

Chemistry Stoichiometry Problem Sheet 2 Answers

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Chemistry Stoichiometry Problem Sheet 2

Chemistry: Stoichiometry – Problem Sheet 2 KEY 9) 2.24×10^{23} molecules I 1 mol I 6.02×10^{23} molecules I 1 mol Cl 1 mol 71 g Cl 546 g Cl 10) 292 g Ag 1 mol Ag 108 g Ag 1 mol Cu 63.5 g Cu

Stoichiometry: Problem Sheet 2 - FREE Chemistry Materials ...

Chemistry: Stoichiometry – Problem Sheet 2. Directions: Solve each of the following problems. Show your work, including proper units, to earn full credit. 1. Camels store the fat tristearin ($\text{C}_{57}\text{H}_{110}\text{O}_6$) in the hump. Besides being a source of energy, the fat is a source of water for the camel because when the fat is burned, the following ...

Chemistry: Stoichiometry - Problem Sheet 2

Stoichiometry Problems Name _____ Chem Worksheet 12-2 Stoichiometry Strategy Amount moles molar mass (g/mol) 22.4 L/mol KNOWN UNKNOWN Mass mass Volume of gas at STP Ions Particles atoms, molecules, formula units 6.02×10^{23} particles/mol Mass grams at STP ...

Stoichiometry Problems Name Chem Worksheet 12-2

Unformatted text preview: Chemistry 11 Stoichiometry Worksheet 2 Name: kw; - V I. Solve the following stoichiometry problems. Write a balanced equation for each. a) How many grams of aluminum are required to produce 415 g of aluminum oxide, through reaction with oxygen gas?

CHEM 11 Stoichiometry Worksheet 2 Answers - Chemistry 11 ...

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Stoichiometry: Problem Sheet 2 Subject: Chemistry Author: John Bergmann & Jeff Christopherson Keywords: stoichiometry, chemical equation Last modified by: jlbrock Created Date: 4/12/2009 7:15:00 PM Category: Stoichiometry Other titles: Stoichiometry: Problem Sheet 2

Stoichiometry: Problem Sheet 2

View Notes - Stoichiometry Problems 2-Key from ENG 1310 at Texas State University. KEY Chemistry: Stoichiometry Problem Sheet 2 Directions: Solve each of the following problems. Show your work,

Stoichiometry Problems 2-Key - Course Hero

Stoichiometry Worksheet #1 Answers 1. Given the following equation: $2 \text{C}_4\text{H}_{10} + 13 \text{O}_2 \rightarrow 8 \text{CO}_2 + 10 \text{H}_2\text{O}$, show what the following molar ratios should be. a. $\text{C}_4\text{H}_{10} / \text{O}_2$ b. $\text{O}_2 \dots$ 2 This problem is slightly different from those above. 10. Given the reaction: $4 \text{NH}_3(\text{g}) + 5 \text{O}_2$

Stoichiometry Worksheet #1 Answers

Practice Problems: Stoichiometry. Balance the following chemical reactions: Hint a. $\text{CO} + \text{O}_2 \rightarrow \text{CO}_2$ b. $\text{KNO}_3 \rightarrow \text{KNO}_2 + \text{O}_2$ c. $\text{O}_3 \rightarrow \text{O}_2$ d. $\text{NH}_4\text{NO}_3 \rightarrow \text{N}_2\text{O} + \text{H}_2\text{O}$ e. $\text{CH}_3\text{NH}_2 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O} + \text{N}_2$ Hint f. $\text{Cr}(\text{OH})_3 + \text{HClO}_4 \rightarrow \text{Cr}(\text{ClO}_4)_3 + \text{H}_2\text{O}$ Write the balanced chemical equations of each reaction:

Chemistry Stoichiometry Problem Sheet 2 Answers

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