

# Nanoscale thermal transport

Lecture 3

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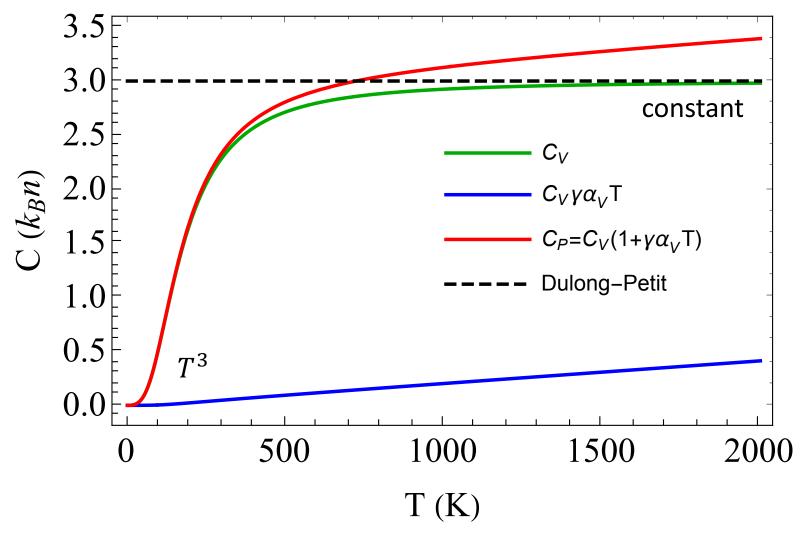
http://rileyhanus.com/science.html

- 1. General discussion about heat capacity
- 2. Obtain the phonon heat capacity
  - a) examine limiting behavior
  - b) low  $\omega$ , high T, low T
- 3. Heat capacity in practice
  - a) Debye level
  - b) Debye temperature
  - c) Speed sound
- 4. Introduction to scattering theory

# Heat capacity

Heat capacity is generally a very boring material property.

Material	Volumetric heat capacity at 25 C $C_{ m V}~({ m J/K~cm^3})$
Liquid Hg	1.9
Liquid water	4.2
Cu	3.45
Glass	2.1
Al	2.4
Graphite	1.53
Diamond	1.8



n: number density of atoms

 $\gamma$ : Gruneisen parameter

 $\alpha_{\rm V}$ : volumetric thermal expansion ( $\alpha_{\rm V} \cong 3\alpha_{\rm L}$ )

## **Derivation set-up**

Heat capacity is the change in energy with temperature

We will look for the *spectral* heat capacity *per unit volume* 

$$C(\omega) = \frac{d E(\omega, T)}{dT}$$

Energy density:  $E(\omega, T) = \hbar \omega n(\omega, T)$ 

Number density of phonons:  $n(\omega, T) = g(\omega) n_{BE}(\omega, T)$ 

### Note:

- $g(\omega)$  is the number of possible states per unit volume. Here, we say it doesn't change with temperature (i.e. the quasi-harmonic approximation). In reality thinks get softer with increasing temperature so  $g(\omega)$  will shift down in  $\omega$  as T increases.
- $n_{\rm BE}(\omega,T)$  is the occupation number (can think of it as an occupation probability), of a state at frequency/energy  $\omega$ , and temperature T.

$$\frac{d E(\omega, T)}{dT} = \hbar \omega g(\omega) \frac{d n_{BE}(\omega, T)}{dT}$$

### **Derivation**

$$n_{\rm BE} = \frac{1}{e^{\frac{\hbar\omega}{k_{\rm B}T}} - 1}$$

Crazy triple chain rule

$$\frac{dn_{\rm BE}}{dT} = \frac{(-1)}{\left(e^{\frac{\hbar\omega}{k_{\rm B}T}} - 1\right)^2} \times e^{\frac{\hbar\omega}{k_{\rm B}T}} \times \frac{(-1)\hbar\omega}{k_{\rm B}T^2} = \frac{\hbar\omega}{k_{\rm B}T^2} \frac{e^{\frac{\hbar\omega}{k_{\rm B}T}}}{\left(e^{\frac{\hbar\omega}{k_{\rm B}T}} - 1\right)^2}$$

$$C(\omega) = \hbar\omega \, g(\omega) \frac{dn_{\rm BE}}{dT} = \hbar\omega \times \frac{3}{2\pi^2} \frac{\omega^2}{v_{\rm p}^2 v_{\rm g}} \times \frac{\hbar\omega}{k_{\rm B}T^2} \frac{e^{\frac{\hbar\omega}{k_{\rm B}T}}}{\left(\frac{\hbar\omega}{e^{\frac{\hbar\omega}{k_{\rm B}T}}-1}\right)^2}$$

$$C(\omega) = \frac{3}{2\pi^2} \frac{\omega^4}{v_{\rm g} v_{\rm p}^2} \frac{\hbar^2}{k_{\rm B} T^2} \frac{e^{\hbar \omega/k_{\rm B} T}}{(e^{\hbar \omega/k_{\rm B} T} - 1)^2}$$

$$Theat capacity per volution of the compact of the compa$$

Heat capacity per volume

### Limiting behavior

Let's look at the low  $\omega$  (and high T) behavior

$$C(\omega) = \frac{3}{2\pi^2} \frac{\omega^4}{v_{\rm g} v_{\rm p}^2} \frac{\hbar^2}{k_{\rm B} T^2} \frac{e^{\hbar \omega/k_{\rm B} T}}{(e^{\hbar \omega/k_{\rm B} T} - 1)^2}$$

If I just plug in  $\omega = 0$ ,  $C(\omega) = 0/0$ .

Not super helpful, what we really want is the limiting behavior.

Define: 
$$x = \frac{\hbar \omega}{k_{\rm B}T}$$
 Taylor expand  $e^x$  about  $x = 0$ .  $e^x \approx 1 + x + \cdots$ 

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$$C(\omega) = \frac{3}{2\pi^2} \frac{\omega^4}{v_p^2 v_g} \frac{\hbar^2}{k_B T^2} \frac{(1)}{(1+x-1)^2} = \frac{3}{2\pi^2} \frac{\omega^4}{v_p^2 v_g} \frac{\hbar^2}{k_B T^2} \left(\frac{k_B T}{\hbar \omega}\right)^2 = k_B \frac{3}{2\pi^2} \frac{\omega^2}{v_p^2 v_g}$$

At high T and or low  $\omega$ 

$$C(\omega) = k_{\rm B} g(\omega)$$

which means  $C \rightarrow cst$  at high T and  $C(\omega) \propto \omega^2$  when the dispersion is linear

# Limiting behavior

Let's look at **low** T behavior

Convenient to convert the integral to be over the dimensionless parameter x.

$$x = \frac{\hbar\omega}{k_{\rm B}T} \qquad C = \frac{3}{2\pi^2} \frac{\hbar^2}{k_{\rm B}T^2} \int_0^{\omega_{\rm max}} \frac{\omega^4}{v_{\rm g} v_{\rm p}^2} \frac{e^{\hbar\omega/k_{\rm B}T}}{(e^{\hbar\omega/k_{\rm B}T} - 1)^2} d\omega$$

$$d\omega = \frac{k_{\rm B}T \, dx}{\hbar}$$

$$x_{\rm max} = \frac{\hbar\omega_{\rm max}}{k_{\rm B}T} \qquad C = \frac{3}{2\pi^2} \frac{k_{\rm B}^4 T^3}{\hbar^3} \int_0^{x_{\rm max}} \frac{x^4}{v_{\rm g} v_{\rm p}^2} \frac{e^x}{(e^x - 1)^2} dx$$

At low T:

- 1. only low frequency modes are populated,  $v_{
  m g}=v_{
  m p}=v_{
  m s}$
- 2.  $x_{\text{max}} \rightarrow \infty$

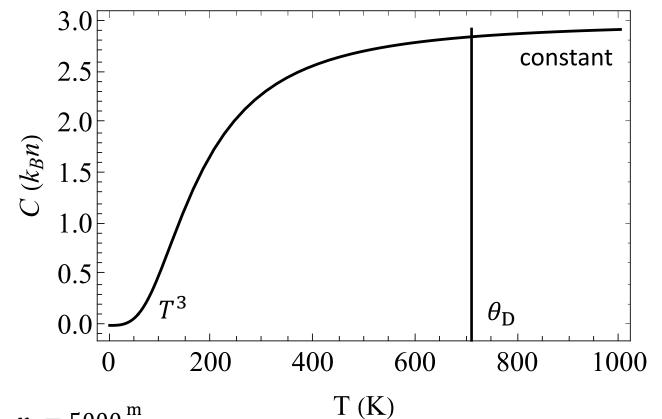
$$\int_{0}^{\infty} \frac{x^4 e^x}{(e^x - 1)^2} dx = \frac{4\pi^4}{15} \qquad C = \frac{3}{2\pi^2} \frac{k_B^4 T^3}{\hbar^3} \left(\frac{4\pi^4}{v_S^3 15}\right)$$

At low T where dispersion is linear:

 $C \propto T^3$ 

# Example model

$$C = \frac{3}{2\pi^2} \frac{\hbar^2}{k_{\rm B} T^2 v_{\rm S}^3} \int_{0}^{\omega_{\rm max}} \frac{\omega^4 e^{\hbar \omega/k_{\rm B} T}}{(e^{\hbar \omega/k_{\rm B} T} - 1)^2} d\omega$$



speed of sound:

$$v_{\rm s} = 5000 \frac{\rm m}{\rm s}$$

volume per atom:

$$V_{\rm at} = \frac{V_{\rm UC}}{N_{\rm at}} = 9 \, {\rm \AA}^3$$
 ( $V_{\rm UC}$ : volume of unit cell,  $N_{\rm at}$ : # atoms in unit cell)

max k-vector (isotropic):  $k_{\mathrm{max}} = \left(\frac{6\pi^2}{V_{\mathrm{at}}}\right)^{1/3} = 1.87~\mathrm{\AA}$ 

Debye frequency:  $\omega_{\mathrm{D}} = \omega_{\mathrm{max}} = v_{\mathrm{s}} k_{\mathrm{max}} = 93.7 \ \mathrm{THz}$ 

Debye temperature:  $k_{\rm B}\theta_{\rm D}=\hbar\omega_{\rm D}$   $\theta_{\rm D}=715~{\rm K}$ 

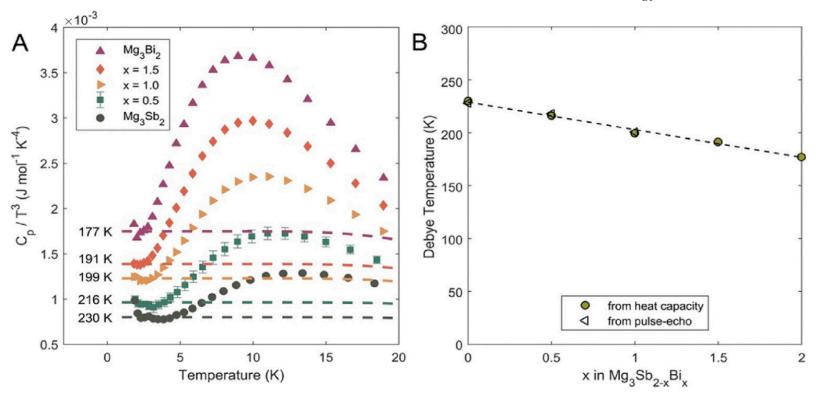
Analyzing low-T data

$$C = \frac{3}{2\pi^2} \frac{k_{\rm B}^4 T^3}{\hbar^3} \left( \frac{4\pi^4}{v_{\rm S}^3 15} \right)$$

$$\frac{C}{T^3} = \left( \frac{2}{5} \right) \frac{\pi^2 k_{\rm B}^4}{\hbar^3 v_{\rm S}^3} = \beta$$

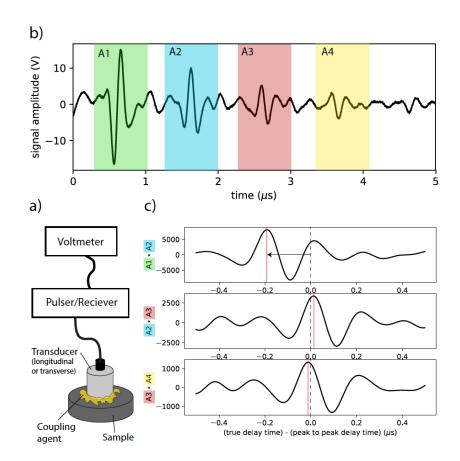
$$\theta_D = \left( \frac{12\pi^4}{5} \frac{n k_{\rm B}}{\beta} \right)^{1/3}$$

 $n = \frac{1}{V_{\rm at}}$ : number density of atoms



Agne, M. T. et al. Heat capacity of Mg3Sb2, Mg3Bi2 and their alloys at high temperature. *Mater. Today Phys.* **6**, DOI 10.1016/j.mtphys.2018.10.001 (2018).

# Speed of sound

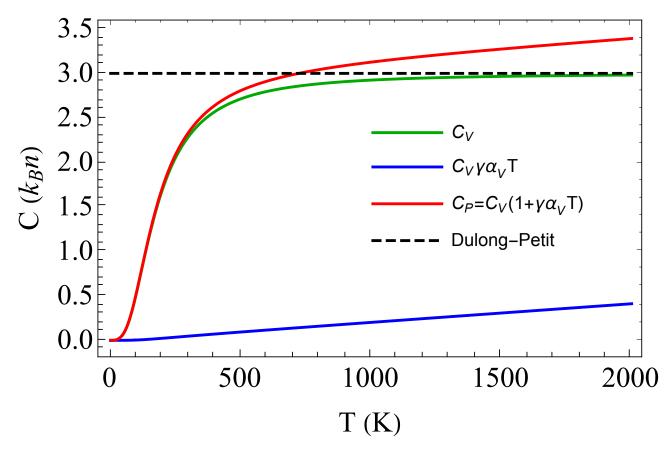


The isotropic average speed of sound, used for thermal conductivity modeling, heat capacity modeling, Debye temperature.

- When the cross-correlation is used to measure the time delay, the error is typically limit by the thickness variation of the sample. Can get better than 1% accuracy, if careful.
- If sample is a randomly oriented polycrystalline, then you are measuring the isotropic average longitudinal and transverse speeds of sound.
- Be mindful of preferred orientation of grains, especially for non-cubic materials.
- Be mindful of sample density. Samples should be very dense, >98%.

$$v_{\rm S} = \left(\frac{1}{3} \left[ \frac{1}{v_{\rm L}^3} + \frac{2}{v_{\rm T}^3} \right] \right)^{-1/3}$$

# Thermal expansion contribution



n: number density of atoms

 $\gamma = B\alpha_{\rm V}/C_{\rm V}$ : Gruneisen parameter

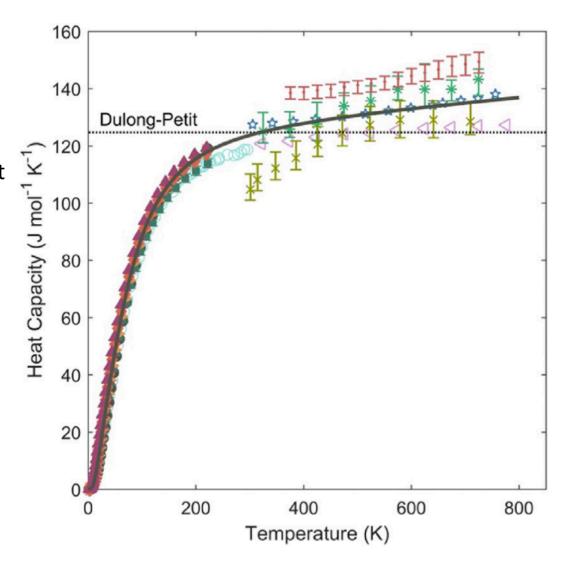
B: isothermal Bulk modulus

 $\alpha_{\rm V}$ : volumetric thermal expansion ( $\alpha_{\rm V}\cong 3\alpha_{\rm L}$ )

$$C_{\rm p} = C_{\rm V}(1 + \gamma \alpha_{\rm V} T)$$

# Analyzing high-T data

- Absolute value of high temperature heat capacity is notoriously hard to measure.
- In contrast low-temperature heat capacity can be very accurate (e.g. physical property measurement system <u>PPMS</u> in heat capacity mode).
- Often times our models are better at predicting it then and one given measurement is.
- The slope of Heat Capacity vs. T at high T can be reasonably accurate.



Agne, M. T. et al. Heat capacity of Mg3Sb2, Mg3Bi2 and their alloys at high temperature. *Mater. Today Phys.* **6**, DOI 10.1016/j.mtphys.2018.10.001 (2018).

# Thermal conductivity considerations

Modeling:

$$\kappa = \int C(\omega) D(\omega) d\omega$$

$$D(\omega) = \frac{1}{3}v_{\rm g}^2(\omega)\tau(\omega)$$

 $C(\omega)$ : spectral heat capacity per

unity volume

 $D(\omega)$ : spectral diffusivity

### Experiment:

$$\kappa = C_{\rm V} D = C_{\rm P} \rho D$$

*D*: thermal diffusivity measured by, for example LFA

 $C_{\rm V}$ : heat capacity per unit volume

 $C_P$ : heat capacity per unit mass

 $\rho$ : mass density

$$C_{\rm V} = \int C(\omega) \, d\omega$$

$$D = \frac{\int C(\omega)D(\omega)d\omega}{\int C(\omega)d\omega}$$

Heat capacity weighted average diffusivity

$$\kappa = \left( \int C(\omega) d\omega \right) \frac{\int C(\omega) D(\omega) d\omega}{\left( \int C(\omega) d\omega \right)} = \int C(\omega) D(\omega) d\omega$$