

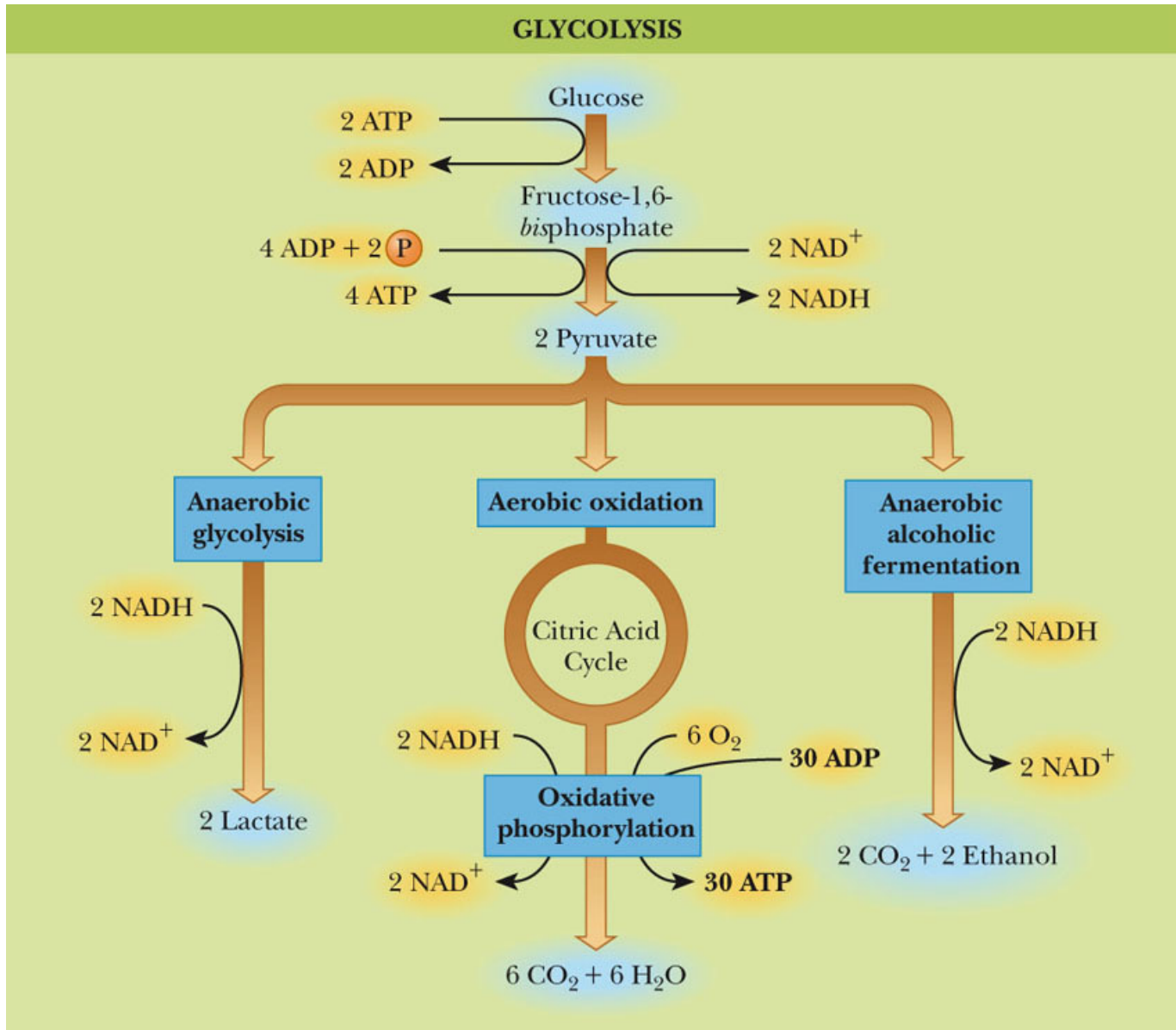
Glycolysis

Chapter 17

The Overall Pathway of Glycolysis

- Glycolysis is the first stage of glucose metabolism
- One molecule of glucose is converted to pyruvate which gives rise to 2 molecules of ATP via substrate level phosphorylation.
- Note that all of the steps of glycolysis are enzymatically catalyzed.
- It plays a key role in the way organisms extract energy from nutrients
- Once Pyruvate is formed, it has one of several fates

Fates of Pyruvate From Glycolysis



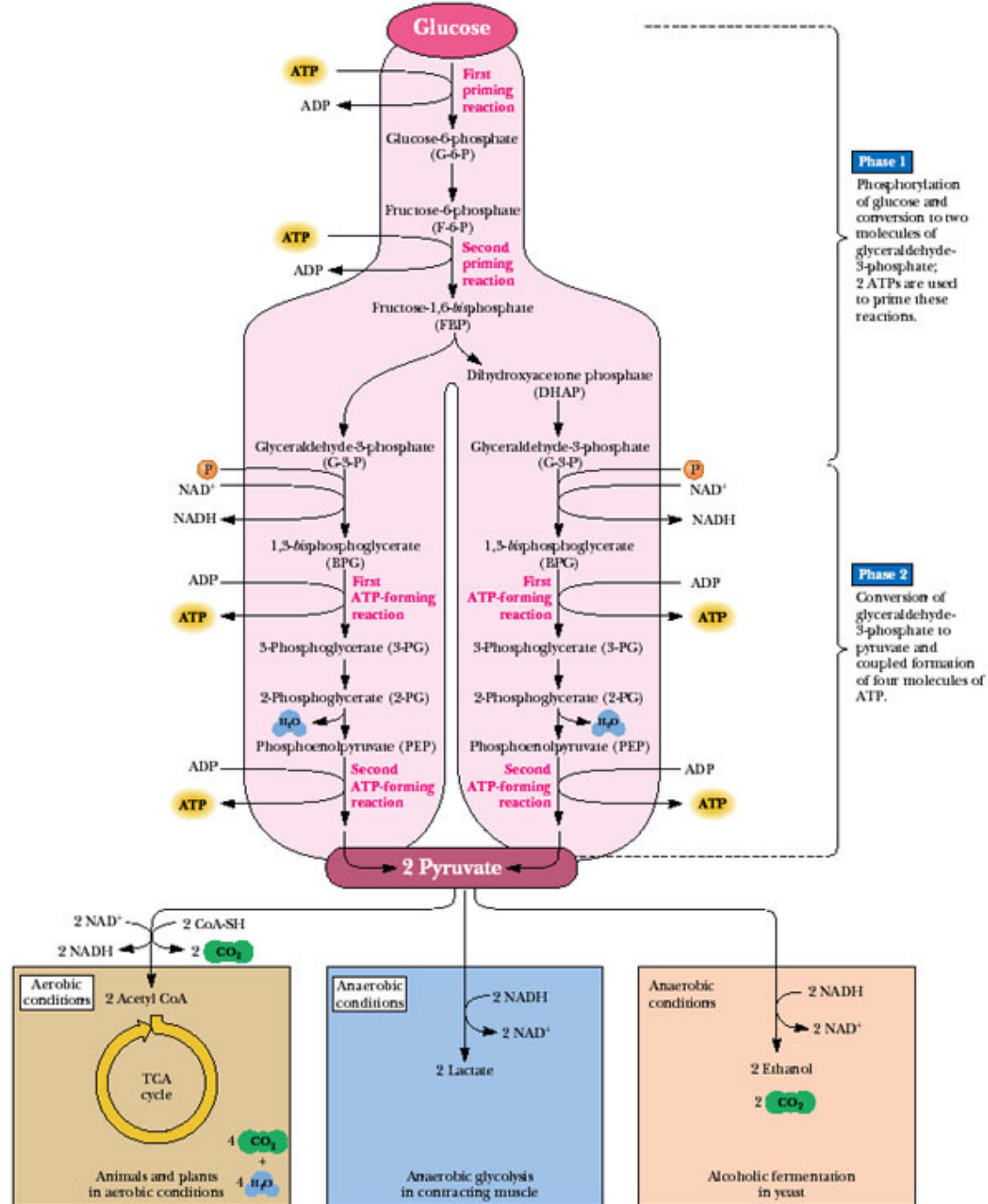


Fig. 17-2, p.465

The Reactions of Glycolysis

- **Phosphorylation** of glucose to give glucose-6-phosphate
- **Isomerization** of glucose-6-phosphate to give **fructose-6-phosphate**
- **Phosphorylation** of fructose-6-phosphate to yield fructose-1,6-bisphosphate
- **Cleavage** of fructose-1,6-bisphosphate to give **glyceraldehyde-3-phosphate** and **dihydroxyacetone phosphate**
- **Isomerization** of dihydroxyacetone phosphate to give **glyceraldehyde-3-phosphate**

The Reactions of Glycolysis (Cont'd)

- **Oxidation** of glyceraldehyde-3-phosphate to give 1,3-bisphosphoglycerate
- **Transfer of a phosphate group** from 1,3-bisphosphoglycerate to ADP to give ATP
- **Isomerization** of 3-phosphoglycerate to give **2-phosphoglycerate**
- **Dehydration** of 2-phosphoglycerate to give **phosphoenolpyruvate**
- **Transfer of a phosphate group** from phosphoenolpyruvate to ADP to give pyruvate

Glycolysis phases

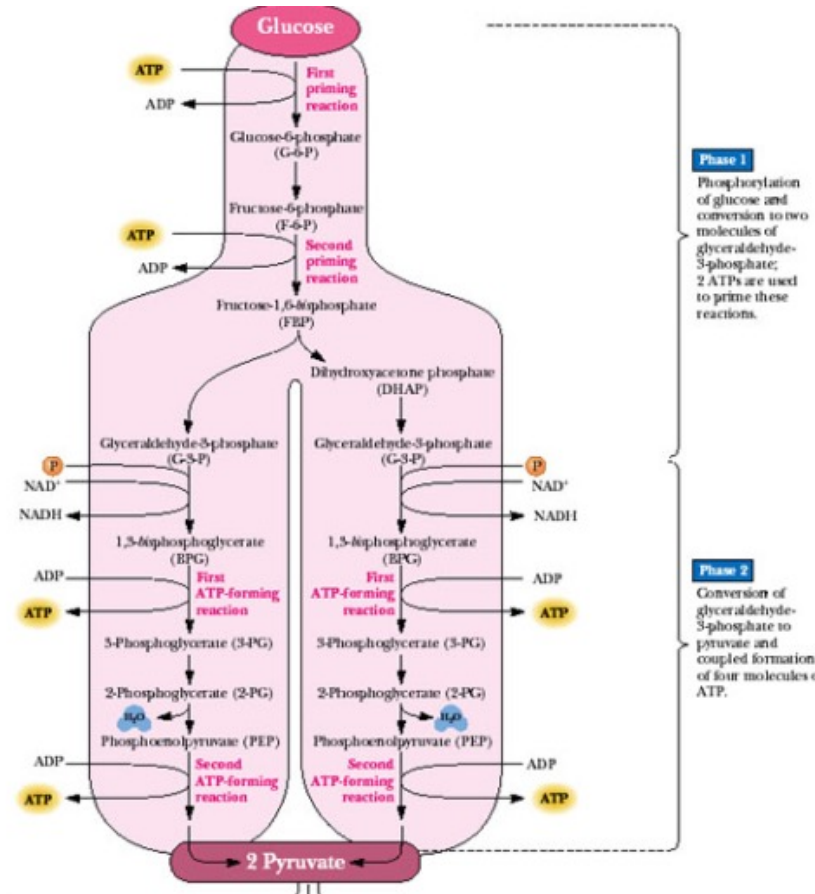
- The pathway of glycolysis can be seen as consisting of 2 separate phases:

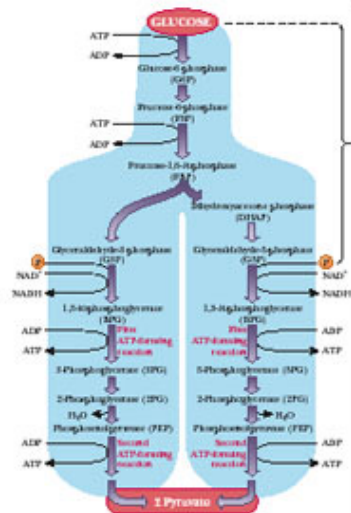
1. The chemical priming phase requiring energy in the form of ATP:

- 2 equivalents of ATP are used to convert glucose to fructose 1,6-bisphosphate (F1,6BP).

2. The energy-yielding phase:

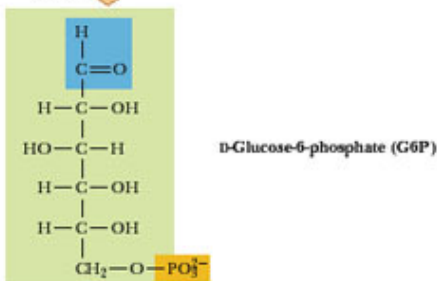
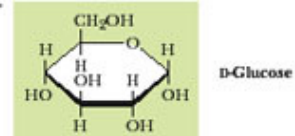
- F1,6BP is degraded to pyruvate, with the production of
- 4 equivalents of ATP and 2 equivalents of NADH.



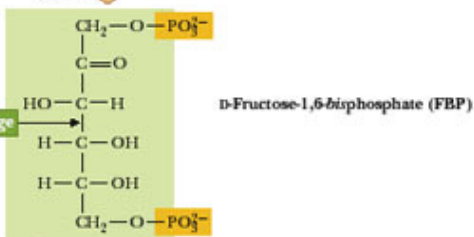
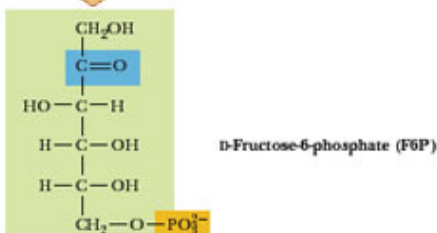


In the first five steps of glycolysis, one 6-carbon molecule of glucose is split into two 3-carbon compounds.

Two molecules of ATP are required for these reactions.

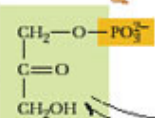


Glucosephosphate isomerase



Aldol cleavage

Aldolase



Triose phosphate isomerase

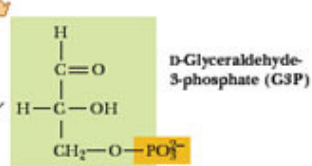
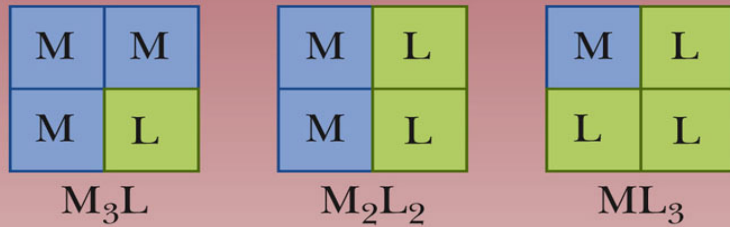


Fig. 17-3, p.468

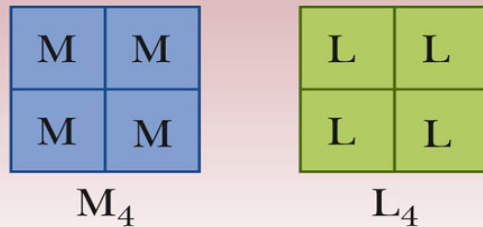
Phosphofructokinase is a key regulatory enzyme

- **Phosphofructokinase (PFK):**
- Exists as a tetramer and subject to allosteric feedback regulation
- The tetramer is composed of L and M subunits
 - M_4 , M_3L , M_2L_2 , ML_3 , and L_4 all exist. Combinations of these subunits are called Isozyme
 - Muscles are rich in M_4 ; the liver is rich in L_4
- ATP is an allosteric effector;
 - high levels inhibit the enzyme
 - low levels activate it
- Fructose-1,6-bisphosphate is also an allosteric effector

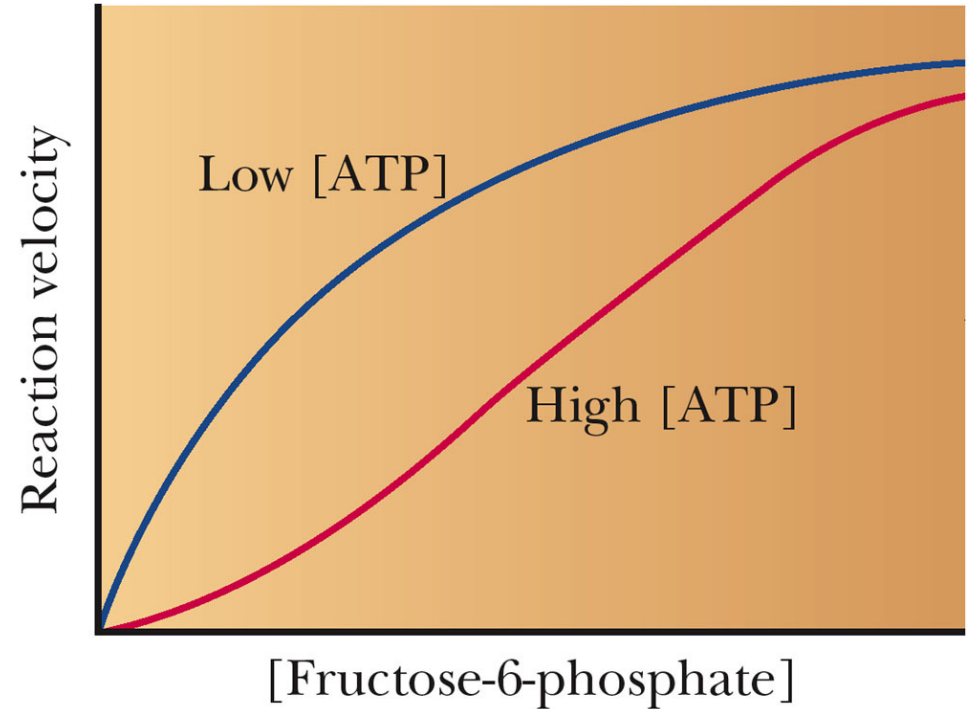
Phosphofructokinase is a key regulatory enzyme



Heterogeneous forms



Homogeneous forms



Summary

- In the first stages of glycolysis, glucose is converted to two molecules of glyceraldehyde-3-phosphate
- The key intermediate in this series of reactions is fructose-1,6-bisphosphate.
- The enzyme that catalyzes this reaction, phosphofructokinase, is subject to allosteric control

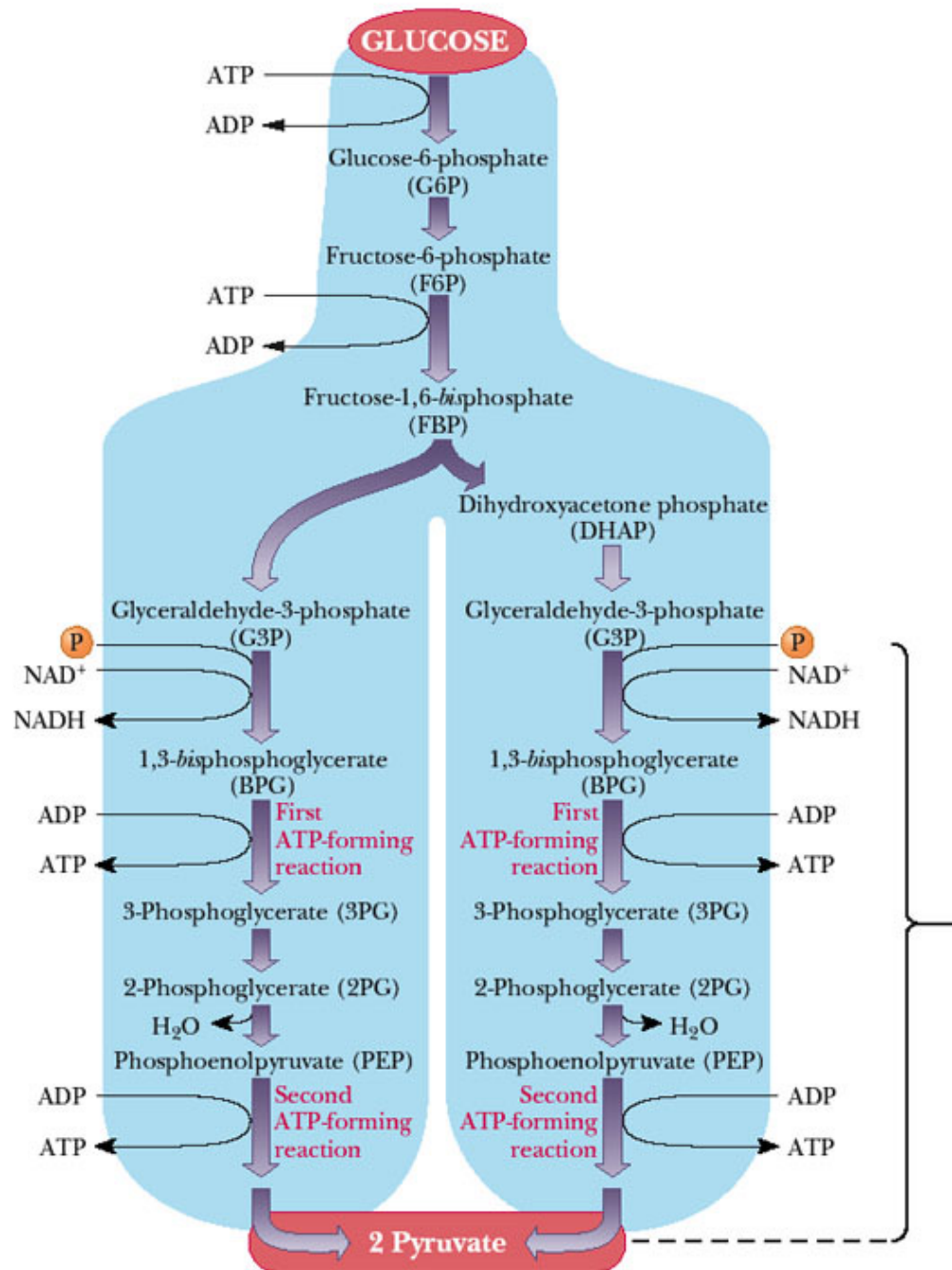


Fig. 17-7a, p.473

In the second phase of glycolysis, glyceraldehyde-3-phosphate is converted to pyruvate.

These reactions yield four molecules of ATP, two for each molecule of pyruvate produced.

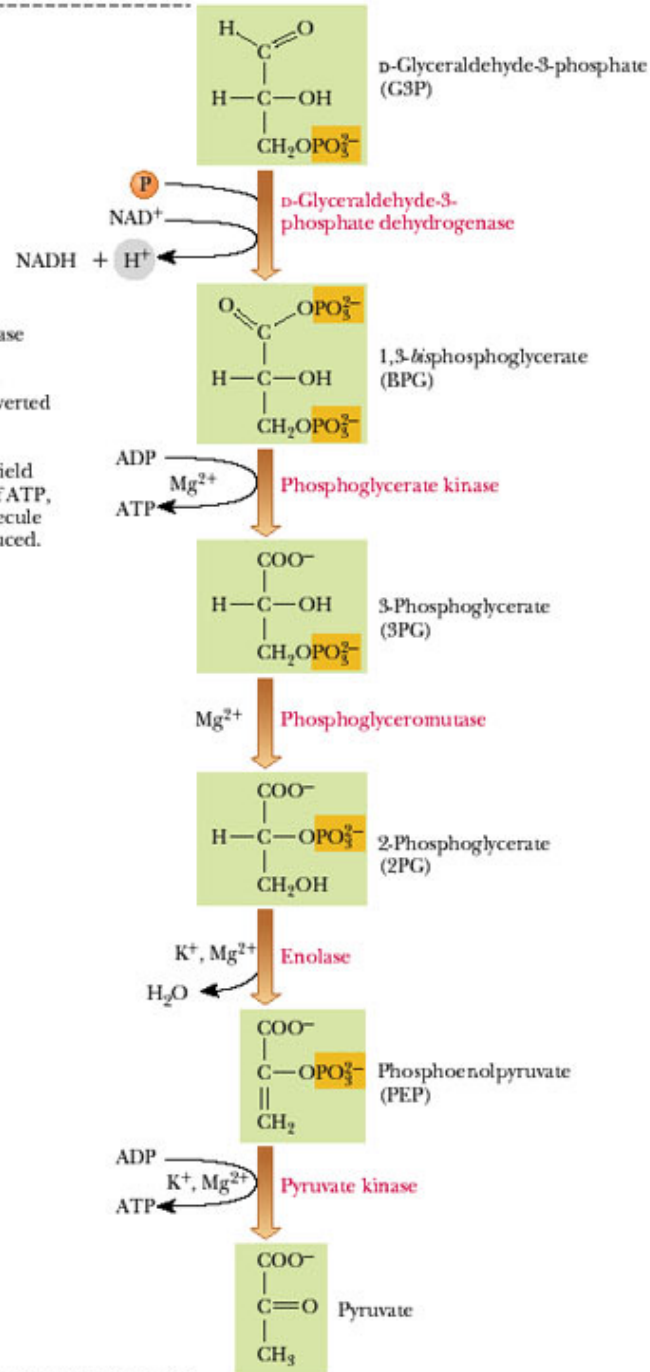


Fig. 17-7b, p.473

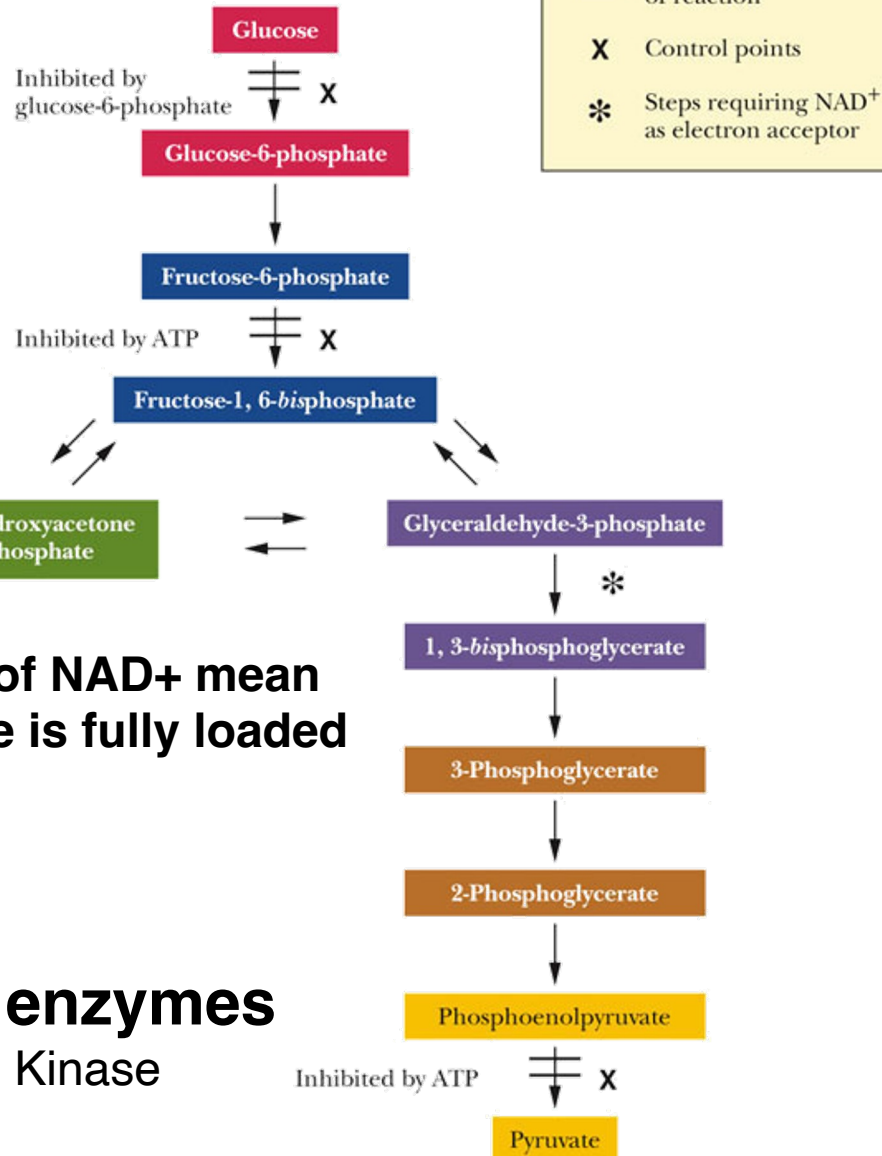
Control Points in Glycolysis

Allosteric enzymes
Hexokinase

Allosteric enzymes
Phosphofructokinase

**Lack of NAD⁺ mean
storage is fully loaded**

Allosteric enzymes
Pyruvate Kinase



Thermodynamics of Glycolysis

Table 15-1

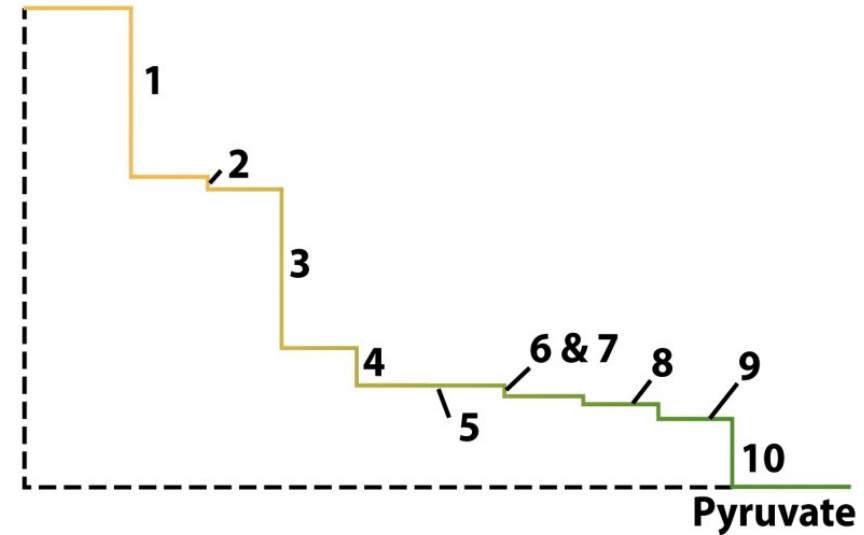
$\Delta G^{\circ'}$ and ΔG for the Reactions of Glycolysis in Heart Muscle^a

Reaction	Enzyme	$\Delta G^{\circ'}$ (kJ · mol ⁻¹)	ΔG (kJ · mol ⁻¹)
1	Hexokinase	-20.9	-27.2
2	PGI	+2.2	-1.4
3	PFK	-17.2	-25.9
4	Aldolase	+22.8	-5.9
5	TIM	+7.9	~0
6 + 7	GAPDH + PGK	-16.7	-1.1
8	PGM	+4.7	-0.6
9	Enolase	-3.2	-2.4
10	PK	-23.0	-13.9

^aCalculated from data in Newsholme, E.A. and Start, C., *Regulation in Metabolism*, p. 97, Wiley (1973).

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Glucose



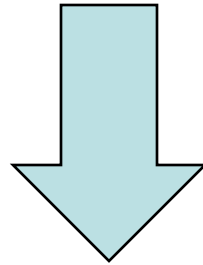
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Summary

- In the final stages of glycolysis, two molecules of pyruvate are produced for each molecule of glucose that entered the pathway
- These reactions involve electron transfer, and the net production of two ATP for each glucose
- There are three control points in the glycolytic pathway

The Energy Derived from Glucose Oxidation

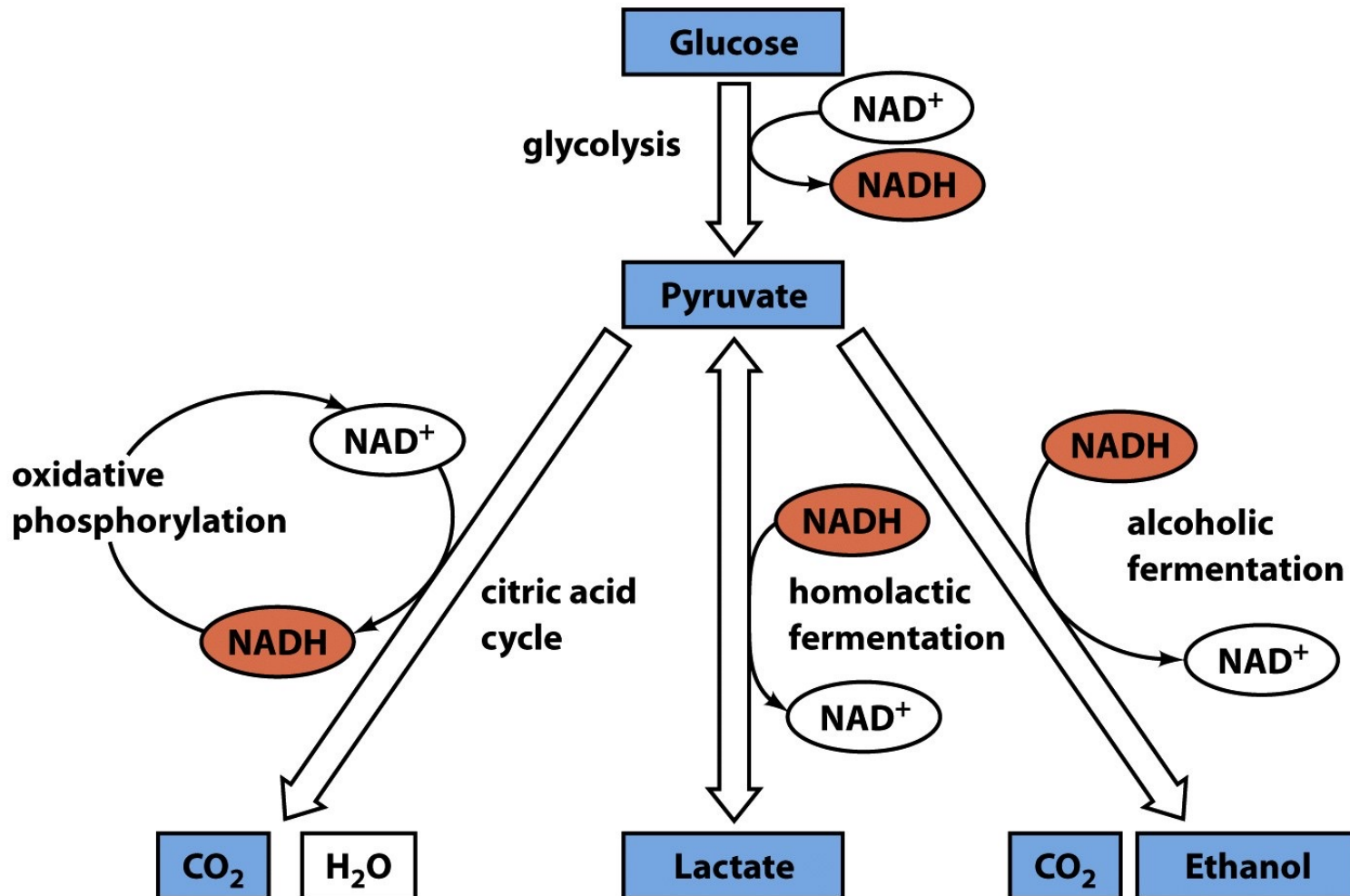
- Aerobic glycolysis of glucose to pyruvate, requires two ATP to activate the process, with production of four ATP and two NADH.



Glycolysis NADH

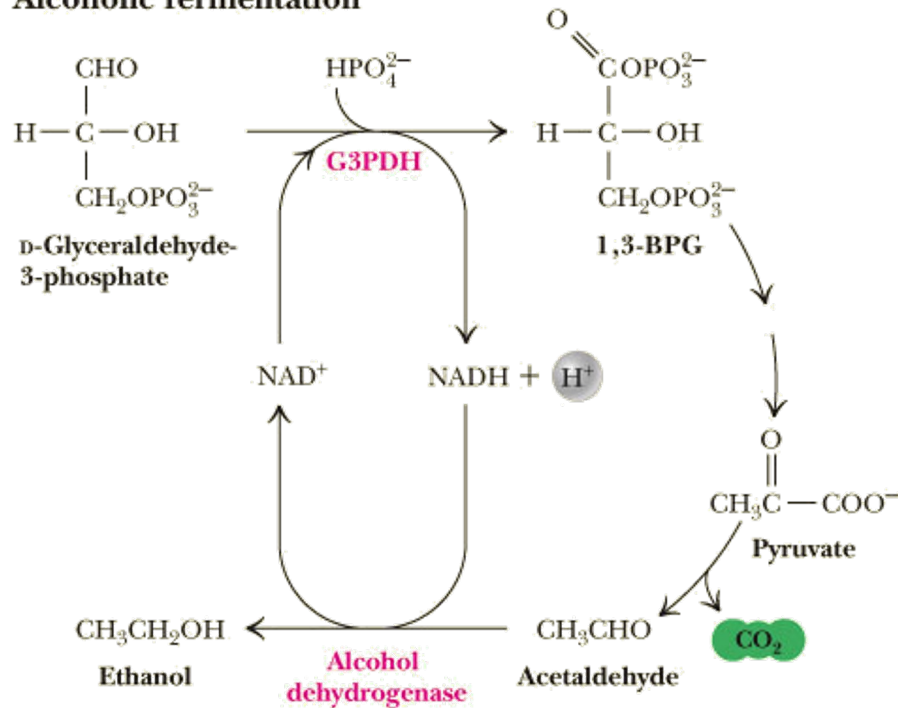
- The NADH generated during glycolysis is used to fuel mitochondrial ATP synthesis via oxidative phosphorylation, producing either two or three equivalents of ATP
- Depending upon whether the glycerol phosphate shuttle or the malate-aspartate shuttle is used to transport the electrons from cytoplasmic NADH into the mitochondria.

Fates of Pyruvate

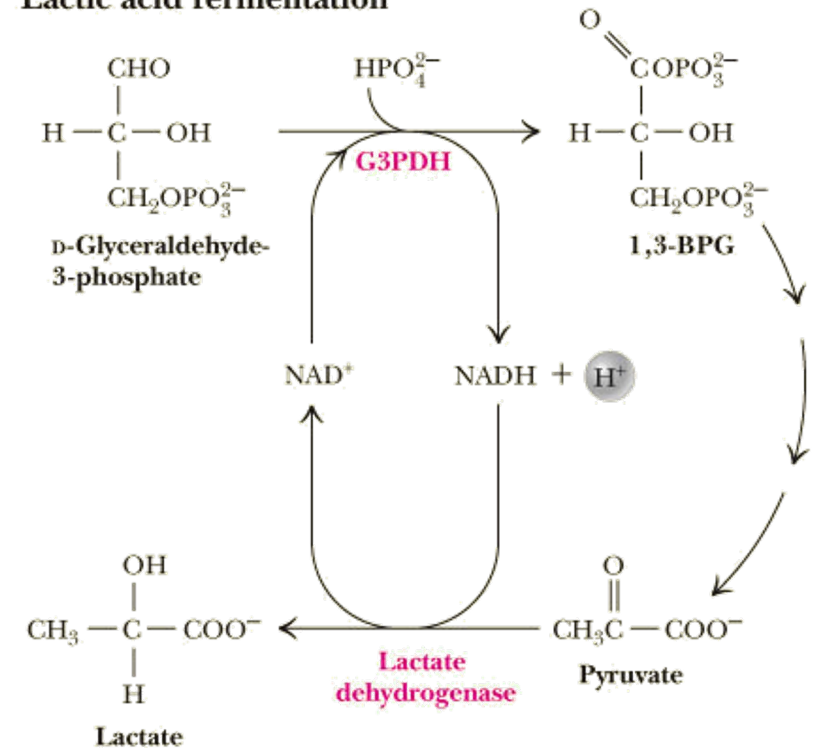


NAD⁺ Needs to be Recycled to Prevent Decrease in Oxidation Reactions

Alcoholic fermentation

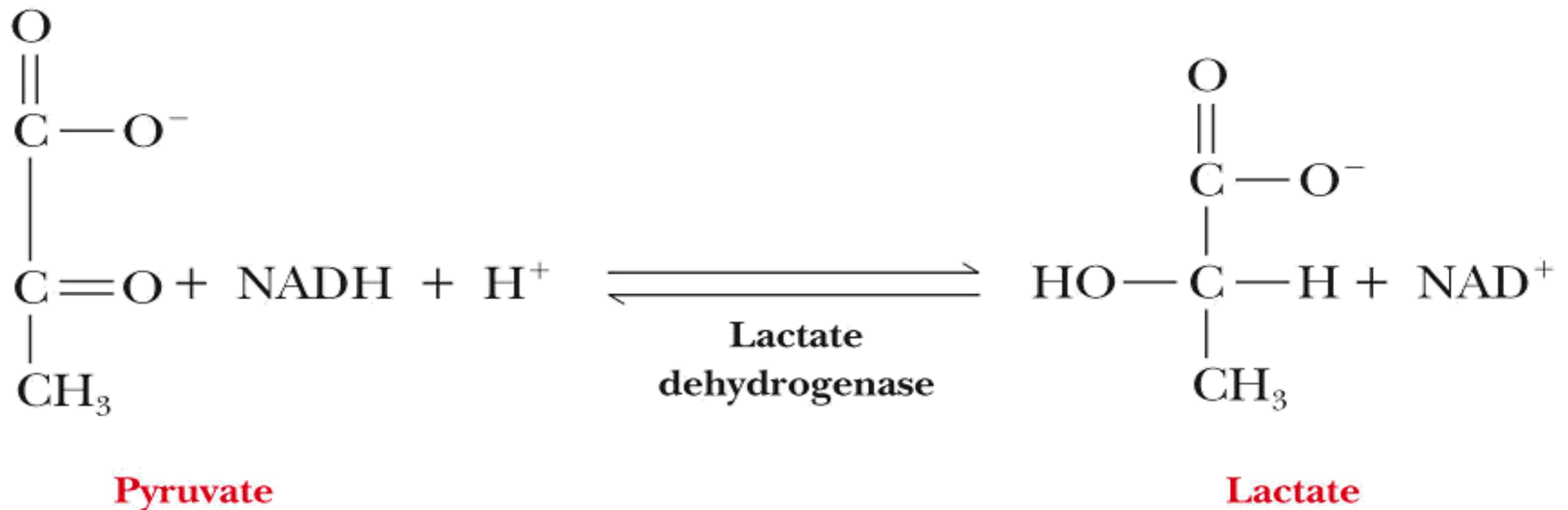


Lactic acid fermentation

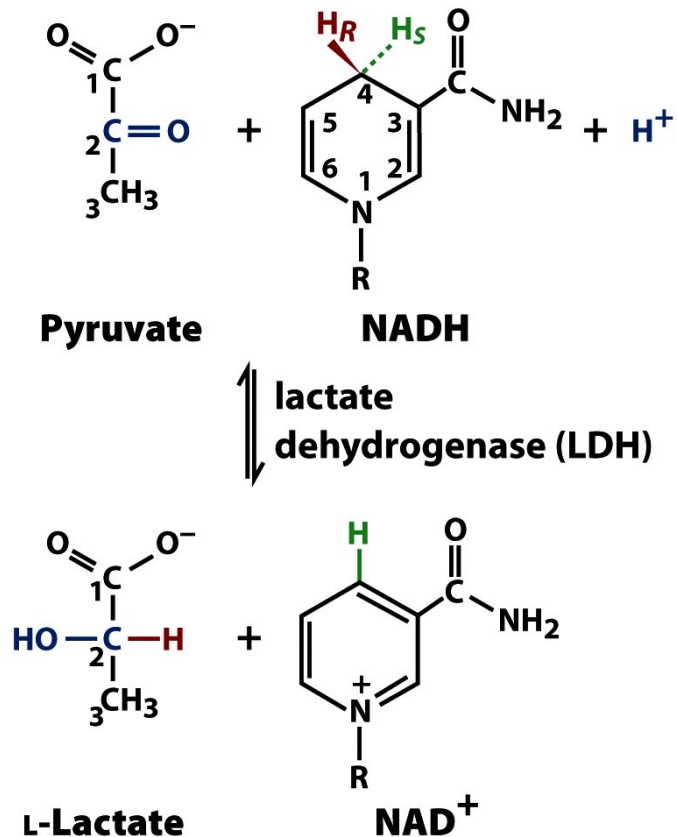


Anaerobic Metabolism of Pyruvate

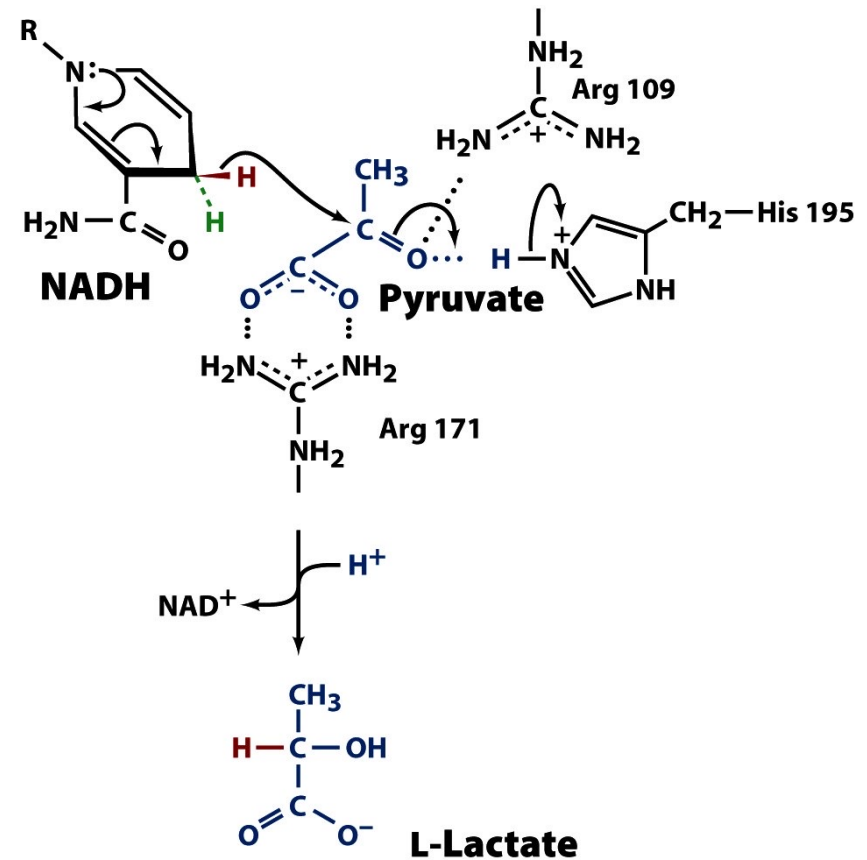
- Under anaerobic conditions, the most important pathway for the regeneration of NAD^+ is reduction of pyruvate to lactic acid
- Lactate dehydrogenase (LDH) is a tetrameric isoenzyme consisting of H and M subunits; H_4 predominates in heart muscle, and M_4 in skeletal muscle



Lactate Dehydrogenase (LDH)



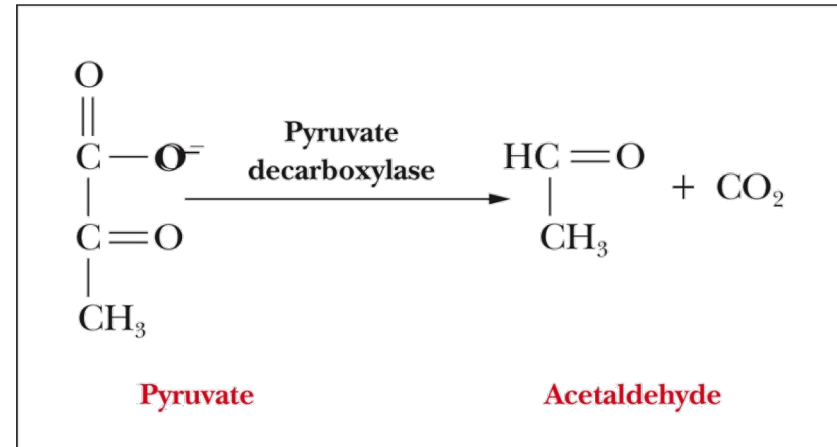
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Alcoholic Fermentation

- Two reactions lead to the production of ethanol:
 - Decarboxylation of pyruvate to acetaldehyde
 - Reduction of acetaldehyde to ethanol
- Pyruvate decarboxylase catalyzes the first reaction
 - This enzyme requires Mg^{2+} and the cofactor Thiamine pyrophosphate (TPP)
- Alcohol dehydrogenase catalyzes the conversion of acetaldehyde to ethanol



Energy Efficiency



To make 2 ATP = $\Delta G^{\circ \prime} = 61 \text{ kJ/mol}$

Therefore, lactate production is $61/196 = 31\%$ efficient

Therefore, ethanol production is $61/235 = 26\%$ efficient

*Under biochemical concentrations, the reactions are actually
>50% efficient*