

# JEE Advanced Exam 2017

## (Paper & Solution)

Date : 21 / 05 / 2017

### PAPER-2

### PART-I (PHYSICS)

#### SECTION – 1 (Maximum Marks : 21)

- This section contains **SEVEN** questions
- Each question has **FOUR** options (A), (B), (C) and (D). **ONLY ONE** of these four options is correct.
- For each question, darken the bubble corresponding to the correct option in the ORS.
- For each question, marks will be awarded in one of the following categories:

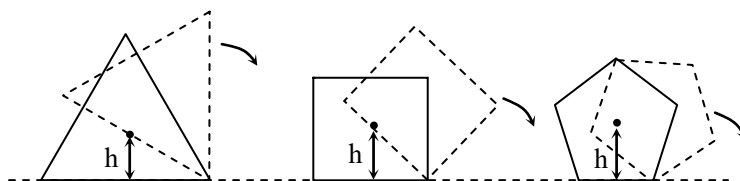
Full Marks	: +3	If only the bubble corresponding to the correct option is darkened.
Zero Marks	: 0	If none of the bubbles is darkened.
Negative Marks	: –1	In all other cases.

**Q.1** A photoelectric material having work-function  $\phi_0$  is illuminated with light of wavelength  $\lambda \left( \lambda < \frac{hc}{\phi_0} \right)$ .

The fastest photoelectron has a de Broglie wavelength  $\lambda_d$ . A change in wavelength of the incident light by  $\Delta\lambda$  results in a change  $\Delta\lambda_d$  in  $\lambda_d$ . Then the ratio  $\Delta\lambda_d/\Delta\lambda$  is proportional to

- (A)  $\lambda_d^3/\lambda$                       (B)  $\lambda_d/\lambda$                       (C)  $\lambda_d^2/\lambda^2$                       (D)  $\lambda_d^3/\lambda^2$

- Q.2** Consider regular polygons with number of sides  $n = 3, 4, 5, \dots$  as shown in the figure. The centre of mass of all the polygons is at height  $h$  from the ground. They roll on a horizontal surface about the leading vertex without slipping and sliding as depicted. The maximum increase in height of the locus of the centre of mass for each polygon is  $\Delta$ . Then  $\Delta$  depends on  $n$  and  $h$  as



- (A)  $\Delta = h \left( \frac{1}{\cos\left(\frac{\pi}{n}\right)} - 1 \right)$       (B)  $\Delta = h \tan^2 \left( \frac{\pi}{2n} \right)$       (C)  $\Delta = h \sin \left( \frac{2\pi}{n} \right)$       (D)  $\Delta = h \sin^2 \left( \frac{\pi}{n} \right)$

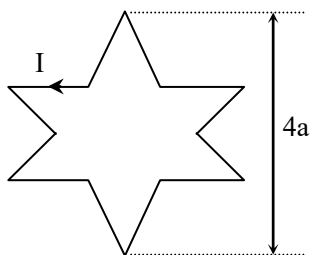
**Q.3** Consider an expanding sphere of instantaneous radius  $R$  whose total mass remains constant. The expansion is such that the instantaneous density  $\rho$  remains uniform throughout the volume. The rate of fractional change in density  $\left(\frac{1}{\rho} \frac{d\rho}{dt}\right)$  is constant. The velocity  $v$  of any point on the surface of the expanding sphere is proportional to

- (A)  $R^3$                       (B)  $R$                       (C)  $\frac{1}{R}$                       (D)  $R^{2/3}$

**Q.4** A person measures the depth of a well by measuring the time interval between dropping a stone and receiving the sound of impact with the bottom of the well. The error in his measurement of time is  $\delta T = 0.01$  seconds and he measures the depth of the well to be  $L = 20$  meters. Take the acceleration due to gravity  $g = 10 \text{ ms}^{-2}$  and the velocity of sound is  $300 \text{ ms}^{-1}$ . Then the fractional error in the measurement,  $\delta L/L$ , is closest to

- (A) 5%                      (B) 3%                      (C) 0.2%                      (D) 1%

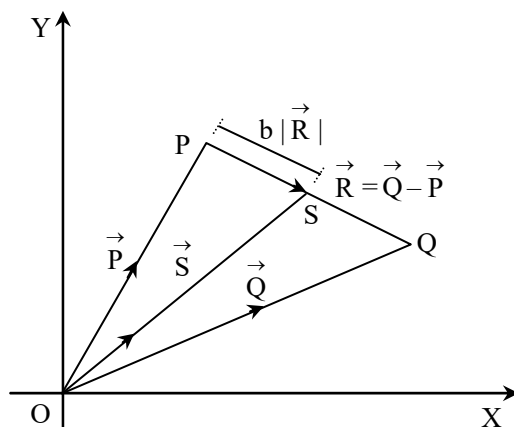
- Q.5** A symmetric star shaped conducting wire loop is carrying a steady state current  $I$  as shown in the figure. The distance between the diametrically opposite vertices of the star is  $4a$ . The magnitude of the magnetic field at the centre of the loop is



- (A)  $\frac{\mu_0 I}{4\pi a} 3[\sqrt{3} - 1]$       (B)  $\frac{\mu_0 I}{4\pi a} 6[\sqrt{3} - 1]$       (C)  $\frac{\mu_0 I}{4\pi a} 3[2 - \sqrt{3}]$       (D)  $\frac{\mu_0 I}{4\pi a} 6[\sqrt{3} + 1]$

- Q.6** A rocket is launched normal to the surface of the Earth, away from the Sun, along the line joining the Sun and the Earth. The Sun is  $3 \times 10^5$  times heavier than the Earth and is at a distance  $2.5 \times 10^4$  times larger than the radius of the Earth. The escape velocity from Earth's gravitational field is  $v_e = 11.2 \text{ km s}^{-1}$ . The minimum initial velocity ( $v_s$ ) required for the rocket to be able to leave the *Sun-Earth system* is closest to (*Ignore the rotation and revolution of the Earth and the presence of any other planet*)
- (A)  $v_s = 42 \text{ km s}^{-1}$       (B)  $v_s = 62 \text{ km s}^{-1}$       (C)  $v_s = 22 \text{ km s}^{-1}$       (D)  $v_s = 72 \text{ km s}^{-1}$

- Q.7** Three vectors  $\vec{P}$ ,  $\vec{Q}$  and  $\vec{R}$  are shown in the figure. Let S be any point on the vector  $\vec{R}$ . The distance between the points P and S is  $b |\vec{R}|$ . The general relation among vectors  $\vec{P}$ ,  $\vec{Q}$  and  $\vec{S}$  is



(A)  $\vec{S} = (1 - b^2)\vec{P} + b\vec{Q}$

(B)  $\vec{S} = (1 - b)\vec{P} + b\vec{Q}$

(C)  $\vec{S} = (1 - b)\vec{P} + b^2\vec{Q}$

(D)  $\vec{S} = (b - 1)\vec{P} + b\vec{Q}$

### SECTION – 2 (Maximum Marks : 28)

- The section contains **SEVEN** questions
- Each question has **FOUR** options (A), (B), (C) and (D) . **ONE OR MORE THAN ONE** of these four option(s) is(are) correct.
- For each question, darken the bubble(s) corresponding to all the correct option(s) in the ORS.
- For each question, marks will be awarded in one of the following categories :  
 Full Marks : +4 If only the bubble(s) corresponding to all the correct option(s) is(are) darkened.  
 Partial Marks : +1 For darkening a bubble corresponding **to each correct option**, provided NO incorrect option is darkened  
 Zero Marks : 0 If none of the bubble is darkened.  
 Negative Marks : -2 In all other cases.
- For example, if (A), (C) and (D) are all the correct options for a question, darkening all these three will get +4 marks; darkening only (A) and (D) will get +2 marks; and darkening (A) and (B) will get -2 marks, as a wrong option is also darkened.

**Q.8** The instantaneous voltages at three terminals marked X, Y and Z are given by

$$V_X = V_0 \sin \omega t,$$

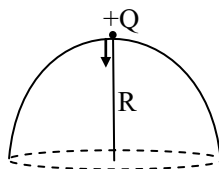
$$V_Y = V_0 \sin \left( \omega t + \frac{2\pi}{3} \right) \text{ and}$$

$$V_Z = V_0 \sin \left( \omega t + \frac{4\pi}{3} \right)$$

An ideal voltmeter is configured to read *rms* value of the potential difference between its terminals. It is connected between points X and Y and then between Y and Z. The reading (s) of the voltmeter will be

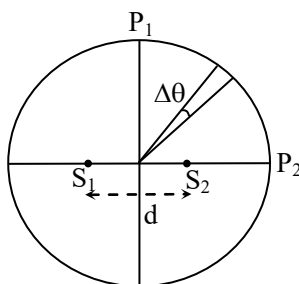
- (A)  $V_{XY}^{\text{rms}} = V_0 \sqrt{\frac{3}{2}}$  (B)  $V_{YZ}^{\text{rms}} = V_0 \sqrt{\frac{1}{2}}$   
 (C) Independent of the choice of the two terminals (D)  $V_{XY}^{\text{rms}} = V_0$

- Q.9** A point charge +Q is placed just outside an imaginary hemispherical surface of radius R as shown in the figure. Which of the following statements is/are Correct?



- (A) Total flux through the curved and the flat surfaces is  $\frac{Q}{\epsilon_0}$   
 (B) The circumference of the flat surface is an equipotential  
 (C) The component of the electric field normal to the flat surface is constant over the surface  
 (D) The electric flux passing through the *curved* surface of the hemisphere is  $-\frac{Q}{2\epsilon_0} \left( 1 - \frac{1}{\sqrt{2}} \right)$

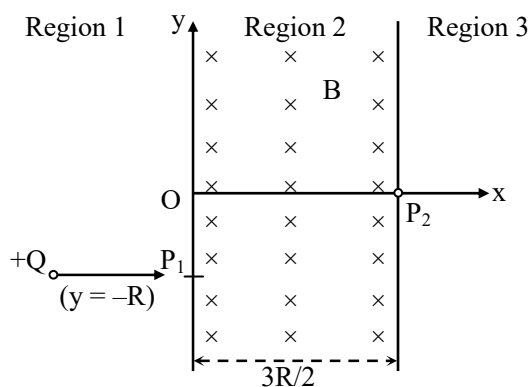
- Q.10** Two coherent monochromatic point sources  $S_1$  and  $S_2$  of wavelength  $\lambda = 600 \text{ nm}$  are placed symmetrically on either side of the center of the circle as shown. The sources are separated by a distance  $d = 1.8 \text{ mm}$ . This arrangement produces interference fringes visible as alternate bright and dark spots on the circumference of the circle. The angular separation between two consecutive bright spots is  $\Delta\theta$ . Which of the following options is/are correct?



- (A) The angular separation between two consecutive bright spots decreases as we move from  $P_1$  to  $P_2$  along the first quadrant
- (B) A dark spot will be formed at the point  $P_2$
- (C) At  $P_2$  the order of the fringe will be maximum
- (D) The total number of fringes produced between  $P_1$  and  $P_2$  in the first quadrant is close to 3000

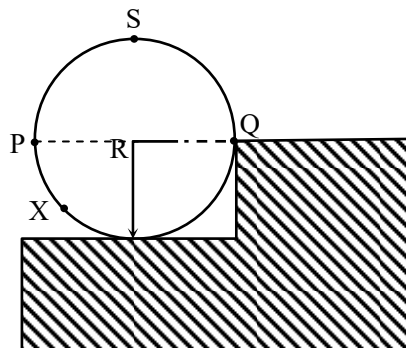


- Q.11** A uniform magnetic field  $B$  exists in the region between  $x = 0$  and  $x = y$  (region 2 in the figure) pointing normally into the plane of the paper. A particle with charge  $+Q$  and momentum  $p$  directed along  $x$ -axis enters region 2 from region 1 at point  $P_1(y = -R)$ . Which of the following option(s) is/are correct ?



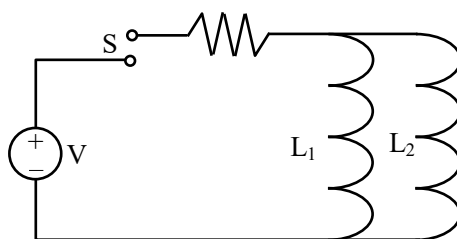
- (A) For a fixed  $B$ , particles of same charge  $Q$  and same velocity  $v$ , the distance between the point  $P_1$  and the point of re-entry into region 1 is inversely proportional to the mass of the particle.
- (B) For  $B = \frac{8}{13} \frac{p}{QR}$ , the particle will enter region 3 through the point  $P_2$  on  $x$ -axis
- (C) When the particle re-enters region 1 through the longest possible path in region 2, the magnitude of the change in its linear momentum between point  $P_1$  and the farthest point from  $y$ -axis is  $p/\sqrt{2}$
- (D) For  $B > \frac{2}{3} \frac{p}{QR}$ , the particle will re-enter region 1

- Q.12** A wheel of radius  $R$  and mass  $M$  is placed at the bottom of a fixed step of height  $R$  as shown in the figure. A constant force is continuously applied on the surface of the wheel so that it just climbs the step without slipping. Consider the torque  $\tau$  about an axis normal to the plane of the paper passing through the point  $Q$ . Which of the following options is/are correct ?



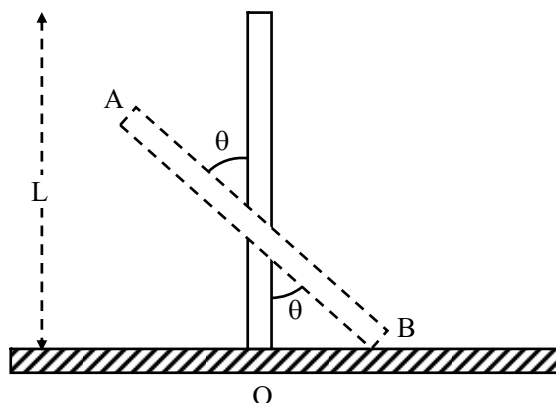
- (A) If the force is applied normal to the circumference at point  $X$  then  $\tau$  is constant  
 (B) If the force is applied tangentially at point  $S$  then  $\tau \neq 0$  but the wheel never climbs the step  
 (C) If the force is applied at point  $P$  tangentially then  $\tau$  decreases continuously as the wheel climbs  
 (D) If the force is applied normal to the circumference at point  $P$  then  $\tau$  is zero

- Q.13** A source of constant voltage  $V$  is connected to a resistance  $R$  and two ideal inductors  $L_1$  and  $L_2$  through a switch  $S$  as shown. There is no mutual inductance between the two inductors. The switch  $S$  is initially open. At  $t = 0$ , the switch is closed and current begins to flow. Which of the following options is/are correct ?



- (A) After a long time, the current through  $L_1$  will be  $\frac{V}{R} \frac{L_2}{L_1 + L_2}$
- (B) The ratio of the currents through  $L_1$  and  $L_2$  is fixed at all times ( $t > 0$ )
- (C) After a long time, the current through  $L_2$  will be  $\frac{V}{R} \frac{L_1}{L_1 + L_2}$
- (D) At  $t = 0$ , the current through the resistance  $R$  is  $\frac{V}{R}$

- Q.14** A rigid uniform bar AB of length  $L$  is slipping from its vertical position on a frictionless floor (as shown in the figure). At some instant of time, the angle made by the bar with the vertical is  $\theta$ . Which of the following statements about its motion is/are correct ?



- (A) Instantaneous torque about the point in contact with the floor is proportional to  $\sin \theta$
- (B) The trajectory of the point A is a parabola
- (C) The midpoint of the bar will fall vertically downward
- (D) When the bar makes an angle  $\theta$  with the vertical, the displacement of its midpoint from the initial position is proportional to  $(1 - \cos \theta)$



### SECTION – 3 (Maximum Marks : 12)

- This section contains **TWO** paragraphs
- Based on each paragraph, there are **TWO** questions
- Each question has **FOUR** options (A), (B), (C) and (D). **ONLY ONE** of these four options is correct
- For each question, darken the bubble corresponding to the correct option in the ORS
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#### PARAGRAPH 1

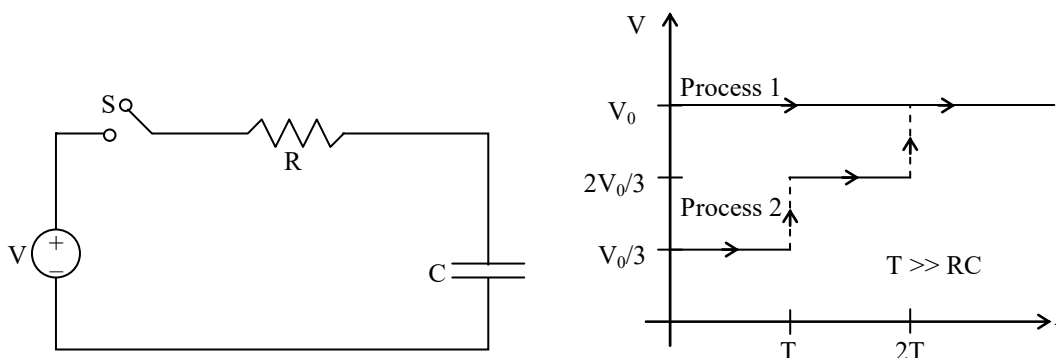
Consider a simple RC circuit as shown in Figure 1.

Process 1 : In the circuit the switch S is closed at  $t = 0$  and the capacitor is fully charged to voltage  $V_0$  (i.e., charging continues for time  $T \gg RC$ ). In the process some dissipation ( $E_D$ ) occurs across the resistance R. The amount of energy finally stored in the fully charged capacitor is  $E_C$ .

Process 2 : In a different process the voltage is first set to  $\frac{V_0}{3}$  and maintained for a charging time  $T \gg RC$ .

Then the voltage is raised to  $\frac{2V_0}{3}$  without discharging the capacitor and again maintained for a time  $T \gg RC$ . The process is repeated one more time by raising the voltage to  $V_0$  and the capacitor is charged to the same final voltage  $V_0$  as in Process 1.

These two processes are depicted in Figure 2.



**Q.15** In Process 1, the energy stored in the capacitor  $E_C$  and heat dissipated across resistance  $E_D$  are related by :

- (A)  $E_C = E_D \ln 2$                       (B)  $E_C = \frac{1}{2} E_D$                       (C)  $E_C = E_D$                       (D)  $E_C = 2E_D$

**Q.16** In Process 2, total energy dissipated across the resistance  $E_D$  is :

- (A)  $E_D = \frac{1}{3} \left( \frac{1}{2} CV_0^2 \right)$                       (B)  $E_D = 3 \left( \frac{1}{2} CV_0^2 \right)$   
 (C)  $E_D = \frac{1}{2} CV_0^2$                       (D)  $E_D = 3 CV_0^2$





## PARAGRAPH 2

One twirls a circular ring (of mass  $M$  and radius  $R$ ) near the tip of one's finger as shown in Figure 1. In the process the finger never loses contact with the inner rim of the ring. The finger traces out the surface of a cone, shown by the dotted line. The radius of the path traced out by the point where the ring and the finger is in contact is  $r$ . The finger rotates with an angular velocity  $\omega_0$ . The rotating ring rolls without slipping on the outside of a smaller circle described by the point where the ring and the finger is in contact (Figure 2). The coefficient of friction between the ring and the finger is  $\mu$  and the acceleration due to gravity is  $g$ .

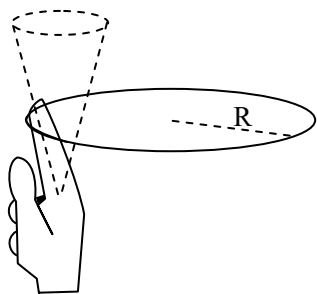


Figure 1

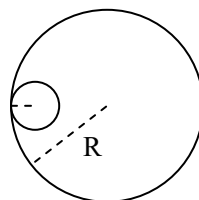


Figure 2

**Q.17** The total kinetic energy of the ring is

(A)  $M\omega_0^2 (R - r)^2$

(B)  $M\omega_0^2 R^2$

(C)  $\frac{3}{2} M\omega_0^2 (R - r)^2$

(D)  $\frac{1}{2} M\omega_0^2 (R - r)^2$

**Q.18** The minimum value of  $\omega_0$  below which the ring will drop down is

(A)  $\sqrt{\frac{2g}{\mu(R-r)}}$

(B)  $\sqrt{\frac{3g}{2\mu(R-r)}}$

(C)  $\sqrt{\frac{g}{\mu(R-r)}}$

(D)  $\sqrt{\frac{g}{2\mu(R-r)}}$

## PART-II (CHEMISTRY)

### SECTION – 1 (Maximum Marks : 21)

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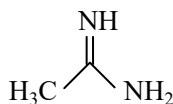
Zero Marks : 0 If none of the bubbles is darkened.

Negative Marks : -1 In all other cases.

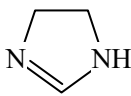
**Q.19** Which of the following combination will produce  $H_2$  gas ?

- (A) Cu metal and conc.  $HNO_3$
- (B) Au metal and NaCN (aq) in the presence of air
- (C) Zn metal and NaOH (aq)
- (D) Fe metal and conc.  $HNO_3$

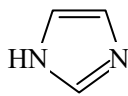
**Q.20** The order of basicity among the following compounds is -



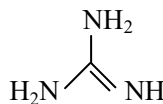
I



II



III



IV

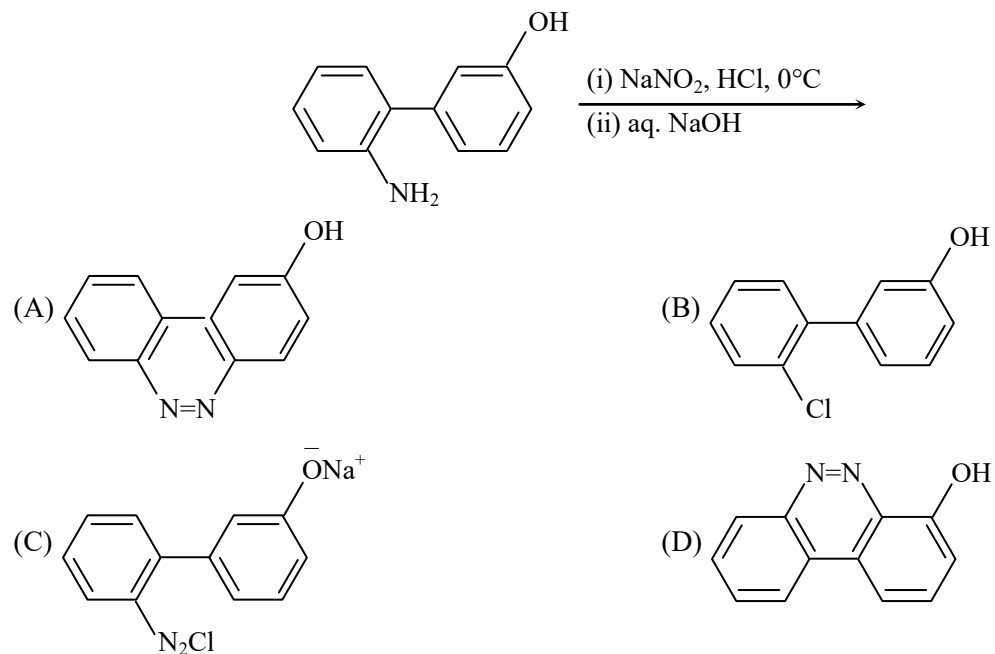
(A) I > IV > III > II

(C) IV > I > II > III

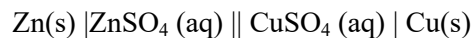
(B) II > I > IV > III

(D) IV > II > III > I

**Q.21** The major product of the following reaction is -



**Q.22** For the following cell,



When the concentration of  $\text{Zn}^{2+}$  is 10 times the concentration of  $\text{Cu}^{2+}$ , the expression for  $\Delta G$  (in  $\text{J mol}^{-1}$ ) is

[F is Faraday constant; R is gas constant; T is temperature;  $E^\circ (\text{cell}) = 1.1 \text{ V}$ ]

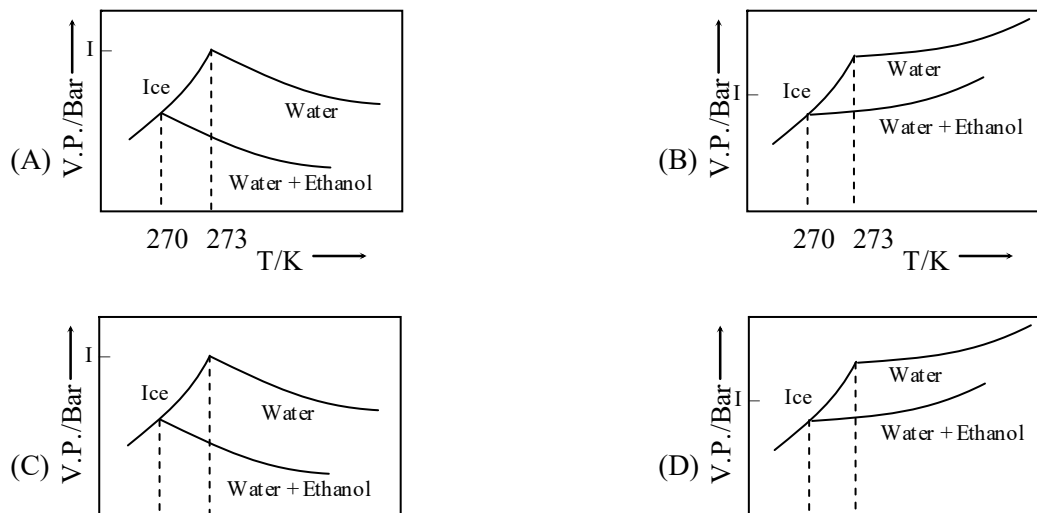
- (A)  $2.303 RT + 1.1 F$       (B)  $2.303 RT - 2.2 F$       (C)  $1.1 F$       (D)  $-2.2 F$

**Q.23** The order of the oxidation state of the phosphorus atom in  $\text{H}_3\text{PO}_2$ ,  $\text{H}_3\text{PO}_4$ ,  $\text{H}_3\text{PO}_3$ , and  $\text{H}_4\text{P}_2\text{O}_6$  is -

- (A)  $\text{H}_3\text{PO}_3 > \text{H}_3\text{PO}_2 > \text{H}_3\text{PO}_4 > \text{H}_4\text{P}_2\text{O}_6$  (B)  $\text{H}_3\text{PO}_4 > \text{H}_3\text{PO}_2 > \text{H}_3\text{PO}_3 > \text{H}_4\text{P}_2\text{O}_6$   
 (C)  $\text{H}_3\text{PO}_2 > \text{H}_3\text{PO}_3 > \text{H}_4\text{P}_2\text{O}_6 > \text{H}_3\text{PO}_4$  (D)  $\text{H}_3\text{PO}_4 > \text{H}_4\text{P}_2\text{O}_6 > \text{H}_3\text{PO}_3 > \text{H}_3\text{PO}_2$

**Q.24** Pure water freezes at 273 K and 1 bar. The addition of 34.5 g of ethanol to 500 g of water changes the freezing point of the solution. Use the freezing point depression constant of water as  $2 \text{ K kg mol}^{-1}$ . The figures shown below represent plots of vapour pressure (V.P.) versus temperature (T). [molecular weight of ethanol is  $46 \text{ g mol}^{-1}$ ].

Among the following, the option representing change in the freezing point is -



**Q.25** The standard state Gibbs free energies of formation of C (Graphite) and C (diamond) at  $T = 298 \text{ K}$  are

$$\Delta_f G^\circ [\text{C}(\text{graphite})] = 0 \text{ kJ mol}^{-1}$$

$$\Delta_f G^\circ [\text{C}(\text{diamond})] = 2.9 \text{ kJ mol}^{-1}$$

The standard state means that the pressure should be 1 bar, and substance should be pure at a given temperature. The converting of graphite [C(graphite)] to diamond [C(diamond)] reduces its volume by  $2 \times 10^{-6} \text{ m}^3 \text{ mol}^{-1}$ . If C(graphite) is converted to C(diamond) isothermally at  $T = 298 \text{ K}$ , the pressure at which C(graphite) is in equilibrium with C(diamond), is

[Useful information:  $1 \text{ J} = 1 \text{ kg m}^2 \text{ s}^{-2}$ ;  $1 \text{ Pa} = 1 \text{ kg m}^{-1} \text{ s}^{-2}$ ;  $\text{bar} = 10^5 \text{ Pa}$ ]

(A) 29001 bar

(B) 1450 bar

(C) 14501 bar

(D) 58001 bar

**SECTION – 2 (Maximum Marks : 28)**

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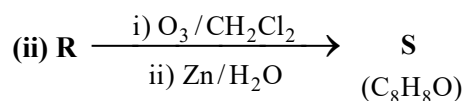
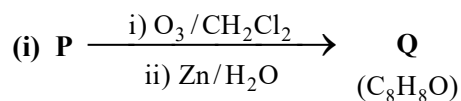
**Q.26** Among the following, the correct statement(s) is(are)

- (A)  $\text{AlCl}_3$  has the three-centre two-electron bonds in its dimeric structure
- (B)  $\text{BH}_3$  has the three-centre two-electron bonds in its dimeric structure
- (C)  $\text{Al}(\text{CH}_3)_3$  has the three-centre two-electron bonds in its dimeric structure
- (D) The Lewis acidity of  $\text{BCl}_3$  is greater than that of  $\text{AlCl}_3$

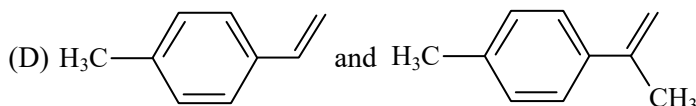
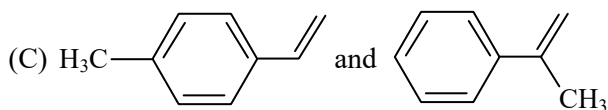
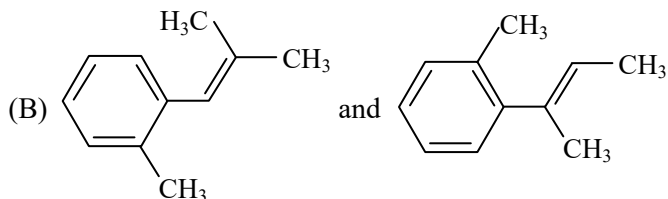
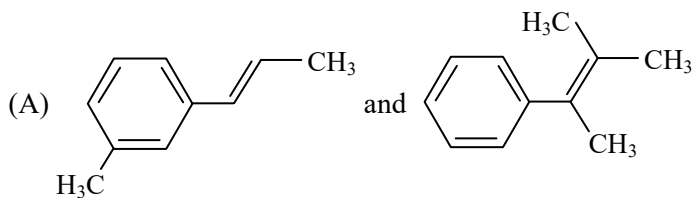


- Q.27** For a reaction taking place in a container in equilibrium with its surroundings, the effect of temperature on its equilibrium constant  $K$  in terms of change in entropy is described by
- (A) With increase in temperature, the value of  $K$  for exothermic reaction decreases because the entropy change of the system is positive
  - (B) With increase in temperature, the value of  $K$  for endothermic reaction increases because the entropy change of the system is negative
  - (C) With increase in temperature, the value of  $K$  for endothermic reaction increases because unfavourable change in entropy of the surroundings decreases
  - (D) With increase in temperature, the value of  $K$  for exothermic reaction decreases because favourable change in entropy of the surroundings decreases

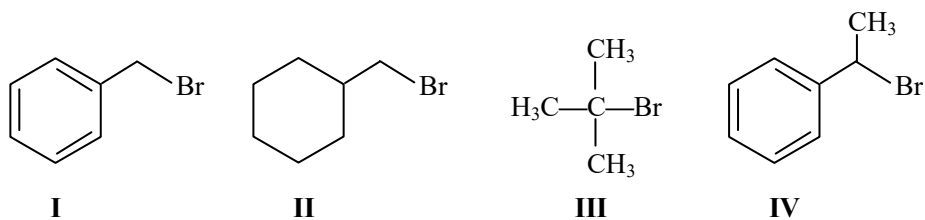
- Q.28** Compounds **P** and **R** upon ozonolysis produce **Q** and **S**, respectively. The molecular formula of **Q** and **S** is  $C_8H_8O$ . **Q** undergoes Cannizzaro reaction but not haloform reaction, whereas **S** undergoes haloform reaction but not Cannizzaro reaction.



The option(s) with suitable combination of **P** and **R**, respectively, is(are)



**Q.29** For the following compounds, the correct statement(s) with respect to nucleophilic substitution reactions is(are)



- (A) **I** and **II** follow  $S_N2$  mechanism  
 (B) Compound **IV** undergoes inversion of configuration  
 (C) The order of reactivity for **I**, **III** and **IV** is: **IV** > **I** > **III**  
 (D) **I** and **III** follow  $S_N1$  mechanism

- Q.30** In a bimolecular reaction, the steric factor  $P$  was experimentally determined to be 4.5. The correct option(s) among the following is(are)
- (A) The value of frequency factor predicted by Arrhenius equation is higher than that determined experimentally
  - (B) Experimentally determined value of frequency factor is higher than that predicted by Arrhenius equation
  - (C) The activation energy of the reaction is unaffected by the value of the steric factor
  - (D) Since  $P = 4.5$ , the reaction will not proceed unless an effective catalyst is used
- Q.31** The correct statement(s) about surface properties is(are)
- (A) Cloud is an emulsion type of colloid in which liquid is dispersed phase and gas is dispersion medium
  - (B) Adsorption is accompanied by decrease in enthalpy and decrease in entropy of the system
  - (C) Brownian motion of colloidal particles does not depend on the size of the particles but depends on viscosity of the solution
  - (D) The critical temperatures of ethane and nitrogen are 563 K and 126 K, respectively. The adsorption of ethane will be more than that of nitrogen on same amount of activated charcoal at a given temperature
- Q.32** The option(s) with only amphoteric oxides is(are)
- (A)  $\text{ZnO}$ ,  $\text{Al}_2\text{O}_3$ ,  $\text{PbO}$ ,  $\text{PbO}_2$
  - (B)  $\text{Cr}_2\text{O}_3$ ,  $\text{BeO}$ ,  $\text{SnO}$ ,  $\text{SnO}_2$
  - (C)  $\text{NO}$ ,  $\text{B}_2\text{O}_3$ ,  $\text{PbO}$ ,  $\text{SnO}_2$
  - (D)  $\text{Cr}_2\text{O}_3$ ,  $\text{CrO}$ ,  $\text{SnO}$ ,  $\text{PbO}$

## SECTION – 3 (Maximum Marks : 12)

- **Type instruction**
- This section contains **TWO** paragraphs.
- Based on each paragraph, there are **TWO** questions.
- Each question has **FOUR** options [A], [B],[C] and [D] **ONLY ONE** of these four options is correct
- For each question, darken the bubble corresponding to the correct option in the ORS
- For each question, marks will be awarded in one of the following categories:

Full marks	:	+3	If only the bubble corresponding to the option is darkened
Zero Marks	:	0	In all other cases

## PARAGRAPH 1

Upon heating  $\text{KClO}_3$  in the presence of catalytic amount of  $\text{MnO}_2$  a gas **W** is formed. Excess amount of **W** reacts with white phosphorus to give **X**. The reaction of **X** with pure  $\text{HNO}_3$  gives **Y** and **Z**.

**Q.33** **Y** and **Z** are, respectively

- |  |  |
|--|--|
| (A) $\text{N}_2\text{O}_5$ and $\text{HPO}_3$          | (B) $\text{N}_2\text{O}_4$ and $\text{HPO}_3$          |
| (C) $\text{N}_2\text{O}_4$ and $\text{H}_3\text{PO}_3$ | (D) $\text{N}_2\text{O}_3$ and $\text{H}_3\text{PO}_4$ |

**Q.34** **W** and **X** are, respectively

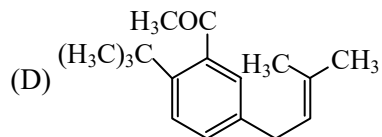
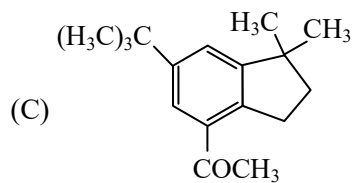
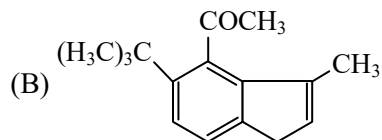
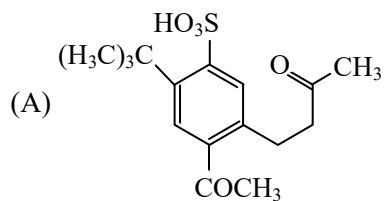
- |  |   |
|--|---|
| (A) $\text{O}_3$ and $\text{P}_4\text{O}_{10}$ | (B) $\text{O}_2$ and $\text{P}_4\text{O}_6$ |
| (C) $\text{O}_2$ and $\text{P}_4\text{O}_{10}$ | (D) $\text{O}_3$ and $\text{P}_4\text{O}_6$ |

## PARAGRAPH 2

The reaction of compound **P** with  $\text{CH}_3\text{MgBr}$  (excess) in  $(\text{C}_2\text{H}_5)_2\text{O}$  followed by additional of  $\text{H}_2\text{O}$  gives **Q**. The compound **Q** on treatment with  $\text{H}_2\text{SO}_4$  at  $0^\circ\text{C}$  gives **R**. The reaction of **R** with  $\text{CH}_3\text{COCl}$  in the presence of anhydrous  $\text{AlCl}_3$  in  $\text{CH}_2\text{Cl}_2$  followed by treatment with  $\text{H}_2\text{O}$  produces compound **S**. [Et in compound **P** is ethyl group]



**Q.35** The Product **S** is



**Q.36** The reactions, **Q** to **R** and **R** to **S**, are

- (A) Friedel-Craft alkylation and Friedel-Craft acylation
- (B) Friedel-Craft alkylation, dehydration and Friedel-Craft acylation
- (C) Dehydration and Friedel-Craft acylation
- (D) Aromatic sulfonation and Friedel-Craft acylation

**PART-III (MATHEMATICS)****SECTION – 1 (Maximum Marks : 21)**

- 
- The section contains **SEVEN** questions
  - Each question has **FOUR** options (A), (B), (C) and (D). **ONLY ONE** of these four options is correct.
  - For each question, darken the bubble corresponding to the correct option in the ORS.
  - For each question, marks will be awarded in one of the following categories:

Full Marks : +3 If only the bubble corresponding to the correct option is darkened.

Zero Marks : 0 If none of the bubbles is darkened.

Negative Marks : -1 In all other cases.

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- Q.37** The equation of the plane passing through the point (1, 1, 1) and perpendicular to the planes  $2x + y - 2z = 5$  and  $3x - 6y - 2z = 7$ , is

(A)  $14x + 2y + 15z = 31$

(B)  $14x - 2y + 15z = 27$

(C)  $-14x + 2y + 15z = 3$

(D)  $14x + 2y - 15z = 1$

- Q.38** How many  $3 \times 3$  matrices  $M$  with entries from  $\{0, 1, 2\}$  are there, for which the sum of the diagonal entries of  $M^T M$  is 5 ?

(A) 162

(B) 198

(C) 126

(D) 135

**Q.39** Three randomly chosen nonnegative integers  $x$ ,  $y$  and  $z$  are found to satisfy the equation  $x + y + z = 10$ . Then the probability that  $z$  is even, is

- (A)  $\frac{1}{2}$  (B)  $\frac{36}{55}$  (C)  $\frac{5}{11}$  (D)  $\frac{6}{11}$

**Q.40** Let  $O$  be the origin and let  $PQR$  be an arbitrary triangle. The point  $S$  is such that

$$\overrightarrow{OP} \cdot \overrightarrow{OQ} + \overrightarrow{OR} \cdot \overrightarrow{OS} = \overrightarrow{OR} \cdot \overrightarrow{OP} + \overrightarrow{OQ} \cdot \overrightarrow{OS} = \overrightarrow{OQ} \cdot \overrightarrow{OR} + \overrightarrow{OP} \cdot \overrightarrow{OS}$$

Then the triangle  $PQR$  has  $S$  as its

- (A) orthocenter (B) incentre (C) centroid (D) circumcentre

- Q.41** If  $y = y(x)$  satisfies the differential equation  $8\sqrt{x}(\sqrt{9+\sqrt{x}})dy = \left(\sqrt{4+\sqrt{9+\sqrt{x}}}\right)^{-1} dx$ ,  $x > 0$  and  $y(0) = \sqrt{7}$ , then  $y(256) =$
- (A) 3 (B) 9 (C) 16 (D) 80

- Q.42** If  $f : \mathbb{R} \rightarrow \mathbb{R}$  is a twice differentiable function such that  $f''(x) > 0$  for all  $x \in \mathbb{R}$  and  $f\left(\frac{1}{2}\right) = \frac{1}{2}$ ,  $f(1) = 1$ , then
- (A)  $\frac{1}{2} < f'(1) \leq 1$  (B)  $f'(1) > 1$  (C)  $0 < f'(1) \leq \frac{1}{2}$  (D)  $f'(1) \leq 0$



- Q.43** Let  $S = \{1, 2, 3, \dots, 9\}$ . For  $k = 1, 2, \dots, 5$ , let  $N_k$  be the number of subsets of  $S$ , each containing five elements out of which exactly  $k$  are odd. Then  $N_1 + N_2 + N_3 + N_4 + N_5 =$
- (A) 126                                      (B) 125                                      (C) 210                                      (D) 252

### SECTION – 2 (Maximum Marks : 28)

- This section contains **SEVEN** questions
- Each question has **FOUR** options (A), (B), (C) and (D). **ONE OR MORE THAN ONE** of these four option(s) is(are) correct.
- For each question, darken the bubble(s) corresponding to all the correct option(s) in the ORS.
- For each question, marks will be awarded in one of the following categories :
 

Full Marks : +4 If only the bubble(s) corresponding to all the correct option(s) is(are) darkened.

Partial Marks : +1 For darkening a bubble corresponding **to each correct option**, provided NO incorrect option is darkened

Zero Marks : 0 If none of the bubbles is darkened.

Negative Marks : -2 In all other cases.
- For example, if (A), (C) and (D) are all the correct options for a question, darkening all these three will get +4 marks; darkening only (A) and (D) will get +2 marks; and darkening (A) and (B) will get -2 marks, as a wrong option is also darkened.

**Q.44** If  $f(x) = \begin{vmatrix} \cos(2x) & \cos(2x) & \sin(2x) \\ -\cos x & \cos x & -\sin x \\ \sin x & \sin x & \cos x \end{vmatrix}$ , then

- (A)  $f(x)$  attains its maximum at  $x = 0$
- (B)  $f'(x) = 0$  at exactly three points in  $(-\pi, \pi)$
- (C)  $f'(x) = 0$  at more than three points in  $(-\pi, \pi)$
- (D)  $f(x)$  attains its minimum at  $x = 0$

**Q.45** Let  $f(x) = \frac{1-x(1+|1-x|)}{|1-x|} \cos\left(\frac{1}{1-x}\right)$  for  $x \neq 1$ . Then

- (A)  $\lim_{x \rightarrow 1^-} f(x) = 0$
- (B)  $\lim_{x \rightarrow 1^+} f(x)$  does not exist
- (C)  $\lim_{x \rightarrow 1^-} f(x)$  does not exist
- (D)  $\lim_{x \rightarrow 1^+} f(x) = 0$

**Q.46** If  $I = \sum_{k=1}^{98} \int_k^{k+1} \frac{k+1}{x(x+1)} dx$ , then

(A)  $I > \frac{49}{50}$

(B)  $I < \log_e 99$

(C)  $I > \log_e 99$

(D)  $I < \frac{49}{50}$

**Q.47** If  $g(x) = \int_{\sin x}^{\sin(2x)} \sin^{-1}(t) dt$ , then

(A)  $g'\left(\frac{\pi}{2}\right) = -2\pi$

(B)  $g'\left(-\frac{\pi}{2}\right) = -2\pi$

(C)  $g'\left(\frac{\pi}{2}\right) = 2\pi$

(D)  $g'\left(-\frac{\pi}{2}\right) = 2\pi$

**Q.48** If the line  $x = \alpha$  divides the area of region  $R = \{(x, y) \in \mathbb{R}^2 : x^3 \leq y \leq x, 0 \leq x \leq 1\}$  into two equal parts, then

- (A)  $2\alpha^4 - 4\alpha^2 + 1 = 0$       (B)  $\alpha^4 + 4\alpha^2 - 1 = 0$       (C)  $\frac{1}{2} < \alpha < 1$       (D)  $0 < \alpha \leq \frac{1}{2}$

**Q.49** If  $f : \mathbb{R} \rightarrow \mathbb{R}$  is a differentiable function such that  $f'(x) > 2f(x)$  for all  $x \in \mathbb{R}$ , and  $f(0) = 1$ , then

(A)  $f(x)$  is decreasing in  $(0, \infty)$

(B)  $f(x) > e^{2x}$  in  $(0, \infty)$

(C)  $f'(x) < e^{2x}$  in  $(0, \infty)$

(D)  $f(x)$  is increasing in  $(0, \infty)$

**Q.50** Let  $\alpha$  and  $\beta$  be nonzero real numbers such that  $2(\cos \beta - \cos \alpha) + \cos \alpha \cos \beta = 1$ . Then which of the following is/are true ?

(A)  $\tan\left(\frac{\alpha}{2}\right) + \sqrt{3} \tan\left(\frac{\beta}{2}\right) = 0$

(B)  $\tan\left(\frac{\alpha}{2}\right) - \sqrt{3} \tan\left(\frac{\beta}{2}\right) = 0$

(C)  $\sqrt{3} \tan\left(\frac{\alpha}{2}\right) - \tan\left(\frac{\beta}{2}\right) = 0$

(D)  $\sqrt{3} \tan\left(\frac{\alpha}{2}\right) + \tan\left(\frac{\beta}{2}\right) = 0$

### SECTION – 3 (Maximum Marks : 12)

- This section contains **TWO** paragraphs.
- Based on each paragraph, there are **TWO** questions.
- Each question has **FOUR** options (A), (B), (C) and (D). **ONLY ONE** of these four options is correct.
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 Full Marks : +3 If only the bubble corresponding to the correct option is darkened.  
 Zero Marks : 0 In all other cases.

#### PARAGRAPH # 1

Let O be the origin, and  $\overrightarrow{OX}$ ,  $\overrightarrow{OY}$ ,  $\overrightarrow{OZ}$  be three unit vectors in the directions of the sides  $\overrightarrow{QR}$ ,  $\overrightarrow{RP}$ ,  $\overrightarrow{PQ}$ , respectively, of a triangle PQR.

**Q.51**  $\left| \overrightarrow{OX} \times \overrightarrow{OY} \right| =$

(A)  $\sin 2R$

(B)  $\sin (P + R)$

(C)  $\sin (Q + R)$

(D)  $\sin (P + Q)$

**Q.52** If the triangle PQR varies, then the minimum value of  $\cos(P + Q) + \cos(Q + R) + \cos(R + P)$  is

(A)  $-\frac{3}{2}$

(B)  $\frac{5}{3}$

(C)  $-\frac{5}{3}$

(D)  $\frac{3}{2}$

---

**PARAGRAPH # 2**

Let  $p, q$  be integers and let  $\alpha, \beta$  be the roots of the equation,  $x^2 - x - 1 = 0$ , where  $\alpha \neq \beta$ . For  $n = 0, 1, 2, \dots$ , let  $a_n = p\alpha^n + q\beta^n$ .

**FACT :** If  $a$  and  $b$  are rational numbers and  $a + b\sqrt{5} = 0$ , then  $a = 0 = b$ .

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**Q.53** If  $a_4 = 28$ , then  $p + 2q =$

(A) 14

(B) 12

(C) 7

(D) 21

**Q.54**  $a_{12} =$

(A)  $a_{11} - a_{10}$

(B)  $a_{11} + a_{10}$

(C)  $a_{11} + 2a_{10}$

(D)  $2a_{11} + a_{10}$