Thermodynamics (MEL2020) **Indian Institute of Technology Jodhpur**

Date: 14th January 2022 **Assignment-2 Instructions:**

Maximum points: 1

- Answer all the questions
- Please write your solutions/explanations on a paper with your handwriting
- Scan all pages as a single pdf file and upload in google classroom before 17th-01-22
- This will give you **1 point** towards your total evaluation,
- Late submission lead to deduction of half point.
- 1. The molar specific volume of a system V is defined as the ratio of the volume of the system to the number of moles of substance contained in the system. Is this an extensive or intensive property? (**0.2 P**)
- 2. Define the isothermal, isobaric, and isochoric processes. (0.2 P)
- 3. What is steady flow process? (0.2 P)
- 4. When a hydrocarbon fuel is burned, almost all of the carbon in the fuel burns completely to form CO₂ (carbon dioxide), which is the principal gas causing the greenhouse effect and thus global climate change. On average, 0.59 kg of CO₂ is produced for each kWh of electricity generated from a power plant that burns natural gas. A typical new household refrigerator uses about 700 kWh of electricity per year. Determine the amount of CO₂ production that is due to the refrigerators in a city with 300,000 households. (0.2 P)
- 5. If you would like do a metabolism (energy) analysis of a person. How would you define the system for this purpose? (0.2 P)