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The common-ion effect says that if you add  $\text{Cl}^-$  to the solution above, then less  $\text{AgCl}$  would dissolve. For example, if you add  $\text{NaCl}$  to a saturated solution of  $\text{AgCl}$ , then some  $\text{AgCl}$  will crash out of solution. Another example: more  $\text{AgCl}$  can dissolve in pure water than in water containing  $\text{Cl}^-$  ions.

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Solutions exist in all three states of matter, gases, liquids, and solids. What Are Some Types of Solutions? Any time that gases do not combine chemically, they exist in a solution. Air is a solution of nitrogen, oxygen, and other gases. Sea water is a solution of water, salt, and other chemical compounds.

### **Chemistry Review of Solutions | Free Homework Help**

Honors Chemistry Solutions Worksheet - Review REVIEW SOLUBILITY 1. \_\_\_\_ is the substance that is dissolved. 2. \_\_\_\_ is the substance that does the dissolving.

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All questions are based on materials that can be found on the Chemistry of Solutions Lecture Main Page. 1. In a solution, what is the general term for the substance that is dissolved? ... Review Questions and Practice Test Questions. Page last updated: 10/2015.

### **Chemistry of Solutions Practice Test Questions**

Solution Properties Review. Choose the correct answer for each question. Show all questions <=> The substance dissolved in a solution is the: ? solvent ? solution ? solute ? mixture; The dissolving medium in a solution is the: ... Some solutions will boil at a temperature below  $100^\circ\text{C}$ , and some of the solutions will boil at a temperature ...

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