1 The Model

In biochemistry, Michaelis Menten kinetics is one of the simplest and best-known models of enzyme kinetics. It is named after German biochemist Leonor Michaelis and Canadian physician Maud Menten. The model can be illustrated as

$$E + S \xrightarrow{k_f} ES \xrightarrow{k_{cat}} E + P$$

Where E is the enzime, S is the substrate, ES is the enzyme binding to the substrate, and P is a product. This leads to the following system of differential equations:

$$\frac{d[S]}{dt} = -k_f[E][S] + k_r[ES]] \tag{1}$$

$$\frac{d[E]}{dt} = -k_f[E][S] + k_r[ES] + k_{cat}[ES] \tag{2}$$

$$\frac{d[ES]}{dt} = k_f[E][S] - k_r[ES] - k_{cat}[ES]$$
(3)

$$\frac{d[P]}{dt} = k_{cat}[ES] \tag{4}$$

(5)

2 Simplifying the Model

Now we make some simplifications:

- If the enzyme is conserved, then $[E] + [ES] = [E]_0$ is a constant.
- In situations where $\frac{d[P]}{dt}$ is constant (3) and (4) imply that $\frac{d[ES]}{dt} = 0$. This is often used as an approximation when $\frac{d[P]}{dt}$ is nearly constant.
- From this it follows that

$$k_{f}[E][S] = (k_{r} + k_{cat})[ES]$$

$$k_{f}([E]_{0} - [ES])[S] = (k_{r} + k_{cat})[ES]$$

$$([E]_{0} - [ES])[S] = \frac{k_{r} + k_{cat}}{k_{f}}[ES]$$

$$[E]_{0}[S] = \left(\frac{k_{r} + k_{cat}}{k_{f}} + [S]\right)[ES]$$

$$[ES] = \frac{[E]_{0}[S]}{\frac{k_{r} + k_{cat}}{k_{f}} + [S]}$$

$$k_{cat}[ES] = \frac{k_{cat}[E]_{0}[S]}{\frac{k_{r} + k_{cat}}{k_{f}} + [S]}$$

$$v = \frac{d[P]}{dt} = \frac{k_{cat}[E]_{0}[S]}{\frac{k_{r} + k_{cat}}{k_{f}} + [S]}$$

• Ignoring the meaning of the constants (for the moment) and focusing on the form, we now have the following sort of relationship between v and S

$$v = \frac{\alpha[S]}{\beta + [S]} \tag{6}$$

That is, the "velocity" of the reaction (rate at which the product is produced) is determined by the concentration of the substrate and constants that do not depend on v or [S].

Equation (6) is not in our favorite linear form, but we can use transforamtions to get into a linear form:

$$\frac{1}{v} = \frac{\beta + [S]}{\alpha [S]} = \frac{1}{\alpha} + \frac{\beta}{\alpha} \frac{1}{[S]}$$

3 Using Data to Fit the Model

This now provides an experimental way to estimate the constants α and β (and from them to infer things about k_f , k_r , and k_{cat} .) We can gather data providing values of v and [S] and fit $\frac{1}{v}$ as a linear function of $\frac{1}{[S]}$. Let's do that using some data from a lab conducted by students at Calvin college.

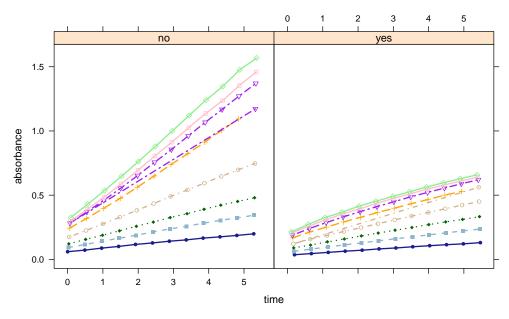
```
mm <- read.csv("http://www.calvin.edu/~rpruim/data/calvin/michaelis-menten.csv")</pre>
summary (mm)
       batch
                                    absorbance
                                                      substrate
                                                                     inhibitor
                       time
 SML-1 A1 : 12
                                          :0.0362
                                                            :0.100
                                                                     no:96
                  Min.
                         :0.00
                                                                     yes:96
 SML-10 B2: 12
                  1st Qu.:1.37
                                  1st Qu.:0.1877
                                                    1st Qu.:0.275
 SML-11 B3: 12
                  Median:2.74
                                  Median :0.3318
                                                    Median : 0.500
 SML-12 B4: 12
                  Mean
                         :2.74
                                          :0.4300
                                                           :1.075
                                  Mean
                                                    Mean
 SML-13 B5: 12
                  3rd Qu.:4.10
                                  3rd Qu.:0.5569
                                                    3rd Qu.:1.625
 SML-14 B6: 12
                  Max.
                         :5.47
                                  Max.
                                          :1.5677
                                                    Max.
                                                            :3.000
 (Other) :120
```

3.1 Dealing with Indirect Measurements

The measurements are a bit indirect. The amount of product is inferred from the absorbance, and v is inferred by the rate of change in absorbance, which should be a roughly linear function of time (in minutes) for each value of [S] (substrate, mM peroxide) if our assumptions are true. Notice too that the experiment was run with and without an inhibitor.

Our first step is to estimate the values of v for each level of substrate by fitting a simple linear model and determining the slope. We begin by plotting the data to confirm that our linearity assumptions seem appropriate.

```
xyplot(absorbance ~ time | inhibitor, data = mm, groups = factor(substrate),
    type = c("p", "l"))
```



Not bad for student-collected data.

Now we need to estimate all those slopes. We can do this all at once with a clever choice of model. For the following analysis, we'll use only the data without the inhibitor.

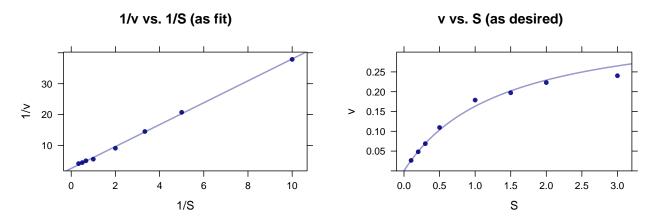
```
mm$S <- factor(mm$substrate)</pre>
mmno <- subset(mm, inhibitor == "no")</pre>
slope.model <- lm(absorbance ~ time * S, data = mmno)</pre>
summary(slope.model)
Call:
lm(formula = absorbance ~ time * S, data = mmno)
Residuals:
    Min
           1Q
                Median
                             3Q
                                    Max
-0.14643 -0.00300 0.00011 0.00270 0.04913
Coefficients:
          Estimate Std. Error t value Pr(>|t|)
(Intercept) 0.06214 0.01075 5.78 1.4e-07
          0.02641 0.00345 7.66 3.7e-11
S0.2
          0.03173 0.01523 2.08 0.04041
          0.05935 0.01526 3.89 0.00021
S0.3
S0.5
          S1
          0.15577 0.01562 9.97 1.1e-15
S1.5
          0.20656 0.01516 13.63 < 2e-16
          0.19780 0.01535 12.88 < 2e-16
S2
          0.22729 0.01538 14.78 < 2e-16
S3
time:S0.2 0.02179 0.00488 4.47 2.6e-05
time:S0.3 0.04235 0.00488 8.68 3.7e-13
time:S0.5 0.08312 0.00488 17.04 < 2e-16
          0.15245 0.00523 29.14 < 2e-16
time:S1
time:S1.5 0.17090 0.00468 36.54 < 2e-16
           0.19665
                     0.00488 40.31 < 2e-16
time:S2
           0.21396
time:S3
                     0.00488
                              43.86 < 2e-16
Residual standard error: 0.0198 on 80 degrees of freedom
Multiple R-squared: 0.998, Adjusted R-squared: 0.998
F-statistic: 2.56e+03 on 15 and 80 DF, p-value: <2e-16
coef(slope.model)
                          S0.2
                                     S0.3
                                                 S0.5
                                                             S1
(Intercept)
                time
                                   0.0594
                                             0.1058
    0.0621
               0.0264
                          0.0317
                                                          0.1558
      S1.5
                           S3 time:S0.2 time:S0.3 time:S0.5
                 S2
    0.2066
             0.1978
                        0.2273
                                  0.0218
                                             0.0424
                                                          0.0831
   time:S1
            time:S1.5
                         time:S2
                                    time:S3
    0.1525
               0.1709
                      0.1966
                                 0.2140
mmSlopes <- data.frame(S = as.numeric(levels(mmno$S)), v = coef(slope.model)["time"] +
   c(`time:S0.1` = 0, coef(slope.model)[paste("time:S", levels(mmno$S)[-1],
       sep = "")]))
mmSlopes
time:S0.1 0.1 0.0264
time:S0.2 0.2 0.0482
time:S0.3 0.3 0.0688
```

```
time:S0.5 0.5 0.1095
time:S1 1.0 0.1789
time:S1.5 1.5 0.1973
time:S2 2.0 0.2231
time:S3 3.0 0.2404
```

3.2 Fitting with Ordinary Least Squares

Now we can fit our Michaelis-Menten model:

```
mmModel \leftarrow lm((1/v) \sim I(1/S), mmSlopes)
summary(mmModel)
Call:
lm(formula = (1/v) ~ I(1/S), data = mmSlopes)
Residuals:
 Min 1Q Median 3Q Max
-0.544 -0.239 0.119 0.209 0.449
Coefficients:
          Estimate Std. Error t value Pr(>|t|)
(Intercept) 2.5915 0.1973 13.1 1.2e-05
I(1/S)
          3.5408 0.0468 75.6 3.6e-10
Residual standard error: 0.41 on 6 degrees of freedom
Multiple R-squared: 0.999, Adjusted R-squared: 0.999
F-statistic: 5.72e+03 on 1 and 6 DF, p-value: 3.61e-10
alpha.hat <- 1/coef(mmModel)[1]</pre>
beta.hat <- coef(mmModel)[2] * alpha.hat</pre>
c(alpha.hat = alpha.hat, beta.hat = beta.hat)
alpha.hat.(Intercept) beta.hat.I(1/S)
                0.386
                           1.366
f <- makeFun(mmModel)</pre>
xyplot(1/v \sim 1/S, mmSlopes, main = "1/v vs. 1/S (as fit)")
g \leftarrow makeFun(f(1/x) \sim x)
plotFun(g(x) \sim x, add = TRUE, col = "navy", alpha = 0.4)
xyplot(v \sim S, mmSlopes, ylim = c(0, 0.3), main = "v vs. S (as desired)")
plotFun(1/f(S) \sim S, add = TRUE, col = "navy", alpha = 0.4)
```



Doing this we see some issues. Although there is a nice tight fit of the transformed data to the least squares regression line, the fit doesn't look nearly as good after back-transforming to the original scales. In particular, errors are much larger for larger values of S. Or thought about the other way around, the way we have fit the model has forced the fit to be extremely good for small values of S at the cost of allowing much poorer fits for larger values of S. This is because the transformations transform the scale for the residuals as well as for the inputs. Note too that the distribution of values of 1/S is not nearly as uniform as it is for S.

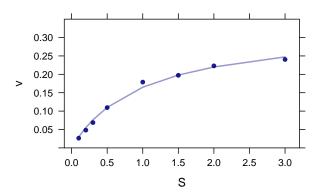
3.3 Fitting with Nonlinear Least Squares

We can do better if we use nonlinear least squares instead of tranforming and using least squares. In non-linear least squares we fit a parameterized function of arbitrary form by determining (well, estimating anyway) the values of the parameters that minimize the sum of the squares of the residuals

$$SS(\alpha, \beta) = \sum_{i=1}^{n} (v_i - f(\alpha, \beta; S_i))^2$$

This approach will be less forgiving of such large residuals for large values of S. We lose something in the exchange, however. It is not longer the case that simple closed-form formulas exist for the estimates. This means that we will need to rely on numerical estimation. Furthermore the estimators are not guaranteed to be unbiased.

```
# we provide nls() with our estimates from above as a starting point.
model.nls <- nls(v ~ alpha * S/(beta + S), data = mmSlopes, start = list(alpha = alpha.hat,
    beta = beta.hat))
summary(model.nls)
Formula: v ~ alpha * S/(beta + S)
Parameters:
      Estimate Std. Error t value Pr(>|t|)
alpha
        0.3297
                   0.0163
                             20.24
beta
        0 9979
                   0 1191
                              8.38
                                   0.00016
Residual standard error: 0.00782 on 6 degrees of freedom
Number of iterations to convergence: 5
Achieved convergence tolerance: 6.84e-06
xyplot(v \sim S, mmSlopes, ylim = c(0, 0.35))
ladd(panel.xyplot(mmSlopes$S, predict(model.nls), type = "1", col = "navy",
    alpha = 0.4))
```



As expected, the fit is much better for larger values of S and we've lost very little for smaller values of S. Now that we have (two sets of) estimates for α and β , we should pause a moment to see what these parameters tell us about the chemistry. Recall Equation (6)

$$v = \frac{\alpha[S]}{\beta + [S]}$$

As [S] increases,

$$\frac{\alpha[S]}{\beta + [S]} \nearrow \alpha$$

so α gives the horizontal asymptote representing the maximum velocity. And if $[S] = \beta$, we get

$$v = \frac{\alpha[S]}{[S] + [S]} = \frac{\alpha}{2} ,$$

so β is the value of [S] that gives half the maximum velocity.

linear least squares fit

0.4 0.3 > 0.2 0.1 0.0 0.5 1.0 1.5 2.0 2.5 3.0 S

non-linear least squares fit

