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160 Guided Reading and Study Workbook Name \_\_\_\_\_ Date \_\_\_\_\_ Class \_\_\_\_\_ CHAPTER 15, Water and Aqueous Systems(continued) 6. Circle the letter next to each sentence that describes a result of the surface tension of water. a. In a full glass of water, the water surface seems to bulge over the rim of the glass.

### **SECTION 15.1 WATER AND ITS PROPERTIES (pages 445-449)**

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solutions when added to water. (i) Write an equation for the reaction of each acid with water. Hydrogen fluoride, HF, with water: Hydrogen bromide, HBr, with water: Chemistry 3.6 AS 91392

Demonstrate understanding of equilibrium principles in aqueous systems Candidates are expected to recognise common strong acids (HCl, HBr, HNO<sub>3</sub>, H<sub>2</sub>SO<sub>4</sub> +

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Water and Aqueous Systems. STUDY. PLAY. mixture in which particle size averages between 1 nm and 1000 nm. colloid. interferes with hydrogen bonding between water molecules. surfactant. colloid of a liquid in a liquid. emulsion. substance that is dissolved in a liquid. solute.

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