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184 Guided Reading and Study Workbook CHAPTER 17, Thermochemistry (continued) 6. In thermochemical calculations, is the direction of heat flow given from the point of view of the system, or of the surroundings? 7. What universal law states that energy can neither be created nor destroyed

### **SECTION 17.1 THE FLOW OF ENERGY HEAT AND WORK (pages 505-510)**

Write your answers where indicated below. System: ... Everything but the mixture of chemicals is the surroundings, but practically speaking, the immediate vicinity of the system. Chapter 17 Thermochemistry 183. Name \_ Date \_ Class \_ CHAPTER17, Thermochemistry ... 184 Guided Reading and Study Workbook @ Name \_ Date \_ Class \_ ~ Units for ...

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Heat = Transfer of Energy The 3 methods that heat can be transferred: Conduction - heat transfers by direct contact Convection - is the process in which heat is carried from one place to another by the bulk movement of a fluid. Radiation- heat transfers by electromagnetic waves.

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Chapter 17 review Important Vocabulary • Thermochemistry- Studies of energy changes that occur during chemical reactions and change state • Chemical potential energy- The energy stored in the chemical bonds of a substance • Heat- Represented by  $q$ , is energy that transfers from one object to another

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