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Behavior of Gases Answer Key. When 3.56 grams of a compound is vaporized at 245°C and 1.20 atm, the resulting gas occupies a volume of 344 mL. Based on this result, what is the molar mass of the compound? (R=0.0821 L•atm/mol•K) Award credit for a student showing work AND arriving at a correct answer, as shown below: [math] [(1.2 atm)...

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Chapter 14 The Behavior of Gases147 SECTION 14.1 PROPERTIES OF GASES(pages 413-417) This section uses kinetic theory to explain the properties of gases. This section also explains how gas pressure is affected by the amount of gas, its volume, and its temperature. Compressibility (pages 413-414) 1. Look at Figure 14.1 on page 413.

SECTION 14.1 PROPERTIES OF GASES(pages 413-417)

P-olai_n each assumption of the kinetic theory of gases in your own words. phrase, or number. , or squeezed into a smaller volume Gases are easily 1 because of the 2 between particles in a' gas. The four variables used to describe a gas are pressure, (P), 3 4. 2 S 5. 8. dec a. g h5 No are c n and number of (n).

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Prentice Hall Chemistry Chapter 14: The Behavior of Gases Chapter Exam Instructions. Choose your answers to the questions and click 'Next' to see the next set of questions.

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a measure of how much the volume of matter decreases under pressure. Gases are easily compressed because. of the space between the particles in a gas. The amount of gas, volume, and temperature are factors that. affect gas pressure. If the temperature is constant, as the pressure of a gas increases, the volume decreases.

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Chapter 1. 1.1 The Scope of Chemistry Section 1.1 Exercises p.1 1.2 Chemistry and You Section 1.2 Exercises p.4 1.3 Thinking Like a Scientist Section 1.3 Exercises p.6 1.4 Problem Solving in Chemistry Section 1.4 Exercises p.9 Chapter 1 Self-Check Activity p.11.

Solutions to Chemistry: Reading and Study Workbook ...

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Chapter 14

Combined Gas Law: PV/T = constant. V1 n1. = V2 n2 ANSWERS. Gas Behavior and Gas Laws Worksheet (answers listed below), Gas Laws and Sig Fig Physical and Chemical Change: Behavior of Gases click. here. Chemistry: Gas Laws Worksheet gas law that describes the behavior of gases in relation to temperature, pressure, and Answers to practice problems.

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Section 3.1 Solids, Liquids, and Gases (pages 68–73) This section explains how materials are classified as solids, liquids, or gases. It also describes the behavior of these three states of matter. Reading Strategy (page 68) Comparing and Contrasting As you read about the states of matter,

Chapter 3 States of Matter Section 3.1 Solids, Liquids ...

• Relate the total pressure of a mixture of gases to the partial pressures of the component gases • Explain how the molar mass of a gas affects the rate at which

05 CTR ch14 7/12/04 8:13 AM Page 347 THE PROPERTIES OF ...

Behavior Of Gases Workbook Answers Lesson 1: Molecular Weights and Mixtures of Gases. In this last section, we'll study how to determine the molecular weight of a gas and also look at one more gas law, Dalton's Law of Partial Pressures, which has to do with mixtures of gases.Lesson 1:

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