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## General Chemistry (121 - 122) | Chemistry & Physics

Page 2 of 11 Chem 121 Final Exam (2) 1) The product of (0.023023 x 4.531) has 2.3 subtracted from it. Correct answer (sig. figs.) in the box. (2) 2) Write down the name (not the symbol) of any transition metal element. (2) 3) The density of aluminum metal is 2.70 g/cm3.Calculate the volume of one mole of aluminum. (3) 4) The dianion of an unknown element contains 54 electrons and 75 neutrons.

#### Chemistry 121 FINAL EXAM - people.ok.ubc.ca

CHEMISTRY 121 FINAL EXAM Record your answers on the machine readable answer sheet provided. Scratch work is to be done on the paper provided with the  $\dots$  500 rnL of a 0.33 M solution was diluted by adding enough water to give 900 mL of solution. Calculate the molarity of the dilute solution. a) 0.18M

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