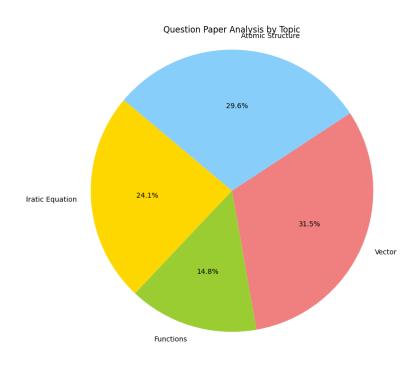
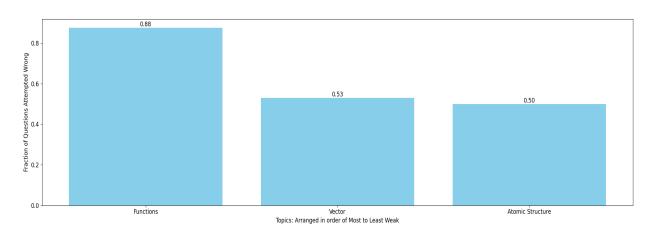
Anubhab Ray Total MLAssist - Personalised DPP

Question Paper Analysis:



Weak Topic Analysis:



Practice Questions:

Functions:

- If range of $f(x) = \frac{2\sin x + 2\sin x + 3}{\sin^2 x + \sin x + 1}$ is [p, q] then 6p 3 q equals 7. 2F(n)+1F(101)
- Let $f(x) = x^2$ and $g(x) = \sin x$ for all $x \in R$. Then the set of all x satisfying 1. (fogogof)(x) = (gogof)(x), where (fog)(x) = f(g(x)), is-[JEE 2011]
 - (A) $\pm \sqrt{n\pi}$, n ∈ {0,1,2,
- (B) $\pm \sqrt{n\pi}$, $n \in \{1, 2, ...\}$
- (C) $\frac{\pi}{2} + 2n\pi$, $n \in \{..., -2, -1,0,1,2,...\}$ (D) $2n\pi$, $n \in \{..., -2, -1,0,1,2,...\}$
- Let $P(x) = x^4 + ax^3 + bx^2 + cx + d$, where $a, b, c, d \in R$. Suppose P(0) = 6, P(1) = 7, P(2) = 88. and P(3) = 9, then find the value of P(4).
- Let $f(x) = \left| \frac{1}{\cos \{x\}} \right|$ where [y] and {y} denote greatest integer and fractional part functions 9. respectively and $g(x) = 2x^2 - 3x(k+1) + k(3k+1)$. If $g(f(x)) < 0 \forall x \in R$ then find the number of integral values of k.

For $x \in \left(0, \frac{3}{2}\right)$, let $f(x) = \sqrt{x}$, $g(x) = \tan x$ and $h(x) = \frac{1-x}{1+x^2}$ If $\varphi(x) = (hof) \circ g(x)$, then $\varphi\left(\frac{\pi}{3}\right)$ is 11. [JEE - Main 2019] equal to

(A) $\tan \frac{\pi}{12}$

(B) $\tan \frac{11\pi}{12}$ (C) $\tan \frac{7\pi}{12}$ (D) $\tan \frac{5\pi}{12}$

Vector:

- If the distance from the point P(1, 1, 1) to the line passing through the points Q(0, 6, 8) and 6. R(-1, 4, 7) is expressed in the form $\sqrt{p/q}$ where p and q are coprime, then the value of $\frac{(p+q)(p+q-1)}{2}$.
- In a \triangle ABC, points E and F divide sides AC and AB respectively so that $\frac{AE}{EC} = 4$ and $\frac{AF}{EB} = 1$. 4. Suppose D is a point on side BC. Let G be the intersection of EF and AD and suppose D is situated so that $\frac{AG}{GD} = \frac{3}{2}$. If the ratio $\frac{BD}{DC} = \frac{a}{b}$, where a and b are in their lowest form, find the value of (a + b).
- 59. Position vectors of the four angular points of a tetrahedron ABCD are A(3, -2, 1); B(3, 1, 5); C(4, 0, 3) and D(1, 0, 0). Acute angle between the plane faces ADC and ABC is (A) $tan^{-1}(5/2)$ (B) $cos^{-1}(2/5)$ (C) $cosec^{-1}(5/2)$ (D) $cot^{-1}(3/2)$
- The pv's of the four angular points of a tetrahedron are $A(\hat{j}+2\hat{k})$; $B(3\hat{i}+\hat{k})$; $C(4\hat{i}+3\hat{j}+6\hat{k})$ & 22. $D(2\hat{i}+3\hat{j}+2\hat{k})$. Find:
 - the perpendicular distance from A to the line BC.
 - the volume of the tetrahedron ABCD.
 - (iii) the perpendicular distance from D to the plane ABC.
 - (iv) the shortest distance between the lines AB & CD.
- The acute angle between the medians drawn from the acute angles of an isosceles right angled 12. triangle is:

(A) cos⁻¹(2/3)

(B) cos⁻¹(3/4) (C) cos⁻¹ (4/5)

(D) none

Atomic Structure:

20.

Column-I

- (A) Electron moving in 2nd orbit in He+ ion
- (P) Radius of orbit in which electron is moving is 0.529 Å
- (B) Electron moving in 3rd orbit in H-atom
- (Q) Total energy of electron is (-)13.6 × 9eV
- (C) Electron moving in 1st orbit in Li+2 ion
- (R) Velocity of electron is

Column-I

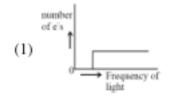
- 40. If the de Broglie wavelength of the electron in nth Bohr orbit in a hydrogenic atom is equal to
- 52. A ball weighing 10 g is moving with a velocity of 90 ms. If the uncertainty in its velocity is 5%, then the uncertainty in its position is _____ × 10 mm. (Rounded off to the nearest integer)

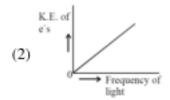
[Given: $h = 6.63 \times 10^{-1}$ Js]

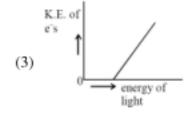
[JEE Main (April) 2021]

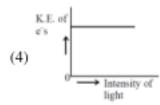
Ans. 1

35. Which of the graphs shown below does not represent the relationship between incident light and the electron ejected from metal surface? [JEE Main (Jan.) 2019]









- Given in hydrogenic atom r_n, V_n, E, K_n stand for radius, potential energy, total energy and kinetic energy in nth orbit. Find the value of U,v,x,y.

 [JEE 2006]
 - (A) $U = \frac{V_n}{K_n}$

(P) 1

 $(B) \qquad \frac{1}{r_n} \propto E^x$

(Q) -2

(C) $r_n \propto Z^y$

(R) −1

(Z = Atomic number)

- (D) v = (Orbital angular momentum of electron in its lowest energy)
- (S) 0

in its lowest energy)

