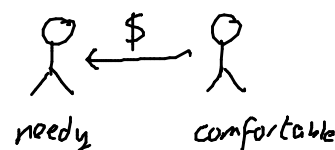


money  $\leftrightarrow$  energy  
happiness  $\leftrightarrow$  entropy

$$\text{need} = \frac{dU}{d\$} \leftrightarrow \frac{1}{T} = \frac{dS}{dU}$$

$$\text{comfort} = \frac{1}{\text{need}} \leftrightarrow T$$

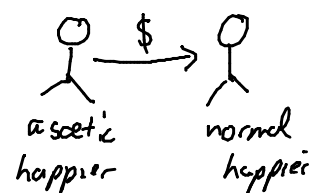


normal: become more comfortable as they become wealthier (get more money)  
 $U \sim T$

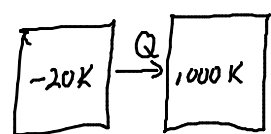
miserly: becomes needier with wealth (eg. planet around sun)

ascetic: (Zen): becomes happier when they lose money

$$\frac{dU}{d\$} < 0 \rightarrow \frac{1}{T} < 0$$

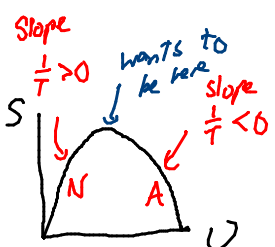


ascetics will always give money to normal people (or misers)

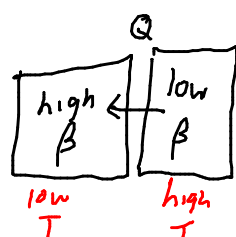
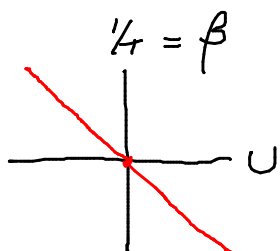
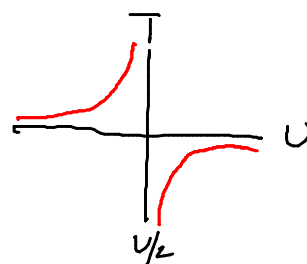


Negative temperatures are hotter than positive temperatures!

eg. Paramagnet



ascetic phases appear when energy is bounded



Heat flows from low  $\beta$  to high  $\beta$  and negative  $\beta$  is lower than positive  $\beta$ .

$$\beta = \frac{1}{T}$$

-1 K  $\leftrightarrow$  -100 K  
-1/K  $\leftrightarrow$  -0.01/K  
lower  $\beta$  hotter  $\leftrightarrow$  colder

# Chapter 5

$U$  is internal energy of system  
but it's not the only energy associated w/ systems.

eg. system in air is holding back the surrounding air  
- if system vanished suddenly, air would rush back in  
potential energy :  $PV$

$$H = U + PV \quad \text{enthalpy of system}$$

e.g. ideal gas

$$H = N \frac{f}{2} kT + NkT = \frac{f+2}{2} NkT$$

$$\Delta H = \Delta U + \Delta(PV)$$

At constant pressure,

$$\Delta H = \Delta U + P\Delta V$$

$$W_c = -P\Delta V$$

$$= Q + \cancel{W_c} + W_{\text{other}} + \cancel{P\Delta V}$$

$$\Delta H = Q + W_{\text{other}} \quad \text{at constant } P$$

↑  
friction

microwave oven, etc

$$C_p = \frac{Q}{\Delta T} = \frac{\Delta H}{\Delta T} = \left( \frac{\partial H}{\partial T} \right)_P \quad C_v = \left( \frac{\partial U}{\partial T} \right)_V$$

$H$  at const  $P$  is analogous to  $U$  at const  $V$ .

When system is annihilated (ignoring  $E=mc^2$ )

$H$  is total energy released

Some can be used as work, but not all

Why? System had some entropy which can't be destroyed.

Heat must flow out  $Q = TS$ .

$H - TS$  can be used as work if you choose  
"free energy"

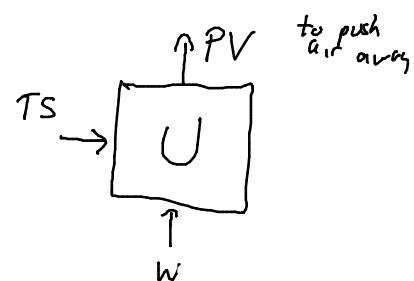
$$F = U - TS \quad \text{Helmholtz free energy}$$

$$G = H - TS \quad \text{Gibbs free energy}$$

in reverse situation if a system is created,

$F$  or  $G$  must be supplied as work

but  $TS$  can be drawn from environment



$$H = U + PV$$

$$F = U - TS$$

$$G = H - TS = F + PV$$