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In a full glass of water, the water surface seems to bulge over the rim of the glass. Water forms nearly spherical drops at the end of an eyedropper. An insect called a water strider is able to "walk" on water.

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water molecules: if the attraction of the water molecules for the positive ion is stronger than the attraction of the negative ions near it in the crystal, the ion leaves the crystal and enters solution. After entering solution, the dissolved ion is surrounded completely by water molecules, which tends to prevent the ion from reentering the ...

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Chapter 15 Chemical Equilibrium. Equilibrium The Concept of Equilibrium Chemical equilibrium occurs when a reaction and its reverse reaction proceed at the same rate. $\text{N}_2\text{O}_4 (\text{g}) \rightleftharpoons 2 \text{NO}_2 (\text{g})$ Equilibrium ... water to give a saturated solution. Example: $\text{Zn} + \text{PbCl}_2 (\text{s}) \rightarrow \text{ZnCl}_2 (\text{aq}) + \text{Pb} (\text{s})$...

Chapter 15 Chemical Equilibrium

and water molecules are also hydrogen bonds (Figure 15.4). Because the attractions between the particles are so similar, the freedom of movement of the ethanol molecules in the water solution is about the same as their freedom of movement in the pure ethanol. The same can be said for the water. Because of this 576 Chapter 15 Solution Dynamics

Chapter 15 Solution Dynamics - An Introduction to Chemistry

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