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Temperature, pressure, and specific volume of both systems are intensive properties. Total mass

and total volume of each system are extensive property. At 100oC and 1 atm, the specific volume v of each system is 1.674 m3/kg. The mass of system (1) is m. 1 = 2 kg and that of system (2) is m2 = 5 kg.

Chemical Engineering Thermodynamics II - Cal Poly Pomona

Chemical Potential, Fugacity, Activity SHR §2.2.1 & Table 2.2 Ps i vapor pressure of species i. y i x i vapor mole fraction of species i. liquid mole fraction of species i. For phase equilibrium, the chemical potential, fugacity, activity (but not activity coefficient or fugacity coefficient) of any species is equal in all phases.

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Thermodynamics - Wikipedia

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Quiz 11 Chemical Engineering Thermodynamics

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figure were observed.

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