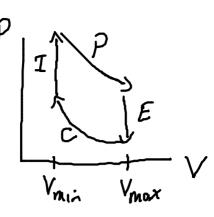
Otto cycle



air-gasoline mix

1) compression (adiabatic)

2) ignition at constant V

3) power (adiabatic)

4) exhaust

Vmax >1 compression ratio

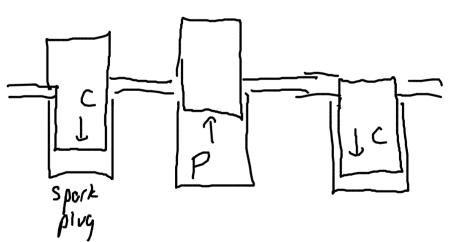
$$\mathcal{T} \in \left[- \left(\frac{V_{\text{min}}}{V_{\text{max}}} \right)^{\gamma - 1} \right]$$

8 : a diabatic exponent

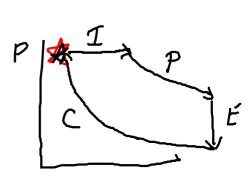
to increase efficiency, compress a lot

but ... If too much, too hot,

& gasoline pre-ignites



Diesel engire



fuel is only added at the no pre-ignition engine & has much larger

Compression ratios fuel can ignite due to temperature of the air-no spork plugs Kefrigerator - transfers heat from cold to hot Te Q Oh Violates and law of Hermodynamics fridge gains $S_{in} = \frac{Q_{in}}{T_{in}} \ge \frac{Q_{in}}{T_{c}}$ $\left(T_{in} \le T_{c}\right)$ $loses S_{out} = \frac{Q_{out}}{T_{out}} \le \frac{Q_{out}}{T_{H}} \left(T_{out} \ge T_{H}\right)$ () = AS = Sin - Sout ≥ Qin - Qour If Qin = Qout, then positive AS = Qin (To -TH)

posity To small TH big

to big the small : 'SS>O not a cycle Need Quit > Qin : I must put work in To Que Quet In Q m + W = Qout Coefficient of Performance $COP = \frac{Q_{in}}{W} = \frac{Q_{in}}{Q_{out} - Q_{in}}$ if COP = 5, new 20 J of work to remove 100 J of heat from cold reservoir 0 > Qin - Qort $\frac{\hat{Q}_{out}}{T_{11}} \geq \frac{\hat{Q}_{1n}}{T_{C}}$ $\frac{T_{c}}{T_{H}} \geq \frac{Q_{in}^{\perp}}{Q_{out}} \qquad \frac{1}{CoP} = \frac{1}{Q_{out}} \leq \frac{1}{T_{H}} = \frac{T_{c}}{T_{H} - T_{c}}$ COP larger IF TH-To 15 Small e.s. Tc = 273K TH = 300K -> COP ≤ 10. A Carnot refregerator is a Carnot engine in reverse: 1) lower T to just below To quickly (2) let host flow in from To isothernow, 3 Paise T to just above The quickly (4) let leat flow out to TA isothernolly