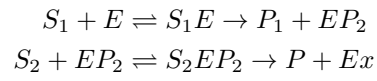


Bootcamp : Maths

Baptiste Rouger

September 19, 2016

Contents



$$V_0 \propto [S_1E] = a \cdot [E] \frac{[S_1]}{[S_1] + K}$$

Strong dissociation constant = lower ones.

Chemistry is an issue of energy and distribution of energy, which laws are described by thermodynamics.