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95 Chemical Kinetics Rate of appearance of P Increase in concentration of P  $P = \text{Time taken } t$  (4.2) Since,  $\Delta[R]$  is a negative quantity (as concentration of reactants is decreasing), it is multiplied with -1 to make the rate of the reaction a positive quantity. Equations (4.1) and (4.2) given above represent the average rate of a reaction,  $r_{av}$ .

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