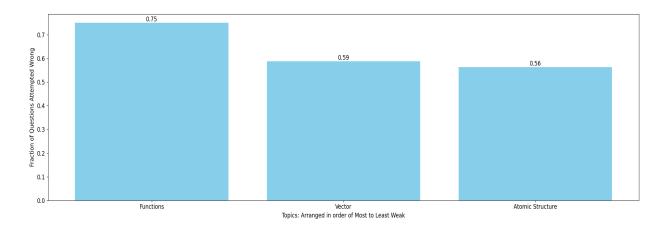
Kalikant Tripathi Total MLAssist - Personalised DPP

Question Paper Analysis:



Weak Topic Analysis:



Practice Questions:

Functions:

Let $f: \left(-\frac{\pi}{2}, \frac{\pi}{2}\right) \to R$ be given by $f(x) = (\log(\sec x + \tan x))^3$. Then, [JEE Ad. 2014] 5.

(A) f(x) is an odd function

(B) f(x) is a one-one function

(C) f(x) is an onto function

(D) f(x) is an even function

Let $f(x) = \frac{3}{2} + \sqrt{x - \frac{7}{4}}$ and g(x) be the inverse function of f(x) then the value of $(f^{-1}og^{-1})(17)$ 6.

is equal to

(A) $\frac{3+\sqrt{61}}{3}$

(B) 242 (C) 17 (D) $\frac{3-\sqrt{61}}{2}$

If the function f: $\mathbf{R} - \{1, -1\} \to \mathbf{A}$ defined by $f(x) = \frac{\mathbf{A}}{1 - x^2}$, is surjective, then A is equal to 16.

[JEE - Main 2019]

(A) $\mathbf{R} - \{-1\}$ (B) $[0, \infty)$ (C) $\mathbf{R} - [-1, 0)$ (D) $\mathbf{R} - (-1, 0)$

Let $f: [0,a] \to S$ be a function defined by $f(x) = 3\cos\frac{x}{x}$. If the largest value of a for which f(x) has 4.

The value of (a + b) is equal to 6.

(A) -2

(C) 0

(D) 1

Vector:

Let $\vec{a} = a_1 \hat{i} + a_2 \hat{j} + a_3 \hat{k}$; $\vec{b} = b_1 \hat{i} + b_2 \hat{j} + b_3 \hat{k}$; $\vec{c} = c_1 \hat{i} + c_2 \hat{j} + c_3 \hat{k}$ be three non-zero vectors such 49. that \vec{c} is a unit vector perpendicular to both $\vec{a} \& \vec{b}$. If the angle between $\vec{a} \& \vec{b}$ is $\frac{\pi}{6}$, then

$$\begin{vmatrix} a_1 & b_1 & c_1 \\ a_2 & b_2 & c_2 \\ a_3 & b_3 & c_3 \end{vmatrix}^2 =$$

(A) 0

- (B) 1
- (C) $\frac{1}{4}(a_1^2 + a_2^2 + a_3^2)(b_1^2 + b_2^2 + b_3^2)$ (D) $\frac{3}{4}(a_1^2 + a_2^2 + a_3^2)(b_1^2 + b_2^2 + b_3^2)(c_1^2 + c_2^2 + c_3^2)$
- 42. If the vectors.

If $\vec{a} = \hat{i} + \hat{j} + \hat{k}$ and $\vec{b} = \hat{i} - 2\hat{j} + \hat{k}$, then the vector \vec{c} such that $\vec{a} \cdot \vec{c} = 2 \& \vec{a} \times \vec{c} = \vec{b}$ is -61.

$$\text{(A)} \ \frac{1}{3} \Big(3 \hat{i} - 2 \hat{j} + 5 \hat{k} \Big) \qquad \text{(B)} \ \frac{1}{3} \Big(- \hat{i} + 2 \hat{j} + 5 \hat{k} \Big) \qquad \text{(C)} \ \frac{1}{3} \Big(\hat{i} + 2 \hat{j} - 5 \hat{k} \Big) \qquad \text{(D)} \ \frac{1}{3} \Big(3 \hat{i} + 2 \hat{j} + \hat{k} \Big)$$

22. Which of the following statement(s) is/are true in respect of the lines

$$\vec{r} = \vec{a} + \lambda \vec{b}$$
; $\vec{r} = \vec{c} + \mu d$ where $\vec{b} \times \vec{d} \neq 0$

- (A) acute angle between the lines is $\cos^{-1} \left(\frac{|\mathbf{b} \cdot \mathbf{d}|}{|\mathbf{b}||\mathbf{d}|} \right)$
- (B) The lines would intersect if [c bd] = [a bd]
- (C) The lines will be skew if [c̄-ā b̄ d̄]≠0
- (D) If the lines intersect at $\vec{r} = \vec{r}_0$, then the equation of the plane containing the lines is $[\vec{r} - \vec{r}_0 \ \vec{b} \ \vec{d}] = 0$

20.	Let A(1, 2	3) B(0, 0,	 C(-1, 1, 1)) are the vertices of a ΔABC.
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(i) The equation of internal angle bisector through A to side BC is

(A)
$$\vec{r} = \hat{i} + 2\hat{j} + 3\hat{k} + \mu (3\hat{i} + 2\hat{j} + 3\hat{k})$$

(B)
$$\vec{r} = (\hat{i} + 2\hat{j} + 3\hat{k}) + \mu(3\hat{i} + 4\hat{j} + 3\hat{k})$$

(C)
$$\bar{r} = \hat{i} + 2\hat{j} + 3\hat{k} + \mu (3\hat{i} + 3\hat{j} + 2\hat{k})$$

(D)
$$\vec{r} = \hat{i} + 2\hat{j} + 3\hat{k} + \mu (3\hat{i} + 3\hat{j} + 4\hat{k})$$

(ii) The equation of median through C to side AB is

(A)
$$\bar{r} = -\hat{i} + \hat{j} + \hat{k} + p (3\hat{i} - 2\hat{k})$$

(B)
$$\vec{r} = -\hat{i} + \hat{j} + \hat{k} + p (3\hat{i} + 2\hat{k})$$

(C)
$$\vec{r} = -\hat{i} + \hat{j} + \hat{k} + p(-3\hat{i} + 2\hat{k})$$

(D)
$$\vec{r} = -\hat{i} + \hat{j} + \hat{k} + p (3\hat{i} + 2\hat{j})$$

(iii) The area (ΔABC) is equal to

(A)
$$\frac{9}{2}$$

(B)
$$\frac{\sqrt{17}}{2}$$

(C)
$$\frac{17}{2}$$

(D)
$$\frac{7}{2}$$

Atomic Structure:

19. For He+ ion, the only INCORRECT combination is

48. In the sixth period, the orbitals that are filled are:

[JEE Main (April) 2020]

The ionziation enthalpy of hydrogen atom is 1.312 x 106 J mol-1. The energy required to excite 13. the electron in the atom from n = 1 to n = 2 is [AIEEE-2008]

(1)
$$8.51 \times 10^5 \text{ J mol}^{-1}$$
 (2) $6.56 \times 10^5 \text{ J mol}^{-1}$ (3) $7.56 \times 10^5 \text{ J mol}^{-1}$ (4) $9.84 \times 10^5 \text{ J mol}^{-1}$

18. The binding energy of e- in ground state of hydrogen atom is 13.6 eV. The energies required to eject out an electron from three lowest states of He* ion will be - (in eV)

52. A ball weighing 10 g is moving with a velocity of 90 ms⁻¹. If the uncertainty in its velocity is 5%, then the uncertainty in its position is_____ × 10⁻¹m. (Rounded off to the nearest integer) [Given: h = 6.63 × 10⁻¹Js] [JEE Main (April) 2021]

Ans. 1