# Feature engineering for chemical compound prediction

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### 1 Abstract

The purpose of this analysis is to identify which variables have a significant effect on melting point of a chemical compound given 21 numerical variables.

#### 1.1 Melting temperature

The melting point of a chemical compound is the temperature at which the compound melts. It is an important property to chemists who aim to develop new compounds that have particular thermophysical behaviour.

In the experiment 60 observations are made with melting point ranging from **68.17** to **220.22**, its median equals **142.07** while mean **142.05**.

## 2 Analysis

#### 2.1 Correlation plot

The very simplest way of observing dependencies between variables is a correlation matrix. Plotting the matrix clearly shows correlation coefficients between a set of variables. The values of coefficients range from -1 to 1, correlations of 0.7 and bigger indicate highly correlated variables.

Correlation plot shows that melting point is highly correlated (0.82) with enthalpy of fusion 2 ( $x_4$ ) and significantly correlated with enthalpy of fusion 1 and polar surface area respectively 0.66 and 0.55.

From the plot we can also observe correlations between other factors what can be useful for further feature engineering if we aim to predict melting point.

Feature	Description
$x_1$	Approximate average width of melting peak
$x_2$	Molecular weight
$x_3$	Enthalpy of fusion 1
$x_4$	Enthalpy of fusion 2
$x_5$	Unit cell density
$x_6$	Partition coefficient
$x_7$	Polar surface area
$x_8$	Molecular volume
$x_9$	Molecular volume from Spartan
$x_{10}$	Number of molecules in the unit cell
$x_{11}$	Unit cell volume
$x_{12}$	Molecular volume/Unit cell volume
$x_{13}$	Molecular dipoles from Hartree
$x_{14}$	Molecular dipoles from Semi
$x_{15}$	Molecular surface area
$x_{16}$	IR frequency of H-bonding
$x_{17}$	Angle of H-bonding
$x_{18}$	Length of H-bonding
$x_{19}$	Torsion angle of $C^1 - S^1 - N^1 - C^7$ bond
$x_{20}$	Number of molecules around one molecule
$x_{21}$	Number of short contacts of one molecule excluding hydrogen bonding

Figure 1: Variables description



Figure 2: Correlation matrix

#### 2.2 Regression analysis

Statistical approach used here will be to select the best regression model from all models that could be fitted. The chosen model would balance complexity and goodness of fit excluding irrelevant variables.

Stepwise algorithm is performed (using forward selection). The algorithm starts with the simplest (intercept) model and then compares the simplest model with others +1 variable models, chooses the one with the smallest Aikake Information Criterion (AIC) to balance complexity and goodness of fit:

$$AIC_i = nlog(\frac{RSS_i}{n}) + 2k \tag{1}$$

Hence the fitted model is:

$$y = -341.65 + 0.65x_1 + 0.26x_2 + 2.53x_4 + 1.04x_{10} + 4.03x_{14} + 97.611x_{18}$$
<sup>(2)</sup>

Note that the algorithm included  $x_4$  but excluded  $x_3$  that has significant correlation with response variable. It may be because of bigger variance of  $x_3$  (var( $x_3$ ) = 1023.528, while var( $x_4$ ) = 98.57) or high correlation between  $x_3$  and  $x_4$  (cor( $x_3, x_4$ ) = 0.915), the algorithm didn't want to overcomplicate the model including two highly correlated variables. However, the plot clearly shows relationship between both  $x_3, x_4$  and response variable.

Note also that according to p-values  $x_2$  is the second most important variable in the model (its p-value equals  $4.72 \times 10^{-6}$ ), the most important is  $x_4$  (its p-value equal to  $4.37 \times 10^{-15}$ ).

It's also worth adding why  $x_7$  was excluded despite significant correlation. Despite one outlier it shows linear relation with dependence variable (look Figure 4), so it's significant when prediciting melting point. However, it could have been excluded due to the fact that there are only 8 unique values so it could be represented as a factor variable not numerical.

We have a few tools to compare this model to different ones. Checking the summary of the model we see that  $R^2 = 0.82$  and  $adj.R^2 = 0.80$ . We can moreover check the root mean square error:

$$RMSE(\theta) = \sqrt{E(\hat{\theta} - \theta)^2}$$
(3)



Figure 3: Melting point/enthalpies plot



Figure 4: Melting point/polar surface area plot

In this case - RSME = 15.15. This values describe how well model fits, it could be compared for example with RMSE of different model to choose which fits better (for example RMSE of intercept only model is 35.76, and full model's 11.84 so RMSE of fitted model is not that big compared with full model bearing in mind that it includes 6 variables not 21). However, prediction and checking goodness of fit may not be adequate since the dataset is small and outliers are harder to detect (and remove). If the dataset was bigger we could also use **cross validation** and so obtain better measure of predictive power of the model.

### 2.3 Boruta algorithm

The algorithm is designed as a wrapper around a random forest classification algorithm. It iteratively removes the features which are proved by a statistical test to be less relevant than random probes (i.e. having smaller Z-scores). [1]



Boruta output

Figure 4: Boruta used on crystal data

The algorithm confirmed what was concluded in the correlation matrix plot. Namely, the plot states that the most important variables are  $x_4$ ,  $x_7$  and  $x_3$ , we have also note  $x_2$ ,  $x_5$  are quite important.

### 3 Conclusion

All in all, I observed that the biggest effect on melting point have enthalpies (especially enthalpy 2) and polar surface area. Except those, also molecular weight and unit cell density have some effect on melting point.

If the aim was to predict the response variable it could be useful to use principal component analysis. PCA reduces number of dimensions by 10 and so 11 components explain 96.3% of variability.

## References

 Miron B Kursa, Aleksander Jankowski, and Witold R Rudnicki. Boruta–a system for feature selection. *Fundamenta Informaticae*, 101(4):271–285, 2010.



Figure 5: Principal component analysis plot