





QUIZ 6. Wednesday November 9, 2016

MECH 221

READ THESE INSTRUCTIONS CAREFULLY BEFORE BEGINNING THE QUIZ. GOOD LUCK!

Duration: Target = 45 minutes. Maximum = 50 minutes.

Materials admitted: Pencil, eraser, straightedge, Mech2 calculator,

Mech2 formula book, one 3x5 inch index card.

There are five short-answer questions and two long-answer questions. *All questions must be answered.*

Provide **all** work and solutions **on these test pages**. Scrap paper is available for your use, but **will not be collected or marked**. An additional mark of up to 5% of the exam value is available for orderly presentation of work

Illegible work, or answers that do not include supporting calculations and explanations will NOT be marked.

BE SURE TO WRITE YOUR NAME ON THE TOP OF ALL EXAM PAGES

NAME:	Section
SIGNATURE:	
STUDENT NUMBER:	

Question	Mark Received	Maximum Mark
SA 1		6
SA 2		6
SA 3		7
SA 4		6
LA 1		25
LA 1		
Presentation		
TOTAL		

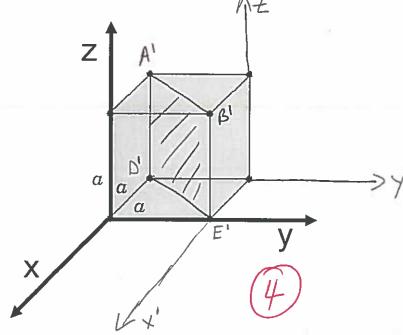
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Part I.

SA 1.

In the unit cell above, sketch the $(1\overline{1}0)$ plane.

If the crystal was body centre cubic, sketch the position of the atoms on the $(1\bar{1}0)$ plane assuming a hard sphere model.



First, it is most convenient to move the coordinate system to x', y' and z' as shown above.

Next, one needs to work backwards in this problem given (hkl). Take the reciprocals of hkl to determine the intercepts, i.e.

x intercept = 1

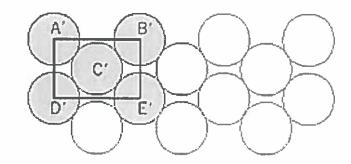
y intercept = -1

z intercept = ∞

This corresponds to the A'B'D'E' plane in the sketch above.

The corresponding positions of atoms on this plane for a BCC crystal is shown below:





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SA 2. Given that iron (Fe) has a body centered cubic crystal, a density of 7.86 gcm⁻³ and a molecular weight of 55.8 gmol⁻¹, determine the diameter of an iron atom (Assume hard sphere model).

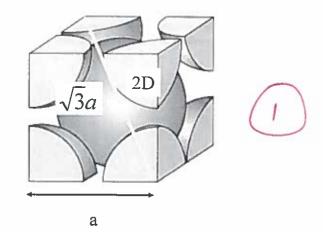
Unit Cell Length

a = lattice parameter

R = atomic radius

$$\sqrt{3}a = 2D$$

$$D = \frac{\sqrt{3}a}{2}$$



Density = ρ = (mass of atoms in unit cell)/(volume of unit cell)

$$\rho = \frac{nA / N_A}{V_c}$$

n = number of atoms per unit cell

A = Atomic weight g/mol

 V_c = unit cell volume = a^3

 N_A = Avogadro's Number (6.023 x 10²³ atoms/mol)

$$3 \qquad \alpha = \sqrt[3]{V_c} = 2.86 \times 10^{-8} \text{cm} \times \frac{m}{100 \text{cm}} \times \frac{nm}{10^{-9} \text{m}} \times \frac{nm}{10^{-9} \text{m}}$$

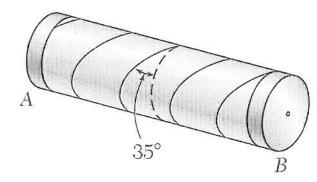
$$= 0.286 \text{ nm}$$

$$D = \sqrt{3} \times 0.286 \text{nm} = 0.248 \text{ nm}$$

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0.248 nm.

SA 3. A thin-walled cylinder of length L, diameter D, wall thickness t and interior pressure p is made by rolling and welding a plate into a 35° spiral, as shown. Derive a compact formula for the stress normal to the weld.



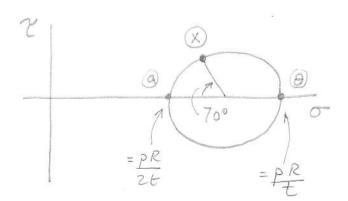
Cylinder stresses:

$$\frac{\sigma_{x} = \frac{\sigma_{x} + \sigma_{y}}{2} + \frac{\sigma_{x} - \sigma_{y}}{2} \cos 2\theta + Y_{xy} \sin^{2} 2\theta}{2} \\
= \frac{\frac{PR}{2t} + \frac{PR}{t}}{2} + \frac{\frac{PR}{2t} - \frac{PR}{t}}{2} \cos (-70^{\circ}) + 0$$

$$\sigma_{\chi} = \frac{3\rho R}{4t} - \frac{\rho R}{4t} \cos 70^{\circ}$$

stress normal to weld

Alternatively, Mohr's Circle could be used:



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SA 4. Why do we use safety factors? What issues do we typically need to consider when choosing an appropriate safety factor?

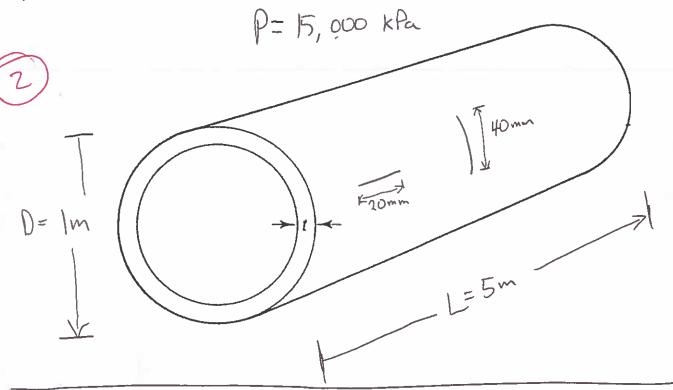
We use safety factors to protect engineering objects from the effects of factor that we cannot adequately consider in our design. Such factors include uncertainties in loads, operating environment and material performance. We choose to use higher safety factors when the consequences of failure are serious, e.g., human injury or large \$ loss. We choose to use lower safety factors when weight ar manufacturing cost are crucial.

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Part II.

- LA 1. A long cylinder made of 4140 steel (Length = 5 m, diameter = 1 m, wall thickness = 10 mm, E = 210 GPa, v = 0.3, yield stress = 1000 MPa) was designed to store nitrogen gas. After the cylinder was filled with gas it was noticed that the interior pressure gradually decreased. So that enough nitrogen gas would remain in the cylinder when needed, it was decided to add some more nitrogen in the cylinder by increasing the initial pressure. However, during this filling, the cylinder exploded at a pressure of 15,000 kPa. (No people or animals were injured in the construction of this question). A subsequent investigation revealed that there had initially been two through-thickness cracks, one parallel to the length of the cylinder with a total length 20 mm and one parallel to the circumference of the cylinder with a total length of 40 mm.
 - i. Sketch the cylinder before it fractured.
 - ii. Did the material of the cylinder yield? Explain your reasoning.
 - iii. Determine which of the cracks was most likely to have been the fracture initiation site. You may assume that Y=1.
 - iv. Estimate the fracture toughness, K_{Ic} , of the steel (in units of MPa \sqrt{m})
 - v. The steel used in the cylinder was strengthened by a combination of deformation at room temperature (strain hardening, reducing the grain size of the steel and solid solution hardening. Explain the basic principles that cause these mechanisms to strengthen the steel.
 - vi. Calculate the changes in length and diameter of the cylinder just before it had exploded.

i)



ii) Calculate hoop and axial stresses.

$$O_H = \frac{PO}{2t}$$
, $O_A = \frac{PO}{4t}$

= 750,000 kPa or 750mPa

The cylinder does not yield since the maximin Stress of 750 mpa is less than the yield Stress (1000 mpa). Note: this ignores the effect of the orthogonal stress.

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(ii) Need to calculate the stress intensity funtar for each of the two cracks.

Crack purelled to the leight, Stress = howp stress

K = OH JTTa

= 750 Wa JTI x 0.01 m

= 132.9 MPa Jun

Crack perpendicion to the lugh, stress = axial stres

K= OA JTIA

= 375MPa VT x0.02m

= 94.0 mA. Jm

As the crack parallel to the lagh of the Cylindr has the highest stress intisity factor, this should be the fractive initiation site.

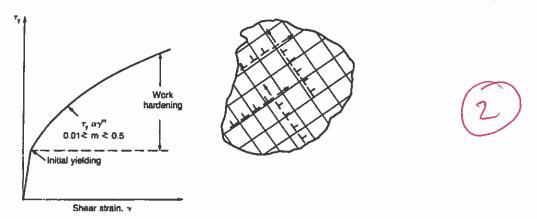
(2) KIC is defined as the stress in tusity at fructure in this case, fructure occurred at a K of 1329 MPa Jun. Therefore, this should be an estate of Kic

FEL= 132.9 MPa Vin

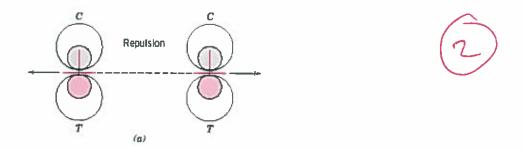
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v. The principle which underlies the strengthening of crystalline materials such as 4140 steel is to make dislocation motion more difficult.

Strain hardening



In this case, dislocation motion is made more difficult due to the interaction between dislocations on parallel or perpendicular planes. Either way, the stress to move the dislocation must increase due to this interaction. For example, there is a repulsive interaction between the two dislocations below. If the dislocation on the left is to move forward the stress will have to be increased due to the interaction with the dislocation on the left.



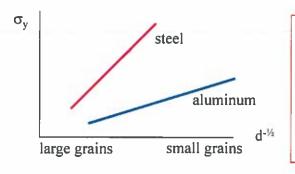
Grain size strengthening

For this case, the grain boundary acts as a barrier to the dislocation since the edge dislocation shown on the next page cannot be transmitted from Grain A to Grain B. This is increases the stress necessary to move the dislocation. Grain size strengthening is given by the Hall-Petch equation:

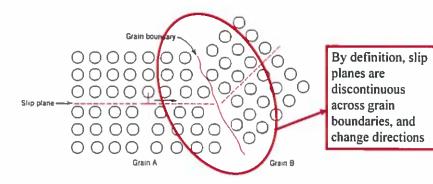
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$$\sigma_y = \sigma_0 + k_y d^{-1/2}$$

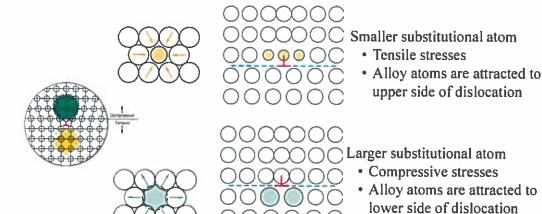


- Grain boundaries act as barrier to dislocations
- As grain size decreases, number of barriers to dislocations increases
- Therefore strength increases



Solid solution hardening

For this case, the elastic stress field of the dislocation interacts with either smaller or larger solute atoms dissolved in the lattice of the host.



- Two means by which there is an interaction
 - · Alloy atoms diffuse to dislocation and pin it down initially
 - Moving dislocation approaches solute atom

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(Vi)
$$\sigma_{\theta} = \frac{PR}{t}$$
 $\sigma_{\alpha} = \frac{PR}{2t}$ $\sigma_{r} = 0$ Cylinder stresses

$$\sigma_a = \frac{PR}{2t}$$

Hooke's Law:

$$\mathcal{E}_{\theta} = \frac{\sigma_{\theta}}{E} - \frac{\sigma_{\theta}}{E} - \frac{\sigma_{\theta}}{E} - \frac{\sigma_{\theta}}{E} = \frac{PR}{2Et} \left(2 - \nu \right)$$

$$\mathcal{E}_{\alpha} = \frac{\sigma_{\alpha}}{\varepsilon} - \nu \frac{\sigma_{\alpha}}{\varepsilon} - \nu \frac{\sigma_{\nu}}{\varepsilon} = \frac{PR}{28\varepsilon} \left(1 - 2\nu \right)$$