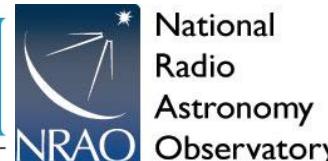


Introduction to Astrochemistry Part 4: Life Seeded from Space? Comets and Exoplanets

Dr. Samantha Scibelli

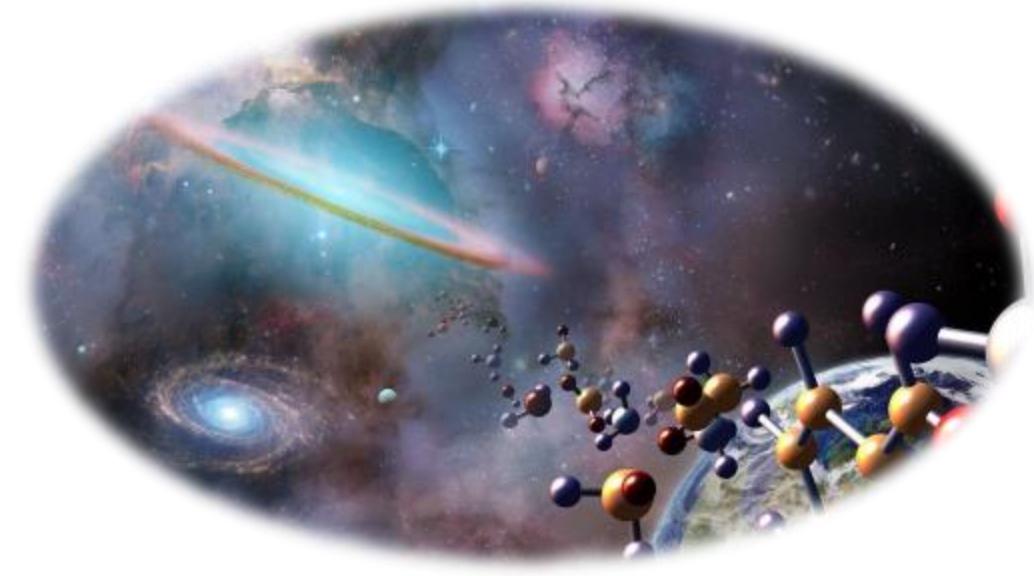
Jansky Fellow at the National Radio Astronomy
Observatory (NRAO)

AAA.org Lecture, June 11th, 2024



Astrochemistry, or “Molecular Astrophysics”

Definition: The study of the formation and destruction of molecules in the Universe, their interaction with radiation, and their feedback on physics of the environments

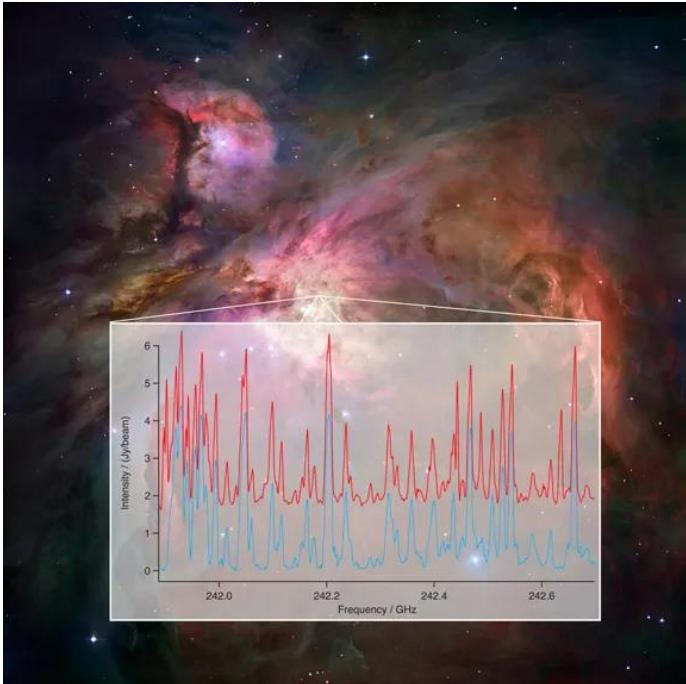


I write about molecules with great diffidence, having not yet rid myself of the tradition that atoms are physics, but molecules are chemistry, but the new conclusions that hydrogen is abundant seems to make it likely that the above mentioned elements H, O, and N will frequently form molecules

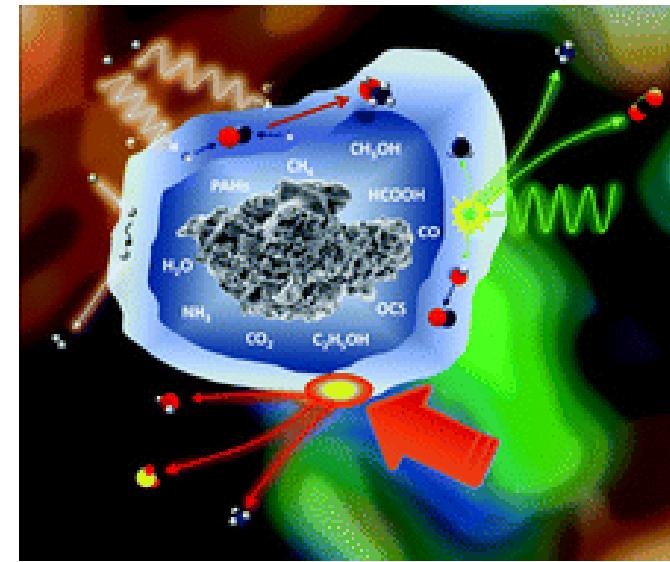
- Sir A. Eddington, 1937

Astrochemistry is an interdisciplinary field! Including, chemistry, physics, astronomy, biology, etc.,

Observations



Modeling



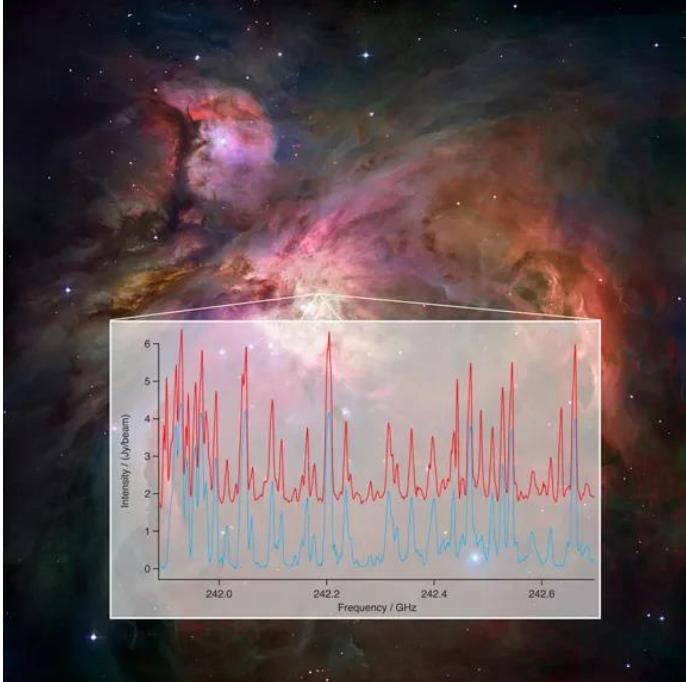
Things an astro**chemist** does

Laboratory



Astrochemistry is an interdisciplinary field! Including, chemistry, physics, astronomy, biology, etc.,

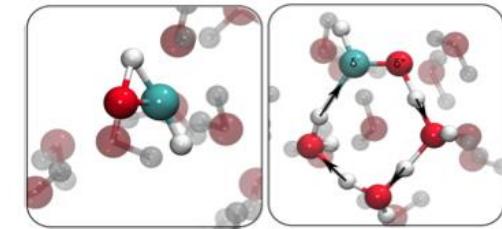
Observations



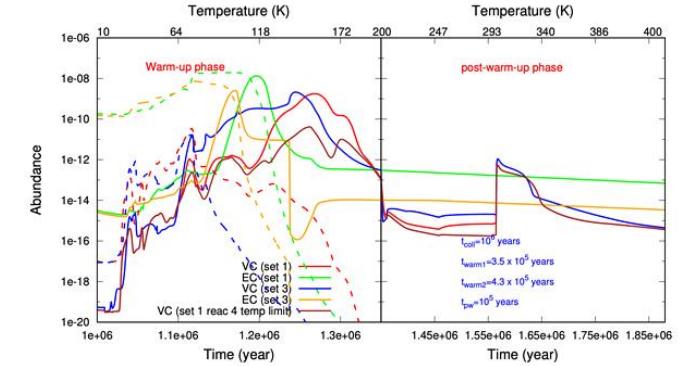
Things an astro**chemist** does

Modeling

Computational chemistry



Simulations of chemical networks



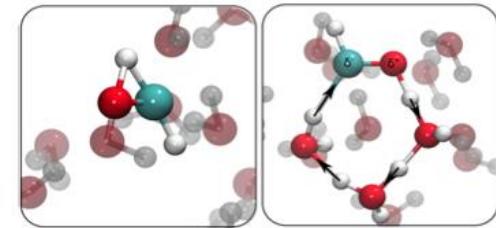
Laboratory



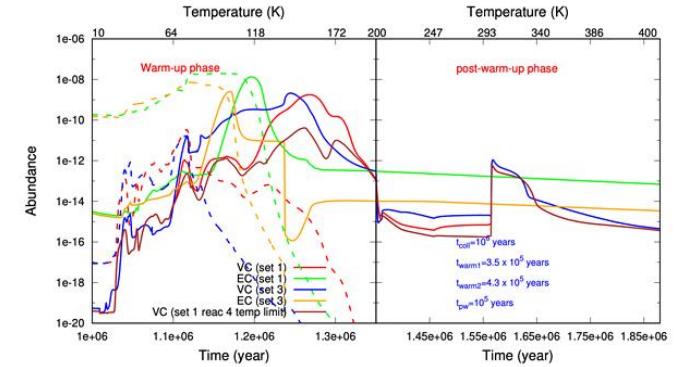
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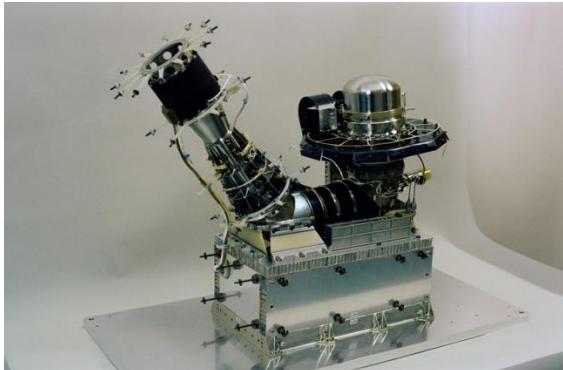


Observations

Millimeter Telescopes



Mass Spectrometers



Things an astro**chemist** does

Laboratory



A Molecular Universe!

2 Atoms	3 Atoms	4 Atoms	5 Atoms	6 Atoms	7 Atoms		
CH	NH	H ₂ O	MgCN	NH ₃	SiC ₃		
CN	SiN	HCO ⁺	H ₃ ⁺	H ₂ CO	CH ₃		
CH ⁺	SO ⁺	HCN	SiCN	HNCO	C ₃ N ⁻		
OH	CO ⁺	OCS	AlNC	H ₂ CS	PH ₃		
CO	HF	HNC	SiNC	C ₂ H ₂	HCNO		
H ₂	N ₂	H ₂ S	HCP	C ₃ N	HO CN		
SiO	CF ⁺	N ₂ H ⁺	CCP	HNCS	HSCN		
CS	PO	C ₂ H	AlOH	HOCO ⁺	HOOH		
SO	O ₂	SO ₂	H ₂ O ⁺	C ₃ O	I-C ₃ H ⁺		
SiS	AlO	HCO	H ₂ Cl ⁺	I-C ₃ H	HMgNC		
NS	CN ⁻	HNO	KCN	HCN ⁺	HCCN ⁺		
C ₂	OH ⁺	HCS ⁺	FeCN	H ₃ O ⁺	CNCN		
NO	SH ⁺	HOC ⁺	HO ₂	C ₃ S	HONO		
HCl	HCl ⁺	SiC ₂	TiO ₂	c-C ₃ H	MgCCH		
NaCl	SH	C ₂ S	CCN	HC ₂ N	HCCNC		
AlCl	TiO	C ₃	SiCSi	H ₂ CN	HCCS		
KCl	ArH ⁺	CO ₂			H ₂ COH ⁺		
AlF	NS ⁺	CH ₂					
PN	HeH ⁺	C ₂ O	HSC				
SiC	VO	MgNC	NCO				
CP		NH ₂	CaNC				
		NaCN	NCS				
		N ₂ O					
8 Atoms	9 Atoms	10 Atoms	11 Atoms	12 Atoms	13 Atoms	PAHs	Fullerenes
HCOOCH ₃	CH ₃ OCH ₃	CH ₃ COCH ₃	HC ₉ N	C ₆ H ₆	C ₆ H ₅ CN	1-C ₁₀ H ₇ CN	C ₆₀
CH ₃ C ₃ N	CH ₃ CH ₂ OH	HOCH ₂ CH ₂ OH	CH ₃ C ₆ H	n-C ₃ H ₇ CN	HC ₁₁ N	2-C ₁₀ H ₇ CN	C ₆₀ ⁺
C ₇ H	CH ₃ CH ₂ CN	CH ₃ CH ₂ CHO	CH ₃ C ₅ H	C ₂ H ₅ OCHO	i-C ₃ H ₇ CN		
CH ₃ COOH	HC ₇ N	CH ₃ C ₅ N	CH ₃ C ₅ H	CH ₃ COOCH ₃	1-C ₅ H ₅ CN		
H ₂ C ₆	CH ₃ C ₄ H	CH ₃ CHCH ₂ O	CH ₃ COCH ₂ OH	2-C ₅ H ₅ CN			
CH ₂ OHCHO	C ₈ H	CH ₃ OCH ₂ OH	C ₅ H ₆				
HC ₆ H	CH ₃ CONH ₂						
CH ₂ CHCHO	C ₈ H ⁻						
CH ₂ CCHCN	CH ₂ CHCH ₃						
NH ₂ CH ₂ CN	CH ₃ CH ₂ SH						
CH ₃ CHNH	HC ₇ O						
CH ₃ SiH ₃	CH ₃ NHCHO						
NH ₂ CONH ₂	H ₂ CCCCCHCH						
HCCCH ₂ CN	H ₂ CCCCHCN						
CH ₂ CHCCH	H ₂ CCCHC ₃ N						

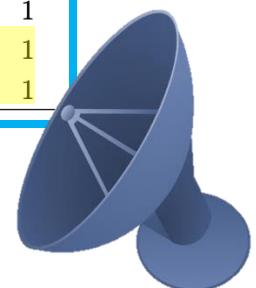
>300 Molecules

McGuire 2022; <https://arxiv.org/pdf/2109.13848>

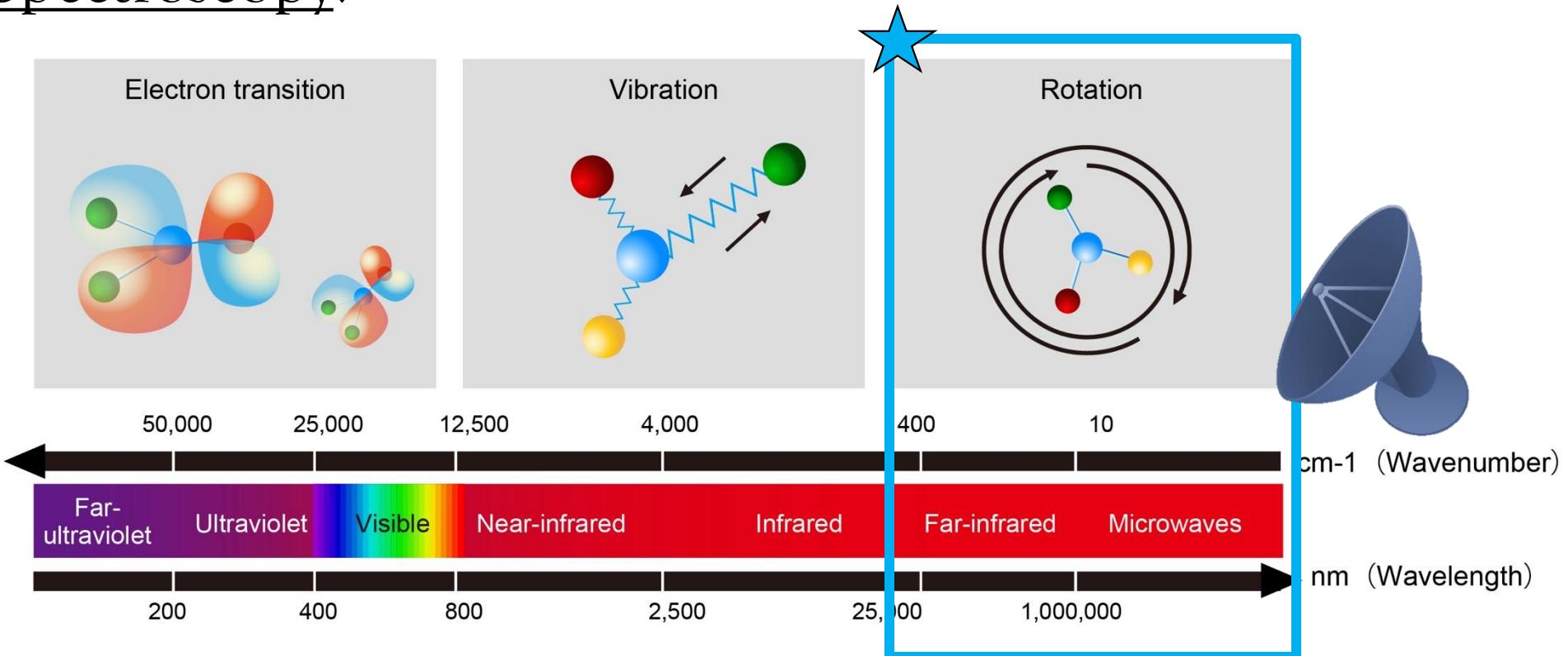
of molecule discoveries per observatory

Facility	#	Facility	#
IRAM 30-m	64	SMA	2
NRAO 36-ft	33	SEST	2
GBT 100-m	28	SOFIA	2
NRAO/ARO 12-m	27	Hat Creek 20-ft	2
Yebes 40-m	19	IRTF	2
Nobeyama 45-m	15	PdBI	2
NRAO 140-ft	13	OVRO	2
Bell 7-m	8	MWO 4.9-m	2
ALMA	8	Hubble	1
SMT	7	IRAS	1
Herschel	7	BIMA	1
Parkes	5	NRL 85-ft	1
FCRAO 14-m	5	ATCA	1
ISO	5	Mitaka 6-m	1
APEX	4	McMath Solar Telescope	1
Onsala 20-m	4	UKIRT	1
KPNO 4-m	4	Odin	1
Effelsberg 100-m	4	FUSE	1
Algonquin 46-m	3	KAO	1
Mt. Wilson	3	Mt. Hopkins 60-in	1
Spitzer	3	Aerobee-150 Rocket	1
Haystack	3	Millstone Hill 84-ft	1
CSO	2	Goldstone	1

> 90% Identified by Radio Astronomy!



Submillimeter and Millimeter Radio Telescopes Identify Molecules via Rotational Spectroscopy!



1) ELECTRONIC STATES

- electrons change levels
- energies in visible, UV

2) VIBRATIONAL STATES

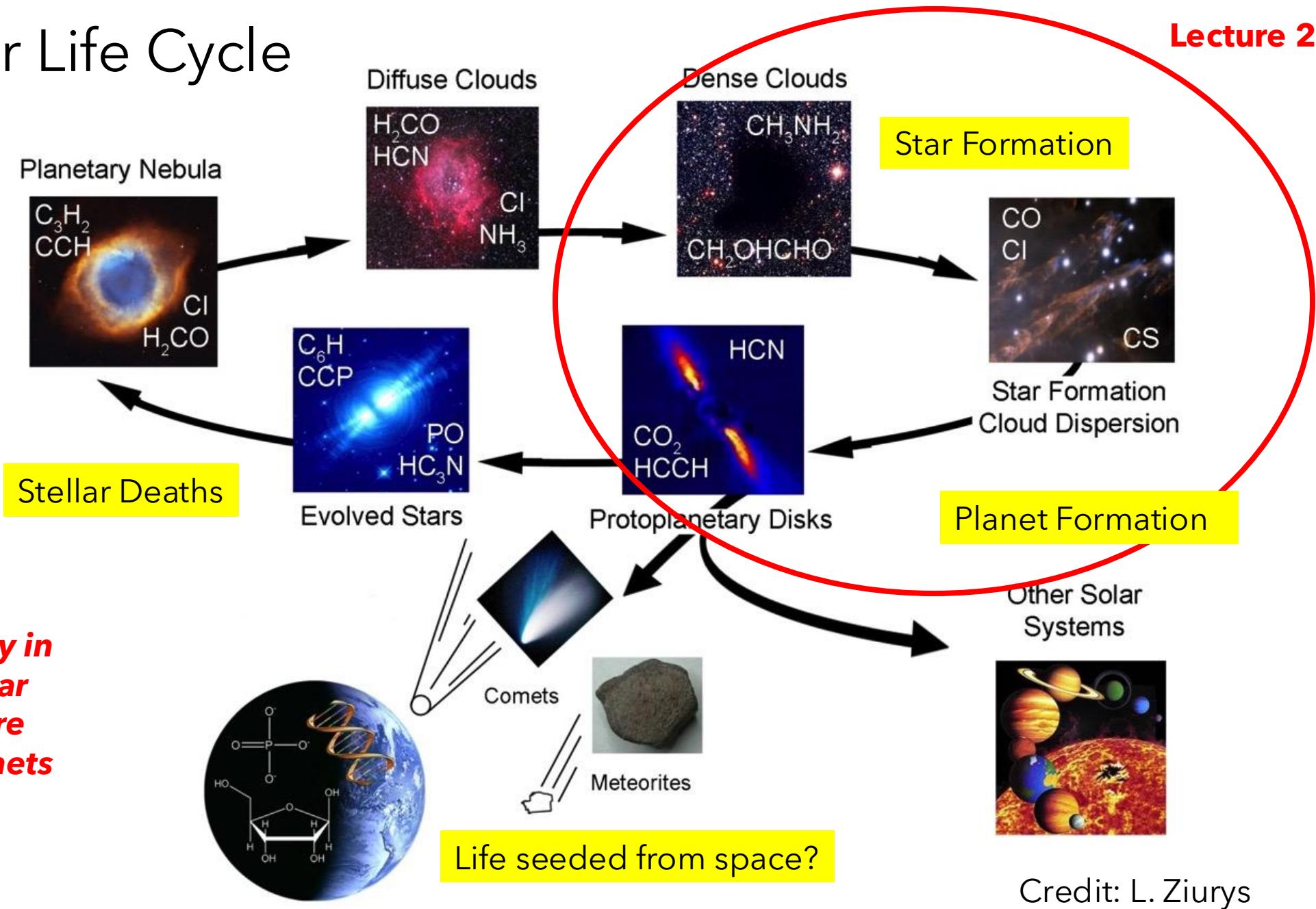
- normal modes of nuclear motions
- occur in infrared region

3) ROTATIONAL STATES

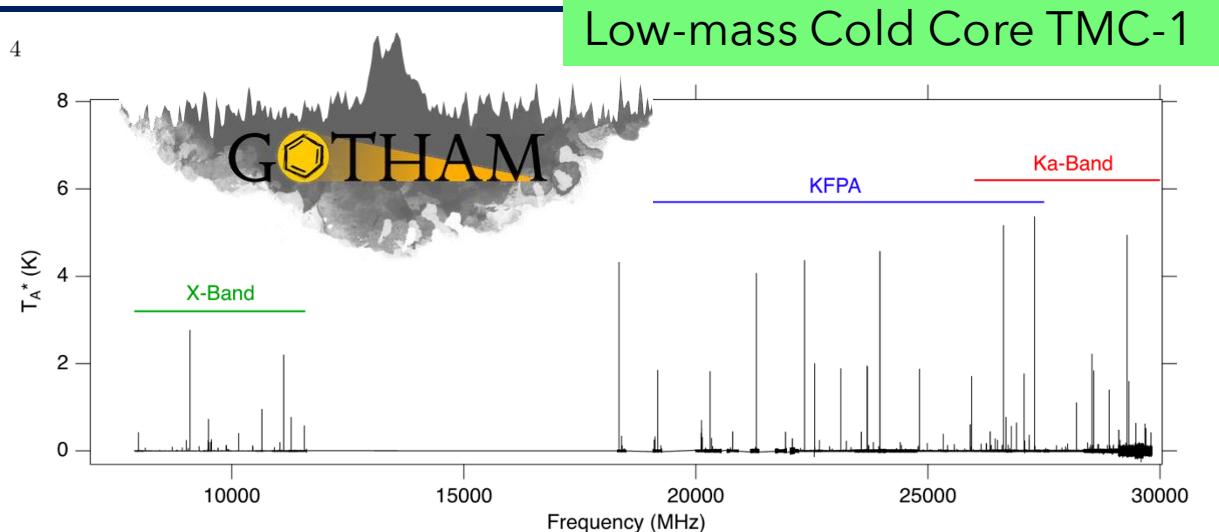
- end-on-end motion of nuclei
- energies in microwave/millimeter-wave regions

- Electronic states have **vibrational/rotational structure**
- Vibrational states have **rotational structure**

Molecular Life Cycle

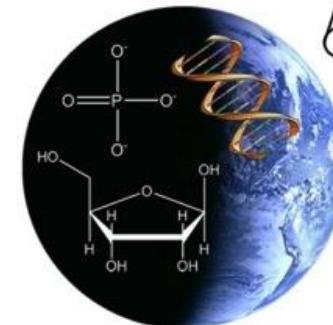


Lecture 2



McGuire et al. 2018, 2020

**Rich chemistry in
the molecular
clouds where
stars and planets
form!**

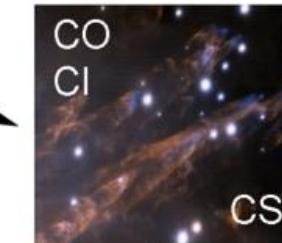


Life seeded from space?

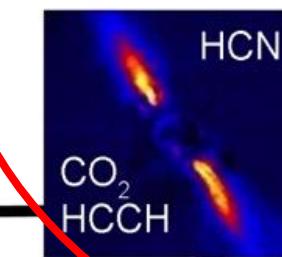
Dense Clouds



Star Formation

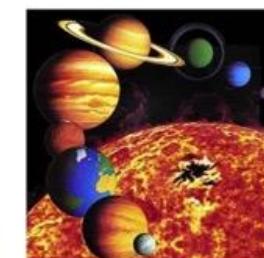


Star Formation
Cloud Dispersion



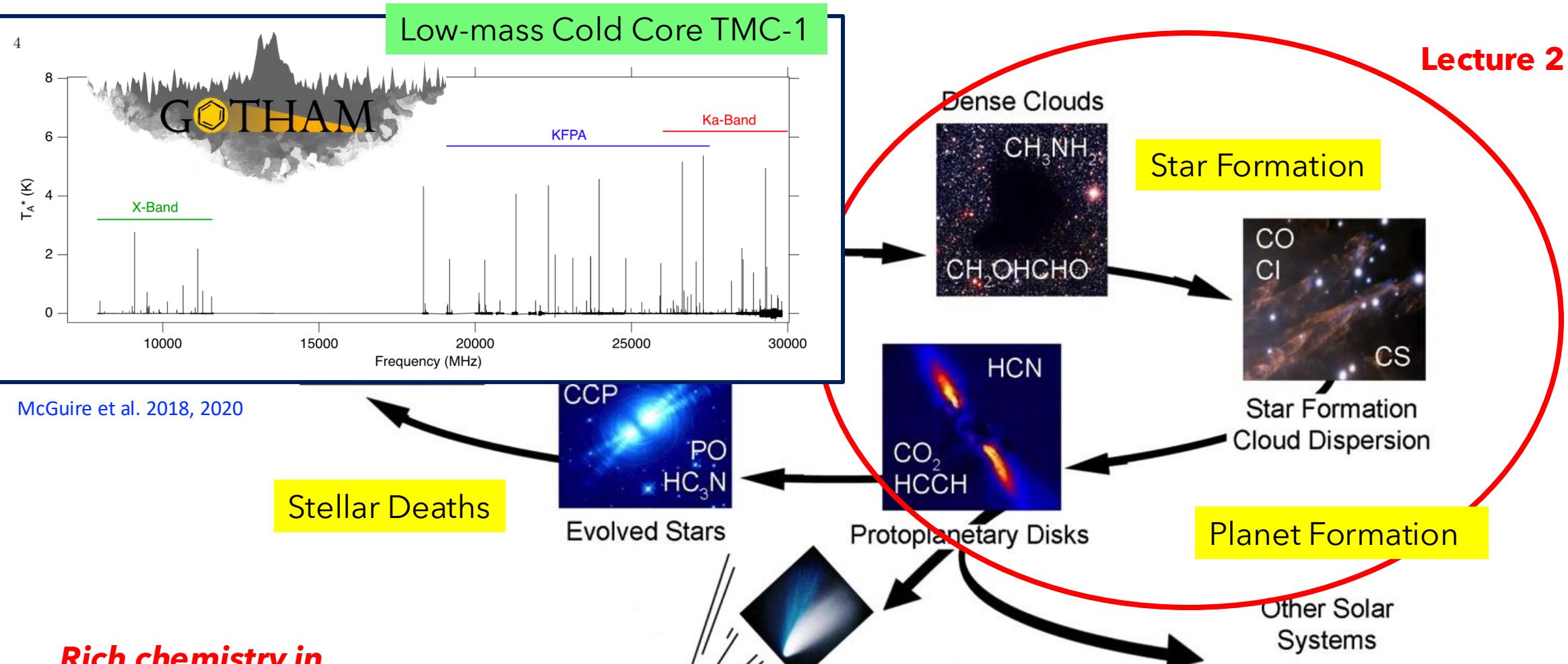
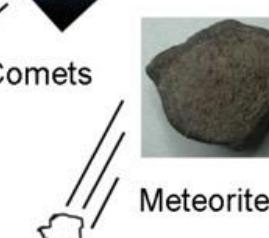
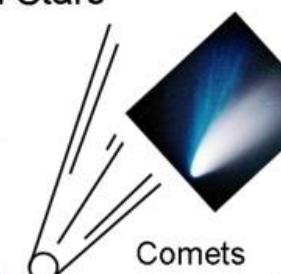
Planet Formation

Other Solar
Systems



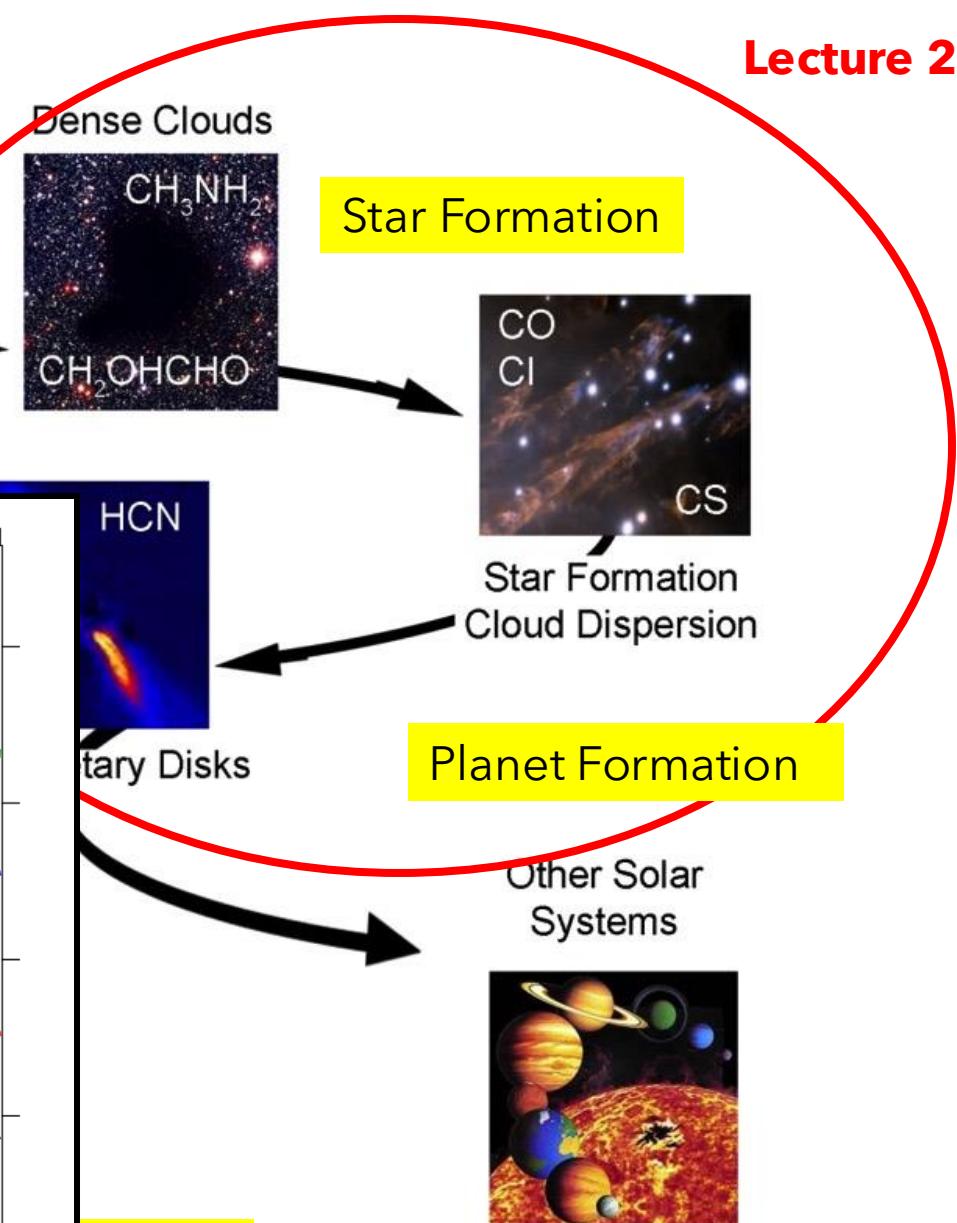
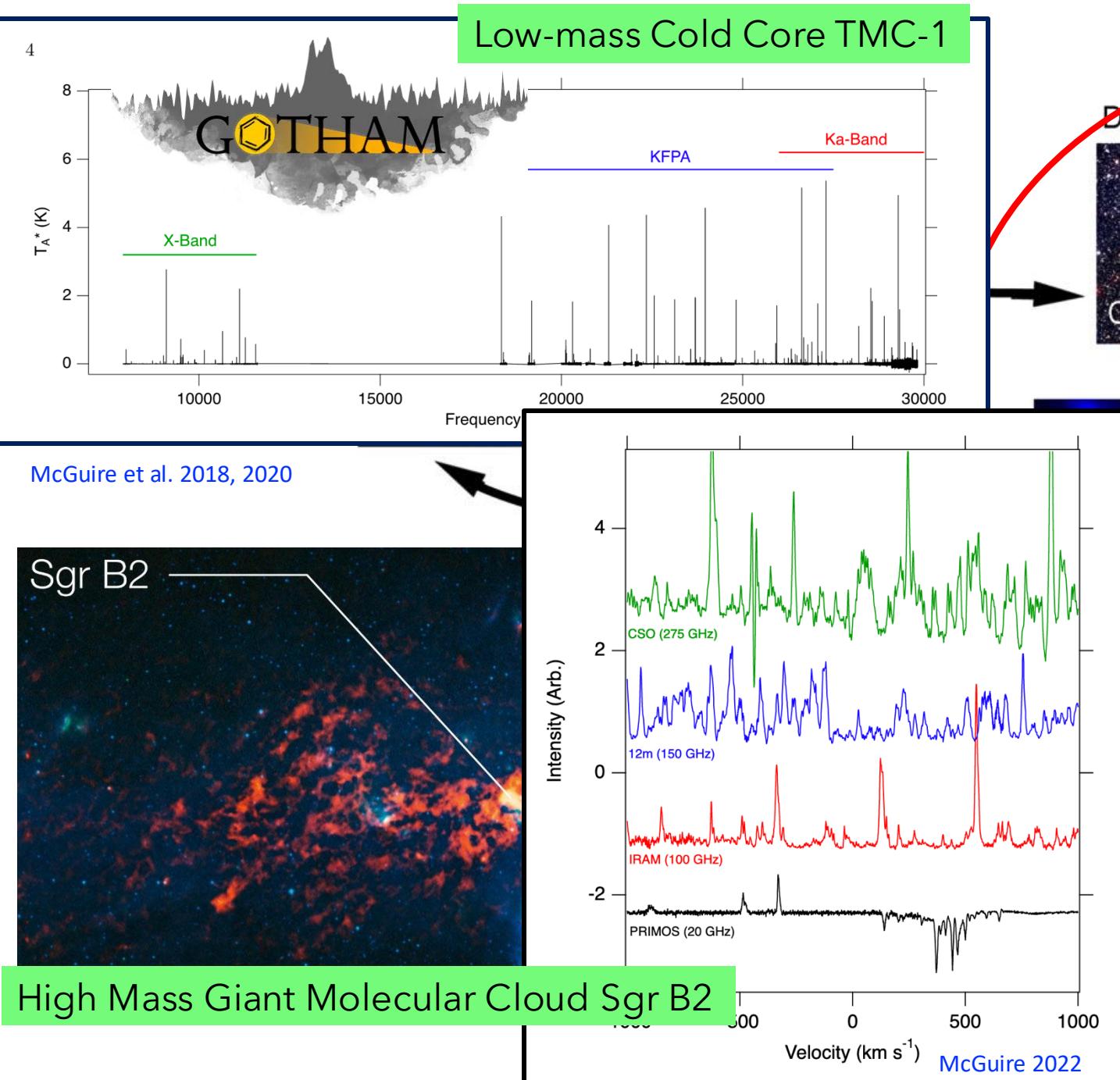
Evolved Stars

Protoplanetary Disks



Credit: L. Ziurys

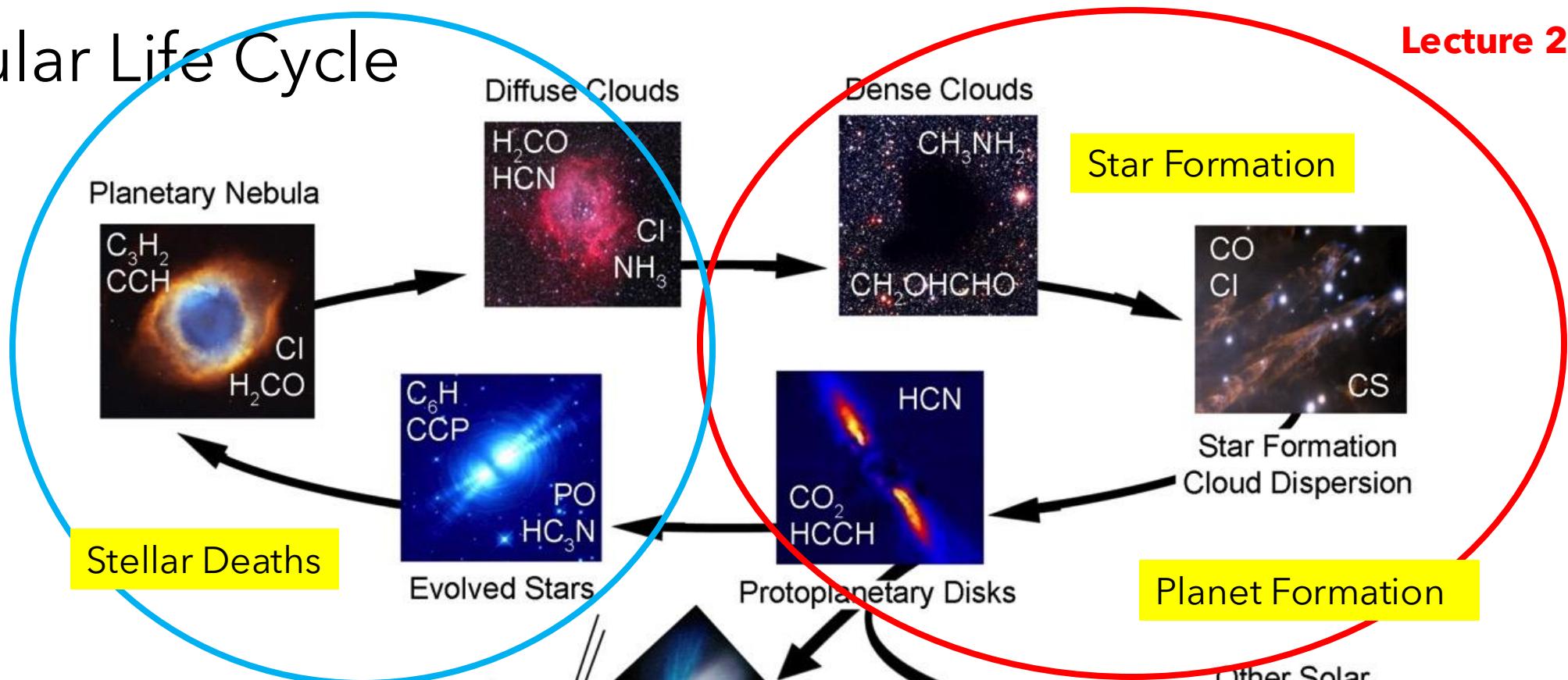
Lecture 2



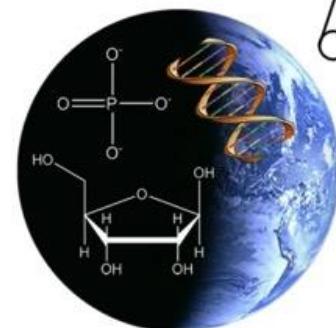
Credit: L. Ziurys

Molecular Life Cycle

Lecture 3



Recycling of material!
Evolved stars provide the material that enrich the diffuse and dense clouds that form stars and planets

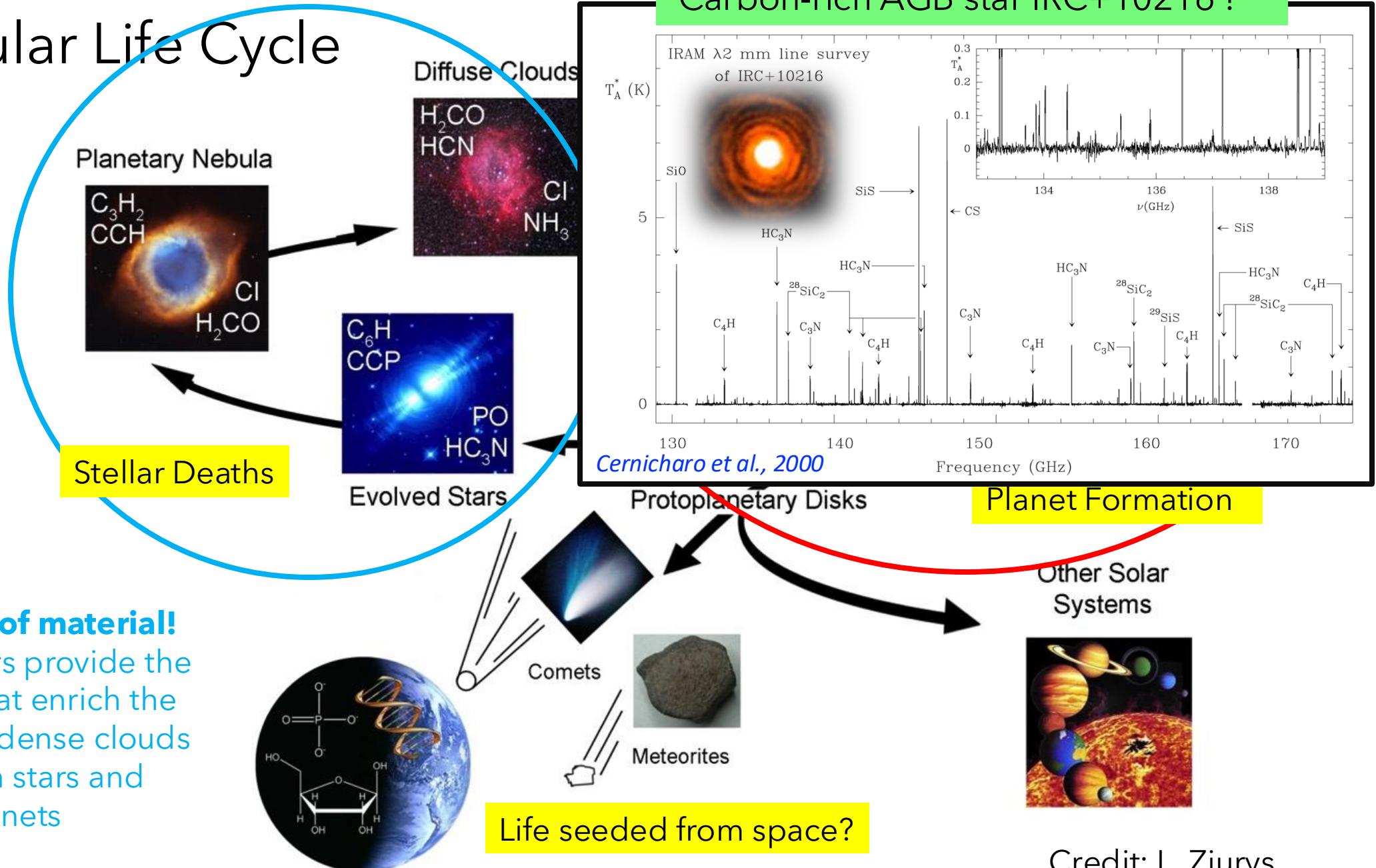


Life seeded from space?

Credit: L. Ziurys

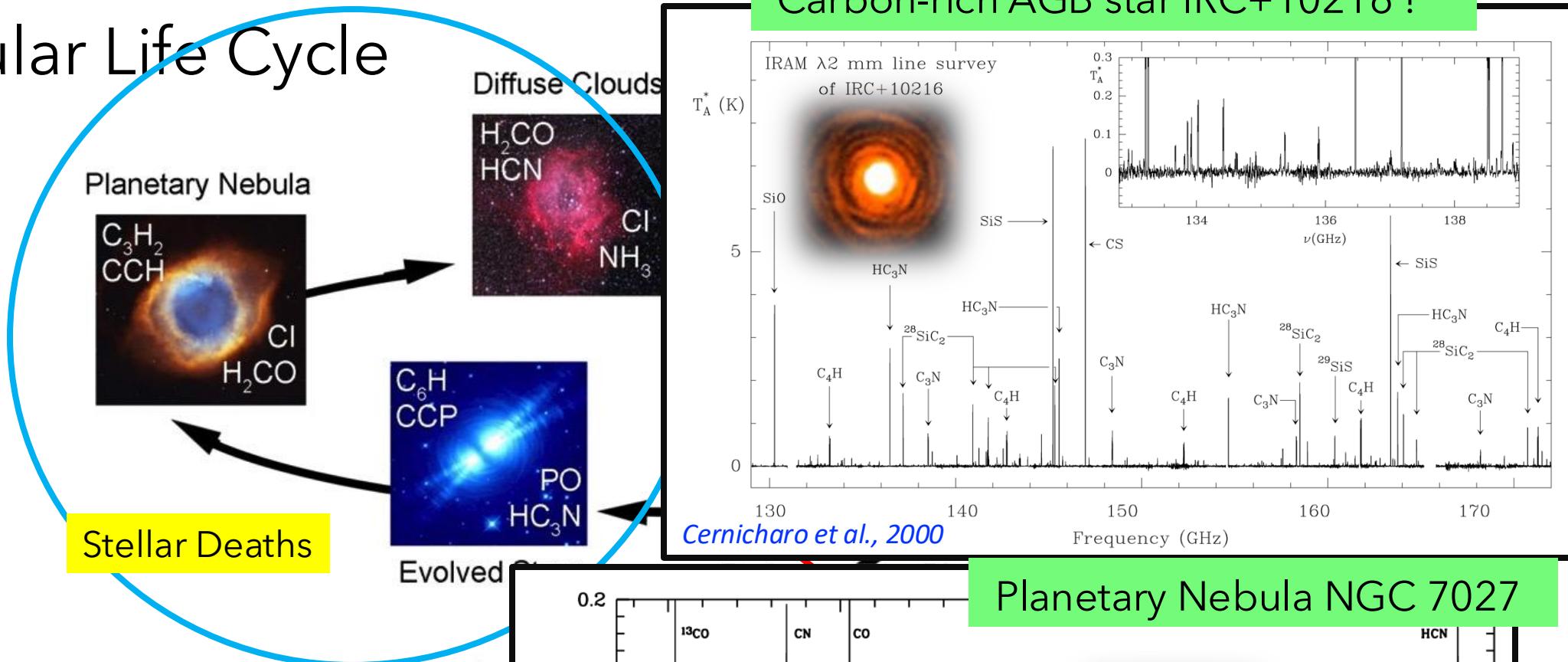
Molecular Life Cycle

Lecture 3



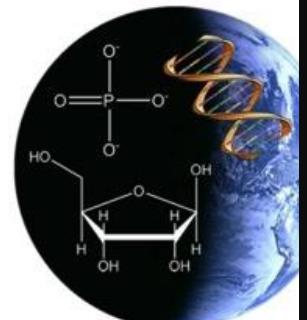
Molecular Life Cycle

Lecture 3



Recycling of material!

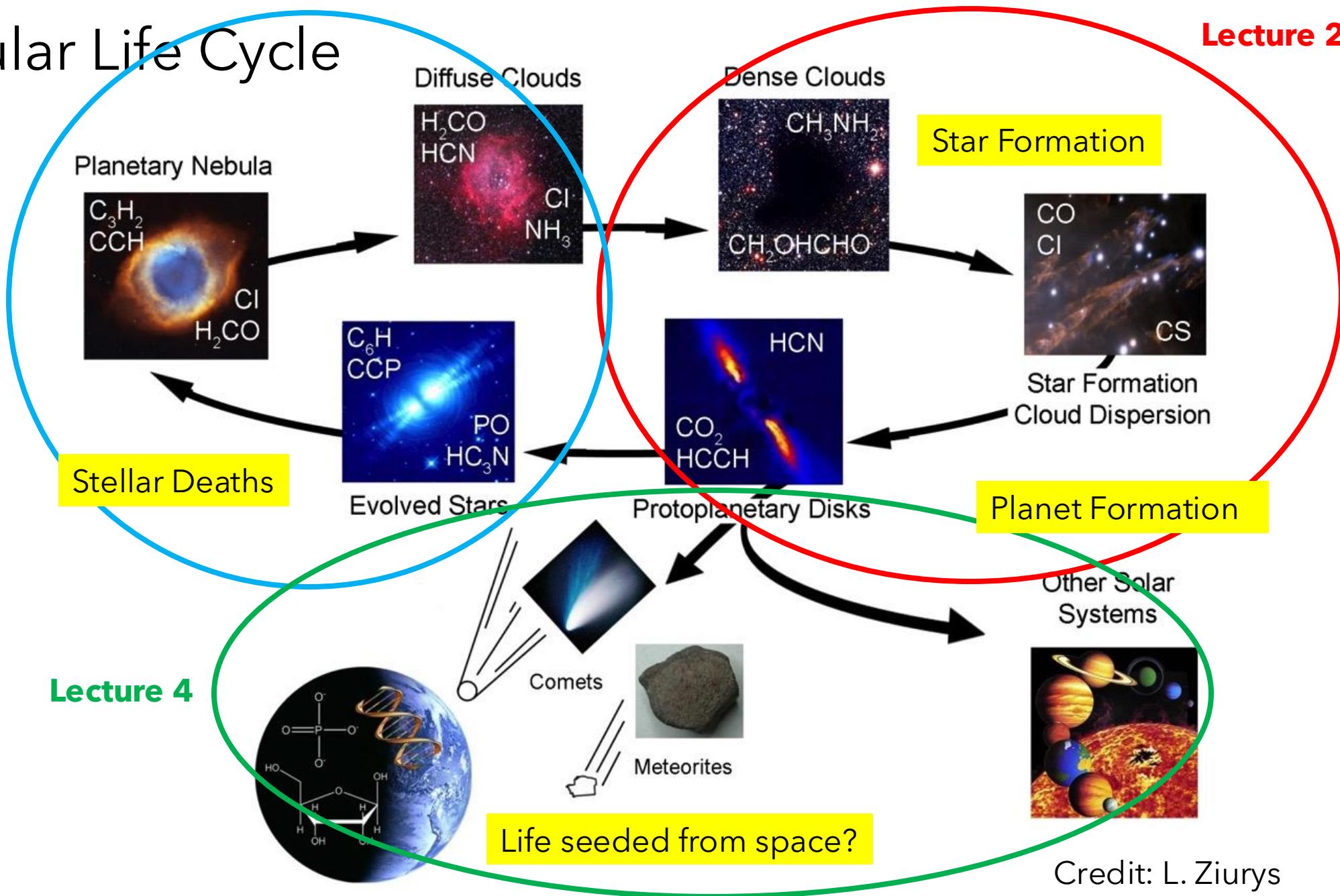
Evolved stars provide the material that enrich the diffuse and dense clouds that form stars and planets



Molecular Life Cycle

Lecture 2

Lecture 3



Credit: L. Ziurys

Comets – Structure and Composition

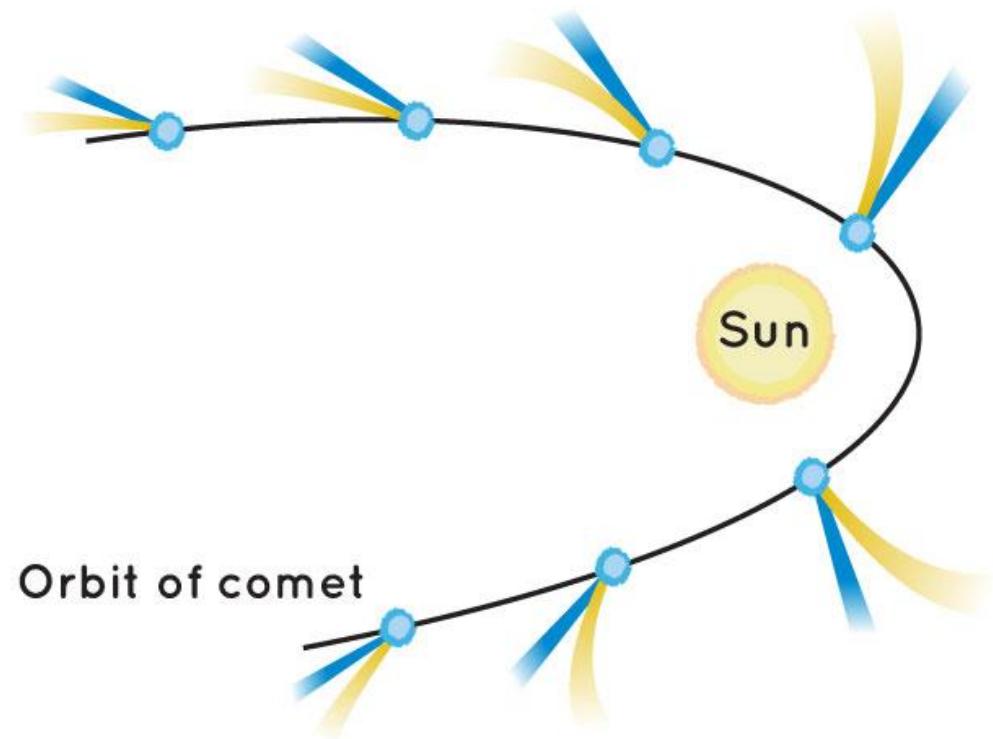
- Small, fragile, **irregularly shaped bodies**
- Composed of mixture of **non-volatile grains** and **frozen gases**
- Structures **diverse and dynamic**
- Over half the material is **ICE**
- A "dirty iceberg" or "dirty snowball"



Credit: L. Ziurys

Comets – Structure and Composition

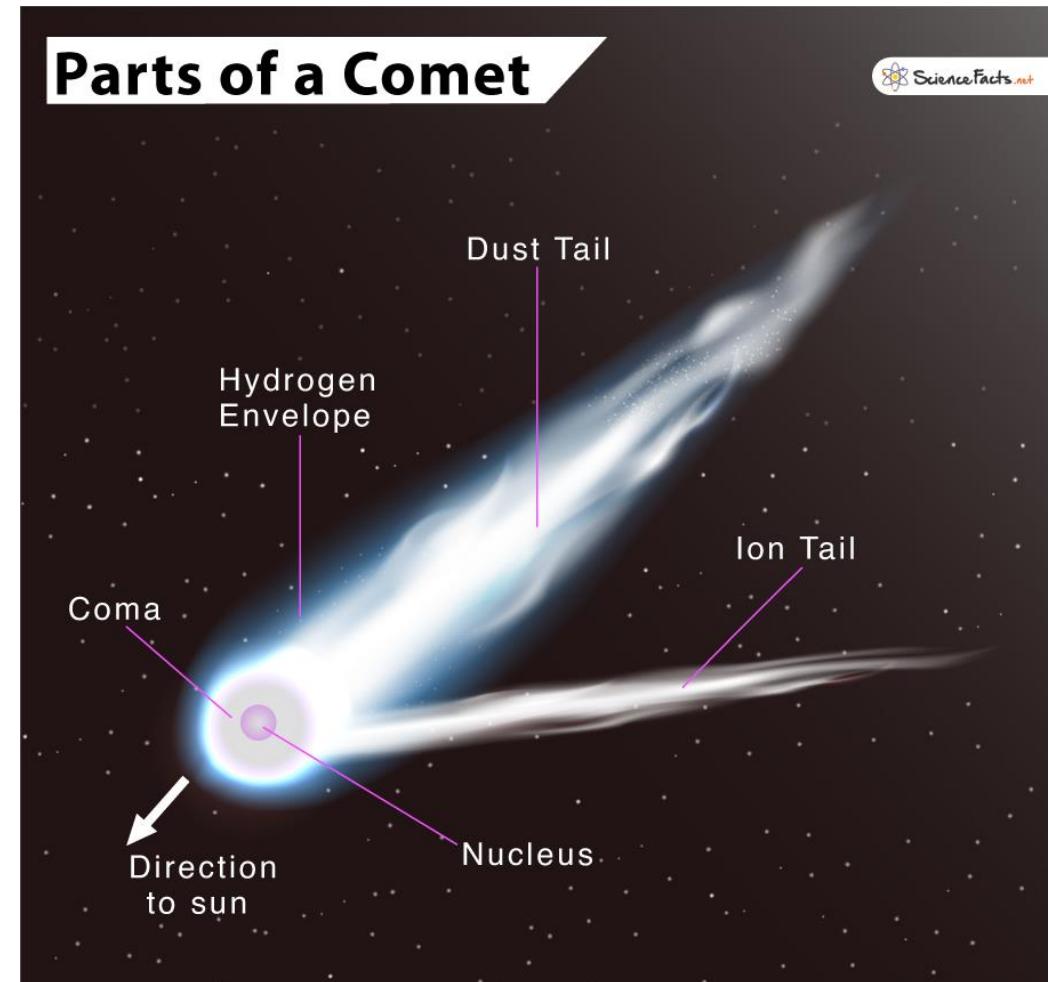
- Highly elliptical orbits: bring them very close to **Sun**
- As approach Sun, comets develop **enormous tails** of luminous material → Called a **coma**
- Extend for **millions of kilometers** from head
- **Usually grows in size and brightness near sun**
- Seen by reflected sunlight
- As coma develops, **dust reflects more sunlight**
- Direction: **away from the Sun**
- When far from Sun: nucleus is cold / frozen solid



Credit: L. Ziurys

Comets – Structure and Composition

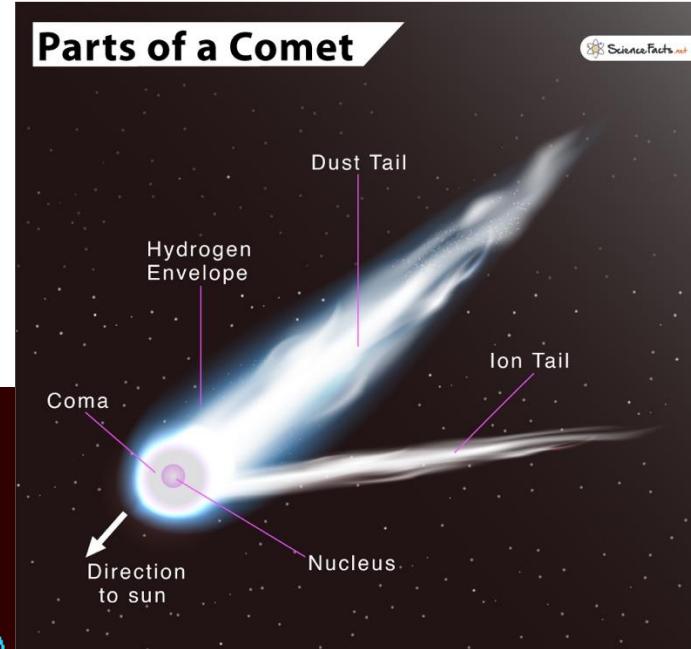
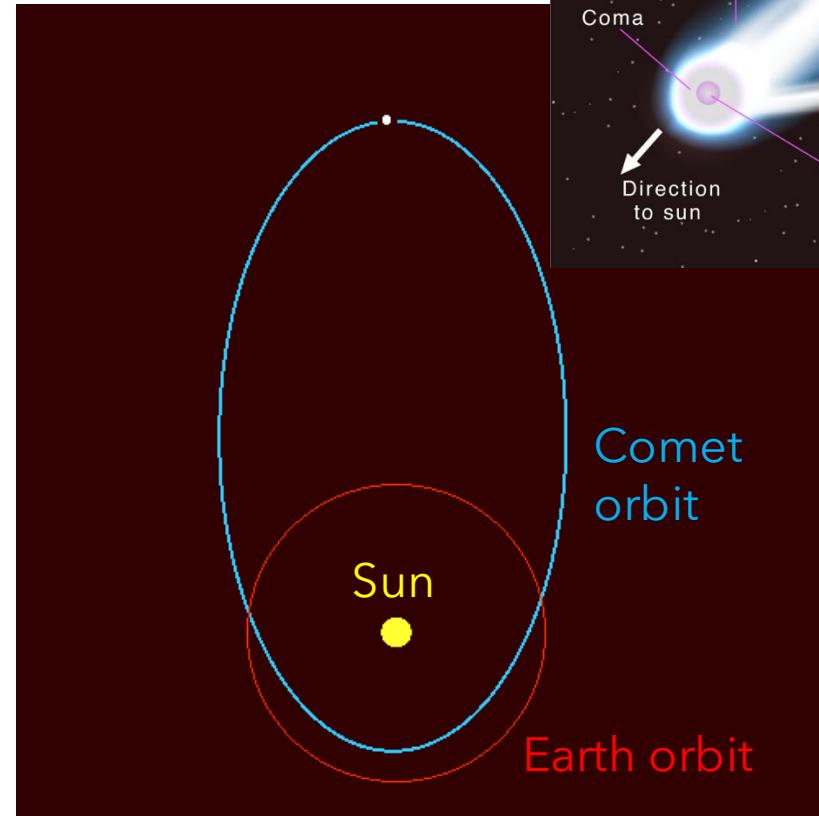
- Gas in coma absorbs **ultraviolet radiation** and begins to **fluoresce**
 - At ~ 5 AU from the Sun, **fluorescence** usually more intense than reflected light.
 - As comet absorbs **UV**, chemical processes release **hydrogen**
- Forms **hydrogen envelope** (not seen from Earth as the emission **absorbed by atmosphere**, instead detected by spacecraft)
- Small, bright nucleus (less than 10 km in diameter) visible in middle of coma
- Coma and the nucleus together constitute head of comet



Credit: L. Ziurys

Comets – Structure and Composition

- Sun's **radiation pressure and solar wind** accelerate materials
- Pushed **away from comet head**
- Acceleration depends on **the size/mass** of materials
- Massive **dust tails** accelerated slowly; thus **curved**
- **Ion tail** is much less massive: accelerated greatly
- **Nearly straight line** extending away from Sun
- Can be swung deeply into space, often **beyond orbit of Pluto**



Credit: L. Ziurys

Comet Origins: **Short Period**

- Two types of comets
 - **Short Period** and **Long Period**
- **Short Period Comets**
 - Arise in the **Kuiper Belt** (Kuiper-Belt objects)
 - Region with **icy bodies** around **Neptune's** orbit
 - ⇒ 50 AU from sun
 - ⇒ over 1,000 KBO's
 - Full orbit around sun and back: 200 yrs.
 - Travel in **plane of solar system**
 - Routinely reappear
 - Typical example: **Halley's Comet**
 - **Processed** material



Credit: L. Ziurys

Comet Origins: **Long Period**

- **Long Period Comets**
 - Originate in **Oort Cloud**
 - Immense **spherical cloud** surrounding Solar System
 - Remnant **molecular cloud**
 - Extends 18 trillion miles from Sun (100,000 AU)
 - **Edge of Solar System:** sun's gravitational influence ends
 - Contains **billions of icy bodies** in solar orbit
- **Passing stars** disturb orbit of a given body in Oort cloud
 - creates a long period comet
 - such comets **weakly bound** to sun
 - often only appear **once**
 - 4,000 yrs. to 30 million yrs.
 - Orbit **perpendicular** to Plane of ecliptic
 - Hale-Bopp a good example
 - Represent **Pristine** Material

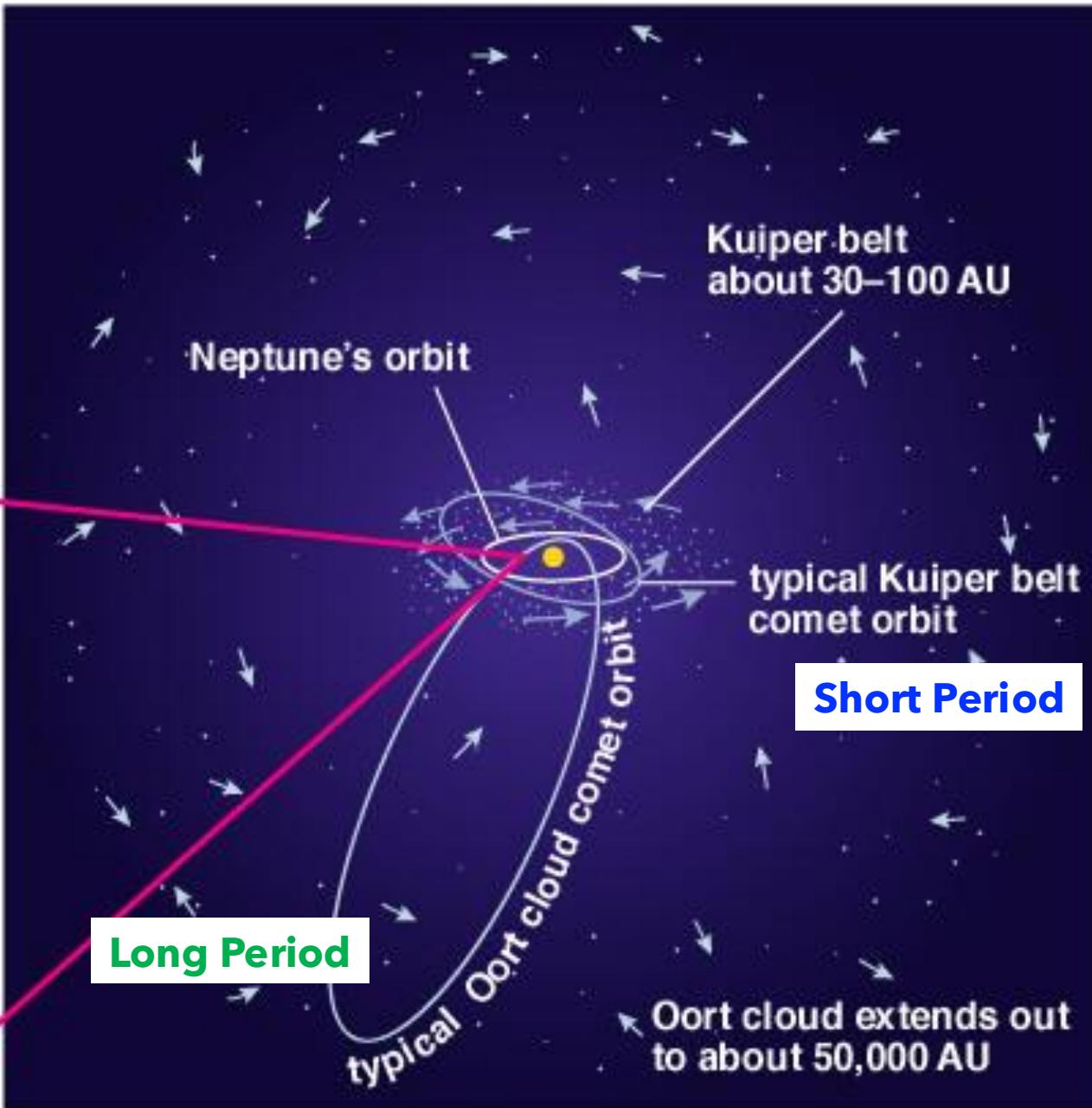


Credit: L. Ziurys

Comet Origins

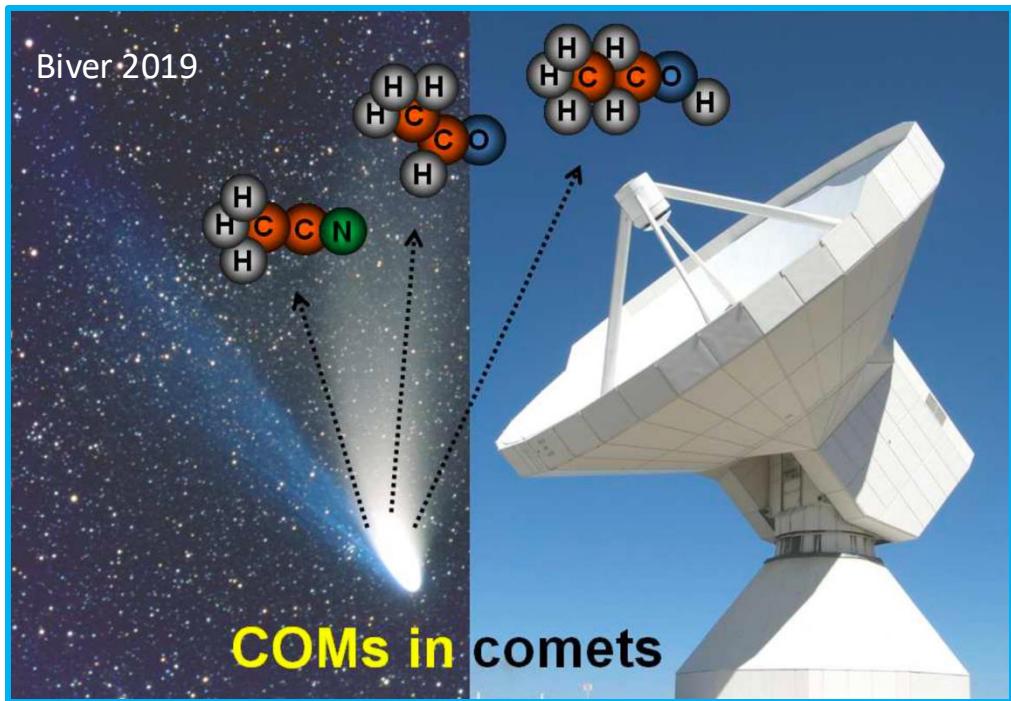
Oort Cloud:

Contains **pristine** material
of original Molecular
Cloud!

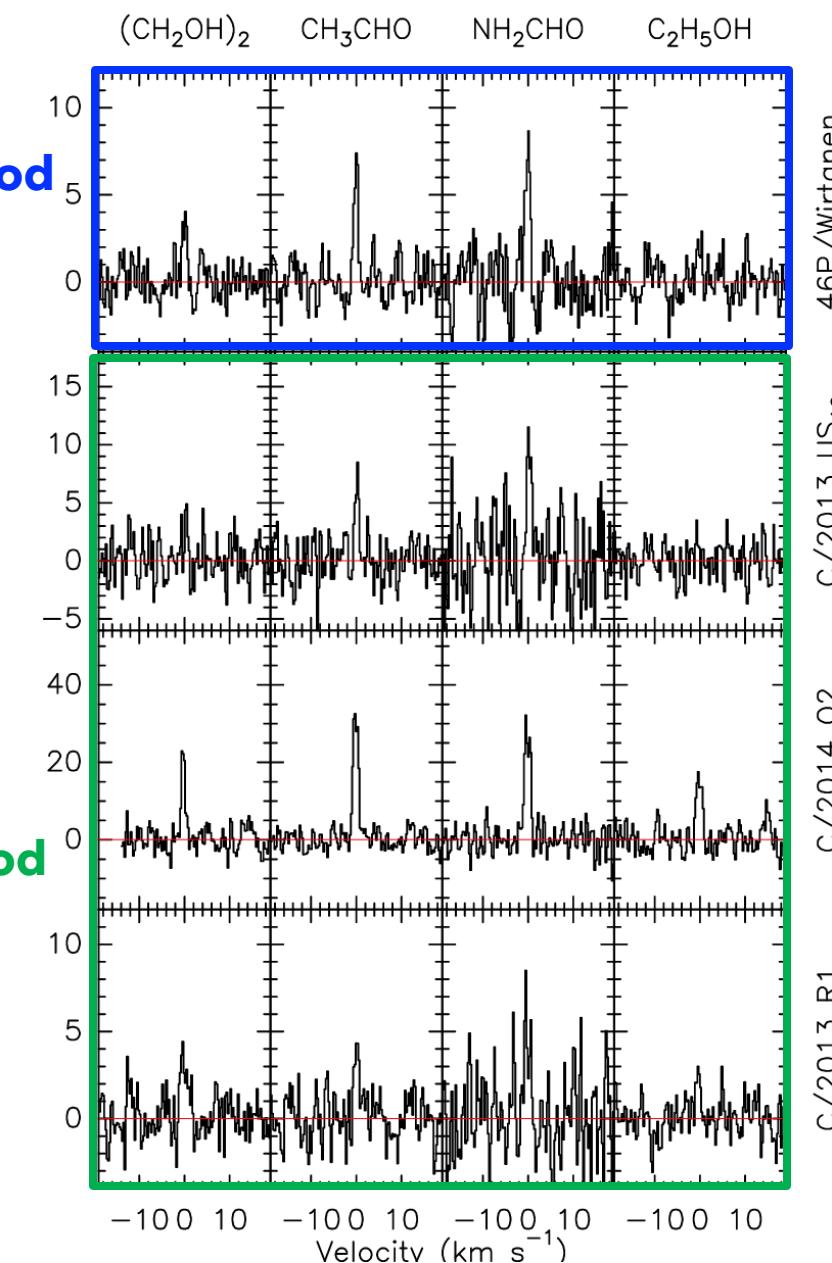


Complex Comet Chemistry!

Rotational spectra of the complex organic molecules (COMs) ethylene glycol, $(\text{CH}_2\text{OH})_2$, acetaldehyde, CH_3CHO , formamide, NH_2CHO , and ethanol, $\text{C}_2\text{H}_5\text{OH}$, obtained from IRAM 30m radio telescope observations of various comets →



Short-period

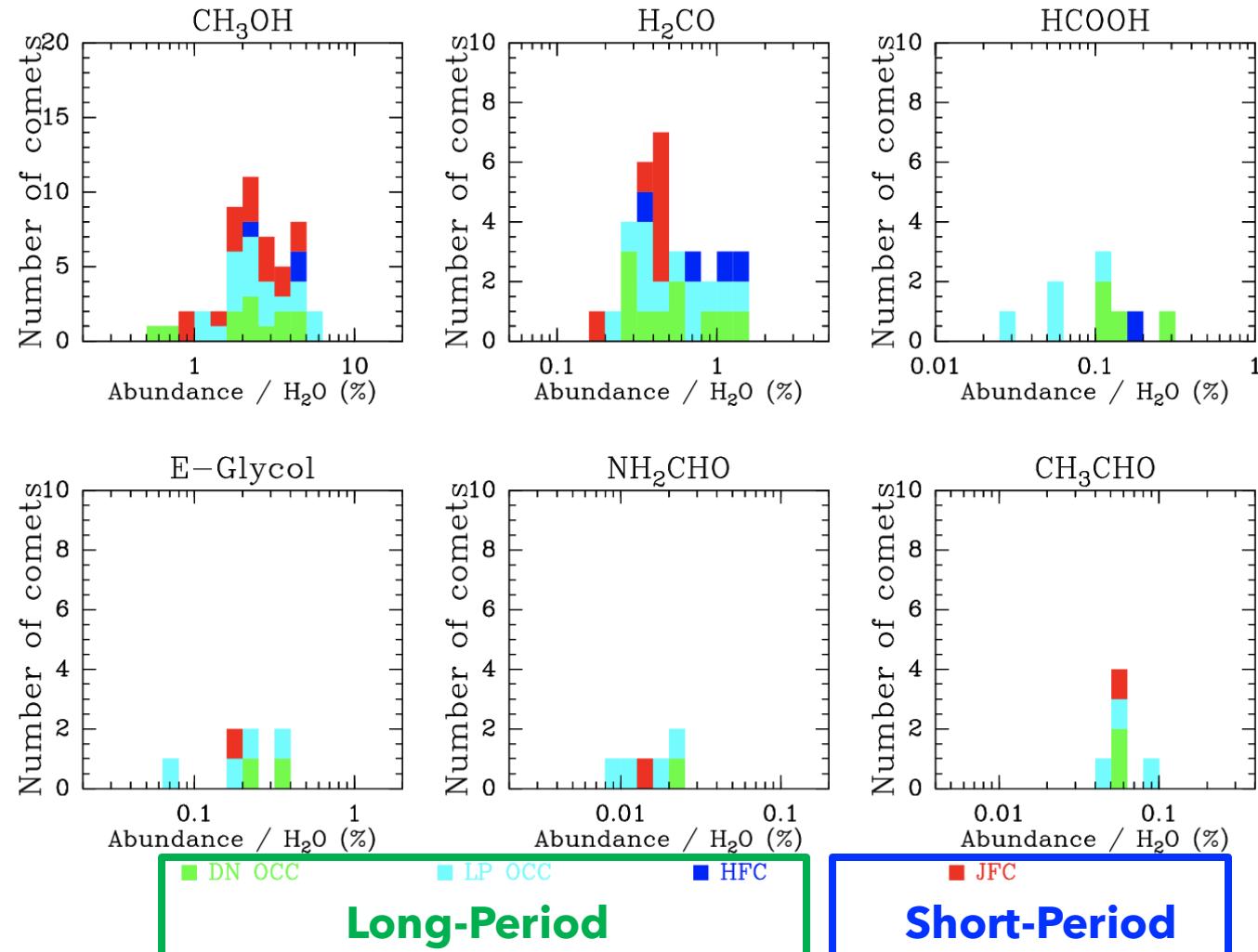


Long-period

Complex Comet Chemistry!

Biver 2019

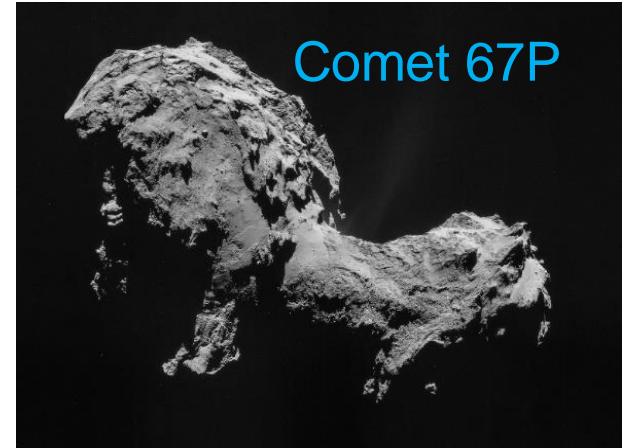
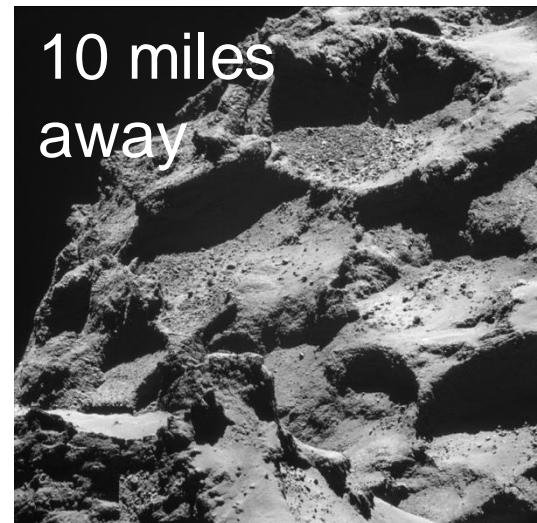
- Abundances in comets measured as production rates with respect to water, H_2O
- Histogram shows the distribution of the abundance relative to water of several COMS detected in more than 6 comets.
- The scatter in **abundances** (relative to water or methanol) of the COMs is less than one order of magnitude and the abundances **do not seem to depend on the dynamical family** (long-period Oort cloud comets, OCC, or short-period Jupiter family comets, JFC) of the comet



Complex Comet Chemistry!

Special Case: the Rosetta Mission to JFC 67P

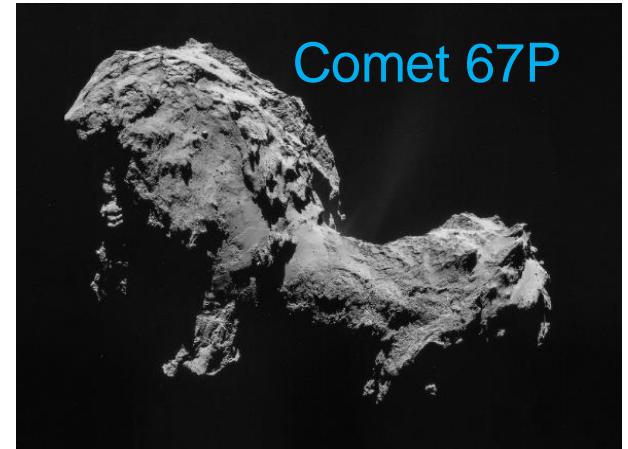
- **Rosetta Mission:** Mission to the Jupiter Family Comet (JFC) **67P/Churyumov-Gerasimenko, or 67P**
- Launched in March 2004 and made a 10-year journey across the solar system to arrive at its target in 2014



Complex Comet Chemistry!

Special Case: the Rosetta Mission to JFC 67P

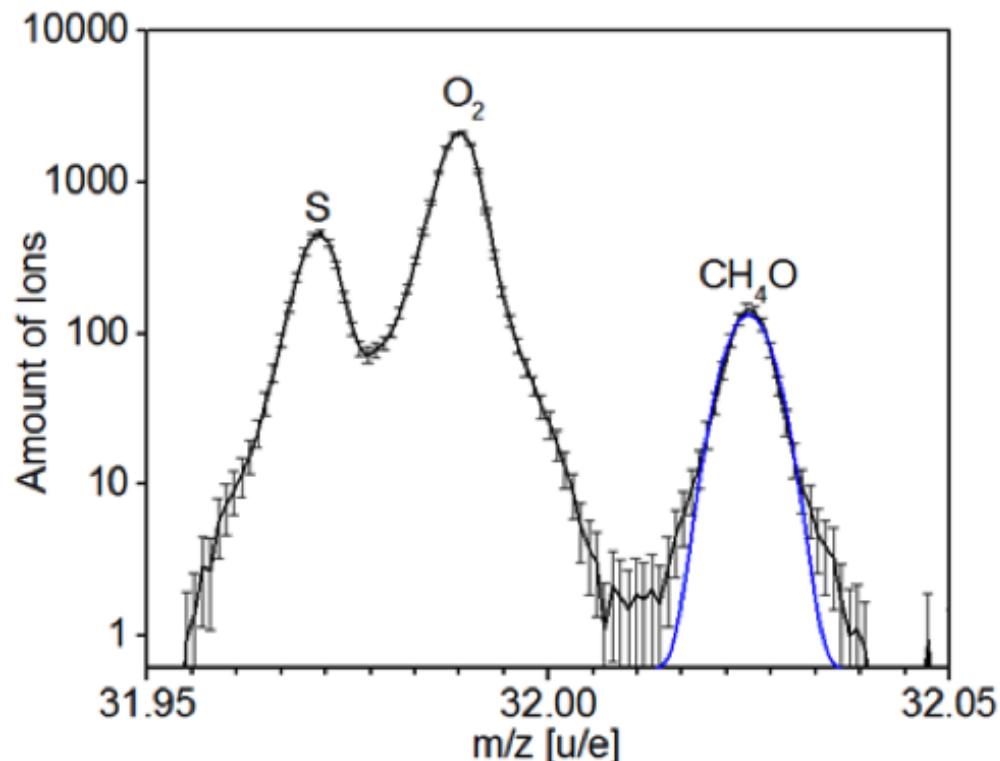
- **Rosetta Mission:** Mission to the Jupiter Family Comet (JFC) **67P/Churyumov-Gerasimenko, or 67P**
- Launched in March 2004 and made a 10-year journey across the solar system to arrive at its target in 2014
- **Philae lander:** Spacecraft landed on **Comet Nucleus**
- Philae lander carried **mass spectrometer** called ROSINA
 - Measured spectrum of comet ‘dust’
 - First detailed, *in situ* measurements of the chemical composition of a comet’s atmosphere, or coma
 - Analyses the various isotopes of atoms such as xenon and rare organic molecules, including sulfur-containing compounds



Complex Comet Chemistry!

Special Case: the Rosetta Mission to JFC 67P

- Mass spectrometer called ROSINA – Rosetta Orbiter Spectrometer for Ion and Neutral Analysis
- Double Focusing Mass Spectrometer (DFMS) designed to detect the cometary volatiles originating from both the cometary bulk material and the ejected particle
 - Uses electron impact ionization to ionize very low energy cometary molecules
 - Molecules separated according to their mass-to-charge ratios (m/z)
- Lots of O_2 – never detected before in comets!

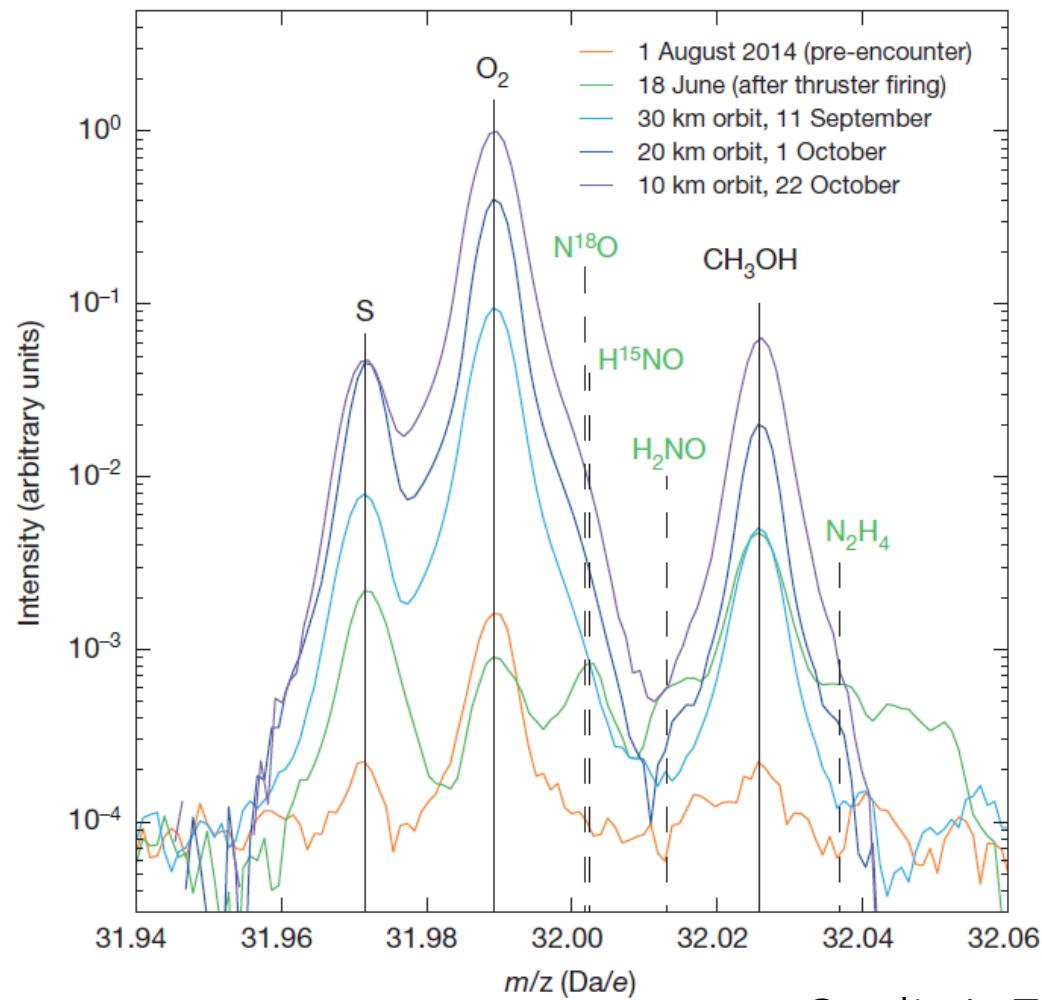


Schuhmann et al., 2019

Complex Comet Chemistry!

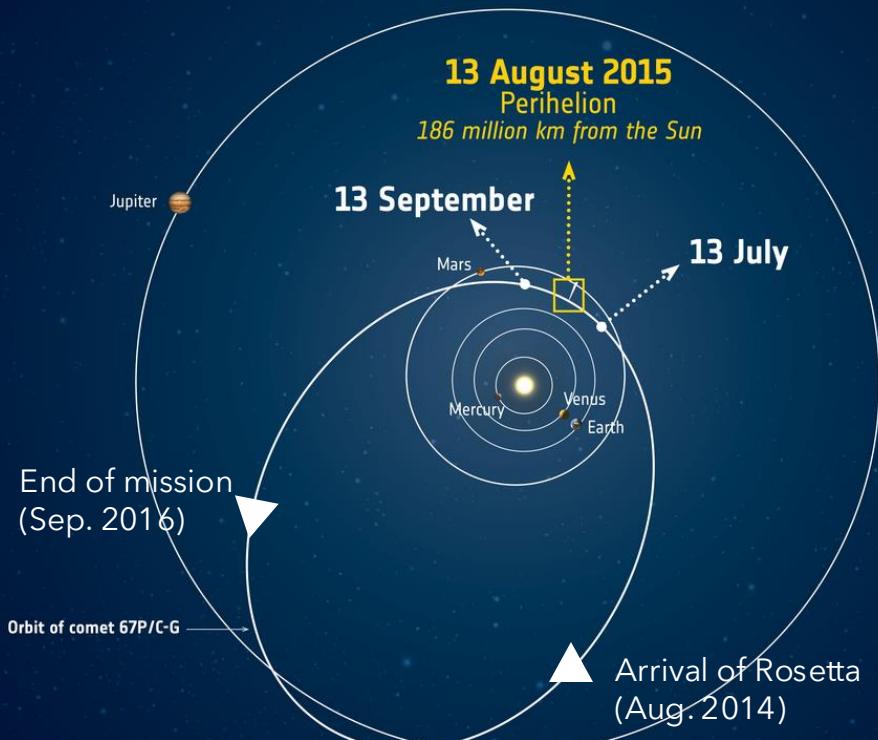
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- Lots of O_2 – never detected before in comets!
 - Dynamical → Closer to the sun enhanced desorption of species



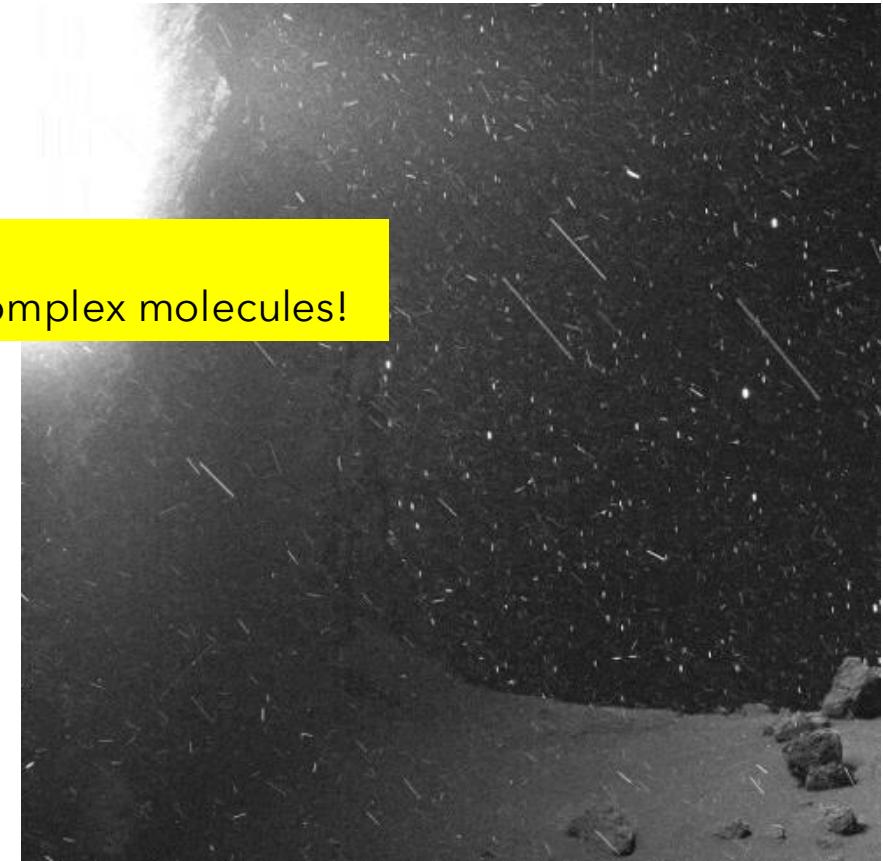
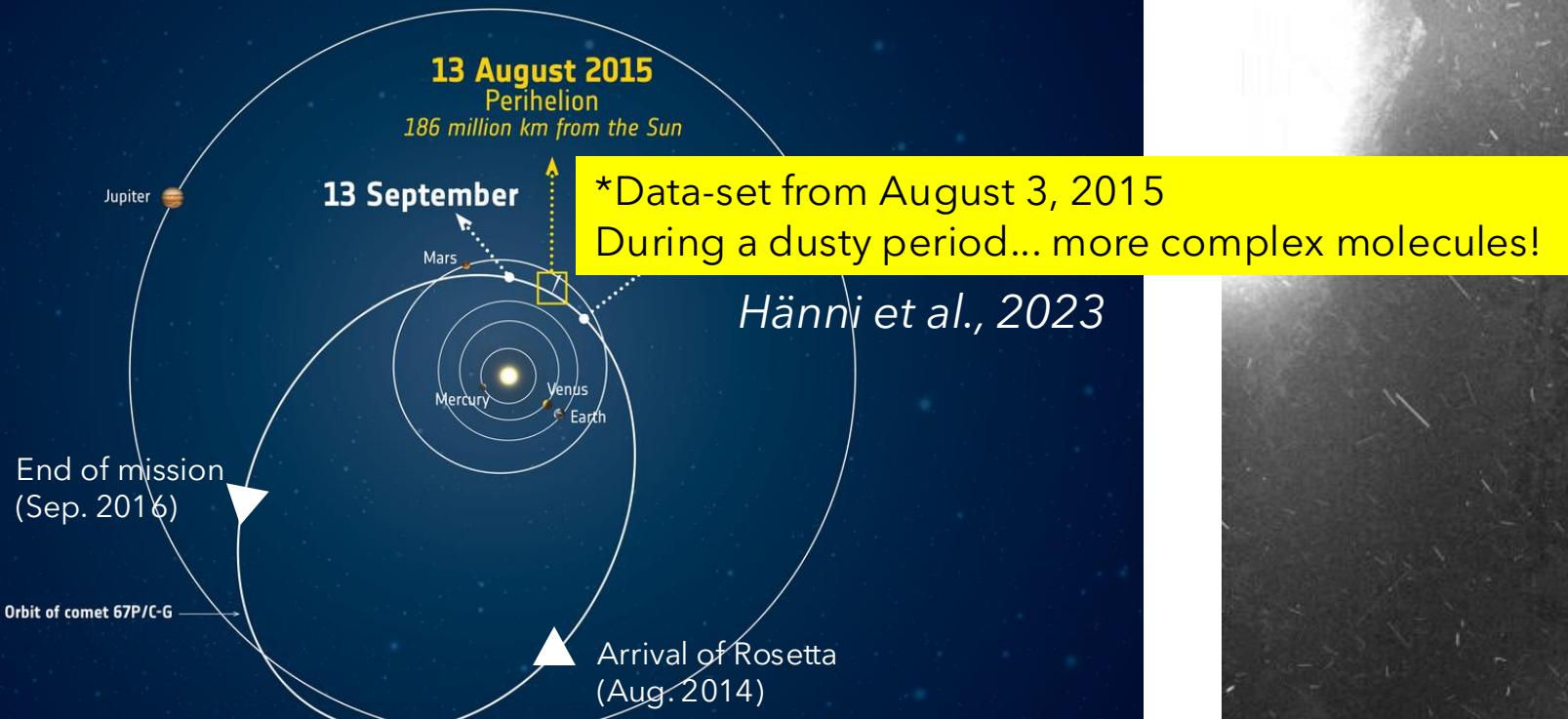
Credit: L. Ziurys

→ COMET 67P/CHURYUMOV–GERASIMENKO
AROUND PERIHELION



landru79 on Twitter/European Space Agency

→ COMET 67P/CHURYUMOV–GERASIMENKO
AROUND PERIHELION

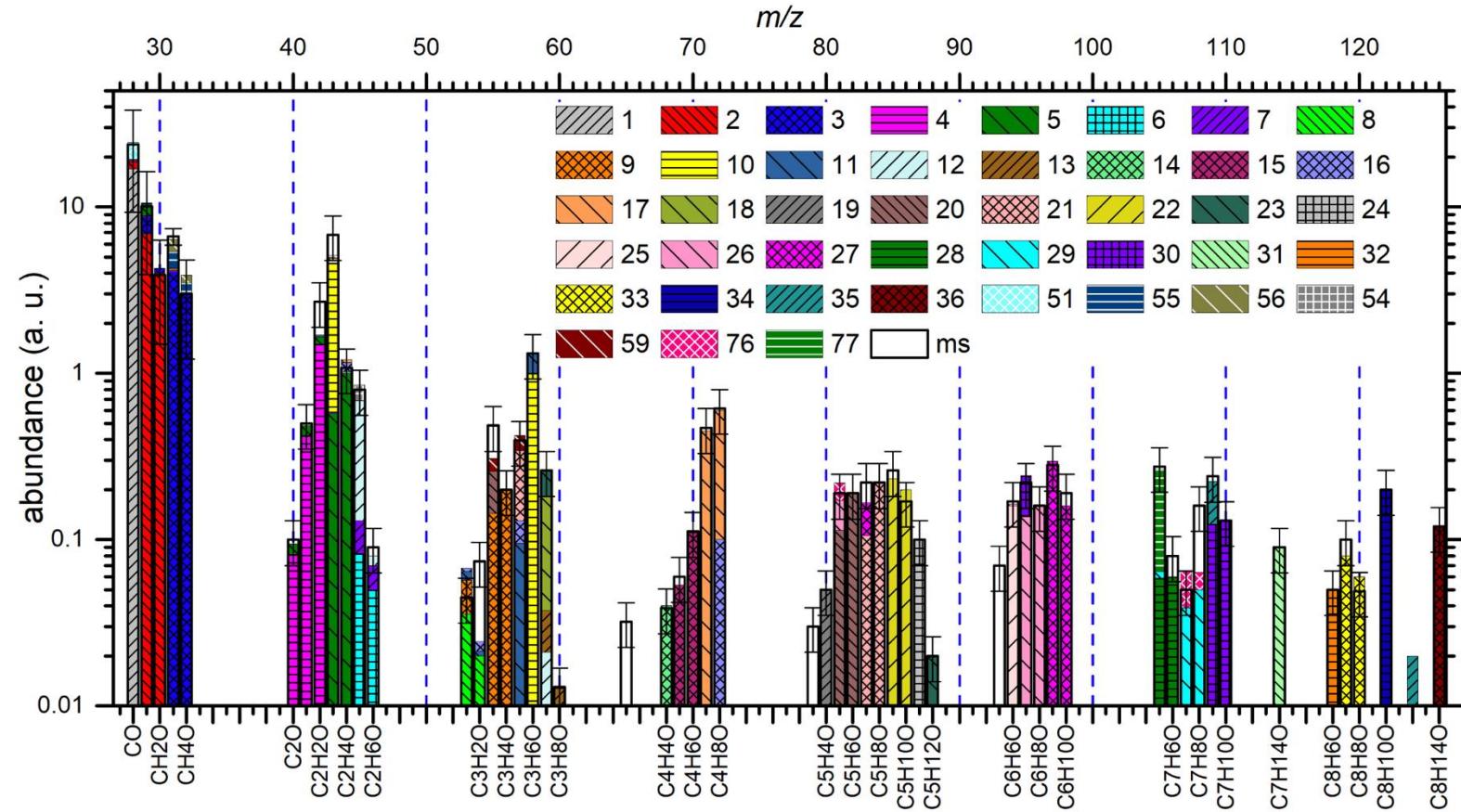


landru79 on Twitter/European Space Agency

Complex Comet Chemistry!

Special Case: the Rosetta Mission to JFC 67P

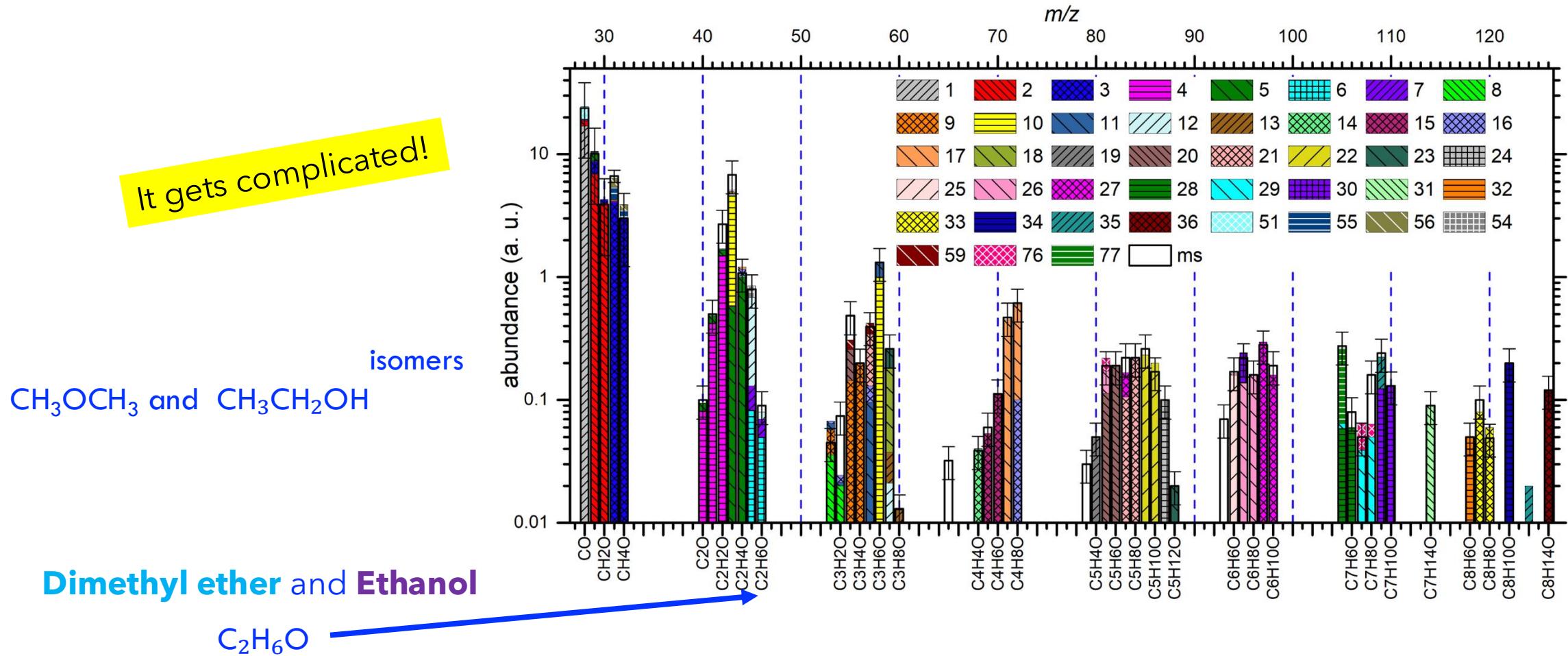
- Double Focusing Mass Spectrometer (DFMS) technique again used, sorting molecules by their **mass-to-charge ratios (m/z)**
- This new study derive a minimum of 63 $C_nH_mO_x$ candidate molecules!
- Though large oxygen bearing species may have similar configurations/masses (i.e., **isomers** CH_3OCH_3 and CH_3CH_2OH), the authors use probability (Occam's razor) to predict what molecules are likely present



Hänni et al., 2023

Complex Comet Chemistry!

Special Case: the Rosetta Mission to JFC 67P



Hänni et al., 2023

→ THE COMETARY ZOO: GASES DETECTED BY ROSETTA



THE LONG CARBON CHAINS

Methane
Ethane
Propane
Butane
Pentane
Hexane
Heptane



THE AROMATIC RING COMPOUNDS

Benzene
Toluene
Xylene
Benzoic acid
Naphtalene



THE KING OF THE ZOO

Glycine (amino acid)



THE "MANURE SMELL" MOLECULES

Ammonia
Methylamine
Ethylamine



THE "POISONOUS" MOLECULES

Acetylene
Hydrogen cyanide
Acetonitrile
Formaldehyde



THE ALCOHOLS

Methanol
Ethanol
Propanol
Butanol
Pentanol



THE VOLATILES

Nitrogen
Oxygen
Hydrogen peroxide
Carbon monoxide
Carbon dioxide



THE TREASURES WITH A HARD CRUST

Sodium
Potassium
Silicon
Magnesium



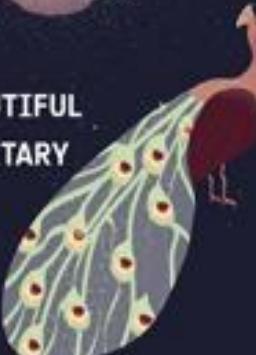
THE "SALTY" BEASTS

Hydrogen fluoride
Hydrogen chloride
Hydrogen bromide
Phosphorus
Chloromethane



THE BEAUTIFUL AND SOLITARY

Argon
Krypton
Xenon



THE "EXOTIC" MOLECULES

Formic acid
Acetic acid
Acetaldehyde
Ethylenglycol
Propylenglycol
Butanamide



THE MOLECULE IN DISGUISE

Cyanogen



→ THE COMETARY ZOO: GASES DETECTED BY ROSETTA



THE LONG CARBON CHAINS
Methane
Ethane
Propane
Butane
Pentane
Hexane
Heptane



THE AROMATIC RING COMPOUNDS
Benzene
Toluene
Xylene
Benzoic acid
Naphthalene



THE KING OF THE ZOO
Glycine (amino acid)



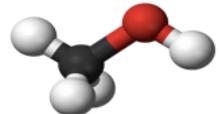
THE "MANURE SMELL" MOLECULES
Ammonia
Methylamine
Ethylamine



THE "POISONOUS" MOLECULES
Acetylene
Hydrogen cyanide
Acetonitrile

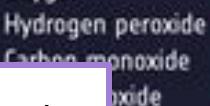


THE ALCOHOLS
Methanol
Ethanol



Methanol
 CH_3OH

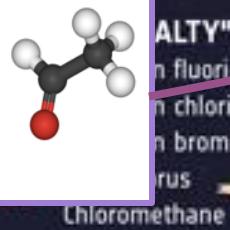
THE VOLATILES
Nitrogen
Oxygen
Hydrogen peroxide
Carbon monoxide
Cyanide



Rich in Complex Organic Molecules (COMs) also found in Star Forming Regions!

T
A
S
P
S
Magnesium

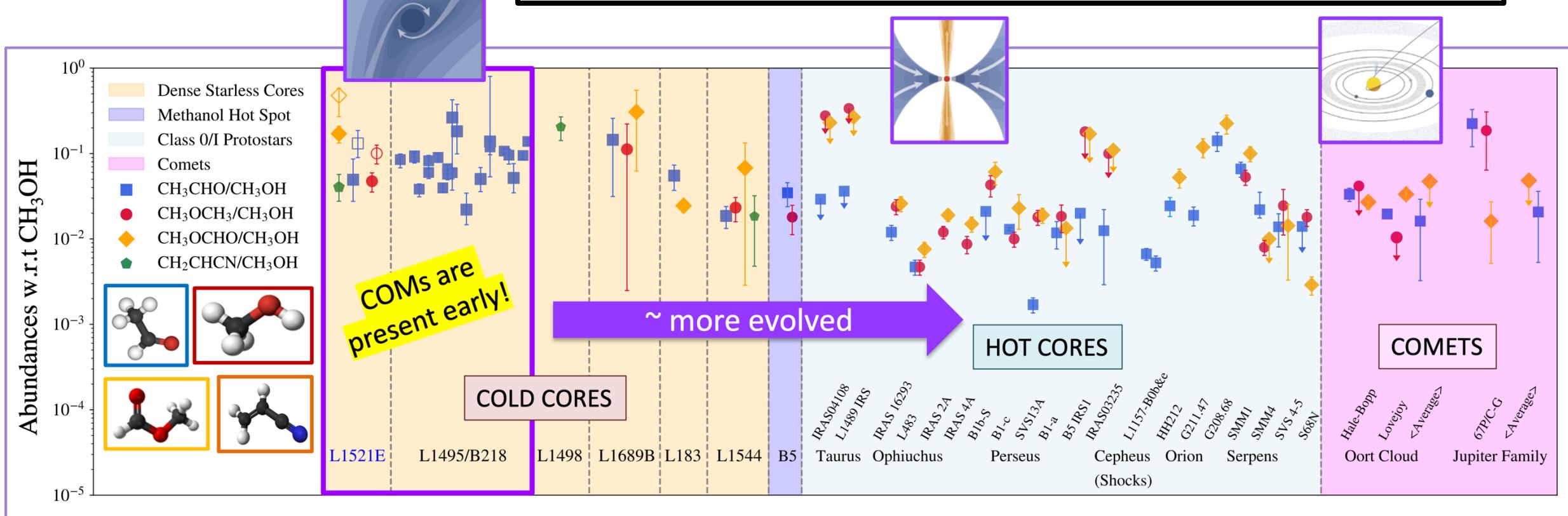
Acetaldehyde
 CH_3CHO



Taurus Molecular Cloud

Complex Comet Chemistry - Link to Star and Planet Formation!

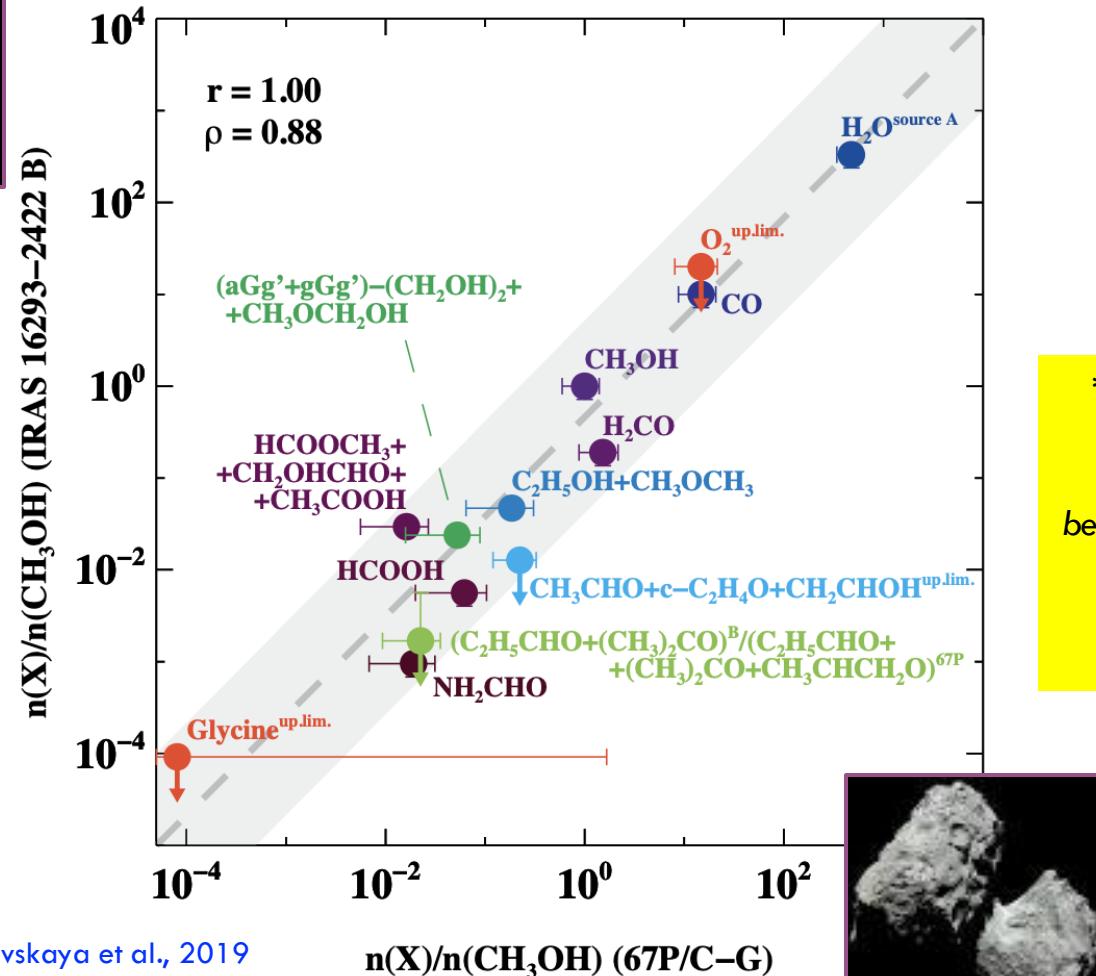
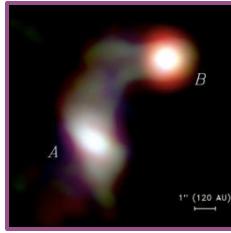
The general agreement in COM abundances across all stages suggests at least some of the COMs formed in starless and prestellar cores are being *inherited* to later stages of low-mass star and planet formation!



[L1495] Scibelli & Shirley 2020, [L1498] Jiménez-Serra et al. 2021, [L183] Lattanzi et al. 2020, [L1544] Jimenez-Serra et al. 2016, [L1689B] Bacmann et al. 2012, [B5] Taquet et al. 2017, [B1-c, B1-bs and S68N] van Gelder et al. 2020, [IRAS2A, IRAS4A] Taquet et al. 2015, López-Sepulcre et al. 2017; [IRAS 16293-2422] Jaber et al. 2014, Jørgensen et al. 2018, [SVS13A] Bianchi et al. 2019, [IRAS03245, B1-a, B5, IRS 1, SVS 4-5, IRAS 04108, L1489 IRS], Graninger et al. 2016, Bergner et al. 2017, [SMMI, SMM4], Lee et al. 2019, Hsu et al. 2020 [HH212, G2II.47, G208.68], Öberg et al. 2011, Taquet et al. 2015, [L483] Jacobsen et al. 2019, [L157-Boe L157-Bob] Codella et al. 2020, [B335] Imai et al. 2016., [Hale-Bop and Lovejoy] Biver & Bockelée-Morvan 2019, [67P/ChuryumovGerasimenko(67/C-G)] Schuhmann et al. 2019 & Rubin et al. 2019

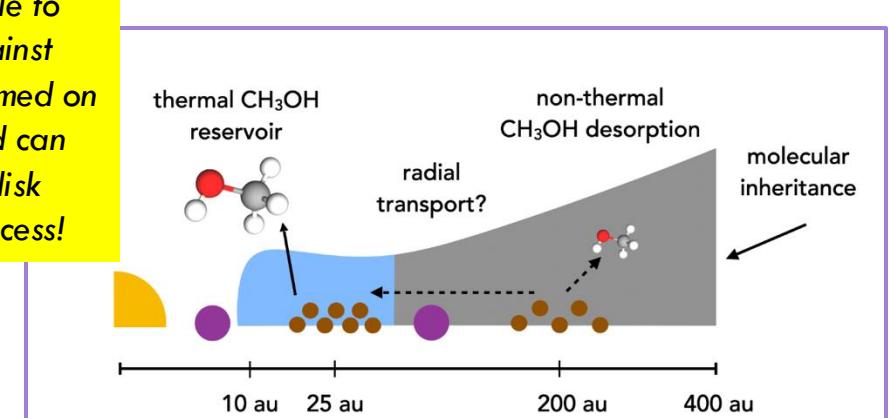
Adapted from Scibelli et al., 2021

Complex Comet Chemistry - Link to Star and Planet Formation!



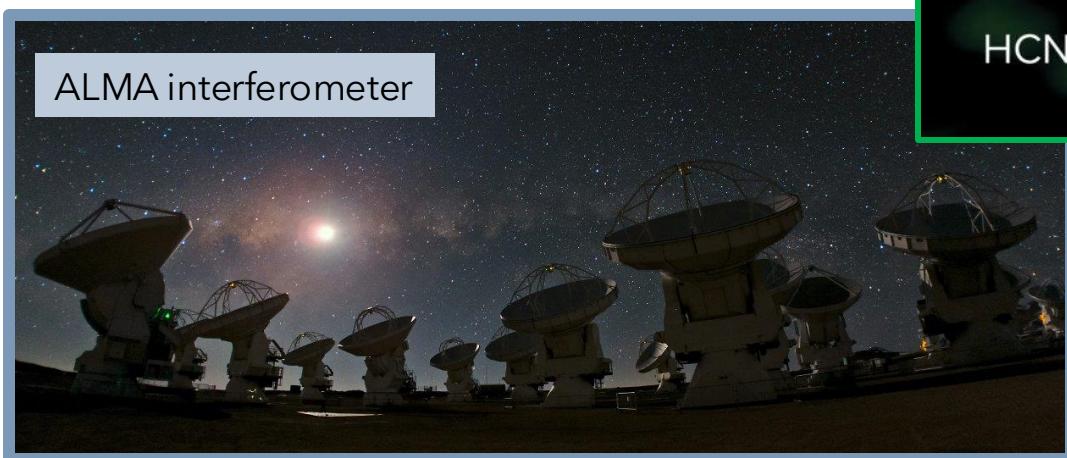
← Additional evidence that molecules observed in protostars and comets could be inherited from the cold **prestellar core** phase!

*CH₃OH is a good tracer molecule to compare against because it is formed on the grains and can survive the disk formation process!

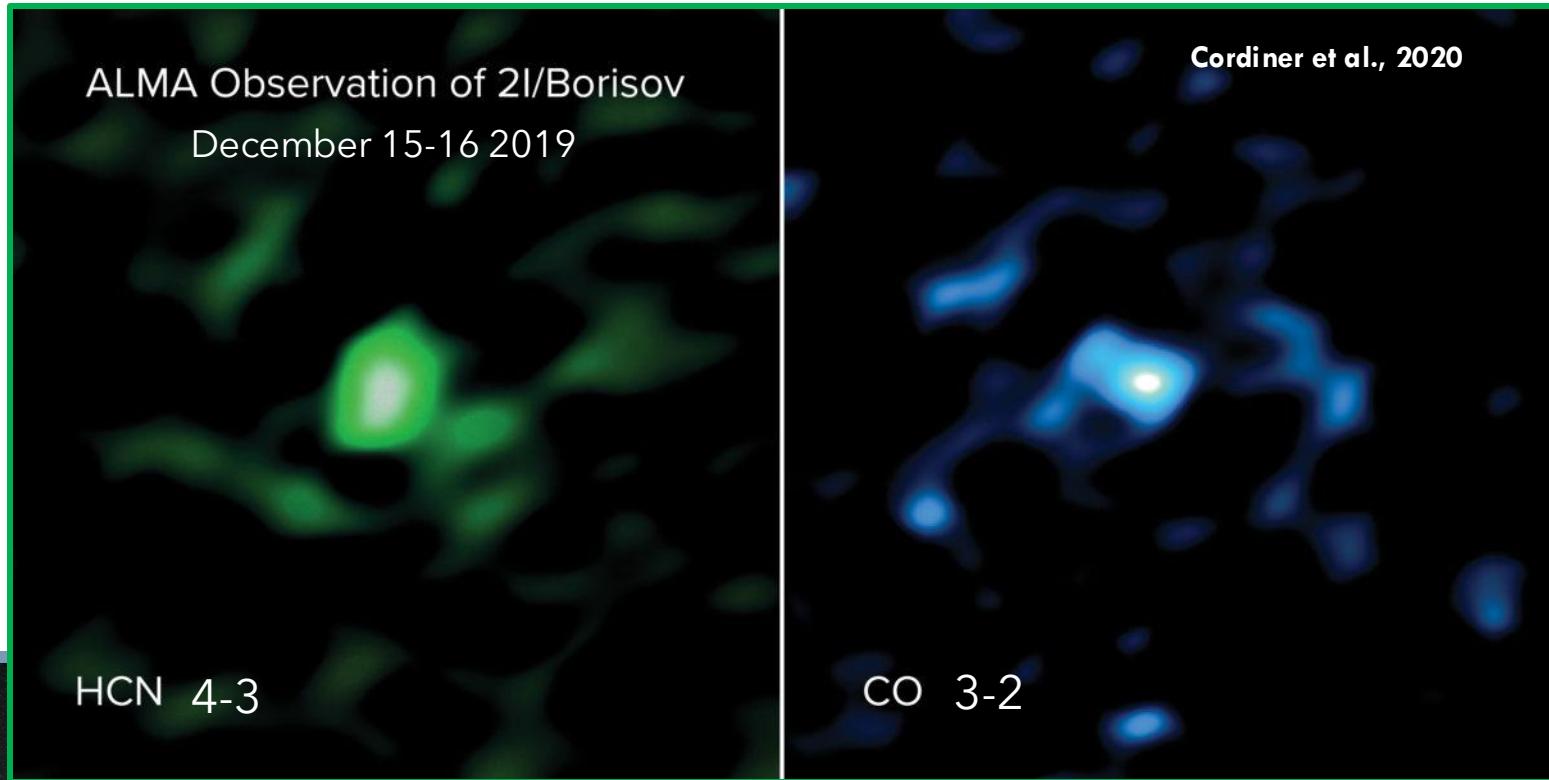


Interstellar Comet Chemistry!

- The first confirmed **interstellar** comet!
- The **HCN** abundance similar to that of comets in our Solar System
- The **CO** abundance is among the highest observed in any comet within 2 au of the Sun!
- 2I/Borisov must have formed in a relatively CO-rich environment in the very cold, outer regions of a distant protoplanetary accretion disk (similar to our our proto-Kuiper belt)



ALMA interferometer



Maps from rotational line spectra

Emergence of Life on Earth, or *Elsewhere*?

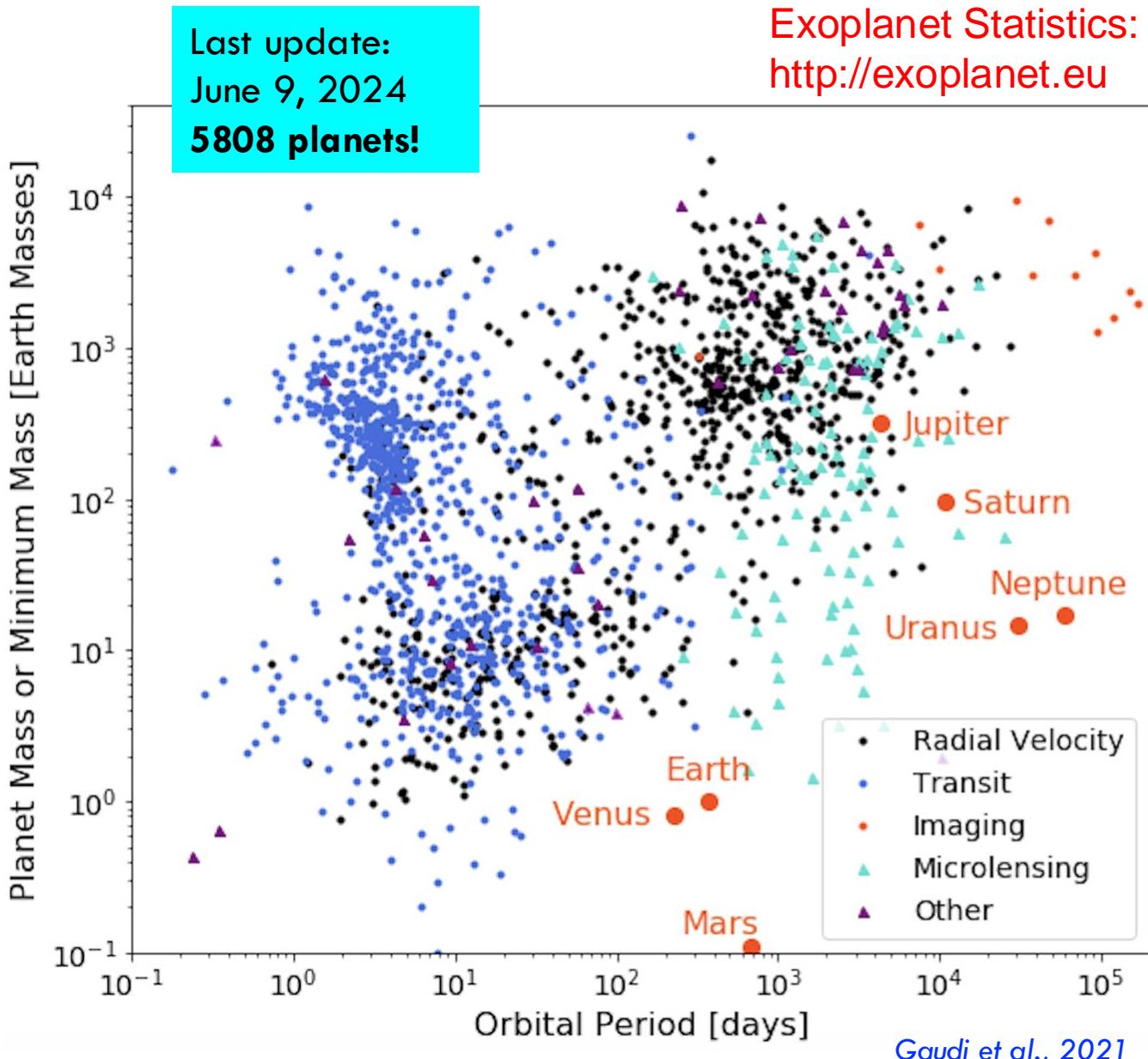
- Some of the complex chemistry formed in molecular clouds are inherited to the later stages of star and planet formation, including comets formed in the debris disks of solar systems, that can then seed the chemistry needed for life on Earth or other planets



EquinoxGraphics.net

Exoplanets

Exoplanet → a **planet**, which is defined as a gravitationally bound body that is not massive enough for nuclear fusion, but large enough to become round under the influence of its own gravitation, **around another star**



Exoplanets

- **Exoplanets typically grouped by size, mass, orbital period as well as atmospheric composition, which allows for detailed studies of the planet's chemistry**



Gas Giants

The size of Saturn or Jupiter, or much larger. They include "hot Jupiters" - scorching planets in close orbits around their stars.

Exoplanet Types



Neptune-Like

Similar in size to our own Neptune and Uranus, with hydrogen or helium-dominated atmospheres. "Mini-Neptunes," not found in our solar system, are smaller than Neptune but larger than Earth.



Super-Earth

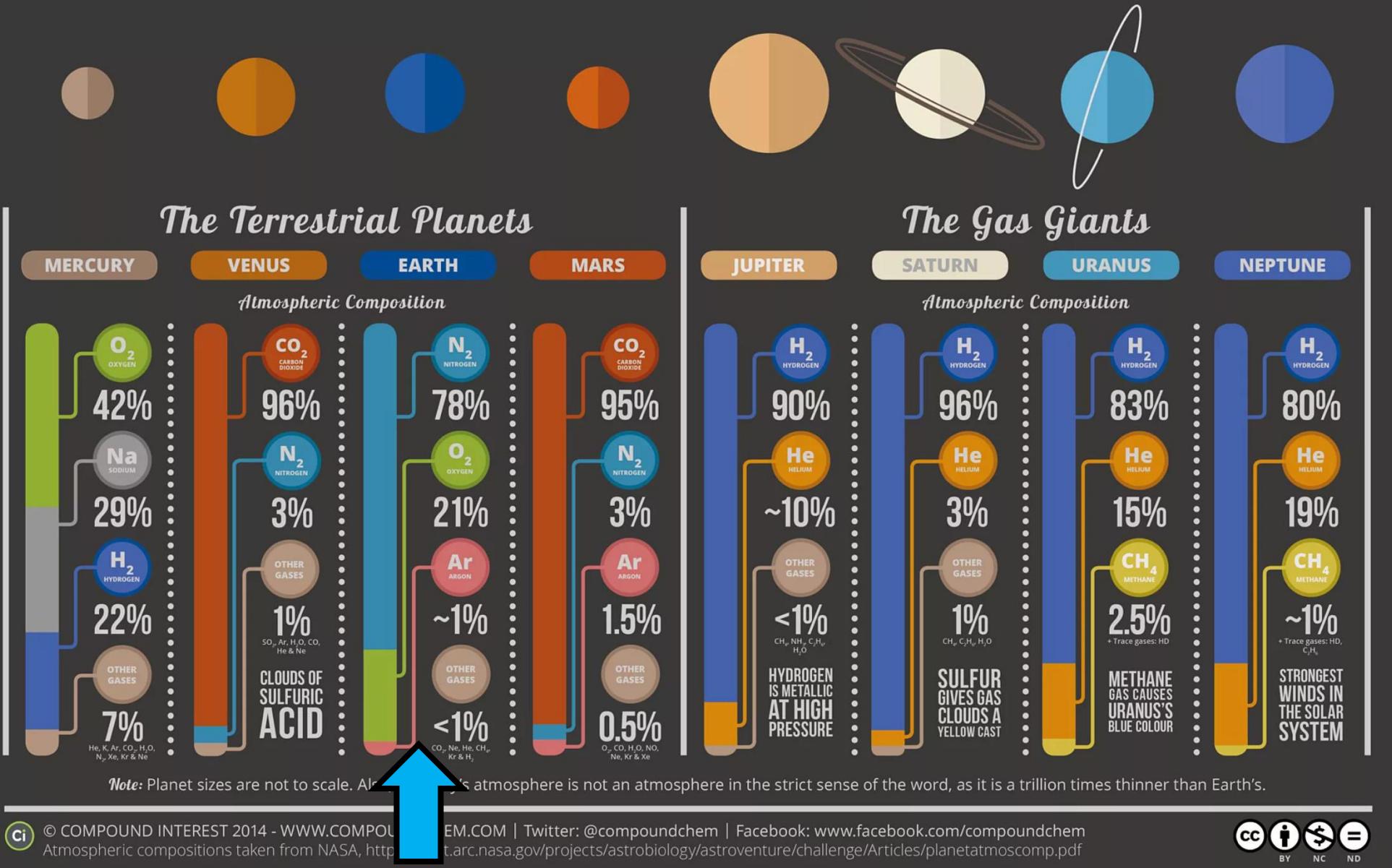
Typically "terrestrial," or rocky, and more massive than Earth but lighter than Neptune. They might or might not have atmospheres.



Terrestrial

Earth-sized or smaller, mostly made of rock and metal. Some could possess oceans or atmospheres and perhaps other signs of habitability.

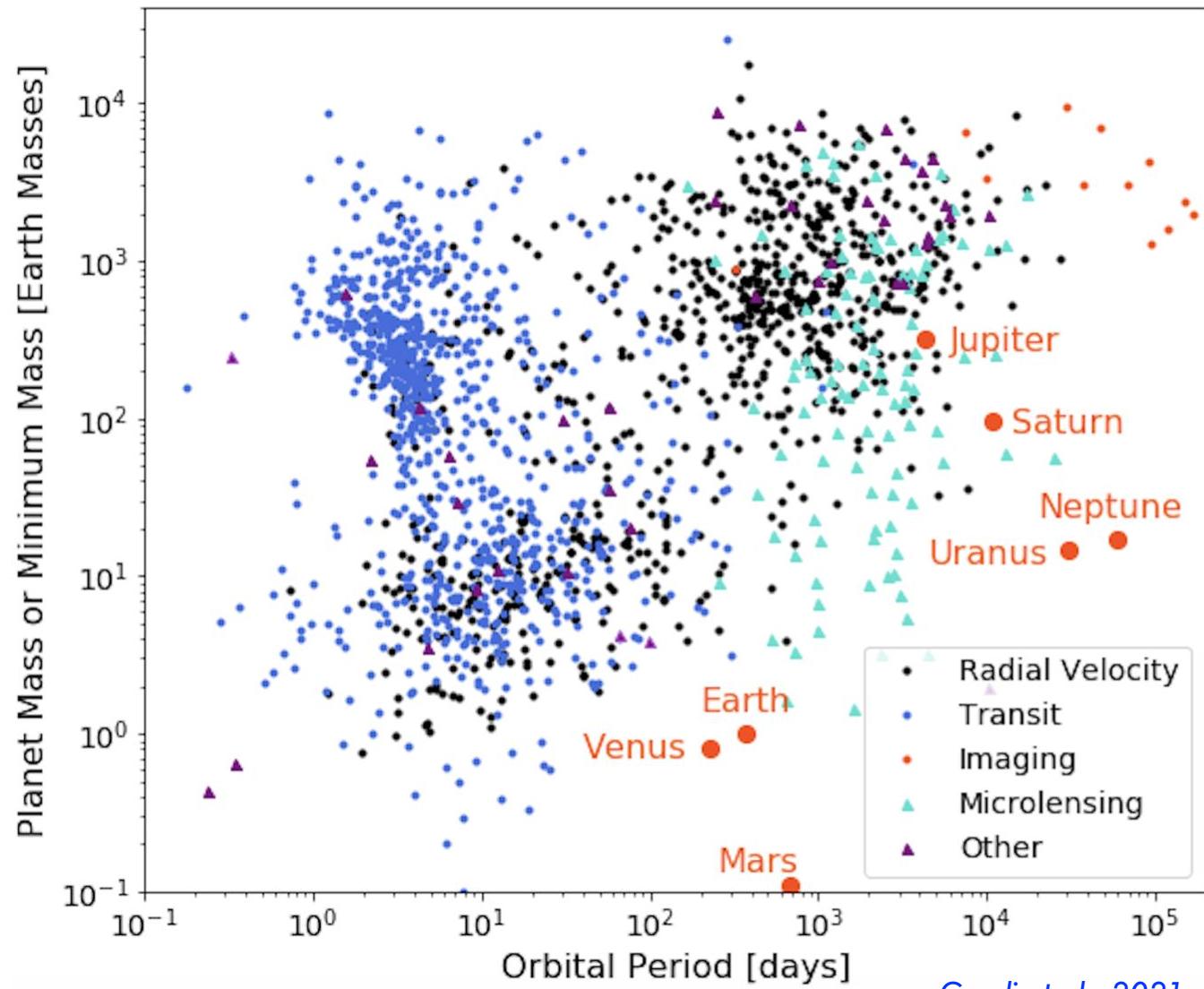
ATMOSPHERES OF THE SOLAR SYSTEM



How does the composition of atmospheres of solar system objects (Earth in particular!) compare to those of exoplanets?

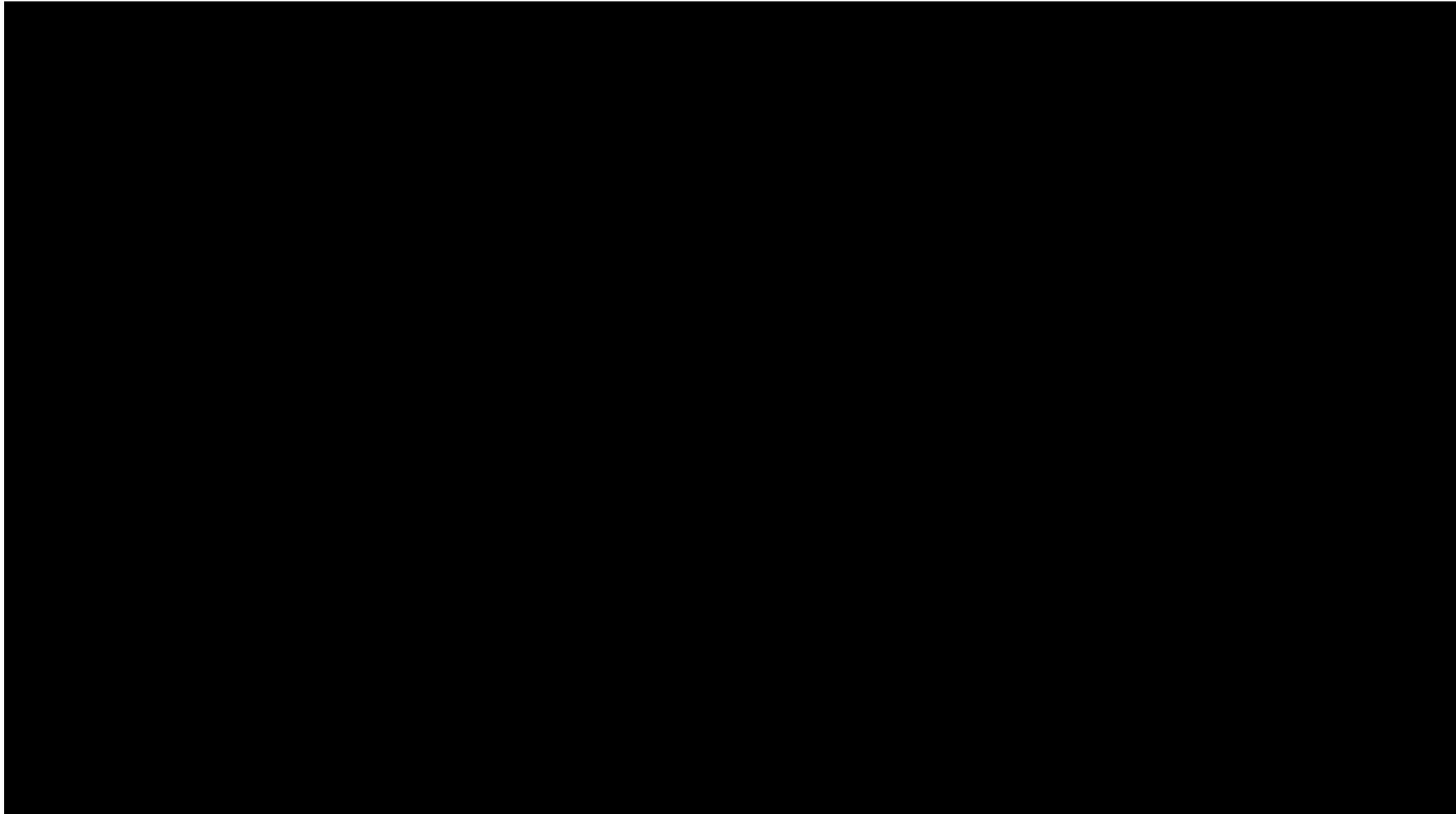
Studying the Atmosphere of Exoplanets: Transit Spectroscopy

- Most planets with relatively short orbital periods discovered by the **transit method** (in blue →)
- When a planet crosses directly between us and its star, we see the star dim slightly because the planet is blocking out a portion of the light

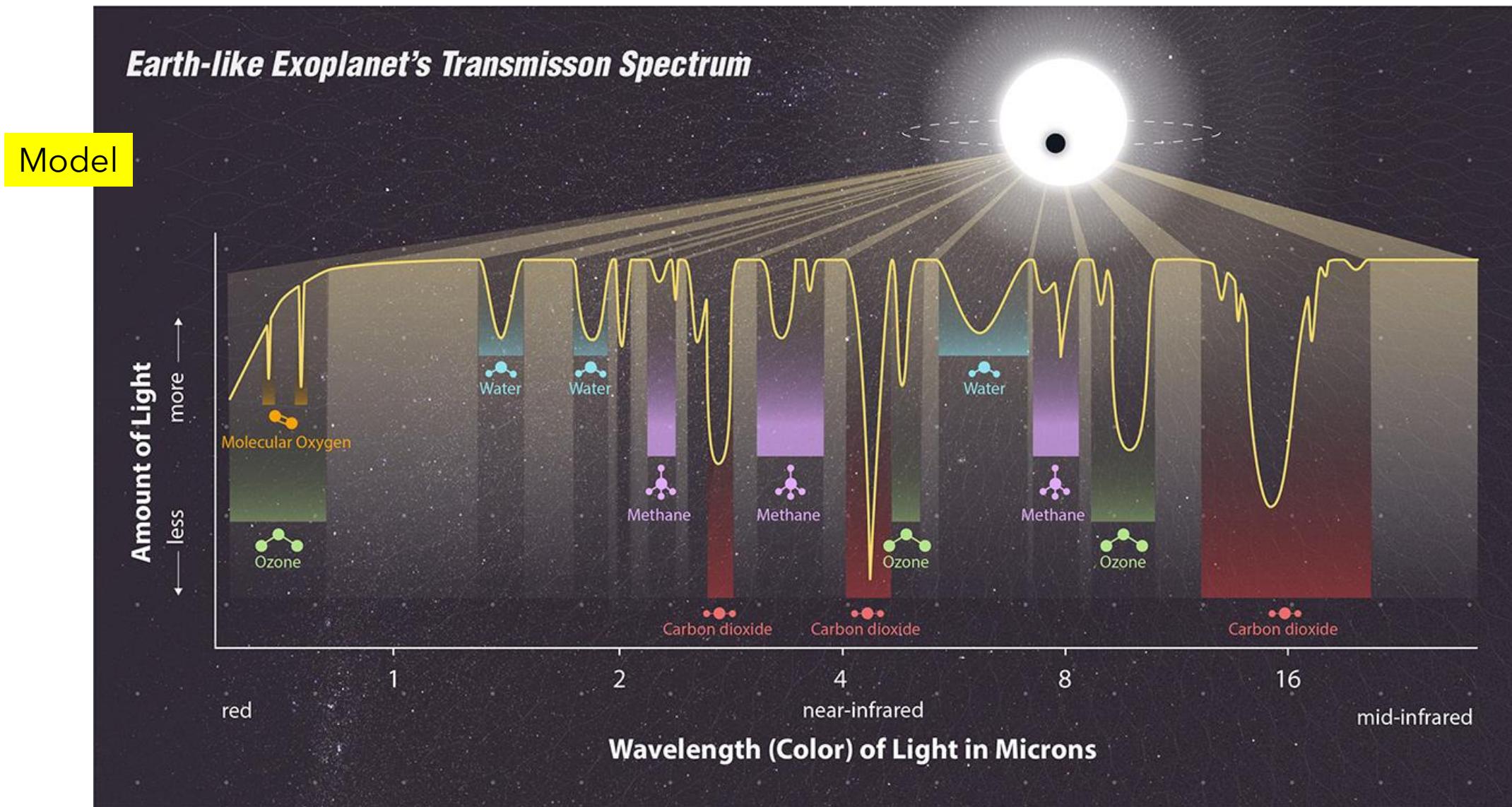


Gaudi et al., 2021

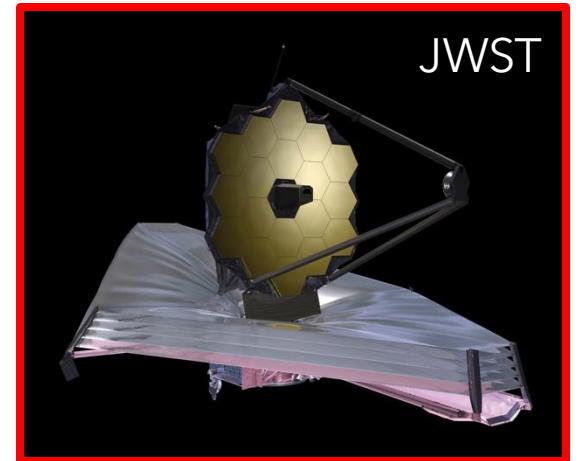
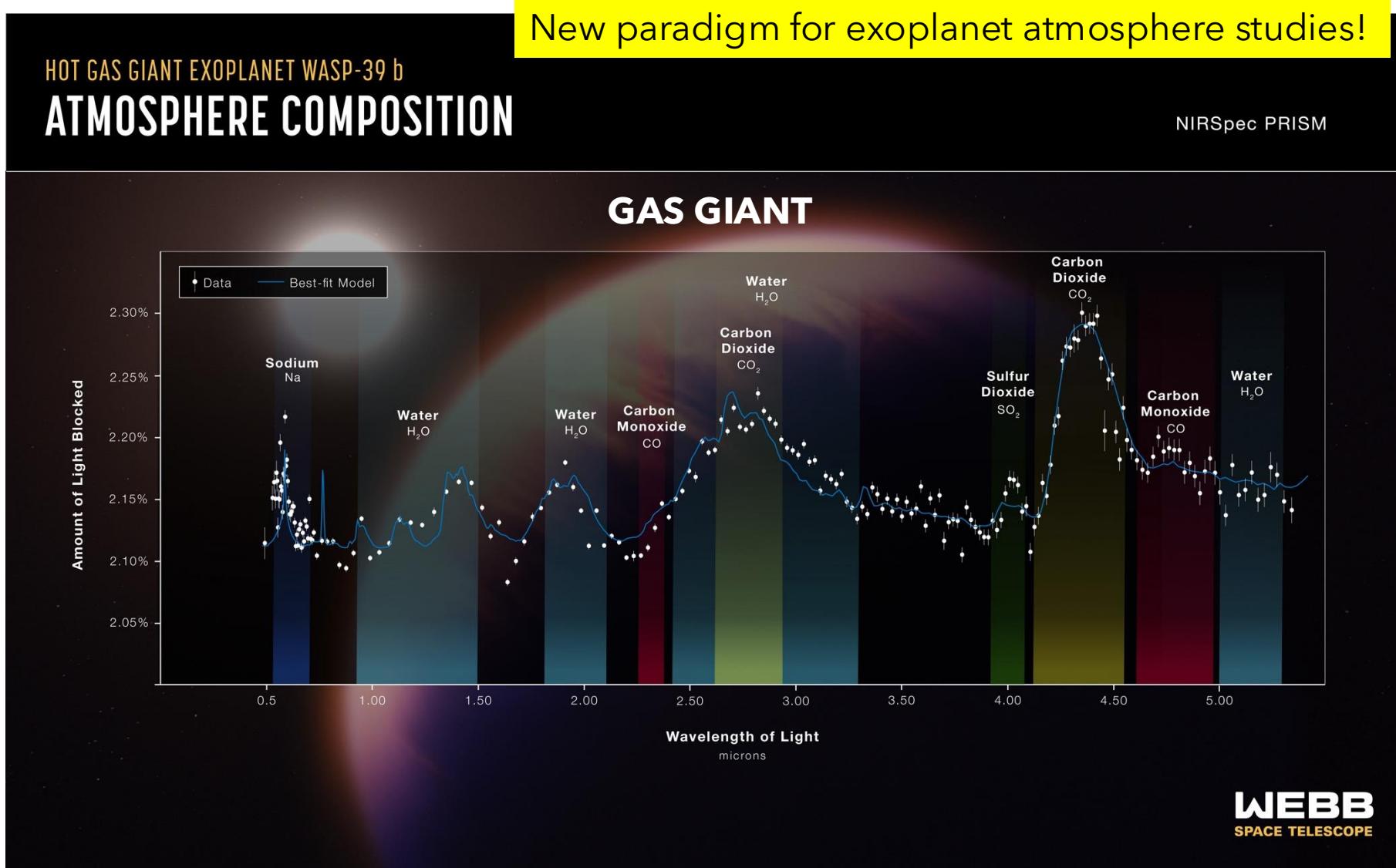
Studying the Atmosphere of Exoplanets: Transit Spectroscopy



Studying the Atmosphere of Exoplanets: Transit Spectroscopy



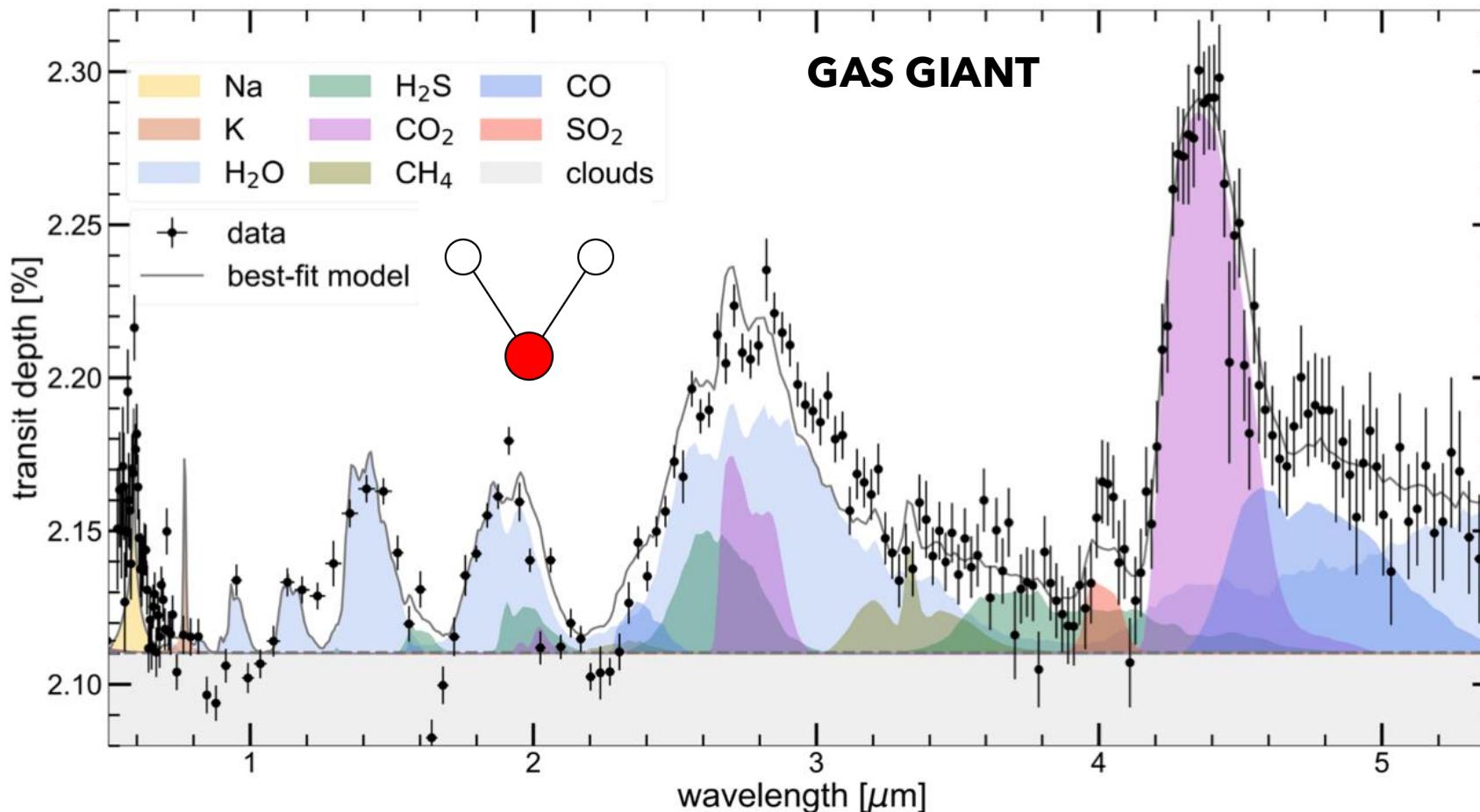
Studying the Atmosphere of Exoplanets: Transit Spectroscopy



← Press release image

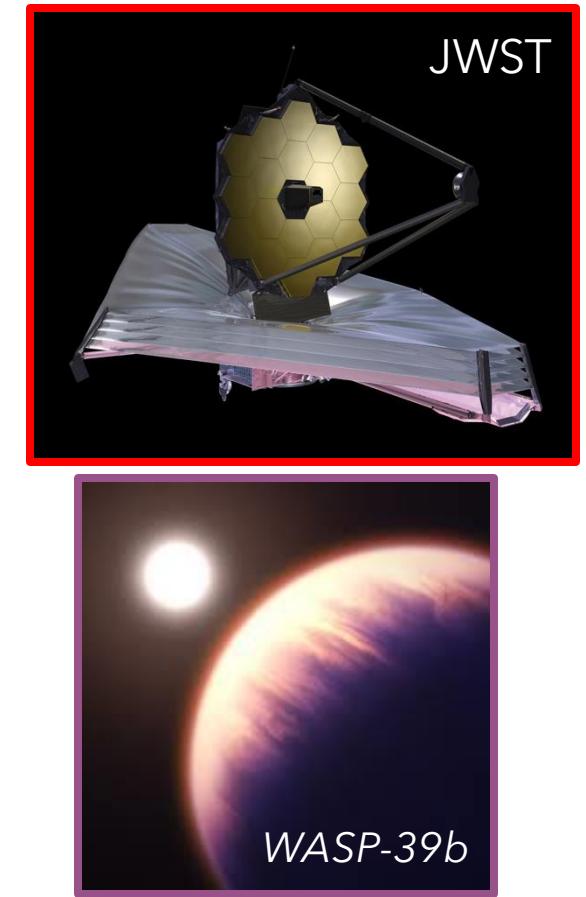
Studying the Atmosphere of Exoplanets: Transit Spectroscopy

New paradigm for exoplanet atmosphere studies!



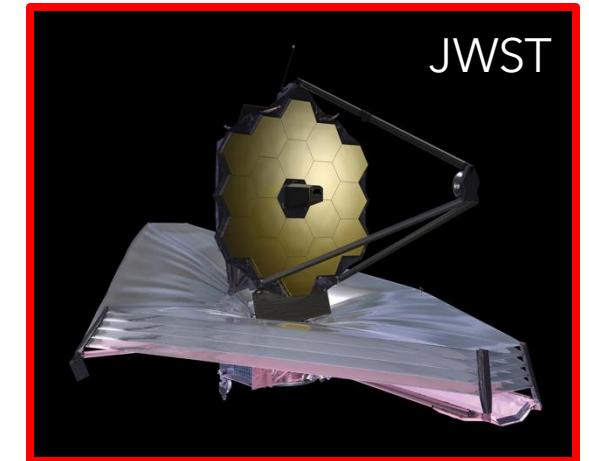
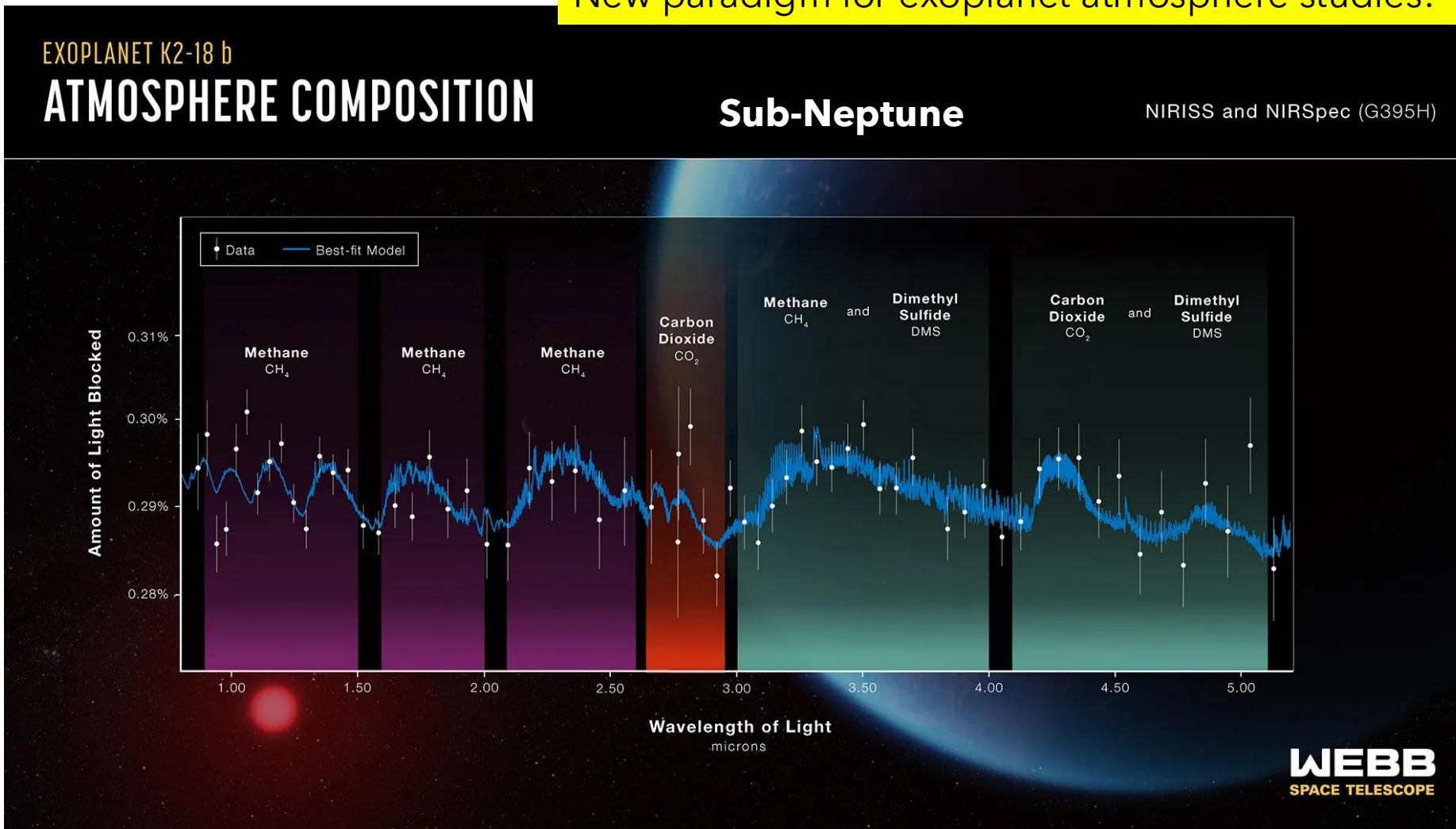
Rustamkulov et al., 2022

Overlapping IR Vibrational modes used to detect molecules!



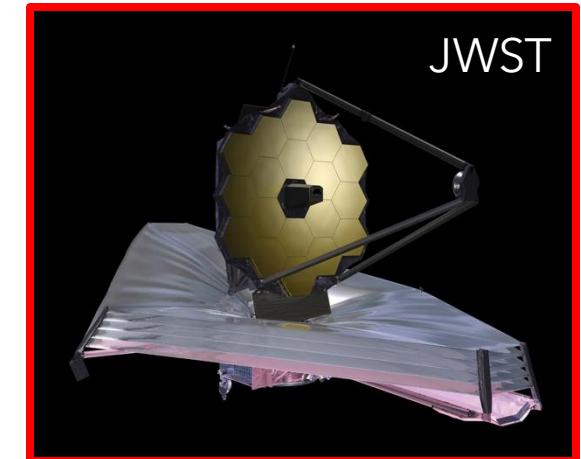
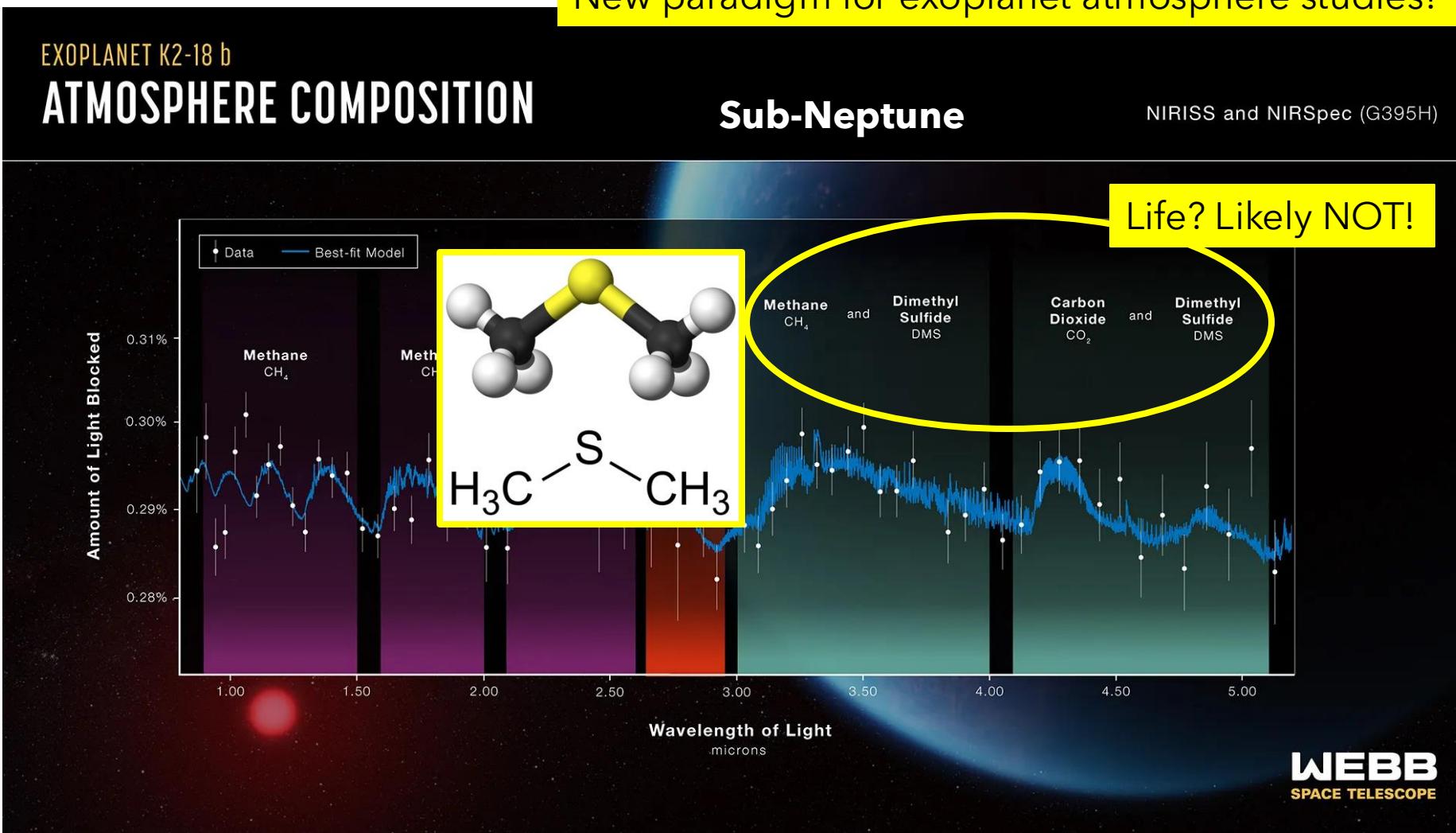
← Published Data

Studying the Atmosphere of Exoplanets: Transit Spectroscopy



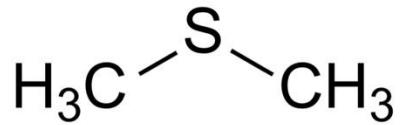
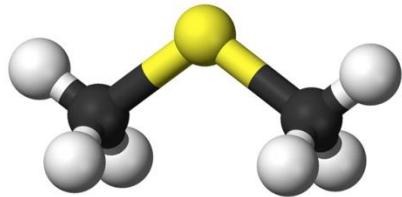
← Press release image

Studying the Atmosphere of Exoplanets: Transit Spectroscopy



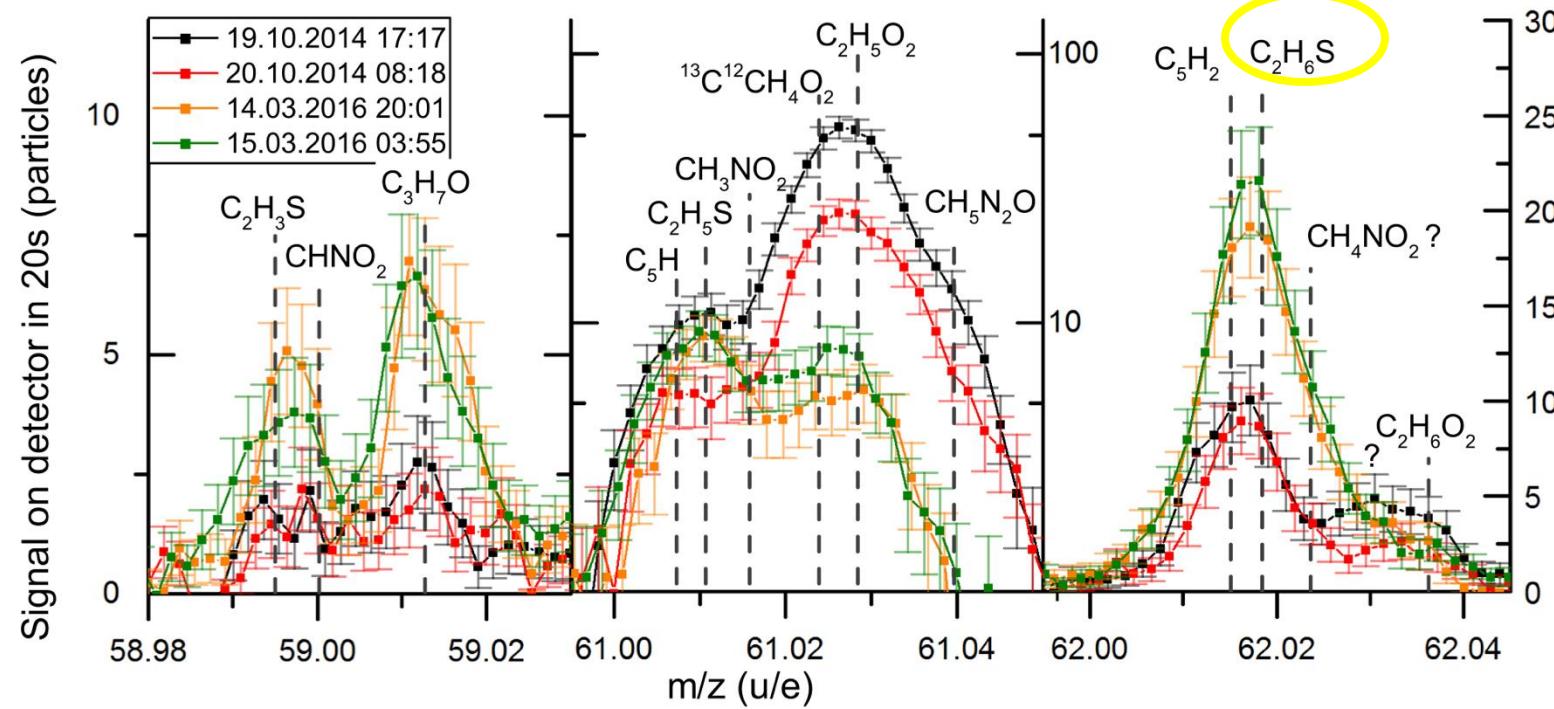
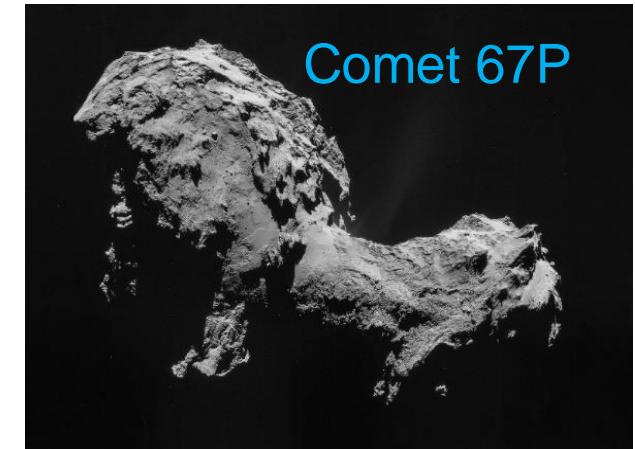
← Press release image

Dimethyl Sulfide

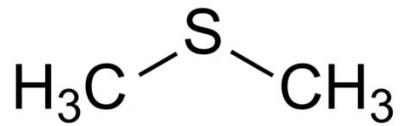
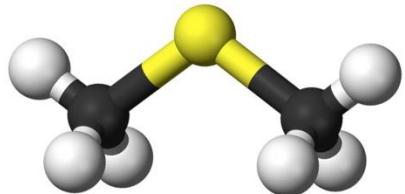


Also detected in 'lifeless' comet 67P!

- Though, with this method still cannot distinguish between the two isomers (C_2H_6S) **ethanethiol** and **dimethyl sulfide**!
- Updated analysis in 2024 claim to confirm its presence
([Hänni et al.,](#))

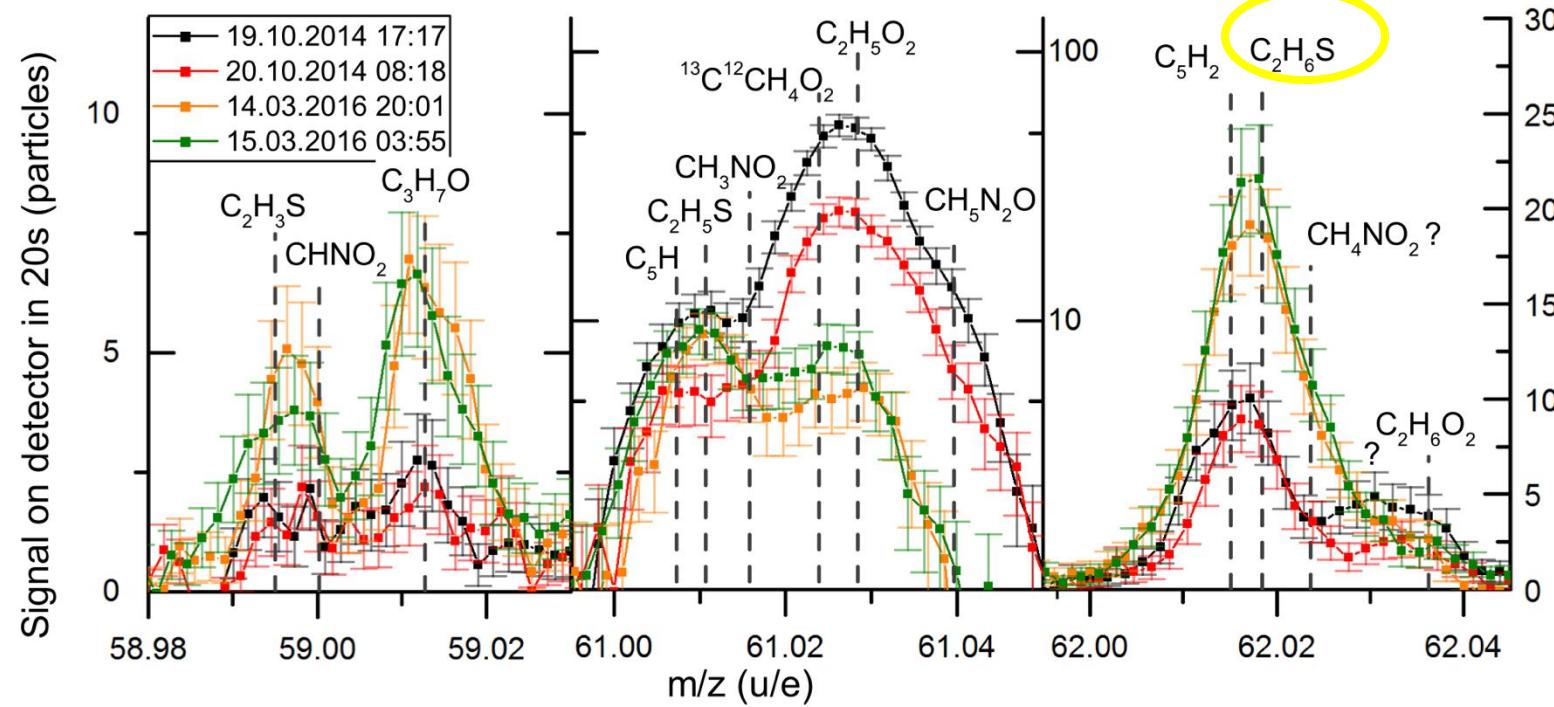


Dimethyl Sulfide

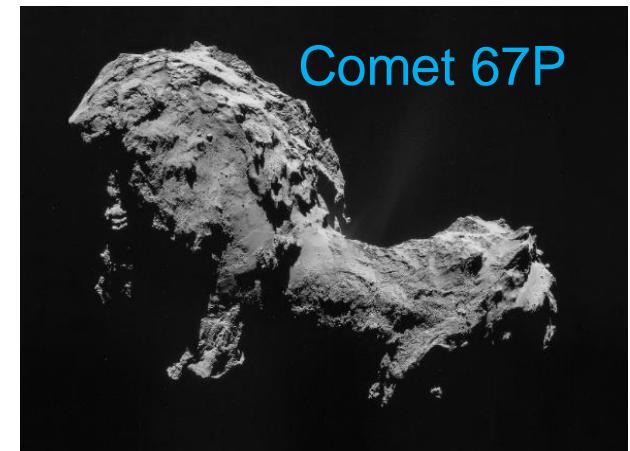


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Calmonte et al., 2016

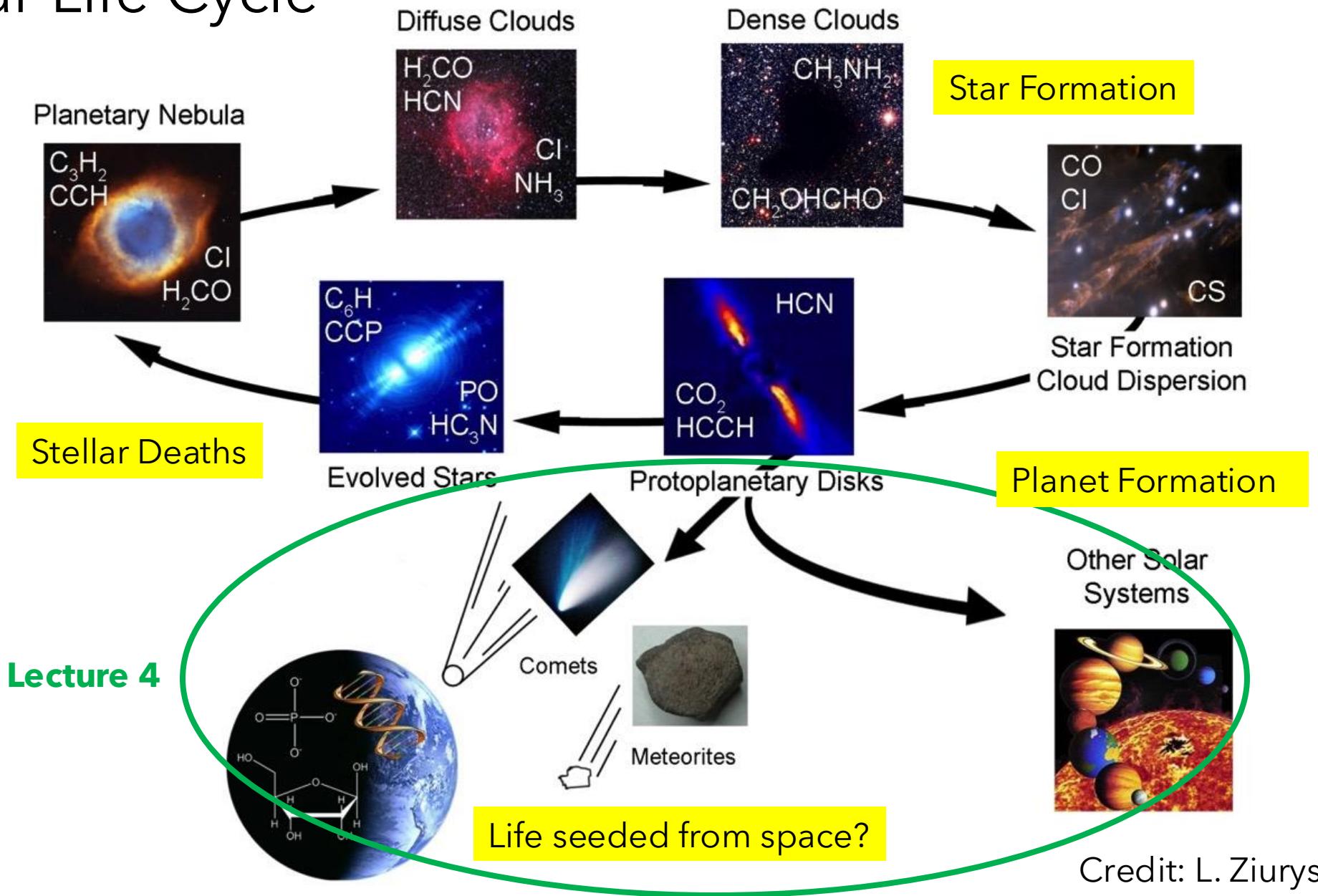


Comet 67P

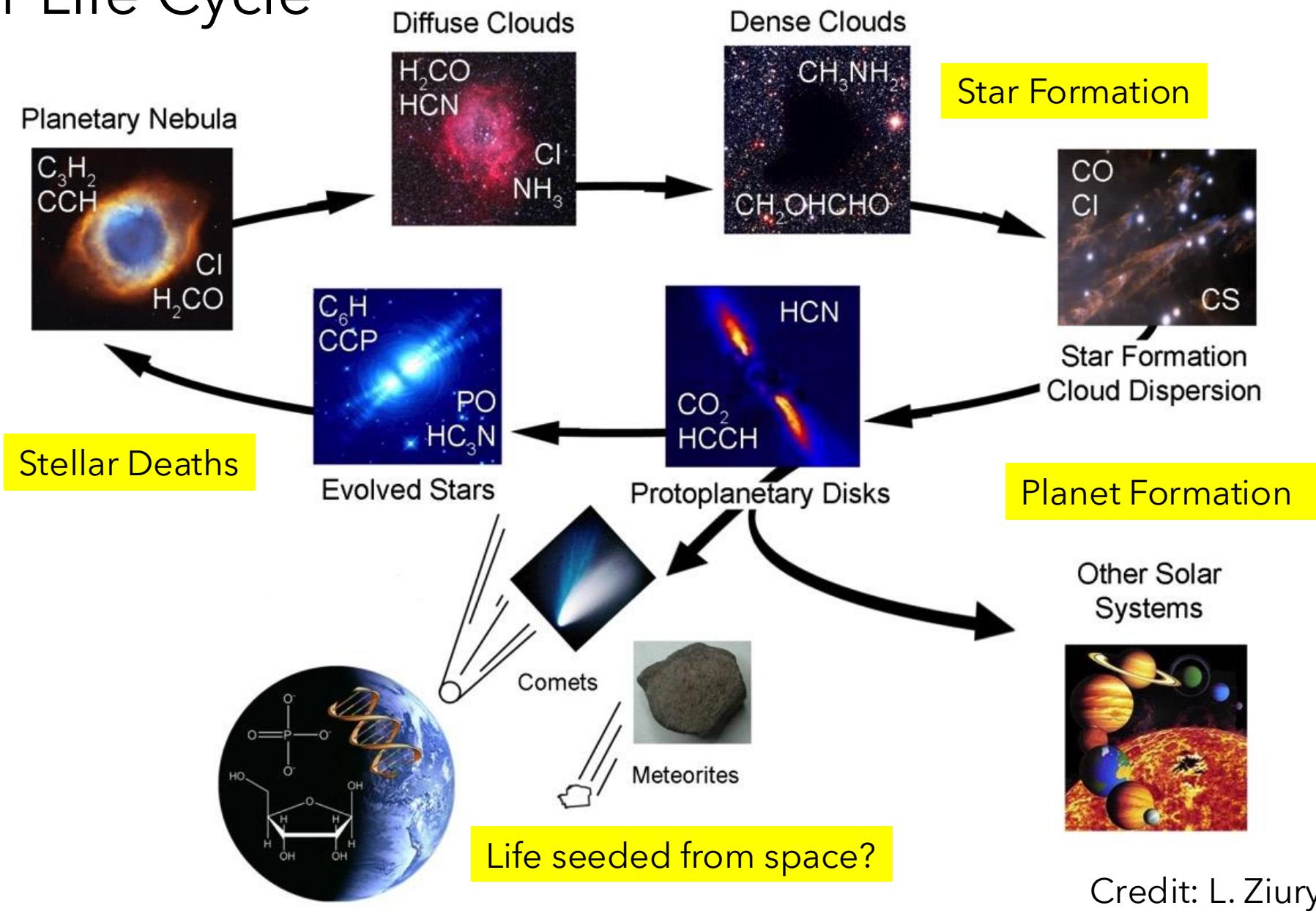
Molecular 'signs of life' have often already been detected in the cold depths of space!

Molecular Life Cycle

The **building blocks for life** on **Earth**, and potentially **other worlds**, were likely seeded from star and planet-forming material (such as **comets!**)



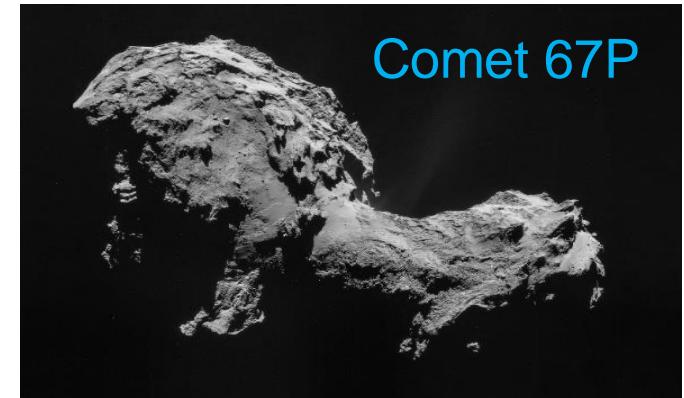
Molecular Life Cycle



Credit: L. Ziurys

SUMMARY:

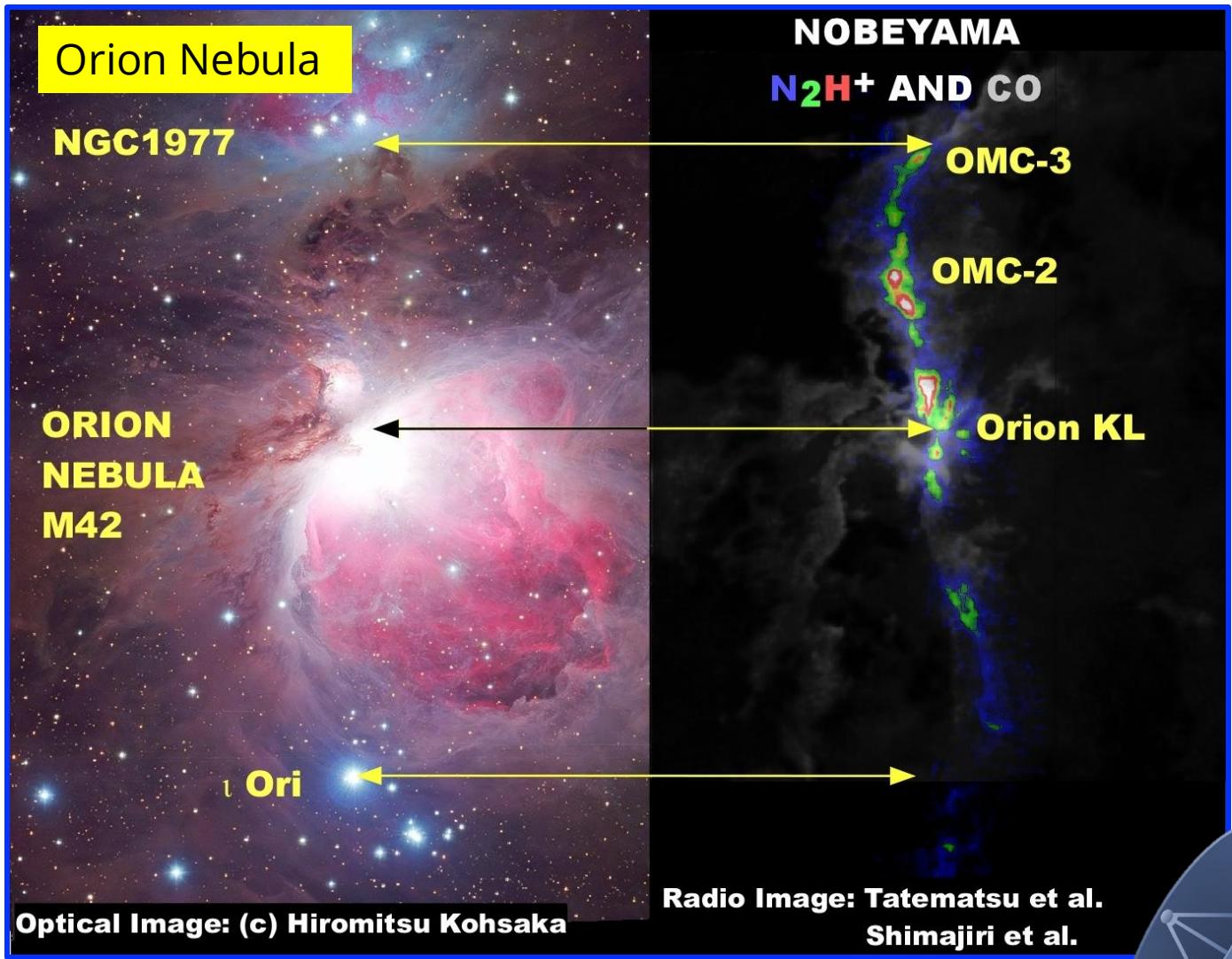
- Reminder! **Submillimeter and millimeter radio telescopes are powerful instruments** that let observational astrochemists (like myself) study the **rotational spectra** of interstellar molecules in high detail!
- Cold molecular clouds are the birthplaces of stars and planets. Within molecular clouds, **H₂ forms on the surfaces of interstellar dust grains** and is released into the gas – this is the start of chemistry in the interstellar medium!
- Dust grains are formed in the circumstellar shells of evolved stars, specifically AGBs! Dust and molecular gas is transported outward in winds, enriching the interstellar medium! **Mass loss from Evolved stars supplies ~85% of the material in the ISM!** The dust and molecules enrich the surrounding **diffuse** gas and eventually the dense gas that goes on to form stars and planets!
- Comets are remnants from when solar systems formed and, **due to their dusty and hydrogen rich comas**, their chemistry can be studied at millimeter wavelengths. Comet 67P is an interesting case, as a fly-by mission was able to measure the composition of the coma via mass spectroscopy. **The abundance of the complex molecules seen in comets is like that seen in other stages of star formation, suggesting inheritance.**
- With the advent of JWST, **exoplanet atmospheres are beginning probed in exquisite detail**. **Do not, however, be fooled by ‘signs of life’ from molecules often detected in the cold depths of space!**





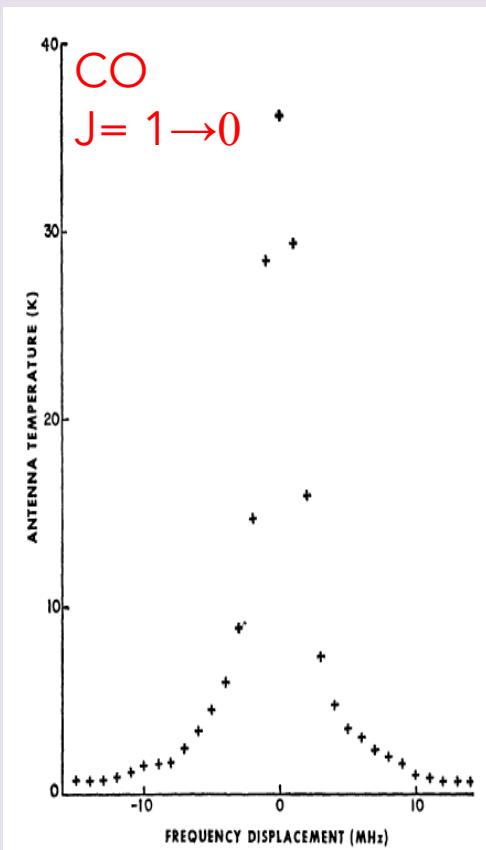
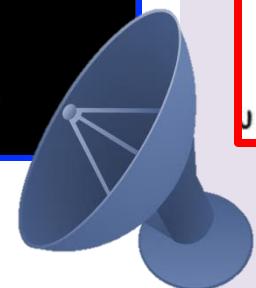
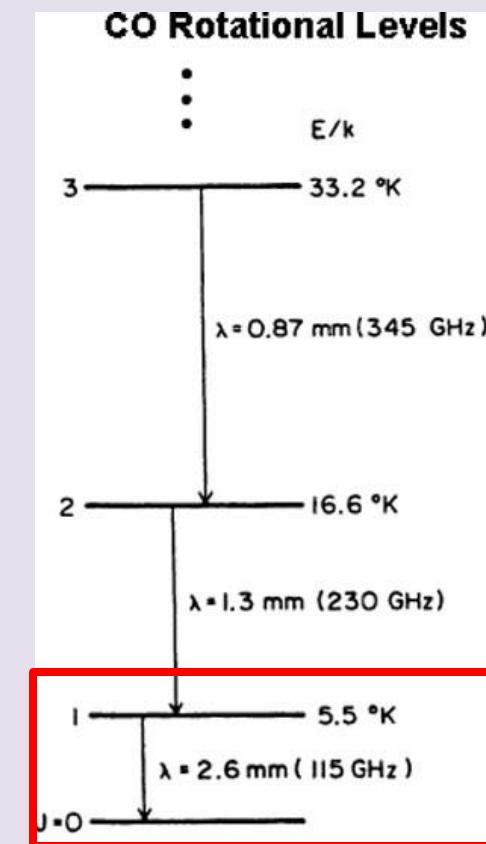
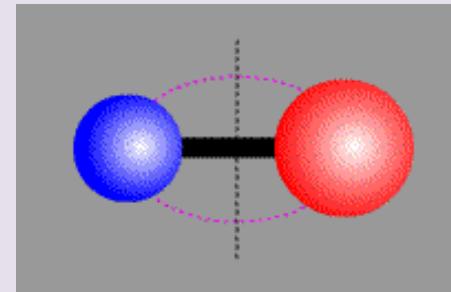
***Questions?
+ Takeaways from
the Class!***





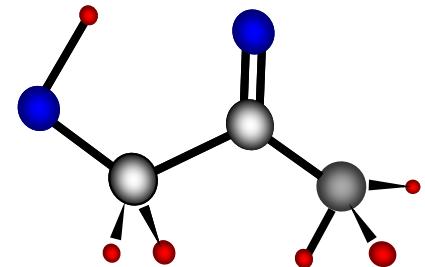
<https://www.nro.nao.ac.jp/~kt/html/kt-e.html>

Discovery of CO
in the Star Forming Region,
Orion KL at 115 GHz
(J = 1 → 0 transition)
in 1970 at **Kitt Peak, Arizona!**



Wilson et al., 1970

Molecule Formation (in Molecular Clouds)

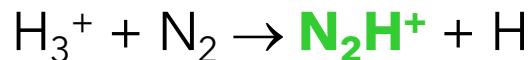
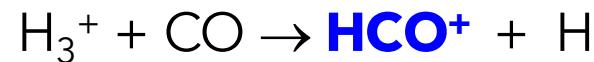
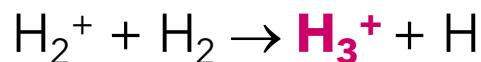
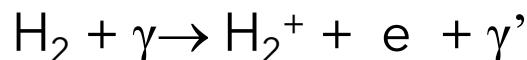


- Typical Conditions in molecular gas:
 - low Densities ($10 - 10^7 \text{ cm}^{-3}$; $< 10^{-12} \text{ torr}$)
 - compared to Earth atmosphere ($\sim 10^{19} \text{ cm}^{-3}$)
 - low Temperatures: $T \sim 10 - 100 \text{ K}$ → **Severely restricts allowed chemical processes!**
 - only two body collisions
 - reactions must be exothermic!

Basic Chemical Scheme:

1) H_2 formed on grain surfaces: $\text{H} + \text{H} + \text{grain} \rightarrow \text{H}_2 + \text{grain}$

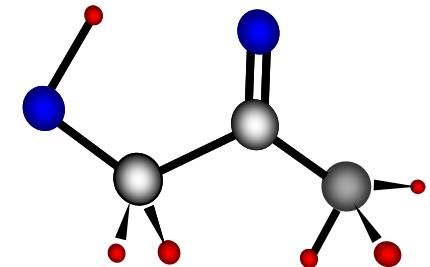
2) Gas-phase reactions initiated by cosmic rays (photons, γ) and proceed via ion-molecule reactions



etc.

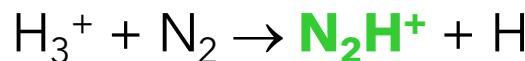
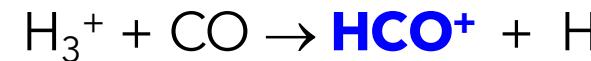
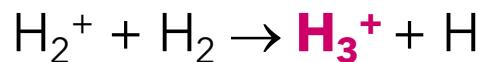
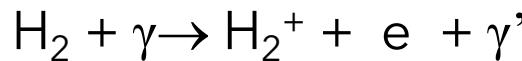
Credit: L. Ziurys

Molecule Formation (in Molecular Clouds)



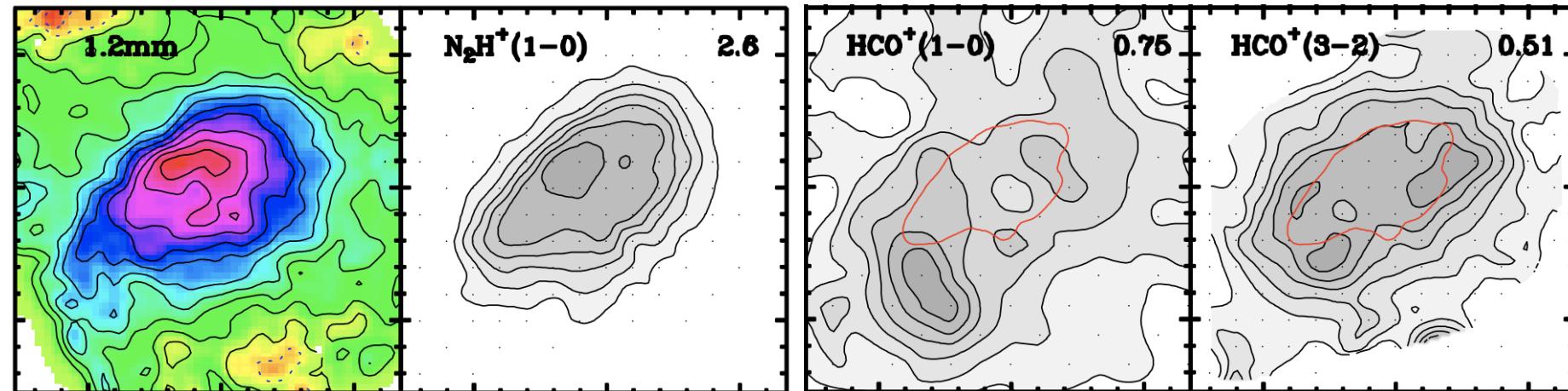
Basic Chemical Scheme:

- 1) H₂ formed on grain surfaces: H + H + grain → H₂ + grain
- 2) Gas-phase reactions initiated by cosmic rays (photons, γ) and proceed via ion-molecule reactions



etc.

Starless core L1498 → CO frozen out on grains thus less HCO⁺



Tafalla 2006

Molecule Formation (in Molecular Clouds)

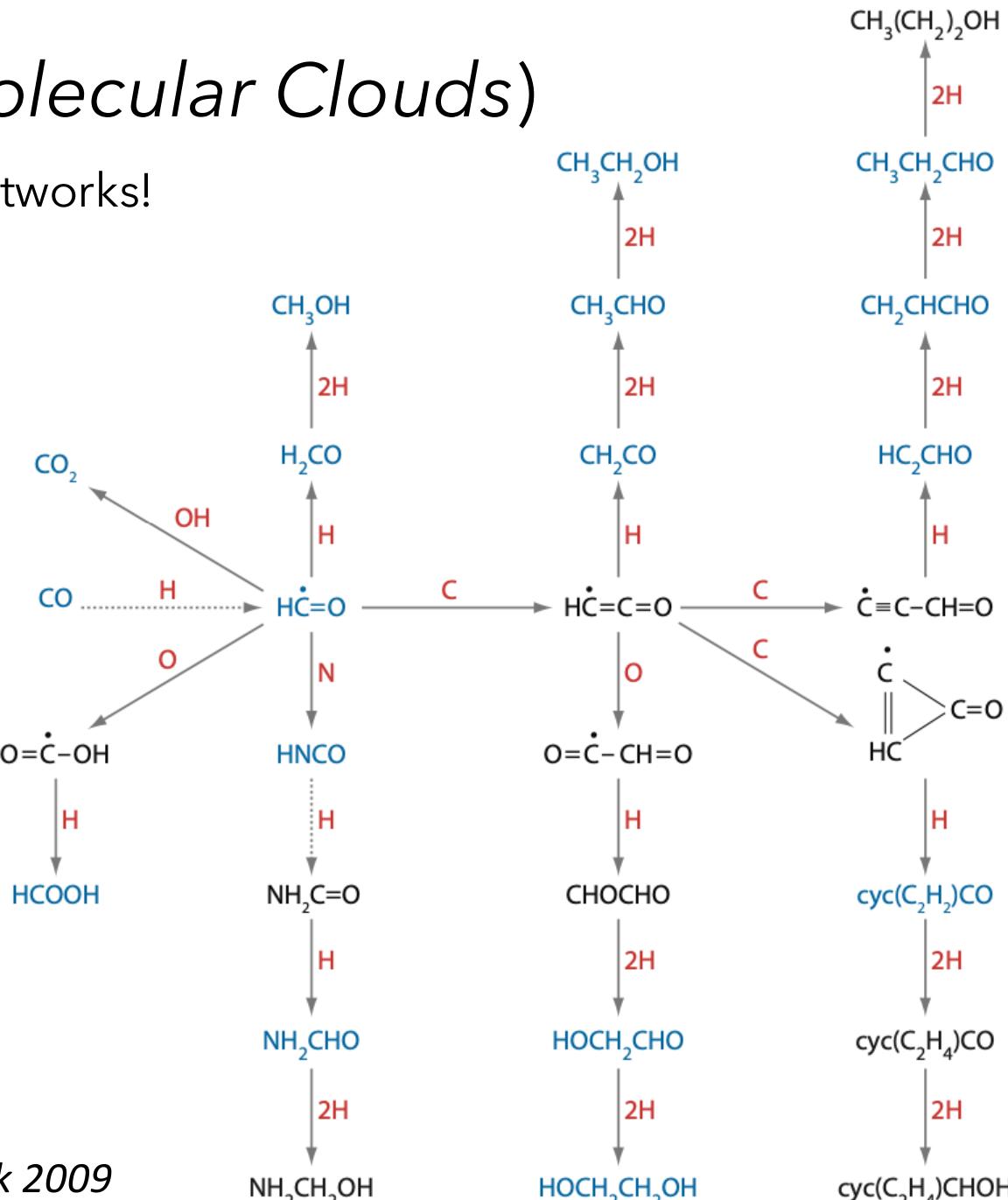
Large chemical reaction networks!

- Organic chemistry on interstellar grains resulting from cold H addition reactions to CO

- Broken arrows indicate reactions with activation energy barriers

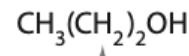
- Where **2H** is shown, a barrier penetration reaction followed by **exothermic** addition

- Molecules in blue detected in star-forming molecular clouds!**



Molecule Formation (in Molecular Clouds)

Large chemical reaction networks!



2H



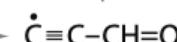
2H



2H



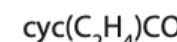
H



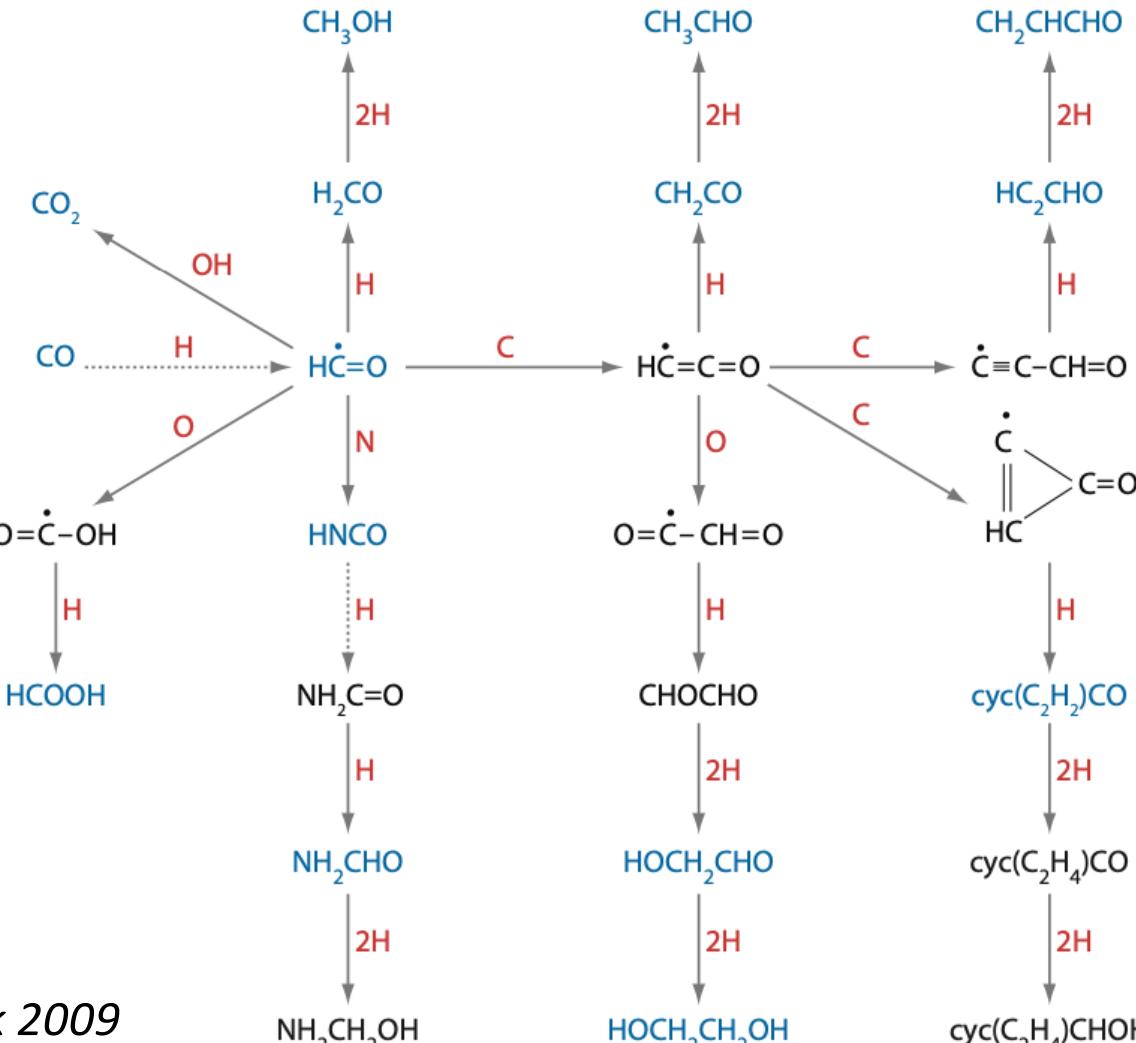
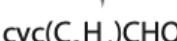
H



2H

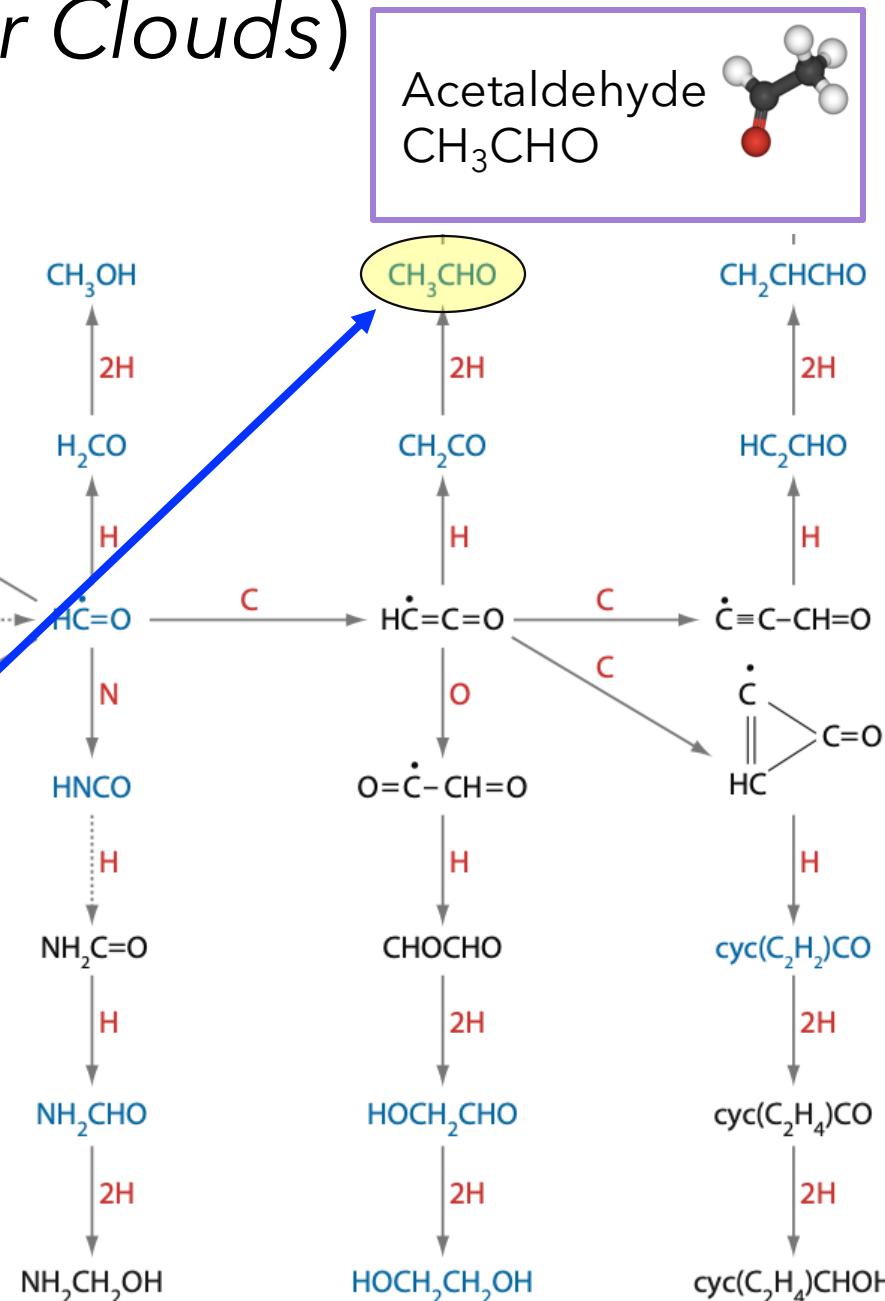
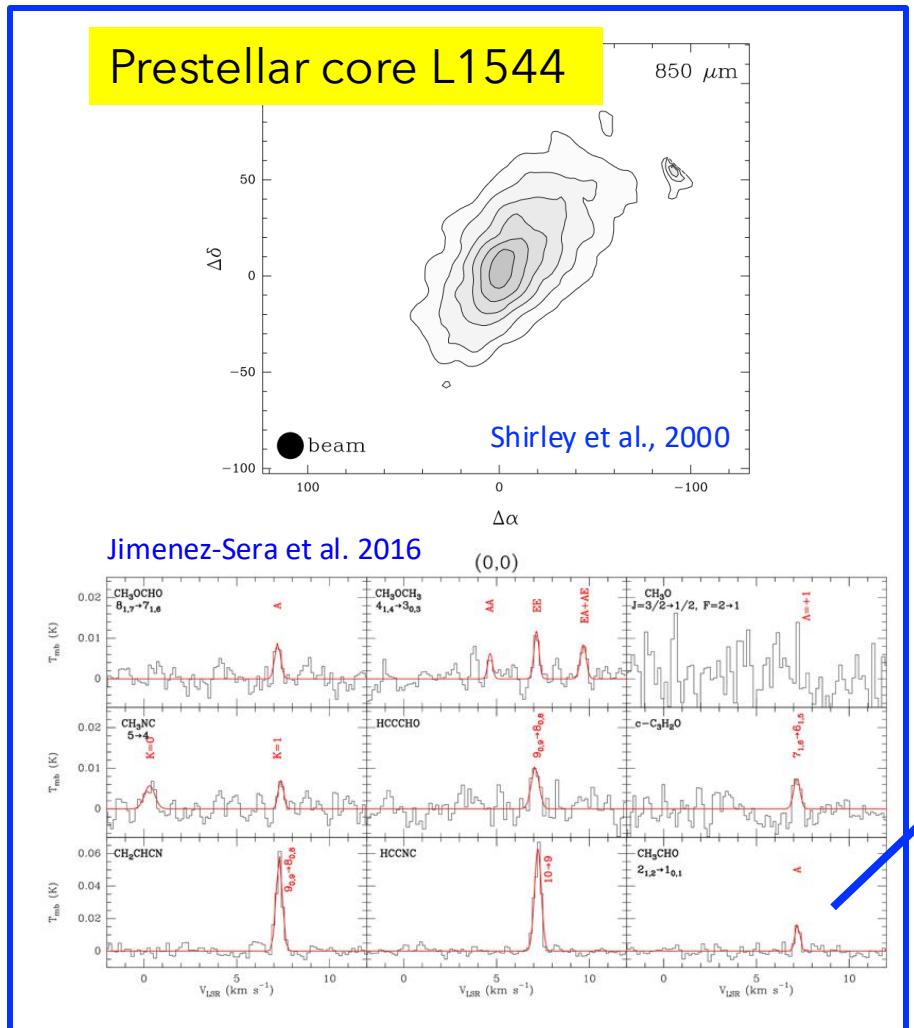


2H

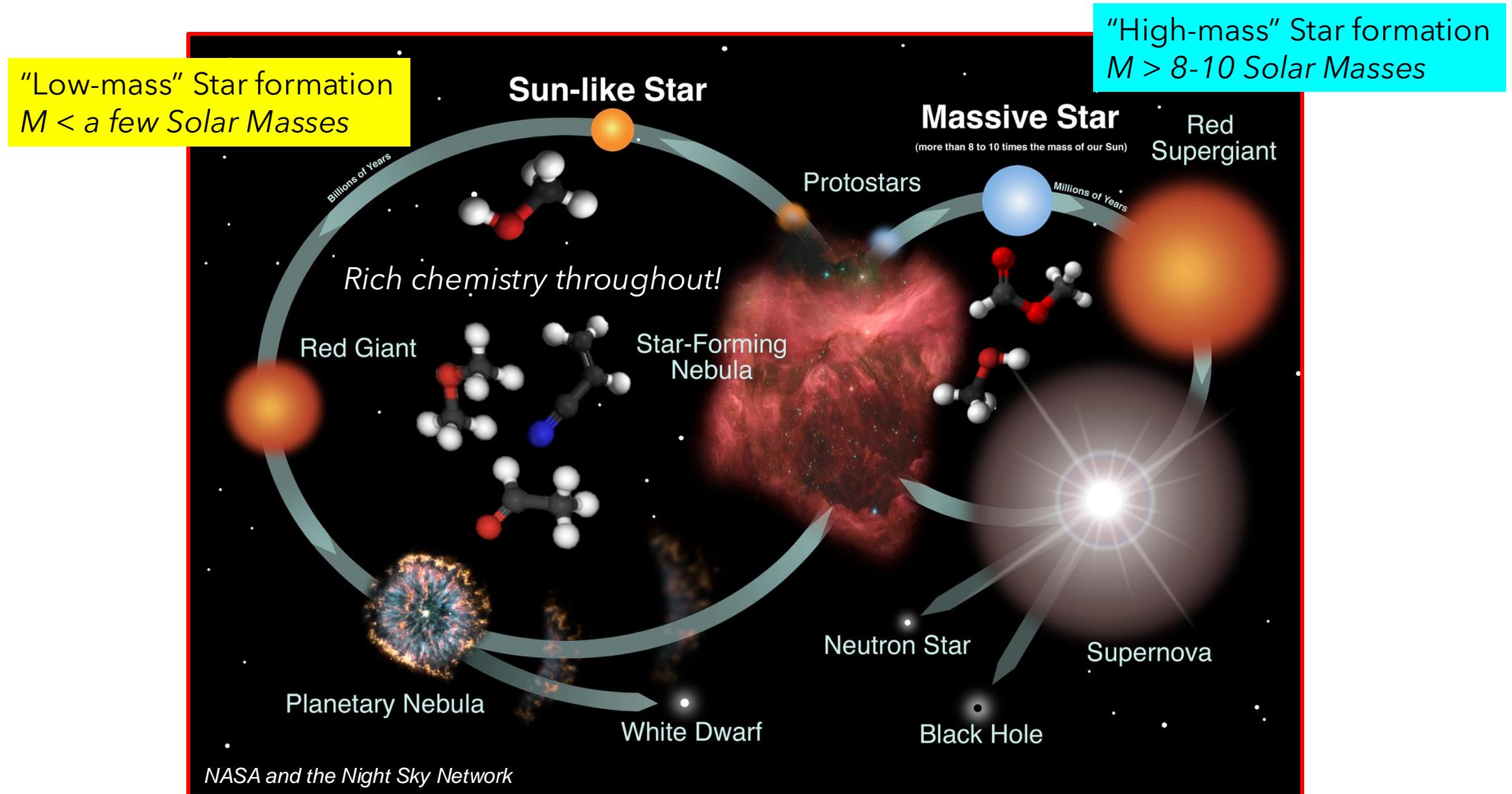


Molecule Formation (in Molecular Clouds)

Large chemical reaction networks!



Evolved Stars – the end stage of star formation



Evolved Stars – the end stage of star formation!

+ Dust Formation!

- **IMPORTANT** in astrochemistry because material is cycled back to the ISM!

- Mass loss from evolved stars

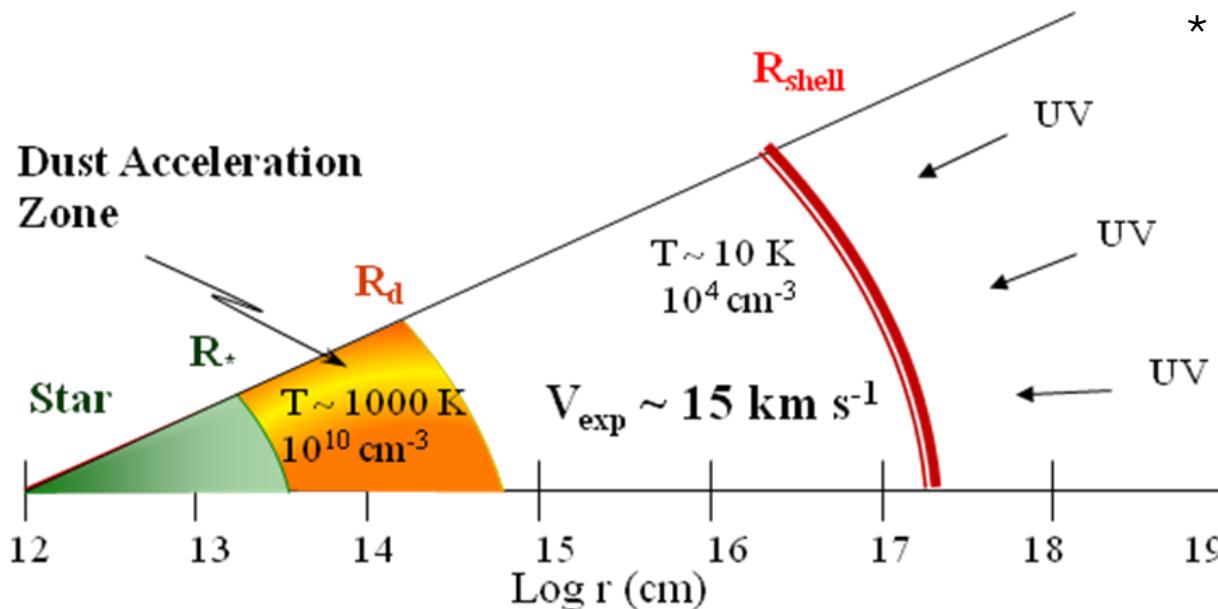
⇒ ***Supplies 85% of material in ISM***

- Material cycled in

circumstellar shells

- Shell is COOL; **Dust grains form**

- Molecules also form there then **transported outward**

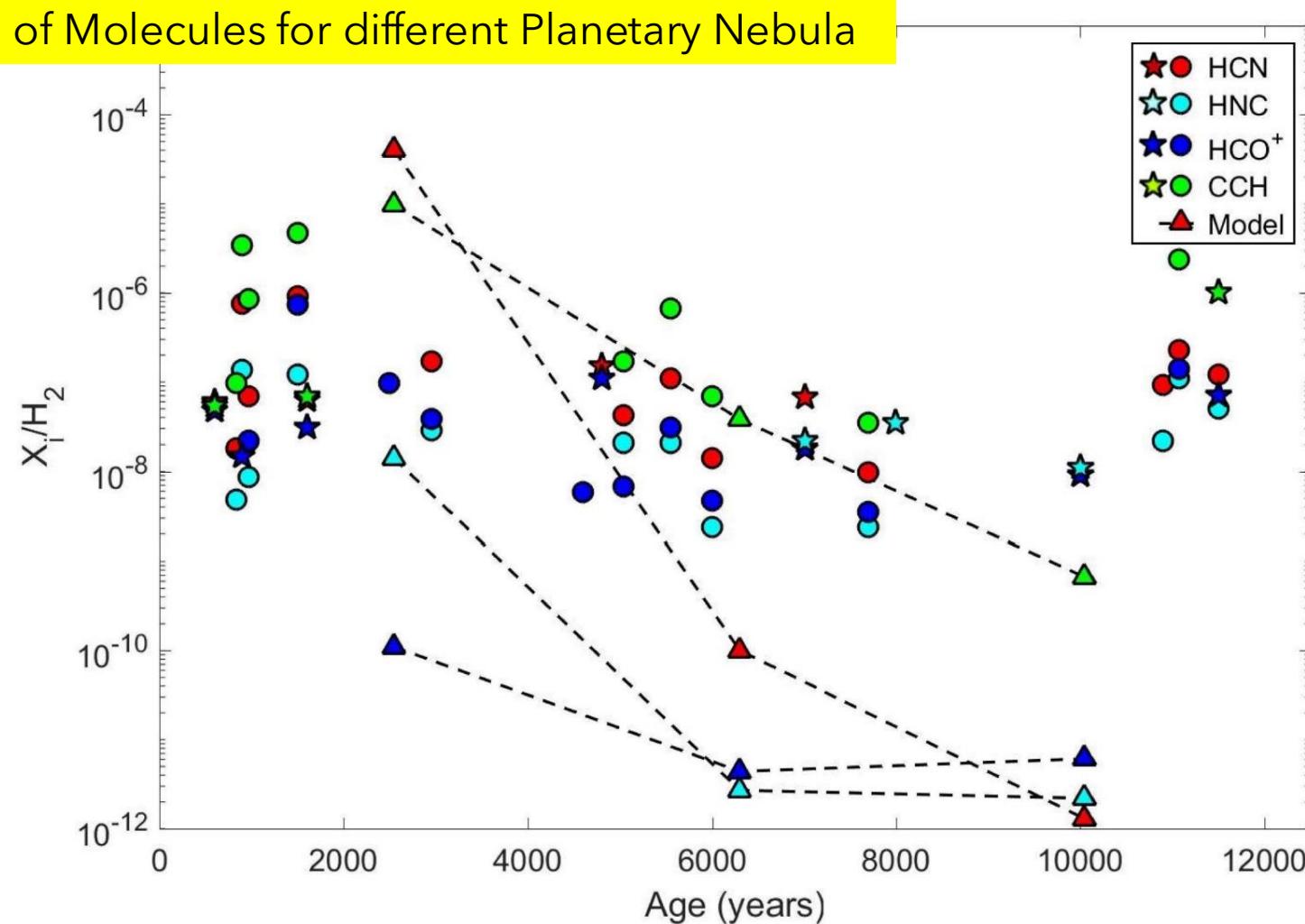


* Universe 99% gas (mostly hydrogen), 1% dust (by mass)

Credit: L. Ziurys

Chemistry in **Planetary Nebula** – Observations vs. Models

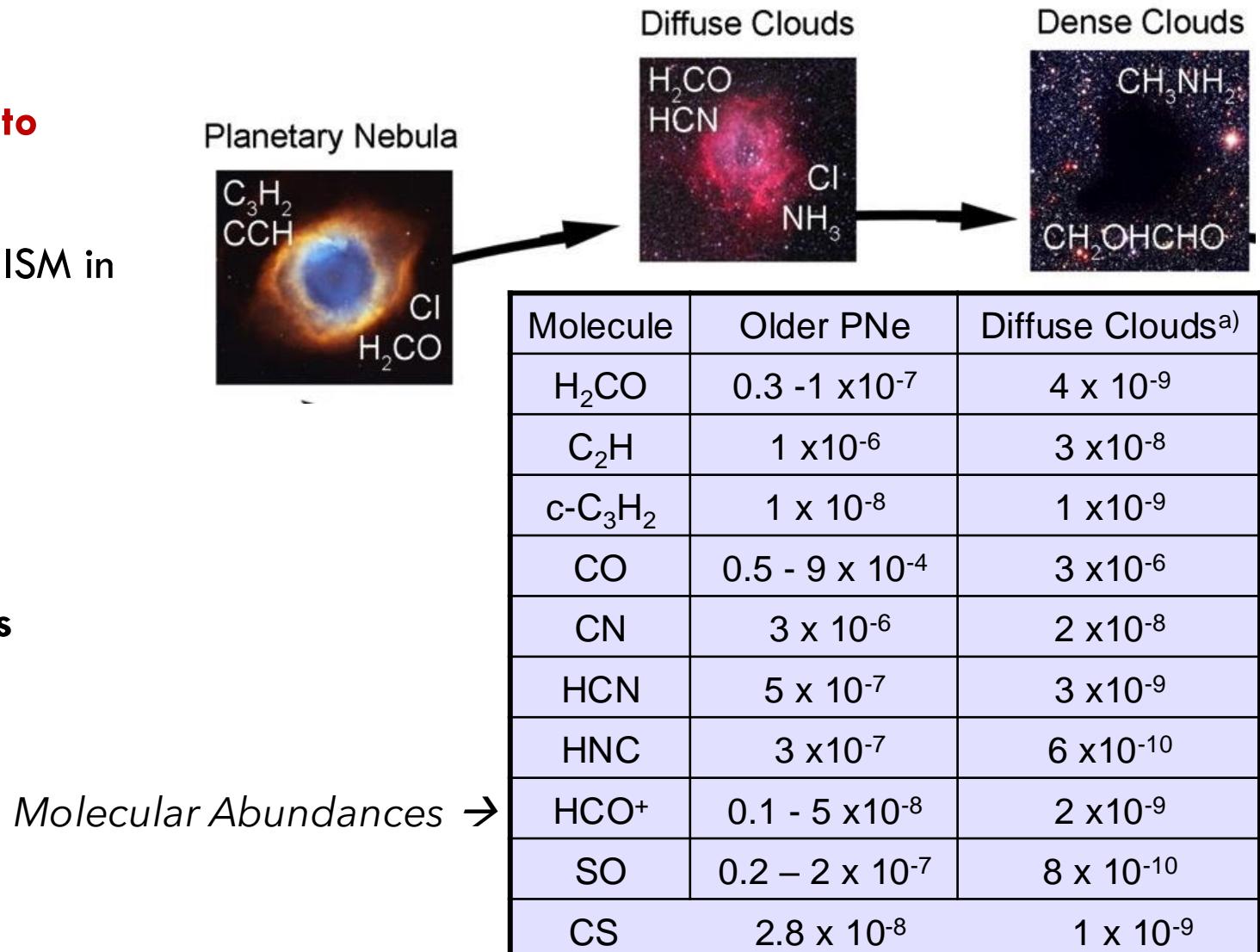
- Molecule abundances stay *constant with age*
- Not destroyed by Photodissociation
- Survive in *dense clumps!* with $n(\text{H}_2) \sim 10^6 \text{ cm}^{-3}$
- Molecular clumps ejected into diffuse ISM
- *Preserve C-enrichment*



Credit: L. Ziurys

Planetary Nebula connection to Diffuse Clouds

- Planetary Nebulae **disperse into diffuse ISM**
- Molecular gas entering diffuse ISM in clumps
- Evidence from **Observations of Diffuse Clouds**
- Diffuse Clouds and Planetary Nebulae similar set of **molecules**



Credit: L. Ziurys

a) Liszt et al. 2006