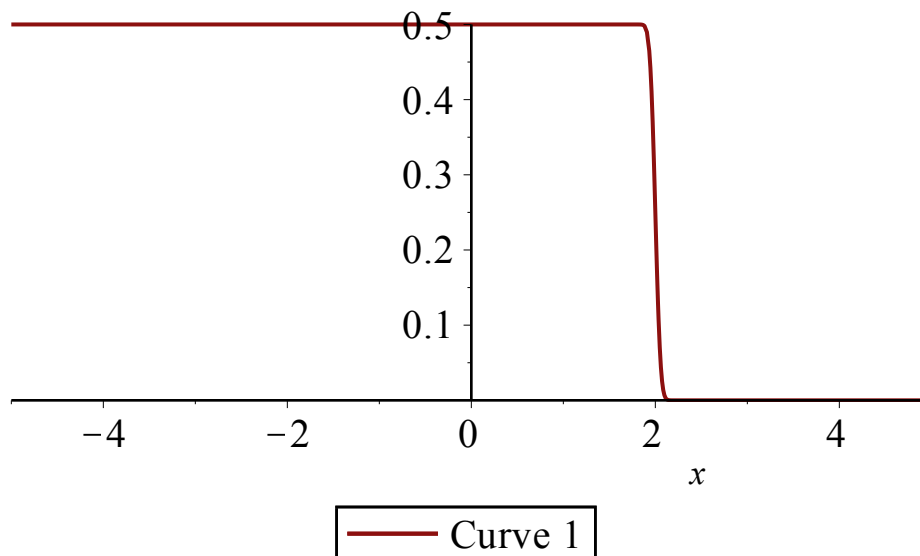
> $\text{evalf}(\text{subs}(\{x_0=1, x=-1000, t=1, Dd=1\}, fC(x,t)), 2); \text{evalf}(\text{subs}(\{x_0=1, x=1000, t=1, Dd=1\}, fC(x,t)), 1);$

$0.50 N_{ZnO}$

$6. 10^{-108578} N_{ZnO}$

We see the concentration is $0.5 * N_{ZnO}$ as it should below x_0 and the second number is practically 0

> $\text{plot}(\text{subs}(\{N_{ZnO}=1, Dd=1, x_0=2, t=0.001\}, fC(x,t)), x=-5..5);$



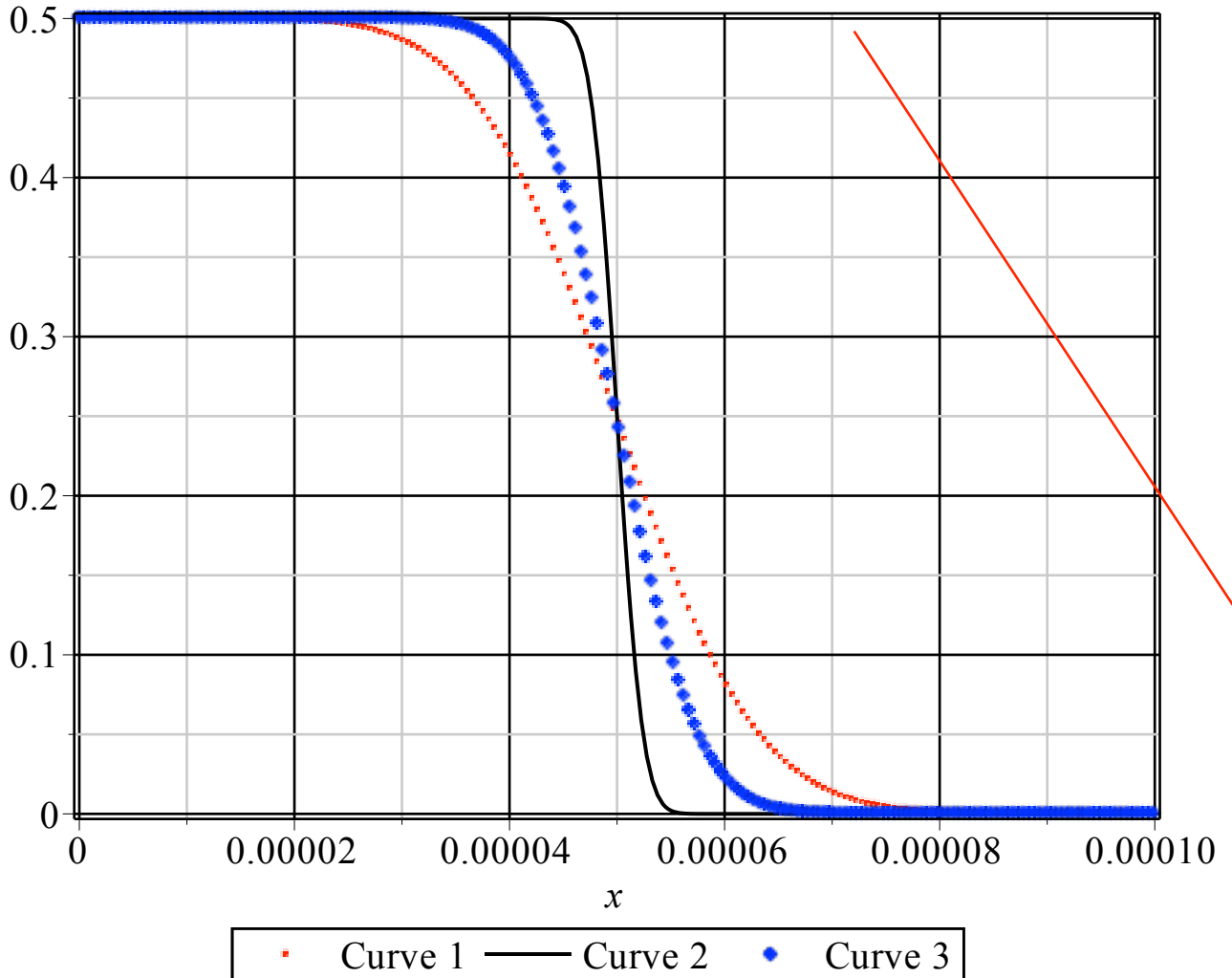
We see the initial conditions are OK.

We have shown what the concentration should look like.

How to find from experimental data?

They were generated by

```
> p1:=plot(subs({N[ZnO]=1,Dd=0.5e-14, x0=5e-5,t=3600*0.1},fC(x,t)),x=0..1e-4,color=black):p2:=plot(subs({N[ZnO]=1,Dd=0.5e-14, x0=5e-5,t=3600*1},fC(x,t)),x=0..1e-4,color=blue):p3:=plot(subs({N[ZnO]=1,Dd=0.5e-14, x0=500e-7,t=3600*3},fC(x,t)),x=0..1e-4,color=red):display({p1,p2,p3});
```



When the scale goes to 0.5 at the surface, it means it is relative to $N[\text{ZnO}]$,

So how to find the diffusivity

Strategy, read off the point where $C(x,t)$ is equal to say 0.4, We could do that for the curves and solve the equation for $C(x,t)$, We could repeat for other values, we could pick a lot of points and do a least square curve fit.

```
> restart;with(RealDomain):
```

We have the theoretical concentration

```
> eqX:=C(x, t) = (1/4)*N[ZnO]*erfc((1/2)*(x-x0)/sqrt(Dd*t));
```

$$eqX := C(x, t) = \frac{1}{4} N_{\text{ZnO}} \operatorname{erfc}\left(\frac{1}{2} \frac{x - x_0}{\sqrt{Dd t}}\right)$$

For the blue curve we read off $C=0.4 \rightarrow x=0.000044726$

```
> eqX1:=subs({C(x,t)=0.4,N[ZnO]=1,x0=5e-5,t=3600*1,x=0.000044726},
```

eqX);

$$eqX1 := 0.4 = \frac{1}{4} \operatorname{erfc} \left(- \frac{7.325000000 \cdot 10^{-10} \sqrt{3600}}{\sqrt{Dd}} \right)$$

> Dd=evalf(solve(eqX1,Dd));

$$Dd = 5.453990619 \cdot 10^{-15}$$

For the red curve we read off C=0.4 ->x=0.000041229, 4522

> eqX2:=subs({C(x,t)=0.4,N[ZnO]=1,x0=5e-5,t=3600*3,x=0.000041229},eqX);

$$eqX2 := 0.4 = \frac{1}{4} \operatorname{erfc} \left(- \frac{4.060648148 \cdot 10^{-10} \sqrt{10800}}{\sqrt{Dd}} \right)$$

> Dd=evalf(solve(eqX2,Dd));

$$Dd = 5.028183320 \cdot 10^{-15}$$

The diffusivity is determined to around 5e-15

which incidentally agree with the value used for generation

Reading of x=0.00004 C=0.578, blue curve

**> eqX3:=subs({C(x,t)=0.476,N[ZnO]=1,x0=5e-5,t=3600*1,x=0.00004},eqX);
Dd=evalf(solve(eqX3,Dd));**

$$eqX3 := 0.476 = \frac{1}{4} \operatorname{erfc} \left(- \frac{1.388888889 \cdot 10^{-9} \sqrt{3600}}{\sqrt{Dd}} \right)$$

$$Dd = 5.012646774 \cdot 10^{-15}$$

**> eqX4:=subs({C(x,t)=0.416,N[ZnO]=1,x0=5e-5,t=3600*3,x=0.00006},eqX);
Dd=evalf(solve(eqX4,Dd));**

$$eqX4 := 0.416 = \frac{1}{4} \operatorname{erfc} \left(\frac{4.629629630 \cdot 10^{-10} \sqrt{10800}}{\sqrt{Dd}} \right)$$

$$Dd = 5.001576832 \cdot 10^{-15}$$

We can look at the derivative

> diff(C(x,t),x)=evalf(subs(x=x0,diff(rhs(eqX),x))/N[ZnO]);Dd=solve(%,Dd);

$$\frac{\partial}{\partial x} C(x, t) = - \frac{0.1410473959}{\sqrt{Dd t}}$$

$$Dd = \frac{0.01989436789}{t \left(\frac{\partial}{\partial x} C(x, t) \right)^2}$$

Red curve Derivati dC/dx=0.5/0.00002519

**> subs({t=3600*3, diff(C(x, t), x)=0.5/0.00002519},Dd =
0.1989436789e-1/(t*(diff(C(x, t), x))^2));**

$$Dd = 4.675442448 \cdot 10^{-15}$$

Could read off any width of C slope vlu from red and blue, say for between C=0.1 and C=0.4 , Delx
plot delx^2 vs time t, slope is diffusivity