## Blackbody Radiation

All matter absorbs and enits electromagnetic radiation over a large frequency spectrum. Since energy is released through this mechanism, we have to take this radiation into account when we work at thermodynamic applications.

One way of doing this is by looking at the spectral flux of an object. This is a measure of irradiance or emission intensity, with units  $win^2Hz^2$  and denoted by  $\Phi(f)$ . So it we are trying to describe a body with one A emitting radiation within a frequency range  $f \rightarrow f + \Delta f$ , over a time period  $\Delta f$ , we can give total energy:  $E = \Phi(f) \land \Delta f \Delta f$ 

when we measure the spectral flue of various bodies, it seems to depend on the material and temperature of the object. However, in the case of a perfect absorbing object, a "blackbody," it is only dependent on temperature.

One way to imagine a blackbody is as a hollow object with a very small hole. If incident radiation posses through the hole, it will bounce around inside and never escape, since the hole is so small. The radiation is 100% absorbed.



The ray will never escape and thus has been perfectly absorbed.

## wier's Low

For a blackbody, the spectral flux depends only on temperature. There will be a peak in spectral flux (and thus a peak in frequency of omitted radiation) at some temperature. For a perfect blackbody, the curves showing intensity emitted against frequency emitted are called Planck's curves.

It can be shown that the peak treamency occurs at:

This can be translated to wavelength as well using  $f\lambda = C$ :

## Stefon's Law

The power emitted per unit over is called the flux and is given by:

or is the Stefen-Bultzmann constant of \$5.67x10-8.

The power emitted is in the form of heat, so me can work out the heat flow of an object using: