VUV AND EUV IRRADIATION OF CH₄ + NH₃ ICE MIXTURES

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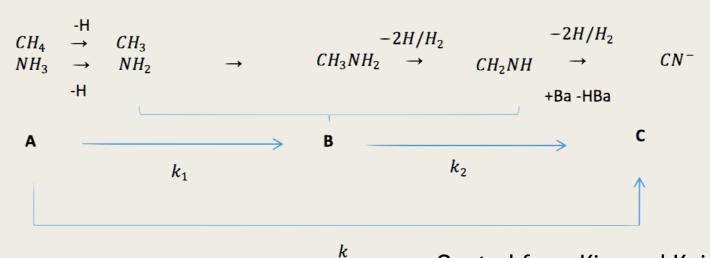
Introduction

Production mechanism of CN⁻

Enthalpy of CH₃NH₂ formation

$$CH_3 + NH_2 \rightarrow CH_3NH_2 \Delta H = -3.64 \text{ eV}$$

Quoted from Kundu et al. (2017)



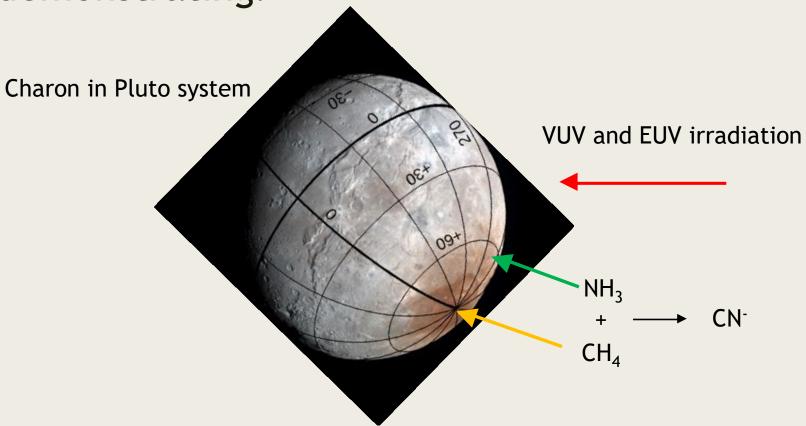
Quoted from Kim and Kaiser (2011)

Production mechanism of CN⁻

Attempts to detect CH₃NH₂:

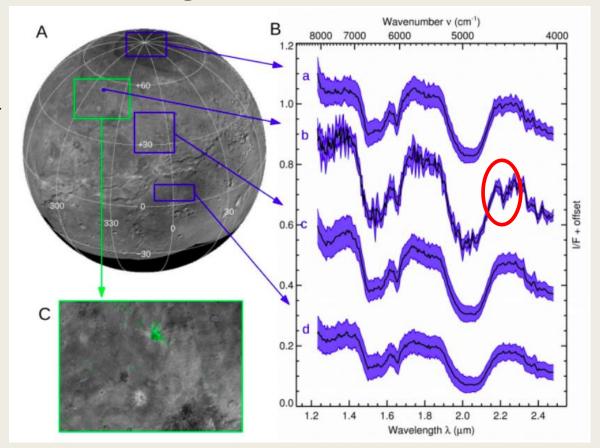
- Different results from 2 e⁻ irradiating experiments
 - 5 keV e⁻ by Kim and Kaiser (2011):
 - The intermediate CH₃NH₂ was detected by TPD
 - 1- 90 eV e⁻ experiment by Kundu et al.(2017)
 - The intermediate CH₃NH₂ cannot be detected by TPD
- How about photons?

What astrophysical environments are we demonstrating?



Ammonia on Organa Crater

 Ammonia was detected all over the surfaces, especially on Organa Crater

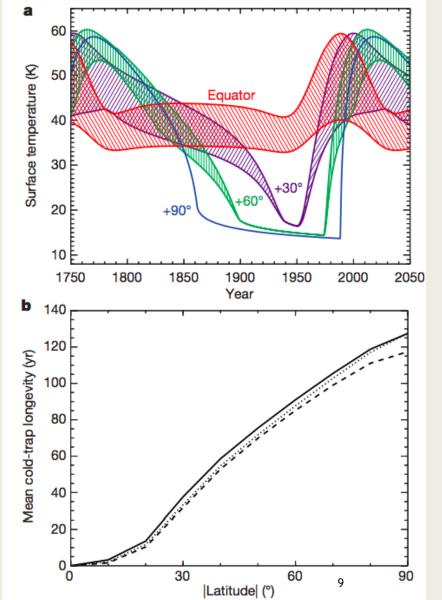


from Grundy et al. (2016)

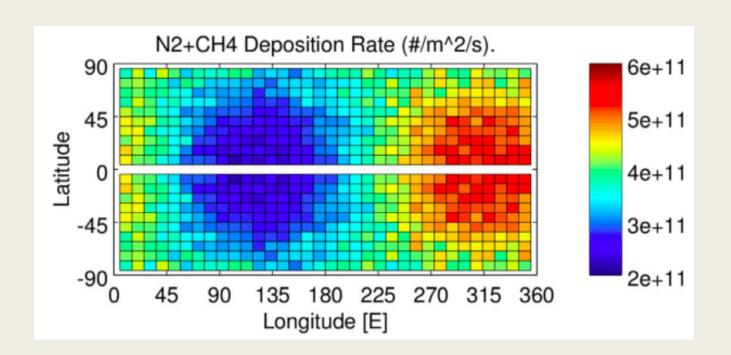
Surface temperatures at different latitudes

- ► Thermal model from Grundy et al. (2016) shows the pole position is below 25 K for 130 years
- Methane can condense on those positions where the temperature is below 25 K.

Quoted from Grundy et al. (2016)



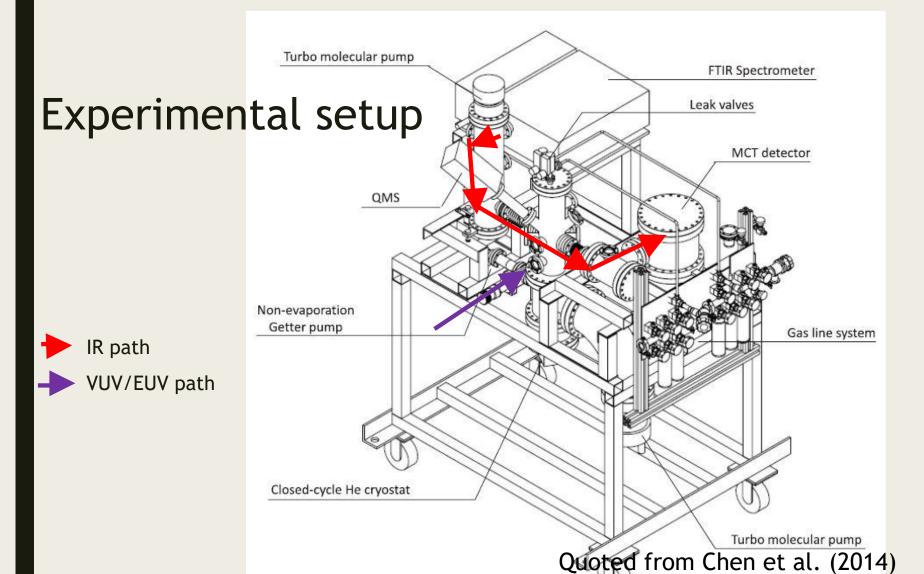
Deposition rate of methane on Charon



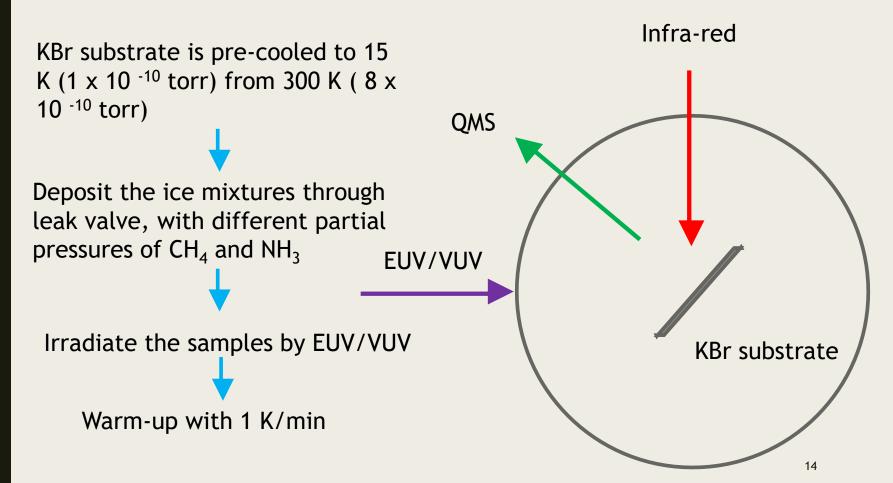
Motivation

- 1. To compare with previous studies
 - Experiment: $CH_4+NH_3=3:2$
 - Confirm mechanism of CN⁻
- 2. To simulate the surface of Charon
 - Experiment: $CH_4 + NH_3 = 1:5, 1:10, 1:20$
 - Variation of photon sources: from VUV to EUV

Methodology



Experimental Protocol

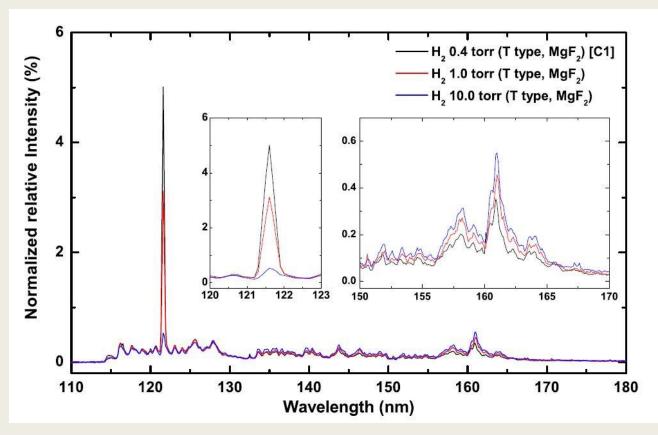


The spectrum of VUV (MDHL) energy

source

H₂ 0.4 torr was adopted

- 19.1% is Ly-α
- average photon energy is 9.27 eV



Quoted from Chen et al. (2014)

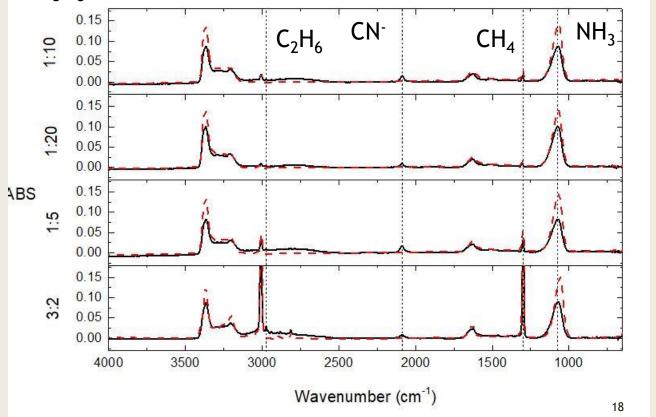
Experimental Configurations

Energetic Source	constituent	Column Density (x10 ¹⁵ molecules cm ⁻²)			
		3:2	1:5	1:10	1:20
VUV (MDHL)	CH₄	900	120	60	30
	NH ₃	600	600	600	600
EUV (30.4 nm)	CH ₄	900	120		
	NH ₃	600	600		

Results

Infra-red spectra

- Infra-red spectra before (red dotted lines) and after (black solid lines)
 VUV irradiation
- CN-, C₂H₆ and C₃H₈ are formed after VUV irradiation



From Beer's Law

Transmittance T(v) is defined by:

$$T(v) = \frac{I(v)}{I_o(v)}$$

Absorbance a(v) is defined by:

$$a(v) = -lnT = -\ln\left(\frac{I(v)}{I_o(v)}\right) = nl\sigma(v)$$

Where n is number density (molecules cm⁻³) l is the path length (cm) and $\sigma(v)$ is the cross-section (cm² molecules ⁻¹)

Column density *N* is defined by:

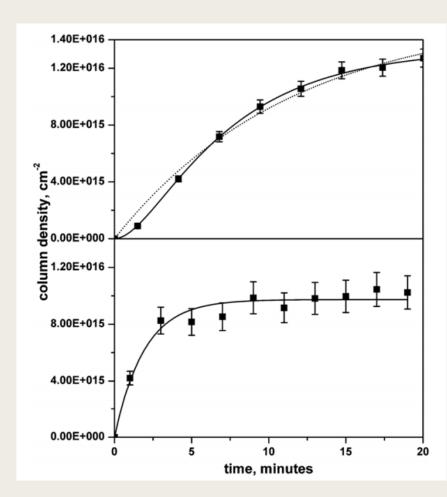
Where N is column density (molecules cm⁻²) A(v) is the absorption strength (Avalue) (cm molecule⁻¹) from literatures

1. Production of CN⁻

- 2 steps/1 step?
- 2 steps rate equation:

1 step rate equation:

$$\qquad [CN^-] = (1 + e^{-kt})[A]_o$$

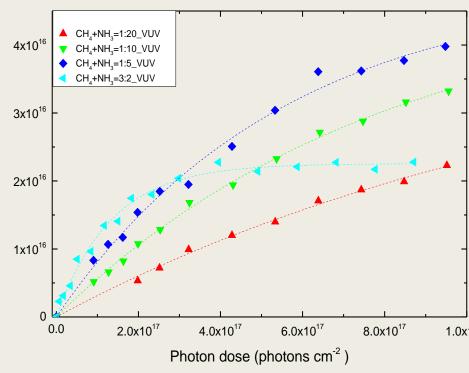


Quoted from Kim and Kaiser (2011)

1. Production of CN⁻

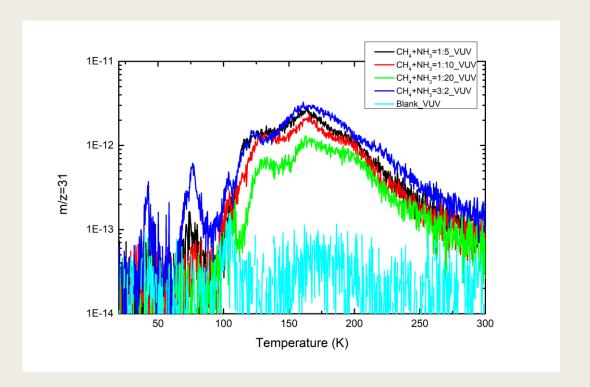
■ 2 steps/1 step?

				_
light source	Ratio of CH₄+NH₃	A (x10 ¹⁶ molecules cm ⁻²) k (x10 ⁻¹⁸ photon ⁻¹) S		
	1:5	4.61±0.18	1.93±0.19	molecules
MDHL	1:10	4.51±0.18	1.33±0.13	mole
MIDHL	1:20	4.75±0.40	0.70±0.09	N(CN')
	3:2	2.24±0.03	8.21±0.70	ž



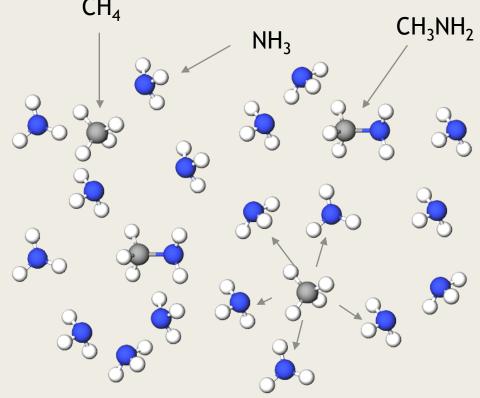
1. Production of CN⁻

Methylamine (CH₃NH₂) with m/z=31 is detected by QMS after isothermal VUV irradiation during warm-up which is the intermediate of the CN⁻



2. The scenario for NH₃ dominating ice mixtures

 Once CH₄ becomes CH₃ radical, it can easily forms methylamine and hence become CN⁻.

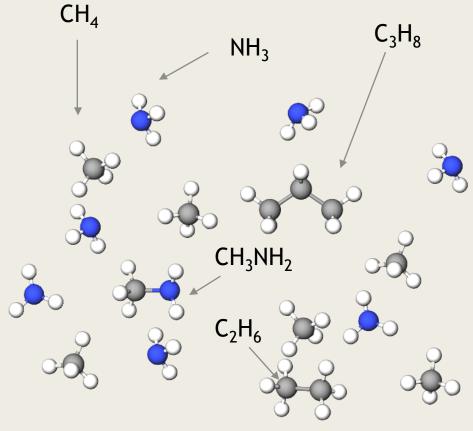


A diagram of $CH_4:NH_3 = 1:5$

2. The scenario for CH₄ dominating ice mixtures

CH₂NH₃ (formed by CH₃ + NH₂) has a competing relationship with C₂H₆ (formed by 2 CH₃) and C₃H₈ (formed by CH₂ + C₂H₆ or C₂H₄ + CH₄)

Once CH₄ becomes CH₃ radical, it reacts with either NH₂ or CH₃ radicals, forming CH₃NH₂ or C₂H₆ respectively

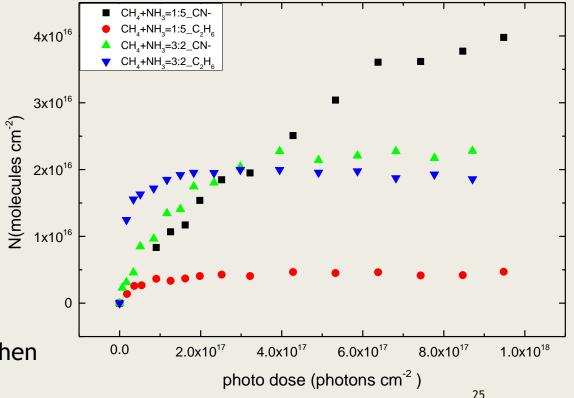


A diagram of $CH_4+NH_3 = 3:2$

2. The relations between CN⁻ and C₂H₆ during VUV irradiations

CH₄:NH₃	C ₂ H ₆ (ML)	CN ⁻ (ML)
3:2 (CH ₄ dominant)	19.1	23
1:5 (NH ₃ dominant)	4.3	49

Concentration of CN^- is not proportional to initial CH_4 when CH_4 is in excess.



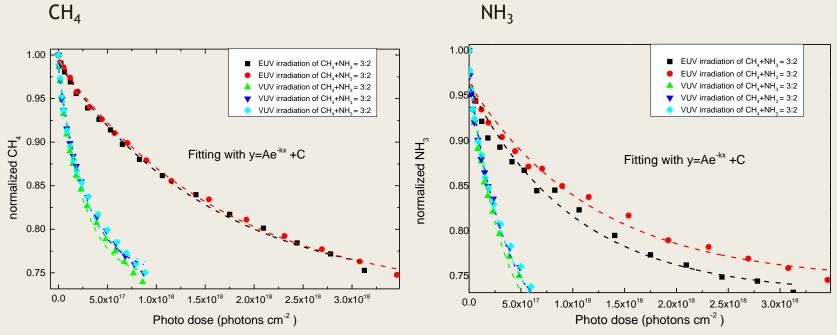
3. Efficiency of EUV (40.1 eV) and VUV (9.27 eV)

Radicals species	CH_4	NH ₃
- 1 H	4.55 eV	4.67 eV
-2 H	4.78 eV	4.38 eV
-3 H	9.19 eV	7.63 eV

(quoted from Kundu et al. (2017))

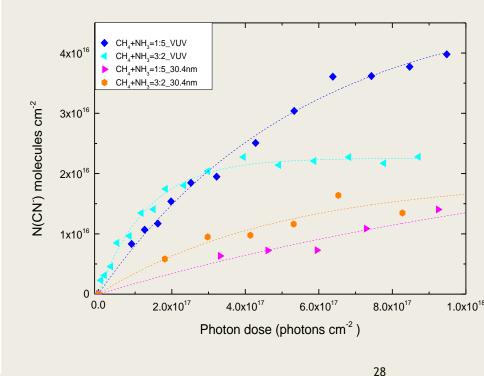
3. CN⁻ formation efficiency of EUV (40.1 eV) and VUV (9.27 eV)

■ Fitting with $y = Ae^{-kx} + C$ (pseudo first order kinetics)



3. CN⁻ formation efficiency of EUV (40.1 eV) and VUV (9.27 eV)

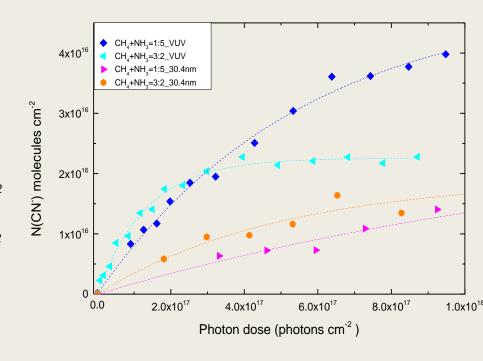
k (photons ⁻¹)	CH ₄ (x 10 ⁻¹⁸)	NH ₃ (x10 ⁻¹⁸)
VUV (MDHL)	3.70±0.18	2.89±0.10
EUV (30.4nm)	0.61±0.03	0.91±0.11
Destruction cross-section ratio	6.06±0.07	3.18±0.12
k (photon ⁻¹)	CH ₄ to NH ₃ 3:2 (x 10 ⁻¹⁸)	CH ₄ to NH ₃ 1:5 (x10 ⁻¹⁸)
VUV (MDHL)	8.21±0.70	1.93±0.19
EUV (30.4nm)	1.92±1.99	0.63±0.37
CN- production ratio	4.28	3.06



Astrophysical implications

Astrophysical implications

- Ly α is the main energy source to produce CN⁻ on Charon
 - 3.06 to 4.28 times more efficient by VUV then EUV
 - VUV flux is 1 order of magnitude more intense than EUV irradiations (Grundy et al. 2016)
 - Ly a exposure: 1.9 x 10⁹ eV cm⁻²
 - EUV exposure: 8.7 x 10⁷ eV cm⁻²

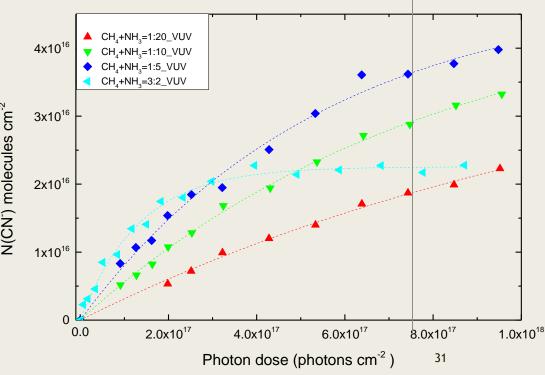


Understand CN- formation after winter on surface of Charon

Surface composition after 1 Pluto winter:

- Ly α exposure: 1.9 x 10⁹ eV cm⁻² s⁻¹ (Grundy et al. 2016)
- = photon dose 7.64x 10 ¹⁷ photons cm⁻²
- = 82-246 ML in 130 earth years

■ CH ₄ deposition rate: 2-6 x 10^7 cm ⁻² $\frac{2}{5}$ s ⁻¹ (Hoey et al. 2017 $\frac{8}{5}$ = 82-246 ML in 130 earth years			
CH ₄ +NH ₃	CN- (ML)	N(CN ⁻) molecules	
1:5	36.6	Z(CN	
1:10	29.5	_	
1:20	18.9		
3:2	22.5		



Conclusion

- 1. Detection of methylamine implies that in
- 2. Concentration of CN⁻ is not proportional to initial CH₄ when CH₄ is in excess.
- 3. The reduced destruction cross-section of EUV 30.4nm irradiation is the main factor of slowing the rate of formations.
- 4. The maximum amount of CN⁻ after Charon winter is simulated experimentally.