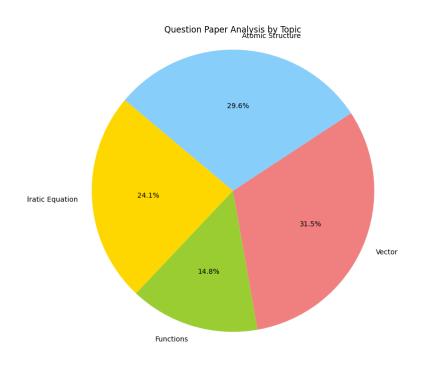
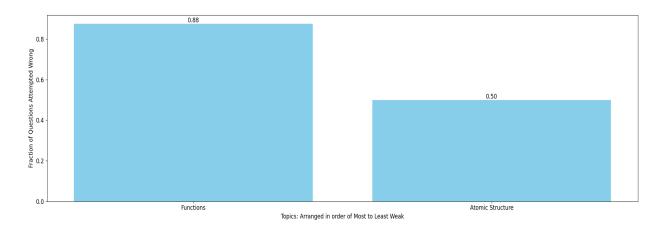
# Shiva gahlod Total MLAssist - Personalised DPP

# **Question Paper Analysis:**



# Weak Topic Analysis:



### **Practice Questions:**

#### **Functions:**

8. Let f(x) = (x + 1)(x + 2)(x + 3)(x + 4) + 5 where x ∈ [-6,6]. If the range of the function is [a, b] where a, b ∈ N then find the value of (a + b).

- 8. If  $F(n+1) = \frac{2\Gamma(11)+1}{2}$ ,  $n = 2, \dots ... \cdot 8F(1) = 2$  then  $\frac{\Gamma(101)}{26}$  equals
- 8. Find the formula for the function fogoh, given  $f(x) = \frac{x}{x+1}$ ;  $g(x) = x^{10}$  and h(x) = x + 3. Find also the domain of this function. Also compute (fogoh)(-1).
- 1. Find the domains of definitions of the following functions:

(Read the symbols [\*] and { \*} as greatest integers and fractional part functions respectively.)

(i) 
$$f(x) = \sqrt{\cos 2x} + \sqrt{16 - x^2}$$

(ii) 
$$f(x) = \log_7 \log_5 \log_3 \log_2 (2x^3 + 5x^2 - 14x)$$

(iii) 
$$f(x) = \ln (\sqrt{x^2 - 5x - 24} - x - 2)$$

(iv) 
$$f(x) = \sqrt{\frac{1-5^x}{7^{-x}-7}}$$

(v) 
$$y = \log_{10} \sin(x-3) + \sqrt{16-x^2}$$

(vi) 
$$f(x) = \log_{100x} \left( \frac{2\log_{10} x+1}{-x} \right)$$

(vii) 
$$f(x) = \sqrt{x^2 - |x|} + \frac{1}{\sqrt{9-x^2}}$$

(viii) 
$$f(x) = \sqrt{(x^2 - 3x - 10) \cdot \ln^2(x - 3)}$$
 (ix)  $f(x) = \sqrt{(5x - 6 - x^2)[\{\ln \{x\}\}]} + \frac{1}{2} \left[ \frac{1}{2}$ 

$$\sqrt{(7x-5-2x^2)} + \left(\ln\left(\frac{7}{2}-x\right)\right)^{-1}$$

(x) 
$$f(x) = \log_{\left[x + \frac{1}{x}\right]} |x^2 - x - 6| + {}^{16-x}C_{2x-1} + {}^{20-3x}P_{2x-5}$$

Let f(x) be a polynomial of degree 3 such that  $f(x) = -\frac{2}{L}$  for k = 2, 3, 4, 5. Then the value of 52 – 32. 10 f(10) is equal to \_\_\_\_\_. [JEE - Main 2021]

### **Atomic Structure:**

The binding energy of e- in ground state of hydrogen atom is 13.6 eV. The energies required to 18. eject out an electron from three lowest states of He\* ion will be - (in eV)

(A) 13.6, 10.2, 3.4

(B) 13.6, 3.4, 1.5

(C) 13.6, 27.2, 40.8 (D) 54.4, 13.6, 6

The ionziation enthalpy of hydrogen atom is 1.312 x 106 J mol-1. The energy required to excite 13. the electron in the atom from n = 1 to n = 2 is

(1) 8.51 x 10<sup>5</sup> J mol<sup>-1</sup> (2) 6.56 x 10<sup>5</sup> J mol<sup>-1</sup> (3) 7.56 x 10<sup>5</sup> J mol<sup>-1</sup> (4) 9.84 x 10<sup>5</sup> J mol<sup>-1</sup>

The difference between the radii of 3rd and 4th orbits of Li2+ is ΔR1. The difference between the 49. radii of  $3^{rd}$  and  $4^{th}$  orbits of  $He^+$  is  $\Delta R_2$ . Ratio  $\Delta R_1$ :  $\Delta R_2$  is: [JEE Main (April) 2020]

(1) 8:3

(2)3:8

(3) 3:2

(4)2:3

For an electron, with n = 3 has only one radial node. The orbital angular momentum of the 64. electron will be

(A) 0

(B)  $\sqrt{6} \frac{h}{2\pi}$  (C)  $\sqrt{2} \frac{h}{2\pi}$  (D)  $3 \left(\frac{h}{2\pi}\right)$ 

35. Which of the graphs shown below does not represent the relationship between incident light and the electron ejected from metal surface? [JEE Main (Jan.) 2019]

