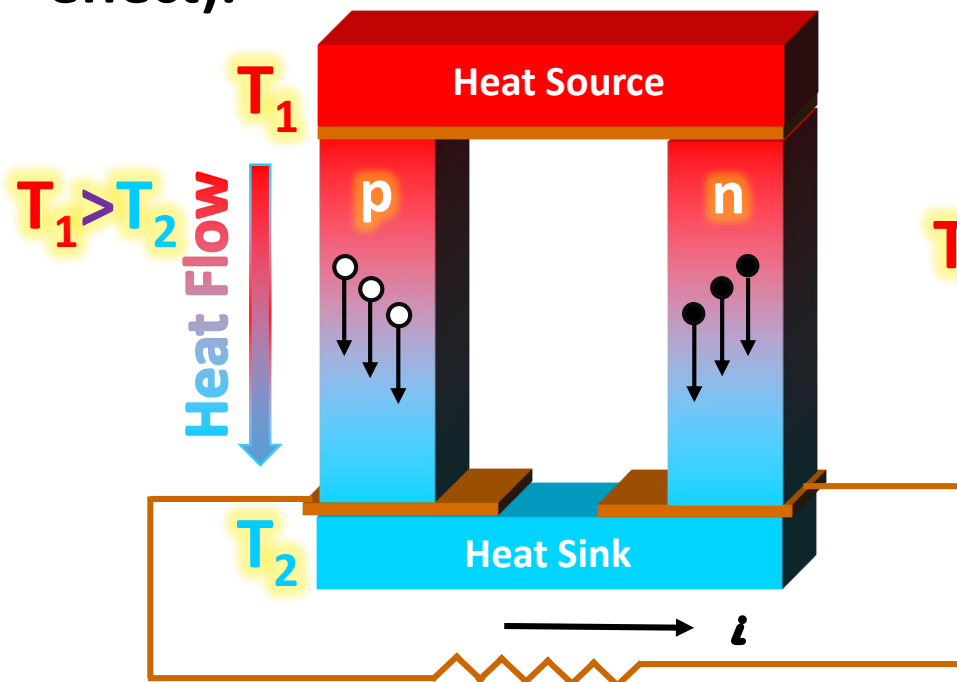


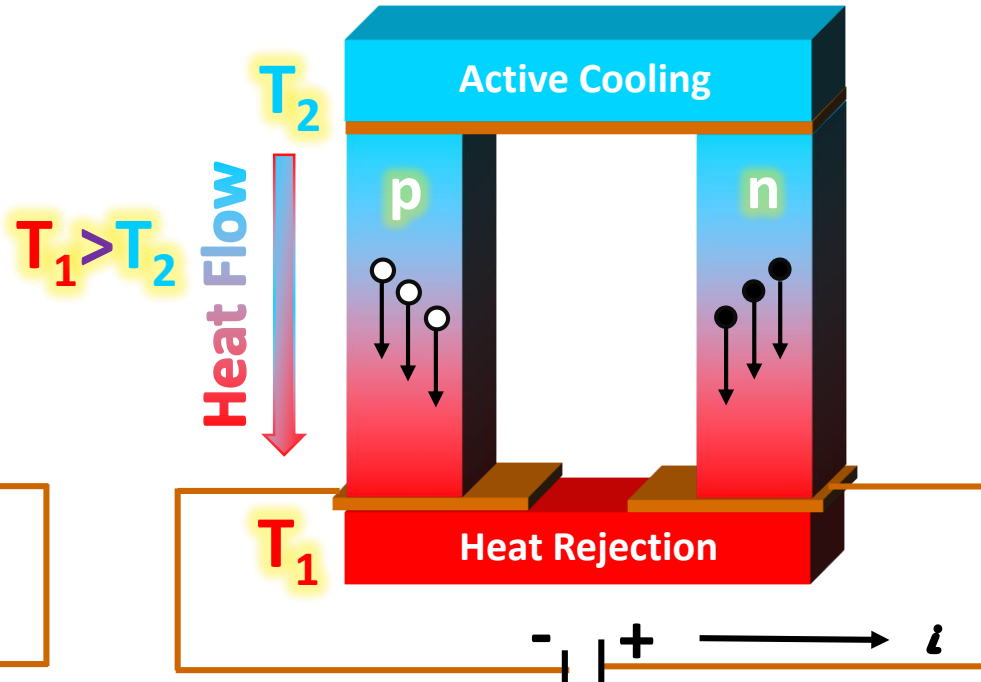
# Computational Exploration of Ultralow Lattice Thermal Conductivity and High Figure of Merit in p-type Bulk $\text{RbX}_2\text{Sb}$ (X=K, Na)

## What are thermoelectric materials?

- Materials that generate electricity due to temperature difference (Seebeck effect) or flow of electricity through which result temperature difference (Peltier effect).



**Seebeck Effect**  
“Electricity from waste-heat”



**Peltier Effect**  
“Refrigeration without any moving part”

## What are thermoelectric materials? (contd.)

- Electricity flows due to voltage difference and voltage difference comes from the difference in electrochemical potential.
- In thermoelectric materials the voltage difference comes from the temperature difference (for Seebeck effect).

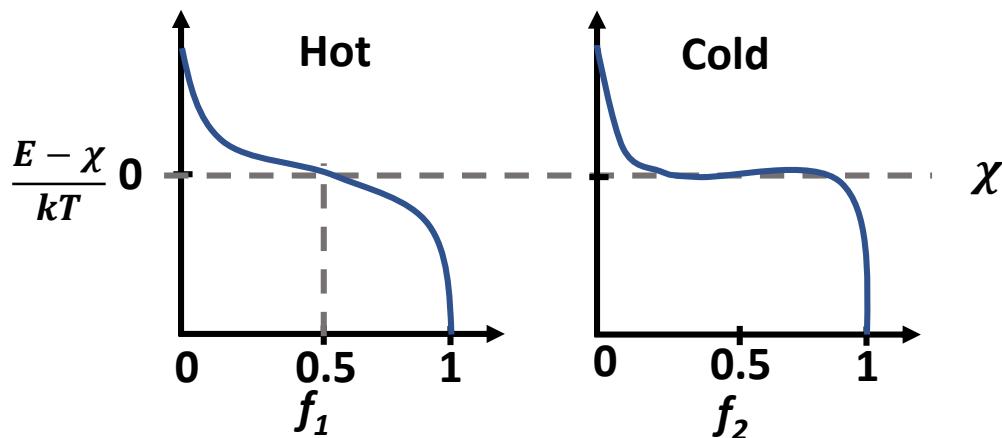
$$I = \frac{1}{q} \int_{-\infty}^{+\infty} dE G(E) (f_1 - f_2)$$

↑  
Current flow

↑  
Conductance

$$f_i = \frac{1}{1 + e^{(E - \chi_i)/kT_i}}$$

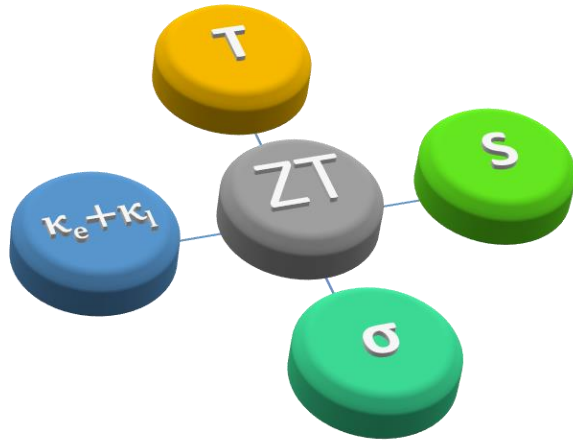
↑ Fermi function      ↑ Electrochemical potential



## Why we are interested in thermoelectric materials?

- Major cause of global warming is the waste heat.
- Almost 75% of the energy produced by the mankind are squandered as waste heat.
- The waste heat can be used as the renewable energy source for the generation of electricity by the aid of the Seebeck effect of thermoelectric materials.

## Figure of merits



Seebeck Coefficient    Electrical conductivity

$$ZT = \frac{S^2 \sigma T}{\kappa_e + \kappa_l}$$

Figure of merit

↓    ↓

← Absolute Temperature

← Lattice Thermal Conductivity

← Electrical Thermal Conductivity

$$S = \frac{\Delta V}{\Delta T}$$

← ability to generate voltage difference due to unit temperature difference

## Interdependencies of the parameters

$$S(T, \chi) = \frac{\int_{-\infty}^{\infty} dE \left( -\frac{\partial f(E, T, \chi)}{\partial E} \right) \sigma(E) (E - \chi)}{qT \int_{-\infty}^{\infty} dE \left( -\frac{\partial f(E, T, \chi)}{\partial E} \right) \sigma(E)} \quad \dots(1)$$

$$\sigma(T, \chi) = \int_{-\infty}^{\infty} dE \left( -\frac{\partial f(E, T, \chi)}{\partial E} \right) \sigma(E) \quad \dots(2)$$

$$\kappa_e = \kappa_0 - T\sigma S^2 = L\sigma T \quad \dots(3)$$

Energy Electrochemical potential

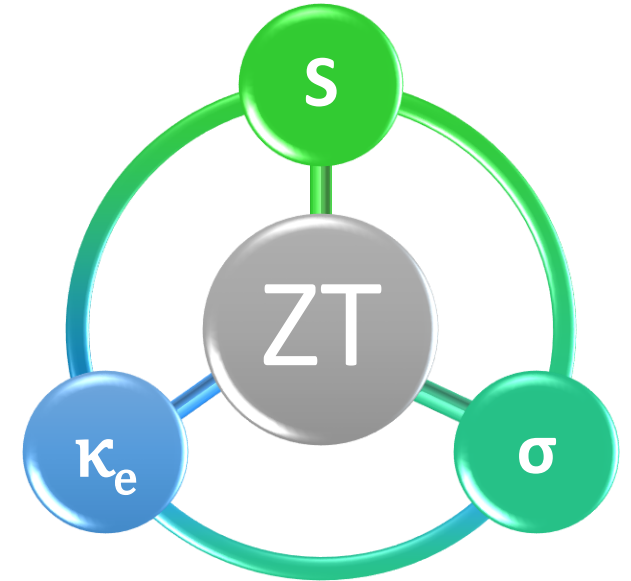


Lorenz Constant



$f(E, T, \chi)$  ← Fermi function.

$\kappa_0$  ← Electrical thermal conductivity under closed circuit condition.



## Contradictory relations between the parameters

$$S = \frac{2k_B^2 T}{3e\hbar^2} \left(\frac{\pi}{3n}\right)^{2/3} m_d^*, \quad \text{where } m_d^* \propto m_b^* = \frac{\hbar^2}{\frac{d^2 E_{edge}}{dk^2}}$$

Charge carrier concentration      DOS effective mass      band effective mass

$$\sigma = \frac{ne^2\tau}{m_b^*} = ne\mu, \quad \text{where } \tau = \frac{m_b^*\mu}{e} \leftarrow \text{Charge carrier mobility}$$

Charge carrier relaxation time

$$\kappa_e = \kappa_0 - T\sigma S^2 = L\sigma T$$

$$ZT = \frac{S^2 \sigma T}{\kappa_e + \kappa_l}$$



**Metal**  
High electrical  
conductivity

+



**Semiconductor**  
High Seebeck  
coefficient

+



**Glass**  
Low thermal  
conductivity

=>

**Good thermoelectric  
material**



# Our present work with bi-alkali antimonides

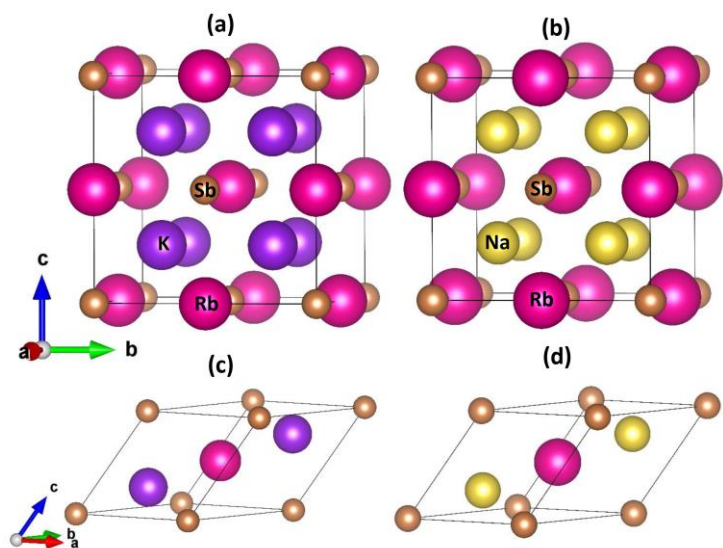


Figure 1: Conventional unit cells (a), (b) and primitive unit cells (c), (d) of  $\text{RbK}_2\text{Sb}$  and  $\text{RbNa}_2\text{Sb}$ , respectively

- ❖ Cubic crystal structure with space group  $\text{Fm-3m}$  (No. 225)
- ❖ 16 atoms per conventional unit cell
- ❖ 4 atoms per primitive unit cell

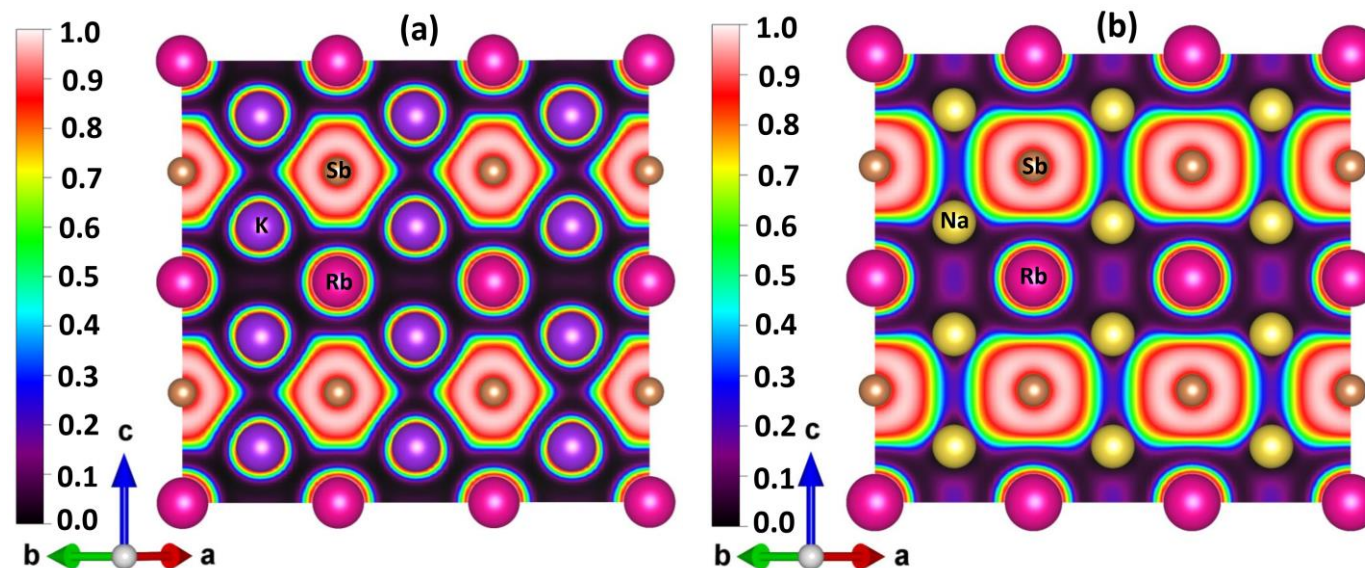


Figure 2: Electron localization function plot along the (110) plane of (a)  $\text{RbK}_2\text{Sb}$  and (b)  $\text{RbNa}_2\text{Sb}$ .

From Figure 2:

- ❖ Negligible ELF values between the atom pairs for both systems
- ❖ Electrons are tightly bound to the atoms resulting in very poor overlaps between the atom pairs
- ❖ Represent ionic characteristic of the bonds in both the systems



## Stability of the systems

- ❖ Negative formation energy (-0.41 and -0.44 eV/atom for RbK<sub>2</sub>Sb and RbNa<sub>2</sub>Sb, respectively): Thermodynamically stable
- ❖ Thermally stable from ab Initio molecular dynamics simulation at 700 K

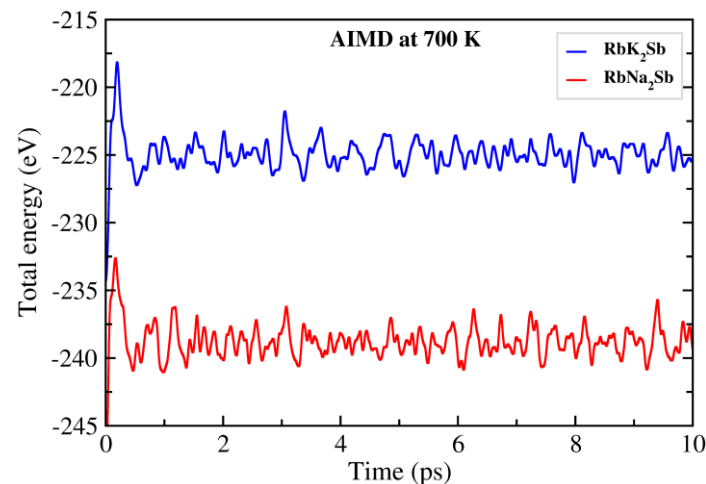


Figure 3: AIMD plot at 700 K up to 10 ps

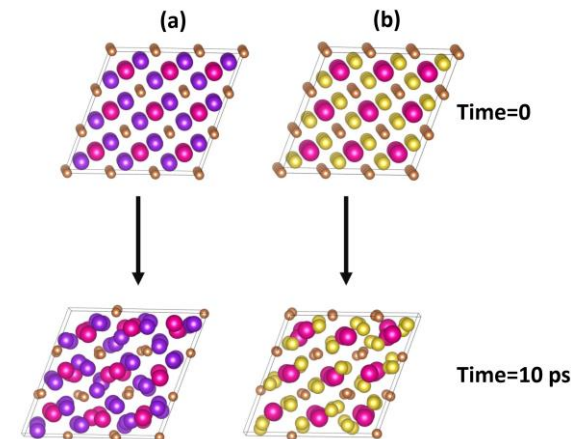


Figure 4: Structures of (a) RbK<sub>2</sub>Sb and (b) RbNa<sub>2</sub>Sb after 10 ps of AIMD simulation.

- ❖ Satisfies the Born stability criteria in terms of elastic stiffness constants: Mechanically stable
- ❑ Born criteria for cubic systems:  $C_{11}-C_{12}>0$ ,  $C_{11}+2C_{12}>0$  and  $C_{44}>0$

## Stability of the systems (contd.)

### ❖ Non-negative phonon frequencies: Dynamically stable

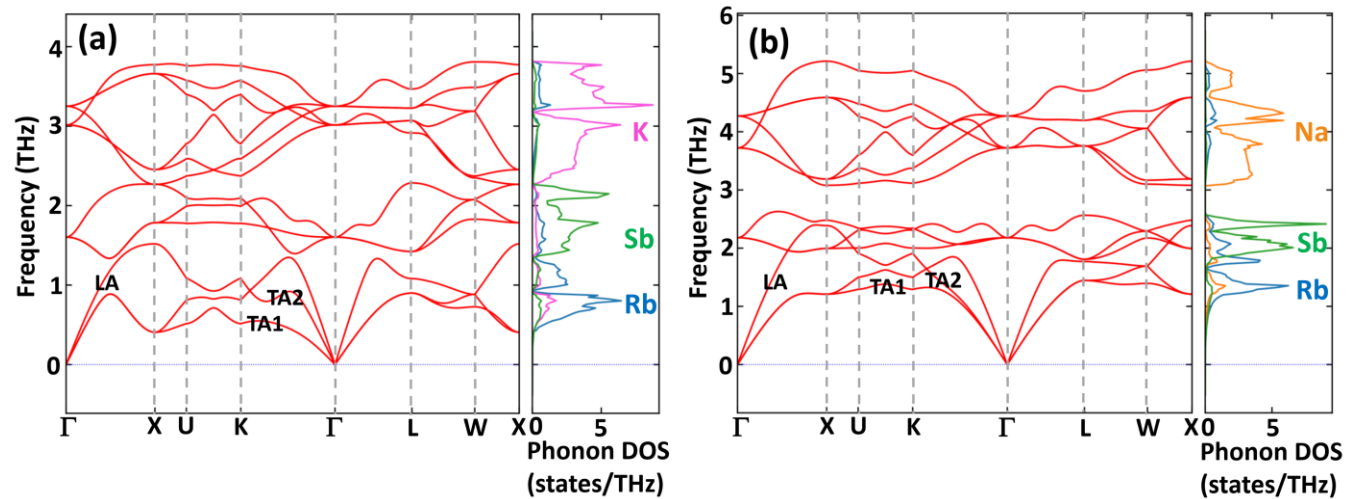


Figure 5: Phonon dispersion and phonon density of state of (a)  $\text{RbK}_2\text{Sb}$ , (b)  $\text{RbNa}_2\text{Sb}$

- 4 atoms per primitive unit cell
- $3 \times 4 = 12$  degrees of freedom: 12 phonon modes = 3 acoustic (2TA+1LA)+9 optical (3 low frequency + 6 high frequency)
- $M_K > M_{Na}$ :  $\omega(\text{RbK}_2\text{Sb}) < \omega(\text{RbNa}_2\text{Sb})$
- Avoided crossing in  $\text{RbK}_2\text{Sb}$  phonon dispersion.
- Flattening of the acoustic bands of the  $\text{RbK}_2\text{Sb}$
- In  $\text{RbNa}_2\text{Sb}$ : low frequency optical and acoustic bands overlap

# Electronic structure

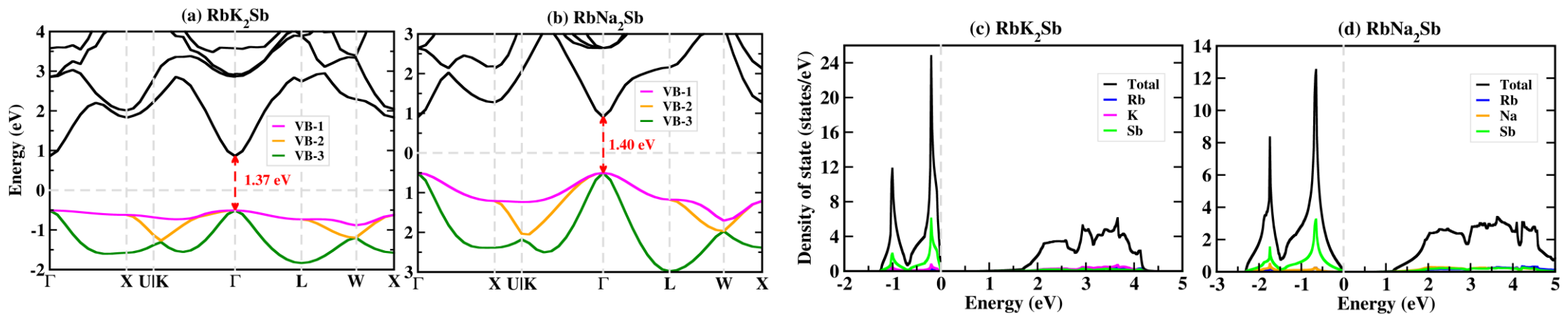


Figure 6: Electronic band structure using HSE06 functional: (a) RbK<sub>2</sub>Sb, (b) RbNa<sub>2</sub>Sb and electronic DOS: (c) RbK<sub>2</sub>Sb, (d) RbNa<sub>2</sub>Sb

VB-1			VB-2			VB-3			CB				
Systems	$\Gamma \rightarrow X$	$\Gamma \rightarrow K$	$\Gamma \rightarrow L$	$\Gamma \rightarrow X$	$\Gamma \rightarrow K$	$\Gamma \rightarrow L$	$\Gamma \rightarrow X$	$\Gamma \rightarrow K$	$\Gamma \rightarrow L$	Systems	$\Gamma \rightarrow X$	$\Gamma \rightarrow K$	$\Gamma \rightarrow L$
RbK <sub>2</sub> Sb	-9.830	-8.780	-2.442	-9.830	-1.826	-2.442	-0.156	-0.169	-0.161	RbK <sub>2</sub> Sb	0.136	0.136	0.128
RbNa <sub>2</sub> Sb	-0.856	-0.907	-0.877	-0.856	-0.830	-0.877	-0.112	-0.114	-0.104	RbNa <sub>2</sub> Sb	0.113	0.117	0.107

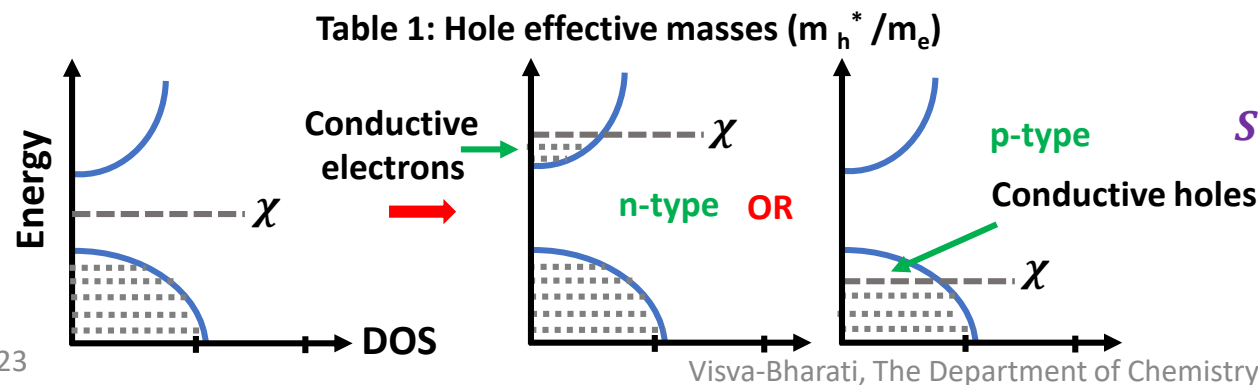


Table 2: Electron effective masses ( $m_e^*/m_e$ )

$$S = \frac{2k_B^2 T}{3e\hbar^2} \left(\frac{\pi}{3n}\right)^{2/3} m_d^*, \quad \text{where } m_d^* \propto m_b^* = \frac{\hbar^2}{\frac{d^2 E_{edge}}{dk^2}}$$

# Thermoelectric properties

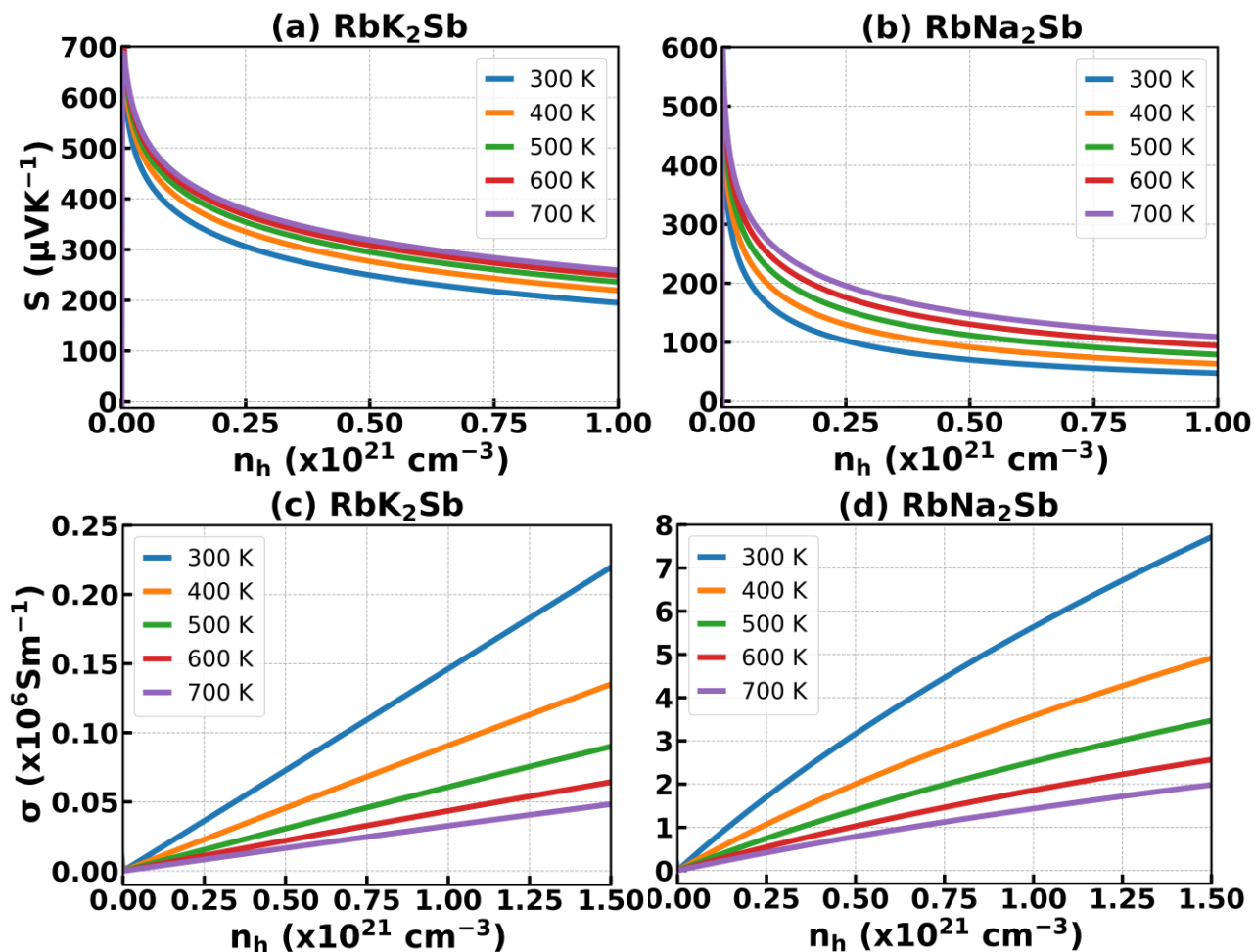


Figure 7: Seebeck coefficient of p-type (a)  $\text{RbK}_2\text{Sb}$ , (b)  $\text{RbNa}_2\text{Sb}$  and electrical conductivity of p-type (c)  $\text{RbK}_2\text{Sb}$ , (d)  $\text{RbNa}_2\text{Sb}$  as a function of hole concentration

	RbK <sub>2</sub> Sb		RbNa <sub>2</sub> Sb	
Temperature (K)	$\mu$ ( $\text{cm}^2\text{V}^{-1}\text{S}^{-1}$ )	$\tau$ (fs)	$\mu$ ( $\text{cm}^2\text{V}^{-1}\text{S}^{-1}$ )	$\tau$ (fs)
300	2.28	12.8	451.38	220
400	1.48	8.3	291.35	142
500	1.05	5.9	209.28	102
600	0.80	4.5	157.98	77
700	0.64	3.6	125.15	61

Table 3: Hole mobilities and relaxation times

$$S = \frac{2k_B^2 T}{3e\hbar^2} \left(\frac{\pi}{3n}\right)^{2/3} m_d^*, \quad \text{where } m_d^* \propto m_b^*$$

$$\sigma = \frac{ne^2\tau}{m_b^*} = \frac{ne^2\mu}{e}, \quad \text{where } \tau = \frac{m_b^*\mu}{e}$$

$$\kappa_e \propto \sigma$$

# Thermoelectric properties (contd.)

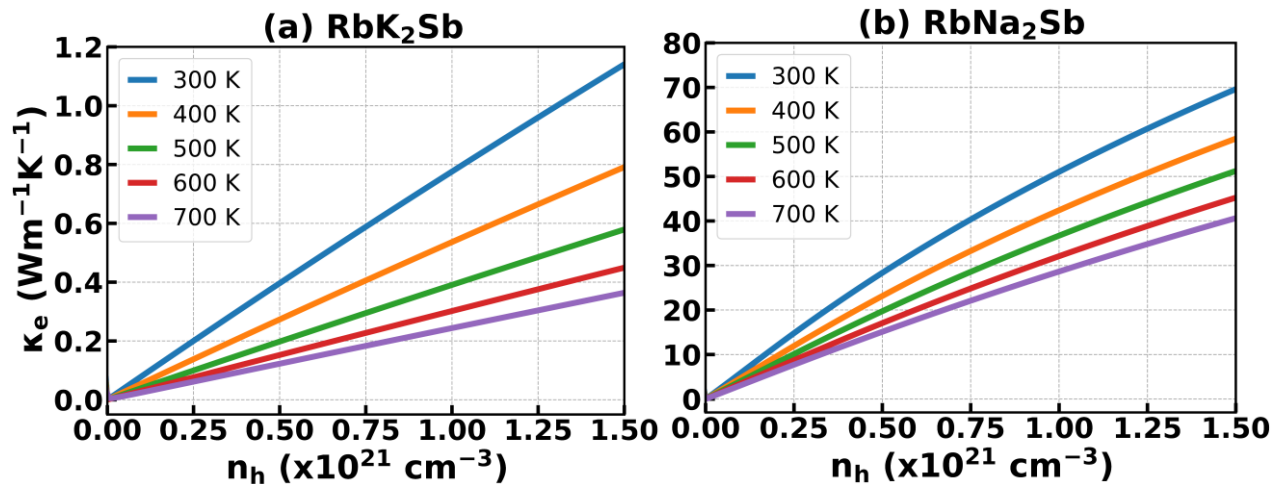


Figure 8: Electronic thermal conductivity of p-type (a) RbK<sub>2</sub>Sb, (b) RbNa<sub>2</sub>Sb as function of hole concentration

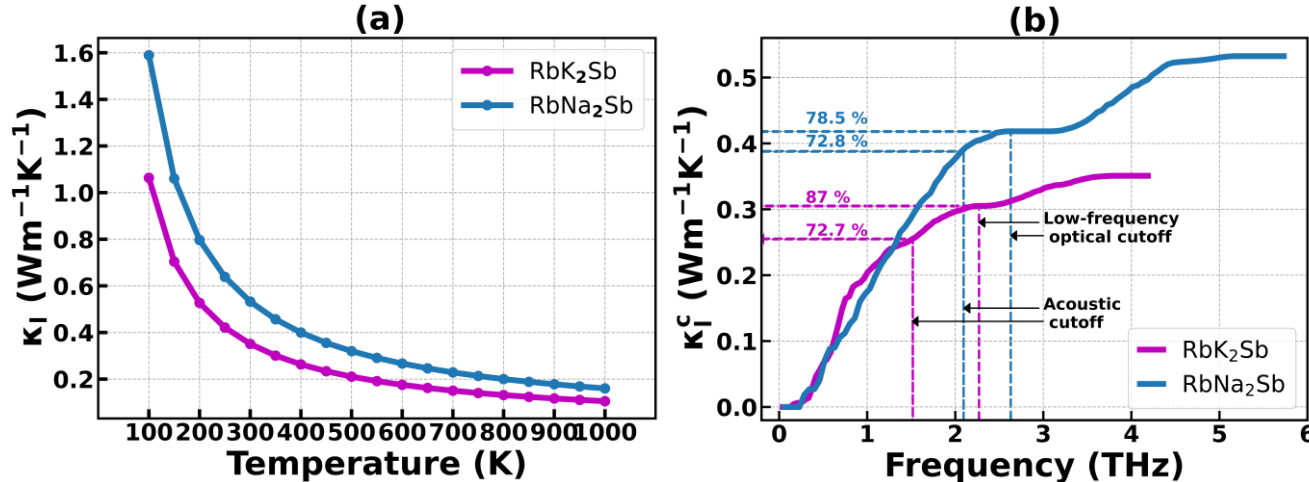


Figure 9: (a) Lattice thermal conductivity as function of temperature (b) Cumulative lattice thermal conductivity as function of frequency

$$ZT = \frac{S^2 \sigma T}{\kappa_e + \kappa_l}$$

$\kappa_e \propto \sigma$   
 Lattice thermal conductivity  
 $\kappa_l(T) = \frac{1}{3} \langle \tau \rangle \langle v_g^2 \rangle C_v(T)$   
 Average relaxation time      Mean square group velocity      Specific heat at constant volume

Systems	300 K	400 K	500 K	600 K	700 K
RbK <sub>2</sub> Sb	0.35087	0.26307	0.21044	0.17536	0.15031
RbNa <sub>2</sub> Sb	0.53243	0.39974	0.31997	0.26673	0.22868

Table 5: Lattice thermal conductivities at different temperatures

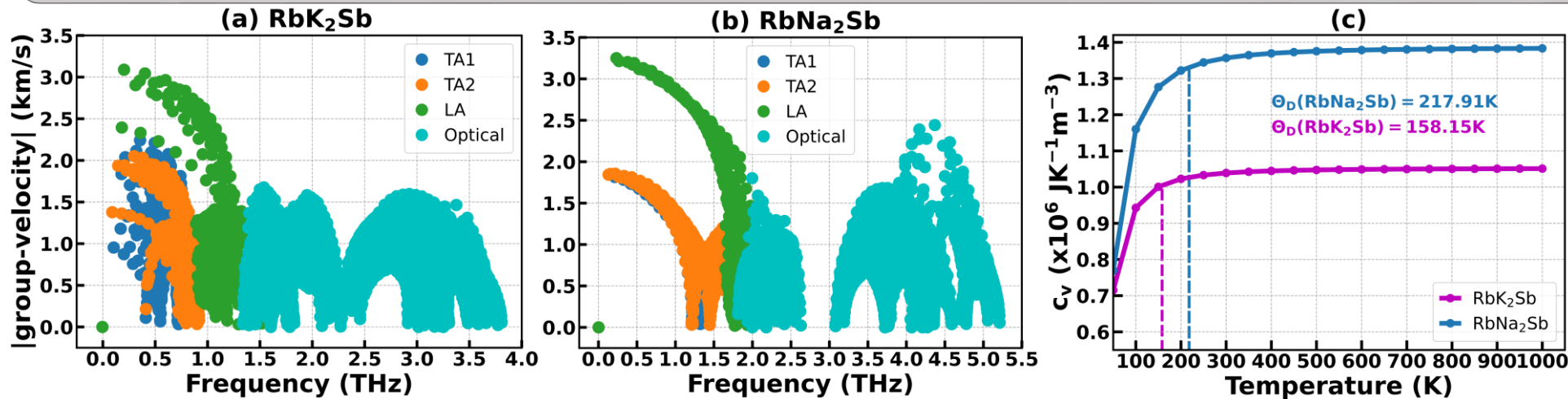
Cumulative lattice thermal conductivity

$$\kappa_l^c(\omega) = \int_0^\omega \frac{1}{N} \sum_s \kappa_{l,s} \delta(\omega_s - \omega') d\omega'$$

Mode specific phonon frequency  
 Contribution to lattice thermal conductivity from phonon mode s



# Thermoelectric properties (contd.)



Debye temperature is the Temperature of the crystal's Highest normal vibrational mode, i.e., above  $\theta_D$  all the phonons get excited.

Figure 10: Mode specific group velocity as a function of phonon frequency for (a)  $\text{RbK}_2\text{Sb}$ , (b)  $\text{RbNa}_2\text{Sb}$ . (c) Specific heat at constant volume as a function of temperature

Systems	$v_{g,LA}(\text{km/s})$	$v_{g,TA1}(\text{km/s})$	$v_{g,TA2}(\text{km/s})$
$\text{RbK}_2\text{Sb}$	3.08	2.24	2.05
$\text{RbNa}_2\text{Sb}$	3.25	1.85	1.86

Table 6: Maximum mode specific group velocities at the vicinity of the  $\Gamma$ -point

Comparable and even lower than many TE materials reported earlier like 2D  $\alpha$ -Te ( $>4 \text{ km/s}$ ),  $\text{Ti}_{3-x}\text{Mo}_x\text{C}_2\text{T}_2$  ( $x=0.5, 1, 1.5, 2.5, 2.5$ , T=OH/-O/-F) ( $>6 \text{ km/s}$ ), 2D BP/MoS2 vdW hetero-bilayer ( $>12 \text{ km/s}$ ) etc.

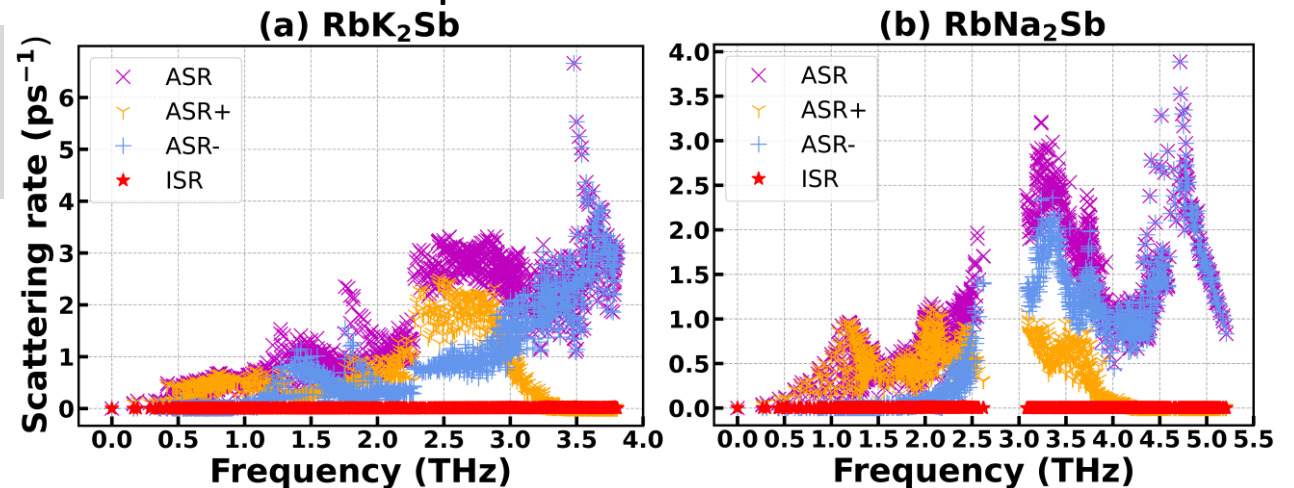


Figure 11: Scattering rates as functions of phonon frequency for (a)  $\text{RbK}_2\text{Sb}$ , (b)  $\text{RbNa}_2\text{Sb}$

## Thermoelectric properties (contd.)

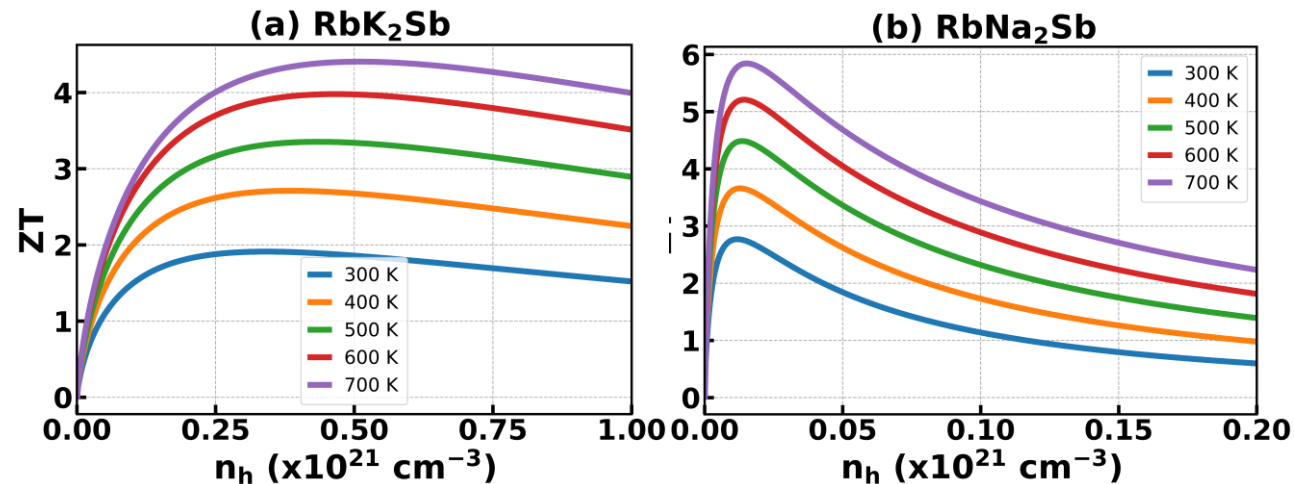


Figure 12:  $ZT$  as function of hole concentration at the temperature range 300-700 K for (a) RbK<sub>2</sub>Sb, (b) RbNa<sub>2</sub>Sb

Systems	300 K	400 K	500 K	600 K	700 K
RbK <sub>2</sub> Sb	1.91	2.71	3.35	3.97	4.40
RbNa <sub>2</sub> Sb	2.76	3.65	4.48	5.20	5.84

Table 7:  $(ZT)_{\text{max}}$  for p-type RbK<sub>2</sub>Sb and RbNa<sub>2</sub>Sb at the temperature range 300-700 K

**$(ZT)_{\text{max}}$  for other full-Heusler compounds reported earlier:**

- n- and p-type Ba<sub>2</sub>AuBi shows  $ZT$  as high as 5 and 2 at 800 K [1]
- Predicted  $ZT$  for n-type Sr<sub>2</sub>BiAu is 4.9 at 700 K [2]
- Predicted  $ZT$  for n-type Sr<sub>2</sub>SbAu is 3.4 at 750 K [2]
- $ZT$  for n-type Sr<sub>2</sub>AsAu is 3.3 at 700 K [3]

**Conclusion:**

The bi-alkali antimonides, we have studied can be considered as better thermoelectric materials as compared to the other full-Heusler compounds



## References and acknowledgments

### References:

- [1] Park, J.; Xia, Y.; Ozoliņš, V. High Thermoelectric Power Factor and Efficiency from a Highly Dispersive Band in  $\text{Ba}_2\text{BiAu}$ . *Phys. Rev. Appl.* **2019**, *11*, 014058.
- [2] Park, J.; Xia, Y.; Ganose, A. M.; Jain, A.; Ozoliņš, V. High Thermoelectric Performance and Defect Energetics of Multipocketed Full-Heusler Compounds. *Phys. Rev. Appl.* **2020**, *14*, 024064.
- [3] Wang, W.; Dai, Z.; Wang, X.; Zhong, Q.; Zhao, Y.; Meng, S. Low Lattice Thermal Conductivity and High Figure of Merit in n-type Doped Full-Heusler Compounds  $\text{X}_2\text{YAu}$  ( $\text{X} = \text{Sr}, \text{Ba}$ ;  $\text{Y} = \text{As}, \text{Sb}$ ). *Int. J. Energy Res.* **2021**, *45*, 20949-20958.

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THANK YOU