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Class 12 Chemistry. Solid State NCERT InText Solution. Question: 1.1 - Why are solids rigid? Answer: The particles of solids are close packed and can only oscillate about their fixed positions. These properties make solids rigid. Question: 1.2 - Why do solids have a definite volume?

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Class 12 Chemistry. Solution In Text Questions 2. Question – 2.5 - Calculate (a) molality, (b) molarity and (c) mole fraction of KI if the density of 20% (mass/mass) aqueous KI is 1.202 g mL⁻¹. Answer: Given, 20% (mass/mass) aqueous solution of KI. This means 20 gm of KI is dissolved in 80 gm of water.

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