

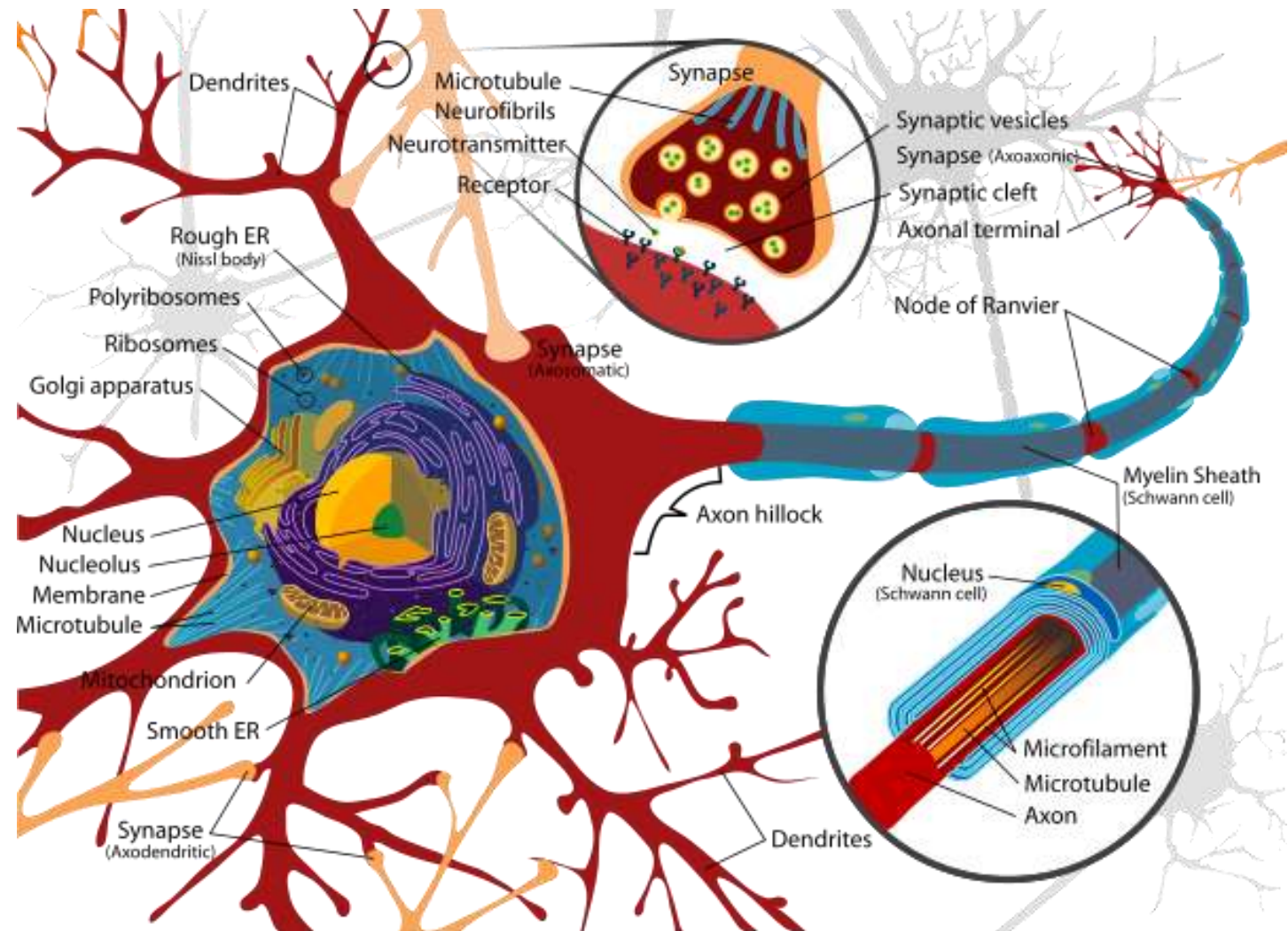
Theoretical & computational  
Neuroscience:

*Programming the Brain*

(BM 6140)

2-credit

# The Neuron

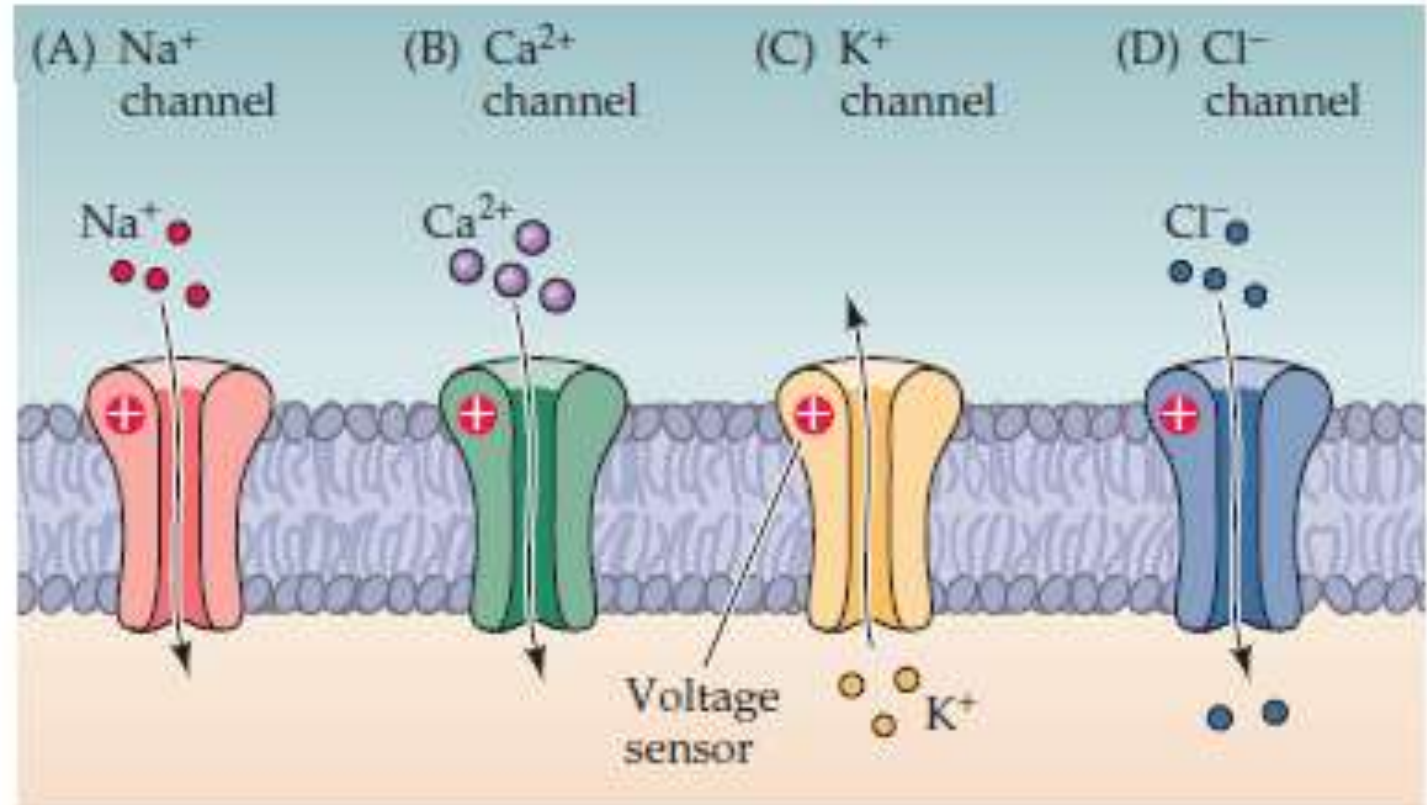


Wikipedia

# Ions and Ion channels

Intra cellular fluids : Extra cellular fluids = 2:1

K<sup>+</sup> abundant                      |                      Na<sup>+</sup> abundant



Purves, Neuroscience

# Ficks law of diffusion

$$J_{\text{diff}} = -D \frac{\partial [C]}{\partial x}, \quad (2.2.1)$$

where  $J$  is diffusion flux (molecules/sec-cm<sup>2</sup>);  $D$  is the diffusion coefficient (cm<sup>2</sup>/sec); and  $[C]$  is the concentration of ion (molecules/cm<sup>3</sup>). The negative sign indicates that  $J$  flows from high to low concentration.

# Drift

$$\begin{aligned} J_{\text{drift}} &= \partial_{el} E \\ &= -\mu z [C] \frac{\partial V}{\partial x}, \end{aligned} \tag{2.2.2}$$

where  $J_{\text{drift}}$  is the drift flux (molecules/sec-cm<sup>2</sup>),  $\partial_{el}$  is electrical conductivity (molecules/V-sec-cm),  $E$  is electric field (V/cm) =  $-\frac{\partial V}{\partial x}$ ,  $V$  is electric potential (V),  $\mu$  is mobility (cm<sup>2</sup>/V-sec),  $z$  is the valence of the ion (dimensionless), and  $[C]$  is the concentration.

# Einstein's relation : connect diffusion constant and mobility

$$D = \frac{kT}{q} \mu, \quad (2.2.3)$$

where  $k$  is Boltzmann's constant ( $1.38 \times 10^{-23}$  joule/°K),  $T$  is absolute temperature (°K), and  $q$  is the charge of the molecule (C).

# Nernst Planck equation : Drift + Diffusion at equilibrium

$$\begin{aligned} J &= J_{\text{drift}} + J_{\text{diff}} \\ &= -\mu z[C] \frac{\partial V}{\partial x} - D \frac{\partial [C]}{\partial x}. \end{aligned}$$

$$J = - \left( \mu z[C] \frac{\partial V}{\partial x} + \frac{\mu kT}{q} \frac{\partial [C]}{\partial x} \right).$$

# Nernst potential

$kT/q = RT/F$  , Flux = 0, Integrate from  $x_{in}$  to  $x_{out}$  ,  $[C]_{in}$  to  $[C]_{out}$  ,  $V_{in}$  to  $V_{out}$

$$V_{in} - V_{out} = \frac{RT}{zF} \ln \frac{[C]_{out}}{[C]_{in}}$$



# What kind of ion distributions will result ?

What will be typical values of  $[C]_{\text{in}} - [C]_{\text{out}}$  &  $V_{\text{in}} - V_{\text{out}}$  ?

# What kind of ion distributions will result ?

What will be typical values of  $[C]_{\text{in}} - [C]_{\text{out}}$  &  $V_{\text{in}} - V_{\text{out}}$  ?

$d[C] = dV = 0$  ? Most natural... discharged battery

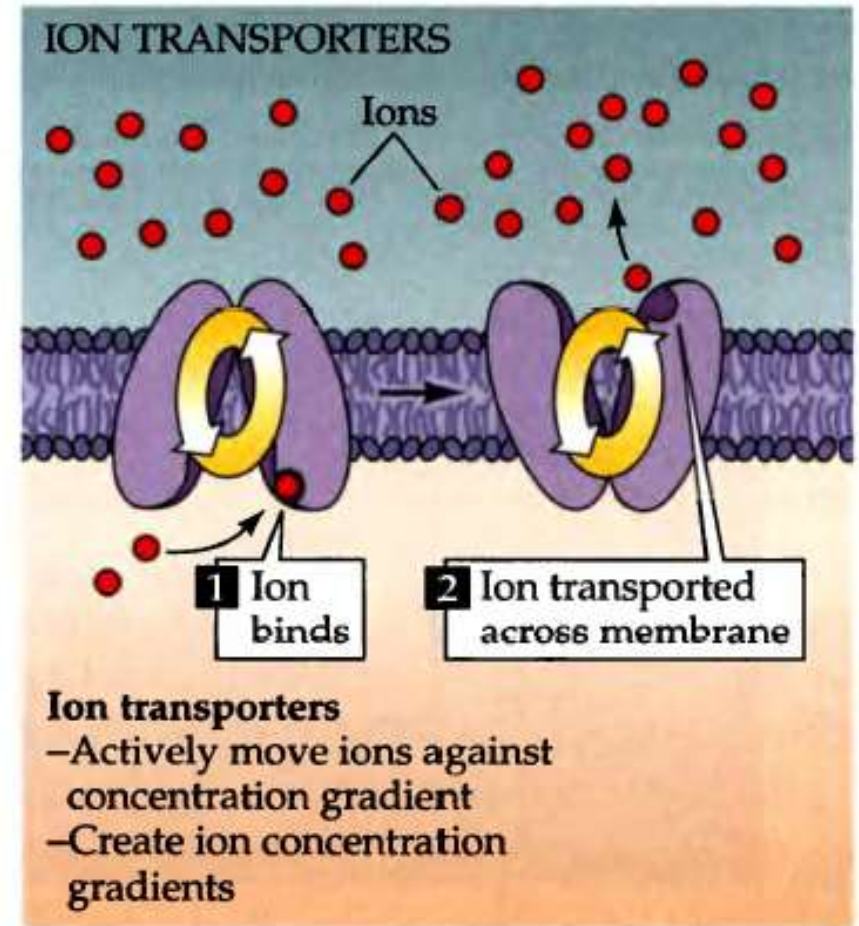
$[K^+]_{\text{in}}/[K^+]_{\text{out}} = [Cl^-]_{\text{in}}/[Cl^-]_{\text{out}}$  Equalize by different kinds of ions

# Ion distribution ?

Mechanisms exist, that establish and maintain asymmetric distribution of ions and hence, keep the cells alive ! 😊 ....

Active : pumps .. Need ATP !!! Food keeping you alive...

Passive : selectively permeable membranes



# GHK

Above derivation assumes movement in fluids.....but we have a membrane with finite thickness here !!

Constant field ... GHK

$$V = \frac{RT}{F} \ln \frac{P_K[K^+]_{\text{out}} + P_{Na}[Na^+]_{\text{out}} + P_{Cl}[Cl^-]_{\text{in}}}{P_K[K^+]_{\text{in}} + P_{Na}[Na^+]_{\text{in}} + P_{Cl}[Cl^-]_{\text{out}}} \quad (2.7.21)$$

Where  $P_x$  is the membrane permeability of ion  $x$

# Calculate membrane potentials

	Inside (mM)	Outside (mM)
<b>Frog muscle</b> (Conway 1957)		
K <sup>+</sup>	124	2.25
Na <sup>+</sup>	10.4	109
Cl <sup>-</sup>	1.5	77.5
Ca <sup>2+</sup>	4.9 <sup>†</sup>	2.1
<b>Squid axon</b> (Hodgkin 1964)		
K <sup>+</sup>	400	20
Na <sup>+</sup>	50	440
Cl <sup>-</sup>	40–150	560
Ca <sup>2+</sup>	0.4 <sup>†</sup>	10

Case 1

$$P_K : P_{Na} : P_{Cl} = 1 : 0.03 : 0.1$$

Case 2

$$P_K : P_{Na} : P_{Cl} = 1 : 15 : 0.1$$

# limitations

Membrane is not the same as an aqueous medium or aqueous pores

Useful only when voltage profiles are not really changing much....and permeability is relatively stable....

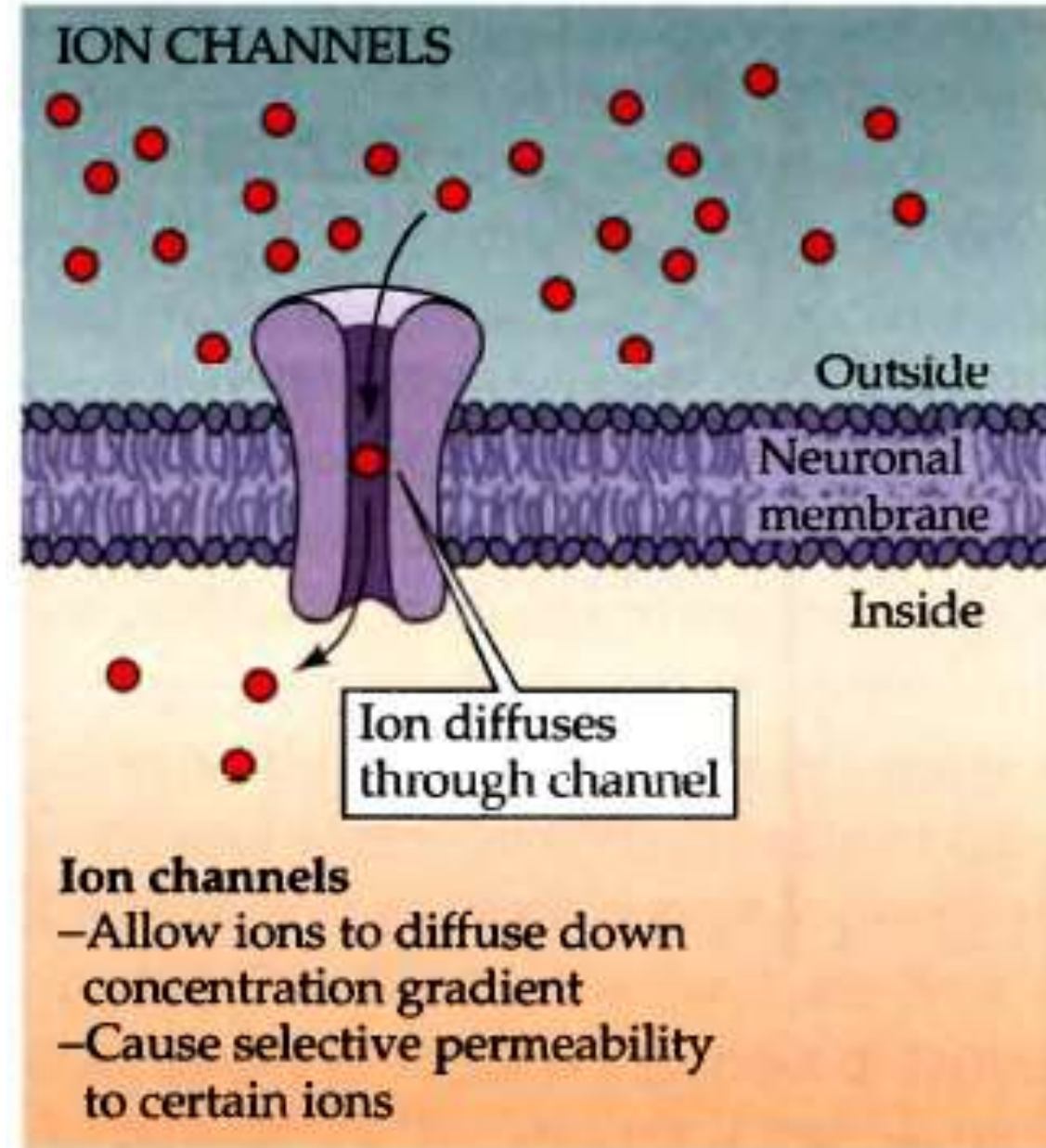
**To model, we need to understand biology !!**

# Equivalent circuit

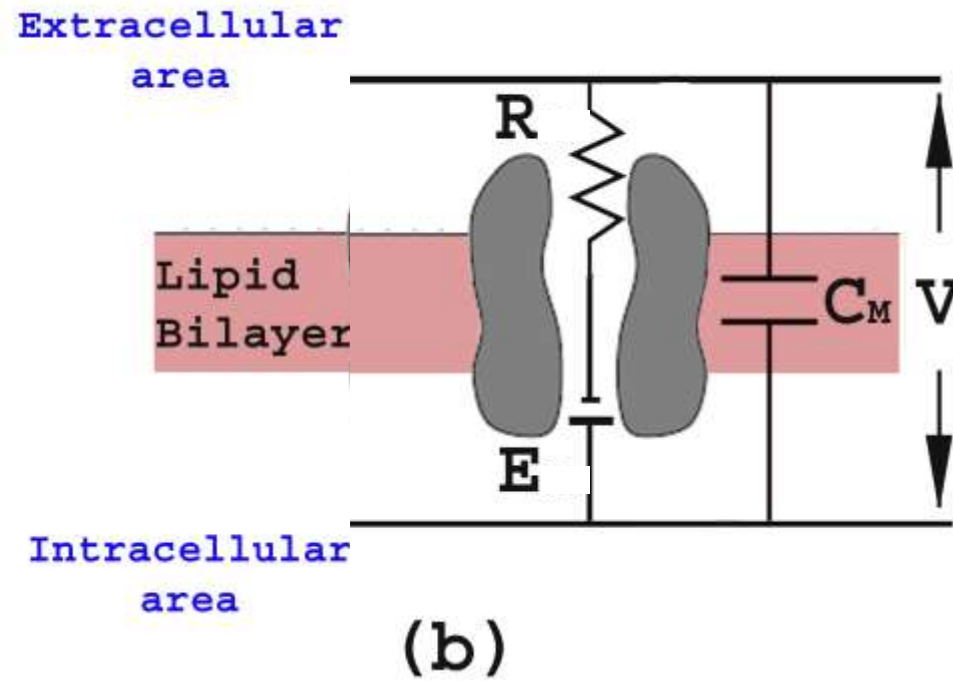
Lipid Bilayer : Insulator, separates charges

Ion channels : Allows ions to flow through  
multiple channels ?

Nernst potential : Ion imbalance at rest

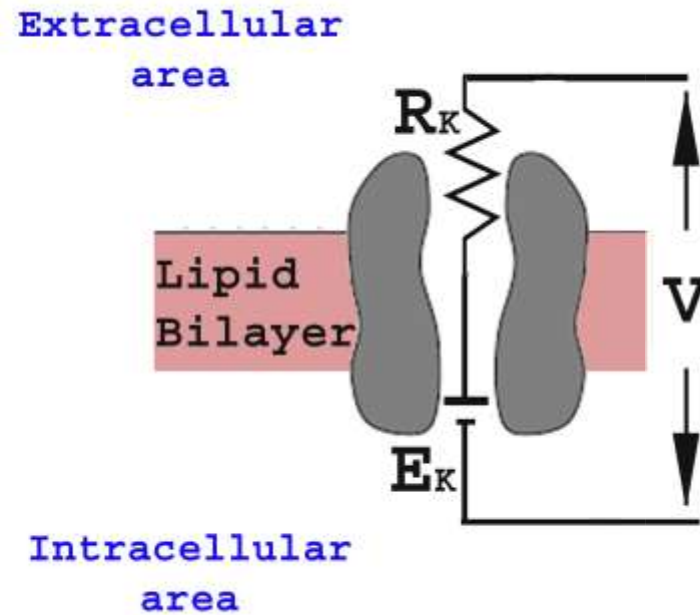


# Equivalent circuit

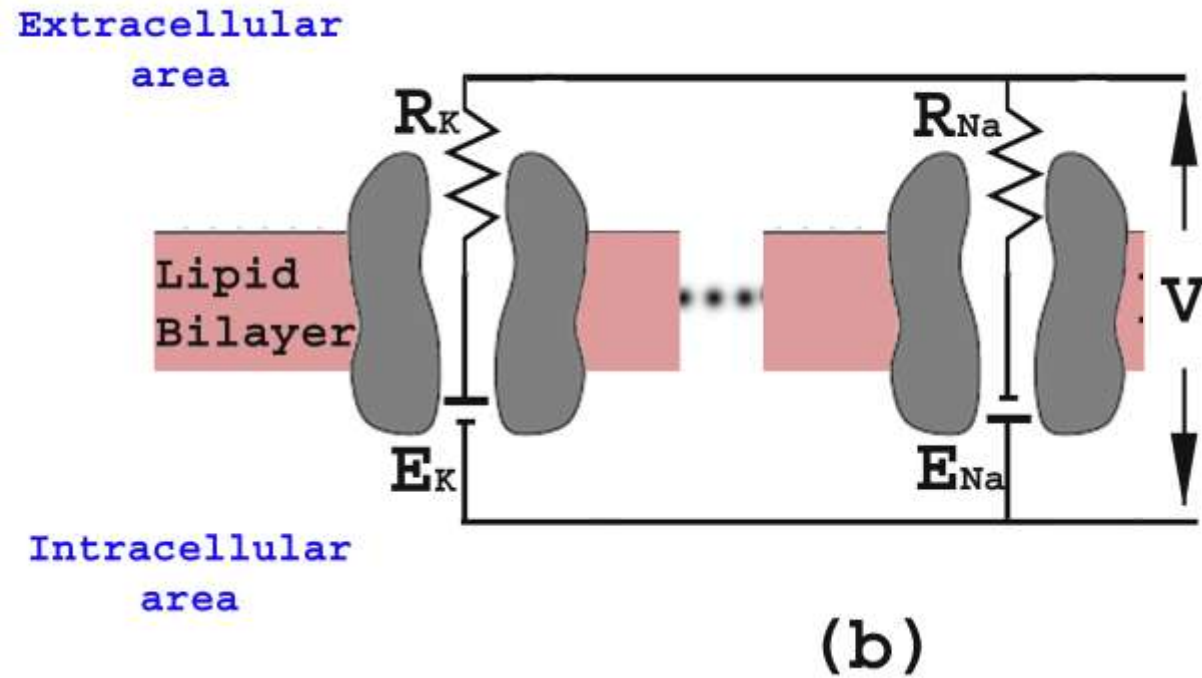




# Modeling an ion channel : a potassium channel



# Add another channel for Sodium



# Parallel conductance model

$$I_{inj} = i_c + i_{Na} + i_K$$

$$I_{inj} = C_m \frac{dV}{dt} + (V - E_K)g_K + (V - E_{Na})g_{Na}$$

