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2. Solutions are _____ mixtures made up of very small particles that are actually molecules, _____, or _____. 3. Solutions are said to be in a _____ phase even though the components may have been in different phases before the solution was formed. 4. Pure gold is 24 carat.

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At the completion of this episode's lesson(s), you should be able to:

- Describe and explain the process of forming a solution using the terms solute and solvent.
- Compare the properties of suspensions, colloids, and solutions.
- List the factors that affect the rate at which a solid solute dissolves in a liquid solvent.

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