

# Determination of Phenol, Cresol isomers in urine by GC-MS

Stephan Drothler

## I. BACKGROUND

In this lab course, artificial urine is investigated for phenol, o-cresol and p-cresol. An external calibration curve is used for the quantification of all three analytes.

## II. METHODS AND MATERIALS

### A. MATERIALS

#### 1) Tools:

- eppendorff-pipettes (200  $\mu\text{L}$ , 1000  $\mu\text{L}$ )
- glas-flask (12 mL)
- GC-MS (HP 6890 (GC) HP 5973 (MS))
- Capillary (HP-5 MS, 30 m0.25 mm I.D., 0.25 m film thickness)
- vacuum chamber
- SPE column (Bond Elut 100 mg C18/OH, Varian)

#### 2) Chemicals:

- methanol (provided by the institute)
- milliQ water (provided by the institute)

### B. METHODS

The injector temperature was set to 250°C with 1  $\mu\text{L}$  of splitless injection volume while the oven temperature was set to 60°C for 3 min, then increased by 5  $\frac{^{\circ}\text{C}}{\text{min}}$  to 80°C and then by 25  $\frac{^{\circ}\text{C}}{\text{min}}$  to 250°C; final temperature: 250°C for 1 min with He as a carrier gas at a flow-rate of 2.5  $\frac{\text{mL}}{\text{min}}$  (Analytical Chemistry 2019). A standard solution mix of phenol, o-cresol and p-cresol [100  $\frac{\mu\text{g}}{\text{mL}}$ ] was already prepared by the supervisor. The working standards were prepared in methanol in a 1 mL GC-vial at final concentrations of 0,1  $\frac{\mu\text{g}}{\text{mL}}$ , 0,25  $\frac{\mu\text{g}}{\text{mL}}$ , 0,5  $\frac{\mu\text{g}}{\text{mL}}$ , 0,75  $\frac{\mu\text{g}}{\text{mL}}$  and 1  $\frac{\mu\text{g}}{\text{mL}}$  of the phenolic compounds. A 10  $\frac{\mu\text{g}}{\text{mL}}$  solution was used to determine the retention times and the appropriate fragments for the SIM mode (Analytical Chemistry 2019). The data processing was performed via R studio (V 1.2.1335) and visualized via the graphrobot web tool (R Core Team 2019, Wang 2019).

## III. RESULTS

The calibration curves of o-cresol (Fig. 2) p-cresol (Fig. 3) and phenol (Fig. 4) provide the variables to the linear equation 1, with m = slope and t = intercept. For the following calculations x = analyte concentration and y = peak area.

$$y = mx + t \quad (1)$$

The sample data (peak area) from Tab III is used to determine the concentration via Equation 1.

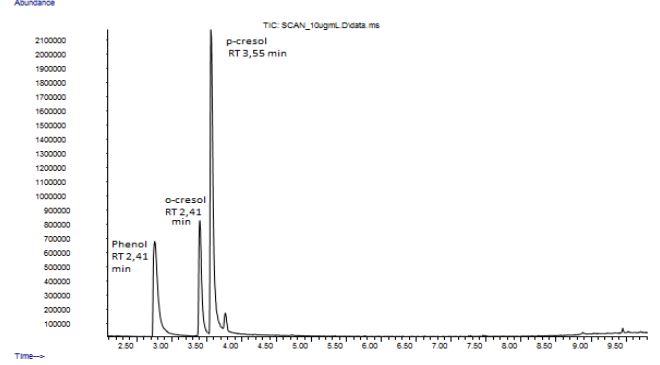


Fig. 1: Chromatogram of the three analytes (10  $\frac{\mu\text{g}}{\text{mL}}$ )

Analyte	RT[ min]	Fragment ion [m/z]
phenol	2,7	65, 66, 94
o-cresol	3,41	77, 107, 108
p-cresol	3,55	77, 107, 108

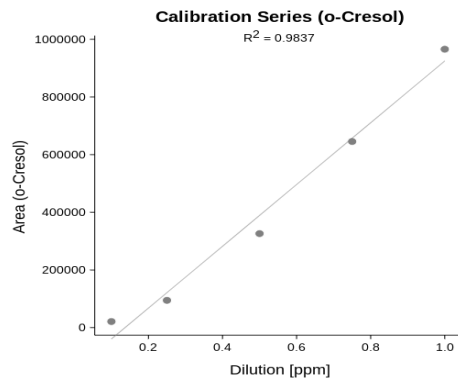
TABLE I: Retention times and fragment ion ratios of the three analytes (Fig. 1).

c [mg/L]	peak number	retention time	Area	analyte
1	1	2,902	48726	phenol
	2	3,474	21211	o-cresol
	3	3,775	23485	p-cresol
25	1	2,849	216298	phenol
	2	3,451	94760	o-cresol
	3	3,635	169866	p-cresol
50	1	2,831	711996	phenol
	2	3,443	326199	o-cresol
	3	3,613	742563	p-cresol
75	1	2,814	1426140	phenol
	2	3,433	645475	o-cresol
	3	3,598	1705651	p-cresol
100	1	2,811	2092602	phenol
	2	3,434	965613	o-cresol
	3	3,594	2653595	p-cresol

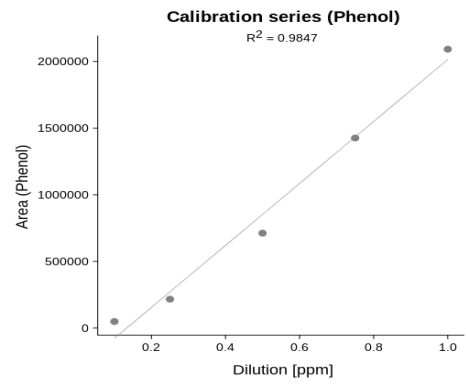
TABLE II: Data for the external calibration of phenol, o-cresol and p-cresol

analyte	RT	area	c [mg/L]
phenol	2,822	1229404	0,66
o-cresol	3,434	596134	0,69
p-cresol	3,602	1317470	0,61

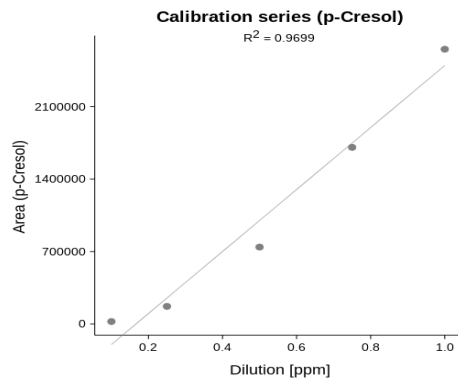
TABLE III: Concentration of the analytes in sample 1 via Equation 1.



**Fig. 2:** External calibration of o-cresol: Slope 1071173.9024, Intercept - 146358.8293



**Fig. 4:** External calibration of phenol: Slope 2325245.5159, Intercept - 309975.2683



**Fig. 3:** External calibration of p-cresol: Slope 2993333.1144, Intercept - 497501.2195

## REFERENCES

- [19] *Analytical Chemistry*. SOPs Lab Course in Instrumental Analytical Chemistry for Molecular Biology. 2019.
- [R C19] R Core Team. *R: A Language and Environment for Statistical Computing*. R Foundation for Statistical Computing. Vienna, Austria, 2019. URL: <https://www.R-project.org/>.
- [Wan19] Lei A Wang. *GraphRobot*. 2019. URL: <https://www.graphrobot.com/>.

## IV. DISCUSSION

The analytes within the sample 1 have a concentration of  $c_{\text{phenol}} = 662 \frac{\text{ng}}{\text{ml}}$ ,  $c_{\text{o-cresol}} = 693 \frac{\text{ng}}{\text{ml}}$  and  $c_{\text{p-cresol}} = 606 \frac{\text{ng}}{\text{ml}}$ .

## V. APPENDIX

### A. Experimental environment

The experiment was performed at TNF Turm JKU, Linz, Austria 06.19 under the supervision of Franz Mlynek.