

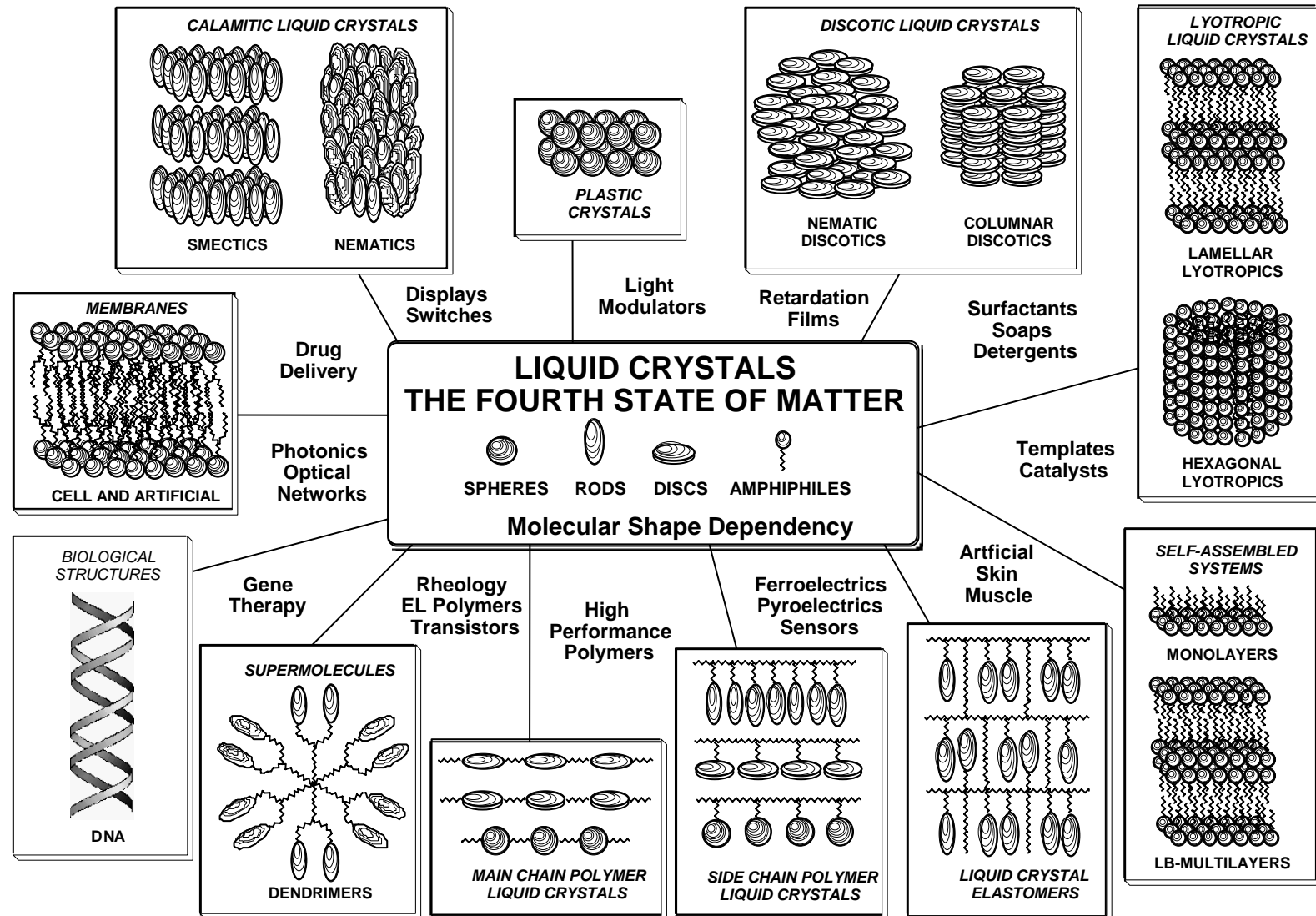
C3A - Display Technology

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Lecture 2

- LC materials
 - The range of LC phases
 - Chemical structures of LC materials
 - Mixtures
- Physical properties
 - Spontaneous ordering
 - Anisotropic properties
 - Elastic constants
- Principles of LCDs
 - Surface alignment
 - Alignment by electric field
 - Competition between surfaces and electric field

Self-organising Soft Matter

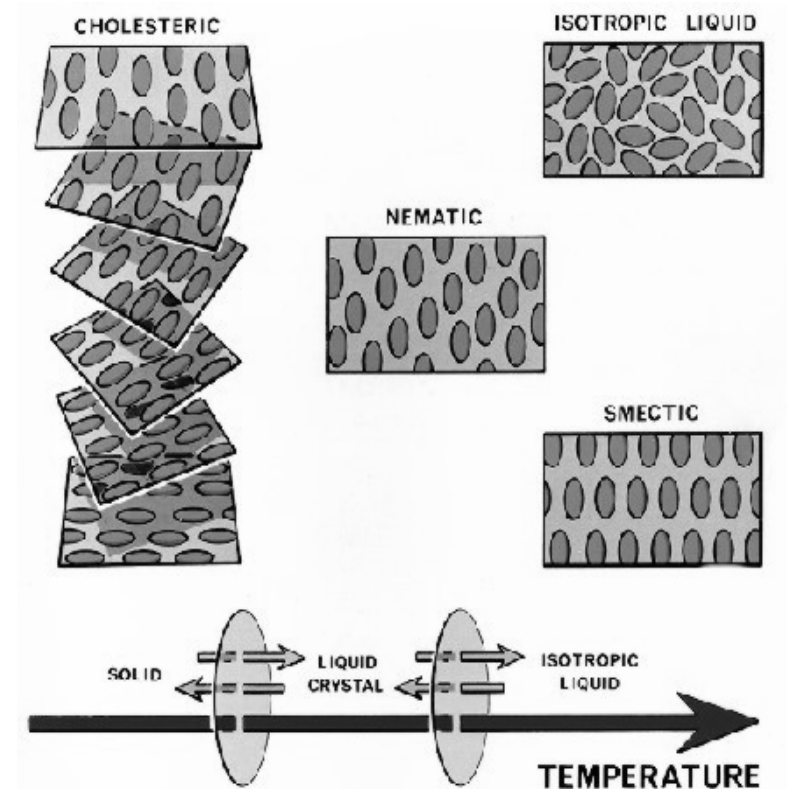


Liquid Crystal Phases

- Occurrence
 - Thermotropic (temperature)
 - Lyotropic (concentration)
- Constituent molecules
 - Small molecules
 - Big molecules (polymers)
 - Rods (calamitic)
 - Disks (discotic)

Thermotropic Calamitic LCs

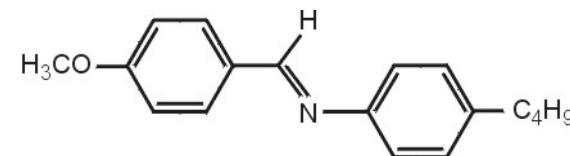
- Finite temperature range
- Isotropic (I)
- Nematic (N)
 - Molecular axes ordered
 - No positional order
- Chiral Nematic (N*)
 - (Cholesteric)
 - Helical rotation (pitch)
- Smectic (S_A , S_B etc)
 - Positional order
 - Many smectic phases



Typical LC Materials

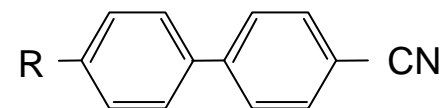
- MBBA

- First room temperature material
- 20 °C to 47 °C
- Totally unstable



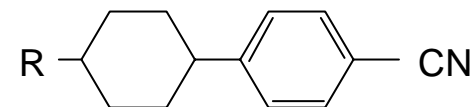
- Cyanobiphenyls (CB)

- First useful LC material
- Chemically & photochemically stable
- Colourless
- Excellent performance in displays
- LC at room temperature but narrow range
 - eg 5CB (R = C₅ H₁₁- : 23 °C to 35 °C)

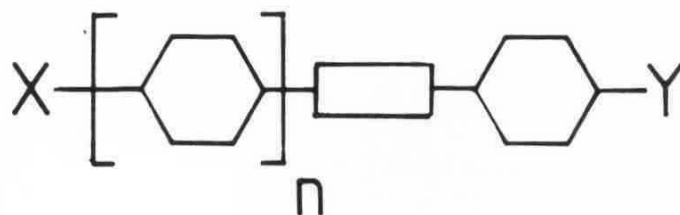


- Phenyl cyclohexanes (PCH)

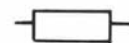
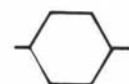
- Second useful LC
 - eg 5PCH (31 °C to 55 °C)



Summary of LC Chemistry



X, Y

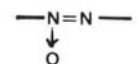


$C_m H_{2m+1}$

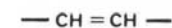
AROMATIC

ALIPHATIC

HETEROCYCLIC



$C_m H_{2m+1} O$

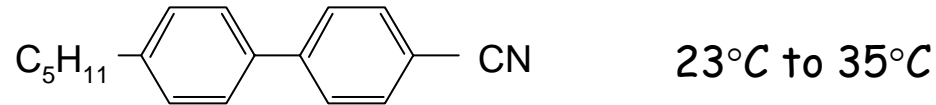


CN

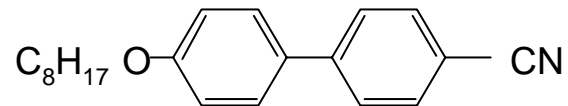
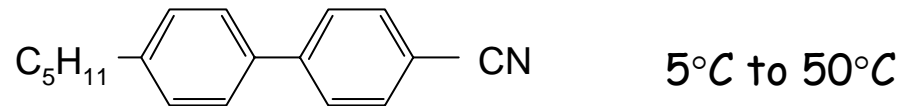


Wider Temperature Range

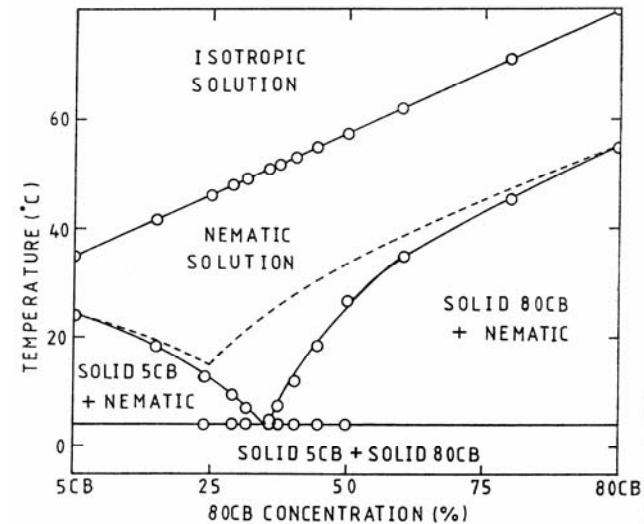
- Single compounds have narrow LC range



- Eutectic binary mixtures have broader LC range



- Binary phase diagrams
 - Easy to find eutectic



Multi-component Eutectic Mixtures

- Must be close to eutectic composition

- Schroder - van Laar Equation $\ln(X_i) = \frac{\Delta H_i}{R} \left(\frac{1}{T_i} - \frac{1}{T} \right)$
where:

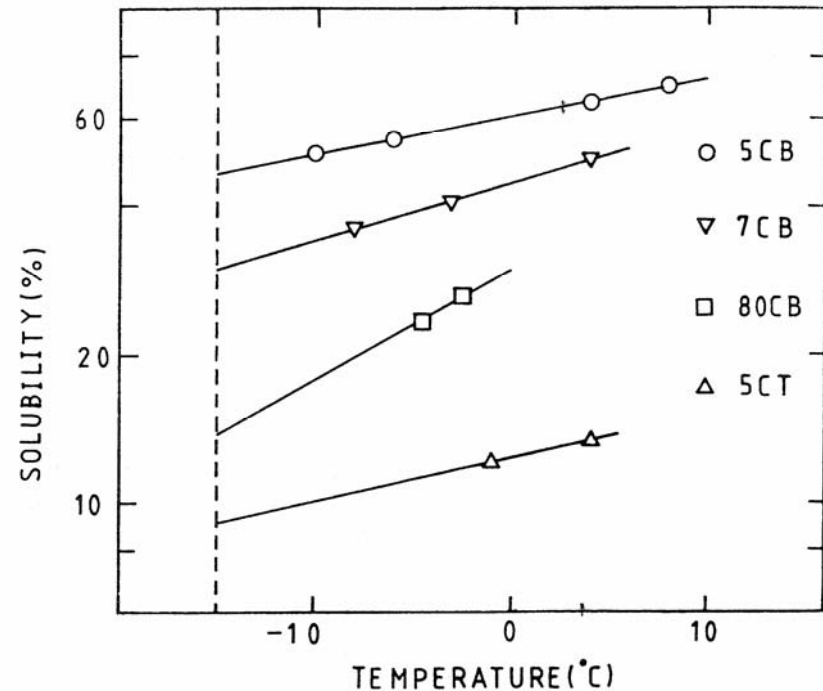
ΔH_i is latent heat of melting of i
 T_i is melting point of component i
 R gas constant

- Solubility equation $\ln(c_i) = a_i T_i + b_i$
where:

c_i is concentration (wt %) of i
 T is temp.,
 a_i and b_i are constants

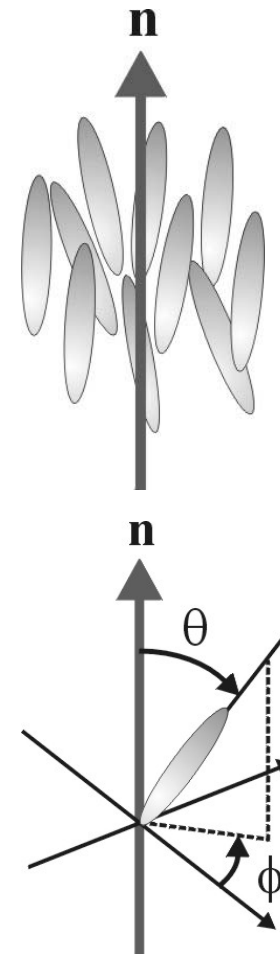
E7 Cyanobiphenyl Mixture

- 4 Components
 - 5CB $24 \Rightarrow 35\text{ }^{\circ}\text{C}$
 - 7CB $30 \Rightarrow 42\text{ }^{\circ}\text{C}$
 - 80CB $54 \Rightarrow 80\text{ }^{\circ}\text{C}$
 - 5CT $130 \Rightarrow 240\text{ }^{\circ}\text{C}$
- First generation
 - $(-15\text{ to }-10) \Rightarrow 60\text{ }^{\circ}\text{C}$
- Second Generation
 - $(-15\text{ to }-13) \Rightarrow 60\text{ }^{\circ}\text{C}$
 - Not wanted my manufacturer!
- Most common LC mixture in LCDs during 1975-1985



Spontaneous Ordering

- Director \mathbf{n} is a unit vector describing average orientation
- Average over:
 - time at a point in space
 - space at a snapshot in time
- Distribution function $f(\theta, \phi) d\Omega$:
 - probability of finding the molecule pointing in the small solid angle $d\Omega = \sin\theta d\theta d\phi$ around the direction (θ, ϕ)
- Nematics have cylindrical symmetry about \mathbf{n}
 - $\Rightarrow f(\theta, \phi) = f(\theta)$
- \mathbf{n} and $-\mathbf{n}$ are equivalent
 - $\Rightarrow f(\theta) \equiv f(\theta - \pi)$ symmetry

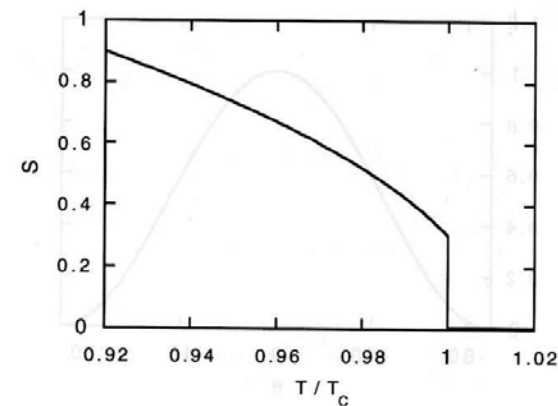


Nematic Order Parameter

- **Definition**
$$S = \frac{1}{2} \langle 3 \cos^2 \theta - 1 \rangle \equiv \frac{\int \frac{1}{2} f(\theta) (3 \cos^2 \theta - 1) d\Omega}{\int \frac{1}{2} f(\theta) d\Omega}$$
- **Perfect order** $f(\theta) = 1$ at $\theta = 0$ and $\theta = \pi$, but $f(\theta) = 0$ when $0 < \theta < \pi$

$$\Rightarrow S \sim \frac{1}{2} \langle 3 \cos^2 \pi - 1 \rangle = 1$$
- **Total disorder** $f(\theta) = 1$ for the whole range of angles $0 \leq \theta \leq \pi$

$$\Rightarrow S \sim \int_0^{2\pi} d\phi \int_0^\pi \frac{1}{2} (3 \cos^2 \theta - 1) \sin \theta d\theta = 0$$
- **Typical nematic LC**
 - $S = 0.3$ to 0.4 at N-I
 - Weakly second order
 - S increases as T lowered
 - S results in anisotropic properties



Anisotropic Physical Properties

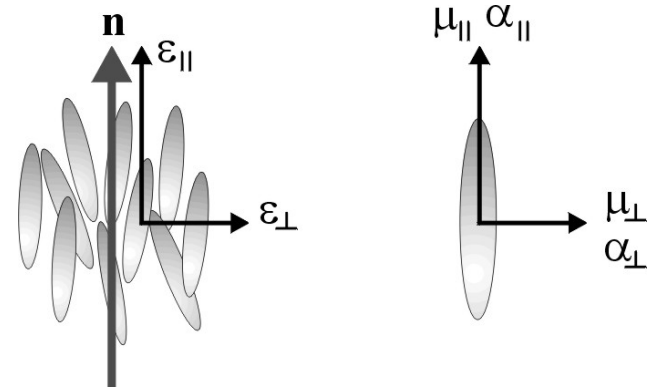
- Dielectric permittivities
- Refractive indices
- Elastic constants
- Viscosities

Dielectric Permittivities - Molecular Contributions

- Two polarisation contributions:

1. Anisotropic molecular polarisabilities (α)
 \Rightarrow contribute up to optical frequencies

2. Permanent molecular dipole moments (μ)
 \Rightarrow field skews molecular distribution
 \Rightarrow net polarisation
 \Rightarrow contributes to permittivity
but only at low frequencies



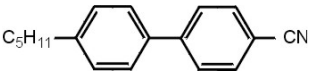
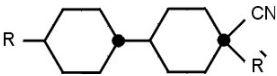
- Langevin theory for an isotropic fluid $\Rightarrow P = N\mu^2 E / 3k_B T$

Dielectric Permittivities - Ordering Effects

- Theory extended to anisotropic fluids (LC) by Maier and Meier

defining $\Delta\epsilon = \epsilon_{\parallel} - \epsilon_{\perp}$ and $\bar{\epsilon} = \frac{1}{3}(\epsilon_{\parallel} + 2\epsilon_{\perp})$

$$\Rightarrow \Delta\epsilon \propto \left[\Delta\alpha + \frac{F}{k_B T} \left(\mu_{\parallel}^2 - \frac{\mu_{\perp}^2}{2} \right) \right] S \quad \text{and} \quad \bar{\epsilon} \propto \bar{\alpha} + \frac{F}{3k_B T} (\mu_{\parallel}^2 + \mu_{\perp}^2)$$

molecular structure	μ_{\parallel}	μ_{\perp}	$\Delta\epsilon$
	4.8	0.0	+15
	0.0	4.0	-5

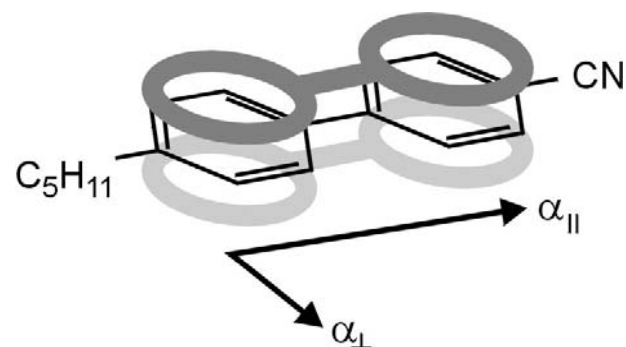
- $\Delta\epsilon$ dominated by dipole moment
- Relaxation at medium frequencies

Refractive Indices - Molecular Contributions

- Polarisability anisotropy

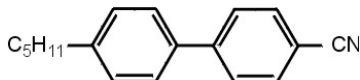
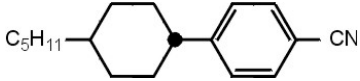
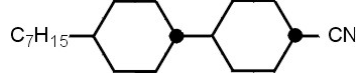
$$\Delta\alpha = \alpha_{||} - \alpha_{\perp}$$

- Delocalised electron clouds in phenyl rings enhance α



- Roughly coplanar phenyl rings in 5CB
 \Rightarrow large $\alpha_{||} \Rightarrow$ large $\Delta\alpha$ and Δn

- Cyclohexane ring
 - No delocalised electron cloud
 - Less planar \Rightarrow smaller $\alpha_{||} \Rightarrow$ smaller $\Delta\alpha$ and Δn

Molecular structure	$\Delta\alpha$	Δn
	19.4	0.194
	16.0	0.125
	11.1	0.045

Refractive Indices - Ordering Effects

- No dipole contributions at optical frequencies
- Use Maier and Meier equations with no dipoles

$$\Rightarrow \Delta\epsilon \propto \Delta\alpha S \quad \text{and} \quad \overline{\epsilon} \propto \overline{\alpha}$$

- From definition of the refractive index $n = \frac{c_{vacuum}}{v_{medium}} = \sqrt{\epsilon}$

$$\Rightarrow \left(n_{\parallel}^2 - n_{\perp}^2\right) \propto \Delta\alpha S \quad \text{and} \quad \overline{n^2} \propto \overline{\alpha}$$

$$\Rightarrow S \Delta\alpha / \overline{\alpha} = \left(n_{\parallel}^2 - n_{\perp}^2\right) / \left(\overline{n^2} - 1\right)$$

Refractive Indices - Variation with Temperature

- Order parameter dominates

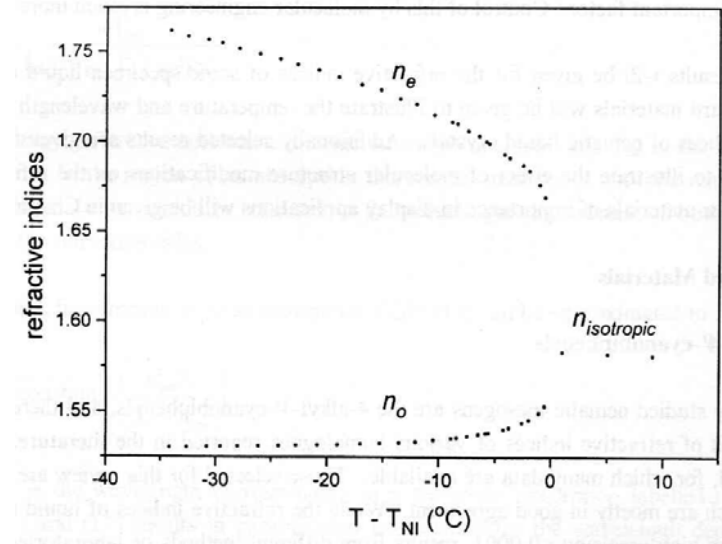
$$n_{\parallel} \propto \left(1 + \bar{\alpha} + \frac{2}{3} \Delta\alpha S\right)^{1/2}$$

$$n_{\perp} \propto \left(1 + \bar{\alpha} - \frac{1}{3} \Delta\alpha S\right)^{1/2}$$

- Haller plot assumes $S = \left(1 - \frac{T}{T_{NI}}\right)^b$

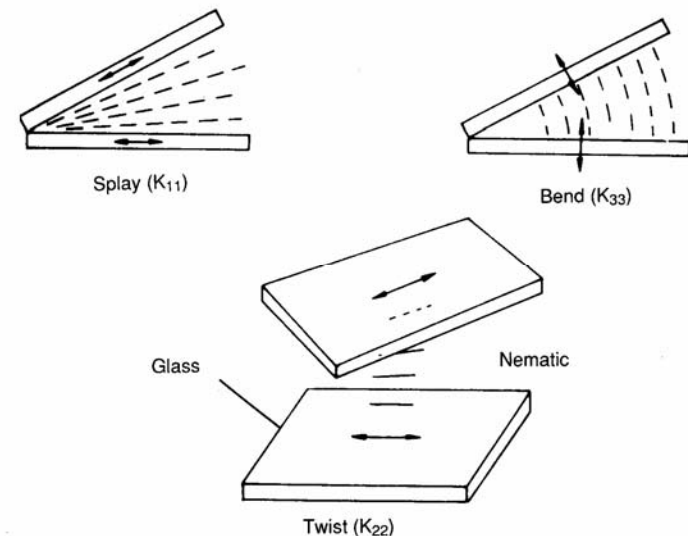
- Plot $\log \left(\frac{n_{\parallel}^2 - n_{\perp}^2}{n^2 - 1} \right)$ versus $\log \left(1 - \frac{T}{T_{NI}} \right)$

\Rightarrow Measurement of order parameter



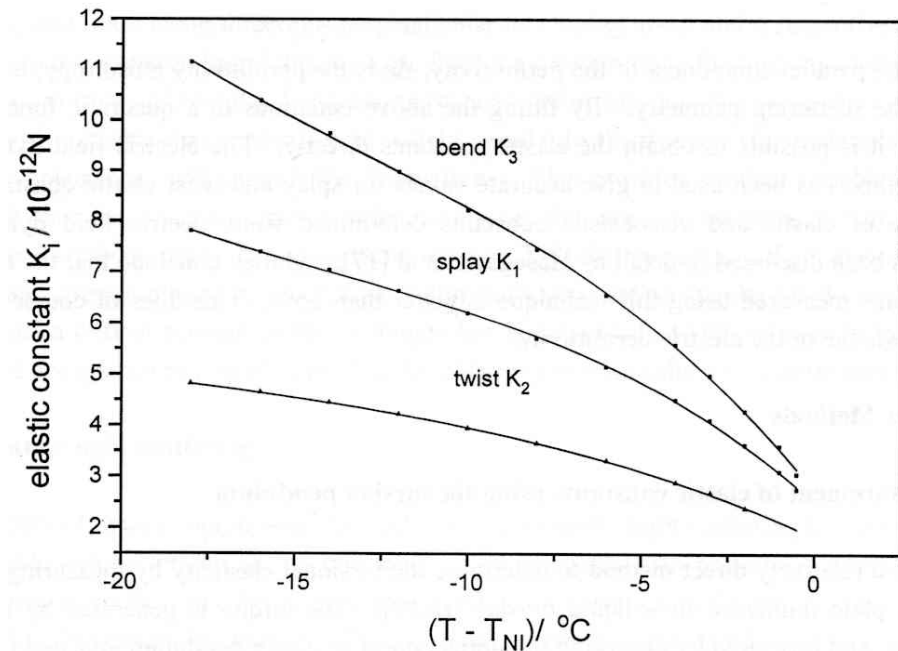
Elastic Constants - Definition

- With no forces applied
⇒ no spatial variation of director
- Elastic torques oppose any distortions
- Long range distortions
 - do not change order S
 - length scale \gg molecular length
- 3 elastic constants describe distortion e
 - splay (K_{11}), twist (K_{11}) and bend (K_{33})
- Used to calculate distortion energy



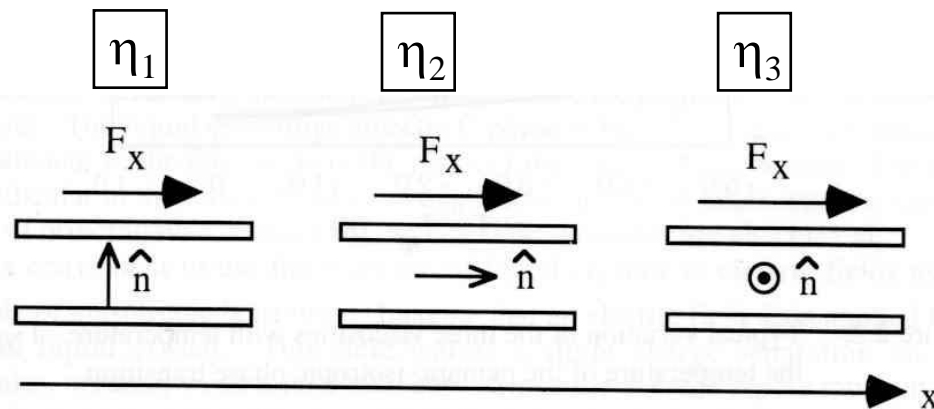
Elastic Constants - Typical Values

- Typically $K_{ii} \sim 10$ pN
 - $5 \text{ pN} < K_{11} < 20 \text{ pN}$
 - $K_{22} / K_{11} \sim 0.5$
 - $0.5 < K_{33} / K_{11} < 3.0$
- Temperature dependence
 - theory suggests $K_{ii} \sim S^2$



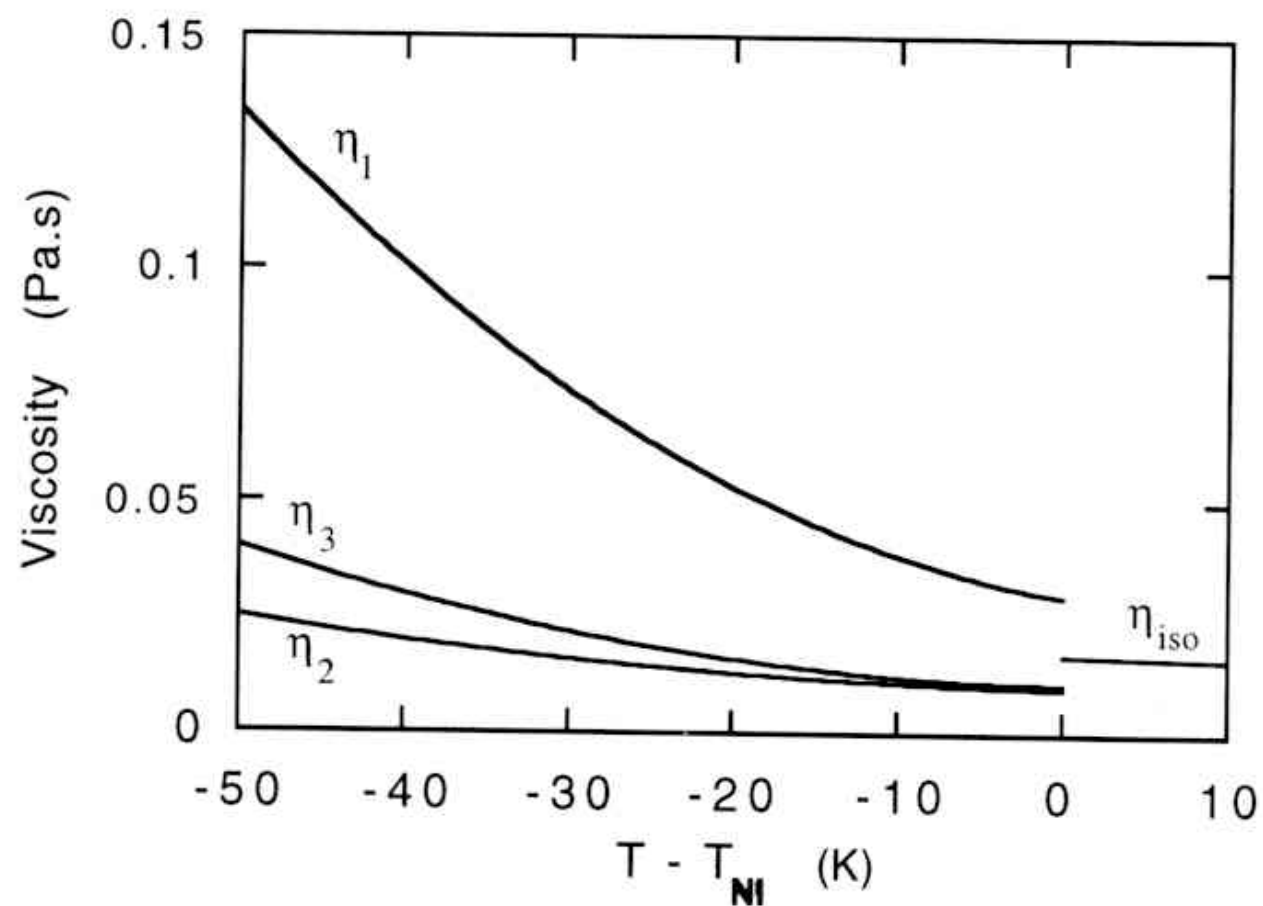
Viscosities - Definitions

- Shear (or Miesowicz) viscosities η_1 , η_2 and η_3



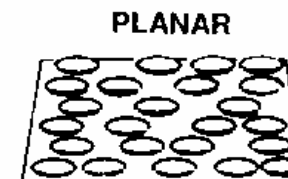
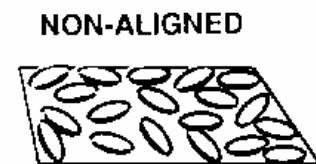
- From force required to move top plate
- 6 (5) Leslie viscosities for nematics
- Shear viscosities are combinations

Viscosities - Some Results



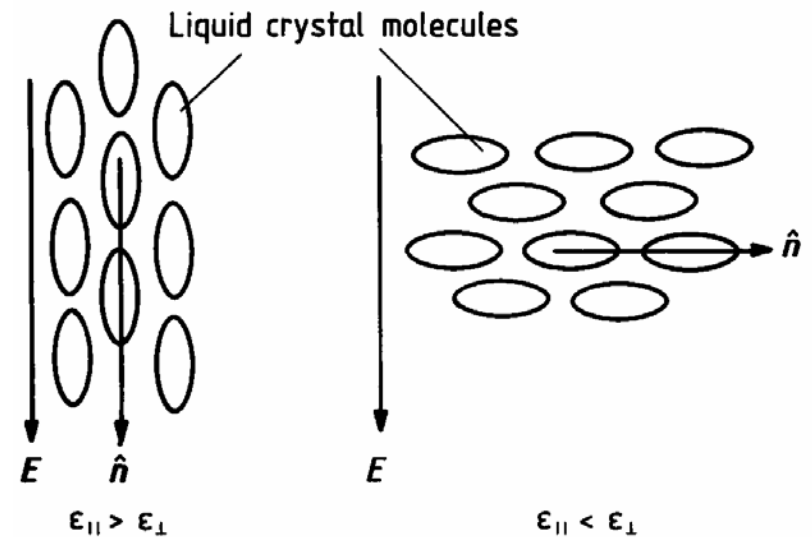
Alignment on Surfaces

- Two types of surface alignment
 - Normal (homeotropic)
 - Surfactants
 - Planar (homogeneous)
 - Oblique evaporation
 - Rubbing polymer coating
- LC between two surfaces
 - Uniform alignment
 - Controlled alignment
 - 'Single crystal' of LC



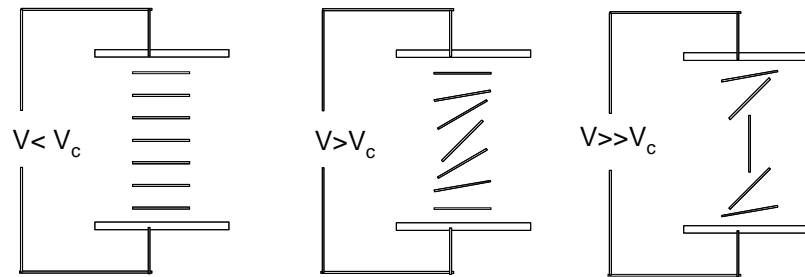
Alignment in Electric Field

- Electric permittivity
 - $\Delta\epsilon = \epsilon_{||} - \epsilon_{\perp}$
 - Dielectric anisotropy ($\Delta\epsilon$)
- Alignment type
 - Sign of $\Delta\epsilon$
- Alignment strength
 - Magnitude of $\Delta\epsilon$



Competition Between Fields and Surfaces

- Fundamental to most LCDs



- Threshold voltage V_c
 - $V < V_c$: nothing happens
 - $V > V_c \Rightarrow$ small reorientation
 - $V \gg V_c \Rightarrow$ strong reorientation
- V_c depends on k_{ii} and $\Delta\epsilon$ and is typically $\sim 1V$

Next Lecture

- Twisted nematic (TN) LCD
 - Device structure
 - Defects
 - Properties
- Passive addressing
 - RMS addressing
 - Alt & Pleshko analysis
- Continuum Theory
 - Static continuum theory
 - Application to switching mechanisms
 - Analytical calculations
 - Numerical calculations