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#### **Ch 22 Hydrocarbon Compounds Workbook Answers**

Chapter 22 Hydrocarbon Compounds237. SECTION 22.1 HYDROCARBONS(pages 693–701) This section describes the bonding in hydrocarbons and distinguishes straight-chain from branched-chain alkanes. It also provides rules for naming branch-chained alkanes.

#### SECTION 22.1 HYDROCARBONS(pages 693-701)

Chapter 22 - Hydrocarbon Compounds. STUDY. PLAY. hydrocarbon. an organic compound that contains only carbon and hydrogen. alkane. a hydrocarbon containing only single covalent bonds; saturated hydrocarbons. straight-chain alkane. a saturated hydrocarbon that contains any number of carbon atoms arranged one after another in a chain.

#### Chapter 22 - Hydrocarbon Compounds Flashcards | Quizlet

Chapter 22: Hydrocarbon Compounds. A hydrocarbon that contains one or more carbon-carbon double covalent bonds. At least one carbon-carbon bond in an alkene is a double covalent bond. Other bonds may be single carbon-carbon bonds and carbon-hydrogen bonds. Ethene (C2H4) is the simplest alkene.

## Chapter 22: Hydrocarbon Compounds Flashcards | Quizlet

The Hydrocarbon Compounds chapter of this Prentice Hall Chemistry Companion Course helps students learn the essential lessons associated with hydrocarbon compounds.

## Prentice Hall Chemistry Chapter 22: Hydrocarbon Compounds ...

Chapter 22 Hydrocarbon Compounds. How is the number of valence electrons in carbon atoms related to the bonds that carbon atoms form? because carbon has four valence electrons, a carbon atom always forms four covalent bonds. What are two possible arrangements of carbon atoms in an alkane?

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## 22.4 Hydrocarbon Rings 22 - mrmiller-science.weebly.com

BIG Idea Organic compounds called hydrocarbons differ by their types of bonds. 21.1 Introduction to Hydrocarbons. MAIN Idea Hydrocarbons are carbon-containing organic compounds that provide a source of energy and raw materials. 21.2 Alkanes. MAIN Idea Alkanes are hydrocarbons that contain only single bonds.

#### **Chapter 21: Hydrocarbons**

3) Hydrocarbons are made of carbon and hydrogen. The alkanes are a hydrocarbon family with the general formula CnH2n + 2, where n is the number of carbon atoms in the compound. The suffix -ane denotes an alkane; adding a prefix to the -ane suffix gives the number of carbon atoms in the straight chain alkane.

#### Chapter 22

CHAPTER 22, Hydrocarbon Compounds (continued) SECTION 22.4 HYDROCARBON RINGS (pages 709-711) This section escribes how to identify and classify cyclic hydrocarbons. It also explains the bonding in benzene. Cyclic Hydrocarbons (page 709) 1. What is a cyclic hydrocarbon? It is a

compound that contains a hydrocarbon ring. 2.

### This section describes the bonding in hydrocarbons and ...

MAIN Idea A halogen atom can replace a hydrogen atom in some hydrocarbons. 22.2 Alcohols, Ethers, and Amines. MAIN Idea Oxygen and nitrogen are two of the most-common atoms found in organic functional groups. 22.3 Carbonyl Compounds. MAIN Idea Carbonyl compounds contain a double-bonded oxygen in the functional group.

#### **Chapter 22: Substituted Hydrocarbons and Their Reactions**

Hydrocarbon Compounds 693 22.1 FOCUS Objectives 22.1.1 Describe the relationship between number of valence electrons and bonding in carbon. 22.1.2 Define and describe alkanes. 22.1.3 Relate the polarity of hydrocar- bons to their solubility.

## 22.1 Hydrocarbons 22 - mrmiller-science.weebly.com

• At least one carbon-carbon bond in an alkene is a double bond. Other bonds may be single carbon-carbon and carbon- hydrogen bonds. • At least one carbon-carbon bond in an alkyne is a triple bond. Other bonds may be single or double carbon-carbon bonds and single carbon-hydrogen bonds.

#### **CHAPTER 22 Study Guide - Quia**

Naming Hydrocarbons Worksheet and Key Write the name of each of the hydrocarbon molecules shown below: 1) 22 2) 3 C 3) 4) 3 5) 6) 7) 3 8) 9) 10)

#### Naming Hydrocarbons Worksheet and Key - saddleback.edu

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