

# Identification of the OF value for maximising the specific impulse of the BT-4 Cygnus engine

ROSARIO DONNARUMMA<sup>1</sup>

<sup>1</sup>University of Naples Federico II, Department of Industrial Engineering

Correspondence: rosar.donnarumma@studenti.unina.it

**ABSTRACT** In this paper a CEA analysis of the BT-4 Cygnus engine is presented. IHI Aerospace of Japan designed and built the BT-4, a pressure-fed liquid rocket engine. The engine uses Monomethylhydrazine fuel and a Nitrogen Tetroxide Oxidizer to produce 450 N of thrust. The propellants are kept in spherical tanks that are helium-pressured. The aim of this project is to evaluate the best mixing ratio to optimise the specific impulse developed by the BT-4 Cygnus. Attention is also drawn to the comparison between the case of efflux from the nozzle in equilibrium and frozen or frozen efflux in the divergent section of the nozzle.

**KEYWORDS** *BT-4 (Cygnus), CEA, Thermal Analysis, Rocket Problem*

## I. INTRODUCTION

In the field of satellite dynamics, small thrusters with liquid propellants are often used to control the attitude of a satellite or to *rendezvous* another object flying in space. Nitrogen tetroxide ( $N_2O_4$ ) is a highly concentrated storable oxidant. It was developed for the LUNAR-A project, however it has since been employed as a liquid apogee engine in various geostationary communications satellites based on Lockheed Martin's A2100 and GEOStar-2 satellite buses.

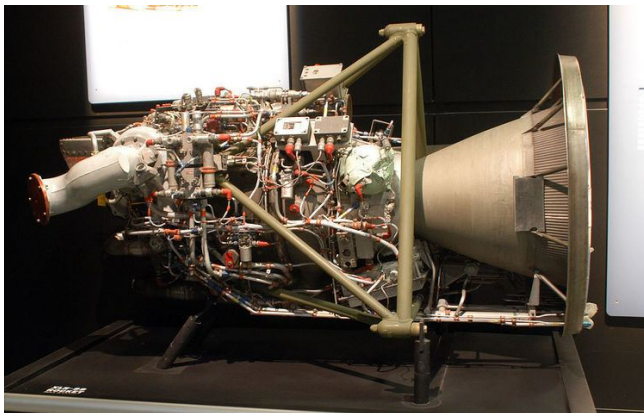


Figure 1: BT-4 (Cygnus)

It was also used on the autonomous cargo spacecraft HTV and Cygnus. For many space applications,

hydrazine monomethyl (*MMH*) and  $N_2O_4$  are a frequently used pair. Despite their known toxicity, due to their ability to provide a high specific impulse and extreme storage stability, this combination is widely used in orbital manoeuvres and launch vehicle propulsion. For re-orientation and minor rendezvous burns, Cygnus' ACS is used.

## II. CASE STUDY AND SOFTWARE IMPLEMENTATION

The hypotheses and data used in this work are shown in tabular form Tab.1.

The analyses were carried out under two conditions: in the case of evolutionary equilibrium or frozen. The most real hypothesis is that of frozen reactions (*frozen*) in the divergent part of the nozzle: there will be many atoms leaving the system, representing lost energy. At this point it is possible to run the CEA software using the data in Tab. 1 and plot the obtained results. The results of the CEA software are reported in the Appendix.

## III. RESULTS

It can be seen from the plot Fig. 5 that, as designed, the maximum impulse is obtained around a mixing ratio of about 1.6. It should be noted that this result has also been obtained in the literature [1]. At this point it is possible to analyse the trend of the concentrations both in the combustion chamber and in the nozzle.

BT-4	
Pressure	1 atm
OF	[1,2,6]
Fuel	$CH_6N_2(L)$ ,MMH
Oxidizer	$N_2O_4$

Table 1: Input Data



Figure 2: The Canadarm2 moves toward the Orbital Sciences Corp. Cygnus commercial cargo craft as it approaches the International Space Station on Jan. 12, 2014. [2]

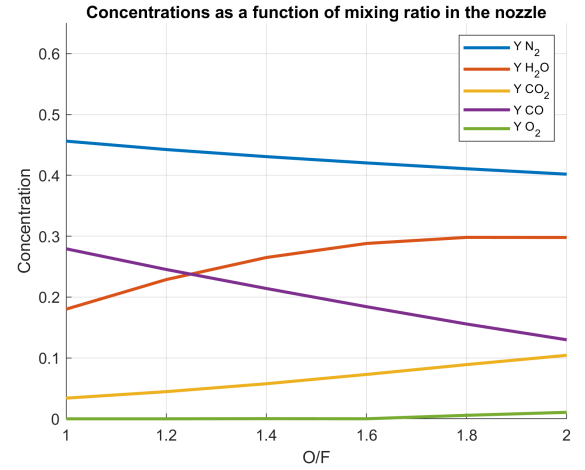


Figure 4: Concentrations as a function of mixing ratio in the nozzle

product of the characteristic speed  $c^*$  and the thrust coefficient  $c_f$ :

$$I_{sp} = c_f \cdot c^* \quad (1)$$

Note that the characteristic velocity, exclusively related to what happens in the combustion chamber, is given by:

$$c^* = f(T_{ch}, \text{Molecular Weight}) \quad (2)$$

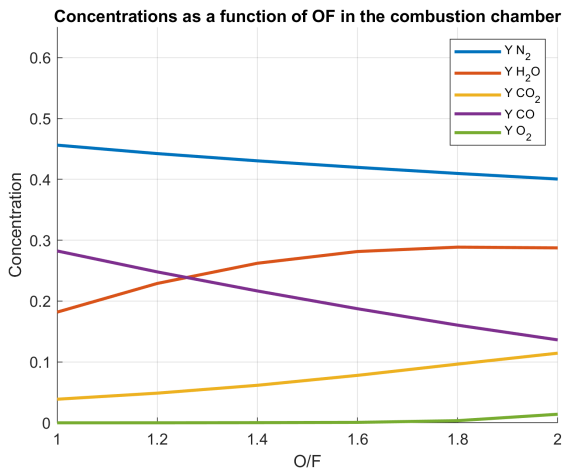


Figure 3: Concentrations as a function of mixing ratio in the combustion chamber

However, it is of fundamental importance to understand the parameters that characterise this trend. This is possible by analysing the trends of the functions that constitute it. The specific impulse is given by the

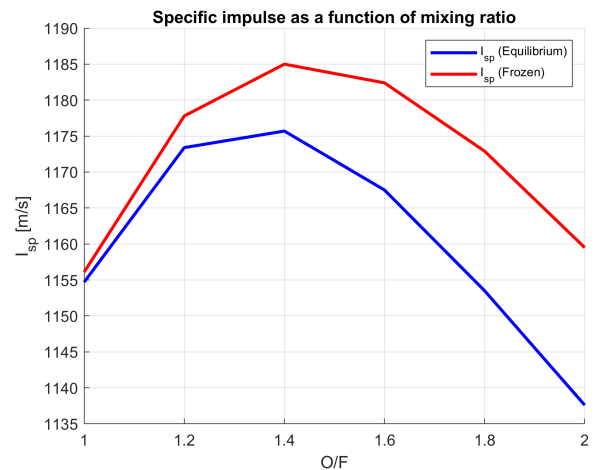


Figure 5: Specific impulse plot



It is also desired to emphasise the difference between the case of efflux in equilibrium and frozen efflux in the divergent part of the convergent-divergent nozzle. It can be seen immediately that the specific impulse in the second case mentioned is actually lower. This is because, when the efflux is frozen, the characteristic time of the chemical reactions  $t_{eq}$  is much greater than the characteristic convective time, i.e. the crossing time  $t_c$ . Under these conditions, the exothermic reactions that would lead to the re-association of the dissociated species generated in the combustion chamber do not occur. Re-association would lead to an increase in the enthalpy of association and thus a consequent increase in temperature. When the efflux is frozen, and therefore this does not occur, the chemical energy to be converted into kinetic energy through the nozzle, and therefore into thrust, is less than in the case of efflux at equilibrium, where instead the exothermic reactions would have the necessary time to start. In conclusion, therefore, the difference in specific impulses in the case of frozen efflux and in the case of efflux at equilibrium will be hardly noticeable for low OF ratios since they will correspond to low temperatures. In this case the dissociated species from combustion would be present in small quantities. If they were to be re-associated, there would be no major energy emissions. A much sharper distinction between the two cases can be seen at high temperatures and thus as the mixing ratio increases.



## APPENDIX. CEA LISTING

## Listing 1: CEA Listing

```
NASA-GLENN CHEMICAL EQUILIBRIUM PROGRAM CEA2, FEBRUARY 5, 2004
BY BONNIE MCBRIDE AND SANFORD GORDON
REFS: NASA RP-1311, PART I, 1994 AND NASA RP-1311, PART II, 1996

*****

### CEA analysis performed on Wed 05-May-2021 16:53:51

# Problem Type: "Rocket" (Infinite Area Combustor)

prob case=dmf_____1812 ro equilibrium frozen

# Pressure (1 value):
p,atm= 10

# Oxidizer/Fuel Wt. ratio (6 values):
o/f= 1, 1.2, 1.4, 1.6, 1.8, 2

# You selected the following fuels and oxidizers:
reac
fuel CH6N2(L),MMH      wt%=100.0000
oxid N2O4              wt%=100.0000

# You selected these options for output:
# short version of output
output short
# Proportions of any products will be expressed as Mass Fractions.
output massf
# Heat will be expressed as siunits
output siunits

# Input prepared by this script:/var/www/sites/cearun.grc.nasa.gov/cgi-bin/CEARU
N/prepareInputFile.cgi

### IMPORTANT: The following line is the end of your CEA input file!
end

THEORETICAL ROCKET PERFORMANCE ASSUMING EQUILIBRIUM

COMPOSITION DURING EXPANSION FROM INFINITE AREA COMBUSTOR

Pin = 147.0 PSIA
CASE = dmf_____

          REACTANT              WT FRACTION      ENERGY      TEMP
          CH6N2(L),MMH          (SEE NOTE)     KJ/KG-MOL     K
FUEL      CH6N2(L),MMH          1.0000000      54200.000     298.150
OXIDANT    N2O4                 1.0000000       0.000        0.000

O/F=      1.00000  %FUEL= 50.000000  R,EQ.RATIO= 2.496406  PHI,EQ.RATIO= 2.496406

          CHAMBER  THROAT
Pinf/P      1.0000  1.8109
P, BAR      10.133  5.5954
T, K        2396.24 2125.90
RHO, KG/CU M 8.5052-1 5.3001-1
H, KJ/KG      588.21 -78.453
U, KJ/KG     -603.12 -1134.17
G, KJ/KG     -32567.9 -29493.9
S, KJ/(KG) (K) 13.8367 13.8367
```



M, (1/n)	16.724	16.743
(dLV/dLP)t	-1.00084	-1.00027
(dLV/dLT)p	1.0193	1.0066
Cp, KJ/(KG) (K)	2.5931	2.4137
GAMMA <sub>s</sub>	1.2475	1.2630
SON VEL, M/SEC	1219.1	1154.7
MACH NUMBER	0.000	1.000

## PERFORMANCE PARAMETERS

Ae/At	1.0000
CSTAR, M/SEC	1655.6
CF	0.6974
Ivac, M/SEC	2069.0
Isp, M/SEC	1154.7

## MASS FRACTIONS

*CO	0.28236	0.27932
*CO <sub>2</sub>	0.03397	0.03875
*H	0.00018	0.00006
*H <sub>2</sub>	0.04508	0.04541
H <sub>2</sub> O	0.18186	0.18015
NH <sub>3</sub>	0.00001	0.00001
*NO	0.00002	0.00000
*N <sub>2</sub>	0.45622	0.45624
*OH	0.00028	0.00006

\* THERMODYNAMIC PROPERTIES FITTED TO 20000.K

NOTE. WEIGHT FRACTION OF FUEL IN TOTAL FUELS AND OF OXIDANT IN TOTAL OXIDANTS

## THEORETICAL ROCKET PERFORMANCE ASSUMING FROZEN COMPOSITION

Pin = 147.0 PSIA

CASE = dmt\_\_\_\_\_

	REACTANT	WT FRACTION (SEE NOTE)	ENERGY KJ/KG-MOL	TEMP K
FUEL	CH <sub>6</sub> N <sub>2</sub> (L), MMH	1.0000000	54200.000	298.150
OXIDANT	N <sub>2</sub> O <sub>4</sub>	1.0000000	0.000	0.000

O/F= 1.00000 %FUEL= 50.000000 R,EQ.RATIO= 2.496406 PHI,EQ.RATIO= 2.496406

	CHAMBER	THROAT
Pinf/P	1.0000	1.8170
P, BAR	10.133	5.5764
T, K	2396.24	2110.20
RHO, KG/CU M	8.5052-1	5.3153-1
H, KJ/KG	588.21	-80.032
U, KJ/KG	-603.12	-1129.15
G, KJ/KG	-32567.9	-29278.3
S, KJ/(KG) (K)	13.8367	13.8367

M, (1/n)	16.724	16.724
Cp, KJ/(KG) (K)	2.3588	2.3119
GAMMA <sub>s</sub>	1.2671	1.2740
SON VEL, M/SEC	1228.6	1156.1
MACH NUMBER	0.000	1.000

## PERFORMANCE PARAMETERS

Ae/At	1.0000
CSTAR, M/SEC	1648.9
CF	0.7011
Ivac, M/SEC	2063.6



Isp, M/SEC 1156.1

#### MASS FRACTIONS

*CO	0.28236	*CO2	0.03397	*H	0.00018
*H2	0.04508	H2O	0.18186	NH3	0.00001
*NO	0.00002	*N2	0.45622	*OH	0.00028

\* THERMODYNAMIC PROPERTIES FITTED TO 20000.K

NOTE. WEIGHT FRACTION OF FUEL IN TOTAL FUELS AND OF OXIDANT IN TOTAL OXIDANTS

#### THEORETICAL ROCKET PERFORMANCE ASSUMING EQUILIBRIUM

#### COMPOSITION DURING EXPANSION FROM INFINITE AREA COMBUSTOR

Pin = 147.0 PSIA

CASE = dmt\_\_\_\_\_

	REACTANT	WT FRACTION (SEE NOTE)	ENERGY KJ/KG-MOL	TEMP K
FUEL	CH6N2 (L) ,MMH	1.0000000	54200.000	298.150
OXIDANT	N2O4	1.0000000	0.000	0.000

O/F= 1.20000 %FUEL= 45.454545 R,EQ.RATIO= 2.080338 PHI,EQ.RATIO= 2.080338

	CHAMBER	THROAT
Pinf/P	1.0000	1.7911
P, BAR	10.133	5.6570
T, K	2691.24	2429.96
RHO, KG/CU M	8.1468-1	5.0545-1
H, KJ/KG	534.74	-153.66
U, KJ/KG	-709.00	-1272.87
G, KJ/KG	-35700.3	-32870.8
S, KJ/(KG) (K)	13.4641	13.4641
M, (1/n)	17.991	18.052
(dLV/dLP)t	-1.00295	-1.00125
(dLV/dLT)p	1.0632	1.0292
Cp, KJ/(KG) (K)	2.9373	2.5904
GAMMA <sub>s</sub>	1.2120	1.2302
SON VEL, M/SEC	1227.8	1173.4
MACH NUMBER	0.000	1.000

#### PERFORMANCE PARAMETERS

Ae/At	1.0000
CSTAR, M/SEC	1708.5
CF	0.6868
Ivac, M/SEC	2127.2
Isp, M/SEC	1173.4

#### MASS FRACTIONS

*CO	0.24795	0.24534
*CO2	0.04460	0.04872
*H	0.00053	0.00024
*H2	0.03340	0.03379
H2O	0.22898	0.22878
NH3	0.00001	0.00001
*NO	0.00023	0.00006
*N2	0.44233	0.44242
*O	0.00003	0.00001
*OH	0.00190	0.00064
*O2	0.00002	0.00000

\* THERMODYNAMIC PROPERTIES FITTED TO 20000.K



NOTE. WEIGHT FRACTION OF FUEL IN TOTAL FUELS AND OF OXIDANT IN TOTAL OXIDANTS

## THEORETICAL ROCKET PERFORMANCE ASSUMING FROZEN COMPOSITION

Pin = 147.0 PSIA

CASE = dmt\_\_\_\_\_

	REACTANT	WT FRACTION (SEE NOTE)	ENERGY KJ/KG-MOL	TEMP K
FUEL	CH6N2 (L) ,MMH	1.0000000	54200.000	298.150
OXIDANT	N2O4	1.0000000	0.000	0.000

O/F= 1.20000 %FUEL= 45.454545 R,EQ.RATIO= 2.080338 PHI,EQ.RATIO= 2.080338

	CHAMBER	THROAT
Pinf/P	1.0000	1.8071
P, BAR	10.133	5.6072
T, K	2691.24	2386.58
RHO, KG/CU M	8.1468-1	5.0838-1
H, KJ/KG	534.74	-158.82
U, KJ/KG	-709.00	-1261.76
G, KJ/KG	-35700.3	-32291.9
S, KJ/(KG) (K)	13.4641	13.4641
M, (1/n)	17.991	17.991
Cp, KJ/(KG) (K)	2.2958	2.2558
GAMMA <sub>s</sub>	1.2520	1.2576
SON VEL,M/SEC	1247.9	1177.8
MACH NUMBER	0.000	1.000

## PERFORMANCE PARAMETERS

Ae/At	1.0000
CSTAR, M/SEC	1692.3
CF	0.6960
Ivac, M/SEC	2114.2
Isp, M/SEC	1177.8

## MASS FRACTIONS

*CO	0.24795	*CO2	0.04460	*H	0.00053
*H2	0.03340	H2O	0.22898	NH3	0.00001
*NO	0.00023	*N2	0.44233	*O	0.00003
*OH	0.00190	*O2	0.00002		

\* THERMODYNAMIC PROPERTIES FITTED TO 20000.K

NOTE. WEIGHT FRACTION OF FUEL IN TOTAL FUELS AND OF OXIDANT IN TOTAL OXIDANTS

## THEORETICAL ROCKET PERFORMANCE ASSUMING EQUILIBRIUM

## COMPOSITION DURING EXPANSION FROM INFINITE AREA COMBUSTOR

Pin = 147.0 PSIA

CASE = dmt\_\_\_\_\_

	REACTANT	WT FRACTION (SEE NOTE)	ENERGY KJ/KG-MOL	TEMP K
FUEL	CH6N2 (L) ,MMH	1.0000000	54200.000	298.150
OXIDANT	N2O4	1.0000000	0.000	0.000

O/F= 1.40000 %FUEL= 41.666667 R,EQ.RATIO= 1.783147 PHI,EQ.RATIO= 1.783147

	CHAMBER	THROAT
Pinf/P	1.0000	1.7708
P, BAR	10.133	5.7219



T, K	2910.05	2674.25
RHO, KG/CU M	8.0132-1	4.9552-1
H, KJ/KG	490.18	-200.92
U, KJ/KG	-774.29	-1355.65
G, KJ/KG	-37671.0	-35270.0
S, KJ/ (KG) (K)	13.1136	13.1136

M, (1/n)	19.135	19.256
(dLV/dLP)t	-1.00692	-1.00366
(dLV/dLT)p	1.1423	1.0808
Cp, KJ/ (KG) (K)	3.5314	2.9980
GAMMAs	1.1815	1.1970
SON VEL,M/SEC	1222.3	1175.7
MACH NUMBER	0.000	1.000

## PERFORMANCE PARAMETERS

Ae/At	1.0000
CSTAR, M/SEC	1739.3
CF	0.6760
Ivac, M/SEC	2157.9
Isp, M/SEC	1175.7

## MASS FRACTIONS

*CO	0.21664	0.21409
*CO2	0.05763	0.06163
*H	0.00094	0.00054
*H2	0.02403	0.02431
H2O	0.26217	0.26503
*NO	0.00112	0.00043
*N2	0.43042	0.43075
*O	0.00027	0.00007
*OH	0.00650	0.00306
*O2	0.00026	0.00007

\* THERMODYNAMIC PROPERTIES FITTED TO 20000.K

NOTE. WEIGHT FRACTION OF FUEL IN TOTAL FUELS AND OF OXIDANT IN TOTAL OXIDANTS

## THEORETICAL ROCKET PERFORMANCE ASSUMING FROZEN COMPOSITION

Pin = 147.0 PSIA

CASE = dmt\_\_\_\_\_

	REACTANT	WT FRACTION (SEE NOTE)	ENERGY KJ/KG-MOL	TEMP K
FUEL	CH6N2 (L) ,MMH	1.0000000	54200.000	298.150
OXIDANT	N2O4	1.0000000	0.000	0.000

O/F= 1.40000 %FUEL= 41.666667 R,EQ.RATIO= 1.783147 PHI,EQ.RATIO= 1.783147

	CHAMBER	THROAT
Pinf/P	1.0000	1.8001
P, BAR	10.133	5.6290
T, K	2910.05	2593.36
RHO, KG/CU M	8.0132-1	4.9953-1
H, KJ/KG	490.18	-211.99
U, KJ/KG	-774.29	-1338.85
G, KJ/KG	-37671.0	-34220.3
S, KJ/ (KG) (K)	13.1136	13.1136

M, (1/n)	19.135	19.135
Cp, KJ/ (KG) (K)	2.2340	2.1992
GAMMAs	1.2415	1.2462
SON VEL,M/SEC	1252.9	1185.0
MACH NUMBER	0.000	1.000





## PERFORMANCE PARAMETERS

Ae/At	1.0000
CSTAR, M/SEC	1711.7
CF	0.6923
Ivac, M/SEC	2135.9
Isp, M/SEC	1185.0

## MASS FRACTIONS

*CO	0.21664	*CO2	0.05763	*H	0.00094
*H2	0.02403	H2O	0.26217	*NO	0.00112
*N2	0.43042	*O	0.00027	*OH	0.00650
*O2	0.00026				

\* THERMODYNAMIC PROPERTIES FITTED TO 20000.K

NOTE. WEIGHT FRACTION OF FUEL IN TOTAL FUELS AND OF OXIDANT IN TOTAL OXIDANTS

## THEORETICAL ROCKET PERFORMANCE ASSUMING EQUILIBRIUM

## COMPOSITION DURING EXPANSION FROM INFINITE AREA COMBUSTOR

Pin = 147.0 PSIA

CASE = dmt\_\_\_\_\_

	REACTANT	WT FRACTION (SEE NOTE)	ENERGY KJ/KG-MOL	TEMP K
FUEL	CH6N2 (L), MMH	1.0000000	54200.000	298.150
OXIDANT	N2O4	1.0000000	0.000	0.000

O/F= 1.60000 %FUEL= 38.461538 R,EQ.RATIO= 1.560254 PHI,EQ.RATIO= 1.560254

	CHAMBER	THROAT
Pinf/P	1.0000	1.7526
P, BAR	10.133	5.7813
T, K	3059.15	2854.26
RHO, KG/CU M	8.0264-1	4.9533-1
H, KJ/KG	452.47	-229.04
U, KJ/KG	-809.93	-1396.20
G, KJ/KG	-38677.4	-36738.2
S, KJ/(KG) (K)	12.7911	12.7911

M, (1/n)	20.148	20.333
(dLV/dLP)t	-1.01302	-1.00809
(dLV/dLT)p	1.2624	1.1735
Cp, KJ/(KG) (K)	4.3967	3.7096
GAMMAs	1.1581	1.1678
SON VEL, M/SEC	1209.2	1167.5
MACH NUMBER	0.000	1.000

## PERFORMANCE PARAMETERS

Ae/At	1.0000
CSTAR, M/SEC	1752.1
CF	0.6663
Ivac, M/SEC	2167.2
Isp, M/SEC	1167.5

## MASS FRACTIONS

*CO	0.18744	0.18422
*CO2	0.07289	0.07795
*H	0.00122	0.00084
*H2	0.01692	0.01689
H2O	0.28144	0.28811



*NO	0.00328	0.00168
*N2	0.41968	0.42043
*O	0.00110	0.00046
*OH	0.01441	0.00869
*O2	0.00161	0.00071

\* THERMODYNAMIC PROPERTIES FITTED TO 20000.K

NOTE. WEIGHT FRACTION OF FUEL IN TOTAL FUELS AND OF OXIDANT IN TOTAL OXIDANTS

#### THEORETICAL ROCKET PERFORMANCE ASSUMING FROZEN COMPOSITION

Pin = 147.0 PSIA

CASE = dmt\_\_\_\_\_

	REACTANT	WT FRACTION (SEE NOTE)	ENERGY KJ/KG-MOL	TEMP K
FUEL	CH6N2 (L) ,MMH	1.0000000	54200.000	298.150
OXIDANT	N2O4	1.0000000	0.000	0.000

O/F= 1.60000 %FUEL= 38.461538 R,EQ.RATIO= 1.560254 PHI,EQ.RATIO= 1.560254

	CHAMBER	THROAT
Pinf/P	1.0000	1.7952
P, BAR	10.133	5.6441
T, K	3059.15	2735.53
RHO, KG/CU M	8.0264-1	4.9998-1
H, KJ/KG	452.47	-246.51
U, KJ/KG	-809.93	-1375.36
G, KJ/KG	-38677.4	-35237.0
S, KJ/(KG) (K)	12.7911	12.7911
M, (1/n)	20.148	20.148
Cp, KJ/(KG) (K)	2.1749	2.1438
GAMMA <sub>s</sub>	1.2342	1.2384
SON VEL,M/SEC	1248.2	1182.4
MACH NUMBER	0.000	1.000

#### PERFORMANCE PARAMETERS

Ae/At	1.0000
CSTAR, M/SEC	1714.0
CF	0.6898
Ivac, M/SEC	2137.1
Isp, M/SEC	1182.4

#### MASS FRACTIONS

*CO	0.18744	*CO2	0.07289	*H	0.00122
*H2	0.01692	H2O	0.28144	*NO	0.00328
*N2	0.41968	*O	0.00110	*OH	0.01441
*O2	0.00161				

\* THERMODYNAMIC PROPERTIES FITTED TO 20000.K

NOTE. WEIGHT FRACTION OF FUEL IN TOTAL FUELS AND OF OXIDANT IN TOTAL OXIDANTS

#### THEORETICAL ROCKET PERFORMANCE ASSUMING EQUILIBRIUM

#### COMPOSITION DURING EXPANSION FROM INFINITE AREA COMBUSTOR

Pin = 147.0 PSIA

CASE = dmt\_\_\_\_\_

	REACTANT	WT FRACTION (SEE NOTE)	ENERGY KJ/KG-MOL	TEMP K
FUEL	CH6N2 (L) ,MMH	1.0000000	54200.000	298.150



OXIDANT N2O4 1.0000000 0.000 0.000  
 O/F= 1.80000 %FUEL= 35.714286 R,EQ.RATIO= 1.386892 PHI,EQ.RATIO= 1.386892

	CHAMBER	THROAT
Pinf/P	1.0000	1.7389
P, BAR	10.133	5.8269
T, K	3148.38	2969.93
RHO, KG/CU M	8.1407-1	5.0194-1
H, KJ/KG	420.15	-245.11
U, KJ/KG	-824.52	-1405.99
G, KJ/KG	-38925.6	-37360.6
S, KJ/(KG) (K)	12.4971	12.4971
M, (1/n)	21.032	21.271
(dLV/dLP)t	-1.02066	-1.01484
(dLV/dLT)p	1.4113	1.3132
Cp, KJ/(KG) (K)	5.4098	4.7357
GAMMA <sub>s</sub>	1.1427	1.1461
SON VEL, M/SEC	1192.6	1153.5
MACH NUMBER	0.000	1.000

## PERFORMANCE PARAMETERS

Ae/At	1.0000
CSTAR, M/SEC	1750.1
CF	0.6591
Ivac, M/SEC	2159.9
Isp, M/SEC	1153.5

## MASS FRACTIONS

*CO	0.16044	0.15575
*CO2	0.08906	0.09644
*H	0.00129	0.00099
HO2	0.00001	0.00001
*H2	0.01188	0.01153
H2O	0.28862	0.29810
*NO	0.00678	0.00431
*N2	0.40970	0.41086
*O	0.00270	0.00154
*OH	0.02372	0.01693
*O2	0.00575	0.00352

\* THERMODYNAMIC PROPERTIES FITTED TO 20000.K

NOTE. WEIGHT FRACTION OF FUEL IN TOTAL FUELS AND OF OXIDANT IN TOTAL OXIDANTS

## THEORETICAL ROCKET PERFORMANCE ASSUMING FROZEN COMPOSITION

Pin = 147.0 PSIA  
 CASE = dmt\_\_\_\_\_

	REACTANT	WT FRACTION (SEE NOTE)	ENERGY KJ/KG-MOL	TEMP K
FUEL	CH6N2 (L) ,MMH	1.0000000	54200.000	298.150
OXIDANT	N2O4	1.0000000	0.000	0.000

O/F= 1.80000 %FUEL= 35.714286 R,EQ.RATIO= 1.386892 PHI,EQ.RATIO= 1.386892

	CHAMBER	THROAT
Pinf/P	1.0000	1.7921
P, BAR	10.133	5.6541
T, K	3148.38	2821.69
RHO, KG/CU M	8.1407-1	5.0686-1
H, KJ/KG	420.15	-267.67
U, KJ/KG	-824.52	-1383.19



G, KJ/KG	-38925.6	-35530.7
S, KJ/ (KG) (K)	12.4971	12.4971
M, (1/n)	21.032	21.032
Cp, KJ/ (KG) (K)	2.1192	2.0907
GAMMA <sub>s</sub>	1.2293	1.2332
SON VEL, M/SEC	1237.0	1172.9
MACH NUMBER	0.000	1.000

## PERFORMANCE PARAMETERS

Ae/At	1.0000
CSTAR, M/SEC	1704.4
CF	0.6881
Ivac, M/SEC	2124.0
Isp, M/SEC	1172.9

## MASS FRACTIONS

*CO	0.16044	*CO <sub>2</sub>	0.08906	*H	0.00129
HO <sub>2</sub>	0.00001	*H <sub>2</sub>	0.01188	H <sub>2</sub> O	0.28862
*NO	0.00678	*N <sub>2</sub>	0.40970	*O	0.00270
*OH	0.02372	*O <sub>2</sub>	0.00575		

\* THERMODYNAMIC PROPERTIES FITTED TO 20000.K

NOTE. WEIGHT FRACTION OF FUEL IN TOTAL FUELS AND OF OXIDANT IN TOTAL OXIDANTS

## THEORETICAL ROCKET PERFORMANCE ASSUMING EQUILIBRIUM

## COMPOSITION DURING EXPANSION FROM INFINITE AREA COMBUSTOR

Pin = 147.0 PSIA  
CASE = dmt\_\_\_\_\_

	REACTANT	WT FRACTION (SEE NOTE)	ENERGY KJ/KG-MOL	TEMP K
FUEL	CH <sub>6</sub> N <sub>2</sub> (L), MMH	1.0000000	54200.000	298.150
OXIDANT	N <sub>2</sub> O <sub>4</sub>	1.0000000	0.000	0.000

O/F= 2.00000 %FUEL= 33.333333 R,EQ.RATIO= 1.248203 PHI,EQ.RATIO= 1.248203

	CHAMBER	THROAT
Pinf/P	1.0000	1.7309
P, BAR	10.133	5.8538
T, K	3192.43	3030.36
RHO, KG/CU M	8.3199-1	5.1280-1
H, KJ/KG	392.14	-254.88
U, KJ/KG	-825.73	-1396.40
G, KJ/KG	-38652.2	-37317.1
S, KJ/ (KG) (K)	12.2303	12.2303

M, (1/n)	21.795	22.072
(dLV/dLP) <sub>t</sub>	-1.02750	-1.02216
(dLV/dLT) <sub>p</sub>	1.5434	1.4629
Cp, KJ/ (KG) (K)	6.2301	5.7575
GAMMA <sub>s</sub>	1.1342	1.1336
SON VEL, M/SEC	1175.3	1137.6
MACH NUMBER	0.000	1.000

## PERFORMANCE PARAMETERS

Ae/At	1.0000
CSTAR, M/SEC	1737.0
CF	0.6549
Ivac, M/SEC	2141.0
Isp, M/SEC	1137.6



## MASS FRACTIONS

*CO	0.13624	0.12983
*CO2	0.10434	0.11442
*H	0.00121	0.00097
HNO	0.00001	0.00000
HO2	0.00003	0.00002
*H2	0.00849	0.00796
H2O	0.28741	0.29791
*N	0.00001	0.00000
*NO	0.01100	0.00798
NO2	0.00001	0.00000
*N2	0.40050	0.40192
*O	0.00475	0.00322
*OH	0.03192	0.02512
*O2	0.01408	0.01064

\* THERMODYNAMIC PROPERTIES FITTED TO 20000.K

NOTE. WEIGHT FRACTION OF FUEL IN TOTAL FUELS AND OF OXIDANT IN TOTAL OXIDANTS

## THEORETICAL ROCKET PERFORMANCE ASSUMING FROZEN COMPOSITION

Pin = 147.0 PSIA

CASE = dmt\_\_\_\_\_

	REACTANT	WT FRACTION (SEE NOTE)	ENERGY KJ/KG-MOL	TEMP K
FUEL	CH6N2 (L) ,MMH	1.0000000	54200.000	298.150
OXIDANT	N2O4	1.0000000	0.000	0.000

O/F= 2.00000 %FUEL= 33.333333 R,EQ.RATIO= 1.248203 PHI,EQ.RATIO= 1.248203

	CHAMBER	THROAT
Pinf/P	1.0000	1.7900
P, BAR	10.133	5.6605
T, K	3192.43	2865.29
RHO, KG/CU M	8.3199-1	5.1785-1
H, KJ/KG	392.14	-280.04
U, KJ/KG	-825.73	-1373.11
G, KJ/KG	-38652.2	-35323.4
S, KJ/(KG) (K)	12.2303	12.2303
M, (1/n)	21.795	21.795
Cp, KJ/(KG) (K)	2.0677	2.0408
GAMMA <sub>s</sub>	1.2262	1.2299
SON VEL, M/SEC	1222.0	1159.5
MACH NUMBER	0.000	1.000

## PERFORMANCE PARAMETERS

Ae/At	1.0000
CSTAR, M/SEC	1687.5
CF	0.6871
Ivac, M/SEC	2102.2
Isp, M/SEC	1159.5

## MASS FRACTIONS

*CO	0.13624	*CO2	0.10434	*H	0.00121
HNO	0.00001	HO2	0.00003	*H2	0.00849
H2O	0.28741	*N	0.00001	*NO	0.01100
NO2	0.00001	*N2	0.40050	*O	0.00475
*OH	0.03192	*O2	0.01408		

\* THERMODYNAMIC PROPERTIES FITTED TO 20000.K



NOTE. WEIGHT FRACTION OF FUEL IN TOTAL FUELS AND OF OXIDANT IN TOTAL OXIDANTS

## References

- [1] Muhalim, Noor Muhammad Feizal B., and Subramaniam Krishnan, "*Design Of Nitrogen-Tetroxide/Monomethyl-Hydrazine Thruster For Upper Stage Application*", Department of Aeronautical Engineering, Faculty of Mechanical Engineering Universiti Teknologi Malaysia, 81310 Skudai, Malaysia
- [2] NASA "*NASA/Expedition 38*", <https://www.flickr.com/photos/nasa2explore/11936279413/>