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A particular gas is enclosed in a cylinder with a moveable piston. It is observed that if the walls are adiabatic, a quasi-static increase in volume results in a decrease in pressure according to the equation $p \propto V^{-\gamma}$) Find the quasi-static work done on the system and the net heat transfer to the system ...

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Hint: You will want to refer to your previous work on problem # 3 above. 7. Thermodynamics and chemical reactions Callen 6.4-1 One half mole of H_2S , three-quarters of a mole of H_2O , 2 moles of H_2 and one mole of SO_2 are allowed to react in a vessel at a fixed temperature of $T = 300\text{K}$ and pressure of $P = 104\text{Pa}$. The components can react by the ...

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Part 1 of lecture 5. Thermodynamics of solutions. Enthalpy of mixing 4:56 Entropy of Mixing 24:14 Gibb's Energy of Mixing (The Regular Solution Model) 39:56 ...

5.1 | MSE104 - Thermodynamics of Solutions

Answers to Problems 381 2. At equilibrium, $\Delta G = 0$, that is, $T_f = T_0 f - 2\gamma(T_0 f / L f)$. As $L f > 0$, the melting point of the crystal will always be less than the melting point $T_0 f$ of the ideal crystal in the absence of a surface.

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