

Energy and Environmental Engineering

CEME 102



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GLOBAL AND NATIONAL ENERGY SCENARIO. (1 hours)

INTRODUCTION TO ENERGY SOURCES (2 hours)

Classification of Energy Sources in terms of Primary and Secondary Sources, Commercial and Non Commercial Sources of Energy; Renewable and Fossil based Sources of Energy;

INTRODUCTION TO FUELS AND ITS PROPERTIES (1 hours)

INTRODUCTION TO VARIOUS ENERGY CONVERSION SYSTEMS (6 hours)

like Power Plant, Pump, Refrigerator, Air Conditioner, Internal Combustion Engine, Solar PV Cell, Solar Water Heating System, Biogas Plant, Wind Turbine System general functioning including their normal rating specifications.

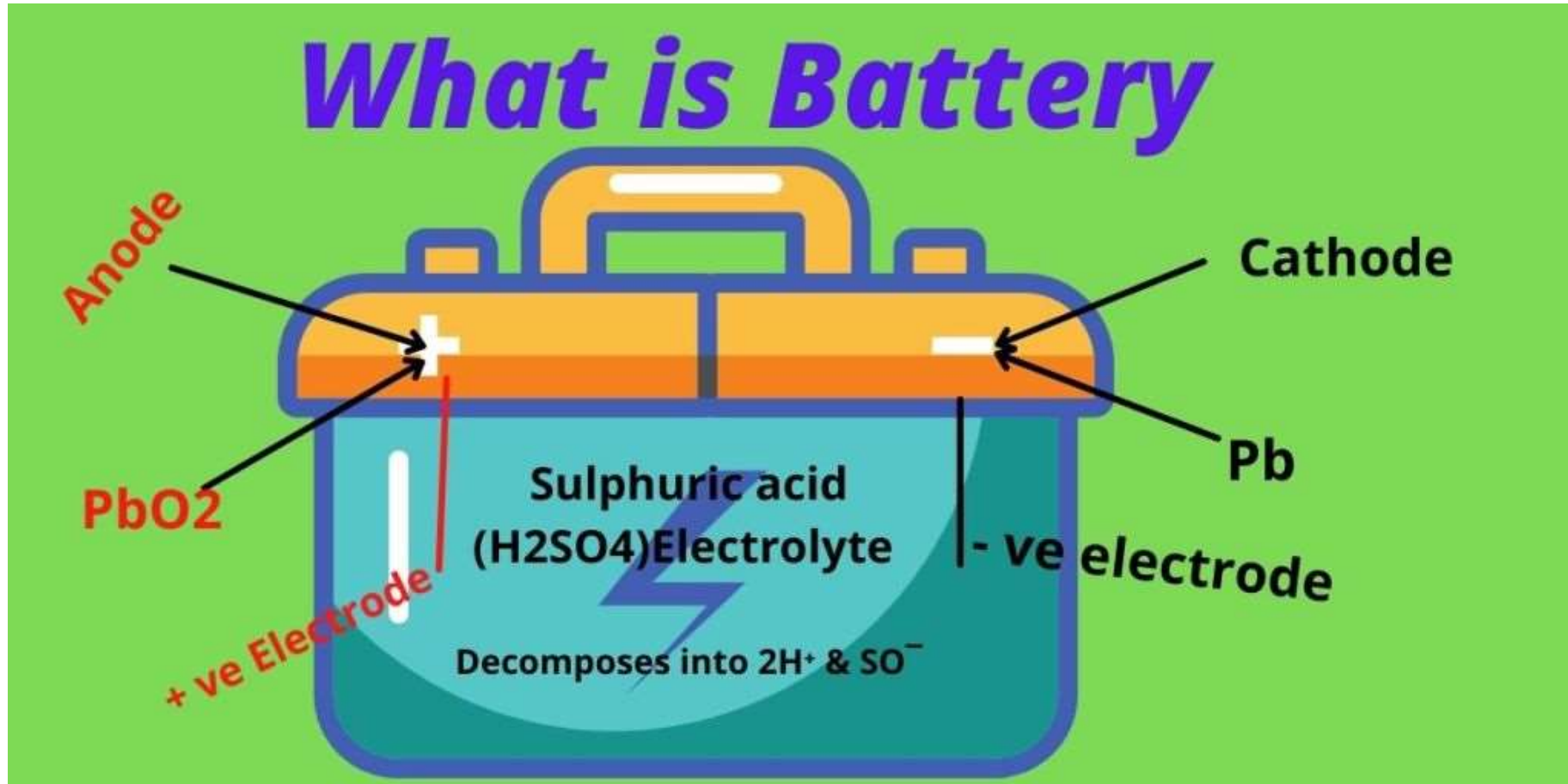
ASPECTS OF ENERGY CONSERVATION AND MANAGEMENT (4 hours)

Energy Conservation Act, Energy Policy of Company; Need for Energy Standards and Labelling; Energy Building Codes.

ENERGY STORAGE IN BATTERIES (2 hours)

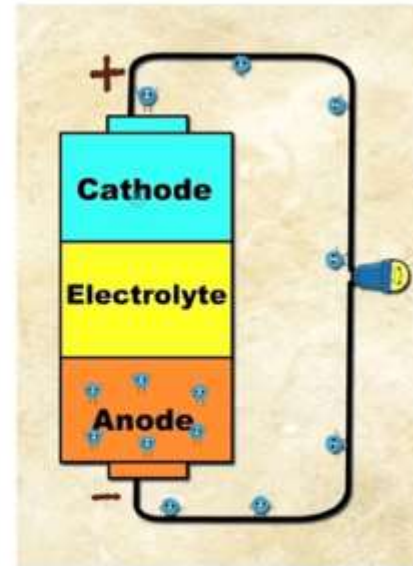
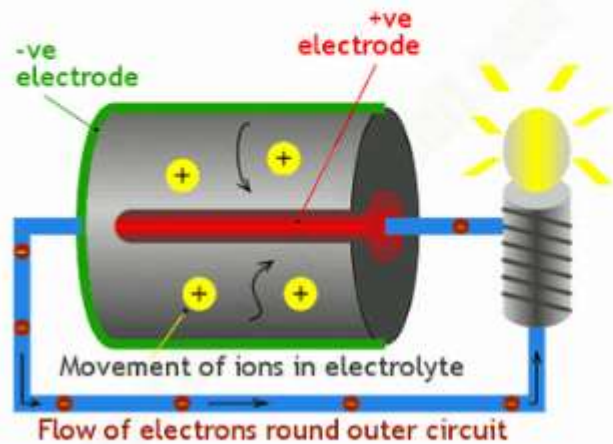
Type of batteries; Electric Vehicles

Battery??



Working of a battery

How Does a Battery Work?



Electrical 4 U

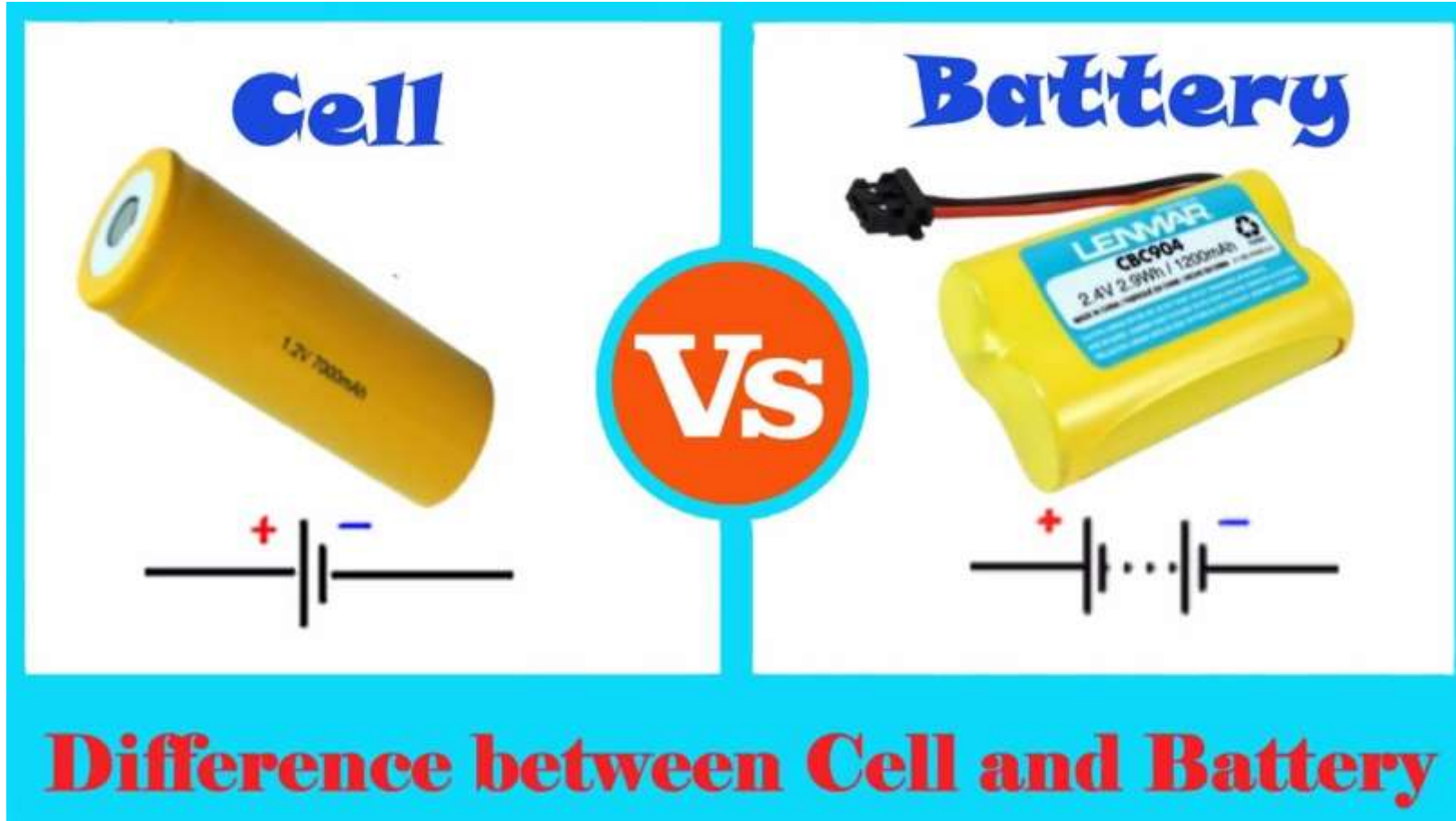
A battery converts chemical energy into electrical energy through a chemical reaction. Within the battery, two different materials called electrodes react with an electrolyte, generating an electric current. One electrode undergoes oxidation (loses electrons), while the other undergoes reduction (gains electrons), creating a flow of electrons through an external circuit, powering devices connected to the battery. This process continues until the reactants are depleted, requiring recharging or replacement of the battery.

Cell vs Battery

CELL

A single unit that generates electrical energy through chemical reactions.

Used in devices requiring low power, such as remote controls and watches.

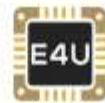


BATTERY

A collection of cells connected in series or parallel to provide a larger voltage and capacity. Functions by combining multiple cells to increase energy storage and output. Utilized in high-power applications like laptops, and electric vehicles.

Types of Batteries

Types of Batteries



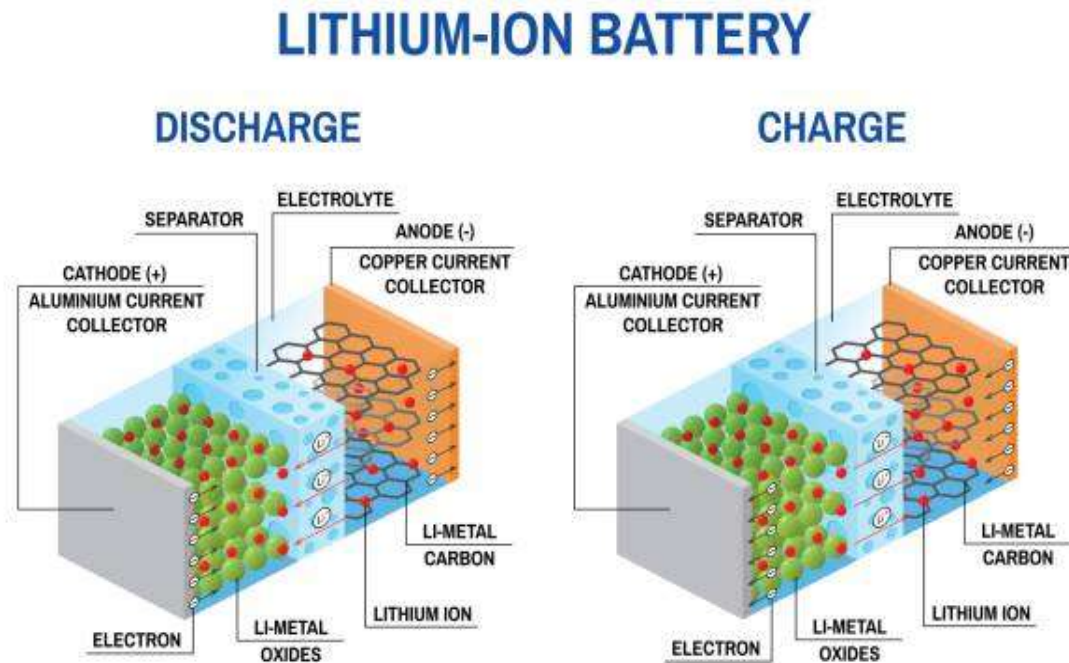
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Classification of Batteries

PRIMARY CELLS VERSUS SECONDARY CELLS

Primary cells are batteries that cannot be recharged or reused	Secondary cells are batteries that can be recharged and reused
Irreversible reactions occur	Reversible reactions occur
Can be used only once	Can be used more than once
Used in portable devices as they produce current immediately	Needs to be charged before use and are used in automobiles
Have lower self-discharge rates and can be used for long term storage of power	Have a higher self-discharge rate compared to primary cells

Lithium-ion Battery

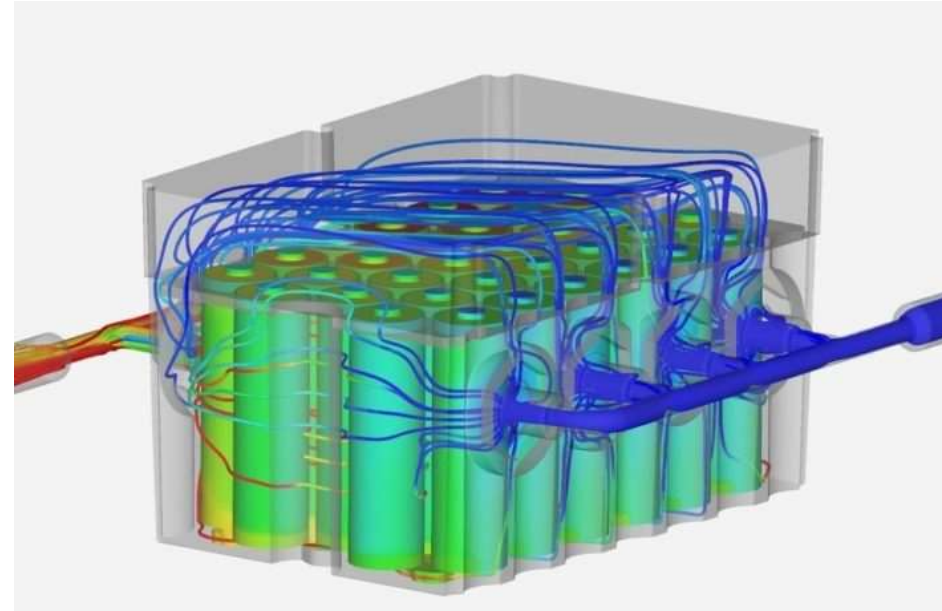
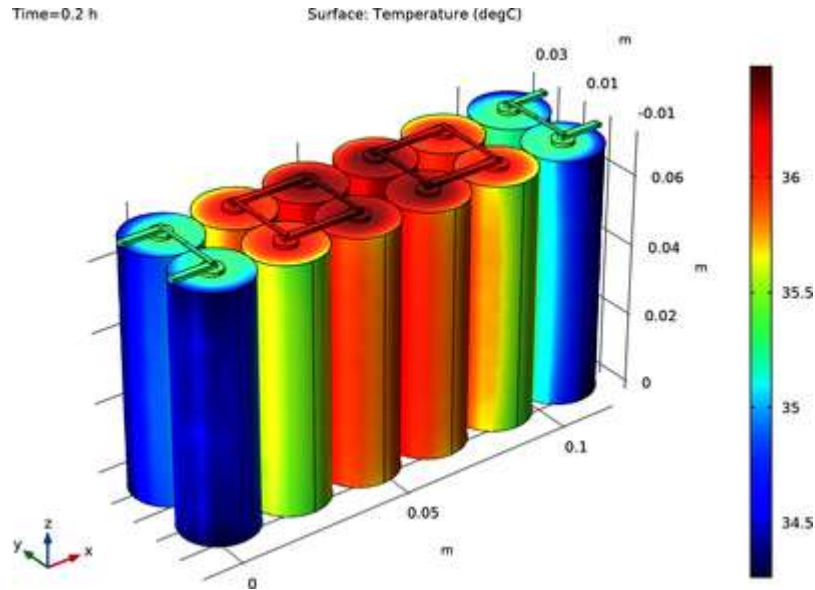


Lithium-ion batteries are a popular choice for portable electronics and electric vehicles due to their high energy density, lightweight design, and long lifespan. They utilize lithium ions moving between positive and negative electrodes to store and release energy efficiently. Their rechargeable nature makes them a sustainable solution for various applications, contributing to the advancement of clean energy technologies.

Application of Batteries



Battery thermal management: Ensuring Optimal Performance



Effective battery thermal management is crucial for enhancing performance, longevity, and safety of battery systems. By maintaining ideal operating temperatures, we prevent overheating, which can degrade battery life and compromise safety. Implementing strategies such as active cooling or heating systems ensures optimal conditions for batteries, maximizing efficiency and reliability. In summary, prioritizing battery thermal management is key to unlocking the full potential of energy storage solutions.