

ShahJalal University of Science & Technology, Sylhet
 Department of Biochemistry and Molecular Biology
 B. Sc. (Hons) 2nd year 1st Semester Examination, 2012
 Course No.: BMB -221 Course title: Enzymology-I
 Credit: 2 Total marks: 70 Time: 2 hours

Instructions:

- Number in the right side indicates the marks of the question
- Answer any two questions from each part (Part A & B)

Part A

- Define: Enzyme, cofactor, apoenzyme and holoenzyme. 6
- Write down the classification of enzymes. 6
- Summarize the key features of the active sites of enzymes. 2.5
- Deduce the Michaelis-Menten equation for single substrate enzyme reaction. 10
- Write down the significance of K_m and V_{max} value. 4.5
- Justify that allosteric enzyme do not obey Michaelis-Menten kinetics. 3
- Justify: 5.5
 - Enzymes have immense catalytic power. 15
 - Binding energy contributes to specificity and catalytic activity of enzymatic reactions. 2.5
 - Weak interactions between enzymes and substrate are optimized in the transition state.
 - Enzymes alter only in reaction rate not the reaction equilibrium-explain.

Part B

- What is enzyme inhibition? Contrast reversible and irreversible enzyme inhibition. 6
- Describe the effects of competitive and non-competitive inhibitors on the kinetics of enzyme reactions. 4
- Describe the functions and uses of enzyme inhibitors. 6
- Describe the effects of pH and temperature on enzyme reaction. 6.5
- Briefly describe the effects of coenzyme and cofactor on enzyme reaction. 5
- Describe how the K_m for an enzyme may be experimentally determined. 5.5
- Write down the factors that control in enzyme assays. 3.5
- Enzymes are high specific"- explain. 3.5
- Write short notes on any three of the followings: 9
 - Double-displacement enzyme (ping-pong) reaction.
 - Enzyme units.
 - Lineweaver-Burk plot.
 - K_m criterion in kinetic perfection.
 - Enzyme assay: equilibrium method.
 - Inhibition.