# A Summary of Analytical Separations

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## 1 High Performance Liquid Chromatography

### 1.1 Introduction

• HPLC stands for "High Performance Liquid Chromatography" or "High Pressure Liquid Chromatography"

#### Advantages

- Analysis of thermally unstable compounds
- Analysis of nonvolatile compounds

### • Major requirement of LC

- Solute solubility in mobile phase
- This is in contrast to GC which require solute volatility

## 1.2 Scope of HPLC

Adsorption chromatography (LSC)

Ion chromatography (IC)

Size-exclusion chromatography (SEC)

Partition chromatography separation of analytes by partitioning, most commonly to a stationary phase bonded to a solid support

Note that this replaces liquid-liquid chromatography with its problems of stripping of stationary phase

Hydrophobic interaction chromatography (HIC) for separation of proteins without denaturation

Hydrophilic interaction chromatography (HILIC) for separation of very polar analytes

Chiral chromatography

Affinity chromatography

## 1.3 Column Efficiency in HPLC

• Recall the van Deemter equation for GLC (commonly called GC) for packed columns:

 $H = A + \frac{B}{u} + Cu \tag{1}$ 

- For longitudinal diffusion in LC, as the  $D_l \approx 10^{-5} D_g$ , the peak broadening due to longitudinal diffusion in mobile phase (liquid) phase in LC is negligible.
- i.e.,  $\frac{B}{u} = \frac{2\gamma D_m}{u}$

## 1.4 Pumps

**High Pressure Pumps** required to force liquid through a densely packed column at constant flow rate, so that  $t_r$  is reproducible.

Requirements of a pump:

- output pressure all the way to 6000 psi, without any leaks; for UHPLC, the pressure is typically  $\approx 18\,000\,\mathrm{psi}$
- have variable flow rates, from  $0.1\,\mathrm{mL/min}$  to  $10\,\mathrm{mL/min}$ ; for UHPLC, the pressure can go all the way down to  $0.01\,\mathrm{mL/min}$
- have a pulse-free output
- have a reproducible flow rate (ours to  $\pm 0.3\%$ )

#### 1.4.1 Reciprocating Pumps

These are the most common today, replacing pneumatic and displacement (syringe) pumps.

A single piston reciprocating pump:

• an eccentric (off-center) can drives a piston back and forth

- its motion is synchronized with operation of check valves which control direction of flow
- Fill stroke: piston cavity fills, drawing solvent through the inlet check valve via suction; the outlet valve is closed
- delivery stroke: mobile phase forced through column as inlet check valve closes and the outlet valve opens. Pulsed flow produces baseline noise that must be damped
- electronic pulse compensation: speeds up piston during refill cycle, reducing the time the piston is not delivering solvent ( $\approx 200\,\mathrm{ms}$ ); this reduces the noisy baseline
- note that the pulse flow pulsates, but the time-averaged flow rate is constant
- the regular pressure surge spikes in the baseline may still be seen at extreme sensitive detector
- an elliptical cam minimize pulsations

#### 1.4.2 Dual Piston Reciprocating Pumps

- essentially 2 single-piston pumps driven by the same motor
- 2 pistons are 180° out of phase, i.e., they are synchronized to provide a continuous flow of mobile phase
- when one is in the fill stroke, the other is on deliver stroke

#### 1.4.3 HPLC solvents (Mobile Phase)

Solvents need to be HPLC or of similar high-grade for a stable baseline and to extend the life of the column, 'cause those things are expensive!

Filtration to remove particulate matter

**Degassing** to remove dissolved gases, especially  $N_2$  and  $O_2$ , by vacuum or He sparging

#### 1.4.4 Pressure Drop Across Column (Backpressure)

• Pressure developed when the liquid mobile phase is pumped through the (packed) HPLC column:

\*enter the equation for backpressure\*

- therefore, for much larger backpressure occurs with smaller particles, more tightly packed column and a more viscous mobile phase
- there is a tradeoff between better performance and higher pressure

### 1.5 Elution Techniques

- 1.6 Injectors
- 1.7 Columns
- 1.8 Detectors
- 1.9 Types of Chromatography in HPLC
- 2 GC-MS

## 3 Important Tips

**Isocratic elution** one solvent, or constant solvent mixture.

**Gradient elution** continuous change of solvent composition to increase eluent strength.

Gradient elution in HPLC is analoguous to temperature programming in gas chromatography.

Increased eluent strength is required to elute more strongly retained solutes.

General elution problem for a complex mixture, isocratic conditions can often be found to produce adequate separation of early-eluting peaks

or late-eluting peaks, but not both. This problem drives us to use gradient elution.

**Note:** Elution strength decreases as the solvent becomes more polar, correct???

**Separation factor,**  $\alpha$  Also called relative retention; for two components, 1 and 2, it is the ratio of their adjusted retention times.

The greater the relative retention, the greater the separation between two components.

Relative retention is fairly independent of flow rate and can therefore be used to to help identify peaks when the flow rate changes. (show equation?)