

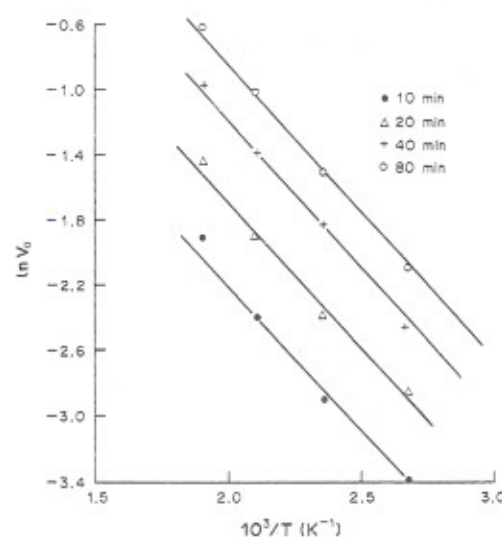
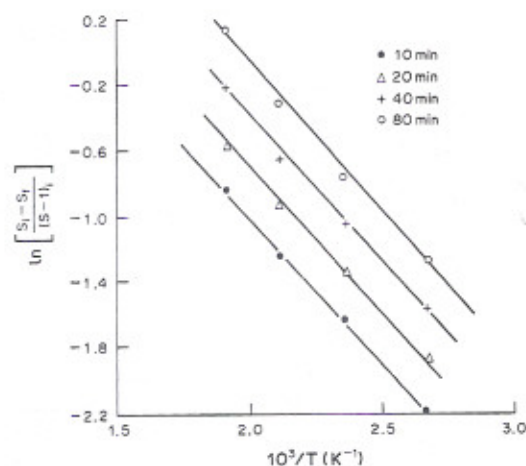
Table 1. The values of annealing rate, V_a , as the rate of change of length and diameter, for different angles θ of incident beam (^{208}Pb at 17 MeV n^{-1}), annealed for 10 min

Serial No.	T ($^{\circ}\text{C}$)	T (K)	$\frac{10^3}{T}$ (K^{-1})	$\theta = 90^{\circ}$		$\theta = 45^{\circ}$				
				D (μm)	$V_a = \frac{dD}{dt}$	Observed length l' (μm)	Actual length $l'/\cos\theta$	$V_a = \frac{dl}{dt}$	D (μm)	$V_a = \frac{dD}{dt}$
1	Unannealed	—	—	$17.50 \pm 0.20^*$	—	20.00 ± 0.25	28.17^{\dagger}	—	$11.00 \pm 0.15^*$	—
2	50	323	3.09	16.25 ± 0.15	0.12	18.50 ± 0.15	26.05	0.21	10.00 ± 0.10	0.10
3	100	373	2.68	15.00 ± 0.12	0.25	16.50 ± 0.15	23.24	0.49	8.50 ± 0.15	0.25
4	150	423	2.36	13.50 ± 0.15	0.40	14.00 ± 0.20	19.72	0.85	6.50 ± 0.10	0.45
5	200	473	2.11	10.00 ± 0.10	0.75	10.00 ± 0.10	14.08	1.41	4.00 ± 0.12	0.70
6	250	523	1.91	6.50 ± 0.15	1.10	5.00 ± 0.10	7.04	2.11	1.00 ± 0.10	1.00

* D_0 . $^{\dagger}l_0$.

etched along with the parent unannealed sample, under the optimum etching conditions of 2.5% HF at room temperature (32°C) with etching time varying from 20 to 60 min.

The mean observed track lengths (l') and diameters (D) were measured using a Carl Zeiss microscope with a resolution of $1 \mu\text{m}$. The actual mean lengths were calculated using the relation $l = l'/\cos\theta$, where θ is the angle of incident beam with respect to the detector surface. Statistical errors were also applied to these measurements for each ion set, they were also made to conform to a standard deviation of 1σ . The track etch rate, V_T , is obtained from the linear portion of the plot of track length vs etching time. The general etch rate, V_G , is found to be independent of the extent of annealing, and hence can be treated as constant for soda-lime glass. We have projected here the curves obtained from the annealing data for only the ^{139}La (14.6 MeV n^{-1}) ion beam, whereas the results for ^{208}Pb (17 and 13.6 MeV n^{-1}) ion beams have been tabulated, as the curves drawn for these ions show correspondence with those for ^{139}La .

FIG. 1. Plot of $\ln V_a$ vs $1/T$ (10^3 K^{-1}) for a soda-lime glass detector using ^{139}La ions.FIG. 2. Plot of $\ln [(S_1 - S_0)/(S_1 - 1)]$ vs $1/T$ (10^3 K^{-1}) for a soda-lime glass detector using ^{139}La ions.

3. RESULTS AND DISCUSSION

The annealing rate, V_a , for different ion sets was calculated as the rate of change of length as well as diameter (Table 1). The plot of $\ln V_a$ against reciprocal temperature, T , yields a straight line for each ion

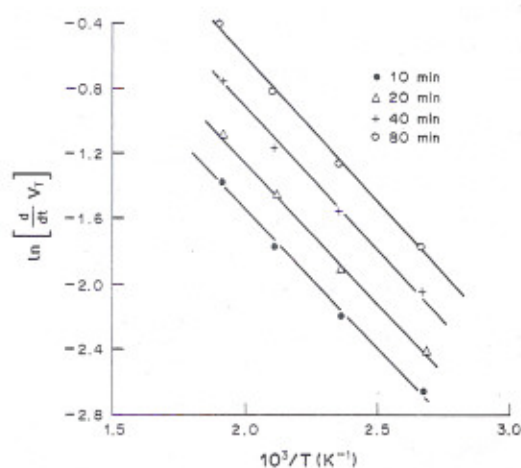
FIG. 3. Plot of $\ln [d/dt (V_r)]$ vs $1/T$ (10^3 K^{-1}) for a soda-lime glass detector using ^{139}La ions.

Table 2. Comparison of E_a , n and A values obtained by using different formulations (see text)

Ion beam	Modgil and Virk formulation			Price <i>et al.</i> formulation			Modified formulation of Modgil and Virk		
	Activation energy E_a (eV)	n value	A value ($\mu\text{m min}^{-1}$)	Activation energy E_a (eV)	n value	A value ($\mu\text{m min}^{-1}$)	Activation energy E_a (eV)	n value	A value ($\mu\text{m min}^{-1}$)
^{139}La (14.6 MeV n^{-1})	0.16	0.64	76–88	0.15	0.55	4.0–4.5	0.15	0.55	2.4–2.7
^{208}Pb (17 MeV n^{-1})	0.16	0.65	75–90	0.15	0.57	5.2–5.6	0.15	0.58	3.5–4.0
^{208}Pb (13.6 MeV n^{-1})	0.16	0.60	73–90	0.15	0.52	3.1–3.7	0.15	0.54	2.8–3.2

set (Fig. 1). The activation energy as calculated using equation (1) is 0.16 eV for all three ions.

Using the Price *et al.* (1987) formulation, the etch-rate ratio, $S = V_T/V_G$, is calculated for all three ions. The plots of the logarithms of the left-hand side of equation (2) against the reciprocal temperature yield straight lines (Fig. 2), and hence a single activation energy. This activation energy turns out to be identical (i.e. 0.15 eV) for all three ions, and thus appears to be only medium-specific. The annealing data for track etch rate, V_T , are also used to fit equation (3). The plot of the logarithm of the left-hand side of equation (3) against the reciprocal temperature again yields a straight line (Fig. 3) and, hence, a single activation energy of 0.15 eV.

Table 2 shows a comparison of values of E_a , n and A for all three ions, using different formulations. Thus, we find that our annealing data for soda-lime glass, using different formulations, give similar results for activation energy, E_a . However, A and n values are found to be quite discordant in these formulations.

The above results further support the concept of a single activation energy as an intrinsic property of the detector. It also proves that in those SSNTDs where general etch rate (V_G) is independent of annealing, our revised formulation is simpler and more suitable.

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