

Certified Reagent Operation Form First-year Chemistry

☐ Pipetting a Liquid ☐ Mass of liquid transferred is measured on a balance and documented properly (assume a density of 1.00 g/mL) ☐ All measured volumes are within ±5% with respect to the target volume ☐ No liquid is spilled on the bench surface ☐ No obvious errors are committed, particularly contamination	 ☐ Weighing a Solid ☐ Measured mass of solid is documented properly ☐ No solid is spilled in the balance ☐ Solid is transferred <i>only</i> off the balance and spillage is minimal ☐ All measured masses are within ±5% with respect to the target mass (or, all masses round to the target mass in the protocol) ☐ No obvious errors are committed
□ Volumetric Preparation of a Solution □ Stock solution or solid solute is transferred to volumetric flask without spillage □ Solvent is added to the volumetric flask to bring the solution volume <i>exactly</i> to the mark. Reject the operation if the bottom of the meniscus is not sitting on the mark! □ Flask is capped and inverted three times to mix the solution □ Completed solution is free of debris and homogeneous	□ Obtaining a Visible Absorption Spectrum □ Deionized water blank is obtained □ Cuvette is filled at least 2/3 full with the solution to be analyzed □ Cuvette is appropriately aligned with the incident light beam □ Wavelength of maximum absorbance (with units) and maximum absorbance value are appropriately documented □ Spectrometer is returned to the LabQuest box at the end of lab
☐ Titration to an Endpoint ☐ Buret is filled and titrant is delivered without spilling ☐ Initial and final volumes are recorded to appropriate precision ☐ Titration is halted <i>exactly</i> at the endpoint (i.e. the endpoint is not overshot) ☐ Re-filling the buret with titrant is not necessary	☐ Heating a Solid with a Bunsen Burner ☐ Apparatus includes crucible with lid, clay triangle, ring stand and ring, Bunsen burner, and tubing ☐ Initial and final masses are recorded to appropriate precision ☐ Bunsen burner is set to produce a flame of moderate size with a clearly visible inner blue cone ☐ Crucible is positioned at the tip of the inner blue cone of the flame
Operator's Name	Signature
Observer's Name	Signature
TA's Signature	
Today's Date	

Submission Instructions. The student performing the operation should scan and submit completed form(s) for the current experiment using the Certified Reagent Operations assignment for the current experiment on Canvas.