# Demo

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Some example calls of NH3/CO2/slurry pH model.

Load all functions

```
source('NH3_mods.R')
Use eqSpec to calculate equilibrium speciation
eq <- eqSpec(tot = c(H. = -0.02, NH3 = 0.1, H2CO3 = 0.1, K. = 0.03,
                      Na. = 0.02, Cl. = 0.03, HAc = 0.0242),
             temp.c = 20, of = 'all')
eq
## $m
##
             Η.
                          NH3
                                      H2C03
                                                      C<sub>02</sub>
                                                                     Κ.
                                                                                  Na.
## 3.906191e-08 8.545425e-04 8.861097e-06 5.316655e-03 3.000000e-02 2.000000e-02
##
            Cl.
                          {\tt HAc}
                                        OH.
                                                     NH4.
                                                                  HCO3.
                                                                                CO3.2
## 3.000000e-02 3.163202e-05 2.958420e-07 9.914546e-02 9.437214e-02 3.023479e-04
##
## 2.416837e-02
##
## $a
##
             Η.
                          NH3
                                      H2C03
                                                      C02
                                                                     Κ.
## 3.165828e-08 8.844623e-04 9.171347e-06 5.502805e-03 2.165735e-02 1.493057e-02
                                                     NH4.
                                                                  HCO3.
##
                          HAc
                                        OH.
## 2.165735e-02 3.273954e-05 2.166467e-07 7.046144e-02 7.225524e-02 9.614624e-05
            Ac.
## 1.814904e-02
##
## $g
##
                    NH3
                            H2C03
                                         C02
                                                     Κ.
                                                               Na.
                                                                         Cl.
                                                                                    HAc
          Η.
## 0.8104643 1.0350126 1.0350126 1.0350126 0.7219115 0.7465283 0.7219115 1.0350126
##
         OH.
                   NH4.
                            HCO3.
                                       CO3.2
## 0.7323053 0.7106875 0.7656417 0.3179987 0.7509419
##
## $i
## [1] 0.1494478
##
## $1.m
##
                    NH3
                            H2C03
                                         C02
                                                                         Cl.
                                                                                    HAc
          Η.
                                                     Κ.
                                                               Na.
## -7.408247 -3.068266 -5.052513 -2.274362 -1.522879 -1.698970 -1.522879 -4.499873
                            HCO3.
                                       CO3.2
         OH.
                   NH4.
## -6.528940 -1.003727 -1.025156 -3.519493 -1.616753
##
```

```
## $1.a
                                       C02
##
                   NH3
                           H2C03
                                                                      Cl.
          Η.
                                                  Κ.
                                                           Na.
                                                                                HAc
  -7.499513 -3.053321 -5.037567 -2.259416 -1.664395
##
                                                     -1.825924 -1.664395 -4.484927
##
         OH.
                  NH4.
                           HCO3.
                                     CO3.2
                                                 Ac.
##
   -6.664248 -1.152048 -1.141131 -4.017068 -1.741146
##
## $1.g
##
            Η.
                       NH3
                                 H2C03
                                               C02
                                                            Κ.
                                                                        Na.
##
  -0.09126614
                0.01494565
                            0.01494565
                                        0.01494565 -0.14151601 -0.12695373
##
           Cl.
                       HAc
                                   OH.
                                              NH4.
                                                         HCO3.
                                                                      CO3.2
##
   -0.14151601
                ##
           Ac.
##
  -0.12439365
##
## $tot
##
        Η.
               NH3
                     H2C03
                                К.
                                       Na.
                                               Cl.
                                                        HAc
            0.1000
                   0.1000 0.0300
                                    0.0200
                                            0.0300
                                                    0.0242
##
  -0.0200
##
##
  $totk
##
             Η.
                         NH3
                                    H2C03
                                                   C<sub>02</sub>
                                                                 Κ.
                                                                              Na.
##
   -0.020000000
                 0.10000000
                              0.094683345
                                           0.005316655
                                                        0.030000000
                                                                     0.020000000
##
            Cl.
                         HAc
   0.03000000
##
                0.024200000
##
## $cb
##
   [1] 1.197548e-09
##
## $i.its
##
  [1] 2
##
## $p.CO2
  [1] 0.1406605
##
## $p.NH3.NH3
## [1] 1.146077e-05
```

Some info on arguments: tot = molalities (concentration in mol/kg water) of "components" (elements here, but not below, each represented by a "master species"), temp.c = temperature in C, of = output format ('all' is the most complete).

Here is some info on the ouput from eqSpec

#### names(eq)

```
## [1] "m" "a" "g" "i" "l.m" "l.a" ## [7] "l.g" "tot" "totk" "cb" "i.its" "p.CO2" ## [13] "p.NH3.NH3"
```

m= molalities of species (conc. in mol/kg water), a= activities of species, g= activity coefficients, i= ionic strength, l.m= log molalities, l.a= log activities, tot = molalities of components (e.g., H2CO3=TIC, NH3=TAN), totk = similar to tot but here H2CO3 (i.e., H2CO3, HCO3-, and CO3-2 species) and CO2 (only CO2 (aq)) are listed separately—totk is only used to get totals for kinetic model where there are two IC components, cb= charge (un)balance, i.its = iterations used for ionic strength to stabilize, p.CO2= equilibrium CO2 partial pressure

```
str(eq)
## List of 13
              : Named num [1:13] 3.91e-08 8.55e-04 8.86e-06 5.32e-03 3.00e-02 ...
    ..- attr(*, "names")= chr [1:13] "H." "NH3" "H2CO3" "CO2" ...
##
              : Named num [1:13] 3.17e-08 8.84e-04 9.17e-06 5.50e-03 2.17e-02 ...
##
    ..- attr(*, "names")= chr [1:13] "H." "NH3" "H2CO3" "CO2" ...
##
              : Named num [1:13] 0.81 1.035 1.035 1.035 0.722 ...
     ..- attr(*, "names")= chr [1:13] "H." "NH3" "H2CO3" "CO2" ...
##
##
   $ i
              : num 0.149
   $ 1.m
##
              : Named num [1:13] -7.41 -3.07 -5.05 -2.27 -1.52 ...
    ..- attr(*, "names")= chr [1:13] "H." "NH3" "H2CO3" "CO2" ...
              : Named num [1:13] -7.5 -3.05 -5.04 -2.26 -1.66 ...
##
    ..- attr(*, "names")= chr [1:13] "H." "NH3" "H2CO3" "CO2" ...
##
   $ 1.g
             : Named num [1:13] -0.0913 0.0149 0.0149 0.0149 -0.1415 ...
##
    ..- attr(*, "names")= chr [1:13] "H." "NH3" "H2CO3" "CO2" ...
##
            : Named num [1:7] -0.02 0.1 0.1 0.03 0.02 0.03 0.0242
   $ tot
    ..- attr(*, "names")= chr [1:7] "H." "NH3" "H2CO3" "K." ...
##
            : Named num [1:8] -0.02 0.1 0.09468 0.00532 0.03 ...
    ..- attr(*, "names")= chr [1:8] "H." "NH3" "H2CO3" "CO2" ...
##
   $ cb
              : num 1.2e-09
##
   $ i.its
              : num 2
  $ p.CO2
            : num 0.141
   $ p.NH3.NH3: num 1.15e-05
eq
## $m
                         NH3
                                    H2C03
                                                   C02
                                                                 Κ.
##
             Η.
## 3.906191e-08 8.545425e-04 8.861097e-06 5.316655e-03 3.000000e-02 2.000000e-02
            Cl.
                         HAc
                                      OH.
                                                  NH4.
                                                              HCO3.
## 3.000000e-02 3.163202e-05 2.958420e-07 9.914546e-02 9.437214e-02 3.023479e-04
##
## 2.416837e-02
##
## $a
##
            Η.
                         NH3
                                   H2C03
                                                   C02
## 3.165828e-08 8.844623e-04 9.171347e-06 5.502805e-03 2.165735e-02 1.493057e-02
                        HAc
                                     OH.
                                                 NH4.
                                                              HCO3.
## 2.165735e-02 3.273954e-05 2.166467e-07 7.046144e-02 7.225524e-02 9.614624e-05
           Ac.
## 1.814904e-02
##
## $g
                           H2C03
##
         Η.
                  NH3
                                       C02
                                                  Κ.
                                                           Na.
## 0.8104643 1.0350126 1.0350126 1.0350126 0.7219115 0.7465283 0.7219115 1.0350126
                 NH4.
                           HCO3.
                                    CO3.2
                                                 Ac.
## 0.7323053 0.7106875 0.7656417 0.3179987 0.7509419
## $i
## [1] 0.1494478
##
## $1.m
##
                   NH3
                           H2C03
                                       C02
                                                  Κ.
                                                                     Cl.
## -7.408247 -3.068266 -5.052513 -2.274362 -1.522879 -1.698970 -1.522879 -4.499873
```

```
##
          OH.
                   NH4.
                              HCO3.
                                         CO3.2
## -6.528940 -1.003727 -1.025156 -3.519493 -1.616753
##
## $1.a
##
           Η.
                     NH3
                              H2C03
                                           C02
                                                       Κ.
                                                                 Na.
                                                                            Cl.
                                                                                       HAc
   -7.499513 -3.053321 -5.037567 -2.259416 -1.664395
                                                           -1.825924 -1.664395 -4.484927
##
##
          OH.
                   NH4.
                              HCO3.
                                         CO3.2
                                                      Ac.
##
   -6.664248 -1.152048 -1.141131 -4.017068 -1.741146
##
##
   $1.g
##
             Η.
                         NH3
                                    H2C03
                                                    C<sub>02</sub>
                                                                  Κ.
                                                                               Na.
   -0.09126614
                 0.01494565
                                            0.01494565 -0.14151601 -0.12695373
##
                               0.01494565
##
            Cl.
                         HAc
                                       OH.
                                                   NH4.
                                                               HCO3.
                                                                            CO3.2
##
   -0.14151601
                 0.01494565 - 0.13530780 - 0.14832132 - 0.11597444 - 0.49757462
##
            Ac.
   -0.12439365
##
##
   $tot
                                   К.
                                                    Cl.
##
                       H2C03
                                                             HAc
        Η.
                NH3
                                           Na.
##
   -0.0200
             0.1000
                      0.1000
                               0.0300
                                       0.0200
                                                0.0300
                                                         0.0242
##
   $totk
##
                                                        C<sub>02</sub>
##
              Η.
                           NH3
                                       H2C03
                                                                        Κ.
                                                                                     Na.
   -0.020000000
                  0.100000000
                                 0.094683345
                                               0.005316655  0.030000000  0.020000000
##
##
             Cl.
                           HAc
##
    0.030000000
                  0.024200000
##
## $cb
   [1] 1.197548e-09
##
##
## $i.its
## [1] 2
##
## $p.CO2
##
   [1] 0.1406605
##
## $p.NH3.NH3
## [1] 1.146077e-05
If you specify a pH, KOH or HAc (acetic acid) will be added to match it, so this call is essentially identical
eq <- eqSpec(tot = c(H. = -0.02, NH3 = 0.1, H2CO3 = 0.1,
                       K. = 0.03, Na. = 0.02, C1. = 0.03,
                       HAc = 0.0),
              temp.c = 20, pH = 7.5, of = 'all')
```

Here we will extract total component concentrations for use in kinetic model (only difference from tot argument in call above is partitioning of H2CO3)

```
itot <- eq$totk
```

itot is initial total concentration of components (including distribution of IC between H2CO3 and CO2, which are not necessarily in equilibrium in simulation below)

```
## H. NH3 H2CO3 CO2 K. Na.
```

```
## -0.020000000 0.100000000 0.118775663 0.006661572 0.030000000 0.020000000  ## Cl. HAc  ## 0.030000000 0.000000000
```

Now for the emission model (note that there is also an equilibrium version eqEmisDiffMod()) Next call predicts changes in solution chemistry as CO2 and NH3 are emitted from surface, assuming no carbonic anhydrase (that what the "kin" is for–kinetic)

May take 30 sec to run

```
pred <- kinEmisDiffMod(c.thk = c(rep(1E-5, 5), rep(1E-4, 5), rep(9.45E-4, 10)),

h.m = 1E-3, p.CO2.a = 4E-6,

times = c(0:5*600, 1:12*3600), temp.c = 20, tot = itot)
```

Some info on arguments: c.thk = a vector of cell thicknesses in m (more and smaller is more accurate and slower), h.m = convective mass transfer coefficient in air for NH3 (m/s), p.CO2.a = ambient CO2 partial pressure (atm), temp.c = temperature in C, tot = molalities of components (usual length 8 vector and MUST be in order: "H.", "NH3", "H2CO3", "CO2", "K.", "Na.", "Cl.", "HAc")

Info on the ouput Complete output

#### names(pred)

```
## [1] "times" "pos" "tot.k" "tot.f" "act" "mol" ## [7] "ph" "kin.CO2" "emis" "exe.time" "n.calls" "pars" ## [13] "summ"
```

Extact times

```
tt <- pred$times
```

position (center of cells) converted to mm

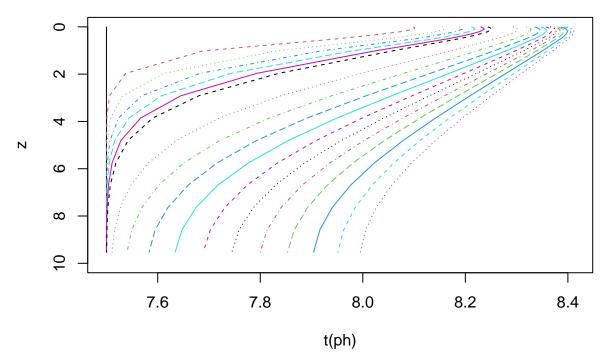
```
z <- pred$pos*1000
```

and pH

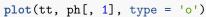
```
ph <- pred$ph
```

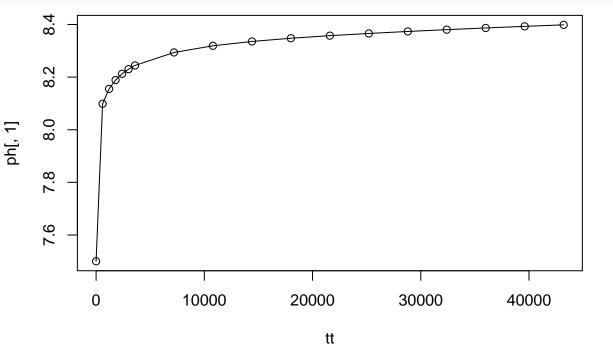
Plot pH

```
matplot(t(ph), z, type = 'l', ylim = c(10, 0))
```



"Surface" pH over time





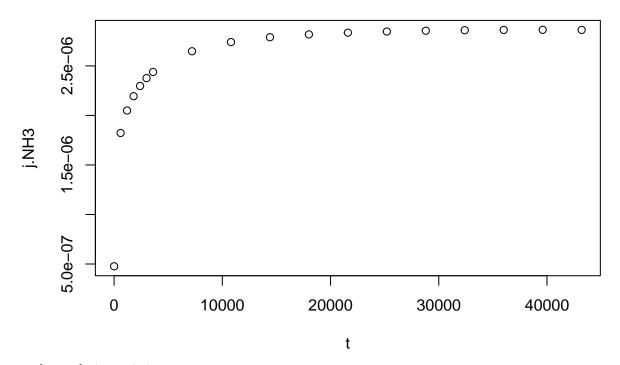
Get fluxes

# pred\$emis

```
## 1800
        1800 2.194284e-06 7.418364e-06 0.0033462950 0.018898793 0.0033462950
        2400 2.297832e-06 6.923425e-06 0.0046954424 0.023191989 0.0046954424
## 2400
## 3000
       3000 2.376499e-06 6.574302e-06 0.0060987331 0.027235969 0.0060987331
## 3600
        3600 2.438783e-06 6.310321e-06 0.0075440466 0.031097688 0.0075440466
## 7200
         7200 2.647830e-06 5.452464e-06 0.0167590255 0.052029393 0.0167590255
## 10800 10800 2.740499e-06 5.037332e-06 0.0264772148 0.070844197 0.0264772148
## 14400 14400 2.789642e-06 4.770487e-06 0.0364398912 0.088469836 0.0364398912
## 18000 18000 2.818012e-06 4.573661e-06 0.0465381728 0.105273518 0.0465381728
## 21600 21600 2.835616e-06 4.415350e-06 0.0567174553 0.121443702 0.0567174553
## 25200 25200 2.847236e-06 4.280142e-06 0.0669481526 0.137089441 0.0669481526
## 28800 28800 2.855131e-06 4.160060e-06 0.0772131110 0.152278756 0.0772131110
## 32400 32400 2.860280e-06 4.051039e-06 0.0875014413 0.167056398 0.0875014413
## 36000 36000 2.863205e-06 3.950480e-06 0.0978043528 0.181456893 0.0978043528
## 39600 39600 2.864170e-06 3.856627e-06 0.1081142411 0.195507529 0.1081142411
## 43200 43200 2.863314e-06 3.768425e-06 0.1184242689 0.209230801 0.1184242689
##
           e.CO2.rel
                         p.C02.s
                                       p.NH3.s
## 0
        0.000000000 0.1762452869 1.147280e-05
        0.007107536 0.0003805818 4.385137e-05
## 1200 0.011347225 0.0003211404 4.930331e-05
## 1800 0.015066334 0.0002908621 5.278353e-05
## 2400 0.018488919 0.0002717232 5.527436e-05
## 3000 0.021712826 0.0002582229 5.716671e-05
## 3600 0.024791433 0.0002480150 5.866494e-05
## 7200 0.041478428 0.0002148424 6.369357e-05
## 10800 0.056477805 0.0001987896 6.592272e-05
## 14400 0.070529166 0.0001884709 6.710485e-05
## 18000 0.083925254 0.0001808597 6.778731e-05
## 21600 0.096816310 0.0001747380 6.821077e-05
## 25200 0.109289272 0.0001695096 6.849030e-05
## 28800 0.121398368 0.0001648661 6.868021e-05
## 32400 0.133179273 0.0001606504 6.880406e-05
## 36000 0.144659513 0.0001567619 6.887443e-05
## 39600 0.155860841 0.0001531326 6.889763e-05
## 43200 0.166801191 0.0001497219 6.887704e-05
Plot NH3 flux
```

plot(j.NH3 ~ t, data = pred\$emis)

7



and cumulative emission

