**Equations:** 

$$t_{c} = \frac{t_{s} + T}{\left(1 + ad_{w}R\frac{t_{s} + T}{bP}\right)} , \qquad (1)$$

where  $t_c$  is the corrected air temperature (°C),  $t_s$ ,  $d_w$  and P are the air temperature (°C), H2O density (g m<sup>-3</sup>) and air pressure (kPa) measured by the Irgason correspondingly; T = 273.15 °K; a and b are the empirical coefficients equal to 0.32 and 18.02 correspondingly.