

## *Chemistry Stoichiometry Problem Sheet 2 Key*

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## Chemistry Stoichiometry Problem Sheet 2

Chemistry: Stoichiometry – Problem Sheet 2 KEY 9)  $2.24 \times 10^{23}$  molecules I 1 mol I  $6.02 \times 10^{23}$  molecules I 1 mol Cl 1 mol  $71 \text{ g Cl}$   $546 \text{ g Cl}$  10)  $292 \text{ g Ag}$  1 mol Ag  $108 \text{ g Ag}$  1 mol Cu 1 mol Ag  $63.5 \text{ g Cu}$

## Stoichiometry: Problem Sheet 2 - FREE Chemistry Materials ...

Chemistry: Stoichiometry – Problem Sheet 2. Directions: Solve each of the following problems. Show your work, including proper units, to earn full credit. 1. Camels store the fat tristearin ( $\text{C}_{57}\text{H}_{110}\text{O}_6$ ) in the hump. Besides being a source of energy, the fat is a source of water for the camel because when the fat is burned, the following ...

## Chemistry: Stoichiometry - Problem Sheet 2

Stoichiometry Problems Name \_\_\_\_\_ Chem Worksheet 12-2 Stoichiometry Strategy Amount moles molar mass (g/mol)  $22.4 \text{ L/mol}$  KNOWN UNKNOWN Mass mass Volume of gas at STP Ions Particles atoms, molecules, formula units  $6.02 \times 10^{23}$  particles/mol Mass grams at STP ...

## Stoichiometry Problems Name Chem Worksheet 12-2

Unformatted text preview: Chemistry 11 Stoichiometry Worksheet 2 Name: kw; - V I. Solve the following stoichiometry problems. Write a balanced equation for each. a) How many grams of aluminum are required to produce  $415 \text{ g}$  of aluminum oxide, through reaction with oxygen gas?

## CHEM 11 Stoichiometry Worksheet 2 Answers - Chemistry 11 ...

Chemistry 12 from stoichiometry problems chem worksheet 12 2, source: colgurchemistry.com. You just compose the precise same 3 letters to on your own. In both circumstances the very initial letter is easy to compose because that will most likely reflect your state of mind right now.

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## Stoichiometry Problems Chem Worksheet 12 2 - Best Free ...

Stoichiometry: Problem Sheet 2 Subject: Chemistry Author: John Bergmann & Jeff Christopherson Keywords: stoichiometry, chemical equation Last modified by: jlbrock Created Date: 4/12/2009 7:15:00 PM Category: Stoichiometry Other titles: Stoichiometry: Problem Sheet 2

## Stoichiometry: Problem Sheet 2

View Notes - Stoichiometry Problems 2-Key from ENG 1310 at Texas State University. KEY Chemistry: Stoichiometry Problem Sheet 2 Directions: Solve each of the following problems. Show your work,

## Stoichiometry Problems 2-Key - Course Hero

Stoichiometry Worksheet #1 Answers 1. Given the following equation:  $2 \text{ C}_4\text{H}_{10} + 13 \text{ O}_2 \rightarrow 8 \text{ CO}_2 + 10 \text{ H}_2\text{O}$ , show what the following molar ratios should be. a.  $\text{C}_4\text{H}_{10} / \text{O}_2$  b.  $\text{O}_2 \dots$  2 This problem is slightly different from those above. 10. Given the reaction:  $4 \text{ NH}_3(\text{g}) + 5 \text{ O}_2$

## Stoichiometry Worksheet #1 Answers

Practice Problems: Stoichiometry. Balance the following chemical reactions: Hint a.  $\text{CO} + \text{O}_2 \rightarrow \text{CO}_2$  b.  $\text{KNO}_3 \rightarrow \text{KNO}_2 + \text{O}_2$  c.  $\text{O}_3 \rightarrow \text{O}_2$  d.  $\text{NH}_4\text{NO}_3 \rightarrow \text{N}_2\text{O} + \text{H}_2\text{O}$  e.  $\text{CH}_3\text{NH}_2 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O} + \text{N}_2$  Hint f.  $\text{Cr}(\text{OH})_3 + \text{HClO}_4 \rightarrow \text{Cr}(\text{ClO}_4)_3 + \text{H}_2\text{O}$  Write the balanced chemical equations of each reaction:

## Practice Problems: Stoichiometry - Department of Chemistry

Chemistry: Stoichiometry – Problem Sheet 2 Directions: Solve each of the following problems. Show

your work, including proper units, to earn full credit. 1.  $\text{CaCl}_2 + \text{AgNO}_3 \rightarrow \text{Ca(NO}_3)_2 + \text{AgCl}$  How many grams of silver chloride are produced when 45 g of calcium chloride react with excess silver nitrate? 2.  $\text{CuO} + \text{H}_2$

**Stoichiometry: Problem Sheet 2 - FREE Chemistry Materials ...**

Chemistry: Stoichiometry - Practice Problem Sheet 2 Directions: Solve each of the following problems. Show your work, including proper units 1.  $\text{CaCl}_2 + \text{AgNO}_3 \rightarrow \text{Ca(NO}_3)_2 + \text{AgCl}$  How many grams of silver chloride are produced when 45 g of calcium chloride react with excess silver nitrate? 2.  $\text{CuO} + \text{H}_2$

**Stoichiometry: Problem Sheet 2 - Weebly**

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**ANSWER KEY for Stoichiometry Review - chemistrygods.net**

CHEMISTRY COMPUTING FORMULA MASS WORKSHEET Problem Set-up example: Find the formula mass of  $\text{Ca(NO}_3)_2$  ... A flow chart for solving stoichiometry problems: I II III IV Sample Problem What mass, in grams, of  $\text{KClO}_3$  is consumed when 90 grams of  $\text{O}_2$  is produced according to the following reaction: ... 3.37 b) 7.02 c) 5.26 d) 2.11 e) 4.21 CHEMISTRY ...

**CHEMISTRY COMPUTING FORMULA MASS WORKSHEET**

Stoichiometry Worksheet W 3222 Everett Community College Student Support Services Program 1) Write a balanced equation for the reaction of sulfuric acid with gallium hydroxide to form water and gallium sulfate: 2) From the equation in part 1, determine the mass of gallium sulfate that

**Stoichiometry Worksheet 2 - Everett Community College**

Chemistry 801: Mole/Mole and Mole/Mass Stoichiometry Problems Instructions Before viewing an episode, download and print the note-taking guides, worksheets, and lab data sheets for that episode, keeping the printed sheets in order by page number.

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Stoichiometry: Mixed Problems (KEY) 1)  $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$  What volume of  $\text{NH}_3$  at STP is produced if 25.0 of  $\text{N}_2$  is reacted with an excess of  $\text{H}_2$ ? 3 3 3 2 3 2 2 2 40.0L  $\text{NH}_3$  1mol  $\text{NH}_3$  22.4L  $\text{NH}_3$  1mol N 2mol  $\text{NH}_3$  28.0g N 25.0g N 1mol N  $\times \times \times =$  2)  $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$  If 5.0g of  $\text{KClO}_3$  is decomposed, what volume of  $\text{O}_2$  is produced at STP? 2

**Stoichiometry: Mixed Problems (KEY)**

$\text{Fe}_2\text{O}_3 + 3\text{H}_2 \rightarrow 2\text{Fe} + 3\text{H}_2\text{O}$  . Calculate how many grams of iron can be made from 16.5 grams of  $\text{Fe}_2\text{O}_3$  by the following equation. Worksheet for Basic Stoichiometry. Part 1: Mole  $\leftrightarrow$  Mass Conversions. Convert the following number of moles of chemical into its corresponding mass in grams. 1. 0.436 moles of ammonium chloride. 2.

**Worksheet for Basic Stoichiometry**

Chapter 6 Balancing and Stoichiometry Worksheet and Key Topics: • Balancing Equations • Writing a chemical equation • Stoichiometry Practice: 1. In the reaction:  $4\text{Li (s)} + \text{O}_2 \text{ (g)} \rightarrow 2\text{Li}_2\text{O (s)}$  a. what is the product? b. what are the reactants? c. what does the "(s)" after the formula of lithium oxide signify?

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