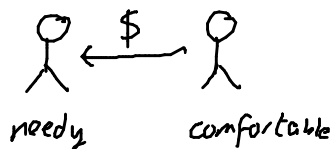


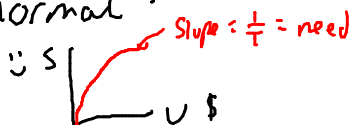
money \leftrightarrow energy
happiness \leftrightarrow entropy

$$\text{need} = \frac{d\ddot{U}}{d\$} \leftrightarrow \frac{1}{T} = \frac{dS}{dU}$$

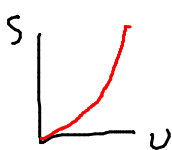
$$\text{Comfort} = \frac{1}{\text{need}} \leftrightarrow T$$



normal: become more comfortable as they become wealthier (get more money)
 $U \sim T$

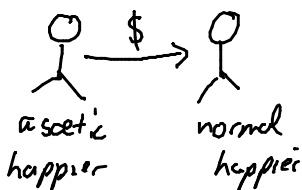


miserly: becomes needier with wealth
(e.g. planet around sun)

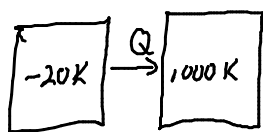


ascetic: (Zen): becomes happier when they lose money

$$\frac{d\ddot{U}}{d\$} < 0 \rightarrow \frac{1}{T} < 0$$

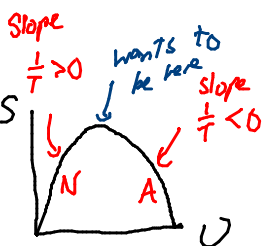


ascetics will always give money to normal people (or misers)

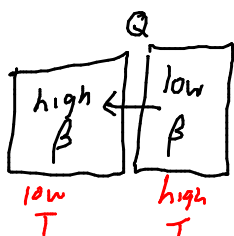
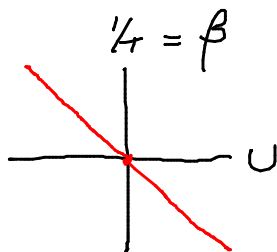
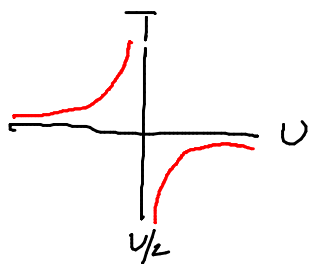


Negative temperatures are hotter than positive temperatures!

e.g. Paramagnet



ascetic phases appear when energy is bounded



Heat flows from low β to high β and negative β is lower than positive β .

$$\beta = \frac{1}{T}$$

$$-1\text{ K} \quad \text{vs} \quad -100\text{ K}$$

$$-1/\text{K}$$

$$-0.01/\text{K}$$

lower β
hotter

colder

Chapter 5

U is internal energy of system
but it's not the only energy associated w/ systems.

eg. system in air is holding back the surrounding air
- if system vanished suddenly, air would rush back in
potential energy : PV

$$H = U + PV \quad \text{enthalpy of system}$$

e.g. ideal gas

$$H = N \frac{f}{2} kT + NkT = \frac{f+2}{2} NkT$$

$$\Delta H = \Delta U + \Delta(PV)$$

At constant pressure,

$$\Delta H = \Delta U + P \Delta V$$

$$= Q + \cancel{W_c} + W_{\text{other}} + \cancel{P \Delta V}$$

$$\Delta H = Q + W_{\text{other}} \quad \text{at constant } P$$

\uparrow
Friction
microwave oven, etc

$$W_c = -P \Delta V$$

$W_{\text{other}} = 0$

$$C_p = \frac{Q}{\Delta T} = \frac{\Delta H}{\Delta T} = \left(\frac{\partial H}{\partial T} \right)_P \quad C_v = \left(\frac{\partial U}{\partial T} \right)_V$$

H at const P is analogous to U at const V .

When system is annihilated (ignoring $E=mc^2$)

H is total energy released

Some can be used as work, but not all

Why? System had some entropy which can't be destroyed.

Heat must flow out $Q = TS$.

$H - TS$ can be used as work if you choose
"free energy"

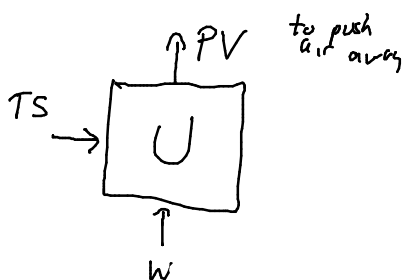
$$F = U - TS \quad \text{Helmholtz free energy}$$

$$G = H - TS \quad \text{Gibbs free energy}$$

in reverse situation if a system is created,

F or G must be supplied as work

but TS can be drawn from environment



$$H = U + PV$$

$$F = U - TS$$

$$G = H - TS = F + PV$$