CH3CHO +1 CH3. + CHO.

 $CH_3 \cdot + CH_3CHO \xrightarrow{k_2} CH_3 \cdot + CO + CH_4$ $CHO \cdot + CH_3CHO \xrightarrow{k_3} CH_3 \cdot + 2CO + H_2$ $2CH_3 \cdot \xrightarrow{k_4} C_2H_6$

- rAC = r, + r2+r3

= k, CAC + k2 CAC CCH3. + k8 CAC CCHO.

= CAC [K,+ k2CCH3. +kg CHO.] - 1

Active Intermediate

CH3. , CHO.

 $\frac{-r_{C}}{-r_{CH_{3}}} = -r_{1} + r_{2}^{\prime} - r_{3}^{\prime} - r_{3}^{\prime} + \frac{1}{2}r_{4}^{\prime}$

0 = -k, CAC - K3 CCHO. CAC + 1 k4 CCH3. -2

- TCHO. = - 1, + 53

0 = - K, CAC + K3 CCHO, CAC

 \neq CCHO, $=\frac{k_1}{k_2}$

$$C_{CH_3} = 2 C_{AC} \left(\frac{k_1}{K_4}\right)^{1/2}$$
 4

