VUV AND EUV IRRADIATION OF CH₄ + NH₃ ICE MIXTURES

Lily Leung

Contents

- Introduction
 - CN⁻ formation mechanisms by electron irradiation at 10-15 K
 - Different results from 2 groups
 - We perform 3:2 CH₄+NH₃ ice mixtures by VUV and EUV photons
 - What astrophysical environments are we demonstrating?
 - Charon
 - NH₃
 - Infra-red spectra shows ammonia on Organa Crater
 - CH₄
 - Surface temperatures at different latitudes
 - We perform 1:5,1:10 and 1:20 CH₄+NH₃ ice mixtures

Contents

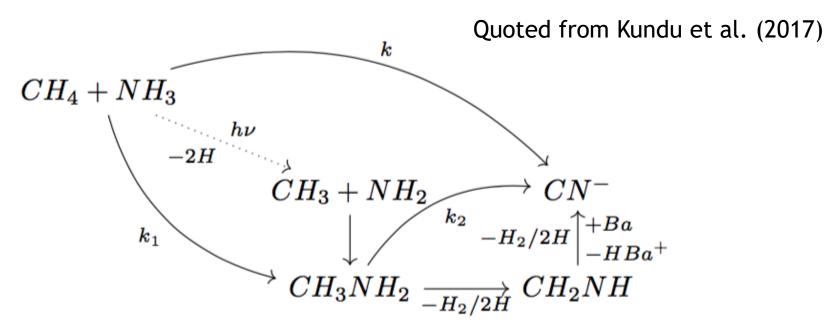
- Methodology
 - Experimental setup
 - The spectrum of VUV (MDHL) energy source
 - Experimental Configurations
- Results
 - Production of CN⁻
 - The relations between CN^- and C_2H_6
 - CN⁻ formation efficiency of EUV (40.8 eV) and VUV (9.27 eV)
- Astrophysical Implications
 - Understand CN⁻ formation after winter on surface of Charon

Introduction

Production mechanism of CN⁻

Enthalpy of CH₃NH₂ formation

$$CH_3 + NH_2 \rightarrow CH_3NH_2 \Delta H = -3.64 \text{ eV}$$



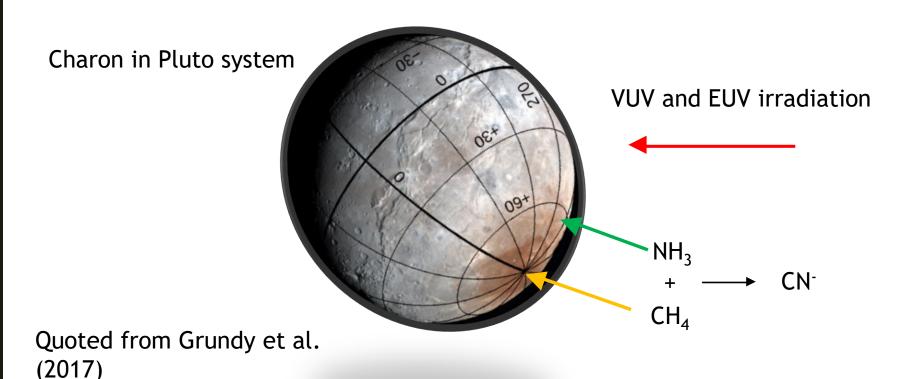
Quoted from Kim and Kaiser (2011)

Production mechanism of CN⁻

Attempts to detect CH₃NH₂:

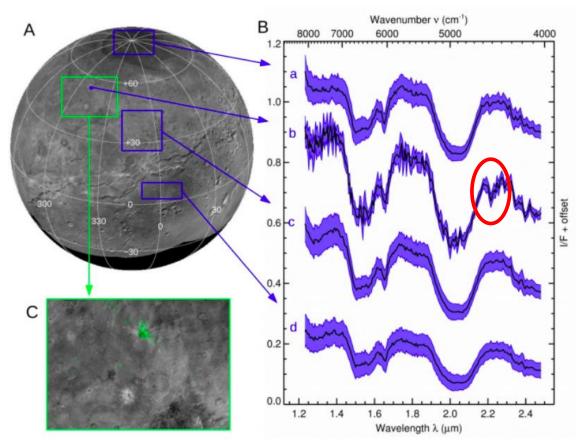
- Different results from 2 e⁻ irradiating experiments
 - 5 keV e⁻ by Kim and Kaiser (2011):
 - The intermediate CH₃NH₂ was detected by TPD
 - 1- 90 eV e⁻ experiment by Kundu et al.(2017)
 - The intermediate CH₃NH₂ cannot be detected by TPD
- How about EUV and VUV photons?

What astrophysical environments are we demonstrating?



Ammonia on Organa Crater

- Ammonia
hydrate
(2.21µm) was
detected all
over the
surfaces,
especially on
Organa Crater

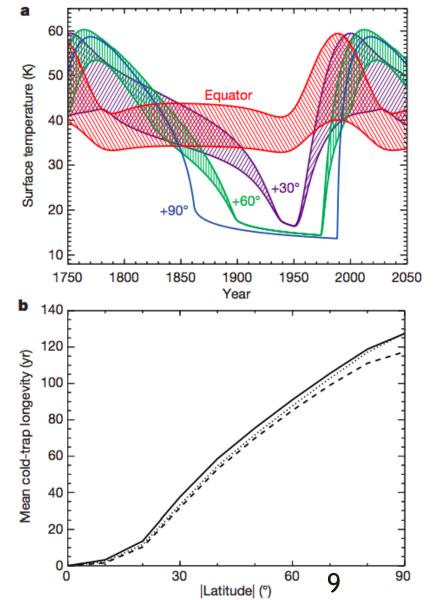


from Grundy et al. (2016)

Surface temperatures at different latitudes

- ►Thermal model from Grundy et al. (2016) shows the pole position is below 25 K for 130 years
- Methane can condense on those positions where the temperature is below 25 K.

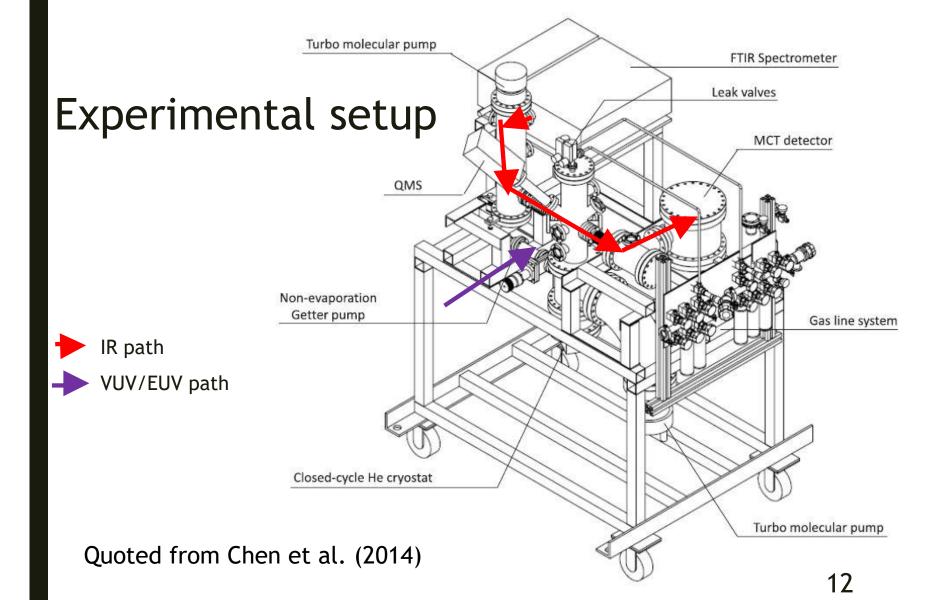
Quoted from Grundy et al. (2016)



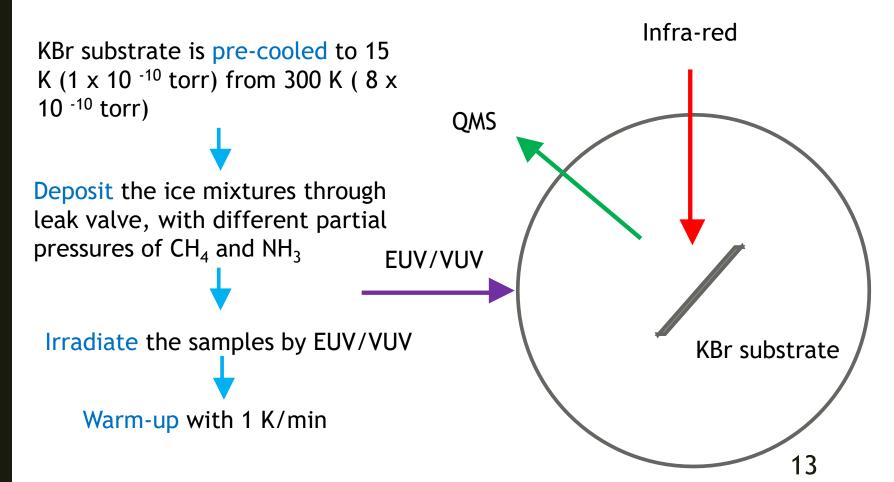
Motivation

- 1. To compare with previous studies
 - Kim and Kaiser ($CH_4:NH_3$ 3:1) and Kundy et al. (2017) ($CH_4:NH_3$ 3:2)
 - We perform experiment of $CH_4+NH_3=3:2$
 - Confirm the mechanism of CN⁻
- 2. To simulate the surface of Charon
 - Experiment: $CH_4:NH_3=1:5, 1:10, 1:20$
 - Variation of photon sources: from VUV to EUV

Methodology



Experimental Protocol



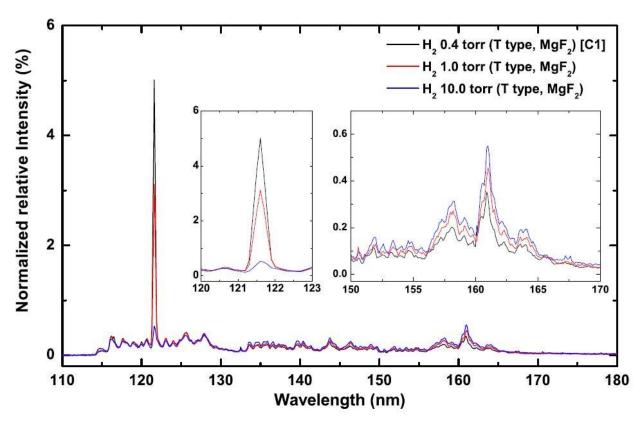
The spectrum of VUV (MDHL) energy

source

 H₂ 0.4 torr was adopted

- 19.1% is Ly-α
- average photon energy is 9.27 eV

• EUV is 40.8 eV (30.4nm) provided by NSRRC



Quoted from Chen et al. (2014)

Experimental Configurations

Energe tic	constituent	Column Density (x10 ¹⁵ molecules cm ⁻²)			
Source		3:2	1:5	1:10	1:20
VUV (MDHL)	CH ₄	900	120	60	30
	NH ₃	600	600	600	600
EUV (30.4 nm)	CH ₄	900	120		
	NH ₃	600	600		

Results

Beer's Law

Transmittance T(v) is defined by:

$$T(v) = \frac{I(v)}{I_o(v)}$$

Absorbance $\tau(v)$ is defined by:

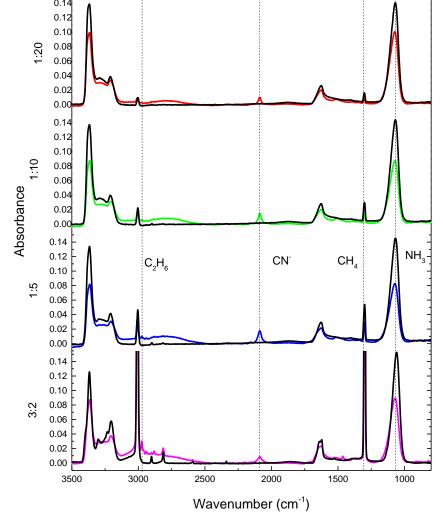
$$\tau(v) = -lnT = -\ln\left(\frac{I(v)}{I_o(v)}\right) = nl\sigma(v)$$

- where n is number density (molecules cm⁻³), l is the path length (cm) and $\sigma(v)$ is the cross-section (cm² molecules ⁻¹)

Column density *N* is defined by:

$$N = \frac{\int \tau(v)dv}{A(v)}$$

 where N is column density (molecules cm⁻²), A(v) is the absorption strength (A-value) (cm molecule⁻¹) from literatures



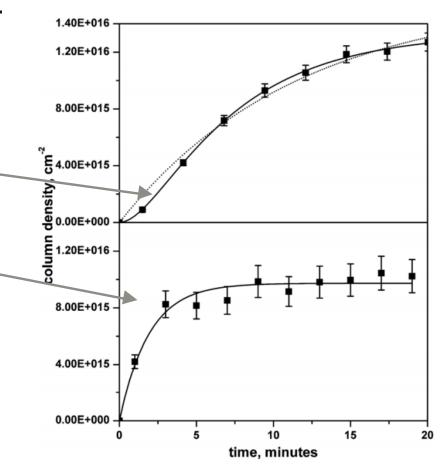
Infra-red spectra before (black lines) and after (coloured lines) VUV irradiation where CN^- , C_2H_6 and C_3H_8 are formed after VUV irradiation.

■ 2 steps/1 step?

2 steps rate equation:

1 step rate equation:

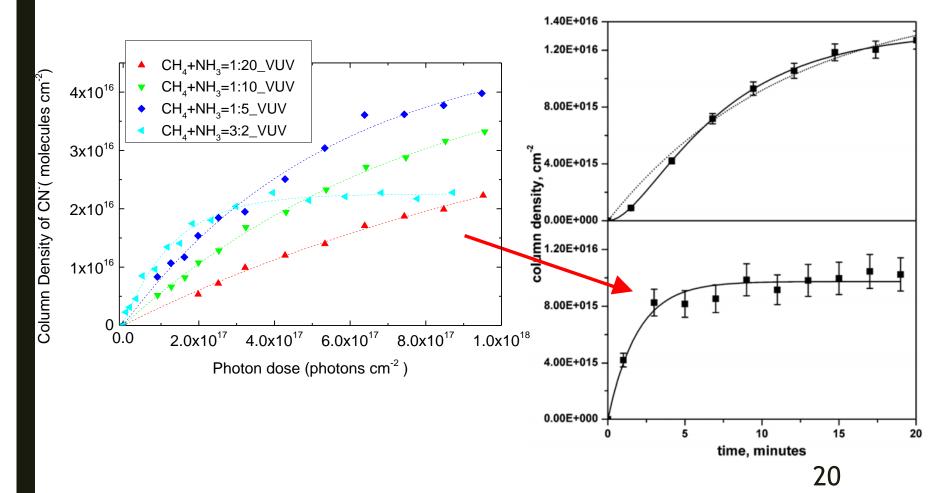
$$\qquad [CN^-] = (1 + e^{-kt})[A]_o$$



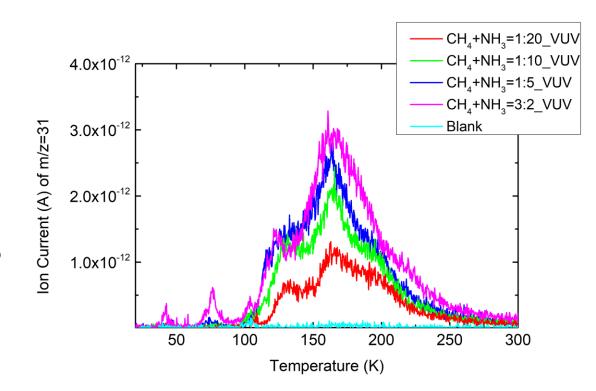
Quoted from Kim and Kaiser (2011)

	Table 3.5: The	fitting results of CN ⁻ by	equation 2.10		
VUV experiments with CH ₄ +NH ₃ ice mixtures					
Ratio	$A (x10^{16} \text{ molecules cm}^{-2})$	$k_1 (x10^{-18} \text{ photon}^{-1})$	$k_2 \text{ (photon}^{-1})$		
1:20	4.75 ± 0.40	0.70 ± 0.09	>1		
1:10	4.51 ± 0.18	1.33 ± 0.13	>1		
1:5	4.61 ± 0.18	1.93 ± 0.19	>1		
3:2	2.24 ± 0.03	8.21 ± 0.70	>1		
	Quotated from				
Ratio	$A(x10^{16} \text{ molecules cm}^{-2})$,	$k_2 \ (\times \ 10^{-3} \ \text{s}^{-1})$		
	$0.1 \ \mu A e^-$ with CH_4+NH_3 ice mixtures				
3:1	1.3 ± 0.0	2.7 ± 0.3	8.9 ± 1.6		
	$1 \mu A e^-$ with $C_n H_{2n+2}$ (n=1-6)+NH ₃ ice mixtures				
2:5	1.0 ± 0.0	8.7 ± 1.3	»1		

A represents the amount of CN⁻ we may obtain when irradiated the ice for infinitely long.

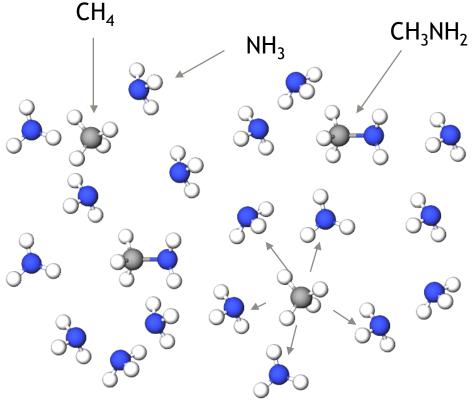


Methylamine (CH₃NH₂) with m/z=31 is detected by QMS



2. The scenario for NH₃ dominating ice mixtures

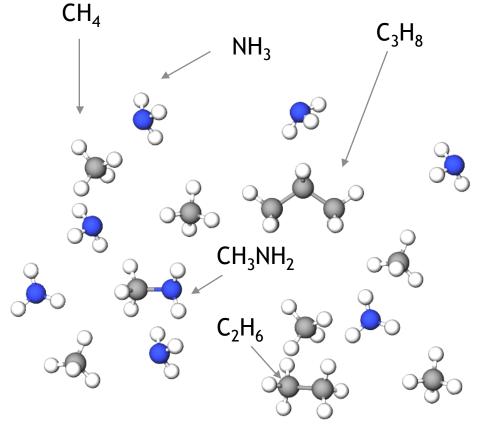
Once CH₄ becomes CH₃ radical, it can easily forms CH₃NH₂ and hence become CN⁻.



A diagram of $CH_4:NH_3 = 1:5$

2. The scenario for CH₄ dominating ice mixtures

CH₂NH₃ (formed by CH₃ + NH₂) has a competing relationship with C₂H₆ (formed by 2 CH₃) and C₃H₈ (formed by CH₂ + C₂H₆ or C₂H₄ + CH₄)

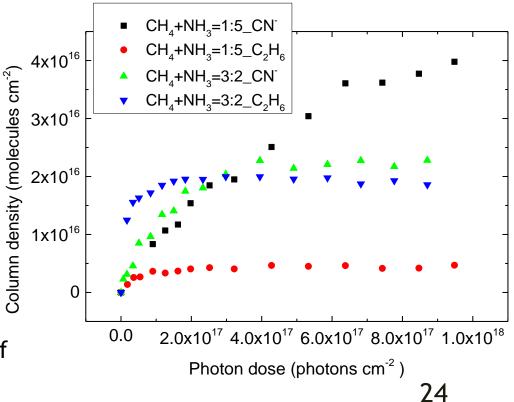


A diagram of $CH_4+NH_3 = 3:2$

2. The relations between CN⁻ and C₂H₆ during VUV irradiations

CH₄:NH₃	C ₂ H ₆ (ML)	CN ⁻ (ML)	Ratio of CN ⁻ to C ₂ H ₆
3:2 (CH ₄ dominant)	19.1	23	1.2
1:5 (NH ₃ dominant)	4.3	49	11.3

Concentration of CN^{-} is not proportional to initial amount of CH_{4} when CH_{4} is in excess.



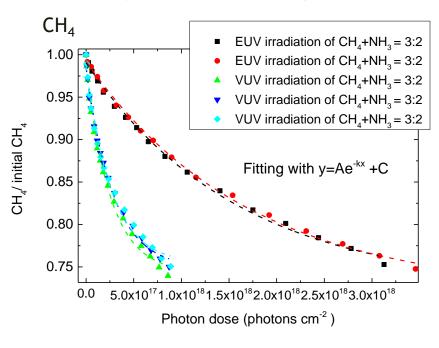
3. Energy needed for forming radicals by EUV (40.1 eV) and VUV (9.27 eV)

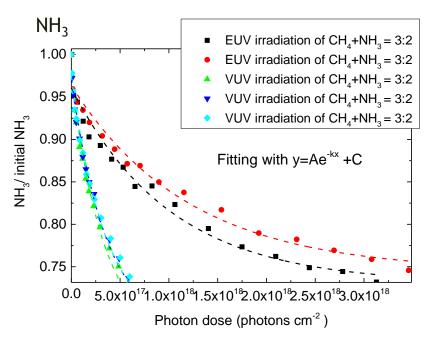
Radicals species	CH ₄	NH ₃
- 1 H	4.55 eV	4.67 eV
-2 H	4.78 eV	4.38 eV
-3 H	9.19 eV	7.63 eV

(quoted from Kundu et al. (2017))

3. Destruction cross-section of EUV (40.1 eV) and VUV (9.27 eV)

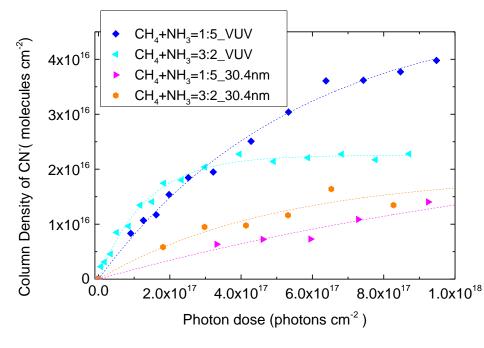
■ Fitting with $y = Ae^{-kx} + C$ (pseudo first order kinetics)





3. CN⁻ formation efficiency of EUV (40.1 eV) and VUV (9.27 eV)

k (photons ⁻¹ cm ²)	CH ₄ (x 10 ⁻¹⁸)	NH ₃ (x10 ⁻¹⁸)	
VUV (MDHL)	3.70±0.18	2.89±0.10	
EUV (30.4nm)	0.61±0.03	0.91±0.11	
Destruction cross-section ratio	6.06±0.07	3.18±0.12	
k (photon ⁻¹ cm ²)	CH ₄ to NH ₃ 3:2 (x 10 ⁻¹⁸)	CH ₄ to NH ₃ 1:5 (x10 ⁻¹⁸)	
VUV (MDHL)	8.21±0.70	1.93±0.19	
EUV (30.4nm)	1.92±1.99	0.63±0.37	
CN ⁻ production ratio	4.28	3.06	



Astrophysical implications

Understand CN⁻ formation after winter on surface of Charon

Surface composition after 1 Pluto winter:

Ly α exposure: 1.9 x 10⁹ eV cm⁻² s⁻¹ (Grundy et al. 2016)

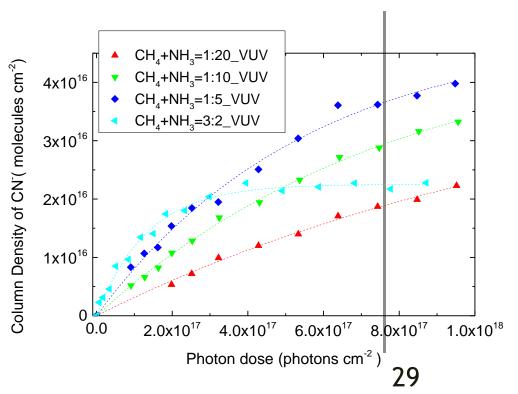
 \rightarrow photon dose: 7.64 x 10 ¹⁷ photons cm⁻²

■ CH₄ deposition rate:

(Hoey et al. 2017)

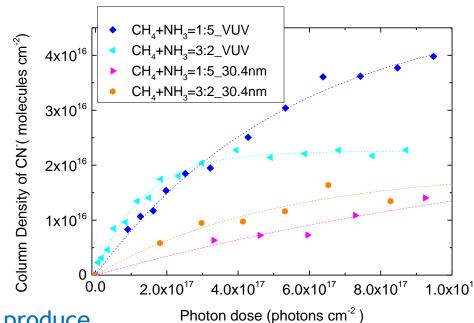
 \rightarrow ~110-150 ML in 130 earth years

CH ₄ +NH	CH ₄ (ML)	CN ⁻
1:5	110	36.6
1:10	60	29.5
1:20	30	18.9
3:2	900	22.5



Astrophysical implications

- VUV is 3.06 to 4.28 times more efficient than EUV
- VUV flux is 1 order of magnitude more intense than EUV irradiations (Grundy et al. 2016)
 - Ly-a exposure: 1.9 x 10⁹
 eV cm⁻² s⁻¹
 - EUV exposure: 8.7 x 10⁷
 eV cm⁻² s⁻¹



Ly- α is the main energy source to produce CN- on Charon

Conclusion

- 1. Detection of methylamine implies that CN⁻ is formed via a 2 step mechanism.
- 2. Concentration of CN⁻ is not proportional to the initial amount of CH₄ when CH₄ is in excess.
 - This implies that we have to experimentally simulate the amount of CN⁻ after Charon winter for further investigations.
- 3. The reduced destruction cross-section of EUV 30.4nm irradiation is the main factor of slowing the rate of formations.
 - This implies that Ly-a is the main energy source to produce CN⁻ on Charon.

Q&A

Production yield and production rates

- The yields should be correlated with initial limiting substances
- Fitting rates are the same

