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In chemistry, diffusion is defined as the movement of matter by the random motions of molecules.

In a gas or a liquid, the molecules are in perpetual motion; for instance, in a quantity of seemingly immobile air, molecules of nitrogen and oxygen are constantly bouncing off each other.

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**Chapter 7 An Introduction to Chemical Reactions**

170 Guided Reading and Study Workbook 4. Look at Figure 16.8 on page 481. Circle the letter of the best procedure for making a 0.50-molar (0.50M) solution in a 1.0-L volumetric flask. a. Add distilled water exactly to the 1.0-L mark, add 0.50 mol of solute, and

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