



Intercalation phase diagram of Mg in V_2O_5 from first principles

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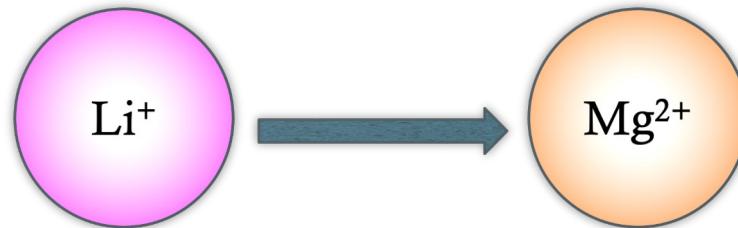
G. Sai Gautam, P. Canepa, A. Abdellahi, A. Urban, R. Malik, G. Ceder
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(submitted)

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V_2O_5 : Critical to cathode design of Mg-batteries

Why Mg?

- Next generation electrical devices benefit from high energy density storage systems
- Superior volumetric capacity for Mg metal anode ($\sim 3833 \text{ mAh/cm}^3$) vs. Li metal anode ($\sim 2046 \text{ mAh/cm}^3$)



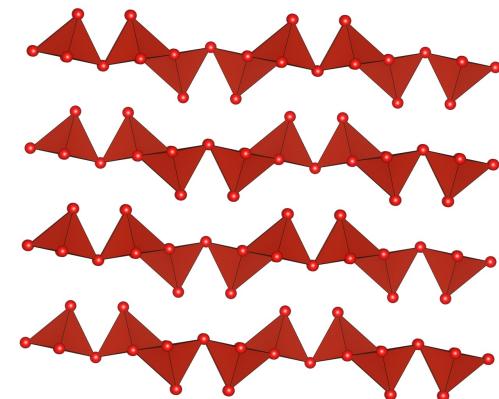
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- Superior volumetric capacity for Mg metal anode ($\sim 3833 \text{ mAh/cm}^3$) vs. Li metal anode ($\sim 2046 \text{ mAh/cm}^3$)
- New chemistry: Cathode design challenge
 - High voltage, high rates, high capacity

Why V_2O_5 ?

- One of only 3 cathodes to reversibly intercalate Mg
 - Others: Chevrel Mo_3S_4 ,^[1] Layered MoO_3 ^[2]
 - Higher voltage and lower volume change in V_2O_5
- Known Li intercalant



1. Aurbach *et al.*, Nature, 2000
2. Gershinsky *et al.*, Langmuir, 2013

How does Mg intercalate into V₂O₅?

Characterize the system through DFT

Ground State hull and Voltage curves

- Benchmark with experiments



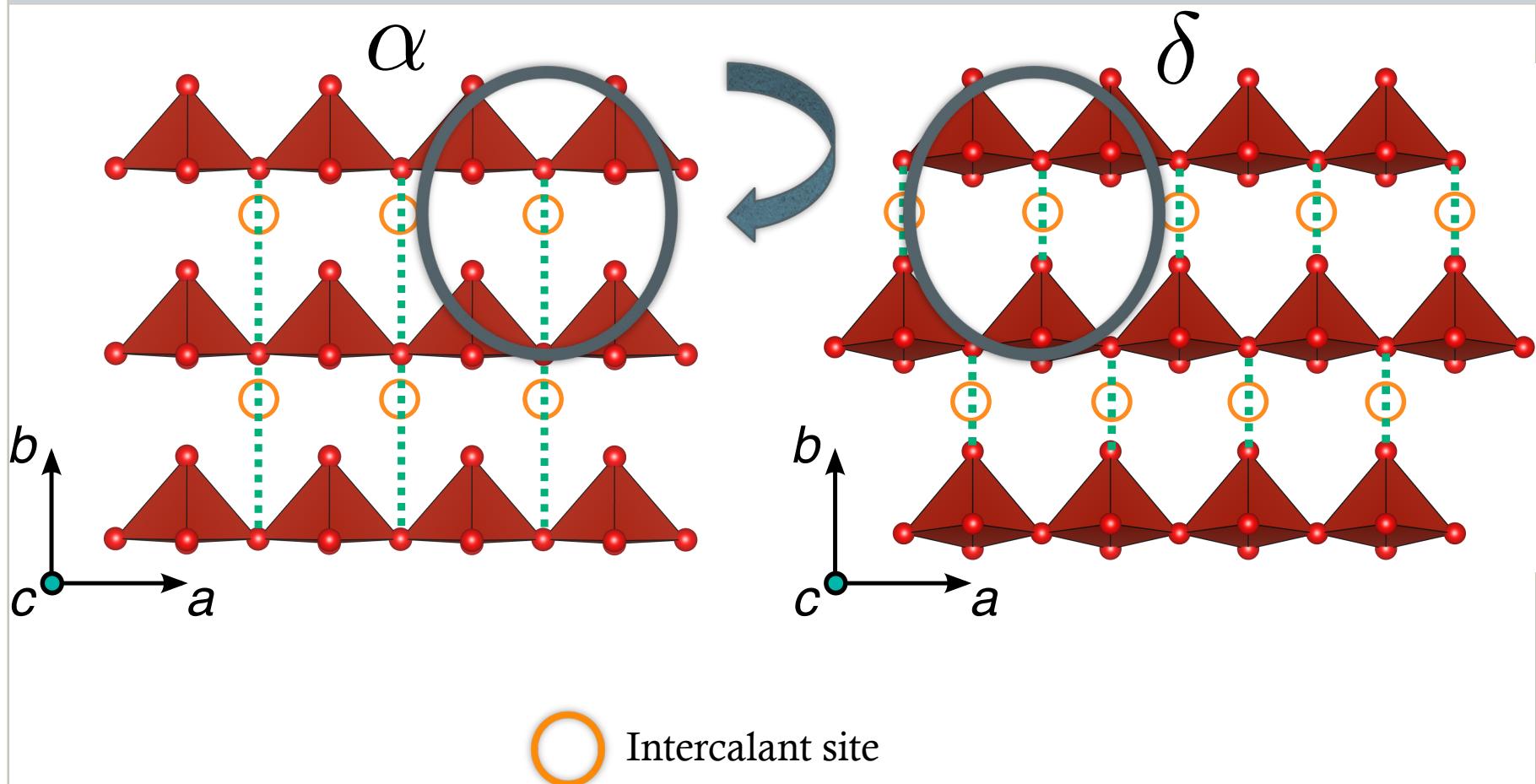
Mg migration barriers

- Determine phase(s) of interest



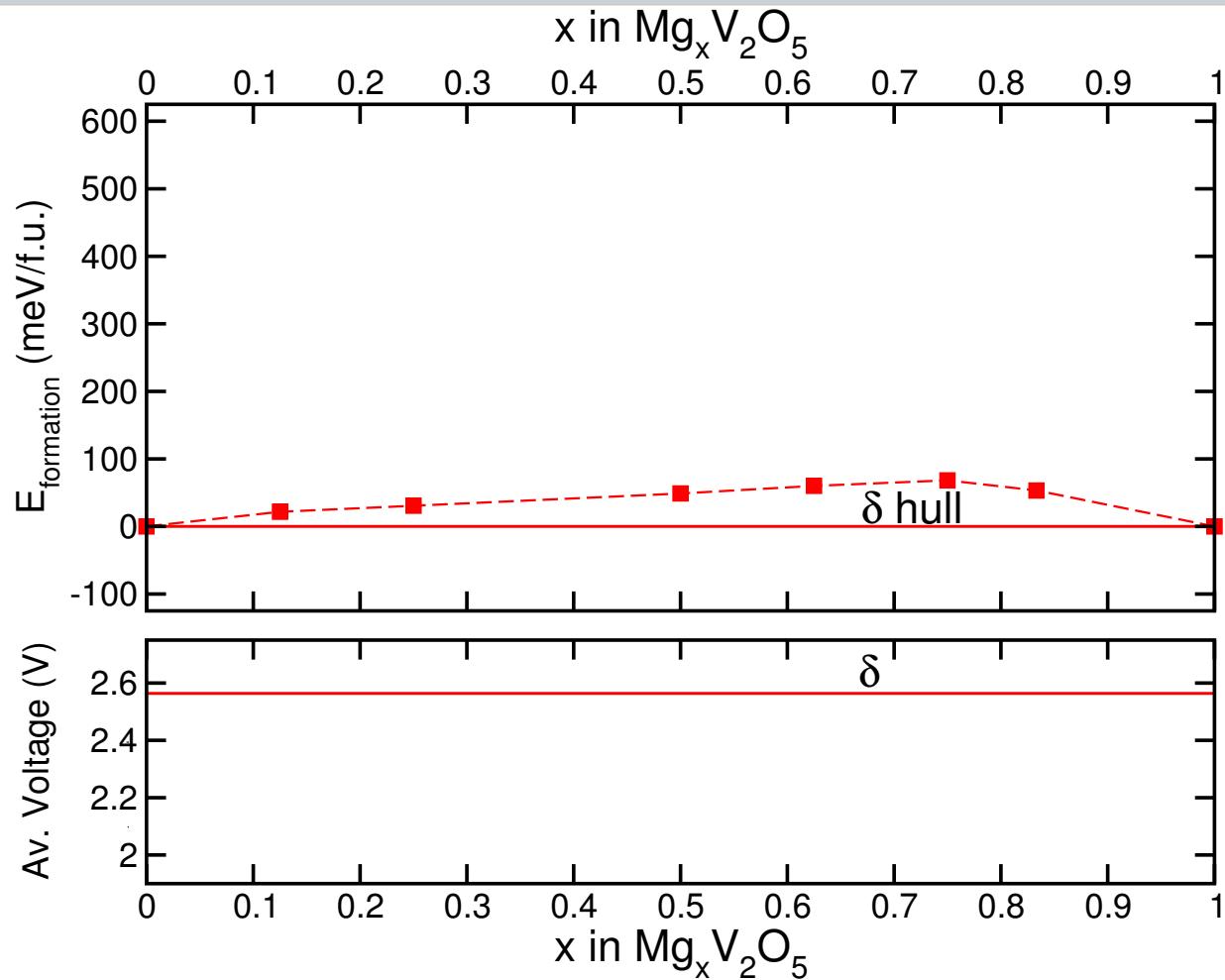
Suggestions to improve performance

Polymorphs of Orthorhombic V_2O_5



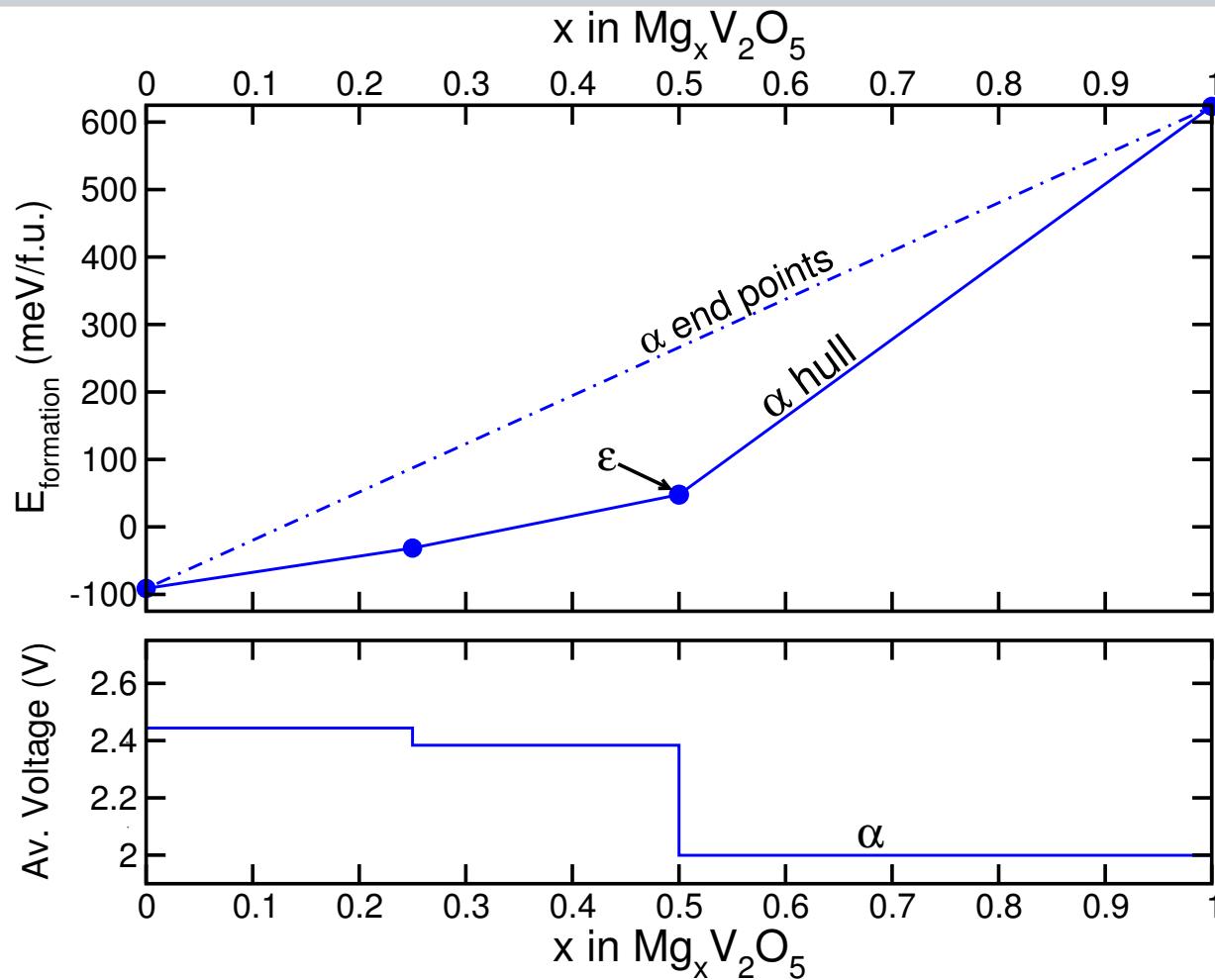
Ground State Hull and Voltages

δ inserts Mg at a higher voltage than α



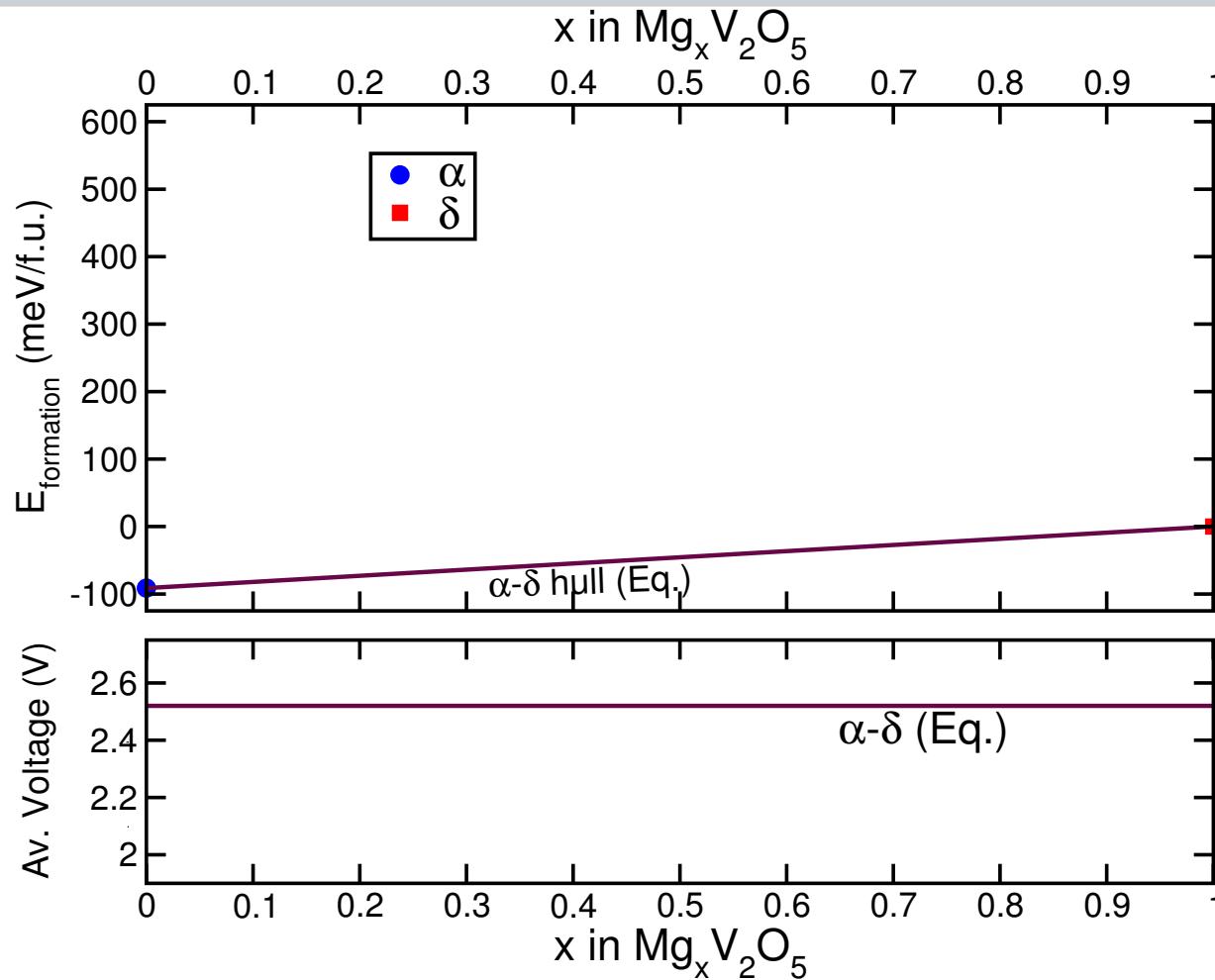
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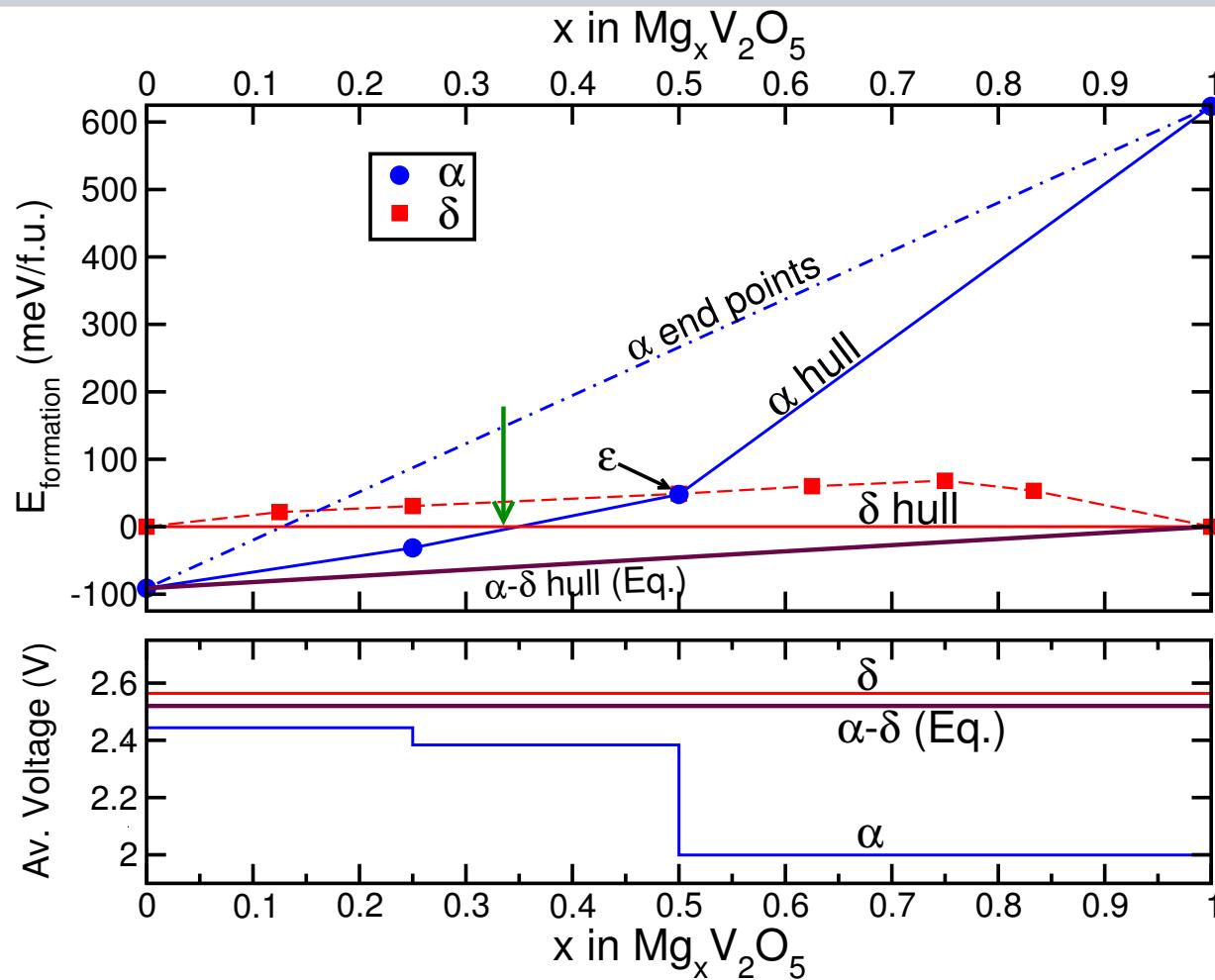
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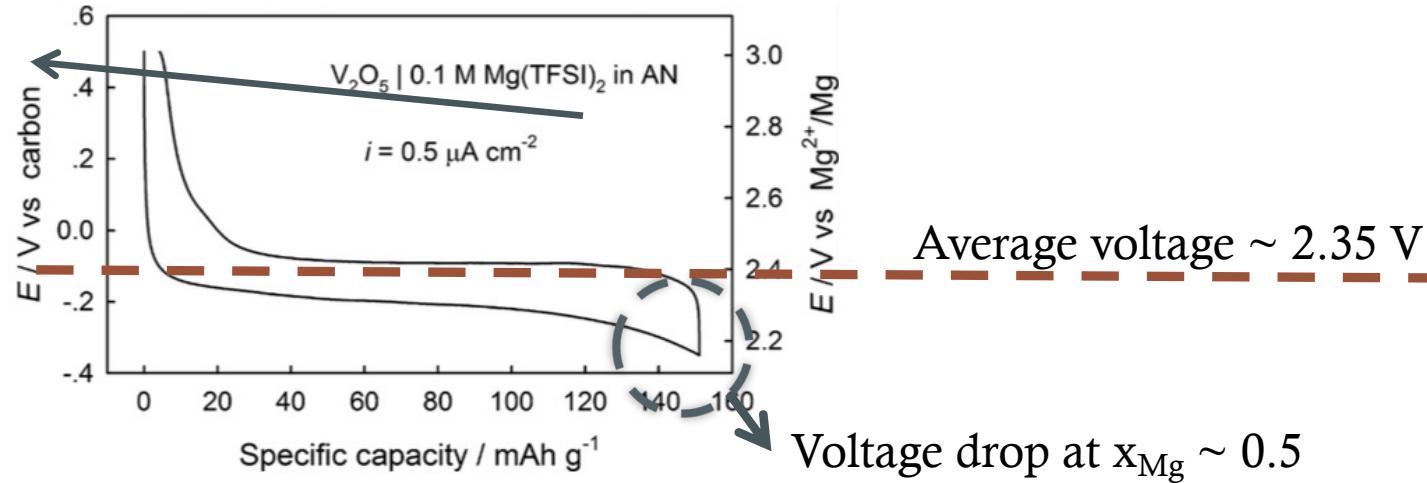
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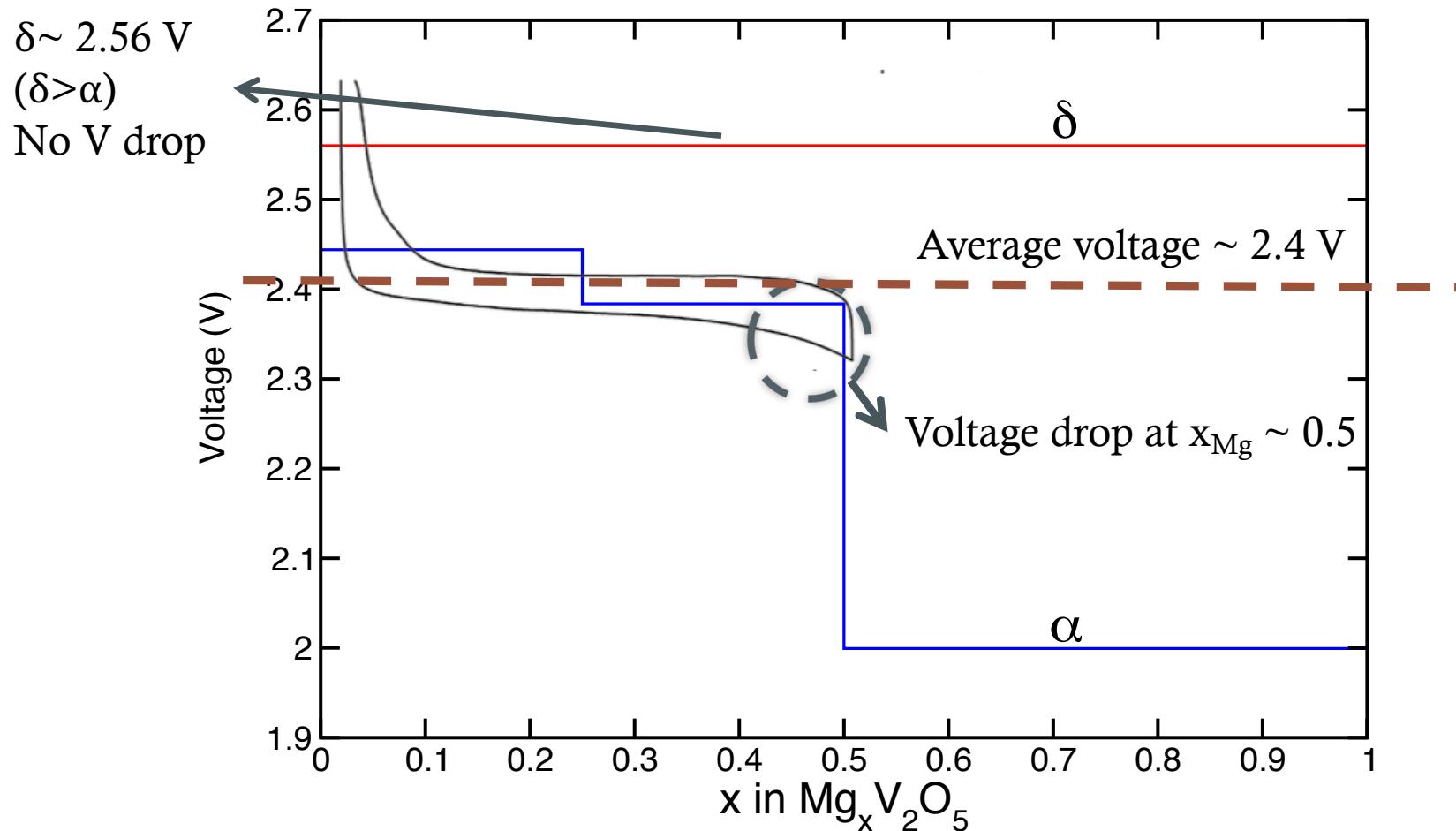
Experimental voltage profile matches α

$\delta \sim 2.56 \text{ V}$
 $(\delta > \alpha)$
No V drop



Experimental voltage curve:
Gershinsky *et al.*, Langmuir, 2013

Experimental voltage profile matches α

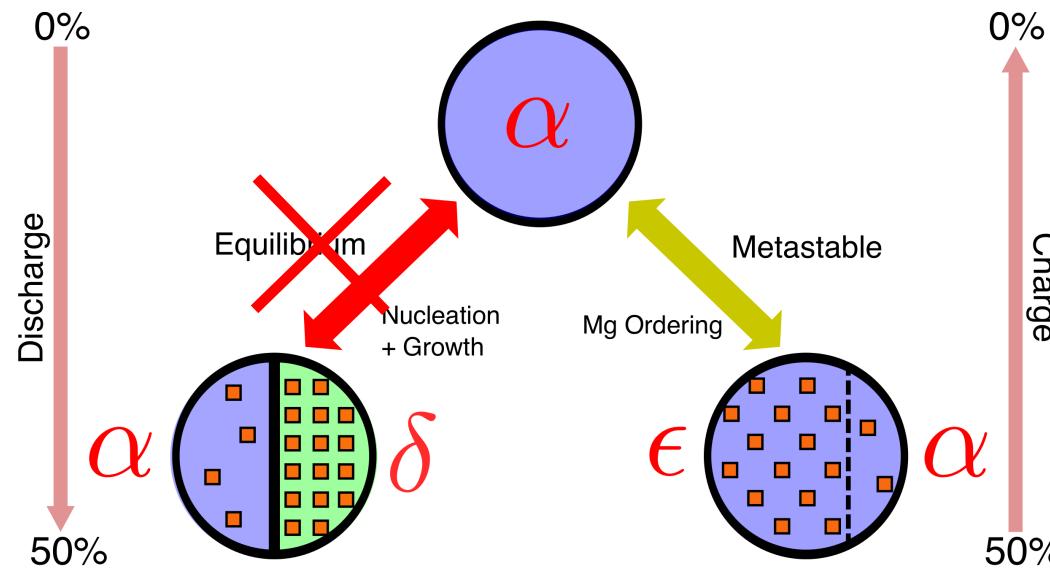


Experimental voltage curve:
Gershinsky *et al.*, Langmuir, 2013

Experiments cycle Mg in α -V₂O₅

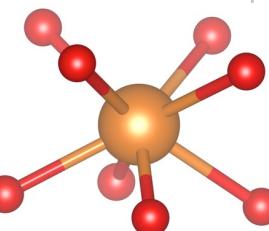
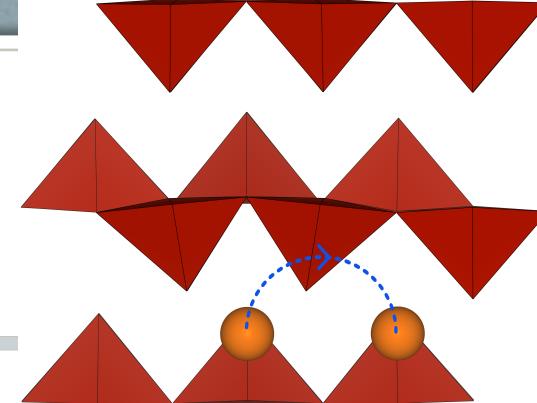
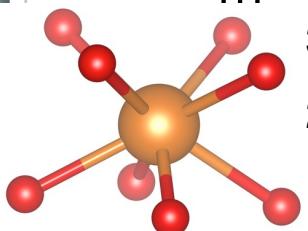
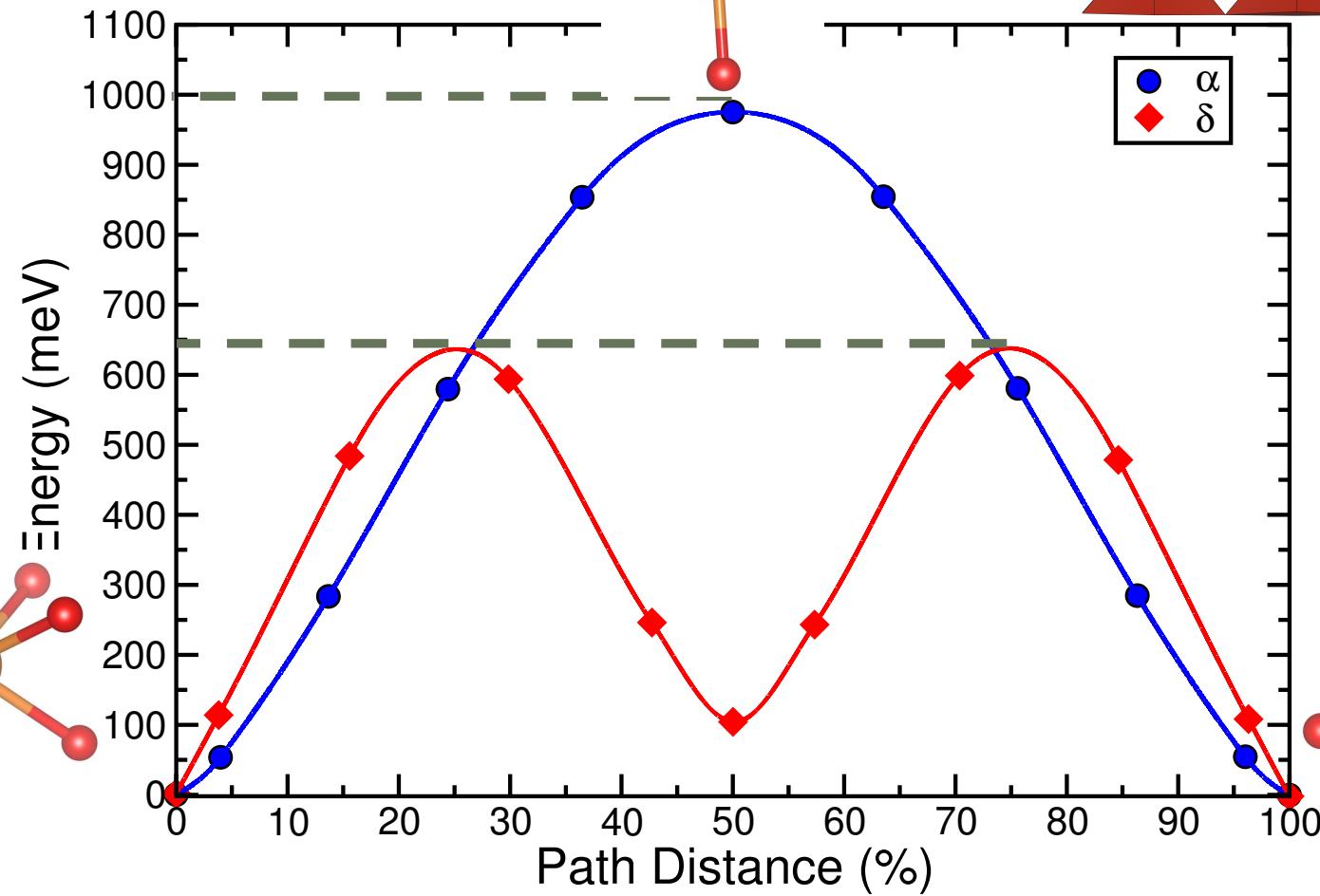
When Mg cycling is started in empty (charged) V₂O₅

- Experimental voltage profile matches better with α
- $\alpha \rightarrow \delta$ transformation requires structural rearrangement
- δ -V₂O₅, if accessed, could be metastable upon Mg cycling
 - δ -MgV₂O₅ has been experimentally synthesized³



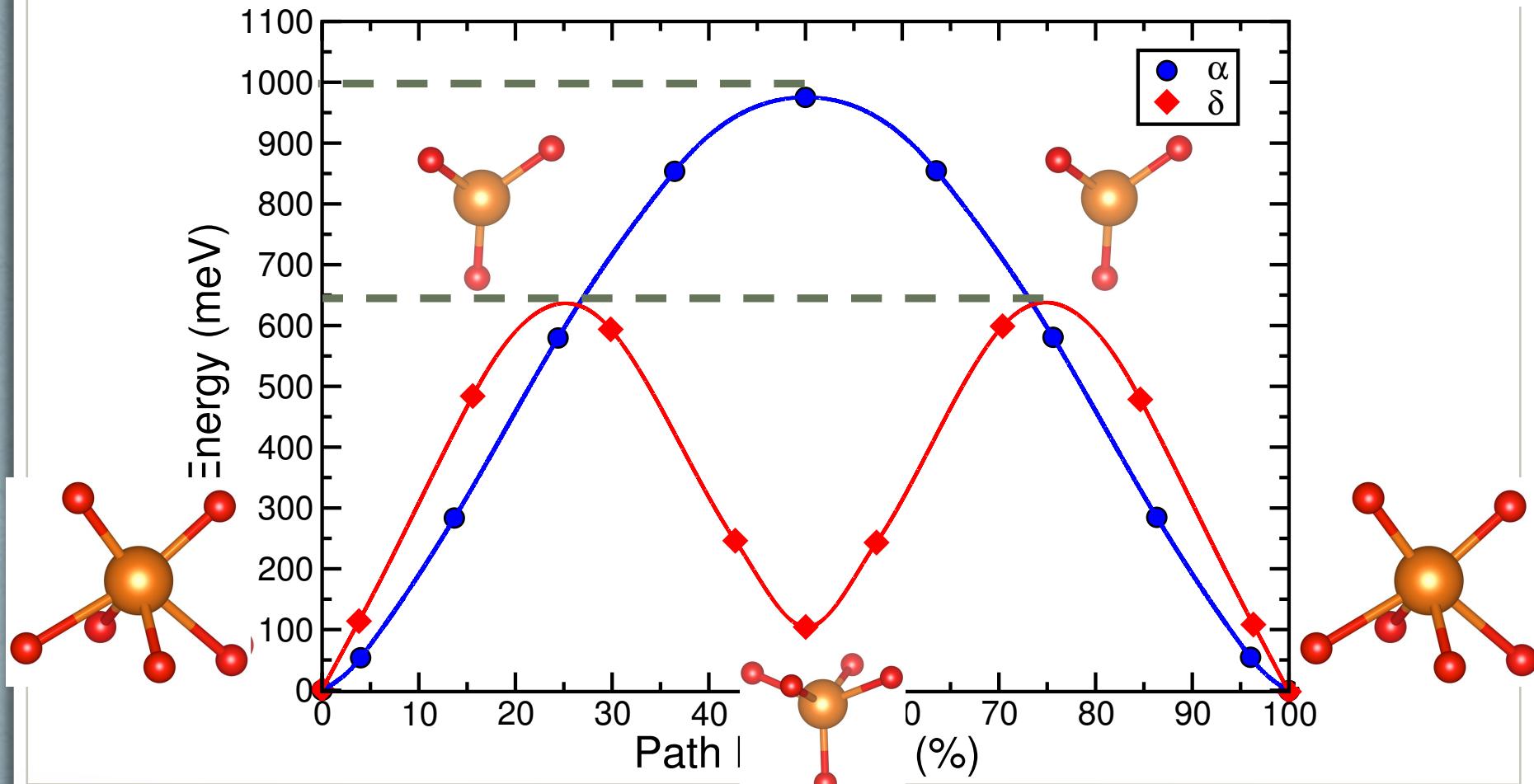
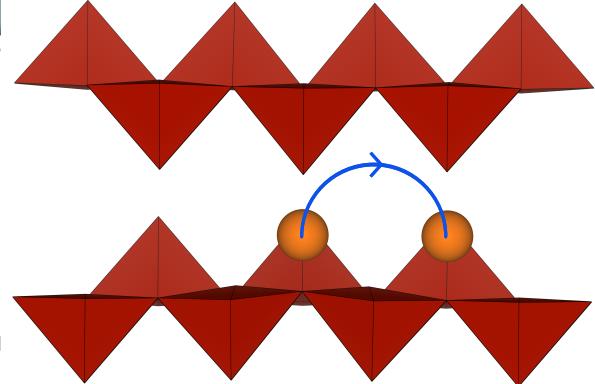
Mg Migration barrier

δ is a better diffuser



Mg Migration barrier

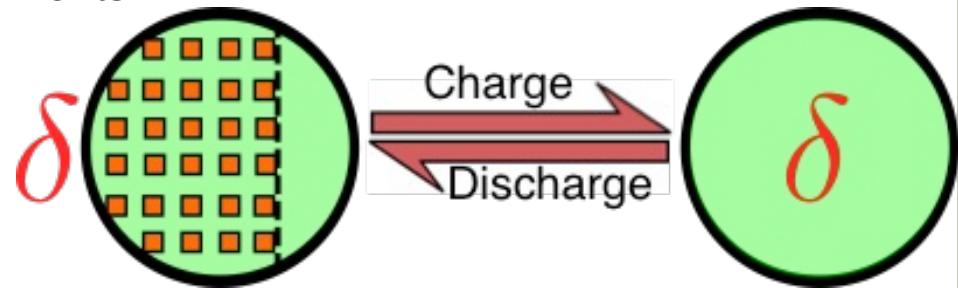
δ is a better diffuser than α



Conclusions

Mg cycling in δ is better than α

- Mg cycling when begun in empty (charged) V_2O_5 stays in α
 - Voltage profile matches with experiments
- δ is better than α
 - Lower Mg migration barrier(s)
 - Higher Mg insertion voltage



Mg cycling when begun in full (discharged) V_2O_5 could stay in δ

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