

Extension of the Polarizable Charge Equilibration Model to Higher Oxidation States with Applications to Ge, As, Se, Br, Sn, Sb, Te, I, Pb, Bi, Po, and At Elements

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1. Parameter Set for PQEq2

Table S1: The electronegativity χ , idempotential J , shell charge Q_c , atomic covalent radius $R_c=R_s$, and spring force constant K_s for PQEq2.

Atom	χ (eV)	J (eV)	Q_c	$R_c=R_s$ (Å)	K_s (kcal/mol/Å ²)
H	4.52719	15.17433	1	0.371	2037.201
He	9.66	29.84	1	1.3	1619.411
Li	3.006	4.772	1	1.557	13.64832
Be	4.877	8.886	1	1.24	59.29709
B	5.11	9.5	1	0.822	109.592
C	5.41477	10.25602	1	0.759	198.8405
N	6.87577	10.26467	1	0.715	301.8761
O	9.7139	14.66128	1	0.669	414.0445
F	9.71494	17.7886	1	0.706	596.1646
Ne	11.04	21.1	1	1.768	842.1173
Na	2.843	4.592	1	2.085	13.77286
Mg	3.951	7.386	1	1.5	31.32676
Al	4.06	7.18	1	1.201	48.8329

Si	4.80466	6.71377	1	1.176	60.04769
P	6.02536	7.86352	1	1.102	91.4776
S	7.01035	8.95326	1	1.047	114.5047
Cl	7.81003	10.23353	1	0.994	152.3228
Ar	9.465	12.71	1	2.108	202.3422
K	2.421	3.84	1	2.586	7.71165
Ca	3.231	5.76	1	2	14.5642
Sc	3.395	6.16	1	1.75	18.65526
Ti	3.47	6.76	1	1.607	22.74409
V	3.65	6.82	1	1.47	26.77933
Cr	3.415	7.73	1	1.402	28.62618
Mn	3.325	8.21	1	1.533	35.32593
Fe	3.76	8.28	1	1.393	39.53139
Co	4.105	8.35	1	1.406	44.27516
Ni	4.465	8.41	1	1.398	48.8329
Cu	3.729	5.002	1	1.434	53.55866

Zn	5.106	8.57	1	1.4	57.75021
Ga	3.641	6.32	1	1.211	40.89454
Ge	4.80386	6.71243	1	1.189	56.86022
As	5.41473	7.5831	1	1.204	77.04494
Se	6.05692	8.40969	1	1.224	88.08056
Br	7.79	8.79002	1	1.141	108.8733
Kr	8.505	11.43	1	2.27	133.6595
Rb	2.331	3.692	1	2.77	7.02929
Sr	3.024	4.88	1	2.415	12.03129
Y	3.83	5.62	1	1.998	14.62836
Zr	3.4	7.1	1	1.758	18.55104
Nb	3.55	6.76	1	1.603	21.15055
Mo	3.465	7.51	1	1.53	25.94248
Tc	3.29	7.98	1	1.5	29.12839
Ru	3.575	8.03	1	1.5	34.58997
Rh	3.975	8.01	1	1.509	38.61206

Pd	4.32	8	1	1.544	69.17994
Ag	4.436	6.268	1	1.622	48.97695
Cd	5.034	7.914	1	1.6	45.11735
In	3.506	5.792	1	1.404	32.55526
Sn	4.62392	6.45025	1	1.354	52.87639
Sb	4.56949	6.45124	1	1.404	50.31268
Te	6.02877	6.80311	1	1.38	60.37522
I	6.1548	7.50909	1	1.333	62.06798
Xe	7.595	9.95	1	2.459	82.11269
Cs	2.183	3.422	1	2.984	5.58842
Ba	2.814	4.792	1	2.442	8.36432
La	2.8355	5.483	1	2.071	10.67729
Ce	2.774	5.384	1	1.925	11.21837
Pr	2.858	5.128	1	2.007	11.77531
Nd	2.8685	5.241	1	2.007	10.57528
Pm	2.881	5.346	1	2	11.03202

Sm	2.9115	5.439	1	1.978	11.52999
Eu	2.8785	5.575	1	2.227	11.98786
Gd	3.1665	5.949	1	1.968	14.13037
Tb	3.018	5.668	1	1.954	13.02211
Dy	3.0555	5.743	1	1.934	13.55362
Ho	3.127	5.782	1	1.925	14.0705
Er	3.1865	5.829	1	1.915	14.62836
Tm	3.2514	5.8658	1	2	15.23228
Yb	3.2889	5.93	1	2.158	15.88822
Lu	2.9629	4.9258	1	1.896	15.16273
Hf	3.7	6.8	1	1.759	20.49776
Ta	5.1	5.7	1	1.605	25.34837
W	4.63	6.62	1	1.538	29.91565
Re	3.96	7.84	1	1.6	34.23337
Os	5.14	7.26	1	1.7	39.06632
Ir	5	8	1	1.866	43.69259

Pt	4.79	8.86	1	1.557	51.08672
Au	4.894	5.172	1	1.618	57.25236
Hg	6.27	8.32	1	1.6	66.14815
Tl	3.2	5.8	1	1.53	43.69259
Pb	3.9	7.06	1	1.444	47.5736
Bi	4.53181	7.32783	1	1.514	44.87347
Po	4.42943	8.40996	1	1.48	48.8329
At	4.53674	8.99576	1	1.47	55.34395
Rn	5.37	10.74	1	2.2	62.65353
Fr	2	4	1	2.3	6.83259
Ra	2.843	4.868	1	2.2	8.67007
Ac	2.835	5.67	1	2.108	10.34466
Th	3.175	5.81	1	2.018	10.34466
Pa	2.985	5.81	1	1.8	13.07337
U	3.341	5.706	1	1.713	13.33589
Np	3.549	5.434	1	1.8	13.38967

Pu	3.243	5.638	1	1.84	13.55362
Am	2.9895	6.007	1	1.942	14.25166
Cm	2.8315	6.379	1	1.9	14.43755
Bk	3.1935	6.071	1	1.9	14.62836
Cf	3.197	6.202	1	1.9	16.19823
Es	3.333	6.178	1	1.9	16.85603
Fm	3.4	6.2	1	1.9	13.95226
Md	3.47	6.22	1	1.9	18.24526
No	3.475	6.35	1	1.9	20.24779
Lr	3.5	6.4	1	1.9	0

2. Electrostatic Interaction Energy Curves

Figure S1: A comparison between QM (LACVP**++¹ or ERMLER**++² and B3LYP³) with PQEq and PQEq2 via the electrostatic interaction dipole energies. A subfigure is depicted for each plot to describe the direction of the dipole scan.