

Across

- √4. The streamline flow of a fluid flows without fluctuations or turbulence
- $\sqrt{7}$. A law that states that the vapour pressure of a solution is reduced in proportion to the amount of solute that is added. It is a German discovery
- √9. A part of a system that is chemically and physically uniform throughout
- $\sqrt{10}$. How many significant figures would the solution of the following problem have? [(5.74) (38.27)/0.001250]
- √16. Creator of the Periodic table
- √18. Capable of igniting constantly in air
- √22. A mechanical device used to move air, gas or vapour at low gauge pressure
- √23. Pressure measuring device containing a thin metal tube flexes and moves a dial as the pressure measured being changes
- √24. Which temperature is defined as the product temperature when the fuel is burned completely in an adiabatic reactor?
- $\sqrt{26}$. Lumen/(m*m) divided by 41 gives the Unit
- $\sqrt{28}$. The amount of energy required to change the temperature of unit mass or mole one

Down

- √1. Partial Vaporization of a liquid of a given composition at a fixed temperature
- $\sqrt{2}$. The property of fluids to form distinct and separate phases under all relative proportions
- $\sqrt{3}$. A system in which one or more of the conditions of the system vary with time.
- $\sqrt{\mathbf{5}}$. A process in which material is neither added to nor removed from the process during its operation
- √6. You will find when you travel back to ancient Rome to weigh medicines
- $\sqrt{8.}$ -228.599 dB(K/W/Hz)(Hint: proportionality factor){Drop one 'n'}
- √11. This is a quick but progressive combustion as distinguished from a detonation which is the instantaneous decomposition of combustible material
- √12. The father of chemical engineering
- √13. 1.9858 BTU.lb/mol/(degree Reumer)
- √14. This scale is used to report specific gravity of petroleum compounds
- √**15.** Avogadro's number multiplied by 453.6 molecules gives
- √17. Physical phenomenon that occurs when gas or liquid molecules are brought into contact

degree

- √30. An economic activity by business that involves the manufacture of products through the conversion of raw materials, production of goods or supply of services
- √31. Check the Beck
- √32. An open system with material entering and/or leaving the system
- √**34.** First university to introduce a 4-year degree in chemical engineering
- √36. A complete or partial reduction in the reactivity of a substance such as enzyme or poisoning of a catalyst
- √37. The partial pressure of water vapour in a non-condensable gas divided by the vapour pressure of water at the temperature of gas
- √38. A relation that relates the partial pressure of one component in the vapor phase to the mole fraction of the same component in the liquid phase
- $\sqrt{39}$. Fathoms multiplied by even prime and smallest odd prime
- √40. A containment wall surrounding a storage tank or process vessel designed to contain its contents in the event of accidental leakage, spillage or a catastrophic failure

- with a solid surface (Solid surface = Check the Beck)
- √19. The system in which no variation of conditions is possible without one phase disappearing
- √20. Brewing a cup of coffee is an example of which process
- √21. Location in a process where a stream of material is first introduced.]
- √25. A documented form of evidence that a process is capable of producing a product to a specified standard or quality. It is used for various manufacturing processes and, in particular, the production of pharmaceutical products
- √27. Cord multiplied by the even prime seven times
- √29. An instrument used to measure the speed of a gas such as air. It comprises of cups or vanes that rotate freely and are linked to tachometer
- √31. The point for which the dew point curve and bubble point curve coincide in a phase diagram
- $\sqrt{33}$. This unit is the same as 1000 litres
- √35. This person taught the first chemical engineering course and introduced the first four-year undergraduate chemical engineering program.