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1 Chapter 8 Solutions, Acids, and Bases Section 8.1 Formation of Solutions Section 8.2 Solubility and Concentration Section 8.3 Properties of Acids and Bases Section 8.4 Strength of Acids and Bases. 2 Section 8.1 Formation of Solutions For a solution to form, one substance must dissolve into another. ...

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Some general properties of acids include sour taste, reactivity with metals, and ability to produce color changes in indicators. Some general properties of bases include bitter taste, slippery feel, and ability to produce color changes in indicators.

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Chapter 8 Solutions, Acids, and Bases Summary 8.1 Formation of Solutions • A solute is a substance whose particles are dissolved in a solution. • The substance in which the solute dissolves is called the solvent. Substances can dissolve in water in three ways—by dissociation,

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A substance that changes color in the presence of an acid or a base: Base: A compound that produces hydroxide ions when dissolved in water; a proton acceptor: Neutralization: A chemical reaction between an acid and a base: Salt: An ionic compound formed when an acid reacts with a base: pH: A measure of the hydronium ion concentration of a ...

Quia - Chapter 8: Solutions, Acids, and Bases

Ch 8 Solutions, Acids and Bases Study Guide Answers 1. What is a homogeneous mixture of two or more substances? Solution 2. For a solution to form, what must one substance do in another? Solute must dissolve in solvent 3. What is the maximum amount of a solute that dissolves in a given amount of solvent at a constant temperature? Solubility 4.

Ch 8 Solutions, Acids and Bases Study Guide Answers

a.Strong bases have a higher concentration of hydronium ions than pure water. b.Strong bases dissociate almost completely in water. c.Strong bases have a pH below 7. d.Examples of strong bases include sodium hydroxide and calcium hydroxide. Weak Acids and Bases (page 248) 12. What happens when weak acids and bases dissolve in water? 13.

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