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Chemistry MCQs Part 1 ~ Pak MCQs

The higher the pH value, the greater the concentration of hydroxide ions and the lower the concentration of hydrogen ions, which means the more alkaline the solution. Let's take a look at a Chemistry MCQ question. Question 1: An unknown aqueous solution has a pH value of about 2.

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1st year chemistry notes new 1 www.allonlinefree.com CHAPTER 9 SOLUTIONS MCQs Q.1 Which of the following solutions has the highest boiling point? (a) 5.85% solution of NaCl (b) 18.0% solution of glucose (c) 6.0% solution of urea (d) all have same boiling point Q.2 Two solutions of NaCl and KCl are prepared separately by

CHAPTER 9 SOLUTIONS MCQs www.allonlinefree

A solution of sucrose (molar mass = 342 g mol⁻¹) has been prepared by dissolving 68.5 g of sucrose in 100 g of water. The freezing point of the solution obtained will be (K_f for water = 1.86 K kg mol⁻¹) [CBSE AIPMT 2010]

Solutions: MCQs | Question - Paper 2

This is the Part #1 of our Chemistry MCQs section. There are few other parts and each one contain 40 Chemistry MCQs. You can answer first and check your answers at the end of the page.

Chemistry MCQs - Part #1 With Answers • Smart Science Pro

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1 1 Chemistry 1st Year Solved MCQ,S Prepared by: Babar Joya ... EXPERIMENTAL TECHNIQUES IN CHEMISTRY. MCQs . Q.1 Science of the chemical characterization is stand under the heading of (a) industrial chemistry (b) experimental chemistry ... Q.7 Premature crystallization of solution of filter paper or in funnel can

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CHEMISTRY Three hours are allotted for this examination. One hour and 30 minutes are allotted for Section I, which consists of multiple-choice questions. For Section II, Part A, 40 minutes are allotted; for Section II, Part B, 50 minutes are allotted. Section I is printed in this examination booklet. Section II is printed in a separate booklet.

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