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142 Guided Reading and Study Workbook CHAPTER 13, States of Matter (continued) SECTION 13.3 THE NATURE OF SOLIDS (pages 396-399) This section describes the highly organized structures of solids, distinguishes between a crystal lattice and a unit cell, and explains how allotropes of an element differ. A Model for Solids (page 396) 1.

Name Date Class STATES OF MATTER 13

Collins CSEC® Chemistry Workbook answers A1 States of matter. 1. a) i) Ammonium chloride (1) ii) Diffusion Diffusion is the movement of particles from an area of higher concentration to an area of lower concentration until the particles are evenly distributed.

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Energy and states of matter. Solids, liquids, and gases are the most common states of matter that exist on our planet. If you would like to compare the three states to one another, click on the comparison animation below. Note the differences in molecular motion of water molecules in these three states.

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