

## *Class Xii Chemistry Ch 2 Solutions*

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2.10 What role does the molecular interaction play in a solution of alcohol and water? Sol. Alcohol and water both have strong tendency to form intermolecular hydrogen bonding. On mixing the two, a solution is formed as a result of formation of H-bonds between alcohol and  $H_2O$  molecules but these interactions are weaker and less extensive than those in pure  $H_2O$ .

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Class XII Chemistry Ch. 2: Solutions Important formulae & Concepts  
1. Mass percentage of a component (w/w) =  $\frac{\text{Mass of component in solution}}{\text{Total mass of solution}} \times 100$   
2. Volume percentage of a component (v/v) =  $\frac{\text{Volume of the component}}{\text{Total volume of solution}} \times 100$   
3. Mole fraction of a component (x) =  $\frac{\text{Number of moles of the component}}{\text{Total number of moles of all components}}$

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