

## MODELLING: DIFFUSION

### Introduction to Diffusion

=> Diffusion is a flow of particles down a concentration gradient

- ↳ Smoke in a room
- Ink in water
- Redistribution of atoms during a phase transition

### Obtaining the equations governing diffusion

Obtaining the equations is outlined in 3 steps:

- ① Continuity Equation (describes conservation of particles)
- ② Write down the rules for the current in terms of concentration gradient (Fick's first law)
- ③ We combine Fick's first law with the continuity equation to eliminate the current of particles and produce an equation just in terms of concentration (Fick's second law)

### Continuity Equation

- The continuity equation describes a conservation law (for diffusion the no. of particles conserved)
- The equation relates some kind of current to some kind of density

↳ For diffusion: Particle current  
Particle density

### Continuity Equation in 1D for particles

- The difference in rates in and out leads to a change in the total number of particles in the box
- Letting the particle currents flow for a time  $\Delta t$

↳ The change in the number of particles of  $a$  in the box is:

$$\Delta n_a w A = \Delta t (I_{a,in} - I_{a,out}) \quad [1]$$

- It is more convenient to work with particle current densities:  $J_{a,in} = I_{a,in} / A$   
 $J_{a,out} = I_{a,out} / A$

$$\therefore \frac{\Delta n_a}{\Delta t} = \frac{1}{w} (J_{a,in} - J_{a,out}) \quad [2]$$

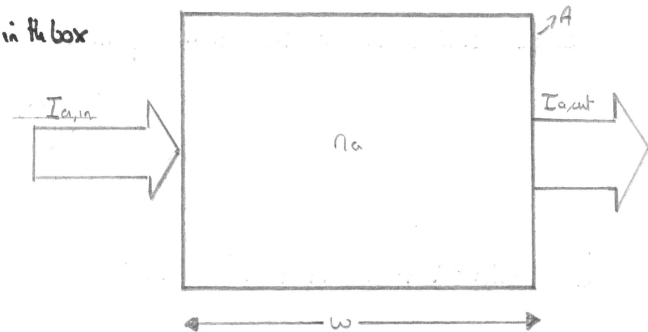
In the limit  $w \rightarrow 0$  and  $\Delta t \rightarrow 0$  we assume the current density is a continuous function of position  $x$ , giving:

$$\frac{\partial n_a}{\partial t} = - \frac{\partial J_a}{\partial x} \quad [3] \quad \text{// Continuity equation}$$

When working with diffusion it is convenient to work with concentrations rather than number densities. Let the total number density of particles in the box be  $p$ . We define the concentration of  $a$  by  $c_a = n_a / p$

∴ We can rewrite [3]:

$$\frac{\partial (p c_a)}{\partial t} = \frac{\partial J_a}{\partial x} \quad [4]$$



$w \Rightarrow$  width

$A \Rightarrow$  cross sectional area

$I_{a,in} \Rightarrow$  flow rate in

$I_{a,out} \Rightarrow$  flow rate out

$n_a \Rightarrow$  number density of  $a$  particles

$p \Rightarrow$  total number density of particles in the box

## MODELLING: Solving Diffusion Equations

The partial differential equation can be tackled in a range of methods and we shall consider one analytical and two numerical ones.

### ANALYTIC SOLUTION

We consider the 1D case:  $\frac{\partial c_a}{\partial t} = D \frac{\partial^2 c_a}{\partial x^2}$  [1]

An easy way to solve this is to reduce the differential equation into an <sup>ODE</sup> algebraic equation by way of the Fourier transform.

$$\frac{\partial c_a}{\partial x} = \frac{1}{2\pi} \int_{-\infty}^{\infty} \tilde{c}_a(g, t) i g e^{i g x} dg$$

$$\frac{\partial^2 c_a}{\partial x^2} = -\frac{1}{2\pi} \int_{-\infty}^{\infty} \tilde{c}_a(g, t) g^2 e^{i g x} dg$$

Substituting back into [1]:  $\frac{1}{2\pi} \int_{-\infty}^{\infty} \left[ \frac{\partial \tilde{c}_a(g, t)}{\partial t} + D g^2 \tilde{c}_a(g, t) \right] e^{i g x} dg = 0$

The term inside the brackets must equal zero:

$$\therefore \frac{\partial \tilde{c}_a(g, t)}{\partial t} = -D g^2 \tilde{c}_a(g, t) \quad [3]$$

Solving gives a general solution:  $\tilde{c}_a(g, t) = e^{-D g^2 t} \tilde{c}_a(g, 0) \quad [4]$

Replacing  $\tilde{c}_a(g, 0)$  with the Fourier transform of  $c_a(x, 0)$

$$\tilde{c}_a(g, 0) = \int_{-\infty}^{\infty} c_a(x, 0) e^{-i g x} dx$$

Taking the inverse F.T. of [4] gives the standard analytical solution:

$$c_a(x, t) = \frac{1}{2\pi} \int_{-\infty}^{\infty} \tilde{c}_a(g, t) e^{i g x} dg$$

$$= \frac{1}{2\pi} \int_{-\infty}^{\infty} e^{-D g^2 t} e^{i g x} \tilde{c}_a(g, 0) dg$$

$$= \frac{1}{2\pi} \int_{-\infty}^{\infty} e^{-D g^2 t} e^{i g x} \int_{-\infty}^{\infty} c_a(x', 0) e^{-i g x'} dx' dg$$

$$= \int_{-\infty}^{\infty} \frac{1}{2\pi} \int_{-\infty}^{\infty} e^{-D g^2 t} e^{-i g(x'-x)} dg c_a(x', 0) dx'$$

$$= \frac{1}{\sqrt{4\pi D t}} \int_{-\infty}^{\infty} e^{-(x-x')^2/4Dt} c_a(x', 0) dx' \quad [5]$$

$$\int_{-\infty}^{\infty} e^{-x g^2 t} e^{-i g x} dg = \sqrt{\frac{\pi}{a}} e^{-x^2/4x}$$

## NUMERICAL METHOD

### Explicit Method

We begin with equation [3] since working with the Fourier transformed concentration allows us to remove derivatives of conc.

We assume we know:  $\tilde{c}_a(y, 0)$

↳ We look to find  $\tilde{c}_a(y, t)$  at later times by stepping forward in time steps of  $\Delta t$

The forward difference approximation is:

$$\frac{\partial \tilde{c}_a(y, t)}{\partial t} \approx \frac{\tilde{c}_a(y, t + \Delta t) - \tilde{c}_a(y, t)}{\Delta t} \quad [9]$$

Substituting [9] into [3] and rearranging gives:

$$\tilde{c}_a(y, t + \Delta t) \approx [1 - Dg^2 \Delta t]^n \tilde{c}_a(y, 0) \quad [10]$$

Moving forward one time step  $\Delta t$  is achieved by multiplying the composition  $\tilde{c}_a(y, t)$  by the constant  $[1 - Dg^2 \Delta t]$

After  $n$  time steps:

$$\tilde{c}(y, n\Delta t) \approx [1 - Dg^2 \Delta t]^n \tilde{c}(y, 0) \quad [11]$$

For such a solution to remain stable we must ensure that it does not diverge.

This is guaranteed, provided that  $|1 - Dg^2 \Delta t| < 1$

∴ This puts an upper limit on the allowed value of the time step:  $\Delta t < \underline{\underline{\frac{2}{Dg^2}}}$

if step size too big solution does not work

The issue arises that if we wish to retain large values of  $y$ , then we must use very small time steps.

### Implicit Method

We can overcome the limitations of the explicit method.

Rewriting [3] as:

$$\frac{\partial \tilde{c}(y, t)}{\partial t} = -Dg^2 \tilde{c}(y, t + \Delta t)$$

Using the finite difference once more:

$$\tilde{c}(y, \Delta t) \approx \frac{\tilde{c}(y, 0)}{1 + Dg^2 \Delta t}$$

the larger  $y$  the smaller  $\Delta t$  must be

Since  $1/(1 + Dg^2 \Delta t) < 1 \forall \text{ } \Delta t$  The method is stable.  
(Not necessarily accurate)

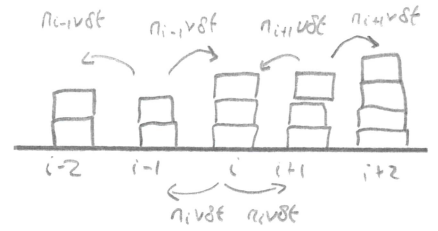
### MODELLING: Kinetic monte carlo to model diffusion

Consider a long line of bins, within each are a number of disks.

Disks can hop into neighbouring bins with the probability per disc per unit time being  $\nu$

If the no. of disks in bin  $i$  is  $n_i$ , then in a time period  $\delta t$  the average number of disks transferred from bin  $i$  to bin  $i+1$  is  $n_i \nu \delta t$ .

The same is true for the average number of disks that will be transferred from bin  $i$  to bin  $i-1$



Considering all additions and subtractions:

$$n_i(t + \delta t) = n_i(t) + n_{i-1} \nu \delta t + n_{i+1} \nu \delta t - 2n_i \nu \delta t \quad [1]$$

Rearranging gives:

$$\frac{n_i(t + \delta t) - n_i(t)}{\delta t} = \nu (n_{i+1} - 2n_i + n_{i-1}) \quad [2]$$

In the limit  $\delta t \rightarrow 0$

$$\frac{\partial n_i}{\partial t} = \nu (n_{i+1} - 2n_i + n_{i-1}) \quad [3]$$

Letting the position of bin  $i$  be  $x_i = ia$ , where  $a$  is the distance between bins, we introduce a function of position  $n(x)$ , satisfying  $n(x_i) = n_i$ .

Substituting  $n(x)$  gives:

$$\frac{\partial n(x_i)}{\partial t} = \nu a^2 \frac{(n(x_i + a) - 2n(x_i) + n(x_i - a)))}{a^2} \quad [4]$$

If  $n$  varies sufficiently slowly with  $x$  we can approximate it by a second order Taylor Expansion

[4] becomes:

$$\frac{\partial n(x_i)}{\partial t} \approx \nu a^2 \frac{\partial^2 n(x_i)}{\partial x^2}$$

// This is the diffusion equation  $\nu a^2 = D$

$\therefore$  We conclude kinetic processes like diffusion can be represented by a process of random hops with known hopping rates.  
The material parameter (diffusion) being related to the hopping rate.