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144 Chemistry: Matter and Change • Chapter 9 Solutions Manual CHAPTER 9 SOLUTIONS MANUAL 24. Challenge $\text{Al(s)} + \text{Pb(NO}_3)_2\text{(aq)} \rightarrow$ 0 Yes. Al is above Pb in the activity series. $2\text{Al(s)} + 3\text{Pb(NO}_3)_2\text{(s)} \rightarrow 0 + 3\text{Pb(s)} + 2\text{Al(NO}_3)_3\text{(aq)}$ Write the balanced chemical equations for the following double-replacement reactions. $\text{LiI(aq)} + \text{AgNO}_3\text{(aq)} \rightarrow$ 25.

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In-chapter exercises. E9.1: The electrophilic carbon is primary, and a carbocation intermediate would not be stabilized by resonance. Therefore it is reasonable to predict an $\text{S}_\text{N}2$ mechanism.. Note: in later chapters, especially in chapters 10 and 12, we will see many more reactions involving ATP.As we will see then, the SAM-forming reaction above is somewhat unusual – most reactions with ...

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Q: What are the concentrations of each of the ions in a saturated solution of PbI_2 , given that the K_{sp} of PbI_2 is 8.5×10^{-9} at 25°C . If 5g PbI_2 are dissolved in water to make 1L of solution at 25°C , would it be sat, unsat, or supersat

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