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Temperature, pressure, and specific volume of both systems are intensive properties. Total mass

and total volume of each system are extensive property. At 100°C and 1 atm, the specific volume  $v$  of each system is 1.674 m<sup>3</sup>/kg. The mass of system (1) is  $m_1 = 2$  kg and that of system (2) is  $m_2 = 5$  kg.

**Chemical Engineering Thermodynamics II - Cal Poly Pomona**

Chemical Potential, Fugacity, Activity  
§2.2.1 & Table 2.2  $P_{s,i}$  vapor pressure of species  $i$ .  $y_i$  vapor mole fraction of species  $i$ .  $x_i$  liquid mole fraction of species  $i$ . For phase equilibrium, the chemical potential, fugacity, activity (but not activity coefficient or fugacity coefficient) of any species is equal in all phases.

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**Thermodynamics - Wikipedia**

Quiz 11 Chemical Engineering Thermodynamics March 30, 2017 Condensation of a natural gas stream (Double mixed refrigerant LNG process provides viable alternative for tropical conditions, Oil & Gas Journal, 100(27) (2002)) results in a liquid stream containing the following compositions.

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figure were observed.

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