Chemquest 15 Answer Key

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15. Complete the following: Elements ending in d2 are in column number 4 or IVB and those ending in d4 are in column number 6 or VIB. Therefore, elements ending in d3 are in column number 5 or VB. 7 Elements ending in d must be in column number 9 or VIIIB. 16. Notice from question 6 that an element that ends in 3s is in the 3rd row. Also, from ...

ChemQuest 15 - Yav Science

 $13.1 \times .12039 = 15.77109 = 15.8.3$ significant figures 5 significant figures Final rounded answer should have 3 significant figures since 3 is the least number of significant figures in this problem. 325.714285714330144560 = 3 significant figures 2 significant figures Final rounded answer should have only 2 significant figures

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15. The charge (in the upper right hand corner of the element symbol) is -1 for a fluorine ion. Why e-neqcuhve Chœcqe isn't it +1 or some other number? 16. What is the charge on every atom? Why is this the charge? Scum? (End f 17. How do you determine the charge on an ion? 06 concl 18. An oxygen ion has a -2 charge.

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(Base the answer to this question on the number of moles of propane that actually get combusted—which is your answer to part a.) 12 moles. For every mole of propane that combusts 3 moles of CO2 are produced, so the number of moles of CO2 that can be produced when 4 moles of propane combusts = 4(3) = 12.

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15 1.6 20 2.3 25 3.2 30 4.2 35 5.6 40 7.4 45 9.8 Critical Thinking Questions 4. Why is it logical to expect that the vapor pressure of water would increase as the temperature increases? As the temperature increases, water is more likely to evaporate and the evaporated molecules move faster, colliding with more force (or pressure). 5.

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12 9. Formulate a definition for each of the following terms. a) element: Matter that is composed of only one kind of atom. b) compound: Matter that is composed of two or more kinds of atoms chemically combined together

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