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A laboratory manager wants to prepare 250.00 mL of a buffer with a pH of 2.813 from equimolar solutions of  $\text{C}_9\text{H}_7\text{O}_4\text{H}$  ( $K_a = 3.000 \times 10^{-4}$ ) and  $\text{LiC}_9\text{H}_7\text{O}_4$ . What volume of the  $\text{C}_9\text{H}_7\text{O}_4\text{H}$  solution would be required? I need help with the math on this problem, I know it has something to do with the solutions being equimolar but I'm brain dead at this point.

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