Quiz 2 (09-02-2022)

Total points 9/10

- This Quiz is for 5 Points
- Each question carries 0.5 point
- No late Submission are allowed (-1 marks for Late Submissions)

The respondent's email (choudhary.21@iitj.ac.in) was recorded on submission of this form.

/	A 300 m3 rigid tank is filled with saturated liquid-vapor mixture of water	1/1
	at 200 kPa. If 25% of the mass is liquid and the 75% of the mass is vapor,	
	the total mass in the tank is *	

- 556 kg
- 331 kg
- 300 kg
- 451 kg

Determine the specific internal energy of water at 200 kPa and 300 oC. * 1/1

- 280.88 kJ/kg
- 200.23 kJ/kg
- 3508.5 kJ/kg
- 2808.8 kJ/kg

X A 1-m3 rigid tank contains 10 kg of water (in any phase or phases) at 170°C. The pressure in the tank is *	0/1
O 2007 kPa	
738 kPa	×
70 kPa	
792 kPa	
Correct answer	
● 792 kPa	
The units of the isothermal compressibility are. *	1/1
Pa^(-1)	✓
m^(-3)	
Option 4	
m^(3)Pa^(-1)	
✓ Steam at 4 bar and 500 oC is *	1/1
Saturated	
Can't say	
○ Wet	
Superheated	✓

✓ Determine the temperature of water at a state of P = 0.5 MPa and h 2890 kJ/kg. *	= 1/1
② 216 oC	✓
200 oC	
250 oC	
○ 150 oC	
✓ The compressibility factor of any gas *	1/1
is always equal to 1	
may be less than, equal to or greater than 1 depending on the nature of gas	✓
is always greater than 1	
is always less than 1	
✓ Ideal gas law is applicable at *	1/1
O Low T, low P	
High T, high P	
O Low T, high P	
High T, low P	~

~	What would be the pressure of compressed water having specific volume (v) 0.0009996 m3/kg at 20 oC? *	1/1
0	30 MPa	
0	50 MPa	
0	20 MPa	
0	5 MPa	✓
~	A solid transformed into vapour without going to the liquid phase at. *	1/1
/	A solid transformed into vapour without going to the liquid phase at. * Below triple point	1/1
00		1/1
000	Below triple point	1/1
	Below triple point Boiling point	1/1

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