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Calculate the average atomic masses. Round all answers to two decimal places. Record your answer with the correct units. 1. Iodine is 80% 1271 17% 1261, and 3%1281. Calculate the average atomic mass of iodine. 2. Calculate the average atomic mass of lithium, which occurs as two isotopes that have the following

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concept of significant figures has the answer. When multiplying and dividing numbers, you need to round your answers to the correct number of significant figures. To round correctly, follow these simple steps: 1) Count the number of significant figures in each number. 2) Round your answer to the least number of significant figures.

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ChemQuest 15 Name: ____ Date: ____ Hour: ____ Information: Relating Electron Configurations to the Periodic Table In this section you will see how the periodic table serves as a road map for writing electron configurations. Get your periodic table out and get ready. Remember that a row on the periodic table

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25 3.2 30 4.2 35 5.6 40 7.4 45 9.8 Critical Thinking Questions 4. Why is it logical to expect that the vapor pressure of water would increase as the temperature increases? As the temperature increases, water is more likely to evaporate and the evaporated molecules move faster, colliding with more force (or pressure). 5. Examine the two ...

Information: Collecting Gas Over Water

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