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Saturated vs. Unsaturated Solutions overall: $\Delta H_{\text{soln}} = \Delta H_1 + \Delta H_2 + \Delta H_3$ when solutions form, there is an increase in disorder increase in entropy solution formation is a spontaneous process sol'n formation may be endothermic or exothermic Thermodynamics of Solution Formation representation of how much solute is present qualitatively: concentrated vs. dilute

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Chapter 12 Section 2 The Solution Process Factors Affecting the Rate of Dissolution •At higher temperatures, collisions between solvent molecules and solvent are more frequent and of higher energy. This helps to disperse solute molecules among the solvent molecules, and speed up the dissolving process.

Chapter 12 Section 1 Types of Mixtures Objectives

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