The probability that the harmonic oscillator is in ground state $E=E_0$ that the harmonic oscillator is in ground state $E=E_0$ is proportional to # microstates of the combined system given that oscillator is in its ground state.

If osc. Hator is in its ground state, $\Omega_0 = 1$ $\Omega = 1 \Omega_R$

$$S = k \ln S \rightarrow \Omega = e^{S/k}$$

Consider two oscillator states A&B, suppose E(A) < E(B)

$$E_R(A) > E_R(B)$$
 because energy is conserved

$$P(A) > P(B)$$
States are

more likely

$$S_{2}(A)/k$$

$$S_{3}(A)/k$$

$$S_{4}(A)/k$$

$$S_{5}(A)/k$$

$$S_{5}(A)/k$$

$$S_{5}(A)/k$$

$$\frac{P(A)}{P(B)} = \frac{\Omega_R(A)}{\Omega_R(B)} = \frac{e^{S_R(A)/k}}{e^{S_R(B)/k}} = e^{[S_R(A) - S_R(B)]/k}$$

$$dS_R = \frac{1}{T} dU_R = -\frac{1}{T} dU_0$$

$$S_R(A) - S_R(B) = -\frac{1}{T} (E(A) - E(B))$$

$$\frac{P(A)}{P(B)} = \frac{-E(A)/kT}{e^{-E(B)/kT}}$$

· Z is a normalization constant $\sum_{\text{all}} \frac{1}{Z} e^{-E(s)/kT} = 1$ ridicultusly useful states

 $e^{-E/kT}$ Boltzmann factor for a given state $P(A) \sim e^{-E(A)/kT}$

if energy increases by kT

probability reduced by /e (to 37% of orging)

KT: characteristic ereigy scale

(a) 300 K, kT = 40 eV

e.g. hydrogen atom

for I mal H, 1-10,000 excited ones

$$Z = \sum_{s} e^{-E_{s}/k\bar{t}}$$
 sum of all Boltzmann factors

sum of all Boltzmann factors

$$\beta = \frac{1}{kT}$$

If ground state is
$$E = 0$$
,

 $Z = 1 + \sum_{\text{excited}} e^{-\beta E_s} \leftarrow \alpha II < 1$

$$\mathcal{L} = 1 + \sum_{\text{excited}} 1$$

12 is a count of all microstates

Z is a reighted court of all microstates, weighted by e

in contact with a Hernel reservoir

$$\langle X \rangle = \frac{1}{2} \sum_{s} X_{s} e^{-\beta E_{s}}$$
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 $\langle E \rangle = \frac{1}{2} \sum_{s} E_{s} e^{-\beta E_{s}}$

Next track: $E_{s} e^{-\beta E_{s}} = \frac{1}{2} \sum_{s} e^{-\beta E_{s}}$
 $\langle E \rangle = \frac{1}{2} \sum_{s} e^{-\beta E_{s}}$

$$\frac{z-\frac{1}{2}\frac{\partial}{\partial\beta}\sum_{s}e^{-\beta E_{s}}}{\left(E\right)^{2}=-\frac{1}{2}\frac{\partial}{\partial\beta}}=-\frac{\partial\ln^{2}}{\partial\beta}=-\frac{\partial\ln^{2}}{\partial\beta}$$