FUNMILAYO ASHLEY OLADAPO 180403059 ELECTRICAL AND ELECTRONICS ENGINEERING EEG 215 ASSIGNMENT 1a: NEEL TEMPERATURE This is the characteristic temperature for antiferromagnetic systems. It is the temperature above which an antiferromagnetic substance Loses its antiferromagnetism and becomes paramagnetic. II TRANSITION TEMPERATURE This is the temperature at which a normal conductor loser its resistivity and becomes a superconductor. iii CURIE LAN This states that the succeptibility of a paramagnetic material is proportional to the reciprocal of temperature in Kelvin (K) where cis a constant MAGNETO-ELASTIC ENERGY Also known as magneto-restrictive energy is the norkdone by the magnetic field against this elastic restoring fora-INTRINSIC BREAKDOWN This is the liberation of electrons from valence bands. Assumptions OF QUANTUM FREE ELECTRON THEORY - In a metal, the available free electrons are fully responsible for electrical conduction.

	GRADE ASHIET PLANT
- The electrons move a constar	nt potential inside the metal.
The electrons move a constant	e metal surface I high potentia
they cannot come out of	THE MONTH SULLEAN THE STATE OF
barrier.	
- Electrons distributed in va	mon energy levels occording
to Paulis exclusion princip	ole.
- ithe on darling second south	and the state of t
C = P = 1.6 x 10 - 8 1 - m 15 0	1 = 16 × 10-3 mm => 16 × 10-6 m
E = 100 Vm-1	$Ne = 9.1 \times 10^{-31} \text{ kg}$ $Nsity = 1.05 \times 10^4 \text{ kgm}^{-3}$
A = 107.9 kg de	nsity = 1.05 ×104 kgm-3
5	BENERALIZED TEMPERATURE
i Electron relaxation time	The second property of the
	1 20 1 20 1 = 6.25 X107
	P 1.6×10-8
	St Copy Land
Carrier concentration, n =	Avogadro's no. x density
	to weight the
6.023 X1023 X1-05 X104	
107.9	T
twotings o i	1 New 3 - 1
T = 6.25 X 10 + X 9.1 X 10-3	
5.8611 X1025 X (1.6 X10	
	Oband Shanes on call
	vala arangan en model nith
ii Electron mobility u	AND THE OWNER OF THE PARTY OF T
M=5 = 6.25 × 107	The second secon
ne 5-864 VID35VI	0.410=19
716 3.8811 X(0X)	6 X 10-19
= 6.6647 m² V	1 100 11884 ATTER C 21
ii Drift velocity Va	mangely = 2 magnitude
	selection and other of T-
	666.47mls
0 00 11 1100 -	006.41m13

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iv Mean free porth
       1=1=1
         T don
                    T. (1-6 × 10-6)2 × 5.8611 × 10
          = 2.1214 ×10-17
v Current
      J=neVa
         = 1-6 × 10-19 × 5-8611×1025 × 666.47
         = 6.25 X109 Am-2 -> 6249995707 Am-2
    A = \frac{\pi}{d} \left( \frac{d}{2} \right)^2 = \frac{\pi}{4} \cdot \left( \frac{16 \times 10^{-6}}{2} \right)^2 = 2.0106 \times 10^{-10} \,\text{m}^2
  J= I -> J=JA = 6249995707 x 2.0106 x 10-10
                       = 1.2566 A
 di
          In a different sheet
                             E-4-460 × 10 19
                 Int & Thory rappe - 3
   E-1968 X 10-19 = - 5 80 (299354 X10-2)
                         2 M-01x211-1 3
ii Probability function F(F) of an electron
IF = 6.23eV
 F(E) = 0.75
    T = 77°C + 273 = 350K
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6-23 ×1.6 ×10-19 = 9.968 × 10-19 J May 53 4 550M
 K = 1-38 × 10-23 JK-1
      F(E) = 1
 T 0.75 = 1
            1+ e[=-9.968 x10-19]
0-75 =
               1+ e = 9.968 × 10-19 ]
     + e^{\left[\frac{E-9.968\times10^{-19}}{483\times10^{-21}}\right]} = 6
                              0.75
      [E-9.968×10-19] = 1

( 4.83×10-21 ] = 1
         E-9-968 X10-19 > In 1
4-83 X10-21
        E-9-968 × 10-19 = -5.306297354 × 10-21
        F = 9-915×10-19 J
           = 6-1968eV
  P = charge
    volume
   = -ze = -3ze
               4XR3
 Lorentz force
   Fi= gt = -zet
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At equilibrium

## DENSITY OF STATES

A parameter of interest in the string of conductivity of metals and semi-conductors is the dennity of State. The Fermi function F(E) gives only the probability of fulling up of electrons on a given energy state. It does not give information about the no. of electrons that can be fulled in a given energy state. To know this we should constain a seertain the no. of available energy state called DENSITY STATE.

Density of state to defined as the no. If energy state per unit volume in an energy interface. It is used to calculate the no. If charge corriers per unit volume of any solid.

NCE) dE = No. of energy states blu E and E+ de

Volume of the metal

NCE) dE = T (8m) 3/2 E 1/2 dE

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