

Aleksandra Królczyk 304296  
Roch Cieślowski 304239

# MKWS Project

## Combustion parameters analysis for a liquid propellant rocket engine Rocketdyne F-1

Warsaw, 08.06.2022r.

---

## 1 Introduction

Rocketdyne F-1 was the first rocket engine that guided humans to the moon. It is indisputably one of the best achievements of humanity, therefore the above-mentioned engine is one of the most famous and most powerful motors of all times. The following work will present approached parameters of combustion for the Rocketdyne F-1 rocket engine. In order to complete the calculations Cantera library shall be used. Results will be compared to the experimental data contained in the training manual provided by The Rocketdyne Division.

## 2 Model

### Assumptions

- Fuel: kerosene (RP-1)
- Oxidizer: oxygen ( $O_2$ )
- Volume of combustion chamber:  $0,7m^3$

Due to the lack of suitable kerosene model in the Cantera's resources, it was necessary to use a surrogate fuel model. Substitute model applied to calculations was the Dagaut model developed by Centre Européen de Recherche et de Formation Avancée en Calcul Scientifique. The surrogate is made up of 74% of  $NC_{10}H_{22}$ , 15% of  $PHC_3H_7$  and 11% of  $CYC_9H_{18}$ . For above substances the following properties have been defined:

Input values used for calculations are shown in the following table:

Input data	Oxidizer	Fuel
State of matter	gas	gas
Pressure [Pa]	$1,2797 \cdot 10^7$	$3 \cdot 10^5$
Temperature [K]	800	300

For analysis four following reservoirs have been used:

- Oxidizer (Oxygen)
- Fuel (Kerosene)
- Igniter (H radicals which ignite mixed flow of fuel and oxidizer)
- Combustor
- Exhaust ( $N_2$  molecules)

Moreover, after computation specific impulse of the rocket has been defined. Equation based on [4]

$$I_{sp} = \sqrt{\frac{k \frac{R}{M_{gas}} T_c}{k - 1} \left[ 1 - \left( \frac{p_e}{p_c} \right)^{\frac{k-1}{k}} \right]} \cdot \frac{1}{g}$$

where:

$k$  - ratio of specific heats,  $\frac{cp}{cv}$

$p_e$  - nozzle exit pressure

$p_c$  - combustion chamber pressure

$T_c$  - combustion chamber temperature

$R$  - exhaust flow universal gas constant

$g$  - gravitational acceleration

Necessary parameters were taken from [1]

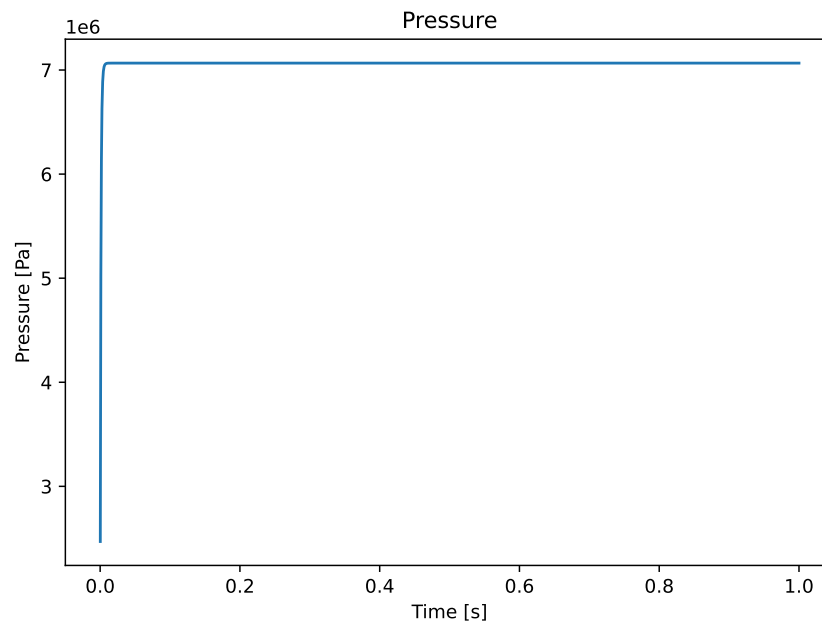
Thrust level (sea level)	1,522,000 pounds	Gas generator mixture ratio	0.416:1
Specific impulse (sea level)	265.3 seconds	Gas generator combustor pressure	980 psia
Total propellant flowrate	5,736 lb/sec (40,644 gpm)	Gas generator temperature	1,453° F
a. Fuel	1,754 lb/sec (15,606 gpm)	Turbine speed	5,492 rpm
b. Oxidizer	3,982 lb/sec (25,038 gpm)	a. Time from turbo-pump initiation to rated speed	5.2 seconds
Mixture ratio	2.27:1	b. Time from cutoff to zero rpm	3.5 seconds
Expansion ratio	16:1	Turbine brake horse-power	53,146 hp
Thrust chamber pressure	1,125 psia	Nozzle extension coolant gas temperature	1,138° F
Thrust chamber temperature	5,970° F	Hydraulic recirculation flowrate	11.6 ±1.1 gpm at 1,500 psig
Thrust chamber exit pressure (16:1)	9.6 psia	Engine dry weight (average)	18,619 pounds
Fuel pump discharge pressure	1,870 psia		
Oxidizer pump discharge pressure	1,602 psia		
Gas generator flowrate (included in total)	167 lb/sec		
a. Fuel	118 lb/sec		
b. Oxidizer	49 lb/sec		

Figure 1-5. Nominal F-1 Engine Parameters

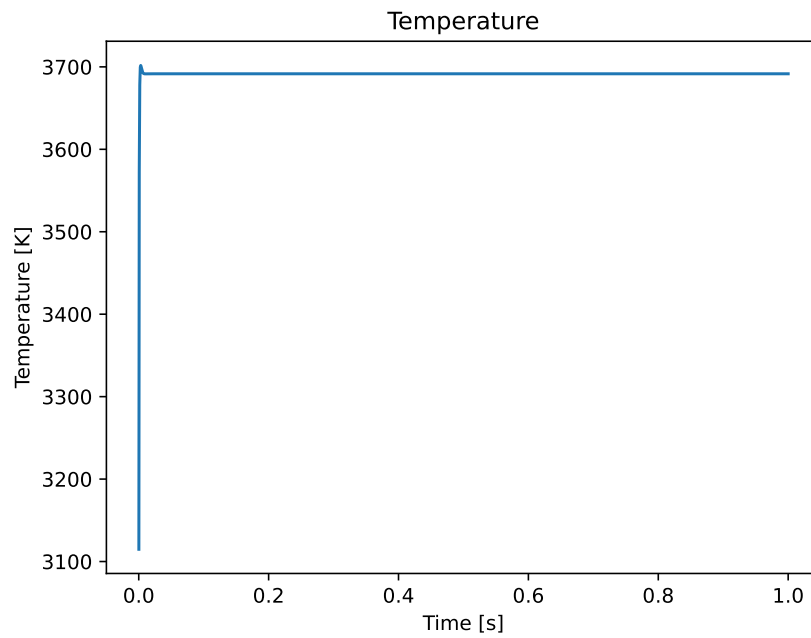
---

## 3 Results

### 3.1 Pressure



### 3.2 Temperature



### 3.3 Specific Impulse

$$I_{sp} = 209.3s$$

---

## 4 Comparison

	Approached	Experimental	Relative Error [%]
Pressure [Pa]	$7.06 \cdot 10^6$	$7 \cdot 10^6$	0.857
Temperature [K]	3691.6	3572.04	3.35
Specific Impulse [s]	209.3	265.3	21.1

As we can see simulation results are acceptably close to real parameters.

## 5 Conclusions

In conclusion, Cantera library is sufficient for laminar flow in constant volume simulations. The reason of difference in Specific Impulse might be the usage of substitute model of kerosene, which change the molecular weight of gas. Moreover, to make the simulation closer to reality we used Oxygen as the oxidizer instead of air, which is recommended in the Dagaut's mechanism. That might be another reason for different molecular mass of gases in the chamber.

## 6 Bibliography

- [1] <https://www.pdf-archive.com/2016/10/21/rocketdyne-f1-engine-manual/preview/page/10/>
- [2] <https://chemistry.cerfacs.fr/en/chemical-database/mechanisms-list/dagauts-mechanism/>
- [3] [https://github.com/lyczeko/MKWS\\_kerosene\\_combustion/blob/master/projekt\\_MKWS/projekt1\\_MKWS2018\\_ALyczek.pdf](https://github.com/lyczeko/MKWS_kerosene_combustion/blob/master/projekt_MKWS/projekt1_MKWS2018_ALyczek.pdf)
- [4] [https://github.com/KrzysiuPietrzak/MKWS\\_project/blob/master/Krzysztof\\_pietrzak\\_MKWS.pdf/](https://github.com/KrzysiuPietrzak/MKWS_project/blob/master/Krzysztof_pietrzak_MKWS.pdf/)

## 7 Github

Link to GitHub account:

[https://github.com/roch00/MKWS2022\\_RocketdyneF1](https://github.com/roch00/MKWS2022_RocketdyneF1)