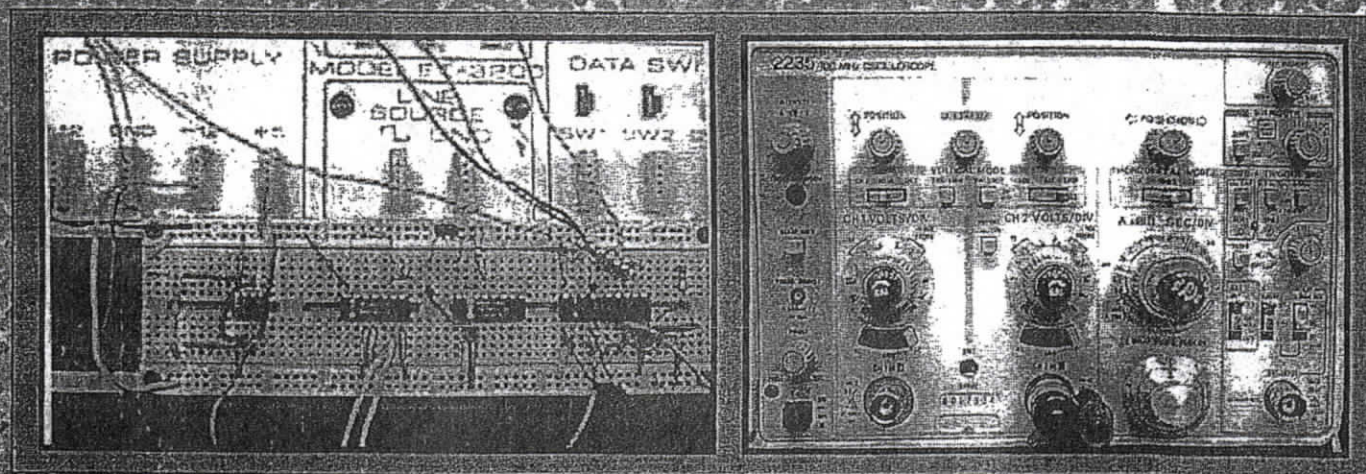


# 29:128 *Electronics* Laboratory Manual

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<u>Lab:</u>	<u>Weeks (typical)*</u>
1. DC Measurements	1
2. AC Measurements	1.5
3. Diode, Power Supplies, Zener, SCR	1
4. Junction Transistor, Part I	1.2
5. Junction Transistor, Part II	1
6. <sup>†</sup> Field Effect Transistor (FET)	1
7. Operational Amplifier	1.3
8. Digital Gates	1
9. Flip-Flops, Shift Register, Decade Counter, Decoder, LED Display	1.5
10. Digital Meets Analog	1
Special project	4

\* An average students will require approximately 3 hours to complete a lab that is indicated as 1 week. Some students will require more than 3 hours, and the TA should make additional lab hours available as needed.

<sup>†</sup> indicates Lab that can be deleted entirely to allow time for a special project

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# PREFACE

## GOAL

The purpose of the experiments described here is to acquaint the student with:

- (1) analog & digital devices
- (2) design of circuits
- (3) instruments & procedures for electronic test & measurement.

The aim is to teach a practical skill that the student can use in the course of his or her own experimental research projects in physics, astronomy, or another science.

At the end of this course, the student should be able to:

- (1) design and build simple circuits of his or her own design.
- (2) use electronic test & measurement instruments such as oscilloscopes, timers, function generators, etc. in experimental research.

## REFERENCE

This manual is intended for use with the following textbook.

Horowitz and Hill  
The Art of Electronics  
2nd Edition, 1989/1990  
Cambridge University Press

## DATA BOOKS

Data books, such as the Texas Instruments or National Instruments references for TTL, CMOS and LINEAR circuits, should be used by the student to check pin designations, outputs, and other data not given in this lab manual.

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## LAB NOTEBOOKS

As a part of training to be a scientist, students should maintain a personal notebook just as a research scientist does. This lab notebook will not be graded, but the student must have one and use it. A lab notebook with a sewn-together binding is preferred.

Here is a guideline for lab notebooks: a notebook should contain sufficient detail so that a year later the experiment could be duplicated exactly.

In the notebook, the student should:

draw a schematic diagram for every circuit that is built

- ☐ label this diagram with
  - part numbers
  - pin designations
  - output/input designations
- ☐ show the major connections to external power supplies, etc.

list the instruments used by type and model

- ☐ include in this list: oscilloscope, multimeters, function generators, etc.

draw the appearance of the oscilloscope display, if used

- ☐ indicate the vertical and horizontal scales, with units

record a table of all measurements

- ☐ include units (e.g. mV) for inputs and outputs
- ☐ record the scale (e.g. 200 mV) of the meter or oscilloscope
- ☐ indicate where on the schematic the measurement is made
- ☐ for digital circuits, this table may be in the form of a truth table

for measurements that have an uncertainty:

- ☐ list more than one measurement as an error check
- ☐ estimate the error bar

If the student does not use a computer to make graphs, he/she will need log graph paper for Labs 2 and 7. Buy some that is semi-log (logarithmic on only one axis) and some that is log-log. For this course, choose graph paper with 5 cycles on the log axes.

# LAB REPORTS

For each Lab, students will individually prepare a lab report for grading.

This report is distinct from the notebook; the notebook is not a substitute.

Reports should be organized as a brief introduction, and then an experimental section that is organized according to the section number.

- indicates where your response is required.

## *Preface*

brief introductory paragraph,  $\approx 30$  words, describing the report's theme

## *Experiments*

REPEAT THE FOLLOWING FOR EACH SECTION

### *Apparatus*

schematic diagram, labeled with:

part numbers (e.g. 1N914 diode)

pin designations (on IC's)

output / input terminals

list of instruments used (e.g., Tektronix 2235 oscilloscope)

### *Procedure*

A maximum of three sentences to explain:

what was measured (e.g., amplifier gain)

what was varied (e.g., oscillator frequency)

how errors were estimated\*

### *Results*

$\Rightarrow$  indicates a response is required

where it is appropriate, response should include:

table and/or graph of results

label each curve

draw smooth curves through data points

label axes and indicate units (e.g., Hz, mV)

\*estimate of errors, for analog measurements

as a column in a table

or as a representative error bar on a graph

sketch of oscilloscope display, if one was used

timing diagram or truth table, for digital circuits

explanation of any problems you encountered

Handwritten reports are adequate. Typewritten reports are unnecessary. Be brief, but write in complete sentences. For graphs, you may use "Graphical Analysis"<sup>†</sup> or other software. If preparing your report consumes significantly more than two hours, talk to your TA to ask if you are doing something that is unnecessarily time consuming.

\* Some instructors may not want error analysis included in the report. Ask to be sure.

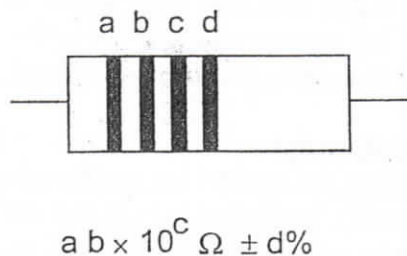
## SYMBOLS USED IN THIS MANUAL

- “ For your information
- ☛ Instructions for you to follow
- ✍ Requires an answer or comment in your lab report

## RESISTOR COLOR CODE

### MULTIPLIERS TOLERANCES

Silver	-2		
Gold	-1		
Black	0	none	20%
Brown	1	silver	10%
Red	2	gold	5%
Orange	3		
Yellow	4		
Green	5		
Blue	6		
Violet	7		
Gray	8		
White	9		



<sup>†</sup> as of 2001, Graphical Analysis was installed on PCs at this location:  
Programs\Vernier Software\GA



# TREATMENT OF EXPERIMENTAL ERRORS

"Error" does not mean a mistake. It means an uncertainty.

In measuring any value, the result is not just one number, such as 5.3 Volts. It is two numbers,  $5.3 \pm 0.1$  Volts. The second number is the experimental uncertainty, or error bar. It usually represents one standard deviation (one sigma) from the first value.

In making an analog measurement with a meter or an oscilloscope, you should record not just the first number, but the error as well.

To decide what value to assign to the error, you must use some judgement. Here are some tips:

## Multimeters:

The manufacturer gives specifications for the precision for the multimeter. The precision of a BK model 388-HD hand-held multimeter is:

DCV	$0.50 \% \text{ rdg} + 1 \text{ LSD}$
ACV	$1.25 \% \text{ rdg} + 4 \text{ LSD}$
F	$2.00 \% \text{ rdg} + 4 \text{ LSD}$
ACA	$1.50 \% \text{ rdg} + 3 \text{ LSD}$
DCA	$1.00 \% \text{ rdg} + 1 \text{ LSD}$
.	$0.75 \% \text{ rdg} + 1 \text{ LSD}$
Hz	$1.00 \% \text{ rdg} + 3 \text{ LSD}$

where rdg is the reading on the display, and LSD is the least significant digit

## Example:

1. You measure a DC voltage and get 5.30 Volt.
2. The uncertainty is given as 0.5 % of the reading, or  $0.005 \times 5.30$ , plus  $1 \times \text{LSD}$ , where the LSD is 0.01 here.
3. So the uncertainty is  $\pm (0.005 \times 5.30 + 1 \times 0.01) = \pm 0.0365 \text{ Volt}$ .
4. Rounding to  $\pm 0.04$ , your result is  $5.30 \pm 0.04 \text{ Volt}$ .

## Analog oscilloscope:

The display has about 8 boxes (divisions) in the vertical direction and 10 boxes horizontally. It also has 5 small tick marks within each division. You can usually measure a value to about  $\pm 0.25$  of a tick mark, i.e.  $\pm 0.05$  divisions.

Example: You use the oscilloscope on the 2 Volt/div scale, and your result is 5.3 Volt. The error bar is  $\pm 0.05$  divisions, so your final result is  $5.3 \pm 0.1 \text{ Volt}$ .



### Propagation of Errors:

When two or more experimental measurements are combined into one by some mathematical operation (such as a product  $P = I V$ ), the two errors add in a non-trivial way.

Example: You measure  $I = 1.2 \pm 0.1$  mA and  $V = 2.5 \pm 0.3$  V. The error on  $P$  is computed using partial derivatives, as:

$$\begin{aligned} dP &= [ (\partial P / \partial I)^2 dI^2 + (\partial P / \partial V)^2 dV^2 ]^{1/2} \\ &= [ V^2 dI^2 + I^2 dV^2 ]^{1/2} \\ &= [ 2.5^2 0.1^2 + 1.2^2 0.3^2 ]^{1/2} = 0.44 \text{ mW} \end{aligned}$$

so the result is  $P = 3.00 \pm 0.44$  mW.

In general: for  $F = F(A, B, C, \dots)$ , the uncertainty  $\delta F$  in terms of the uncertainties  $\delta A$ ,  $\delta B$  etc. is given by:  $dF^2 = S [ (\partial F / \partial A)^2 dA^2 + (\partial F / \partial B)^2 dB^2 + \dots ]$ . Note that errors add through their *squares*.

In this lab course, you are asked to use propagation of errors only twice, both times in Lab 1, in order to develop this skill. These two instances are indicated in Lab 1 with a \*. Otherwise, to save time, do not perform a propagation of errors analysis.

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# TO THE INSTRUCTOR

## HOW ONE CAN USE THIS MANUAL

This manual is intended for a one-semester course. The instructor probably will skip some experiments or parts of experiments due to lack of time.

The first few experiments, analog circuits, will challenge the student the most.

One option is to skip some of the Labs and then finish with a special project. This special project is a circuit invented by the student; it requires about four weeks. The student must begin planning a special project several weeks before this four-week period begins.

## UPDATING THIS MANUAL

This manual for the University of Iowa's Department of Physics and Astronomy has evolved over the years in response to changes in technology and textbooks:

- The earliest version used vacuum tubes.
- The Second Edition was written by Stanley Shawhan and John Benson in 1975. After 17 years it was out of date.
- In 1992 it was retyped and edited by Edward R. McCliment.
- Beginning in 1993 it was re-written by John A. Goree, with the addition of new experiments and illustrations, and a new format. It has been updated each year since 1993.

## TO THE TA:

Before the students do their experiments, you must:

- do the experiment yourself
- inventory parts to be sure they are all available
- set up each lab table.

If enough parts are available, leave your setup in working order for students to examine when they have difficulty.

You will need to schedule additional time in the lab to accommodate students who are unable to complete their work in 3 hours.

Students need their graded lab reports within a week of handing them to you. If you take longer to grade them, students will not know whether they were writing their report as expected, or whether they are omitting required information or spending too much time on unnecessary efforts in writing the report.