

1 The Michaelis-Menten Model

From *The Analysis of Biological Data* by Whitlock and Shluter, page 488–89

The problem with the curve in the left panel of Figure 18.8-1 [using an interpolater like `spliner()`] is that it would probably do a **terrible job of predicting** any *new* observations obtained from the same population **because the curve does not describe the general trend**. Such a complicated curve is **hard to justify biologiclly**...

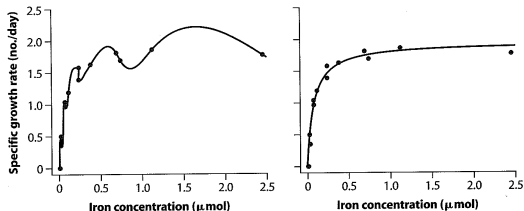


Figure 17.8-1 Population growth rate of a species of phytoplankton in culture in relation to the concentration of iron in the medium (the data are from Sunda and Huntsman 1997). The curve in the left panel is an arbitrarily complex function that passes through all of the data points. The curve in the right panel is a Michaelis–Menten curve that fits the data more simply.

reason to think that all the peaks and dips in this curve truly reflect the effects of iron on the growth of phytoplankton?

Greater simplicity, as demonstrated by the fitted curve in the right panel of Figure 17.8-1, solves both of these problems. The data are the same as in the left panel but this time we've fit the much simpler function,

$$Y = \frac{aX}{b + X}$$

This is the *Michaelis–Menten* equation used frequently in biochemistry. The curve

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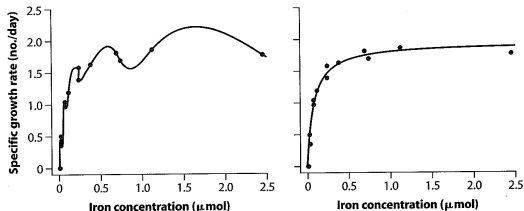


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Where does this come from?

1.1 Some background

In biochemistry, Michaelis-Menten kinetics is one of the simplest and best-known models of enzyme kinetics. It is named after German biochemist Leonor Michaelis and Canadian physician Maud Menten. The reaction can be illustrated as



where E is the enzyme, S is the substrate, ES is the enzyme binding to the substrate, and P is a product.

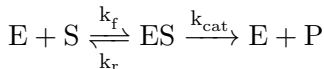
Question: How can we model this?

1.2 Building a Mathematical Model



- $\frac{d[E]}{dt}$ depends on ??
- $\frac{d[S]}{dt}$ depends on ??
- $\frac{d[ES]}{dt}$ depends on ??
- $\frac{d[P]}{dt}$ depends on ??

This leads to the following system of differential equations:



$$\frac{d[S]}{dt} = -k_f[E][S] + k_r[ES] \quad (1)$$

$$\frac{d[E]}{dt} = -k_f[E][S] + k_r[ES] + k_{cat}[ES] \quad (2)$$

$$\frac{d[ES]}{dt} = k_f[E][S] - k_r[ES] - k_{cat}[ES] \quad (3)$$

$$\frac{d[P]}{dt} = k_{cat}[ES] \quad (4)$$

2 Simplifying the Model

Now we make some simplifications:

- If the enzyme is conserved, then $[E] + [ES] = [E]_0$ is a constant.
- In situations where $\frac{d[P]}{dt}$ is constant (3) and (4) imply that $\frac{d[ES]}{dt} = 0$. This is often used as an approximation when $\frac{d[P]}{dt}$ is nearly constant.

- From this it follows that

$$k_f[E][S] = (k_r + k_{cat})[ES]$$

$$k_f([E]_0 - [ES])[S] = (k_r + k_{cat})[ES]$$

$$([E]_0 - [ES])[S] = \frac{k_r + k_{cat}}{k_f}[ES]$$

$$[E]_0[S] = \left(\frac{k_r + k_{cat}}{k_f} + [S] \right) [ES]$$

$$[ES] = \frac{[E]_0[S]}{\frac{k_r + k_{cat}}{k_f} + [S]}$$

$$k_{cat}[ES] = \frac{k_{cat}[E]_0[S]}{\frac{k_r + k_{cat}}{k_f} + [S]}$$

$$v = \frac{d[P]}{dt} = \frac{k_{cat}[E]_0[S]}{k_f + k_r + [S]}$$

- From

$$v = \frac{d[P]}{dt} = \frac{k_{cat}[E]_0[S]}{\frac{k_r+k_{cat}}{k_f} + [S]}$$

ignoring the meaning of the constants (for the moment) and focusing on the form, we now have the following sort of relationship between v and $[S]$

$$v = \frac{\alpha[S]}{\beta + [S]} \quad (5)$$

That is, the “velocity” of the reaction (rate at which the product is produced) is determined by the concentration of the substrate and constants that do not depend on v or $[S]$.

Equation (5) is not in our favorite linear form, but we can use transformations to get into a linear form:

$$\frac{1}{v} = \frac{\beta + [S]}{\alpha[S]} = \frac{1}{\alpha} + \frac{\beta}{\alpha} \frac{1}{[S]}$$

3 Using Data to Fit the Model

This now provides an experimental way to estimate the constants α and β (and from them to infer things about k_f , k_r , and k_{cat} .) We can gather data providing values of v and $[S]$ and fit $\frac{1}{v}$ as a linear function of $\frac{1}{[S]}$. Let's do that using some data from a lab conducted by students at Calvin college.

```
mm <-
```

```
read.csv("http://www.calvin.edu/~rpruim/data/cal  
summary(mm)
```

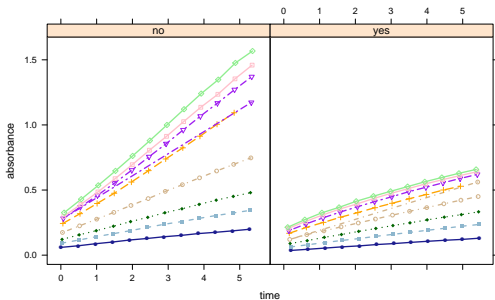
	batch	time	absorbance
SML-1	A1 : 12	Min. :0.00	Min. :0.036
SML-10	B2: 12	1st Qu.:1.37	1st Qu.:0.188
SML-11	B3: 12	Median :2.74	Median :0.332
SML-12	B4: 12	Mean :2.74	Mean :0.430
SML-13	B5: 12	3rd Qu.:4.10	3rd Qu.:0.557
SML-14	B6: 12	Max. :5.47	Max. :1.568
(Other)	:120		

3.1 Dealing with Indirect Measurements

The measurements are a bit indirect. The amount of product is inferred from the absorbance, and v is inferred by the rate of change in absorbance, which should be a roughly linear function of **time** (in minutes) for each value of $[S]$ (**substrate**, mM peroxide) if our assumptions are true. Notice too that the experiment was run with and without an inhibitor.

Our first step is to estimate the values of v for each level of **substrate** by fitting a simple linear model and determining the slope. We begin by plotting the data to confirm that our linearity assumptions seem appropriate.

```
xyplot(absorbance ~ time | inhibitor, data =  
mm, groups = factor(substrate),  
type = c("p", "l"))
```



Not bad for student-collected data.

Now we need to estimate all those slopes. We can do this all at once with a clever choice of model. For the following analysis, we'll use only the data without the inhibitor.

```
mm$S <- factor(mm$substrate)
mmno <- subset(mm, inhibitor == "no")
slope.model <- lm(absorbance ~ time * S, data
= mmno)
summary(slope.model)
```

Call:


```
lm(formula = absorbance ~ time * S, data = mmno)
```

Residuals:

Min	1Q	Median	3Q	Max
-0.14643	-0.00300	0.00011	0.00270	0.04913

Coefficients:

	Estimate	Std. Error	t value	Pr(> t)
(Intercept)	0.06214	0.01075	5.78	1.4e-07
time	0.02641	0.00345	7.66	3.7e-11
S0.2	0.03173	0.01523	2.08	0.04041
S0.3	0.05935	0.01526	3.89	0.00021
S0.5	0.10578	0.01528	6.92	1.0e-09

S1	0.15577	0.01562	9.97	1.1e-15
S1.5	0.20656	0.01516	13.63	< 2e-16
S2	0.19780	0.01535	12.88	< 2e-16
S3	0.22729	0.01538	14.78	< 2e-16
time:S0.2	0.02179	0.00488	4.47	2.6e-05
time:S0.3	0.04235	0.00488	8.68	3.7e-13
time:S0.5	0.08312	0.00488	17.04	< 2e-16
time:S1	0.15245	0.00523	29.14	< 2e-16
time:S1.5	0.17090	0.00468	36.54	< 2e-16
time:S2	0.19665	0.00488	40.31	< 2e-16
time:S3	0.21396	0.00488	43.86	< 2e-16

Residual standard error: 0.0198 on 80 degrees of

Multiple R-squared: 0.998, Adjusted R-squared: 0.
F-statistic: 2.56e+03 on 15 and 80 DF, p-value:

```
coef(slope.model)
```

(Intercept)	time	S0.2	S0.3
0.0621	0.0264	0.0317	0.0594
S1.5	S2	S3	time:S0.2
0.2066	0.1978	0.2273	0.0218
time:S1	time:S1.5	time:S2	time:S3
0.1525	0.1709	0.1966	0.2140

```
mmSlopes <- data.frame(S =  
as.numeric(levels(mmno$S)), v =  
coef(slope.model)["time"] +  
  c(`time:S0.1` = 0,  
coef(slope.model)[paste("time:S",  
levels(mmno$S)[-1],  
  sep = "")]))
```

mmSlopes

	S	v
time:S0.1	0.1	0.0264
time:S0.2	0.2	0.0482
time:S0.3	0.3	0.0688
time:S0.5	0.5	0.1095

```
time:S1    1.0 0.1789
time:S1.5  1.5 0.1973
time:S2    2.0 0.2231
time:S3    3.0 0.2404
```

3.2 Fitting with Ordinary Least Squares

Now we can fit our Michaelis-Menten model:

```
mmModel <- lm((1/v) ~ I(1/S), mmSlopes)
summary(mmModel)
```

Call:

```
lm(formula = (1/v) ~ I(1/S), data = mmSlopes)
```

Residuals:

Min	1Q	Median	3Q	Max
-0.544	-0.239	0.119	0.209	0.449

Coefficients:

	Estimate	Std. Error	t value	Pr(> t)
(Intercept)	2.5915	0.1973	13.1	1.2e-05
I(1/S)	3.5408	0.0468	75.6	3.6e-10

Residual standard error: 0.41 on 6 degrees of fr

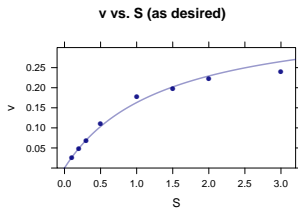
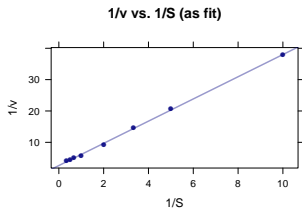
Multiple R-squared: 0.999, Adjusted R-squared: 0.
F-statistic: 5.72e+03 on 1 and 6 DF, p-value: 3

```
alpha.hat <- 1/coef(mmModel)[1]  
beta.hat <- coef(mmModel)[2] * alpha.hat  
c(alpha.hat = alpha.hat, beta.hat = beta.hat)
```

alpha.hat.(Intercept)	beta.hat.I(1/S)
0.386	1.366

```
f <- makeFun(mmModel)  
xyplot(1/v ~ 1/S, mmSlopes, main = "1/v vs.  
1/S (as fit)")
```

```
g <- makeFun(f(1/x) ~ x)
plotFun(g(x) ~ x, add = TRUE, col = "navy",
alpha = 0.4)
xyplot(v ~ S, mmSlopes, ylim = c(0, 0.3),
main = "v vs. S (as desired)")
plotFun(1/f(S) ~ S, add = TRUE, col = "navy",
alpha = 0.4)
```

Doing this we see some issues. Although there is a nice tight fit of the transformed data to the least squares regression line, the fit doesn't look nearly as good after back-transforming to the original scales. In particular, errors are much larger for larger values of S . Or thought about the other way around, the way we have fit the model has forced the fit to be extremely good for

small values of S at the cost of allowing much poorer fits for larger values of S . This is because the transformations transform the scale for the residuals as well as for the inputs. Note too that the distribution of values of $1/S$ is not nearly as uniform as it is for S .

3.3 Fitting with Nonlinear Least Squares

We can do better if we use nonlinear least squares instead of transforming and using least squares. In nonlinear least squares we fit a parameterized function of arbitrary form by determining (well, estimating anyway) the values of the parameters that minimize the

sum of the squares of the residuals

$$SS(\alpha, \beta) = \sum_{i=1}^n (v_i - f(\alpha, \beta; S_i))^2$$

This approach will be less forgiving of such large residuals for large values of S . We lose something in the exchange, however. It is no longer the case that simple closed-form formulas exist for the estimates. This means that we will need to rely on numerical estimation. Furthermore the estimators are not guaranteed to be unbiased.

we provide nls() with our estimates from above as a starting point.

```
model.nls <- nls(v ~ alpha * S/(beta + S),  
data = mmSlopes, start = list(alpha =  
alpha.hat,  
beta = beta.hat))  
summary(model.nls)
```

Formula: $v \sim \alpha * S / (\beta + S)$

Parameters:

	Estimate	Std. Error	t value	Pr(> t)
alpha	0.3297	0.0163	20.24	9.4e-07

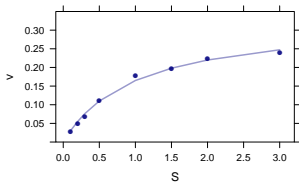
beta	0.9979	0.1191	8.38	0.00016
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Residual standard error: 0.00782 on 6 degrees of freedom

Number of iterations to convergence: 5

Achieved convergence tolerance: 6.84e-06

```
xyplot(v ~ S, mmSlopes, ylim = c(0, 0.35))  
ladd(panel.xyplot(mmSlopes$S,  
predict(model.nls), type = "l", col = "navy",  
alpha = 0.4))
```



As expected, the fit is much better for larger values of S and we've lost very little for smaller values of S .

Now that we have (two sets of) estimates for α and β , we should pause a moment to see what these parameters tell us about the chemistry. Recall Equation

(5)

$$v = \frac{\alpha[S]}{\beta + [S]}$$

As $[S]$ increases,

$$\frac{\alpha[S]}{\beta + [S]} \nearrow \alpha$$

so α gives the horizontal asymptote representing the maximum velocity. And if $[S] = \beta$, we get

$$v = \frac{\alpha[S]}{[S] + [S]} = \frac{\alpha}{2} ,$$

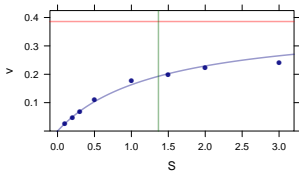
so β is the value of $[S]$ that gives half the maximum velocity.

```
xyplot(v ~ S, mmSlopes, ylim = c(0, 1.1 *  
alpha.hat), main = "linear least squares  
fit")  
plotFun(1/f(S) ~ S, add = TRUE, col = "navy",  
alpha = 0.4)  
ladd(panel.abline(h = alpha.hat, col = "red",  
alpha = 0.4))  
ladd(panel.abline(v = beta.hat, col =  
"darkgreen", alpha = 0.4))  
xyplot(v ~ S, mmSlopes, ylim = c(0, 1.1 *  
alpha.hat), main = "non-linear least squares
```



```
fit")
ladd(panel.xyplot(mmSlopes$S,
predict(model.nls), type = "l", col = "navy",
alpha = 0.4))
ladd(panel.abline(h = coef(model.nls)[1], col
= "red", alpha = 0.4))
ladd(panel.abline(v = coef(model.nls)[2], col
= "darkgreen", alpha = 0.4))
```

linear least squares fit



non-linear least squares fit

