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1.8. Ionic solids conduct electricity in molten state but not in solid state. Ans: In solid state, the ions cannot move, they are held by strong electrostatic forces of attraction. So, ionic solids do not conduct electricity in solid state. However, in the molten state, they dissociate to give free ions and hence conduct electricity.

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NCERT Solutions for Class 12 Chemistry Chapter 1 The Solid State. When the crystal of a group 14 element such as Si or Ge is doped with a group 15 element such as P or As, an n-type semiconductor is generated. Si and Ge have four valence electrons each. In their crystals, each atom forms four covalent bonds.

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CBSE Class-12 Subject Chemistry NCERT Solutions Chapter - 1 The Solid State NCERT Solutions For Class 12 Chemistry Chapter 1 The Solid State NCERT TEXTBOOK QUESTIONS SOLVED 1.1. Why are solids rigid? Ans: The constituent particles in solids have fixed positions and can oscillate about their mean positions. Hence, they are rigid. 1.2. Why do solids have [...]

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NCERT Solutions class 12 Chemistry The Solid State part 2. 14. If the radius of the octahedral void is r and radius of the atoms in close packing is R , derive relation between r and R . Ans. A sphere with centre O , is fitted into the octahedral void as shown in the above figure.

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