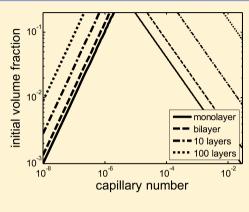
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# Prediction of Coating Thickness in the Convective Assembly Process

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ABSTRACT: Convective assembly is a coating method to fabricate thin films with ordered particle structures that can be used extensively for biochemical sensors, data storage devices, optical devices, and other applications. The fluid flow into or through the close-packed region causes the convective assembly, and it is important to understand the formation mechanism of the close-packed region. In this paper, the length of the close-packed region was predicted, and the dimensionless coating thickness as well as the dimensionless length of the close-packed region was found to be the functions of only three dimensionless variables: two capillary numbers and the initial volume fraction. From the modeling results, coating process regime maps that predict the dimensionless coating thickness in terms of the dimensionless variables were created. In addition, the length of the close-packed region was measured under various coating conditions to validate the model prediction. The experiments firmly supported the model predictions.



#### 1. INTRODUCTION

Particulate coatings are commonly used for many industrial products such as solar cells, batteries and so on. The properties and performance of the coatings depend on the microstructure, and it is important to understand the microstructural change during drying, which can be characterized by the particle distribution, the degree of alignment, and the coating thickness. A number of modeling and experimental studies that analyzed the microstructural change of the particulate coatings have been done. Both modeling and experimental studies emphasized the formation and the role of the close-packed region. The close-packed region is the region where the particles are closely packed with solvent filling in the interparticle void space. The capillary pressure of the meniscus in the close-packed region and the fluid flow into or through the close-packed region have a strong influence on the formation of the microstructure.

Routh and Zimmerman<sup>9</sup> and Cardinal et al.<sup>10</sup> predicted the particle distribution in the vertical direction of a substrate and analyzed the formation of the close-packed region during drying by solving one-dimensional conservation equation. By applying the lubrication approximation, Routh and Russel<sup>3</sup> predicted the particle distribution and the coating thickness in the horizontal direction and analyzed the close-packed region formed from the edge of the film during drying. Maki and Kumar<sup>11</sup> predicted the accumulation of particles at the liquid air interface as well as at the edge of the droplet. Deegan et al. 12,13 showed that a ring-like deposit was produced due to the capillary flow from the center to the edge of the drop induced by the differential evaporation rates across the drop. Dimitrov and Nagayama<sup>5</sup> explained the role of the evaporation in the close-packed region for the convective assembly and predicted the coating thickness.

Using Cryo-SEM, Cardinal et al.<sup>10</sup> observed a well-ordered particle structure in the close-packed region near the coating surface under the evaporation dominant drying condition. Salamanca et al.<sup>4</sup> observed the movement of the boundaries of the close-packed region in the horizontal direction during drying by magnetic resonance microscopy. The movement of the boundary of the close-packed region near the meniscus was observed in the convective assembly.<sup>5</sup> Meng et al.,<sup>7</sup> Gasperino et al.,<sup>8</sup> and Brewer et al.<sup>14</sup> analyzed the crystallization mechanism in the close-packed region by microscopic observation and simulation, respectively; however, the mechanism is not fully understood yet.

Convective assembly is a representative method for the development of a well-ordered structure in the close-packed region by the fluid flow into or through the close-packed region. The general procedure is shown in Figure 1. By placing or lifting slowly a nearly vertical substrate in the bath of a dilute suspension during drying, a thin film with an ordered particle structure develops. It may well be considered as the dip coating of a dilute suspension at a very low processing speed. In analyzing the dip coating process, capillary number *Ca* defined as the relative strength between the viscous stress and the capillary pressure is important.

$$Ca = \frac{\mu(E + u_s)}{\gamma} \tag{1}$$

Here,  $\mu$  is the solvent viscosity,  $\gamma$  is the solvent surface tension, E is the evaporation rate, and  $u_s$  is the drawing speed (E has the same unit as  $u_s$ ). For convenience, the properties of the dilute suspension was assumed to be the same as the properties of the

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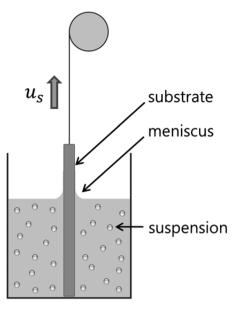


Figure 1. Schematic of the convective assembly.

solvent.  $Ca_0$  is Ca when the drawing speed is zero. In this article, a low processing speed means that Ca is small and a high processing speed means that Ca is large.

For a large Ca, the viscous stress, which is induced by the movement of the substrate, dominates the dip coating process. This corresponds to the typical dip coating process condition, and the wet coating thickness is determined by the well-known Landau–Levich equation.<sup>17</sup> During drying, the particle volume fraction of the film developed by the Landau–Levich deposition increases from the initial volume fraction,  $\phi_0$ , to the particle volume fraction of the random close packing structure,  $\phi_r = 0.64$ . The dry coating thickness,  $h_{\rm LL,dry}$ , is the wet coating thickness,  $0.945l_cCa^{2/3}$ , multiplied by  $\phi_0/0.64$  as follows:

$$\frac{h_{\rm LL,dry}}{l_c} = 1.48Ca^{2/3}\phi_0 \tag{2}$$

where  $l_c = (\gamma/\rho g)^{1/2}$  is the capillary length, which represents the height of the capillary rise ( $\rho$  is the solvent density and g is the acceleration of the gravity). As Ca increases, the coating thickness increases in proportion to  $Ca^{2/3}$ . Ouriemi and Homsy<sup>18</sup> and Dixit and Homsy<sup>19</sup> found that the presence of surface-adsorbed hydrophobic particles results in a slightly different power law from the Landau–Levich equation. However, the effect of interfacial elasticity was neglected in this work. For the convective assembly to develop a well-ordered structure, the particle volume fraction after drying was assumed to be the particle volume fraction of the close-packed lattice structure,  $\phi_m = 0.74$ . In the Landau–Levich deposition, which cannot develop three-dimensional close-packed lattice structure as in the convective assembly, the particle volume fraction after drying is less than 0.74. Here, we assumed a random close packing at  $\phi_r = 0.64$ .

For a small *Ca*, the film is formed by the convective assembly. Figure 2 shows the idealized view of the close-packed region explaining the mechanism of the convective assembly. The particle packing front is the lower boundary of the close-packed region, and the drying front is the upper boundary of the close-packed region. The wet coating thickness, which is the thickness of the close-packed region, equals the coating

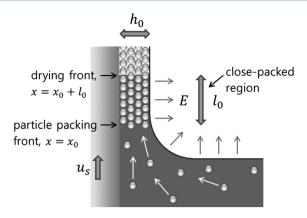


Figure 2. Idealized view of the close-packed region.

thickness of the dry film located above the drying front.  $l_0$  and  $h_0$  represent the length of the close-packed region and the coating thickness of the convective assembly, respectively. When Ca is small enough, the close-packed region is formed on the substrate near the meniscus at the early stage of drying and the fluid flow into or through the close-packed region due to evaporation in the close-packed region causes the convective assembly. Therefore, the coating thickness of the convective assembly is mainly affected by the evaporation in the close-packed region, not the viscous stress.

packed region, not the viscous stress.

Le Berre et al.,<sup>21</sup> Brewer et al.,<sup>22</sup> and Berteloot et al.<sup>23</sup> measured the coating thickness of the dip coating of colloidal suspensions in a wide range of *Ca* from the convective assembly to the Landau–Levich deposition. They showed that the coating thickness decreases as *Ca* increases at the small *Ca* region where the film is formed by the convective assembly, while the coating thickness increases with *Ca* at the large *Ca* region where the film is formed by the Landau–Levich deposition.

To predict the coating thickness of the convective assembly, it is important to estimate the length of the close-packed region. Dimitrov and Nagayama<sup>5</sup> explained the role of evaporation in the close-packed region and predicted the coating thickness. Berteloot et al.<sup>23</sup> predicted the coating thickness using the scale of the meniscus size. However, an accurate prediction could not be made because the length of the close-packed region was either assumed to be constant or neglected. The coating thickness of the convective assembly was measured experimentally under various coating conditions, <sup>6,15,24–26</sup> but there have been few attempts to observe the length of the close-packed region as a function of the coating conditions.

In this study, the length of the close-packed region of the convective assembly was predicted, and the dimensionless length of the close-packed region and the dimensionless coating thickness were shown to be the functions of only three dimensionless variables: Ca,  $Ca_0$ , and  $\phi_0$ . From the modeling results, coating process regime maps that predict the dimensionless coating thickness in terms of the dimensionless variables were created. In addition, the effects of the initial volume fraction and the evaporation rate on the length of the close-packed region were measured experimentally to validate the performance of the modeling.

#### 2. THEORY

It is important to predict the length of the close-packed region in order to understand the convective assembly process. As the

solvent flows through the close-packed region and experiences a pressure drop, the flow can be described by Darcy's law. Then the length of the close-packed region and the coating thickness can be obtained by solving the particle balance equation, the solvent balance equations and Darcy's law in the close-packed region.

To model  $l_0$  and  $h_0$ , several assumptions were required. First, evaporation in the close-packed region was assumed to have a strong influence on the film formation. The validity of this assumption was demonstrated in the discussion on the mechanism of the convective assembly in Figure 2. Second, because  $h_0$  is small relative to the substrate width, the edge effect was neglected. Third, the pressure drop by gravity was assumed to be negligible. In the close-packed region, the maximum capillary pressure of the spherical-cap meniscus in the pore within the close-packed particles of radius a is approximated as  $10\gamma/a = 2.2$  MPa, with a = 100 nm for alcohol.<sup>27</sup> This pressure is equivalent to the pressure drop caused solely by gravity with the height of a column of solvent being 290 m, which is much larger than the length scale (~mm) of the close-packed region in the experiments. So the pressure drop by gravity can be neglected in the convective assembly with the particle size being smaller than or equal to several micrometers. Fourth, the particle volume fraction was assumed to be uniform in the bath during drying because the size of the bath is large relative to that of the coated film. Fifth,  $h_0$  was assumed to have a continuous value in the modeling, though the film has a discontinuous coating thickness because it forms layers of the close-packed particles. Also, the particles were assumed to be monodisperse hard spheres.

By equating the volumetric flow rate of the particle from the bulk solution into the particle packing front and the rate of particle accumulation, the particle balance equation at the particle packing front,  $x = x_0$ , can be established.

$$\phi_0 h_0 w \overline{u}_x(x_0^-) = (\phi_m - \phi_0) h_0 w (E + u_s)$$
(3)

The x-axis is parallel to the substrate.  $x_0$  is the position of the particle packing front, and  $x_0 + l_0$  is the position of the drying front.  $x_0^-$  is the position beneath the particle front, and  $x_0^+$  is the position above the particle packing front. w is the width of the substrate. The particle volume fraction in the close-packed region equals  $\phi_m$  because the particles form a close-packed lattice structure in the close-packed region.  $\overline{u}_x$  is the flow rate of the fluid in the x-direction averaged over the film thickness.  $\overline{u}_x$ is the flow rate of the solution below the close-packed region and becomes the flow rate of the solvent in the close-packed region where the particle movement is restrained.  $\overline{u}_r(x_0^-)$  is the flow rate of the solution beneath the particle packing front. On the right-hand side of eq 3, the particles are accumulated at the particle packing front at a rate of  $E + u_s$  because the substrate is drawn at a rate of  $u_s$ , and the level of the coating solution descends at the rate of E by evaporation. Then an expression for  $\overline{u}_{x}(x_{0}^{-})$  can be obtained as follows:

$$\overline{u}_{x}(x_{0}^{-}) = \frac{(\phi_{m} - \phi_{0})}{\phi_{0}}(E + u_{s})$$
(4)

By equating the volumetric flow rate of the solvent from the bulk solution into the particle packing front and the difference between the volumetric flow rate of the solvent from the particle packing front toward the close-packed region, and the rate of reduction of the solvent due to the descent of the particle packing front, the solvent balance equation at the particle packing front,  $x = x_0$ , can be set up as follows:

$$(1 - \phi_0)h_0w\overline{u}_x(x_0^-)$$

$$= (1 - \phi_m)h_0w\overline{u}_x(x_0^+) - (\phi_m - \phi_0)h_0w(E + u_s)$$
 (5)

where  $\overline{u}_x(x_0^+)$  is the flow rate of the solvent above the particle packing front. After utilizing  $\overline{u}_x(x_0^-)$  from eq 4 in eq 5,  $\overline{u}_x(x_0^+)$  can be derived as follows:

$$\overline{u}_{x}(x_{0}^{+}) = \frac{(\phi_{m} - \phi_{0})}{(1 - \phi_{m})\phi_{0}} (E + u_{s})$$
(6)

When the volume of the solvent evaporation in the close-packed region is approximated as being equal to the volumetric flow rate of the solvent right above the particle packing front, the solvent balance equation in the closed-packed region,  $x_0 < x < x_0 + l_0$ , can be established.

$$l_0 w E = (1 - \phi_m) h_0 w \overline{u}_x (x_0^+)$$
 (7)

Then  $h_0$  in eq 7 can be solved using  $\overline{u}_x(x_0^+)$  from eq 6.

$$h_0 = \frac{\phi_0}{(\phi_m - \phi_0) \left(1 + \frac{u_s}{E}\right)} l_0 \tag{8}$$

Assuming  $\phi_0$  is small and substituting 0.74 for  $\phi_{\it m}$ , the following can be obtained.

$$h_0 = 1.35 \frac{\phi_0}{\left(1 + \frac{u_s}{E}\right)} l_0 \tag{9}$$

Dimitrov and Nagayama<sup>5</sup> obtained the coating thickness of the convective assembly as follows:

$$k = \frac{\beta l_0 E \phi_0}{1.21 a v_c^{(k)} (1 - \phi_0)} \tag{10}$$

where k is the number of layers;  $\beta$  is the ratio between the velocity of a particle in solution and the fluid velocity, and  $\nu_c^{(k)}$  is the growth rate of the k-layer array. As the particles form the close-packed lattice structure in the close-packed region, the coating thickness can be expressed as  $h_0 = (2\sqrt{6/3})ka + (2-(2\sqrt{6/3}))a$ . Assuming  $h_0 \gg (2-(2\sqrt{6/3}))a$ ,  $h_0$  is approximated as  $(2\sqrt{6/3})ka$ . After substituting  $E+u_s$  for  $\nu_c^{(k)}$ , 1 for  $\beta$  and  $(2\sqrt{6/3})ka$  for  $h_0$  in eq 10 and assuming  $\phi_0$  is small, the same result with eq 9 can be obtained.  $l_0$  is included in both eq 9 and eq 10. However,  $l_0$  is not a constant and needs to be a function of the material properties and coating conditions.

The solvent flow through the close-packed region can be described by Darcy's law.  $l_0$  can be modeled by equating the amount of pressure drop in the close-packed region calculated by Darcy's law and the maximum capillary pressure of the spherical-cap meniscus in the pore of the close-packed region.

The flow rate of the solvent in the closed-packed region,  $x_0 < x < x_0 + l_0$ , can be obtained from the continuity equation of the solvent. By equating the divergence of the volumetric flow rate of the solvent and the evaporation rate of the solvent, the continuity equation of the solvent in the close-packed region can be derived.

$$\frac{\partial}{\partial x}[(1-\phi_m)h_0w\overline{u}_x(x)] = -wE \tag{11}$$

Because the flow rate of the solvent is expressed by eq 6 above the particle packing front and is zero at the drying front, the boundary conditions on eq 11 are

$$\overline{u}_x = \frac{(\phi_m - \phi_0)}{(1 - \phi_m)\phi_0} (E + u_s) \quad \text{at} \quad x = x_0^+$$
(12)

$$\overline{u}_x = 0 \quad \text{at} \quad x = x_0 + l_0 \tag{13}$$

Then  $\overline{u}_x(x)$  becomes

$$\overline{u}_{x}(x) = \frac{(E + u_{s})(\phi_{m} - \phi_{0})}{(1 - \phi_{m})\phi_{0}} \left[ 1 - \frac{(x - x_{0})}{l_{0}} \right]$$
(14)

The solvent flow through the close-packed region is described by Darcy's law, adapted from Routh and Russel.<sup>3</sup> Because in the close-packed region Reynolds number,  $Re = (\rho \overline{u}_x a/\mu)$ , is on the order of  $10^{-6}$  with  $E = 10^{-7}$  m/s and a = 100 nm, it is valid to apply Darcy's law.

$$\frac{\partial p}{\partial x} = -\frac{\mu}{k_p} (1 - \phi_m) (\overline{u}_x - u_s) = -\frac{75\mu \phi_m^2}{2a^2 (1 - \phi_m)^2} (\overline{u}_x - u_s)$$
(15)

where p is the pressure and  $k_p$  is the permeability.  $k_p$  is the function of  $\phi_m$  and a according to the Carman–Kozeny equation. Because the capillary pressure has a maximum value at  $x = x_0 + l_p$  where the pressure gradient becomes zero and the maximum capillary pressure of the meniscus in the close-packed region can be approximated as  $10\gamma/a$ ,

$$p_c(x_0 + l_p) \approx \frac{10\gamma}{a} \tag{16}$$

where  $p_c$  is the capillary pressure, and  $l_p$  is the length between the particle packing front and the position where the pressure gradient is zero in the close-packed region. According to eq 15, the flow rate of the solvent equals  $u_s$  at the position where the pressure gradient is zero. By substituting  $x = x_0 + l_p$  and  $\overline{u}_x = u_s$  into eq 14,  $l_p/l_0$  can be obtained.

$$\frac{l_p}{l_0} = 1 - \frac{(1 - \phi_m)\phi_0}{(\phi_m - \phi_0)\left(1 + \frac{E}{u_s}\right)}$$
(17)

If  $\phi_0$  is small, the second term of the right-hand side is negligible and  $l_p$  is approximately equal to  $l_0$ . So the maximum capillary pressure appears near the drying font. By equating the amount of the pressure drop calculated by integrating eq 15 from  $x = x_0$  to  $x = x_0 + l_p$  after using  $\overline{u}_x(x)$  from eq 14 into eq 15 and the maximum capillary pressure, the following equation is obtained:

$$-\int_{x_0}^{x_0+l_p} -\frac{75\mu\phi_m^2}{2a^2(1-\phi_m)^2} \left\{ \frac{(E+u_s)(\phi_m-\phi_0)}{(1-\phi_m)\phi_0} \times \left[1-\frac{(x-x_0)}{l_0}\right] - u_s \right\} dx = \frac{10\gamma}{a}$$
(18)

By solving both eq 17 and eq 18, the following equation is derived:

$$l_{0} = \frac{8a\gamma(1-\phi_{m})^{3}}{15\mu E\left(1+\frac{u_{s}}{E}\right)\phi_{m}^{2}\left[1-\frac{(1-\phi_{m})\phi_{0}}{(\phi_{m}-\phi_{0})\left(1+\frac{E}{u_{s}}\right)}\right]^{2}}\frac{\phi_{0}}{(\phi_{m}-\phi_{0})}$$
(19)

Finally, after inserting  $l_0$  in eq 19 into eq 8,  $h_0$  is obtained as a function of the material properties and the coating conditions.

$$h_{0} = \frac{8a\gamma(1 - \phi_{m})^{3}}{15\mu E\left(1 + \frac{u_{s}}{E}\right)^{2}\phi_{m}^{2}\left[1 - \frac{(1 - \phi_{m})\phi_{0}}{(\phi_{m} - \phi_{0})\left(1 + \frac{E}{u_{s}}\right)}\right]^{2}} \frac{\phi_{0}^{2}}{(\phi_{m} - \phi_{0})^{2}}$$
(20)

If  $\phi_0$  is small, the dimensionless length of the close-packed region and the dimensionless coating thickness can be expressed simply as functions of three dimensionless variables: Ca,  $Ca_0$ , and  $\phi_0$ .

$$\frac{l_0}{a} = 0.023 \frac{\gamma}{\mu(u_s + E)} \phi_0 = 0.023 Ca^{-1} \phi_0 \tag{21}$$

$$\frac{h_0}{a} = 0.031 \frac{\gamma E}{\mu (u_s + E)^2} \phi_0^2 = 0.031 C a_0 C a^{-2} \phi_0^2$$
(22)

When  $u_s = 0$ , eq 22 yields

$$\frac{h_0}{a} = 0.031 C a_0^{-1} \phi_0^{\ 2} \tag{23}$$

#### 3. EXPERIMENTAL METHODS

A model system of monodisperse silica particles in alcohol was used to compare the modeling results to the experiments. While keeping the substrate fixed,  $l_0$  was measured under various coating conditions with different  $\phi_0$  and E.

Monodisperse silica particles of 200 nm  $\pm$  7% in diameter were synthesized using the Stober method.<sup>29</sup> Figure 3 shows

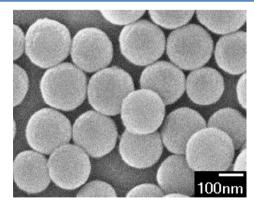


Figure 3. SEM image of the silica particles.

the SEM image of the particles. The particles were suspended in alcohol. Eighteen milliliters of coating solution were added in a 30 mm  $\times$  30 mm  $\times$  30 mm glass cell and dried in a sealed box at 22 °C. Sealed boxes of different sizes, 8 L, 40 L, and 150 L, were used to control the evaporation rate. Because the experiments were performed only for  $u_s = 0$ , the inner wall of the glass cell could be substituted for the substrate. While the particles were being coated on the inner wall of the glass cell,  $l_0$  was measured by taking the images of the wall using a CCD

camera. Because the solution concentration in the glass cell increases as drying proceeds,  $l_0$  was measured when the free surface descended to an equal length of 5 mm. E was determined by measuring the level of the free surface from the images.

Figure 4 shows the image of the inner wall of the glass cell with a black background. The close-packed region looks

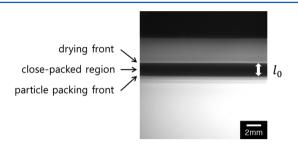


Figure 4. Image of the close-packed region.

substantially transparent because it is a thin film with solvent filled in the interparticle void space. A dried film (white color) is formed above the close-packed region. Below the close-packed region, the coating solution in the glass cell looks white. So the black area between the two white areas is the close-packed region. The upper boundary of the black area is the drying front, and the lower boundary is the particle packing front.

## 4. RESULTS AND DISCUSSION

Modeling Results. The dimensionless length of the closepacked region and the dimensionless coating thickness are the functions of the three dimensionless variables: Ca,  $Ca_0$ , and  $\phi_0$ . Comparing eq 22 and eq 23 with eq 2, the dimensionless coating thicknesses of the convective assembly and the Landau-Levich deposition have the same form, but are different only in the values of the exponents of the dimensionless variables. In the Landau-Levich equation, the dimensionless coating thickness is proportional to  $Ca^{2/3}$  and  $\phi_0$ . In the convective assembly, the dimensionless coating thickness is proportional to  $Ca_0^{-1}Ca^{-2}$  for the moving substrate, eq 22, or  $Ca_0^{-1}$  for the static substrate, eq 23, and  $\phi_0^2$ . The Landau– Levich equation is applied when the close-packed region cannot be developed and the viscous stress mainly affects the dip coating process because Ca is large. The coating thickness of the Landau-Levich deposition increases as Ca increases. On the other hand, the convective assembly is applied when the close-packed region is formed on the substrate near the meniscus at the early stage of drying because Ca is small. As Ca increases, the coating thickness of the convective assembly decreases because  $l_0$  is decreased due to the increase in the pressure drop or the decrease in the maximum capillary pressure in the close-packed region.

Comparing eq 22 with eq 9 or Dimitrov and Nagayama's result (eq 10),  $l_0$ , which is a constant in eq 9 and eq 10, is replaced by  $l_0$  from eq 21. Additionally,  $h_0$  now depends not only on the coating conditions, E,  $u_s$ , and  $\phi_0$ , but also on the material properties of the suspension, a,  $\gamma$ , and  $\mu$ .

In eq  $\bar{2}2$ ,  $\bar{h}_0$  is shown as a function of the material properties  $(a, \gamma, \mu)$  and the coating conditions  $(E, u_s, \phi_0)$ . When  $u_s < E$  or  $u_s = 0$ ,  $l_0$  decreases with the increase in the pressure drop in the close-packed region as  $\mu$ , E, and  $u_s$  increase, and thus,  $h_0$  decreases. When  $u_s > E$ ,  $h_0$  decreases as  $\mu$  and  $u_s$  increase, but  $h_0$ 

increases with E according to eq 22. This is because when  $u_s > E$ , the pressure drop is predominantly determined by  $u_s$  rather than E, and an increase in E increases the amount of the evaporation and enhances the particle packing at the particle packing front. As  $\gamma$  increases, the maximum capillary pressure increases, and  $l_0$  as well as  $h_0$  increases. As a increases,  $l_0$  increases because the pressure drop is reduced by  $1/a^2$  while the maximum capillary pressure is reduced by 1/a. Therefore,  $h_0$  increases. As  $\phi_0$  increases, both  $l_0$  and  $h_0$  increase because more particles accumulate at the particle packing front.

**Coating Process Regime Maps.** Using the modeling result for the static substrate, eq 23, a coating process regime map that predicts the coating thickness in terms of the dimensionless variables,  $\phi_0$  and  $Ca_0$ , was created, as shown in Figure 5. The coating thickness of the film formed by the

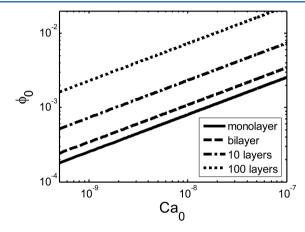
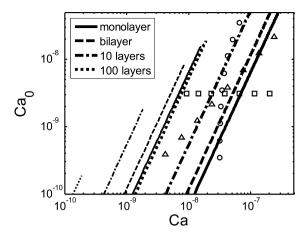


Figure 5. Coating process regime map for the static substrate.

convective assembly is discontinuous because the film has a structure consisting of the stacking of layers of hexagonally close-packed particles. Thus, the coating thickness was expressed as the number of layers, n, instead of the dimensionless coating thickness in the coating process regime map. The film has a monolayer when  $(h_0/a) = 2$ , a bilayer when  $(h_0/a) = 3.633$ , and n layers when  $(h_0/a) = 2 + (2\sqrt{6}/3)(n - 1)$ 1). When n is an integer, the film with uniform coating thickness of n layers is formed because the growth rate of the film balances the difference in the velocities of substrate withdrawal and the descent of the surface of coating solution.<sup>5</sup> However, when n is not an integer, a difference between the growth rate of the film and the velocity of substrate withdrawal relative to the descent of the surface exists. The meniscus between the surface of coating solution and the particle packing front changes with time, so uneven film<sup>22</sup> with average coating thickness of n layers or stripe pattern<sup>30,31</sup> appears. However, the mechanism of the film formation is not fully understood yet. The different shapes of the lines represent the coating conditions for the monolayer, bilayer, 10 layers, and 100 layers, respectively. The slopes of the lines are 1/2 because the dimensionless coating thickness is proportional to  $\phi_0^2$  and  $Ca_0^{-1}$ . As  $\phi_0$  increases or  $Ca_0$  decreases,  $h_0$  increases.

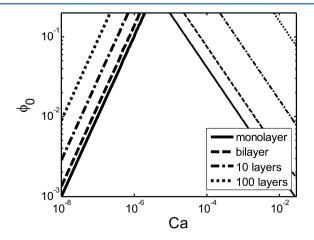
In eq 22, the dimensionless coating thickness is expressed as the function of the three dimensionless variables, Ca,  $Ca_0$ , and  $\phi_0$ . Because  $Ca_0$  and Ca are not independent, a coating process regime map for the moving substrate was created in terms of  $Ca_0$  and Ca while changing  $\phi_0$  shown in Figure 6. The thick lines represent the coating conditions for  $\phi_0 = 0.01$  and the thin lines for  $\phi_0 = 0.001$ . The different shapes of the lines represent



**Figure 6.** Coating process regime map for the moving substrate:  $\phi_0 = 0.01$  (thick lines),  $\phi_0 = 0.001$  (thin lines), the effect of E on  $h_0$  ( $\bigcirc$ ), the effect of  $\gamma$  or  $\mu$  on  $h_0$  ( $\triangle$ ), the effect of  $u_s$  on  $h_0$  ( $\square$ ).

the coating conditions for the monolayer, bilayer, 10 layers, and 100 layers. In the upper left corner of the map where Ca is smaller than  $Ca_0$ , the map was not drawn because  $u_s < 0$ . The slopes of the lines are 2 because the dimensionless coating thickness is proportional to  $Ca_0$  and  $Ca^{-2}$ . If  $\phi_0$  is reduced, Caneeds to be reduced in order to form a film of the same coating thickness at a constant  $Ca_0$ . Therefore, the regime for  $\phi_0$  = 0.001 lies on the left-hand side of the regime for  $\phi_0$  = 0.01, and  $h_0$  increases as  $\phi_0$  increases for the same  $Ca_0$  and Ca. Circle markers show the effect of E on  $h_0$ . The value of E increases as  $Ca_0$  and Ca increase. As E increases,  $h_0$  increases when  $u_s > E$ , but  $h_0$  decreases when  $u_s < E$ . Triangle markers show the effect of  $\gamma$  or  $\mu$  on  $h_0$ . The value of  $\gamma$  decreases or the value of  $\mu$ increases as  $Ca_0$  and Ca increase. As  $\gamma$  decreases or  $\mu$  increases,  $h_0$  decreases. Square markers show the effect of  $u_s$  on  $h_0$ . The value of  $u_s$  increases as Ca increases, and  $h_0$  decreases as  $u_s$ increases.

Using eq 2 for the Landau–Levich deposition and eq 22 for the convective assembly, a coating process regime map in terms of the dimensionless variables,  $\phi_0$  and Ca, with a constant  $Ca_0$  was created, as shown in Figure 7. The thick lines represent the coating conditions of the films being formed by the convective assembly and the thin lines by the Landau–Levich deposition. The different shapes of the lines represent the coating



**Figure 7.** Coating process map for the moving substrate when  $Ca_0$  is fixed: Convective assembly (thick lines), Landau–Levich deposition (thin lines).

conditions for the monolayer, bilayer, 10 layers and 100 layers.  $Ca_0$  was fixed at  $1 \times 10^{-9}$ .  $h_0$  in eq 22 is nondimensionalized by a, and the dimensionless coating thickness is independent of a. However,  $h_{LL,dry}$  in eq 2 is nondimensionalized by  $l_c$  and if it is nondimensionalized by a in order to fit in the same graph with eq 22, the dimensionless coating thickness will be inversely proportional to a. Therefore, a was fixed at 100 nm only for eq 2. The coating thickness of the Landau-Levich deposition for an arbitrary a can be obtained by multiplying the coating thickness predicted by the coating process regime map by 100 nm/a. If a = 200 nm and the coating condition is located in the convective assembly regime (upper left part of the map), we can use the coating thickness predicted by the coating process regime map. However, if the coating condition is located in the Landau-Levich deposition regime (upper right part of the map), we need to multiply the coating thickness predicted by the coating process regime map by (100 nm/200 nm) = 0.5. The slopes of the thick lines in the convective assembly regime are 1 because the coating thickness is proportional to  $\phi_0^2$  and  $Ca^{-2}$ , and the slopes of the thin lines in the Landau–Levich deposition regime are -2/3 because the coating thickness is proportional to  $\phi_0$  and  $Ca^{2/3}$ . At a small Ca, the film is formed by the convective assembly with the close-packed lattice structure, and the coating thickness decreases with Ca. If Ca continues to increase, the coating condition falls into the Landau-Levich deposition regime. At the onset of the Landau-Levich deposition regime, where a transition between the convective assembly and the Landau-Levich deposition is realized, the close-packed monolayer can be developed.<sup>22</sup> However, except at this transition, the film does not form a well-ordered structure as in the convective assembly, and the coating thickness increases as Ca increases. As  $\phi_0$  increases, the coating thickness increases both in the convective assembly regime and in the Landau-Levich deposition regime. Using this coating process regime map, one can predict the coating mechanism (convective assembly or Landau-Levich deposition), the degree of particle alignment (close-packed lattice structure or less ordered structure) and the coating thickness according to  $\phi_0$  and Ca under certain dip coating process conditions.

## **■ EXPERIMENTAL SECTION**

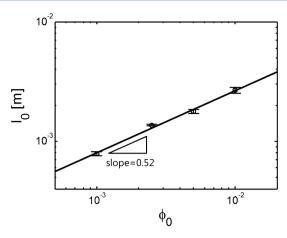
When  $u_{\rm s}=0$ ,  $l_0$  was measured under the various coating conditions with different  $\phi_0$  and E using a model system of monodisperse silica particles in alcohol. Table 1 shows the experimentally explored coating

Table 1. Experimentally Explored Coating Conditions

$\phi_0$	<i>E</i> [m/s]	$Ca_0$	$l_0$ [m]
0.001	$9.7 \times 10^{-8}$	$5.2 \times 10^{-9}$	$0.8 \times 10^{-3}$
0.0025	$9.7 \times 10^{-8}$	$5.2 \times 10^{-9}$	$1.4 \times 10^{-3}$
0.005	$9.7 \times 10^{-8}$	$5.2 \times 10^{-9}$	$1.8 \times 10^{-3}$
0.01	$9.7 \times 10^{-8}$	$5.2 \times 10^{-9}$	$2.7 \times 10^{-3}$
0.0025	$3.6 \times 10^{-8}$	$1.9 \times 10^{-9}$	$2.9 \times 10^{-3}$
0.0025	$6.1 \times 10^{-8}$	$3.3 \times 10^{-9}$	$2.0 \times 10^{-3}$
0.0025	$9.7 \times 10^{-8}$	$5.2 \times 10^{-9}$	$1.4 \times 10^{-3}$

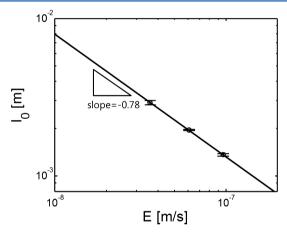
conditions and the average values of the measured  $l_0$ . In the first four coating conditions  $l_0$  increased with  $\phi_0$ , and in the next,  $l_0$  decreased with E.

Figure 8 shows the experimental results of the dependence of  $l_0$  on  $\phi_0$ . The error bars represent the standard deviations of  $l_0$ . The slope of the graph is 0.52 and  $l_0$  was found to be proportional to  $\phi_0^{0.52}$ . According to eq 21,  $l_0$  is proportional to  $\phi_0$  when  $u_s = 0$ . Both the



**Figure 8.** Dependence of the length of the close-packed region on the initial volume fraction.

experiments and modeling showed the trend of  $l_0$  increasing with the increase in  $\phi_0$ . However, in the experiments, the exponent of  $\phi_0$  was measured to be smaller than the modeling. Figure 9 shows the



**Figure 9.** Dependence of the length of the close-packed region on the evaporation rate.

experimental results on the effects of E. The slope of the graph is -0.78 and  $l_0$  was proportional to  $E^{-0.78}$ . In eq 21 of the modeling,  $l_0$  is proportional to  $E^{-1}$  when  $u_s=0$ . Both the experiments and modeling showed the trend of  $l_0$  decreasing with the increase in E. However, in the experiments, the absolute value of the exponent of E was measured to be smaller than the model prediction. In the experiments on the effects of  $\phi_0$  and E, the absolute values of the exponents of both variables were measured to be smaller than the modeling. In addition, the difference between the experiments and the modeling was larger for the dependence on  $\phi_0$  than that on E.

One of the possible causes for the differences is that  $l_0$  could be measured to be larger than the actual length in the experiments. In Figure 2, the particle packing front is located slightly above the surface of the coating solution due to the meniscus being formed in the position where the surface of the coating solution and the substrate meet.  $l_0$  was determined experimentally by measuring the length of the transparent area between the two white areas shown in Figure 4 without observing the particles in the close-packed region directly. Therefore,  $l_0$  could be measured larger than the actual length by including part of the meniscus. If  $l_0$  was assumed to be measured  $0.2 \times 10^{-3}$  m larger,  $l_0$  could be recalculated to be proportional to  $\phi_0^{0.61}$  and  $E^{-0.87}$ , which becomes more consistent with the modeling results. Furthermore, if we assume that  $l_0$  was measured  $0.43 \times 10^{-3}$  m larger, the experiments and the modeling would coincide on the effect of E on  $l_0$ . It is possible that a greater part of the meniscus was included in

determining  $l_0$  for the coating solution with a low  $\phi_0$  because the coating solution with a low  $\phi_0$  is more transparent than a high  $\phi_0$ . Therefore, the effect of  $\phi_0$  on  $l_0$  could be measured to be much smaller.

If the pressure drop is proportional to the nth power of the flow rate (n less than 1), the modeling can predict the experimental results more closely. Darcy's law that was used to determine  $l_0$  in the modeling describes the pressure drop when the Newtonian fluid goes through the packed bed composed of the monodisperse hard spheres. Because the model system used in the experiments was not a perfect monodisperse hard sphere, by assuming that the pressure drop in the close-packed region was proportional to the nth power of the flow rate,  $l_0$  could be rederived. When  $u_s = 0$ , Darcy's law was modified according to the assumptions above.

$$\frac{\partial p}{\partial x} = -\frac{\mu}{k_{p,n}(a)} (1 - \phi_m)^n \overline{u}_x^n \tag{24}$$

where  $k_{p,n}$  is the permeability of the modified Darcy's law. Because  $k_p$  from eq 15 is the function of a,  $k_{p,n}$  was also assumed to be the function of a. The dependence of the permeability on  $\phi_m$  was not considered because  $\phi_m$  is a constant. By substituting  $u_s = 0$  into eq 17,  $l_p = l_0$  was recovered. By equating the amount of the pressure drop calculated by integrating eq 24 from  $x = x_0$  to  $x = x_0 + l_0$  after utilizing  $\overline{u}_x(x)$  from eq 14 and the maximum capillary pressure, the following equation was obtained:

$$-\int_{x_0}^{x_0+l_0} -\frac{\mu(1-\phi_m)^n}{k_{p,n}(a)} \left\{ \frac{E(\phi_m-\phi_0)}{(1-\phi_m)\phi_0} \left[ 1 - \frac{(x-x_0)}{l_0} \right] \right\}^n dx$$

$$= \frac{10\gamma}{a}$$
(25)

After solving  $l_0$  in eq 25 and assuming  $\phi_0$  is small,  $l_0$  was derived as

$$l_0 = \frac{\gamma f_n(a)}{\mu} \frac{\phi_0^n}{E^n} \tag{26}$$

where  $f_n(a)=(10(n+1)/a\phi_m{}^n)k_{p,n}(a)$ . Inserting  $l_0$  from eq 26 into eq 9,  $h_0$  was obtained.

$$h_0 = \frac{1.35 \gamma f_n(a)}{\mu} \frac{\phi_0^{n+1}}{E^n}$$
 (27)

In eq 26,  $l_0$  is proportional to  $\phi_0^n$  and  $E^{-n}$ . If n = 0.52, the modeling results can explain the experimental results of the effect of  $\phi_0$ . Also, if n = 0.78, the modeling can describe well the experimental results of the effect of E.

# 5. CONCLUSIONS

In this work, the length of the close-packed region and the coating thickness were predicted by combining Darcy's law with the particle and solvent balance equations in the closepacked region. The dimensionless length of the close-packed region and the dimensionless coating thickness were found to be the functions of only three dimensionless variables: Ca, Ca<sub>0</sub>, and  $\phi_0$ . From the modeling results, coating process regime maps that predict the dimensionless coating thickness in terms of the dimensionless variables were created. By using the coating process regime map which combines the Landau-Levich deposition regime altogether, one can predict the mechanism (convective assembly or Landau-Levich deposition), the degree of particle alignment and the coating thickness in terms of Ca and  $\phi_0$  under an arbitrary dip coating process. In addition, the effects of the initial volume fraction and the evaporation rate on the length of the close-packed region were observed experimentally to validate the model prediction. Both the experiments and modeling showed the same tendency that the length of the close-packed region increased with an increase

in the initial volume fraction or a decrease in the evaporation rate. However, the measured absolute values of the exponents were found to be smaller than the model prediction. The discrepancy may be attributed to the inclusion of the meniscus during the measurement of the length of the close-packed region. When the pressure drop was assumed to be proportional to the *n*th power of the flow rate, the modeling could fit the experimental results.

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#### Notes

The authors declare no competing financial interest.

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