

**Computer Vision and Statistical Estimation Tools for In Situ,  
Imaging-based Monitoring of Particulate Populations**

by

Paul A. Larsen

A dissertation submitted in partial fulfillment  
of the requirements for the degree of

DOCTOR OF PHILOSOPHY

(Chemical Engineering)

at the

UNIVERSITY OF WISCONSIN-MADISON

2007

© Copyright by Paul A. Larsen 2007

All Rights Reserved

To Jenny

# **Computer Vision and Statistical Estimation Tools for In Situ, Imaging-based Monitoring of Particulate Populations**

Paul A. Larsen

Under the supervision of Professor James B. Rawlings

At the University of Wisconsin–Madison

Solution crystallization is a commonly used but often poorly controlled process for separating or purifying chemical species in the pharmaceutical, chemical, and food industries. The development of effective solid-phase monitoring technology is a critical step to enable better understanding and control of crystallization processes. Video imaging is a promising technology offering the potential to monitor critical solid-phase properties, including particle size distribution (PSD), shape distribution, and, in some cases, polymorphic fraction.

To address the challenges associated with effective use of video imaging for particulate processes, this thesis focuses on the following areas:

1. Developing image analysis algorithms that enable segmentation of noisy, in situ video images of crystallization processes.
2. Developing statistical estimators to overcome the sampling biases inherent in imaging-based measurement.
3. Characterizing the reliability and feasibility of imaging-based particle size distribution measurement given imperfect image analysis.

We have developed two image analysis algorithms. The first algorithm is designed to extract particle size and shape information from in situ images of suspended, high-aspect-ratio

crystals. This particular shape class arises frequently in pharmaceutical and specialty chemical applications and is problematic for conventional monitoring technologies that are based on the assumption that the particles are spherical. The second algorithm is designed to identify crystals having more complicated shapes. The effectiveness of both algorithms is demonstrated using *in situ* images of crystallization processes and by comparing the algorithm results with results obtained by human operators. The algorithms are sufficiently fast to enable real-time monitoring for typical cooling crystallization processes.

We have derived a maximum likelihood estimator to estimate the particle size distribution of needle-like particles. We benchmark the estimator against the conventional Miles-Lantu joul approach using several case studies. For needle-like particles, the MLE provides better estimates than the Miles-Lantu joul approach, but the Miles-Lantu joul approach can be applied to a wider class of shapes. Both methods assume perfect image segmentation, or that every particle appearing in the image is identified correctly.

Given that perfect image segmentation is a reasonable assumption only at low solids concentrations, we have derived a descriptor that correlates with the reliability of the imaging-based measurement (i.e. the quality of the image segmentation) based on the amount of particle overlap. Also, we have developed a practical approach for estimating the number density of particles for significant particle overlap and imperfect image analysis. The approach is developed for mono-disperse particle systems.

Finally, this thesis demonstrates the feasibility of reconstructing a particle size distribution from imaging data for a well-studied industrial crystallization process and realistic imaging conditions.



# Acknowledgments

I am indebted to a great many people for the opportunity to come to the University of Wisconsin and for the positive experience I have had while studying here. I am indebted first to God, who has given me life, health, and the ability to think and be creative. I feel a debt of gratitude to my parents and grandparents for their work and sacrifice to give me a life full of opportunity and happiness. Grandpa Arch in particular had a strong desire to pursue a PhD but was unable. I know he is pleased I have had the opportunity.

I am grateful to my advisor, Jim Rawlings, for giving me a rich and rewarding graduate experience. He has taught me how to “first think clearly, then write clearly,” how to seek and value experts without trusting them blindly, and how to make sense out of complicated and confusing problems. Jim has motivated me with high expectations but also has been supportive of my family situation. I will miss the opportunity to work with him so closely on challenging problems.

I am grateful for Nicola Ferrier’s invaluable guidance in developing effective image analysis algorithms. Nicola also served as my wife’s adviser and has been a great friend to our family. Lian Yu has freely given of his time and resources to help me carry out and analyze crystallization experiments. His graduate student, Jun Huang, has also been a great help and good friend. Professor Chuck Dyer of the Computer Science department also has provided helpful advice.

David Dahl, previously my friendly next-door neighbor and currently an assistant professor of statistics at Texas A&M, has given invaluable statistics consulting. I am also indebted to Professor Antonio Torralba of the MIT Computer Science and Artificial Intelligence Laboratory for the use of his LabelMe database and software.

Philip Dell’Orco at GlaxoSmithKline has helped me considerably by contributing the video

imaging equipment and pharmaceutical material. Hiroya Seki and Shigeharu Katsuo of the Mitsubishi Chemical Company have given me the chance to work on industrial projects that provided excellent learning opportunities. I am also grateful to other members of the Texas-Wisconsin Modeling and Control Consortium for financial support.

Despite being a long and often frustrating process, the commercialization of the SHARC algorithm has been a source of excitement and satisfaction for me. I am grateful to John Hardiman and Marnie Matt at WARF and our patent attorney Stephen Roe at Lathrop Clark for their work in patenting SHARC and M-SHARC and licensing SHARC. Eric Hukkanen, Gregor Hsiao, Paul Barrett, Ben Smith, and Nilesh Shah at Mettler-Toledo have each played important roles in this process.

I am grateful to many of my fellow graduate students for friendship and assistance. I've enjoyed immensely the opportunity to associate, both through church and through the department, with Matt Tingey, George Huber, Tommy Knotts, Ethan Mastny, Nat Fredin, Clark Miller, and Peter Ferrin. Mike Benton has been a thoughtful friend and a constant source of good conversation. The past and present members of the Rawlings group—Brian Odelson, Eric Haseltine, Aswin Venkat, Ethan Mastny, Murali Rajamani, Brett Stewart, and Rishi Amrit—have been excellent coworkers and friends. I only regret that my time with Brett and Rishi has been so short.

Mary Diaz has made my work life much more pleasant with her endless supply of candy and plastic utensils, her willing assistance with administrative details, and her friendship. I wish her and her sons Joshua and Jeremiah all the best.

Ethan Mastny and Murali Rajamani deserve special mention. Ethan has been my sounding board, my one-man audience for countless practice talks, my neighbor, and my friend. Murali has been my math consultant, my Linux troubleshooter, my office buddy, and the source of many fun conversations. I will miss them both terribly.

Finally, I am grateful most of all to my wife Jenny. Her optimism, enthusiasm, wisdom, and encouragement has made possible the happiness our family has enjoyed these past five years. I am excited and comforted to have her at my side as we move to Michigan and begin a new phase

of life. I'm grateful to our children, Beth, Benjamin, and Sophia, for making us laugh, reminding us what's really important, and bringing joy and purpose to our life.

PAUL ARCHIBALD LARSEN

*University of Wisconsin–Madison*

*July 2007*



# Contents

<b>Abstract</b>	ii
<b>Acknowledgments</b>	v
<b>List of Tables</b>	xv
<b>List of Figures</b>	xvii
<b>Notation</b>	xxiii
<b>Chapter 1 Introduction</b>	1
1.1 Project motivation and research objectives . . . . .	1
1.2 Thesis overview . . . . .	4
<b>Chapter 2 Literature Review</b>	7
2.1 Crystallization overview and terminology . . . . .	7
2.2 Conventional practice in industry . . . . .	9
2.2.1 Process development . . . . .	9
2.2.2 Controlled, measured, and manipulated variables . . . . .	14
2.3 Recent advances in crystallization technology . . . . .	15
2.3.1 Spectroscopic and laser-based monitoring . . . . .	15
2.3.2 Imaging-based monitoring . . . . .	19
2.3.3 Manipulated variables for crystal shape and polymorphic form . . . . .	22

2.3.4	Modeling and prediction of crystal size, shape, and polymorphic form . . . . .	23
2.3.5	Control . . . . .	24
2.4	The future of crystallization . . . . .	25
<b>Chapter 3</b>	<b>Crystallization Model Formulation and Solution</b>	<b>27</b>
3.1	Model formulation . . . . .	27
3.1.1	Population balance . . . . .	27
3.1.2	Mass balance . . . . .	29
3.1.3	Energy balance . . . . .	30
3.2	Model solution . . . . .	30
3.2.1	Method of moments . . . . .	30
3.2.2	Orthogonal collocation . . . . .	31
<b>Chapter 4</b>	<b>Experimental and Simulated Image Acquisition</b>	<b>35</b>
4.1	Crystallizer and imaging apparatus . . . . .	35
4.1.1	Crystallizer . . . . .	35
4.1.2	Data acquisition . . . . .	36
4.1.3	Video image acquisition . . . . .	38
4.1.4	Operating procedure . . . . .	39
4.2	Chemical systems . . . . .	41
4.2.1	Industrial pharmaceutical . . . . .	41
4.2.2	Industrial photochemical . . . . .	41
4.2.3	Glycine . . . . .	42
4.3	Artificial image generation . . . . .	44
4.3.1	Stochastic process model . . . . .	44
4.3.2	Imaging model . . . . .	45
4.3.3	Justifications for two-dimensional system model . . . . .	47

<b>Chapter 5 Two-dimensional Object Recognition for High-Aspect-Ratio Particles</b>	<b>49</b>
5.1 Image analysis algorithm description . . . . .	50
5.1.1 Overview . . . . .	50
5.1.2 Linear feature detection . . . . .	52
5.1.3 Identification of collinear line pairs . . . . .	54
5.1.4 Identification of parallel line pairs . . . . .	56
5.1.5 Clustering . . . . .	58
5.2 Experimental results . . . . .	59
5.2.1 Algorithm accuracy . . . . .	61
5.2.2 Algorithm speed . . . . .	65
5.3 Conclusion . . . . .	70
<b>Chapter 6 Three-dimensional Object Recognition for Complex Crystal Shapes</b>	<b>71</b>
6.1 Model-based recognition algorithm . . . . .	72
6.1.1 Preliminaries . . . . .	72
6.1.2 Linear feature detection . . . . .	75
6.1.3 Perceptual grouping . . . . .	76
6.1.4 Model-fitting . . . . .	77
6.1.5 Summary and example . . . . .	81
6.2 Results . . . . .	83
6.2.1 Visual evaluation . . . . .	84
6.2.2 Comparison with human analysis . . . . .	84
6.2.3 Algorithm speed . . . . .	91
6.3 Conclusions . . . . .	92
<b>Chapter 7 Statistical Estimation of PSD from Imaging Data</b>	<b>95</b>
7.1 Previous work . . . . .	96
7.2 Theory . . . . .	98

7.2.1	PSD Definition . . . . .	98
7.2.2	Sampling model . . . . .	99
7.2.3	Maximum likelihood estimation of PSD . . . . .	100
7.2.4	Confidence Intervals . . . . .	103
7.3	Results . . . . .	104
7.3.1	Case study 1: mono-disperse particles . . . . .	106
7.3.2	Case study 2: uniform distribution . . . . .	108
7.3.3	Case study 3: normal distribution . . . . .	109
7.3.4	Case study 4: uniform distribution with particles larger than image . . . . .	111
7.4	Conclusion . . . . .	112
<b>Chapter 8 Assessing the Reliability of Imaging-based, Number Density Measurement</b>		<b>115</b>
8.1	Previous work . . . . .	116
8.2	Theory . . . . .	117
8.2.1	Particulate system definition . . . . .	117
8.2.2	Sampling and measurement definitions . . . . .	117
8.2.3	Descriptor for number density reliability . . . . .	118
8.2.4	Estimation of number density . . . . .	120
8.3	Image analysis methods summary . . . . .	122
8.4	Results . . . . .	123
8.4.1	Descriptor comparison: solids concentration versus overlap . . . . .	123
8.4.2	Estimation of number density . . . . .	126
8.5	Conclusion . . . . .	130
<b>Chapter 9 High-resolution PSD Measurement for Industrial Crystallization</b>		<b>133</b>
9.1	Crystallizer model and imaging summary . . . . .	134
9.2	Results . . . . .	136
9.2.1	Process and imaging simulations . . . . .	137

9.2.2	Absolute PSD measurement . . . . .	139
9.2.3	Measurements for product quality . . . . .	140
9.2.4	Discussion . . . . .	144
9.3	Conclusion . . . . .	147
<b>Chapter 10 Conclusion</b>		<b>149</b>
<b>Appendix A Derivations for Maximum Likelihood Estimation of PSD</b>		<b>153</b>
A.1	Maximum likelihood estimation of PSD . . . . .	153
A.2	Derivation of probability densities . . . . .	155
A.2.1	Non-border particles . . . . .	155
A.2.2	Border particles . . . . .	158
A.3	Validation of marginal densities . . . . .	163
<b>Bibliography</b>		<b>173</b>
<b>Vita</b>		<b>185</b>



# List of Tables

5.1	SHARC parameter values used to analyze images from pharmaceutical crystallization experiment. . . . .	60
5.2	Comparison of results obtained from nine different persons manually sizing the same ten images. . . . .	63
5.3	Comparison of mean sizes obtained from manual sizing of crystals by a human operator and from automatic sizing by SHARC. . . . .	64
5.4	Computational requirements for analyzing image sets from pharmaceutical crystallization experiment . . . . .	68
5.5	Computational requirements for SHARC to achieve convergence of particle size distribution mean and variance. . . . .	69
6.1	Summary of comparison between M-SHARC results and human operator results for in situ video images obtained at low, medium, and high solids concentrations . .	88
6.2	Average cputime required for M-SHARC to analyze single image for three different image sets of increasing solids concentration . . . . .	92
8.1	Parameters used to simulate imaging of particle population at a given solids concentration. . . . .	123
8.2	Parameter values used to analyze artificial images of overlapping particles. . . . .	124
9.1	Parameters used to simulate industrial batch crystallization process . . . . .	135

9.2 Parameters used to simulate imaging of particle population using industrial video imaging probe. . . . .	136
--	-----

# List of Figures

1.1	Images of crystal populations. . . . .	2
2.1	Depiction of solute concentration and temperature trajectories for a generic cooling crystallization process . . . . .	8
2.2	Photograph of production-scale, continuous crystallizer for ammonium sulfate . . .	10
2.3	Photograph of production-scale, continuous crystallizer for sodium chlorate . . . .	11
2.4	Photograph of batch crystallizer used for high potency drug manufacturing . . . .	12
2.5	Photographs of internals and exterior of multi-purpose batch crystallizer used for pharmaceutical and specialty chemical manufacturing . . . . .	13
2.6	Depiction of effect of disturbances on supersaturation trajectory for a batch cooling crystallization . . . . .	13
2.7	Comparison of particle size measurements obtained using laser backscattering versus those obtained using imaging . . . . .	18
4.1	Experimental setup for obtaining in situ crystallization images. . . . .	36
4.2	Imaging system wiring. . . . .	39
4.3	Chemical structure of glycine . . . . .	42
4.4	Images illustrating morphology of $\alpha$ -glycine crystallized in water at room temperature. . . . .	43
4.5	Images illustrating morphology of $\alpha$ -glycine crystallized in water at 55 °C. . . . .	43

4.6	Images illustrating morphology of $\gamma$ -glycine crystallized in water at room temperature. . . . .	43
4.7	Depiction of the perspective projection of a cylindrical particle onto the image plane . . . . .	46
4.8	Depiction of CCD image. . . . .	46
5.1	Step-by-step example of SHARC algorithm applied to an <i>in situ</i> image of suspended pharmaceutical crystals . . . . .	51
5.2	Depiction of shifted gradient direction quantizations used to label pixels . . . . .	53
5.3	Step-by-step example of finding linear features using Burns line finder and blob analysis . . . . .	54
5.4	Depiction of variables used for line pair classification scheme. . . . .	55
5.5	Depiction of valid and invalid parallel line pairs . . . . .	57
5.6	Step-by-step example of clustering procedure for valid parallel pairs . . . . .	59
5.7	Temperature trajectory and image acquisition times for pharmaceutical crystallization experiment . . . . .	60
5.8	Algorithm performance on example image from video image set 3 . . . . .	61
5.9	Algorithm performance on example image from video image set 4 . . . . .	61
5.10	Algorithm performance on example image from video image set 5 . . . . .	62
5.11	Algorithm performance on example image from video image set 6 . . . . .	62
5.12	Comparison of cumulative number fractions obtained from manual and automatic sizing of crystals for video image sets 3–6 . . . . .	65
5.13	Comparison of crystals sized manually and using SHARC . . . . .	66
5.14	Zoomed-in view of crystals that SHARC failed to identify correctly . . . . .	67
6.1	Parameterized, wire-frame model for glycine crystals . . . . .	73
6.2	Depiction of the perspective projection of the glycine model onto the image plane . . . . .	74
6.3	Depiction of different viewpoint-invariant line groups (VIGs) used by M-SHARC . . . . .	76
6.4	Depiction of correspondence hypotheses . . . . .	78

6.5	Depiction of variables used in mismatch calculation for a single line correspondence.	80
6.6	Step-by-step example of M-SHARC algorithm applied to image of $\alpha$ -glycine crystal.	83
6.7	M-SHARC segmentation results for selected images acquired at low solids concentration (13 min. after appearance of crystals).	85
6.8	M-SHARC segmentation results for selected images acquired at medium solids concentration (24 min. after appearance of crystals).	85
6.9	M-SHARC segmentation results for selected images acquired at high solids concentration (43 min. after appearance of crystals).	86
6.10	Illustration of comparison between human operator results and M-SHARC results	87
6.11	Comparison of Human and M-SHARC cumulative distribution functions	89
6.12	Results of linear feature detection for selected crystals missed by M-SHARC	90
7.1	Depiction of methodology for calculating Miles-Lantuéjoul correction factors for particles of different lengths observed in an image of dimension $b \times a$	98
7.2	Example images for simulations of various particle populations	105
7.3	Comparison of sampling distributions for different PSD estimators: mono-disperse population	106
7.4	Fraction of confidence intervals containing the true parameter value as a function of confidence level: mono-disperse population	107
7.5	Relative efficiencies of Miles-Lantuéjoul and maximum likelihood estimators as a function of particle size and sample size: uniformly-distributed population	108
7.6	Fraction of confidence intervals containing the true parameter value as a function of confidence level: uniformly-distributed population and large sample size	109
7.7	Fraction of confidence intervals containing the true parameter value as a function of confidence level: uniformly-distributed population and small sample size	110
7.8	Analytical sampling distributions for the various size classes of a discrete normal distribution	111

7.9	Relative efficiencies of Miles-Lantuéjoul and maximum likelihood estimators as a function of particle size and sample size: normally-distributed population . . . . .	112
7.10	Comparison of sampling distributions for different PSD estimators: particles larger than image . . . . .	113
8.1	Geometric representation of admissible area, or region in which a particle is overlapped by another particle . . . . .	119
8.2	Likelihood of observing $n$ non-overlapping particles with example images of different number densities that give the same number of non-overlapping particles . .	121
8.3	Comparison of images generated for two different mono-disperse particle populations at the same solids concentration and at the same level of overlap . . . . .	124
8.4	Comparison of average number of overlaps per crystal for images simulated at constant overlap and at constants solids concentration . . . . .	125
8.5	Comparison of percentage of particles missed by automated image analysis for images simulated at constant overlap and at constants solids concentration . . . . .	125
8.6	Examples of synthetic images generated at various particle sizes and degrees of overlap . . . . .	127
8.7	Results of number density estimation using Miles-Lantuéjoul method for various particle sizes and various levels of particle overlap . . . . .	128
8.8	Data and model prediction for number of particles with length $\leq 0.1a$ identified per image by automated image analysis . . . . .	129
8.9	Data and model prediction for number of particles with lengths $\leq 0.3a$ or $\leq 0.5a$ identified per image by automated image analysis . . . . .	130
8.10	Ratio of estimated number density and true number density versus image difficulty using SHARC data and empirical correction factors calculated for each different particle size . . . . .	131
9.1	Comparison of simulation results for optimal and linear temperature trajectories . .	137

9.2 Examples of images generated at various times during optimal cooling simulation . . . . .	138
9.3 Examples of images generated at various times during linear cooling simulation . . . . .	139
9.4 Evolution of measured and estimated number-based PSD for optimal cooling and perfect image analysis. . . . .	140
9.5 Evolution of measured and estimated weight PSD for optimal cooling and perfect image analysis . . . . .	141
9.6 Evolution of measured and estimated weight PSD for optimal cooling and image analysis using SHARC . . . . .	142
9.7 Estimated ratios of nuclei mass to seed crystal mass for optimal and linear cooling . .	143
9.8 Estimated mean crystal sizes for optimal and linear cooling . . . . .	143
9.9 Estimated coefficients of variation for optimal and linear cooling . . . . .	144
A.1 Depiction of hypothetical system of vertically-oriented particles randomly and uniformly distributed in space. . . . .	156
A.2 Depiction of geometrical properties used to derive the non-border area function $A_{nb}(l, \theta)$ . . . . .	157
A.3 Depiction of hypothetical system of vertically-oriented particles randomly and uniformly distributed in space. . . . .	160
A.4 Depiction of non-border area for arbitrary length and orientation. . . . .	161
A.5 Comparison of theoretical and simulated marginal densities for randomly-oriented, monodisperse particles of length 0.5 and measured by partitioning [0.1 0.9] into ten bins. Results are for non-border particles. . . . .	164
A.6 Comparison of theoretical and simulated marginal densities for randomly-oriented, monodisperse particles of length 0.5 and measured by partitioning [0.1 0.9] into ten bins (results are shown only for bins 1–4 because the probability of observing a border length in size class 5 or above is zero). Results are for border particles. . . . .	165

A.7 Comparison of theoretical and simulated marginal densities for randomly-oriented particles distributed uniformly on [0.1 0.9] and measured by partitioning [0.1 0.9] into ten bins. Results are for non-border particles. . . . .	166
A.8 Comparison of theoretical and simulated marginal densities for randomly-oriented particles distributed uniformly on [0.1 0.9] and measured by partitioning [0.1 0.9] into ten bins. Results are for border particles. . . . .	167
A.9 Comparison of theoretical and simulated marginal densities for randomly-oriented particles distributed normally and measured by partitioning [0.1 0.9] into 10 bins. Results are for non-border particles. . . . .	168
A.10 Comparison of theoretical and simulated marginal densities for randomly-oriented particles distributed normally and measured by partitioning [0.1 0.9] into 10 bins. Results are for border particles. . . . .	169
A.11 Comparison of theoretical and simulated marginal densities for randomly-oriented particles distributed uniformly on [0.4 2.0] and measured by partitioning [0.4 1.0] into 9 bins with a 10th bin spanning $[1.0 \sqrt{2}]$ . Results are for non-border particles. .	170
A.12 Comparison of theoretical and simulated marginal densities for randomly-oriented particles distributed uniformly on [0.4 2.0] and measured by partitioning [0.4 1.0] into 9 bins with a 10th bin spanning $[1.0 \sqrt{2}]$ . Results are for border particles. . . .	171

# Notation

## Upper Case Letters

$A$	surface area of slurry exposed to crystallizer jacket
$A\{J\}$	area of domain $J$
$A_{ij}$	$i, j$ th element of collocation first derivative weight matrix
$A_{FP}$	projected area of IA false positives
$A_H$	projected area of IA hits
$A_M$	projected area of IA misses
$A_{NB}$	area of region inside which a particle does not touch the image border
$A_{ovp}$	area of region in which two particles of specific shape and orientation overlap
$B^\circ$	crystal nucleation rate density at size $L_0$
$\hat{C}$	solution phase concentration (mass solute/mass solvent)
$\hat{C}_{\text{sat}}$	saturation concentration (mass solute/mass solvent)
$D$	dimensionless parameter giving average number of particle overlaps
$D_c$	set of data lines for which correspondence has been identified with wire-frame model lines
$E$	set of edges defining wire-frame model
$E^P$	set of projected wire-frame model lines
$E_g$	crystal growth activation energy
$E_J$	edge $J$ in wire-frame model
$E^2$	Euclidean plane
$F$	cumulative distribution function for particle orientation
$G$	crystal growth rate

$H$	cumulative distribution function for particle length
$\Delta H_c$	heat of crystallization
$I$	two-dimensional image
$J$	two-dimensional domain parameterized by $(z, n, \theta_n)$
$K$	random variable giving the number of times a particle is overlapped
$L$	characteristic crystal length
$\bar{L}$	mean particle length
$L_0$	initial size of nucleated crystals
$L_i$	random variable giving length of particle $i$
$\tilde{L}_i$	length of line $i$
$L_j$	$j$ th collocation location on length domain
$L_J$	length of $J$ th wire-frame model line
$L_{\max}$	size of the largest particle in the population
$\tilde{L}_{\max}$	length of longest line in parallel line pair
$\tilde{L}_{\min}$	length of shortest line in parallel line pair
$L_N(t)$	size of largest nucleated crystal
$L_{S_l}(t)$	size of smallest seed crystal
$L_{S_u}(t)$	size of largest seed crystal
$L_V$	length of virtual line
$L_i^P$	length of projection of line $i$ onto virtual line
$L(\rho)$	likelihood function
$M$	Miles-Lantuéjoul weighting factor
$M_J$	vector pointing from origin of image coordinate system to midpoint of $J$ th wire-frame model line
$M_T$	solids concentration
$N$	number of images
$\tilde{N}_c$	random variable giving number of crystals in vicinity of imaging volume
$N_H$	number of hits by IA

$N_M$	number of misses by IA
$N_{FP}$	number of false positives by IA
$P_i$	set of points comprising particle $i$
$Q(x)$	sampling region centered at point $x$
$Q_k$	volumetric flow rate of $k$ th stream
$Q_P$	parallel line pair quality
$R_g$	universal gas constant
$R$	lower limit on resolution for camera lens
$\mathbf{R}_x$	rigid-body rotation matrix for transformation from world coordinate frame to camera coordinate frame
$S$	slurry system
$S$	set of data lines
$\hat{S}$	relative supersaturation
$\tilde{S}$	parallel line pair significance
$\mathbf{S}$	vector of breaks between discrete particle size classes
$T$	temperature
$T$	number of discrete particle size classes
$\mathbf{T}$	translation vector with elements $(t_x, t_y, t_z)$ for transformation from world coordinate frame to camera coordinate frame
$T_0$	initial temperature
$T_j$	jacket temperature
$\hat{\mathbf{T}}_J$	unit tangent vector to $J$ th wire-frame model line
$U$	overall heat transfer coefficient
$V$	slurry volume
$V$	set of vertices defining wire-frame model
$V_I$	imaging volume
$X$	random variable giving number of observations of completely isolated, non-border particles
$\tilde{X}$	random variable giving number of particles identified by image analysis algorithm

$\mathbf{X}$	three-dimensional point in model coordinate frame
$\mathbf{X}_c$	three-dimensional point in camera coordinate frame with elements $(X_c, Y_c, Z_c)$
$\mathbf{X}_k$	random vector giving numbers of non-border particles of various size classes observed in image $k$
$X_{ik}$	random variable giving number of non-border particles in size class $i$ observed in image $k$
$\mathbf{X}_\Sigma$	random vector giving total number of non-border particles of various size classes observed in $N$ images
$\mathbf{X}_w$	three-dimensional point in world coordinate frame with elements $(X_w, Y_w, Z_w)$
$\mathbf{X}_{wi}$	three-dimensional random vector with elements $(X_{wi}, Y_{wi}, Z_{wi})$ giving centroid location for particle $i$ in the world coordinate frame
$\mathbf{X}_K[\mathbf{p}_m]$	vertex of wire-frame model in model coordinate frame
$\mathbf{Y}_k$	random vector giving numbers of border particles of various observed lengths observed in image $k$
$\mathbf{Y}_\Sigma$	random vector giving total number of border particles of various observed lengths observed in $N$ images
$Y_{ik}$	random variable giving number of border particles with observed length in size class $i$ observed in image $k$

## Lower Case Letters

$a$	horizontal image dimension
$a_p$	area of the two-dimensional projection of a particle
$a_S$	coefficient for quadratic function representing seed subpopulation
$b$	supersaturation order of nucleation
$b$	vertical image dimension
$c_i$	solute concentration of species $i$
$\tilde{c}_i$	chord length of crystal $i$
$c_i^*$	saturation concentration of species $i$
$c_p$	slurry heat capacity

$c_v$	number-based coefficient of variation
$c_{v_w}$	weight-based coefficient of variation
$\Delta c$	saturation concentration ( $c_i - c_i^*$ )
$d_f$	depth of field
$d_{F,h}$	horizontal Feret diameter
$d_{F,v}$	vertical Feret diameter
$d_{\text{PD}}$	perpendicular distance between two lines
$d_{\text{EP}}$	distance between endpoints of two lines
$\text{diff}_i$	difference between mean particle size values calculated by different operators
$e_{1i}$	first endpoint of $i$ th data line
$e_{2i}$	second endpoint of $i$ th data line
$f(L, t)$	particle size distribution (PSD)
$f_c$	camera focal length
$f_k$	PSD of $k$ th flow stream
$f_N(L, t)$	PSD of subpopulation of nucleated crystals
$\bar{f}_N(\zeta, t)$	PSD of subpopulation of nucleated crystals on scaled length domain
$f_S(L, t)$	PSD of subpopulation of seed crystals
$f_{S_0}$	initial PSD of seed subpopulation
$g$	supersaturation order of growth
$g(D)$	empirical function
$h$	parametric particle length density function
$\hat{h}$	conversion from solvent mass to slurry volume
$h(\rho, \theta)$	model prediction of number of particles identified by IA
$h_m$	crystal body height for wire-frame crystal model
$j$	third moment order of nucleation
$k_a$	area shape factor
$k_g$	growth rate constant
$k_b$	nucleation rate constant

$k_u$	pixel horizontal scaling factor
$k_v$	volume shape factor
$k_v$	pixel vertical scaling factor
$l$	length of particles in mono-disperse population
$l(p)$	log likelihood function
$l_i$	projected length of crystal $i$
$l_j(L)$	Lagrange interpolation polynomial of degree $j$
$m$	length-to-pixel ratio
$m_N$	nucleus-grown crystal mass
$m_S$	seed-grown crystal mass
$m_X$	Poisson distribution parameter for $X$
$m_{\tilde{X}}$	Poisson distribution parameter for $\tilde{X}$
$m_{X_i}$	Poisson parameter for distribution of $X_i$
$m_{Y_j}$	Poisson parameter for distribution of $Y_j$
$l_j$	length of $j$ th data line
$\mathbf{m}_j$	vector pointing from origin of image coordinate system to midpoint of $j$ th data line
$m_S(t)$	mass of seed-grown subpopulation
$n_c$	number of collocation points
$n$	number of observations (particles)
$n$	total number of particles in system
$n_b$	number of buckets for Burns line finder
$n_c$	number of collocation points
$n_d$	number of data points
$\hat{n}_l$	estimate of number of lines identified in image
$n_\nabla$	size of Sobel gradient operator
$\mathbf{p}$	vector of internal wire-frame model parameters and viewpoint parameters
$\mathbf{p}_m$	internal parameters for wire-frame model
$p_{\text{iso}}$	probability that a given particle is completely isolated

$p_{\text{ovp}}$	probability that a given particle is overlapped by a second, given particle
$p_{XY}$	joint probability density for $X$ and $Y$
$p_{X_i}$	probability density for $X_i$
$p_{Y_i}$	probability density for $Y_i$
$q$	discrete, relative PSD
$r$	radius of cylindrical particles
$s_p$	perimeter of the two-dimensional projection of a particle
$t$	time
$t_m$	crystal pyramid height for wire-frame crystal model
$t(\alpha, N - 1)$	Student's t-distribution for confidence level $\alpha$ and number of samples $N$
$t_x$	translation in x-direction for transformation from world coordinate frame to camera coordinate frame
$\hat{\mathbf{t}}_j$	unit tangent vector to $j$ th data line
$\hat{\mathbf{t}}_j^\perp$	unit vector perpendicular to $j$ th data line
$s_N$	sample standard deviation
$u$	horizontal pixel coordinate
$u_0$	value of horizontal pixel coordinate corresponding to $\mathbf{x}_c = (0, 0)$
$v$	vertical pixel coordinate
$v_0$	value of vertical pixel coordinate corresponding to $\mathbf{x}_c = (0, 0)$
$v_{\max}$	number of vertical CCD pixels
$w$	width of particles in mono-disperse population
$w_m$	crystal body width for wire-frame crystal model
$\mathbf{w}$	two-dimensional pixel coordinate vector with elements $(u, v)$
$w_i$	projected width of crystal $i$
$x$	measured number of isolated, non-border particles, or the realization of $X$
$\tilde{x}$	number of particles identified by IA, or the realization of $\tilde{X}$
$x_i$	horizontal centroid of line $i$
$x_{\min}$	lower bound on centroid location of particles in the x-dimension

$x_{\max}$	upper bound on centroid location of particles in the x-dimension
$\mathbf{x}_c$	two-dimensional point in image coordinates with elements $(x_c, y_c)$
$\mathbf{x}$	two-dimensional point in image coordinates with elements $(x, y)$
$\mathbf{x}_k$	realization of random variable $\mathbf{X}_k$
$x_V$	horizontal centroid of virtual line
$y_i$	vertical centroid of line $i$
$\mathbf{y}_k$	realization of random variable $\mathbf{Y}_k$
$y_V$	vertical centroid of virtual line
$z$	center point of two-dimensional domain
$z_0$	distance from camera lens to imaging volume

## Greek Letters

$\alpha$	confidence level
$\alpha_m$	assumed orientation in depth for wire-frame model projection
$\alpha_i$	parameter for distribution function of $X_i$
$\beta_{ij}$	parameter for distribution function of $Y_j$
$\delta$	Dirac delta function
$\epsilon_{ \nabla }$	gradient magnitude threshold
$\epsilon_A$	pixel area threshold
$\epsilon_{AR}$	aspect ratio threshold for parallel line grouping
$\epsilon_{\theta_C}$	orientation difference threshold for collinear line grouping
$\epsilon_{\theta_P}$	orientation difference threshold for parallel line grouping
$\epsilon_{PD}$	threshold on perpendicular distance between two lines
$\epsilon_{EP}$	threshold on end-point distance between two lines
$\epsilon_Q$	parallel line pair quality threshold
$\zeta$	scaled particle size on [0,1] domain
$\theta_i$	orientation of line $i$
$\theta_n$	parameters necessary to specify two-dimensional domain of class $n$

$\theta_V$	orientation of virtual line
$\theta_x$	particle orientation around x-axis of world coordinate frame
$\theta_z$	particle orientation around camera's optical axis
$\Theta_{zi}$	random variable giving orientation of particle $i$ around z-axis of world coordinate frame
$\lambda$	Expected number of particle per image
$\tilde{\lambda}$	Poisson distribution parameter for $\tilde{N}_c$
$\mu_i$	$i$ th moment of the PSD
$\mu_{N_i}$	$i$ th moment of the PSD for nucleus-grown crystals only
$\mu_{S_i}$	$i$ th moment of the PSD for seed-grown crystals only
$\rho$	number density of particles in mono-disperse population
$\rho$	discrete absolute PSD
$\hat{\rho}$	estimate of number density
$\hat{\rho}$	maximum likelihood estimate of $\rho$ using only non-border particle measurements
$\rho_A$	area number density of particles in mono-disperse population
$\hat{\rho}_b$	maximum likelihood estimate of $\rho$ using border and non-border particle measurements
$\rho_c$	crystal density
$\rho_i$	number density of particles in size class $i$
$\hat{\rho}_{ML}$	Miles-Lantuéjoul estimate of $\rho$
$\sigma$	saturation concentration $((c_i - c_i^*)/c_i^*)$
$\Phi$	objective function for optimization
$\Phi_p$	stochastic spatial process describing particle population in vicinity of imaging volume
$\chi^2(\alpha, n - 1)$	chi-squared distribution for confidence level $\alpha$ and sample size $n$
$\Omega_L$	particle length domain
$\Omega$	admissible area, or the expectation of $A_{\text{ovp}}$



# Chapter 1

## Introduction <sup>1</sup>

### 1.1 Project motivation and research objectives

Crystallization plays a critical role in numerous industries for a variety of reasons. In the semiconductor industry, for example, crystallization is used to grow long, cylindrical, single crystals of silicon with a mass of several hundred kilograms. These gigantic crystals, called boules, are sliced into thin wafers upon which integrated circuits are etched. Prior to etching, crystallization is used to grow thin layers of crystalline, semiconductor material onto the silicon wafer using a process called chemical vapor deposition. In the food industry, crystallization is often used to give products the right texture, flavor, and shelf life. Crystallization is used to produce ice cream, frozen dried foods, chewing gum, butter, chocolate, salt, cheese, coffee, and bread [47]. These examples highlight the utility of crystallization in creating solids with desirable and consistent properties.

Crystallization is also widely used to separate and purify chemical species in the commodity, petrochemical, specialty, and fine-chemical industries. In fact, DuPont, one of the world's largest chemical manufacturers, estimated in 1988 [43] that approximately 70% of its products pass through a crystallization or precipitation stage. Crystallization is used in the pharmaceutical industry to identify structure for use in drug design, to isolate chemical species from mixtures of reaction products, and to achieve consistent and controlled drug delivery. The vast majority of pharmaceuticals are manufactured in solid, generally crystalline, form.

Despite crystallization's long history and widespread use, this process remains difficult to

---

<sup>1</sup>Portions of this chapter appear in Larsen, Patience, and Rawlings [65]

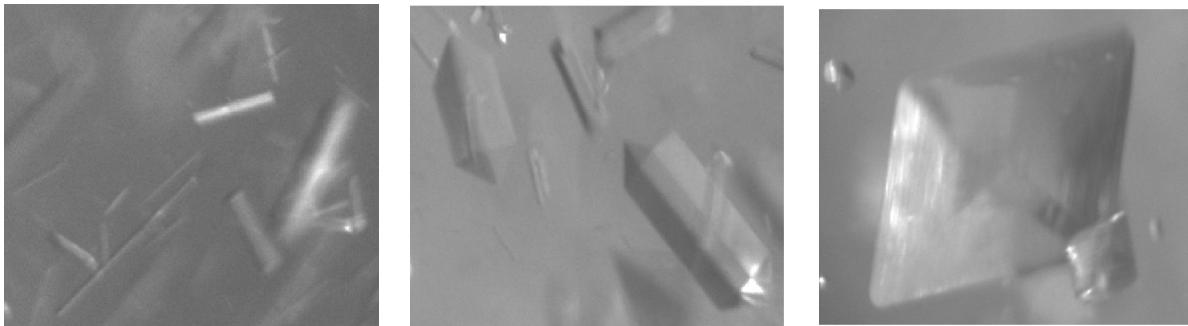


Figure 1.1: Images of crystal populations.

understand and control. To appreciate the challenges associated with this process, consider the images of different crystal populations shown in Figure 1.1. The needle-like crystals on the left are the active ingredient for a Parkinson's disease treatment. The crystals in the image exhibit a wide range of sizes and aspect ratios, indicating the distributed nature of crystallization processes. This feature is one of the basic challenges associated with any dispersed-phase process. Many of the key crystallizer states, including crystal size, shape, and purity, are distributed or vary over the crystal population. The evolution of these states is affected by a variety of complex phenomena, including nucleation, growth, agglomeration, and breakage. The sizes and shapes of the crystals affect the efficiency of downstream processes such as solid-liquid separation, drying, mixing, milling, granulation, and compaction. In some cases, particularly for chemicals having low solubility or low permeability, the crystal size and shape affect product properties such as bioavailability and tablet stability. Control of chemical purity is important for food and pharmaceutical products intended for consumption and for semiconductor devices requiring highly consistent properties.

The remaining two images in Figure 1.1 show crystals of glycine, an amino acid of interest to the pharmaceutical community both as an excipient in pharmaceutical formulations and as an active ingredient. The prismatic crystal shape in the center image corresponds to the  $\alpha$  polymorphic form of glycine, while the bipyramidal shape in the other image corresponds to the  $\gamma$  form. These images demonstrate that even molecules as simple as glycine exhibit polymorphism,

or the ability to crystallize into different crystal structures. Polymorphism must be controlled because the polymorphic form affects product stability, hygroscopicity, saturation concentration, dissolution rate, and bioavailability. The development of increasingly complex compounds in the pharmaceutical and specialty chemical industries makes polymorphism a commonly observed phenomenon for which control is essential. The recent disaster at Abbott Labs [10], in which the appearance of an unknown polymorphic form of ritonavir in drug formulations threatened the supply of the life-saving AIDS treatment Norvir, illustrates both the importance and difficulty of controlling polymorphism.

Robust control of the solid-phase properties requires that they be measured. However, conventional particle size distribution (PSD) monitoring technologies, such as laser diffraction and laser backscattering, are based on assumptions of particle sphericity and thus do not provide the monitoring capability necessary to achieve on-line PSD control for systems in which the particles are highly non-spherical [136, 16]. Additionally, laser backscattering cannot measure the shape of individual particles and therefore cannot measure the distribution of particles between different shape classes (e.g. number of needles relative to number of spheres) nor shape factor distributions (e.g. distribution of aspect ratios).

The limitations inherent in laser-scattering-based monitoring technologies motivate the use of imaging-based methods, which allow direct visualization of particle size and shape. Obtaining quantitative information from imaging-based methods, however, requires image segmentation. Image segmentation means separating the objects of interest (e.g. the particles) from the background. Most commercial, imaging-based, on-line particle size and shape analyzers solve the segmentation problem by imaging the particulate slurry as it passes through a specially-designed flow cell under controlled lighting conditions [3, p. 167]. The images acquired in this way can be segmented using simple thresholding methods. The drawback is that this approach requires sampling, which is inconvenient, possibly hazardous, and raises concerns about whether the sample is representative of the bulk slurry.

This thesis is focused on developing image segmentation algorithms that enable robust

segmentation of noisy, *in situ* images and statistical estimation methods that overcome the biases inherent in imaging-based measurement. These tools are expected to aid practitioners in developing effective, imaging-based monitoring technology, resulting in improved understanding and control of industrial crystallization processes.

## 1.2 Thesis overview

The thesis is organized as follows. Preliminary material is given in Chapters 2–4. Chapter 2 describes conventional industrial practices for designing and controlling crystallization processes. Chapter 2 also reviews the state-of-the-art in crystallization design and control, with particular emphasis given to sensor technology. Chapter 3 presents the batch crystallization model used in this study and describes the methods used to solve the model. Chapter 4 describes the experimental apparatus used to conduct crystallization experiments and obtain *in situ* images. Chapter 4 also discusses the simulation methods used to generate artificial *in situ* images.

Chapters 5 and 6 describe two novel algorithms that enable robust image analysis for noisy, *in situ* images. The first algorithm, called SHARC (Segmentation for High Aspect Ratio Crystals), can be used to find high-aspect-ratio crystals, a specific shape class that arises frequently in pharmaceutical applications. This shape class is particularly problematic for standard image analysis routines because it results in a high degree of particle overlap. The second algorithm, called M-SHARC (Model-based SHApE Recognition for Crystals), is designed to identify and distinguish between multiple shape classes, thereby enabling the measurement of polymorphic fraction for systems in which the polymorphs exhibit different shapes. Chapters 5 and 6 also provide an evaluation of the algorithms in terms of their computational requirements and their accuracy relative to measurements obtained by human operators.

Chapter 7 develops a maximum likelihood estimator for estimating the PSD from imaging data and demonstrates how to obtain confidence intervals for the measured PSD using bootstrapping. We benchmark the estimator against the conventional Miles-Lantuéjoul approach. For

needle-like particles, our estimator provides better estimates than the Miles-Lantuéjoul approach, but the Miles-Lantuéjoul approach can be applied to a wider class of shapes. Both methods assume perfect image segmentation, or that every particle appearing in the image is identified correctly.

Chapter 8 develops a descriptor that correlates with the reliability of the imaging-based measurement (i.e. the quality of the image segmentation) based on the amount of particle overlap. Chapter 8 demonstrates that both the Miles-Lantuéjoul and maximum likelihood approaches discussed above underestimate the number density of particles and develops a practical approach for estimating the number density of particles for significant particle overlap and imperfect image analysis. The approach is developed for mono-disperse particle systems.

Chapter 9 applies the tools developed in previous chapters to a well-studied batch crystallization process of an industrial photochemical, demonstrating the feasibility of imaging-based PSD measurement for industrial crystallization processes. Chapter 9 also demonstrates the ability to monitor important product quality parameters, such as the ratio of nuclei mass to seed mass, that cannot be monitored by conventional technologies.

Finally, Chapter 10 summarizes the contributions of this dissertation, presents conclusions, and provides suggestions for future work.



## Chapter 2

# Literature Review<sup>1</sup>

## 2.1 Crystallization overview and terminology

Crystallization is the formation of a solid state of matter in which the molecules are arranged in a regular pattern. Crystallization can be carried out by a variety of methods, but the concepts and terminology relevant to most crystallization processes can be understood by examining the method of solution crystallization. In solution crystallization, the physical system consists of one or more solutes dissolved in a solvent. The system can be *undersaturated*, *saturated*, or *supersaturated* with respect to species  $i$ , depending on whether the solute concentration  $c_i$  is less than, equal to, or greater than the *saturation concentration*  $c_i^*$ . Crystallization occurs only if the system is supersaturated. The *supersaturation level* is the amount by which the solute concentration exceeds the saturation concentration, and is commonly expressed as  $\sigma = \frac{c_i - c_i^*}{c_i^*}$ ,  $S = \frac{c_i}{c_i^*}$ , or  $\Delta c = c_i - c_i^*$ . The supersaturation level can be increased either by lowering the saturation concentration (for example, by cooling as depicted in Figure 2.1) or by increasing the solute concentration (by evaporating the solvent, for example).

Crystallization moves a supersaturated solution toward equilibrium by transferring solute molecules from the liquid phase to the solid, crystalline phase. This process is initiated by *nucleation*, which is the birth or initial formation of a crystal. Nucleation occurs, however, only if the necessary activation energy is supplied. A supersaturated solution in which the activation energy is too high for nucleation to occur is called *metastable*. As the supersaturation level increases, the

---

<sup>1</sup>Portions of this chapter appear in Larsen, Patience, and Rawlings [65]

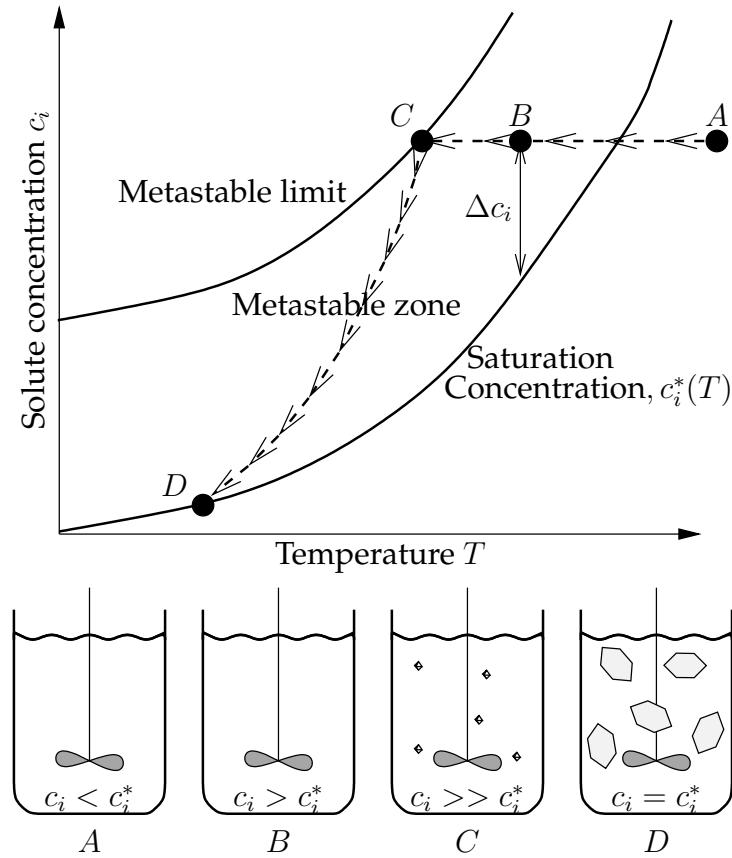


Figure 2.1: Depiction of a cooling, solution crystallization process. The process begins at point  $A$ , at which the solution is undersaturated with respect to species  $i$  ( $c_i < c_i^*$ ). The process is cooled to point  $B$ , at which the solution is supersaturated ( $c_i > c_i^*$ ). No crystals form at point  $B$ , however, because the activation energy for nucleation is too high. As the process cools further, the supersaturation level increases and the activation energy for nucleation decreases. At the metastable limit (point  $C$ ), spontaneous nucleation occurs, followed by crystal growth. The solute concentration decreases as solute molecules are transferred from the liquid phase to the growing crystals until equilibrium is reached at point  $D$ , at which  $c_i = c_i^*$ .

activation energy decreases. Thus spontaneous nucleation, also called *primary nucleation*, occurs only at sufficiently high levels of supersaturation, and the solute concentration at which this nucleation occurs is called the *metastable limit*. Since primary nucleation is difficult to control reliably,

primary nucleation is often avoided by injecting crystal *seeds* into the supersaturated solution.

Crystal nuclei and seeds provide a surface for crystal growth to occur. Crystal growth involves solute molecules attaching themselves to the surfaces of the crystal according to the crystalline structure. Crystals suspended in a well-mixed solution can collide with each other or with the crystallizer internals, causing crystal attrition and breakage that results in additional nuclei. Nucleation of this type is called *secondary nucleation*.

The rates at which crystal nucleation and growth occur are functions of the supersaturation level. The goal of crystallizer control is to balance the nucleation and growth rates to achieve the desired crystal size objective. Often, the size objective is to create large, uniformly sized crystals. Well-controlled crystallization processes operate in the *metastable zone*, between the saturation concentration and the metastable limit, to promote crystal growth while minimizing undesirable nucleation.

## 2.2 Conventional practice in industry

The objective of every industrial crystallization process is to create crystals that meet specifications on size, shape, composition, and internal structure. This objective is achieved using a variety of methods and equipment configurations depending on the properties of the chemical system, the end-product specifications, and the production scale. Continuous crystallizers, such as those shown in Figures 2.2 and 2.3, are typically used for large-scale production, producing hundreds of tons per day. In the specialty chemical, fine chemical, and pharmaceutical industries, batch crystallizers (see Figures 2.4 and 2.5) are often used to produce low-volume, high-value-added chemicals.

### 2.2.1 Process development

The first step in developing a control system for solution crystallization is to determine the saturation concentration and metastable limit of the target species over a range of temperatures,

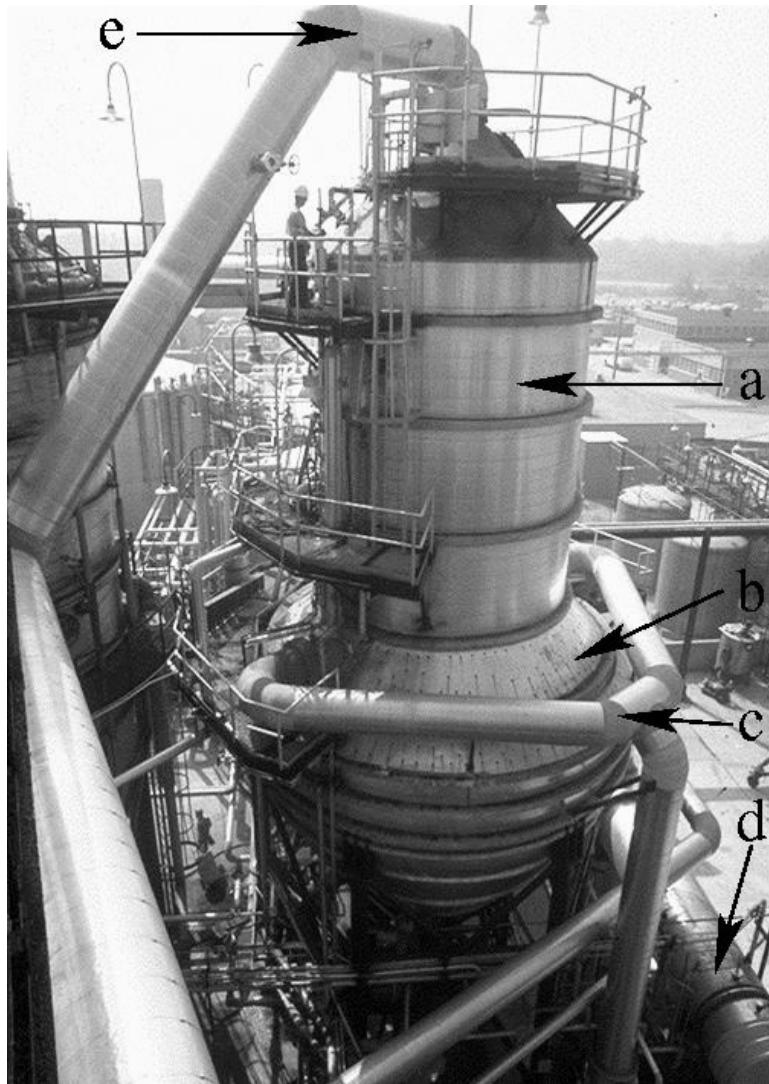


Figure 2.2: Production-scale draft tube crystallizer. This crystallizer is used to produce hundreds of tons per day of ammonium sulfate, commonly used as fertilizer or as a precursor to other ammonium compounds. The crystallizer body (a) widens at the lower section (b) to accommodate the settling region, in which small crystals called fines are separated from the larger crystals by gravitational settling. The slurry of saturated liquid and fines in the settling region is continuously withdrawn (c), combined with product feed, and passed through a heater (d) that dissolves the fines and heats the resulting solution prior to returning the solution to the crystallizer. The heat generated by crystallization is removed as the solvent evaporates and exits through the top of the crystallizer (e), to be condensed and returned to the process. Larger crystals are removed continuously from the bottom of the crystallizer. Image courtesy of Swenson Technology, Inc.



Figure 2.3: Production-scale draft tube baffle crystallizer. This crystallizer is used to produce hundreds of tons per day of sodium chlorate, which is commonly used in herbicides. Image courtesy of Swenson Technology, Inc.

solvent compositions, and pH's. The saturation concentration, also called solubility, represents the minimum solute concentration for which crystal growth can occur. The metastable limit, on the other hand, indicates the concentration above which undesirable spontaneous nucleation occurs (see the "Crystallization tutorial" sidebar). Spontaneous nucleation, which yields smaller, non-uniform crystals, can be avoided by injecting crystal "seeds" into the crystallizer to initialize crystal growth. The saturation concentration and metastable limit provide constraints on the operating conditions of the process and determine the appropriate crystallization method. For example, chemical systems in which the solubility is highly sensitive to temperature are crystallized

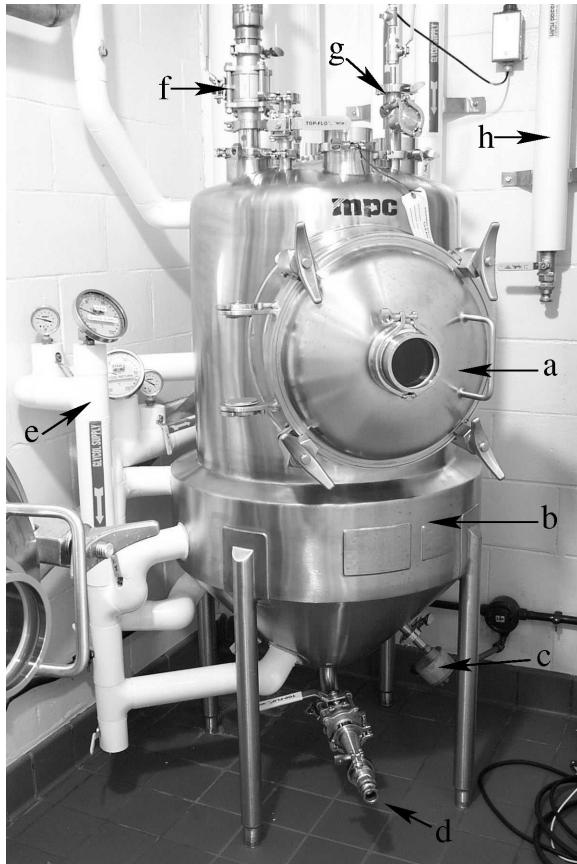


Figure 2.4: Small crystallizer used for high potency drug manufacturing. The portal (a) provides access to the crystallizer internals. The crystallizer widens at the lower section (b) to accommodate the crystallizer jacket, to which coolant (e) and heating fluid (h) lines are connected. Mixing is achieved using an impeller driven from below (c). The process feed enters from above (f) and exits below (d). The temperature sensor is inserted from above (g). Image courtesy of Ferro Pfanziehl Laboratories, Inc.

using cooling, while systems with low solubility temperature dependence employ anti-solvent or evaporation crystallization. Automation tools greatly reduce the amount of time, labor, and material previously required to characterize the solubility and metastable limit, enabling a wide range of conditions to be tested in a parallel fashion [11].

Once a crystallization method and solvents are chosen, kinetic studies are carried out on a

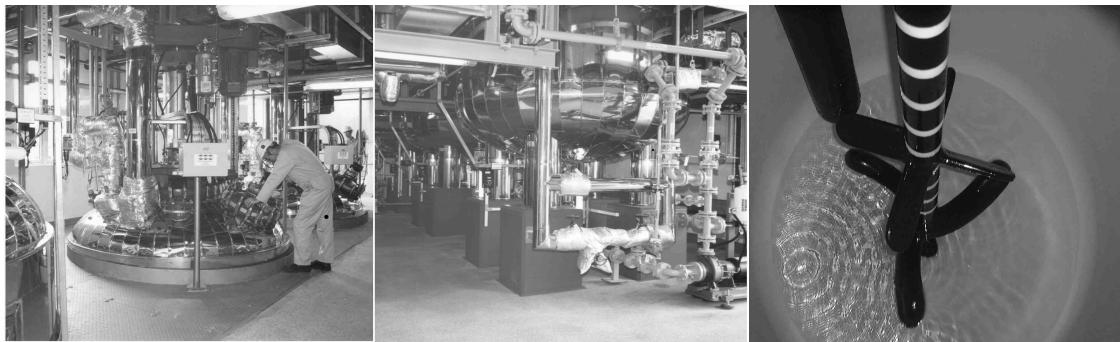


Figure 2.5: Upper section (top image), lower section (center image), and internals of batch crystallizer, showing the impeller and temperature sensor. This crystallizer is used for contract pharmaceutical and specialty chemical manufacturing. Images courtesy of Avecia, Ltd.

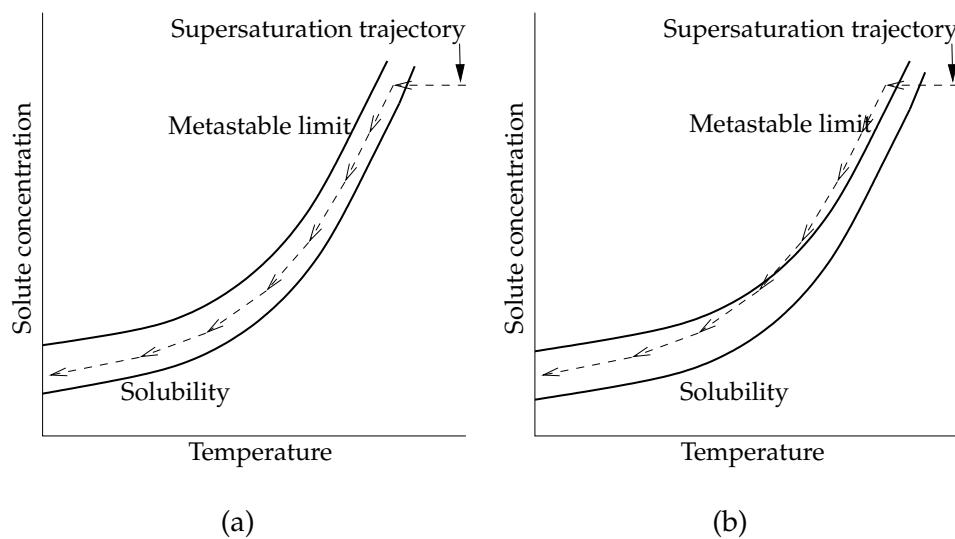


Figure 2.6: Batch cooling crystallization. In this illustration, the process is cooled until it becomes supersaturated and crystallization can occur. As the solute species deposit onto the forming crystals, the solute concentration decreases. Supersaturation is therefore maintained by further cooling. As shown in (a), a well-controlled crystallization process operates in the metastable zone between the saturation concentration and metastable limit, balancing the nucleation and growth rates to achieve the desired crystal size distribution. As depicted in (b), disturbances such as impurities can shift the metastable zone, resulting in undesired nucleation that substantially degrades the resulting particle size distribution.

larger scale (tens to hundreds of milliliters) to characterize crystal growth and nucleation rates and to develop an operating policy (see Figure 2.6) that is robust to variations in mixing, seeding, and impurity levels. These studies minimize the difficulty in scaling up the process several orders of magnitude to the pilot scale. The operating policy is usually determined semi-quantitatively, using trial-and-error or statistical-design-of-experiment approaches. Process robustness is achieved by adopting a conservative operating policy at low supersaturation levels that minimize nucleation events and thus achieve larger, more uniform crystals. Operating at low supersaturation levels, far from the metastable limit, is important because the metastable limit is difficult to characterize and is affected by various process conditions that change upon scaleup, such as the size and type of vessel or impeller.

### **2.2.2 Controlled, measured, and manipulated variables**

The primary concern of most industrial crystallization processes is generating crystals with a particle size distribution (PSD) that enables efficient downstream processing. The controlled variable for most crystallization processes, however, is the supersaturation level, which is only indirectly related to the PSD. The supersaturation level affects the relative rates of nucleation and growth and thus determines the PSD. Because of its dependence on temperature and solution composition, the supersaturation level can be manipulated using various process variables such as the flow rate of the cooling medium to the crystallizer jacket and the flow rate of anti-solvent to the crystallizer.

Process development studies use a wide range of measurement technology. This technology includes, for example, turbidity probes to detect the presence of solid material, laser scattering to characterize particle size distributions, and spectroscopic or absorbance probes to measure solute concentrations. However, large-scale, industrial crystallizers rarely have these advanced measurements available. In fact, controllers for most industrial crystallizers rely primarily on temperature, pressure, and flow rate measurements.

## 2.3 Recent advances in crystallization technology

The above discussion illustrates the limited technology used to control industrial crystallization processes. The obstacles that have hindered the implementation of advanced PSD control in industry, however, are being overcome by recent advances in measurement and computing technology [17],[103].

With improved control technology, additional challenges can be addressed. One challenge is to control shape, which, like PSD, affects the efficiency of downstream processes such as solid-liquid separation, drying, mixing, milling, granulation, and compaction. In some cases, particularly for chemicals having low solubility or low permeability, the crystal size and shape affect product properties such as bioavailability and tablet stability. Chemical purity must also be controlled, especially for food and pharmaceutical products intended for consumption and for semiconductor devices requiring highly consistent properties.

Perhaps the most difficult and important challenge is controlling *polymorphism*, which is the ability of a chemical species to crystallize into different crystal structures. The polymorphic form affects product characteristics, including stability, hygroscopicity, saturation concentration, dissolution rate, and bioavailability. The development of increasingly complex compounds in the pharmaceutical and specialty chemical industries makes polymorphism a commonly observed phenomenon for which control is essential. The recent disaster at Abbott Labs [10], in which the appearance of an unknown polymorphic form of ritonavir in drug formulations threatened the supply of the life-saving AIDS treatment Norvir, illustrates both the importance and difficulty of controlling polymorphism. In the following sections, we describe recent advances that impact industrial crystallizer control.

### 2.3.1 Spectroscopic and laser-based monitoring

One of the major challenges in implementing feedback control for crystallization processes is the lack of adequate online sensors for measuring solid-state and solution properties. The United

United States Food and Drug Administration's (FDA) Process Analytical Technology initiative, aimed at improving pharmaceutical manufacturing practices [138], has accelerated the development and use of more advanced measurement technology. We describe several recently developed sensors for achieving better control and understanding of crystallization processes.

### **ATR-FTIR Spectroscopy**

Attenuated total reflectance Fourier transform infrared (ATR-FTIR) spectroscopy imposes a laser beam on a sample and measures the amount of infrared light absorbed at different frequencies. The frequencies at which absorption occurs indicate which chemical species are present, while the absorption magnitudes indicate the concentrations of these species. As demonstrated in [32] and [31], ATR-FTIR spectroscopy can be used to monitor solute concentration in a crystallization process *in situ*.

ATR-FTIR spectroscopy offers advantages over prior techniques, such as refractometry, densitometry, and conductivity measurements, for measuring solute concentration. Refractometry works only if there is a significant change in the refractive index with solute concentration and is sensitive to air bubbles. Densitometry requires sampling of the crystal slurry and filtering out the crystals to accurately measure the liquid-phase density. This sampling process involves an external loop that is sensitive to temperature fluctuations and subject to filter clogging. Conductivity measurements, which are useful only for electrolytes, require frequent re-calibration. ATR-FTIR spectroscopy overcomes these problems and can measure multiple solute concentrations. Calibration of ATR-FTIR is usually rapid [72] and thus well suited for batch processes and short production runs. In [119], linear chemometrics is applied to estimate solute concentration with high accuracy (within 0.12%). Several applications for which ATR-FTIR monitoring is useful are described in [37].

Unfortunately, ATR-FTIR spectroscopy is considerably more expensive than the alternatives. Another drawback of ATR-FTIR is the vulnerability of the IR probe's optical material to chemical attack and fouling [30].

### Raman spectroscopy

Raman spectroscopy imposes a monochromatic laser beam on a sample and measures the amount of light scattered at different wavelengths. The differences in wavelength between the incident light and the scattered light is a fingerprint for the types of chemical bonds in the sample. Raman spectroscopy has been used to make quantitative polymorphic composition measurements since 1991 [26]. This technology has been applied to quantitative, *in situ* polymorphic composition monitoring in solution crystallization since 2000 [125]–[91].

Raman spectroscopy is well suited to *in situ* polymorphism monitoring for several reasons. Specifically, Raman analysis does not require sample preparation; the Raman signal can be propagated with fiber optics for remote sensing; and Raman sampling probes are less chemically sensitive than ATR-FTIR probes [30]. In addition, this technique can be used to monitor the solid and liquid phases simultaneously [34],[50].

Like ATR-FTIR, Raman-based technologies are expensive. Furthermore, calibration of the Raman signal for quantitative polymorphic composition measurements can be difficult because the signal intensity is affected by the particle size distribution. Hence Raman's utility for quantitative monitoring depends on corrections for particle-size effects [93].

### Near-Infrared Spectroscopy

Near-infrared (NIR) spectroscopy is also used to quantitatively monitor polymorphic composition [90]. Like Raman, NIR is well suited for *in situ* analysis. The main drawback of NIR is that calibration is difficult and time consuming. In some cases, however, coarse calibration is sufficient to extract the needed information [38].

### Laser backscattering

Laser backscattering-based monitoring technology, such as Lasentec's FBRM probe, has proven useful for characterizing particle size and for determining saturation concentrations and metastable

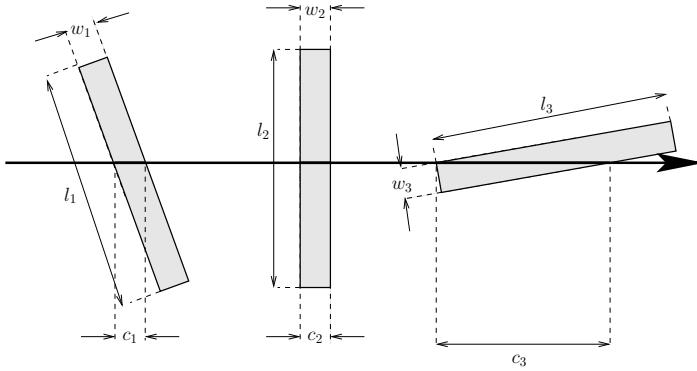


Figure 2.7: Comparison of crystal size measurements obtained using laser backscattering versus those obtained using vision. Laser backscattering provides chord lengths ( $\tilde{c}_1, \tilde{c}_2, \tilde{c}_3$ ) while vision-based measurement provides projected lengths ( $l_1, l_2, l_3$ ) and projected widths ( $w_1, w_2, w_3$ ). The chord-length measurement for each particle depends on its orientation with respect to the laser path (depicted above using the red arrow), while the projected length and width measurements are independent of in-plane orientation. Size measurements from both techniques are affected by particle orientation in depth.

limits [8],[9]. This sensor measures particle *chord lengths* (see Figure 2.7) by moving a laser beam at high velocity through the sample and recording the *crossing times*, that is, the time durations over which light is backscattered as the laser passes over particles. The chord length of each particle traversed by the laser is calculated as the product of the laser's velocity and the crossing time of the particle. This technique allows rapid, calibration-free acquisition of thousands of chord-length measurements to robustly construct a chord length distribution (CLD). Laser backscattering technology can be applied *in situ* under high solids concentrations.

Because laser-backscattering provides a measurement of only chord length, this technique cannot be used to measure particle shape directly. Laser-backscattering therefore cannot measure the distribution of particles between different shape classes (e.g. number of needles relative to number of spheres) nor shape factor distributions (e.g. distribution of aspect ratios). Also, inferring the PSD from the CLD involves the solution of an ill-posed inversion problem. Although

methods for solving this inversion problem are developed in [107, 133, 73], these methods depend on assumptions regarding particle shape. Successful application of these methods has been demonstrated experimentally only for spheres [51] and octahedra [133], but has been demonstrated for high-aspect-ratio particles using simulations only [73].

### 2.3.2 Imaging-based monitoring

Video microscopy can be used to characterize both crystal size and shape. Furthermore, for chemical systems in which the polymorphs exhibit different shapes, such as glycine in water, video microscopy can be used to monitor polymorphic composition [19]. Obtaining all three of these measurements using a single probe reduces cost and simplifies the experimental setup. Video microscopy is also appealing because interpretation of image data is intuitive. Obtaining quantitative information from video images, however, requires image segmentation. Image segmentation means separating the objects of interest (e.g. the particles) from the background.

#### Commercial, imaging-based monitoring instruments

Most commercial, imaging-based, on-line particle size and shape analyzers solve the segmentation problem by imaging the particulate slurry as it passes through a specially-designed flow cell under controlled lighting conditions [3, p. 167]. Several commercial instruments of this type have recently become available, such as Malvern's Sysmex FPIA 3000 and Beckman-Coulter's RapidVUE (see [3] for a survey of other imaging-based instruments). The drawback of these instruments is that they require sampling, which is inconvenient, possibly hazardous, and raises concerns about whether the sample is representative of the bulk slurry [3, 7]. One exception is the Mettler Toledo Lasentec Particle Vision and Measurement (PVM) in situ probe. This probe is packaged with automatic image analysis software that does not give suitable results for most systems. The utility of in situ video microscopy has been limited primarily to qualitative monitoring because the nature of in situ images, which contain blurred, out-of-focus, and overlapping particles, has precluded the successful application of image analysis to automatically quantify particle size and shape [16].

### Challenges for segmentation of in situ images

Robust and efficient segmentation of in situ images is challenging for several reasons. First, in situ imaging typically requires illumination by reflected light in order to handle high solids concentrations. Thus, the particles appear in the image with non-uniform color and intensity such that thresholding methods are ineffective. The use of reflected light can also result in poorly-defined particle outlines, thus limiting the robustness of methods based on closing the particle outlines, such as the technique proposed by [20]. Segmentation is simplified considerably if particles are imaged using transmitted light because the particle outlines are easily distinguished [98]. Second, the crystals can have a large variation in size and are randomly-oriented in 3-D space, which means the projections of the crystals onto the imaging plane can take on a wide variety of shapes. Hough-transform-based methods [42, p. 587], which have been applied extensively to segment images of circular or elliptical particles and droplets, involve exhaustive searches over the particle location and shape parameter space and are therefore computationally infeasible for randomly-oriented particles of complex shapes. The image segmentation problem is further complicated by particle agglomeration, overlap, breakage, and attrition, which result in occluded or poorly-defined particle boundaries.

Methods for segmenting in situ images have been developed for circular particles [113] and elliptical particles [48]. In [20] a technique is presented for automatic segmentation of in-process suspension crystallizer images, but the technique was demonstrated only on images that appear to have been acquired at low solids concentration where there are no overlapping particles and the particles' edges are well-defined. Kaufman and Scott [62] used an in situ fluorescence imaging method that caused the liquid phase of a fluidized bed to fluoresce while leaving the coal particles opaque, thus enabling a gray-level threshold method to detect particle edges. However, for dense particle volume fractions, manual intervention was required to determine which of the segmented particles could be used for sizing.

## Monitoring dynamic crystal populations

In situ video imaging is used by Scholl et al. [112] to observe the solvent-mediated polymorphic transformation of L-glutamic acid from the  $\alpha$  form with prismatic morphology to the  $\beta$  form with needle-like morphology. Monnier et al. [86] use off-line image analysis to characterize the final relative PSD of adipic acid crystals. Pollanen et al. [97] use video microscopy and automated image analysis to characterize the sizes and shapes of sulphathiazole crystals. The application of video imaging to these crystallization processes, however, is limited to qualitative monitoring or characterization of end-product properties.

Only a few studies have used video imaging for on-line, quantitative monitoring of crystal population dynamics. Patience et al. [95] used video microscopy to monitor the evolution of crystal size mean and standard deviation for needle-like pharmaceutical particles. The slurry was sampled periodically and allowed to settle on a microscope stage, and the images were analyzed manually. Calderon De Anda et al. [19] used non-invasive imaging and automatic image analysis to quantify the polymorphic fraction during the transformation from  $\alpha$  to  $\beta$  L-glutamic acid. In a follow-up study, Dharmayat et al. [27] monitored on-line the L-glutamic acid transformation using both on-line video microscopy and X-ray diffraction, but were unable to compare quantitatively the measurements of the two methods because the video microscopy technique failed at higher solids concentrations and the X-ray diffraction method was not sufficiently sensitive at low solids concentrations. Qu et al. [101] use in-line imaging to monitor the evolution of crystal length and width CDFs at 10 to 20 minute intervals, determining width and length growth rates based on the mean particle size.

Besides these crystallization applications, video imaging has been utilized to monitor dynamics of other particulate processes. Watano and Miyanami [127] demonstrated on-line monitoring of the median diameter and shape factor for a wet granulation process using in situ video imaging and automatic image analysis. Blandin et al. [14] used non-invasive imaging of an agglomeration process and automatic image analysis to track the evolution of the particle size num-

ber fractions at 30 minute intervals for about 4 hours. They validated their automated image analysis method using comparisons with manual image analysis. Hukkanen et al. [52] used *in situ* video microscopy and automated image analysis to monitor various PSD moments during the early stages of a suspension polymerization processes.

### 2.3.3 Manipulated variables for crystal shape and polymorphic form

In advanced control implementations, the quality variables of interest (crystal size, shape, form, and purity) are indirectly controlled using manipulated variables that affect the supersaturation level in the crystallizer. For example, cooling crystallizers manipulate the crystallizer temperature to change the saturation concentration of the crystallizing species. Anti-solvent crystallizers change the saturation concentration by manipulating the solvent composition.

Controlling the supersaturation level provides only limited control over the resulting crystal shape distribution, size distribution, and polymorphic form. Progress is being made in this area as well, however. Several research groups are investigating additives that bind to selected crystal faces to inhibit growth of the faces, thereby promoting a desired crystal shape or polymorphic form [128],[13]. These additives, which are similar to the target-crystallizing species, can be incorporated onto a growing crystal face. The remaining exposed portion of the additive consists of chemical groups that hinder further growth on that face. Additives are also used as nucleation promoters or inhibitors to obtain a desired polymorphic form [129]. Templates, such as single-crystal substrates, are also being investigated as a means for manipulating nucleation events to obtain desired polymorphs [85].

Additives are used in industry to help understand the effect of impurities on crystal growth and polymorphism [13]. As yet, additives and templates are not widely used as manipulated variables in industrial crystallizer control schemes. To use these methods as manipulated variables for advanced control, models are needed to describe the effect of the manipulated variables on crystal size, shape, and form.

### 2.3.4 Modeling and prediction of crystal size, shape, and polymorphic form

#### Process Modeling

Crystallizers have highly nonlinear, complex dynamics including multiple steady states, open-loop instability, and long time delays; hence low-order, linear models are often inadequate for control purposes. Furthermore, nonlinear black box models, such as neural networks, are also inadequate in many cases because batch crystallizers have such a large operating region. Online, optimal control of dynamic crystallization is thus enhanced by the ability to efficiently simulate the underlying physical model, which is a system of partial integro-differential equations that couples mass and energy balances with a population balance describing the evolution of the crystal population's PSD.

The evolution of the PSD often involves sharp, moving fronts that are difficult to simulate efficiently. Several software packages are now available for simulating crystallization processes. The commercial software package PARSIVAL [135] is designed to handle the entire class of crystallizer configurations and crystallization phenomena. AspenTech's process simulation software has specific tools for crystallization simulation and troubleshooting. The crystallizer simulator of the GPROMS package can be interfaced with computational fluid dynamics (CFD) software, such as FLUENT or STAR-CD. DYNOCHEM solves the population balance equations by applying the commonly used method of moments, a model reduction technique. Researchers have developed techniques that extend the applicability of the method of moments to systems involving length-dependent crystal growth [79],[88], and these developments are being incorporated directly into CFD software packages [111].

#### Shape Modeling

Models and methods for predicting crystal shape based solely on knowledge of the internal crystal structure are available in software packages such as CERIUS<sup>2</sup> and HABIT [23]. These methods provide accurate shape predictions for vapor-grown crystals but, for solution-grown crystals, do

not take into account the effects of supersaturation, temperature, solvent, and additives or impurities. Current shape-modeling research is focused on accounting for these effects [131]–[106].

### **Polymorphism Modeling**

The ability to predict the crystal structure for a given molecule in a given environment would represent a major advance in drug development. Although significant progress has been made in making these predictions [99],[25], this problem is far from being solved. Most approaches seek to find the crystal structure that corresponds to the global minimum in lattice energy, that is, the most thermodynamically stable form at zero Kelvin. These approaches neglect entropic contributions arising at higher temperatures as well as kinetic effects due to the experimental crystallization conditions. Current polymorphism modeling methods thus cannot reliably predict the polymorphs that are observed experimentally.

#### **2.3.5 Control**

The developments described above impact the way crystallizer control is approached in industry. In particular, the development of better measurement technology enables the application of simple but effective crystallizer control strategies. Most of these strategies focus on feedback control of supersaturation using concentration (by means of ATR-FTIR) and temperature measurements to follow a predefined supersaturation trajectory [74]–[40]. This approach is attractive because it can be implemented without characterizing the nucleation and growth kinetics, using only the saturation concentration and metastable zone width data. Furthermore, this approach results in a temperature profile that can be used in large-scale crystallizers that do not have concentration-measurement capabilities.

Laser backscattering measurements are used to control the number of particles in the system [29]. This control strategy alternates between cooling and heating stages, allowing the heating stage to continue until the particle count number measured by FBRM returns to its original value upon seeding, indicating that all fine particles generated by secondary nucleation have been dis-

solved. In [96], online crystal shape measurements obtained by means of optical microscopy and automated image processing are used to manipulate impurity concentration and thereby control crystal habit. The control schemes employed in these experimental studies are basic (usually PID or on-off control), illustrating that, given adequate measurement technology, simple control schemes can often provide an adequate level of control capability.

More sophisticated control methods have also been demonstrated for batch crystallizers. Experimental results obtained in [84]–[134] demonstrate product improvement by using predictive, first-principles models to determine open-loop, optimal cooling and seeding policies. Closed-loop, optimal control of batch crystallizers is demonstrated using simulations in [105],[114].

For continuous crystallizers, various model-based, feedback controllers have been suggested. In [124], an  $H_\infty$  controller based on a linearized distributed parameter model is shown to successfully stabilize oscillations in a simulated, continuous crystallizer using measurements of the overall crystal mass, with the flow rate of fines (small crystals) to the dissolution unit as the manipulated variable. In [114], a hybrid controller combining model predictive control with a bounded controller is used to ensure closed-loop stability for continuous crystallizers.

## 2.4 The future of crystallization

As the chemical, pharmaceutical, electronics, and food industries continue to develop new products, crystallization will enjoy increasingly wide application as a means to separate and purify chemical species and create solids with desirable properties. Sensor technology for crystallizers will continue to improve, especially given the current emphasis by the FDA's Process Analytical Technology initiative. Industries that have used batch crystallizers to produce low-volume, high-value-added chemicals might choose to move to continuous crystallizers to reduce operating costs and enable more flexible and compact process design. Shape modeling and molecular modeling tools offer tremendous potential for enabling robust process design and control, and these tools can be expected to advance rapidly given the amount of interest and research in this area. These

developments will impact the way in which crystallization processes are designed and will enable more effective control of size distribution, shape, and polymorphic form, leading to the creation of crystalline solids that are useful for a wide variety of applications.

## Chapter 3

# Crystallization Model Formulation and Solution

This chapter presents the batch crystallization model used to demonstrate the value of the imaging-based monitoring methods developed in this thesis. First, the mass, energy, and population balances describing the crystallizer are presented. Next, methods for solving the model are presented. The first method, called the method of moments, is useful for simulating important characteristics of the particle size distribution, such as the total number of particles per unit volume or the mean particle size. The second method, orthogonal collocation on moving finite elements, simulates the evolution of the particle size distribution directly.

### 3.1 Model formulation

The physical model of a batch crystallizer is a system of partial integro-differential equations that couples mass and energy balances with a population balance describing the evolution of the crystal population's PSD. This section presents these equations.

#### 3.1.1 Population balance

The population balance describes the population of crystals dispersed in the continuous liquid phase. A complete description of the crystal population would include solid-phase properties such as purity, polymorphic form, shape, and size. All of these properties can be included in the

population balance formulation. Since the focus of this thesis is on measuring the particle size distribution, however, we restrict ourselves here to a population balance that models only the size distribution of the crystal population.

Let  $f(L, t)$  be the particle size distribution, or the number of particles per unit volume with characteristic size  $L$  at time  $t$ . In the absence of crystal breakage and agglomeration, and assuming all crystals are nucleated at a negligibly small size  $L_0$ , the evolution of  $f$  is given by

$$\frac{\partial fV}{\partial t} + V \frac{\partial Gf}{\partial L} = VB^\circ \delta(L - L_0) - \sum_k Q_k f_k \quad (3.1)$$

in which  $V$  is the crystallizer slurry volume,  $G$  is the crystal growth rate,  $B^\circ$  is the nucleation rate density,  $\delta$  is the Dirac delta function,  $Q_k$  is the volumetric flow rate of the  $k$ th stream, and  $f_k$  is the particle size distribution of the  $k$ th stream. Letting  $L_0 \rightarrow 0$ , Equation (3.1) is equivalent to

$$\frac{\partial fV}{\partial t} + V \frac{\partial Gf}{\partial L} = - \sum_k Q_k f_k \quad (3.2)$$

with the boundary condition

$$f(0, t) = \frac{B^\circ}{G(L=0)} \quad (3.3)$$

For the batch crystallization processes modeled in this work, we assume  $V$  is constant,  $G$  is independent of particle size, and there are no input or output streams. The model used in this study is therefore given by

$$\frac{\partial f}{\partial t} + G \frac{\partial f}{\partial L} = 0 \quad (3.4)$$

with the same boundary condition as above.

To complete the model, the kinetic expressions for the nucleation and growth rates are required. The rates of these processes are known to depend on the supersaturation, or the degree to which the saturation concentration in the continuous phase is exceeded. To represent the supersaturation, define  $\hat{S}$  as the relative supersaturation, given by

$$\hat{S} = \frac{\hat{C} - \hat{C}_{\text{sat}}}{\hat{C}_{\text{sat}}} \quad (3.5)$$

In this study, the growth rate is assumed to follow the standard semi-empirical power law

$$G = k_g \hat{S}^g \quad (3.6)$$

in which  $k_g$  and  $g$  are growth rate constants that must be determined experimentally. The nucleation rate density is similarly given by

$$B^\circ = k_b \hat{S}^b \mu_3^j \quad (3.7)$$

in which  $k_b$ ,  $b$ , and  $j$  are constants to be determined experimentally, and  $\mu_3$  is the third moment of the PSD.

### 3.1.2 Mass balance

The growth and nucleation rates depend on the degree of supersaturation, which depends on both the saturation concentration and the solute concentration. Assuming, as above, that growth rate is size-independent and the system is closed (no input or output streams), the mass balance for the solute concentration is

$$\frac{d\hat{C}}{dt} = -3\rho_c k_v \hat{h} G \int_0^\infty f L^2 dL \quad (3.8)$$

in which  $\hat{C}$  is the solute concentration,  $\rho_c$  is the crystal density,  $k_v$  is a shape factor defined such that  $k_v L^3$  gives the volume of a crystal of characteristic length  $L$ ,  $\hat{h}$  converts solvent mass to slurry volume.  $\hat{C}$  is given in terms of mass of solute per total mass of mother liquor. The initial condition is given by  $\hat{C} = \hat{C}_0$ .

The saturation concentration  $\hat{C}_{\text{sat}}$  is determined experimentally. In this study, the saturation concentration is a quadratic function of temperature:

$$\hat{C}_{\text{sat}}(T) = 0.185 - 2.11 \times 10^{-2} T + 7.46 \times 10^{-4} T^2 \quad (3.9)$$

### 3.1.3 Energy balance

The energy balance gives the evolution of the system temperature, therefore completing the model formulation. Again assuming a closed system, the bulk temperature is determined from

$$\rho V c_p \frac{dT}{dt} = -3\rho_c k_v V \Delta H_c G \int_0^\infty f L^2 dL - UA(T - T_j(t)) \quad (3.10)$$

in which  $V$  is the slurry volume,  $c_p$  is the slurry heat capacity,  $T$  is the bulk temperature,  $\Delta H_c$  is the heat of crystallization,  $U$  is the overall heat transfer coefficient,  $A$  is the surface area of the slurry exposed to the crystallizer jacket, and  $T_j$  is the jacket temperature. The initial condition is  $T = T_0$ .

For the simulations used in this study, the energy balance is unnecessary as we assume perfect temperature control. That is, the temperature follows a fixed temperature trajectory without deviation.

## 3.2 Model solution

### 3.2.1 Method of moments

The method of moments is commonly used for simulating particulate populations. This method, formulated in the early 1960's by Hulbert and Katz [53], does not solve the population balance equations directly, but rather determines the moments of the population. The  $i$ th moment of a population with PSD  $f$  and characteristic length  $L$  is defined as

$$\mu_i = \int_0^\infty f L^i dL \quad (3.11)$$

Although the moments do not uniquely determine the PSD [102], in many cases the moments provide sufficient information to solve practical problems of interest, such as parameter estimation. The moments have useful physical interpretations. For example,  $\mu_0$  is the total number of particles per unit volume,  $\mu_1$  is the total length of particles per unit volume,  $k_a \mu_2$  is the total particle surface area per unit volume, and  $k_v \mu_3$  is the total volume of particles per unit volume. Furthermore,

these quantities are required to complete the mass and energy balances given in Equations (3.8) and (3.10).

To apply the method of moments to the model used in this study, we recall that for a batch crystallizer with size-independent growth and no input or output streams, the population balance is given by

$$\frac{\partial f}{\partial t} = -G \frac{\partial f}{\partial L} + B^\circ \delta(L - L_0) \quad (3.12)$$

Multiplying this equation by  $L^i$  and integrating over all sizes results in

$$\frac{\partial}{\partial t} \int_0^\infty f L^i dL = -G \int_0^\infty \frac{\partial f}{\partial L} L^i dL + \int_0^\infty B^\circ \delta(L - L_0) L^i dL \quad (3.13)$$

Applying integration by parts to the first term on the right hand side of this equation gives

$$-G \int_0^\infty \frac{\partial f}{\partial L} L^i dL = -G f L^i \Big|_0^\infty + iG \int_0^\infty f L^{i-1} dL$$

Assuming a finite number of crystals at size 0 and zero crystals of infinite size, the first term on the right hand side equals zero. Substituting these relationships into Equation (3.13) and letting  $L_0 \rightarrow 0$  gives

$$\frac{d}{dt} \int_0^\infty f L^i dL = iG \int_0^\infty f L^{i-1} dL + B^\circ 0^i \quad (3.14)$$

In terms of the moments, Equation (3.14) can be expressed as

$$\frac{d\mu_0}{dt} = B^\circ \quad (3.15)$$

$$\frac{d\mu_i}{dt} = G\mu_{i-1} \quad i = 1, 2, 3 \quad (3.16)$$

The method of moments therefore provides a set of coupled differential equations that can be solved efficiently by standard ODE solvers.

### 3.2.2 Orthogonal collocation

The orthogonal collocation method, as explained in [123], consists of approximating the model solution at each time step as an  $n$ th order polynomial. In our case, the PSD  $f$  is approximated as

$$f(L, t) = \sum_{j=1}^{n_c} f(L_j, t) l_j(L) \quad (3.17)$$

in which  $l_j(L)$  is the Lagrange interpolation polynomial of degree  $j$ ,  $n_c$  is the number of collocation points, and  $f(L_j, t)$  is the PSD evaluated at length  $L_j$ . Given this formulation, the spatial derivative in Equation (3.4) can be calculated as a linear combination of the model solution values at  $n_c$  collocation locations along the crystal size domain, i.e.

$$\frac{df}{dL} \Big|_{L_i} = \sum_{j=1}^{n_c} A_{ij} f_j \quad (3.18)$$

$$(3.19)$$

in which  $f_j = f(L_j, t)$  and  $A_{ij} = (dl_j/dL)|_{L_i}$  is an element of the derivative weight matrix. Furthermore, the integral of any function of the PSD over the domain  $\Omega_L$  can be calculated using quadrature:

$$\int_{\Omega_L} g(f) dL = \sum_{j=1}^{n_c} Q_j g(f_j) \quad (3.20)$$

in which  $Q_j$  is the  $j$ th quadrature weight. The derivative and integral weight matrices can be computed using the COLLOC function within OCTAVE. The matrices are computed for the domain  $[0,1]$ , so the model must be transformed onto the domain  $[0,1]$ .

Neglecting crystal agglomeration and breakage, we expect the PSD to be non-zero on two distinct domains along the length scale. The first domain corresponds to the subpopulation of nucleated crystals while the second domain corresponds to the subpopulation of seeded crystals. The limits of these domains change with time. In the following, we describe how the PSD can be simulated using collocation on two distinct finite elements with time-varying domains.

### Nucleated crystals

Let  $f_N(L, t)$  denote the PSD of the subpopulation of nucleated crystals, and let  $L_N(t)$  be the size of the largest nucleated crystal, the evolution of which is given by

$$\frac{dL_N}{dt} = G$$

with  $L_N(0) = 0$ . Implementing the collocation method requires a change from the fixed coordinate  $L$ , which denotes absolute particle size, to the moving coordinate  $\zeta$ , which denotes a scaled particle

size on the domain [0,1]. Noting that the size domain of the nucleated crystals is bounded by [0  $L_N$ ] (assuming crystals are nucleated at negligible size), we define the transformed variable  $\zeta(L, t) = L/L_N(t)$ . Equation (3.4) can be transformed to the new coordinate system as follows:

$$\begin{aligned} f_N(L, t) &= \bar{f}_N(\zeta(L, t), t) \\ \frac{\partial f_N}{\partial t} \Big|_L &= \frac{\partial \bar{f}_N}{\partial \zeta} \frac{\partial \zeta}{\partial t} + \frac{\partial \bar{f}_N}{\partial t} \\ \frac{\partial f_N}{\partial L} \Big|_t &= \frac{\partial \bar{f}_N}{\partial \zeta} \frac{\partial \zeta}{\partial L} \end{aligned}$$

It is straightforward to show that

$$\begin{aligned} \frac{\partial \zeta}{\partial t} &= -\frac{\zeta G}{L_N} \\ \frac{\partial \zeta}{\partial L} &= \frac{1}{L_N} \end{aligned}$$

These relationships can be substituted into Equation (3.4) to give the transformed population balance

$$\frac{\partial \bar{f}_N}{\partial t} = \frac{G(\zeta - 1)}{L_N} \frac{\partial \bar{f}_N}{\partial \zeta} \quad (3.21)$$

with boundary condition  $\bar{f}_N(0, t) = B^\circ/G$ .

Applying the collocation equations to the transformed population balance results in the following set of DAEs:

$$\frac{d\bar{f}_{N_i}}{dt} = \frac{G(\zeta_i - 1)}{L_N} \sum_{j=1}^{n_c} A_{ij} \bar{f}_{N_j}, \quad i = 2, \dots, n_c \quad (3.22)$$

$$\bar{f}_{N_1} = \frac{B^\circ(t)}{G(t)} \quad (3.23)$$

in which  $\bar{f}_{N_j} = \bar{f}_N(\zeta_j, t)$  and  $\zeta_j$  is the  $j$ th collocation location. The initial conditions are

$$\bar{f}_{N_i}|_{t=0} = \frac{B^\circ(0)}{G(0)}, \quad i = 1, \dots, n_c \quad (3.24)$$

This formulation is problematic at  $t = 0$  because  $L_N(0) = 0$ . Equation (3.22) can be integrated, however, by noticing that  $\partial \bar{f}_N / \partial \zeta$  also equals zero at  $t = 0$ . Thus, L'Hôpital's rule can be applied to replace the problematic ratio  $\sum_j A_{ij} \bar{f}_{Nj} / L_N$  with  $\sum_j A_{ij} \frac{d\bar{f}_{Nj}}{dt} / G$  for small  $t$ .

### Seed crystals

Let  $f_S(L, t)$  denote the PSD of the subpopulation of seed crystals. Let  $L_{S_u}(t)$  and  $L_{S_l}$  be the characteristic lengths of the largest and smallest seed crystals, respectively. The evolution of  $L_{S_u}$  and  $L_{S_l}$  are given by

$$\begin{aligned}\frac{dL_{S_u}}{dt} &= G \\ \frac{dL_{S_l}}{dt} &= G\end{aligned}$$

with initial values  $L_{S_u}(0) = L_{S_{u0}}$  and  $L_{S_l}(0) = L_{S_{l0}}$  assumed known. The initial seed distribution is assumed to be a symmetric, quadratic function that equals zero at  $L_{S_{l0}}$  and  $L_{S_{u0}}$ :

$$f_{S_0}(L) = a_S(L^2 - L(L_{S_u} + L_{S_l}) + L_{S_u}L_{S_l})$$

in which  $a_S$  is a constant determined by solving the equation

$$\rho_c k_v V \int_{L_{S_{l0}}}^{L_{S_{u0}}} f_S L^3 dL = m_S(t = 0) \quad (3.25)$$

in which  $m_S(t = 0)$  is the mass of injected seeds. Equation (3.25) provides proper initialization of the mass balance by ensuring that the mass corresponding to the third moment of the seed distribution equals the mass of seeds injected into the crystallizer.

With the assumption of size-independent growth,  $f_S(L, t)$  can be calculated simply by shifting the initial seed distribution  $f_{S_0}(L)$  based on the value of  $L_{S_u}$ :

$$f_S(L, t) = f_{S_0}(L_{S_{u0}} - (L_{S_u}(t) - L)) \quad (3.26)$$

## **Chapter 4**

# **Experimental and Simulated Image**

## **Acquisition**

This chapter describes the experimental and simulation methods used to obtain imaging data. First, the crystallizer and data acquisition hardware used to obtain *in situ* crystallization images are described along with experimental procedures. The different chemical systems for which imaging data are acquired are discussed next. Finally, the simulation methods used to generate artificial images are presented.

### **4.1 Crystallizer and imaging apparatus**

#### **4.1.1 Crystallizer**

The experimental setup for the crystallization experiments is depicted in Figure 4.1. The crystallizer is a 500 mL, flat-bottomed, jacketed, glass vessel (Wilmad-LabGlass, LG-8079C). Mixing is achieved in the crystallizer using a 3.8 cm marine-type stainless steel impeller driven by a motor controller with a speed range of 0 to 1250 revolutions per minute. A stainless steel draft tube is used to enhance mixing. The crystallizer temperature is controlled using automatic feedback control.

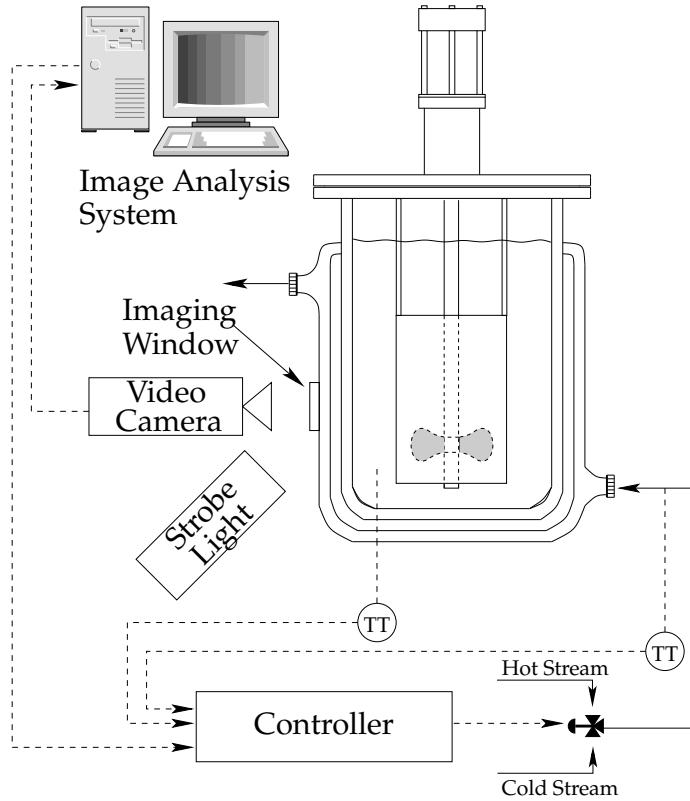


Figure 4.1: Experimental setup for obtaining in situ crystallization images.

#### 4.1.2 Data acquisition

Data is acquired using a custom-built PC connected via the serial port to National Instruments FieldPoint data I/O modules. The same PC also acquires imaging data using PCI frame grabber cards. This section describes this hardware and the necessary software in detail.

#### PC

The computer has an AMD Athlon XP 2600 2.08 GHz processor, an Epox 8RDA3i motherboard, and 1 GB RAM. The computer is equipped with a 40 GB Seagate IDE hard drive, a 37 GB SCSI hard drive, an additional 80 GB IDE hard drive, and a Samsung CD-writer. The operating system is Windows 2000.

## **LabVIEW**

National Instrument Measurement and Automation Explorer software (version 4.0) is used to configure all I/O devices and interfaces. Once configured, the devices can be used for data acquisition and control by setting up Virtual Instruments (VIs) using National Instruments LabVIEW software (version 8.0). The VIs are programmed using the LabVIEW G Language and provide a convenient operator interface for controlling and configuring experiments. For the experiments relevant to this thesis, a VI called “Batch Crystallization Video.vi” has been set up that enables batch cooling crystallization for a specified temperature trajectory while acquiring and analyzing images at specified intervals. The setpoint trajectory tracking temperature controller is an LQR-PID cascade controller and is described in [94, p. 147].

## **Signal conditioning with Compact FieldPoint**

The signals from the various sensors are conditioned using National Instruments Compact FieldPoint (CFP) signal conditioning modules. Wire connections are made using CFP connector blocks (cFP-CB-1). All connector blocks and conditioning modules are connected to a backplane (cFP-1804), which transfers the signals to the PC via the serial port.

## **Temperature**

Temperature measurements are made using Omega 3-wire,  $100\Omega$ , platinum resistance temperature detectors (RTDs) (model PR-13-2-100-1/8). A CFP signal conditioning module (cFP-RTD-122) provides a 0.25 mA excitation current to each RTD and scales the voltage output signal.

## **pH**

Measurements of pH are made using a double-junction, flat membrane pH electrode with epoxy body (Weiss Research, PHF-0281-3B). The electrode mV signal is amplified using a pH preamplifier (Newport Electronics, PHAMP-1) to -2 to 2 V and connected to the CFP signal conditioning

module (cFP-AIO-600). The voltage signal is converted to pH using the LabVIEW VI “pH meter.vi.”

#### 4.1.3 Video image acquisition

The imaging system used in this study was developed by researchers at GlaxoSmithKline and consists of a monochrome CCD video camera (Sony XC-55) synchronized with a xenon strobe light (Active Silicon, Model No VS-200-30) and connected to the PC via a frame grabber (National Instruments, PCI-1410). Images are acquired at a rate of 30 frames per second. The camera gives images of 480 x 640 pixels and is fitted with a lens (Moritex, x2 magnification) providing a 280  $\mu\text{m}$  depth of field and 2.48 x 1.87 mm field of view. The camera and strobe are placed to the side of the crystallizer roughly forty-five degrees apart using Manfrotto positioning arms.

The images are acquired through an “optical flat” attached to the side of the vessel to minimize distortion due to the curved surface of the vessel. The optical flat is created by cutting a 1-in. mounting square (Scotch) to form a U shape. This U-shaped double-sided adhesive is attached (with the U upright) on one side to the crystallizer outer body, and on the other side to a 1 in (25 mm sq) microscope slide cover glass (Corning, No. 1). The small gap created by this arrangement is filled with Permount and allowed to dry.

The camera and strobe are synchronized using a special-purpose cable (National Instruments, Model IMAQ A8055) that consists of a main cable connecting the camera to the frame grabber with break-out cables to access trigger lines and power supply lines. The wiring system is shown in Figure 4.2, in which the IMAQ A8055 cable is shown as a dashed line. NI-IMAQ driver software is required to trigger image acquisition and strobing. The VI “imacquire\_continuous.vi” demonstrates image acquisition with synchronized strobing.

Although not used for the studies described in this thesis, an additional frame grabber (National Instruments, PCI-1405) is installed in the PC that enables image acquisition from RGB video devices, such as video cameras mounted to microscopes.

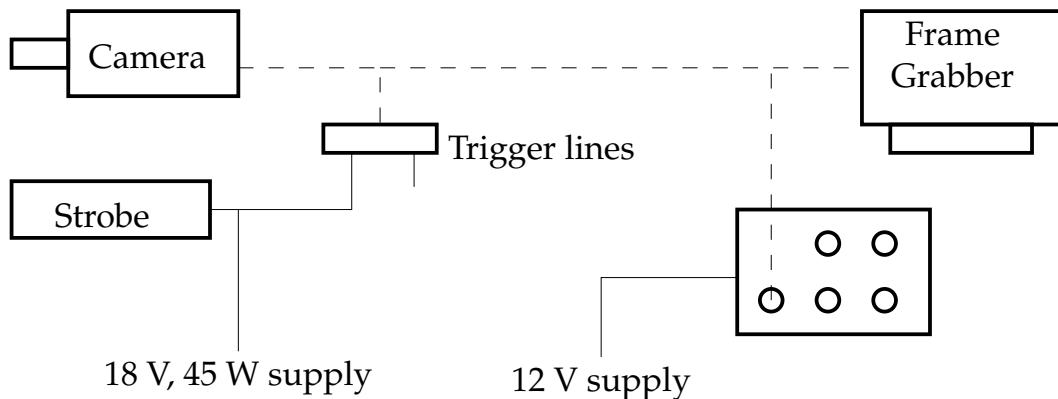


Figure 4.2: Imaging system wiring.

#### 4.1.4 Operating procedure

The following steps describe how the apparatus described above is operated for batch crystallization.

1. Open the VI “Batch Crystallization Video.vi” (hereafter referred to as simply “VI”) in LabVIEW 8.0. On the front panel, ensure all toggle switches are off (manual mode) and that the manual control valve position is specified as closed (i.e. fraction open = 0). Start the VI to apply the specified control valve position. Running the VI in manual mode is necessary also to set the initial valve position required for subsequent PID controller calculations.
2. Load crystallizer with appropriate amounts of solvent and crystallizing species. Generally 600 mL of solvent are used. For the industrial pharmaceutical experiments, 17.7 g solid material are added. For the glycine experiments, the vessel is charged with sufficient solid material to achieve 20% supersaturation at the desired crystallization temperature.
3. Attach lower and upper jacket ports to jacket inlet and outlet streams, respectively.
4. Turn on impeller. Check slurry visually to ensure impeller speed is sufficiently high to produce a well-mixed suspension. A speed of 500-600 RPM is typically used for the industrial pharmaceutical and glycine experiments.

5. Turn on 3-way 0.5 in. control valve (Badger Meter, Model 1002), water supply, and gear pump (Baldor, Model 220/56C). The control valve should be closed such that only cold water is fed to the crystallizer jacket initially, thereby allowing time to adjust the camera and strobe before the solid material dissolves.
6. Adjust position of camera and strobe, using “imacquire\_continuous.vi” to preview image quality. Superior image quality is generally achieved if the camera is focused at the slurry/glass interface with the strobe at about a forty-five degree angle to the camera.
7. Turn on the electric heater (Advantage Sentra, Model S-925) and set the hot water setpoint. The setpoint is 80 °C for the industrial pharmaceutical and RT glycine experiments. For glycine crystallization at higher temperature, a setpoint of 120 ° is used.
8. Set inputs for VI:
  - Cooling profile input file
  - Filename for temperature, transmittance, and control valve position data
  - Imaging parameters:
    - Number of images per acquisition
    - Time between acquisitions
    - Image analysis algorithm
    - Filename for image analysis data
9. Stop VI, if running. Turn “PID Control” toggle switch on. Set “Manual Jacket Temp Set-point” to desired temperature (60 °C for industrial pharmaceutical, 55 °C for RT glycine, 65 °C for high temperature glycine). Run VI, allowing solid material to dissolve.
10. Stop VI. Turn “MPC Control”, “Store Data” and “Acquire Images” toggle switches on. Run VI. With these toggle switches on, the LQR-PID cascade setpoint trajectory tracking con-

troller attempts to follow the pre-specified temperature trajectory given in the cooling profile input file, and all imaging and other data is stored on the hard drive according to the specified data files.

11. Inject seeds at appropriate time. For the industrial pharmaceutical, 1.4 g of seeds are injected 160 minutes after the cooling process begins.

## 4.2 Chemical systems

Three different chemical compounds are considered in this work, including an industrial pharmaceutical, an industrial photochemical, and glycine. This section describes important features associated with the crystallization of these compounds.

### 4.2.1 Industrial pharmaceutical

The industrial pharmaceutical is a proprietary compound manufactured by GlaxoSmithKline. In this study, the compound is crystallized in iso-propyl alcohol (IPA) and water (93/7 vol%), producing a polymorph with parallelepiped, needle-like shape. This compound has been studied extensively by Patience et al. [94, 95] and is used for image analysis algorithm validation in Chapter 5.

### 4.2.2 Industrial photochemical

The industrial photochemical is a proprietary compound manufactured by Kodak in Rochester, New York. The compound is crystallized in heptane and has a parallelepiped shape that impedes accurate size distribution measurement. The nucleation and growth kinetics of this compound have been characterized by Matthews [81]. This system is studied in Chapter 9.

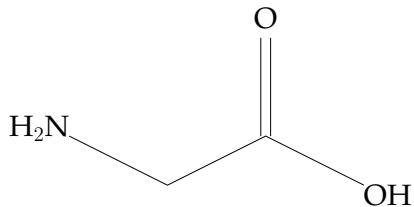


Figure 4.3: Chemical structure of glycine.

#### 4.2.3 Glycine

Glycine ( $\text{C}_2\text{H}_5\text{NO}_2$ ) is an amino acid with a simple, non-chiral structure, as shown in Figure 4.3. Glycine is of interest to the pharmaceutical community, both as an excipient in pharmaceutical formulations and as an active ingredient. Glycine exists in three polymorphic forms:  $\alpha$  [60],  $\beta$  [56], and  $\gamma$  [57]. It is known that crystallization of glycine in water under basic or acidic pH produces the stable  $\gamma$  form while crystallization under neutral pH produces the metastable  $\alpha$  form [137, 120]. Both industrial and academic researchers have proposed process design and control strategies for effective batch crystallization of the desired glycine polymorph [29, 87].

The solution-mediated transformation of  $\alpha$  to  $\gamma$  has been investigated [109], as has the transformation of  $\beta$  to  $\alpha$  [36]. In this thesis, glycine is used for image analysis algorithm validation because of its many complex shapes. Crystallized in aqueous solution ( $\text{pH} = 6.2$ ) at room temperature, the  $\alpha$  form has a prismatic morphology as shown in Figure 4.4. Crystallized under the same conditions except at higher temperature ( $50^\circ\text{C}$ ), the  $\alpha$  form has a bullet or pencil-head shape, as shown in Figure 4.5. Crystallized in basic conditions ( $\text{pH} = 8.3$ ), the  $\gamma$  form has a bi-pyramidal shape, as shown in Figure 4.6.

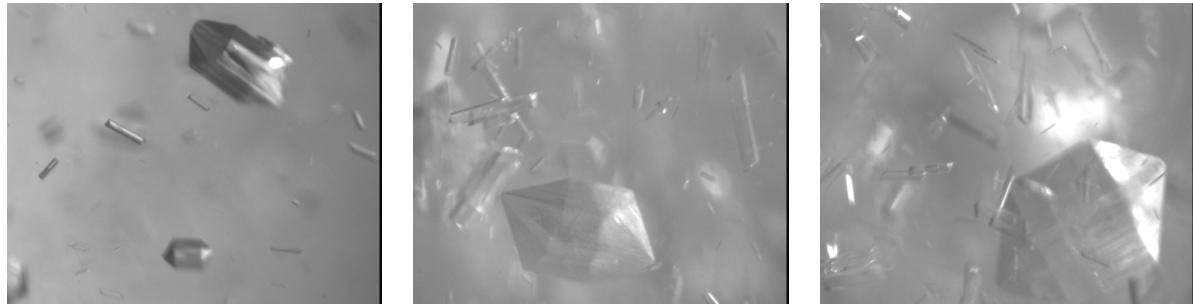


Figure 4.4: Images illustrating morphology of  $\alpha$ -glycine crystallized in water at room temperature.



Figure 4.5: Images illustrating morphology of  $\alpha$ -glycine crystallized in water at 55 °C.

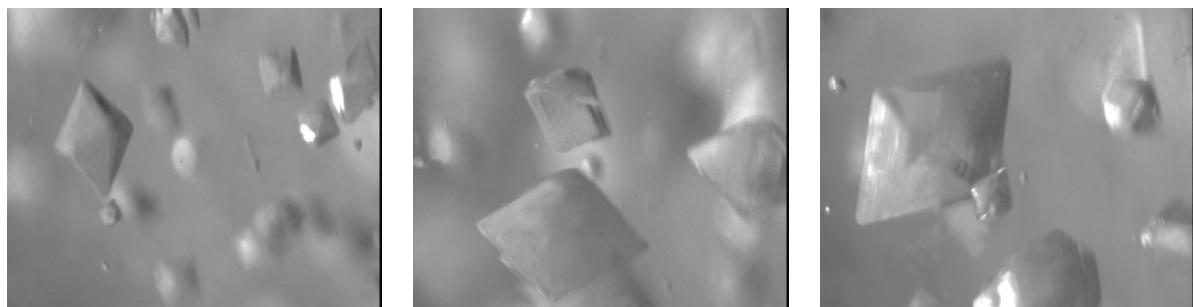


Figure 4.6: Images illustrating morphology of  $\gamma$ -glycine crystallized in water at room temperature.

### 4.3 Artificial image generation

This section describes the methods used to generate artificial images.

#### 4.3.1 Stochastic process model

Consider a slurry  $S$  of volume  $V$  in which a solid phase of discrete particles is dispersed in a continuous fluid phase. Let  $L$  be the characteristic length of a particle and define a shape factor  $k_v$  such that the volume of a single particle is given by  $k_v L^3$ . Let  $f(L)$  denote the PSD, or the number of particles of characteristic length  $L$  per unit volume slurry. Let  $V_I \in S$  denote an imaging volume, and let  $I$  denote an image created by perspective projection of  $V_I$  onto a two-dimensional image plane. Let  $a$  and  $b$  denote the horizontal and vertical dimensions of  $V_I$ , or the field of view, and let  $d_f$  denote the depth dimension of  $V_I$ , or the depth of field. Thus, the volume of  $V_I$  is  $abd_f$ .

To generate an artificial image, we simulate the particle population in a local region surrounding the imaging volume  $V_I$ . In this region, we model the particle population as a three-dimensional stochastic process  $\Phi_p = (\mathbf{X}_{wi}, L_i, \Theta_{zi})$  on  $\mathbb{R}^3 \times \mathbb{R}^+ \times (-\pi/2, \pi/2]$  for  $i = 1, \dots, \tilde{N}_c$ .  $\mathbf{X}_{wi} = (X_{wi}, Y_{wi}, Z_{wi})$  gives the location of the centroid for particle  $i$  in the world coordinate frame,  $L_i$  gives the length,  $\Theta_{zi}$  gives the orientation around the z-axis of the world coordinate frame, and  $\tilde{N}_c$  gives the number of particles.  $\mathbf{X}_{wi}$ ,  $L_i$ ,  $\Theta_{zi}$ , and  $\tilde{N}_c$  are distributed independently of each other.  $X_{wi}$ ,  $Y_{wi}$ ,  $Z_{wi}$ , and  $\Theta_{zi}$  are distributed uniformly on  $[x_{\min}, x_{\max}]$ ,  $[y_{\min}, y_{\max}]$ ,  $[z_{\min}, z_{\max}]$ , and  $(-\pi/2, \pi/2]$ , respectively.  $L_i$  has probability density function  $h$  and corresponding cumulative distribution function  $H$ , given by

$$H(L) = \begin{cases} 0 & L \leq R \\ \int_R^L f(l)dl / \int_R^{L_{\max}} f(l)dl & R < L \leq L_{\max} \\ 1 & L > L_{\max} \end{cases}$$

in which  $R$  is the lower limit of resolution of the camera and  $L_{\max}$  is the size of the largest particle in the population.  $\tilde{N}_c$  has a Poisson distribution with parameter  $\tilde{\lambda} = \lambda(x_{\max} - x_{\min})(y_{\max} - y_{\min})/ab$ ,

in which  $\lambda$  is the expected number of crystals per image, calculated from the PSD using

$$\lambda = V_I \int_R^\infty f(L) dL$$

The size of the local region surrounding the imaging volume is defined by  $(x_{\min}, x_{\max}) = (-0.5L_{\max}, a + 0.5L_{\max})$  and  $(y_{\min}, y_{\max}) = (-0.5L_{\max}, b + 0.5L_{\max})$ , in which  $L_{\max}$  is defined as the size of the largest particle in the population.

### 4.3.2 Imaging model

Each particle is a convex, three-dimensional domain  $P_i \in V$ . To model the imaging process,  $P_i$  is projected onto an imaging plane using a camera model. This projection is computed by first applying rigid-body rotations and translations to change each point  $\mathbf{X}_w$  in  $P_i$  from the world coordinate frame to the camera coordinate frame:

$$\mathbf{X}_c = \mathbf{R}_z \mathbf{R}_y \mathbf{R}_x \mathbf{X}_w + \mathbf{T} \quad (4.1)$$

in which  $\mathbf{R}_z$ ,  $\mathbf{R}_y$ , and  $\mathbf{R}_x$  are rigid-body rotation matrices, which are functions of the in-plane orientation  $\theta_z$  and the orientations in depth  $\theta_y$  and  $\theta_x$ , respectively.  $\mathbf{T} = (t_x, t_y, t_z)$  is a translation vector. Next, each point is projected onto the image plane according to some imaging model. Under perspective projection with a pinhole camera, the transformation from a 3-D point  $\mathbf{X}_c = (X_c, Y_c, Z_c)$  in camera coordinates to an image point  $\mathbf{x}_c = (x_c, y_c)$  is given by

$$x_c = \frac{f_c}{Z_c} X_c, \quad y_c = \frac{f_c}{Z_c} Y_c \quad (4.2)$$

in which  $f_c$  is the focal length of the camera. Figure 4.7 depicts the perspective projection of a cylindrical particle onto the image plane. Finally, to model CCD imaging, the image plane coordinates  $\mathbf{x}_c$  must be converted to pixel coordinates  $\mathbf{w} = (u, v)$  using

$$u = u_0 + k_u x_c, \quad v = v_0 + k_v y_c \quad (4.3)$$

in which  $(u_0, v_0)$  corresponds to  $\mathbf{x}_c = (0, 0)$  and  $k_u$  and  $k_v$  provide the necessary scaling based on pixel size and geometry. The CCD image is depicted in Figure 4.8. For our purposes, the

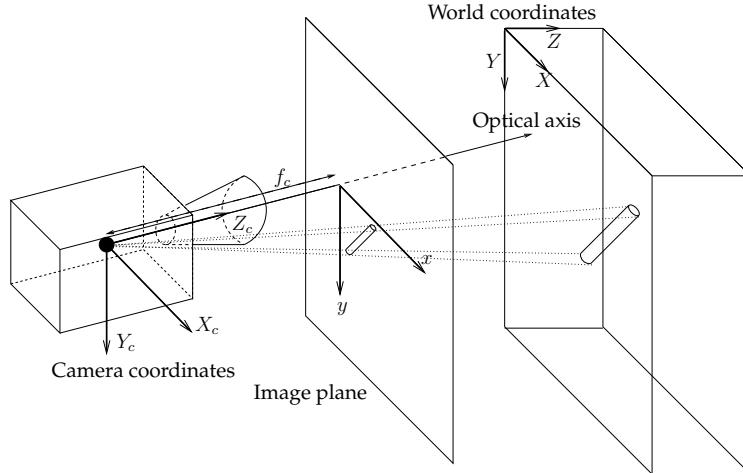


Figure 4.7: Depiction of the perspective projection of a cylindrical particle onto the image plane. For simplicity, the image plane is displayed in front of the camera.

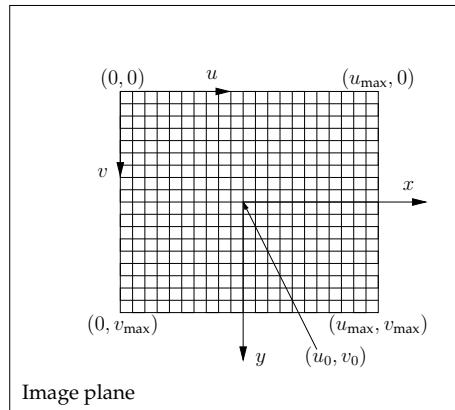


Figure 4.8: Depiction of CCD image.

projection of  $P_i$  onto the CCD array is simplified considerably by assuming the world coordinate frame and camera coordinate frame differ only by a translation in the z-direction. Thus,  $X_c = X_w$  and  $Y_c = Y_w$ . Furthermore, the “weak perspective” projection model can be used because the depth of the imaging volume is small relative to the distance of the imaging volume from the camera. Thus,  $f_c/Z_c$  and  $t_z$  can be assumed constant for all objects. Finally, we can assume that  $(u_0, v_0) = (0, 0)$  and that the pixels are square such that  $k_u = k_v$ . Given these assumptions, the projection of a point  $X_w$  onto the CCD array is given simply by  $(u, v) = (mX_1, mY_1)$ , where

$$m = k_u f_c / Z_c.$$

### 4.3.3 Justifications for two-dimensional system model

The assumptions justifying our use of a two-dimensional process to model a three-dimensional system are as follows. First of all, we assume the camera is positioned a fixed distance  $z_0$  from the imaging volume, and that  $d_f \ll z_0$ . This assumption means the particles in the imaging volume are projected onto the image plane according to the weak perspective projection model. In other words, the projected particle lengths measured in the image coordinate system can be related to the true projected particle lengths by applying a constant magnification factor  $m$ , without regard for the distance of the particle from the camera. Secondly, we assume all particles are oriented in a plane orthogonal to the camera's optical axis. This assumption, together with the weak perspective assumption, essentially reduces the 3-D process to a 2-D process, thereby simplifying the analysis considerably. These assumptions are not used only for convenience, however, but rather to reflect the actual conditions under which in situ imaging measurements are made in practice. To obtain useful in situ images in high solids concentrations, the camera must have a small depth of field and be focused only a small depth into the particulate slurry. It seems reasonable, therefore, to expect the shear flow at the slurry-sensor interface to cause the particles to align orthogonal to the interface, and thus orthogonal to the camera's optical axis.



## Chapter 5

# Two-dimensional Object Recognition for High-Aspect-Ratio Particles <sup>1</sup>

Suspension crystallization processes often result in crystals having a high aspect ratio, a shape commonly described as needle-like, rod-like, or acicular. High-aspect-ratio crystals are particularly commonplace in the specialty chemical and pharmaceutical industries. As discussed in Section 2.3.1, conventional PSD monitoring technologies, such as laser diffraction and laser backscattering, are based on assumptions of particle sphericity [136, 89] and therefore do not provide the monitoring capability necessary to achieve on-line PSD control for systems in which the particles are highly non-spherical. Several researchers have developed imaging-based methods for sizing elongated crystals [100, 110, 94, 95], but none of these methods are sufficiently automated to be suitable for on-line monitoring and control. Commercially available, imaging-based particle size and shape analyzers require sampling, which is inconvenient, possibly hazardous, and raises concerns about whether the sample is representative of the bulk slurry [3, 7]. The utility of in situ video microscopy has been limited primarily to qualitative monitoring because the nature of in situ images, which contain blurred, out-of-focus, and overlapping particles, has precluded successful image segmentation.

This chapter demonstrates robust and efficient segmentation for in situ images of high-aspect-ratio particles using a novel image analysis algorithm. We show that the algorithm's PSD measurements are consistent with measurements obtained through manual image analysis by

---

<sup>1</sup>Portions of this chapter appear in Larsen, Rawlings, and Ferrier [69]

human operators. The accuracy of the measured PSD, therefore, is established only with respect to the PSD measured by human operators. The absolute accuracy of the measured PSD is the subject of Chapters 7– 9. The chapter is organized as follows. Section 5.1 describes the algorithm, and Section 5.2 presents the experimental studies used to evaluate the algorithm’s accuracy and speed. Our findings are summarized in Section 5.3.

## 5.1 Image analysis algorithm description

This section describes the image analysis algorithm developed for the purpose of analyzing *in situ* images of high-aspect-ratio crystals. The algorithm is referred to in this thesis as SHARC (**S**egmentation for **H**igh-**A**spect-**R**atio **C**rystals) and has been implemented in MATLAB 7.1. We present first an overview of SHARC and then describe in more detail each of SHARC’s components.

### 5.1.1 Overview

The SHARC algorithm is built on the assumption that a needle-shaped crystal can be modeled geometrically as a group of two or more spatially-proximate lines with similar orientation and length. The SHARC algorithm searches for image features satisfying this model in the following manner: First, SHARC detects linear features in the image, referred to as “elementary line segments” or ELSs. Next, SHARC identifies collinear line pairs (lines that appear to belong to a single crystal edge but have been broken up due to background noise, particle overlap, or crystal defects) and creates a representative line, called a “base line,” for each pair. Given both the ELSs and the base lines, SHARC identifies pairs of spatially-proximate, parallel lines of similar length. Finally, SHARC identifies consistent groups of parallel lines and clusters the constituent lines in each of these groups as belonging to a single crystal. The properties (e.g. length, aspect ratio) of these line clusters are used as estimates of the properties of the crystals in the image. Figure 5.1 shows the result of applying these steps to a small section of an *in situ* image of needle-like

pharmaceutical crystals.

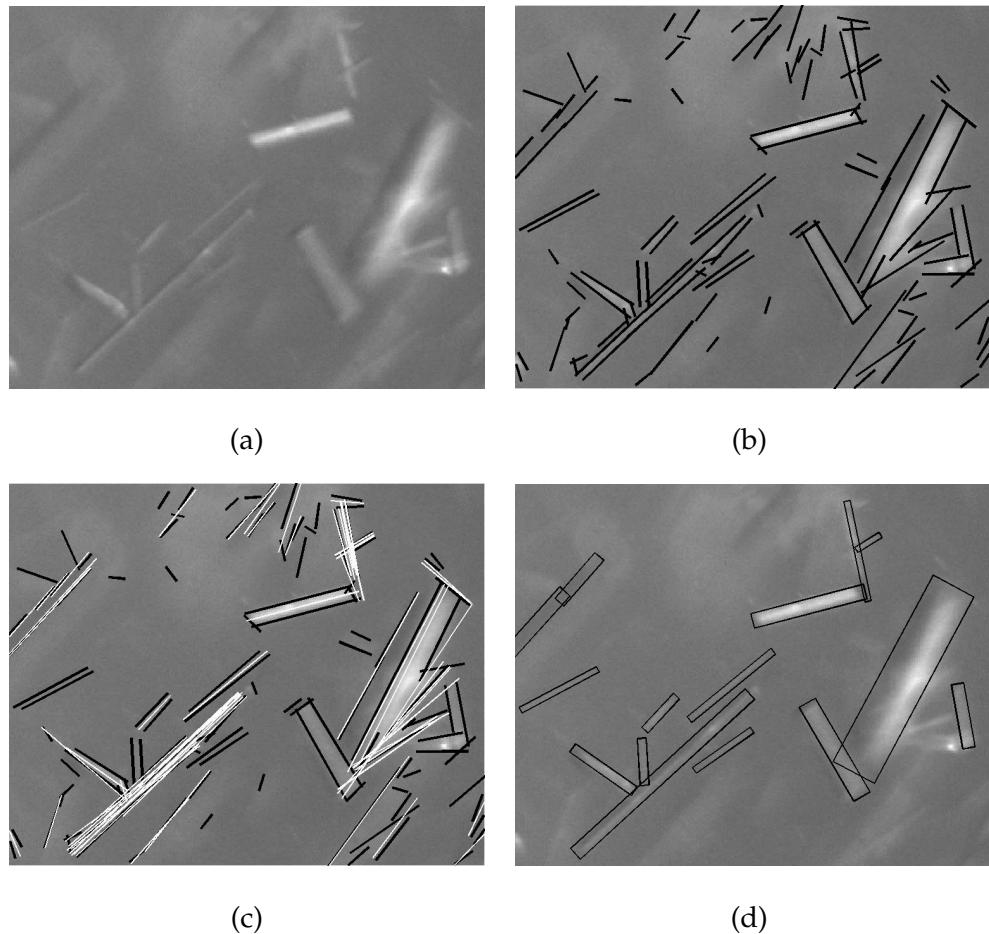


Figure 5.1: Example of SHARC algorithm applied to an in situ image of suspended pharmaceutical crystals. (a) A region of interest in the original image. (b) Linear features (ELSs) extracted from the original image. (c) ELSs (black lines) and lines representing each collinear line pair (white lines). (d) Representative rectangles for clusters of spatially-proximate parallel lines with roughly equal length. The lengths, widths, and aspect ratios of the rectangles are used as the crystal size and shape measurements.

### 5.1.2 Linear feature detection

Line segments are commonly used as inputs to higher-level processes in machine vision, and many different methods have been developed for extracting line segments from images (see [58] for a review of these methods). SHARC uses the Burns line finder [18]. For our application, the Burns line finder is advantageous over the popular Hough transform-based methods for several reasons. First, the Burns line finder is scale-independent, that is, it finds short lines just as easily as it finds long lines. The Burns line finder also has lower computation and memory requirements than the Hough transform and finds line endpoints more easily. The Burns line finder is unique in that it detects lines on the basis of image intensity gradient *direction*, whereas most line-finders are based on image intensity gradient *magnitude*. The Burns line finder is therefore able to detect subtle linear features that would be missed by other line finders. This feature also means that its performance is relatively insensitive to variations in contrast and brightness. This property is important for crystallization imaging because, as crystallization occurs, the increasing solids concentration causes more reflected light to reach the camera CCD, resulting in image intensity variations. Such variations do not affect the performance of the Burns line finder.

We have modified slightly the Burns algorithm to enhance its performance for our particular application, incorporating some of the speed-up suggestions given in [61]. Our implementation consists of the following steps:

1. Calculate the direction and magnitude of the image intensity gradient at each pixel using a Sobel gradient operator of size  $n_\nabla \times n_\nabla$ .
2. For each pixel with gradient magnitude above a small threshold  $\epsilon_{|\nabla|}$ , assign a gradient direction label by coarsely quantizing the pixel's gradient direction into one of  $n_b$  sets of ranges, or "buckets," as depicted in Figure 5.2.
3. Apply a connected components algorithm (CCA) to group identically-labeled, adjacent (including diagonally adjacent) pixels into "line support regions," as depicted in Figure 5.3.

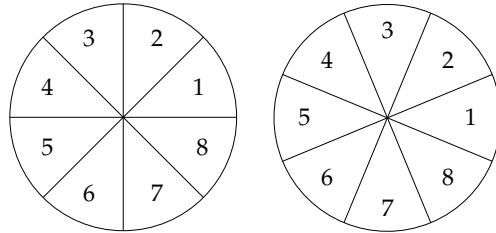


Figure 5.2: Depiction of different eight-bucket gradient direction quantizations used to label pixels. For the quantization on the left, pixels having gradient direction in the range of 0 to 45 degrees are labeled as “1”, pixels with gradient direction in the range of 45 to 90 degrees are labeled as “2”, and so forth. Quantization effects are mitigated by applying a second quantization, such as that shown on the right, and subsequently resolving any conflicts between the results given by each quantization.

4. Filter line support regions that have a pixel area less than some pre-defined threshold  $\epsilon_A$ .
5. Fit a line to each remaining line support region, as depicted in Figure 5.3.

To eliminate quantization effects, steps 2–4 are performed twice before proceeding to step 5, each time using a different quantization in which the gradient direction partitioning is shifted by half the bucket size, as shown in Figure 5.2. This procedure results in each pixel being associated with two different line support regions, and this conflict is resolved through a voting process designed to select the interpretation that results in the longest possible line support regions [18]. To reduce computation time, SHARC carries out this voting process on the basis of the pixel areas of the conflicting line support regions, which in almost every case gives the same results as voting based on length.

The line-fitting method used in step 5 is standard blob analysis, available in most image analysis packages. Blob analysis fits a line to each region of pixels by determining an ellipse having the same geometric moments as the region, as depicted in Figure 5.3. The ellipse’s major axis length, minor axis length, orientation, and centroid are used respectively as the length, width, orientation, and center of the corresponding line.

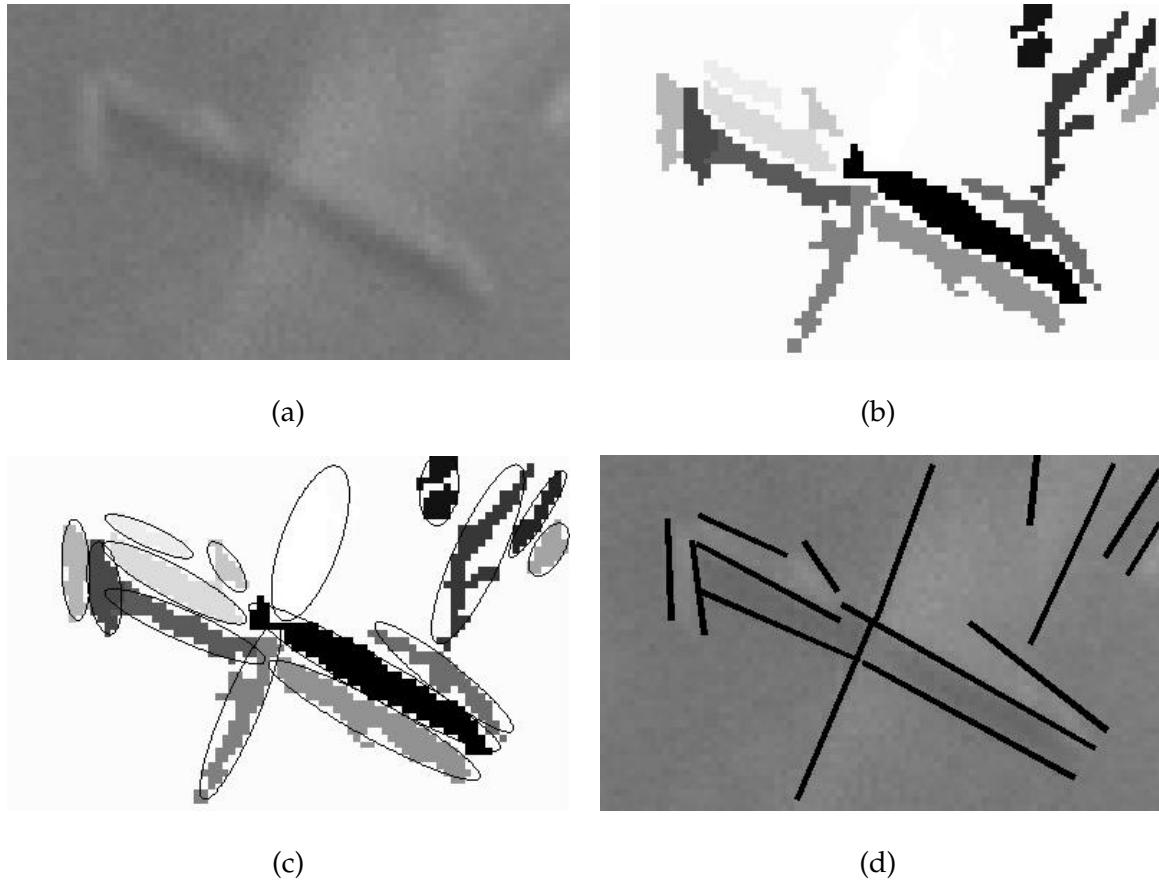


Figure 5.3: Example of finding linear features using Burns line finder and blob analysis. (a) Grayscale image. (b) Regions of pixels having similar gradient orientation, determined using the Burns line finder. (c) Best-fit ellipses for each region of pixels, determined using blob analysis. (d) Major axes of the best-fit ellipses imposed on the original grayscale image.

### 5.1.3 Identification of collinear line pairs

During linear feature extraction, single edges are commonly broken up into multiple collinear lines. This problem is common for systems of needle-like crystals because the particles are often touching or overlapping. Therefore, SHARC groups these collinear lines prior to searching for groups of parallel lines having similar length and orientation. The problem of collinear line grouping has been studied extensively. Jang and Hong [58] compare and evaluate a number of the available methods. SHARC uses a straightforward, computationally inexpensive method de-

veloped by [33]. Etemadi's method involves projecting ELSs of similar orientation onto a common line to determine if the lines satisfy simple spatial-proximity thresholds given by

$$|\theta_1 - \theta_2| < \epsilon_{\theta_C}, \quad d_{PD} < \epsilon_{PD}(w_1 + w_2), \quad d_{EP} < \epsilon_{EP}(\tilde{L}_1^P + \tilde{L}_2^P) \quad (5.1)$$

in which  $d_{PD}$  is the perpendicular distance between the two lines and  $d_{EP}$  is the distance between their nearest projected endpoints.  $\theta_i$  and  $w_i$  are, respectively, the orientation and width of line  $i$  calculated using blob analysis, and  $\tilde{L}_i^P$  is the projected length of line  $i$ , calculated as described below.  $\epsilon_{\theta_C}$ ,  $\epsilon_{PD}$ , and  $\epsilon_{EP}$  are user-specified thresholds.

The perpendicular and endpoint distances and projected lengths are calculated, as depicted in Figure 5.4, by projecting the two lines onto a "virtual line" whose position  $(x_V, y_V)$  and orien-

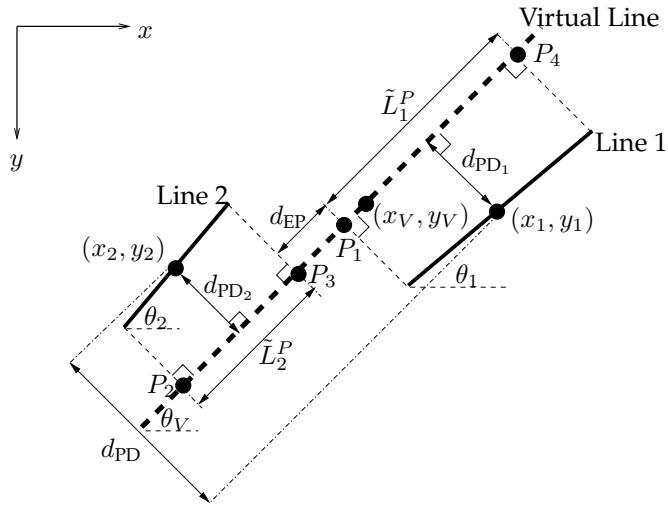


Figure 5.4: Depiction of variables used for line pair classification scheme.

tation ( $\theta_V$ ) are length-weighted averages of the positions and orientations of the constituent ELSs, given by

$$x_V = \frac{\tilde{L}_1 x_1 + \tilde{L}_2 x_2}{\tilde{L}_1 + \tilde{L}_2}, \quad y_V = \frac{\tilde{L}_1 y_1 + \tilde{L}_2 y_2}{\tilde{L}_1 + \tilde{L}_2}, \quad \theta_V = \frac{\tilde{L}_1 \theta_1 + \tilde{L}_2 \theta_2}{\tilde{L}_1 + \tilde{L}_2} \quad (5.2)$$

in which  $\tilde{L}_i$ ,  $\theta_i$ ,  $x_i$ , and  $y_i$  are the length, orientation, horizontal centroid, and vertical centroid of ELS  $i$ , calculated using blob analysis. Given the position and orientation of the virtual line, the perpendicular distance between the two ELSs can be calculated as the sum of the perpendicular

distances of the ELSs centroids from the virtual line. The length of the virtual line,  $L_V$ , is defined as the length of the shortest possible line containing all four projected endpoints.

For each line pair satisfying the collinearity criteria, the corresponding virtual line becomes a base line and is subsequently used in the identification of parallel pairs, as described in the following section.

#### 5.1.4 Identification of parallel line pairs

Following collinear line pair identification, SHARC identifies pairs of parallel lines, or lines that have similar orientation, are spatially proximate, and exhibit a high degree of overlap when projected onto a common line. These line pairs satisfy the following criteria:

$$|\theta_1 - \theta_2| < \epsilon_{\theta_P}, \quad d_{PD} < \frac{1}{\epsilon_{AR}} \tilde{L}_{max}, \quad Q_P > \epsilon_Q \quad (5.3)$$

in which  $\epsilon_{\theta_P}$ ,  $\epsilon_{AR}$ , and  $\epsilon_Q$  are user-specified thresholds for orientation difference, aspect ratio, and pair “quality.”  $d_{PD}$  is the perpendicular distance between the two lines, and  $L_V$  is the length of the virtual line, as defined in the previous section.  $\tilde{L}_{max}$  is the length of the longest line in the pair, and  $Q_P$  quantifies the “quality” of the pair. The quality metric used by SHARC, and suggested in [33], is based on the degree of overlap of the two parallel lines, calculated using

$$Q_P = \frac{\tilde{L}_1^P + \tilde{L}_2^P}{2L_V} \quad (5.4)$$

in which the projected lengths  $\tilde{L}_i^P$  are computed as described in Section 5.1.3. This metric is simple to compute and scale-independent, depending only on the relative lengths of the lines. Overlapping parallel pairs give a  $Q_P$  between 0.5 and 1, the latter value representing a perfectly overlapping pair.

If the parallel pair includes a base line comprising two collinear ELSs, it is possible that the two lines in the parallel pair share an ELS, in which case the pair is invalid and is discarded. Figure 5.5 depicts examples of valid and invalid parallel pairs.

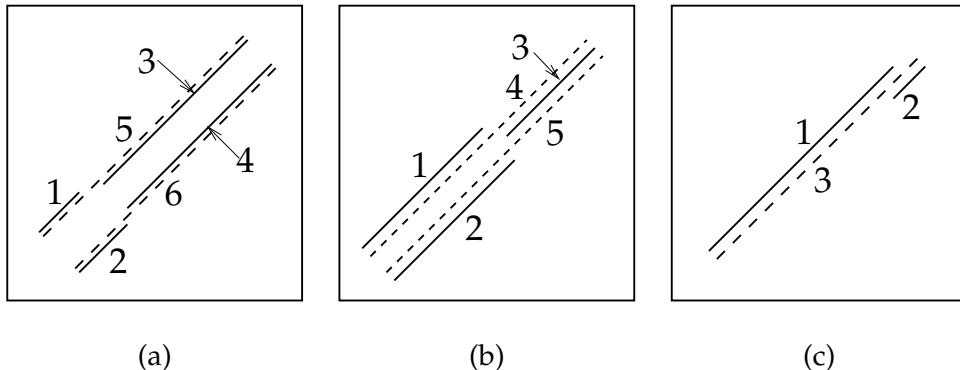


Figure 5.5: Depiction of valid and invalid parallel line pairs. The solid lines represent ELSs and the dotted lines represent base lines (lines arising from instances of collinearity). In (a), the base lines 5 and 6 form a valid parallel pair, and the ELSs 3 and 4 also form a valid parallel pair. In (b), the parallel lines 4 and 5 are an invalid parallel pair because they both depend upon ELS 3. Similarly, in (c), base line 3 and ELS 1 form an invalid pair because both depend on ELS 1.

Each parallel pair is ranked according to its significance, calculated as

$$\tilde{S} = \frac{\tilde{L}_{min}^2}{\tilde{L}_{max}} \quad (5.5)$$

in which  $\tilde{S}$  is the significance level and  $\tilde{L}_{min}$  and  $\tilde{L}_{max}$  are, respectively, the lengths of the shorter and longer lines in the pair. This significance measure is used to account for the fact that longer lines are less likely to have arisen by accident or due to noise and should thus be considered more significant. The significance ranking is used to order the subsequent line clustering procedure but affects the results only when there is a conflict between two high quality pairs, or when two high quality pairs are mutually exclusive. These conflicts arise since SHARC identifies parallel pairs using both the ELSs and the base lines (for example, ELSs 3 and 4 in Figure 5.5(a) form a valid parallel pair but are also involved indirectly in the parallel pair of base lines 5 and 6). If a given ELS is involved in two conflicting parallel pairs, the significance ranking is used in the line clustering process to favor the interpretation that leads to the longer crystal.

### 5.1.5 Clustering

The objective of clustering is to group those lines that appear to belong to a single crystal. The clusters are formed by (1) identifying the most significant parallel pair on the basis of Equation (5.5), (2) recursively identifying all other lines that are parallel-paired with at least one of the lines in the current cluster, and (3) removing from the list of parallel pairs any pairs that include an ELS or base line associated with the newly-formed group. This process is iterated until the list of parallel pairs is empty, after which all lines that are not included in any of the formed groups are discarded.

The properties of each line cluster are calculated using the method described in Section 5.1.3, generalized to an arbitrary number of lines. That is, the cluster orientation is the length-weighted average of all lines in the cluster, the cluster length is the length of the shortest possible line containing all projected endpoints for all lines in the cluster, and the cluster width is the largest possible perpendicular distance between the centroids of all lines in the cluster. Clusters having an aspect ratio below the user-defined threshold  $\epsilon_{AR}$  are discarded.

Figure 5.6 illustrates the clustering procedure using the set of lines extracted in Figure 5.3. Figure 5.6(a) shows all lines involved in at least one valid parallel pair, including both ELSs and base lines. The valid parallel pairs for this example are (10,14), (12,15), (18,19), (18,20), and (19,21). Lines 18, 19, 20, and 21 are base lines comprising the collinear pairs (9,15),(10,12),(12,14), and (14,15), respectively. Figure 5.6(b) shows that pair (18,20) has the highest significance and is therefore analyzed first in the clustering order. Figure 5.6(c) shows the result of recursively identifying all lines that are paired with lines in the cluster. That is, line 19 is identified in the first recursion due to its pairing with line 18, and line 21 is identified in the second recursion due to its pairing with line 19. The rectangle in Figure 5.6(d) indicates the length, width, and orientation of the line cluster. The parallel pairs (10,14) and (12,15), each of which has at least one of its members involved in the newly-formed grouping, are removed from the list of parallel pairs.

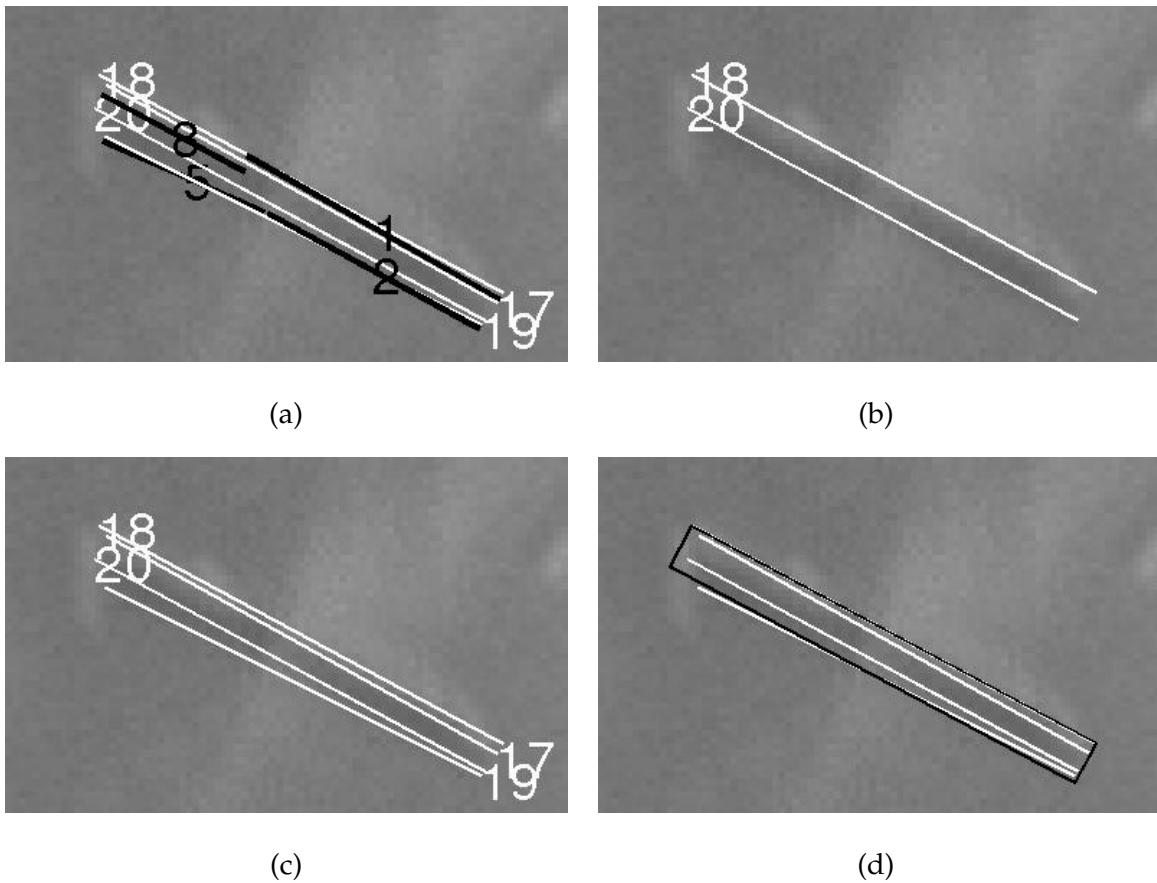


Figure 5.6: Example of clustering procedure for valid parallel pairs. (a) ELSs (dark) and base lines (light) involved in at least one valid parallel pair. (b) The pair with the highest significance. (c) Lines that are parallel-paired (either directly or indirectly) with either of the lines in the highest significance pair. (d) The bounding box calculated for the line cluster.

## 5.2 Experimental results

To evaluate SHARC's performance, a seeded, pharmaceutical crystallization was carried out during which several sets of video images were acquired. The images in each set were acquired over a few seconds only, such that we assume the properties of the crystal population are constant for each image set. The temperature profile (following seed injection) for this crystallization is shown in Figure 5.7. The time at which each set of video images was acquired is indicated in Figure 5.7 by a vertical line and labeled as  $t_j$ , the subscript indicating the image set number. The mixing speed

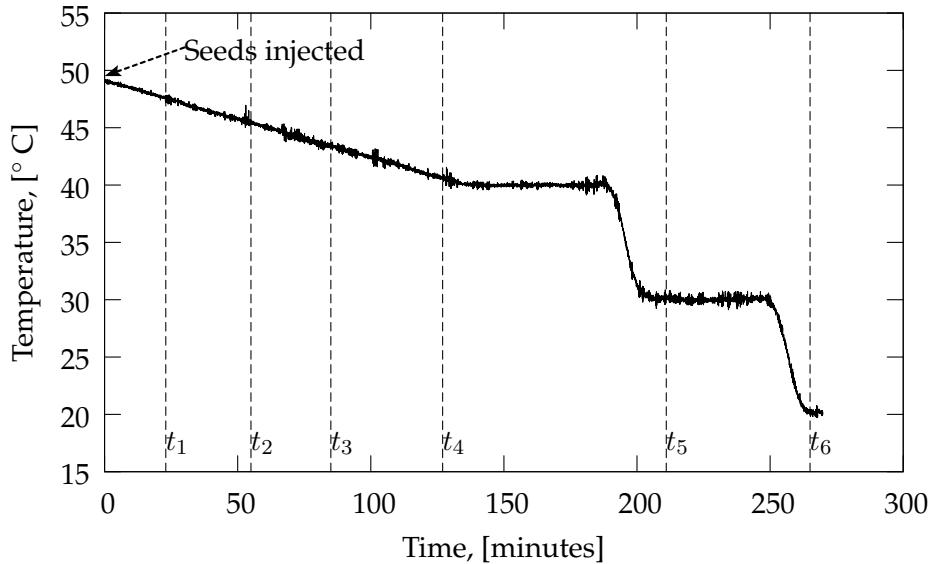


Figure 5.7: Temperature trajectory for crystallization experiment. The vertical lines indicate the times at which sets of video images were acquired.

Line finder		Collinearity		Parallelism	
Parameters		Thresholds		Thresholds	
$n_\nabla$	5	$\epsilon_{\theta_C}$	20 degrees	$\epsilon_{\theta_P}$	5 degrees
$\epsilon_{ \nabla }$	1	$\epsilon_{EP}$	0.5	$\epsilon_Q$	0.85
$n_b$	6 buckets	$\epsilon_{PD}$	0.5	$\epsilon_{AR}$	4.5
$\epsilon_A$	20 pixels				

Table 5.1: SHARC parameter values used to analyze images from pharmaceutical crystallization experiment.

for this experiment was 550 RPM (1.1 m/s tip speed), sufficiently fast that the crystals could not be tracked from one video frame to the next. The same parameter values were used to analyze all images (see Table 5.2).

The following sections assess SHARC's suitability for on-line monitoring and control with respect to both accuracy and speed.

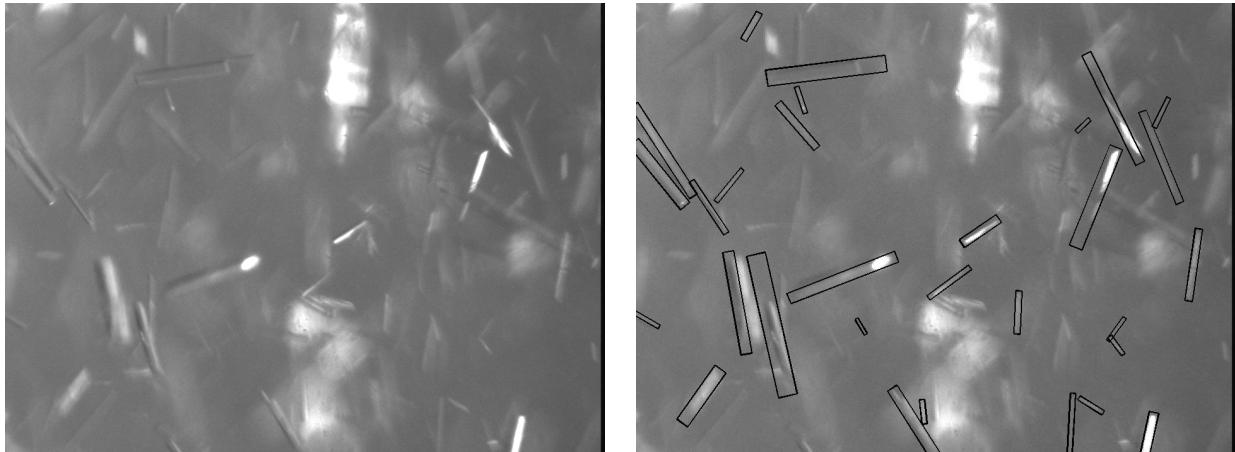


Figure 5.8: Algorithm performance on example image (set 3, frame 1).

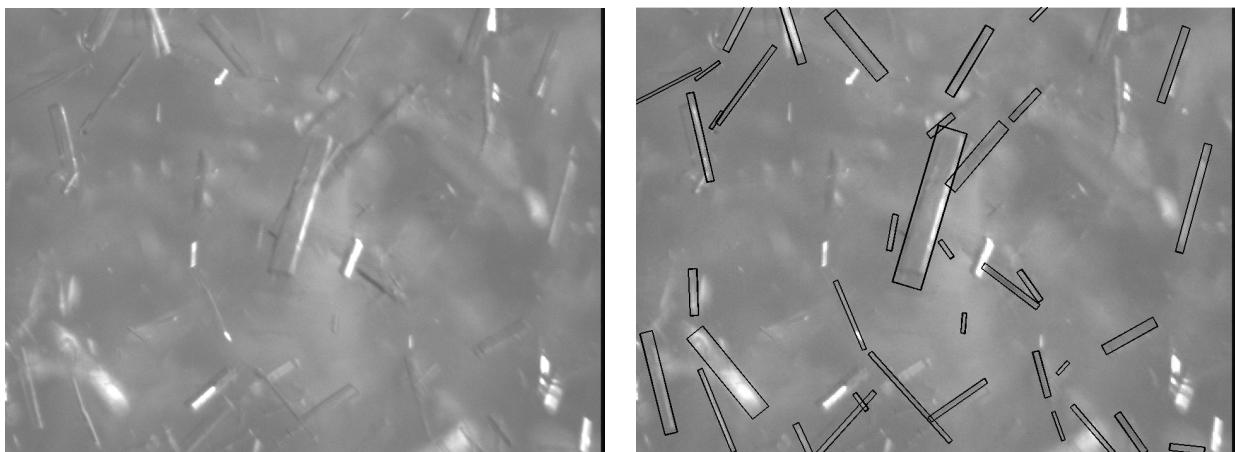


Figure 5.9: Algorithm performance on example image (set 4, frame 0).

### 5.2.1 Algorithm accuracy

#### Visual evaluation

Figures 5.8–5.11 show SHARC’s performance on selected images taken from video sets 3, 4, 5, and 6. These figures demonstrate SHARC’s effectiveness for images having poor contrast and uneven background. These figures also demonstrate SHARC’s ability to detect crystals with poorly-defined edges, varying intensity levels, and a certain level of particle overlap.

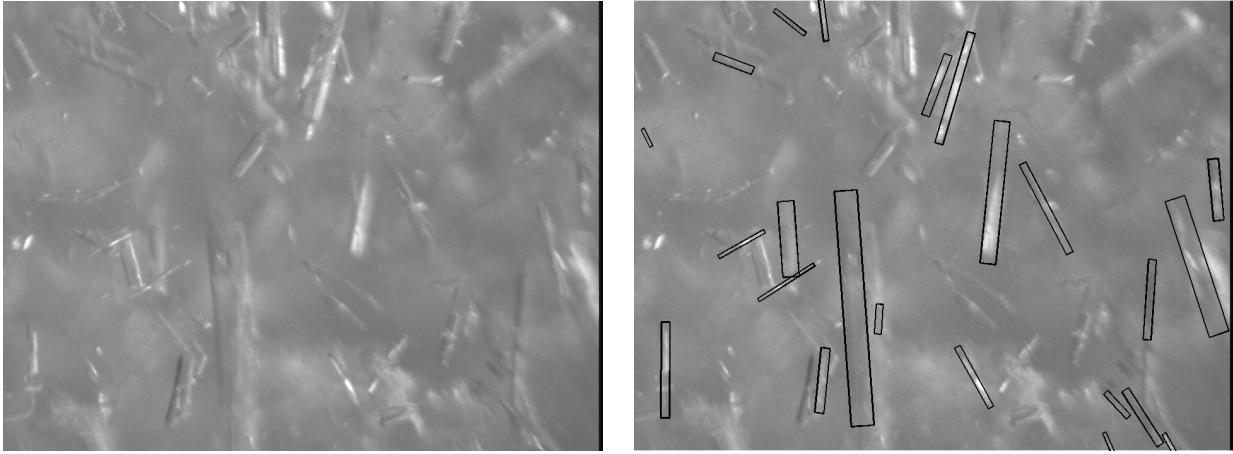


Figure 5.10: Algorithm performance on example image (set 5, frame 5).

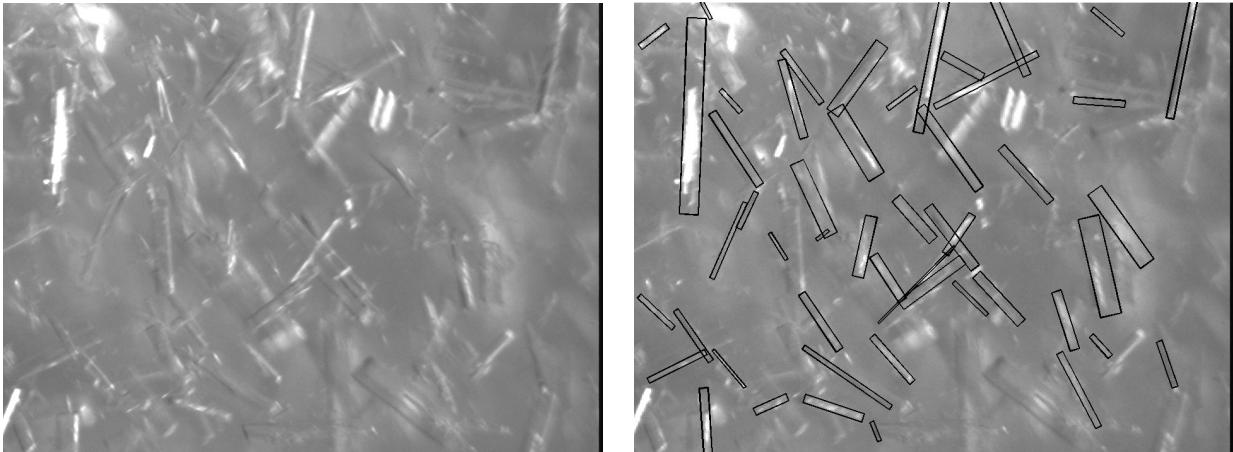


Figure 5.11: Algorithm performance on example image (set 6, frame 3).

### Comparisons with manual sizing

We evaluate SHARC's accuracy by comparing its PSD measurements with measurements obtained through manual image analysis by human operators. Although the human vision system is clearly more reliable than current computer vision systems, manual image analysis introduces an undesirable level of subjectiveness into the measurement. The subjectiveness of manual sizing is magnified for *in situ* images because the crystals often appear blurry, overlapping, and out-of-focus. Thus, it can be difficult to decide whether or not a given crystal is sufficiently in focus and well-defined to be sized accurately. To assess the subjectiveness involved in manually sizing in

Operator	1	2	3	4	5	6	7	8	9	Mean	SD
Mean size [ $\mu\text{m}$ ]	224	300	314	295	276	268	328	276	272	284	30
# of crystals	460	230	140	350	150	300	200	330	500	290	130
ave. % diff (Eq. 5.6)	26	10	13	10	9	10	17	9	9	12	

Table 5.2: Comparison of results obtained from nine different persons manually sizing the same ten images.

situ images and determine what constitutes “good” agreement between SHARC’s measurement and a manual measurement, we asked nine different people to manually size crystals for the same ten images from image set 3. We confirmed that ten images was sufficient to achieve convergence of the overall mean size measurement for each operator. Table 5.2 shows the overall mean size calculated for all ten images for each operator, the total number of crystals found by each operator, and the average percent difference in overall mean size between each operator and the other operators. This latter value is calculated for operator  $i$  using the equation

$$\text{diff}_i = \frac{100}{N_{\text{op}} - 1} \sum_{j=1}^{N_{\text{op}}} \frac{|\bar{x}_i - \bar{x}_j|}{\frac{\bar{x}_i + \bar{x}_j}{2}} \quad (5.6)$$

Table 5.2 shows that the mean size varied by as much as 37% between operators, and the number of crystals found varied by over 100% between operators, illustrating the large degree of subjectiveness associated with manual sizing of in situ images. However, the relatively small standard deviation in overall mean size indicates that manual sizing constitutes a reasonably reliable standard provided the measurement is performed by a sufficient number of operators.

Based on the results shown in Table 5.2, we define “good” agreement between SHARC’s results and a set of manual results to mean that their individual means are within approximately 12% of their combined mean value. Table 5.3 shows that the SHARC algorithm determines a mean crystal size within 2% of that found by the nine manual operators, and Figure 5.12 shows a good match between the cumulative distribution function found by SHARC and the cumulative distribution functions of the nine manual operators.

Image set	Mean Size, [ $\mu\text{m}$ ]		
	Manual	Automatic	% Difference
3	283	284	0.3
4	290	286	1.2
5	306	278	9.3
6	355	283	22.5

Table 5.3: Comparison of mean sizes obtained from manual sizing of crystals by a human operator and from automatic sizing by SHARC.

To evaluate SHARC's ability to maintain accuracy for the duration of the experiment, twenty-five images from sets 4, 5, and 6 were analyzed both manually (by a single operator) and using SHARC. Table 5.3 shows that SHARC maintains good agreement with the results obtained manually for sets 4 and 5, but its performance declines somewhat for set 6. These same conclusions can be drawn from Figure 5.12, which compares the cumulative distribution functions obtained using both methods.

Figure 5.12 and Table 5.3 indicate that, as the solids concentration and degree of crystal attrition increase, SHARC either fails to identify a significant percentage of the larger crystals or erroneously identifies smaller crystals. Figure 5.13 shows the results of manually sizing an image from set 6 compared with the results given by SHARC. SHARC identifies several smaller crystals not identified manually as well as only portions of some of the bigger crystals. These misidentifications explain why SHARC's mean size is less than the mean size obtained manually.

Some of the differences between SHARC's results and the manual results can be attributed to the subjectiveness associated with manual sizing. However, crystals 2, 11, and 15 are clearly misidentifications. Figure 5.14 shows a zoomed-in view of these crystals and demonstrates the results of each step in SHARC. For these cases, particle overlap and attrition interferes with the detection of the crystal edges to such an extent that SHARC is unable to detect important instances of collinearity. These particular misidentifications can be corrected by relaxing SHARC's collinear-

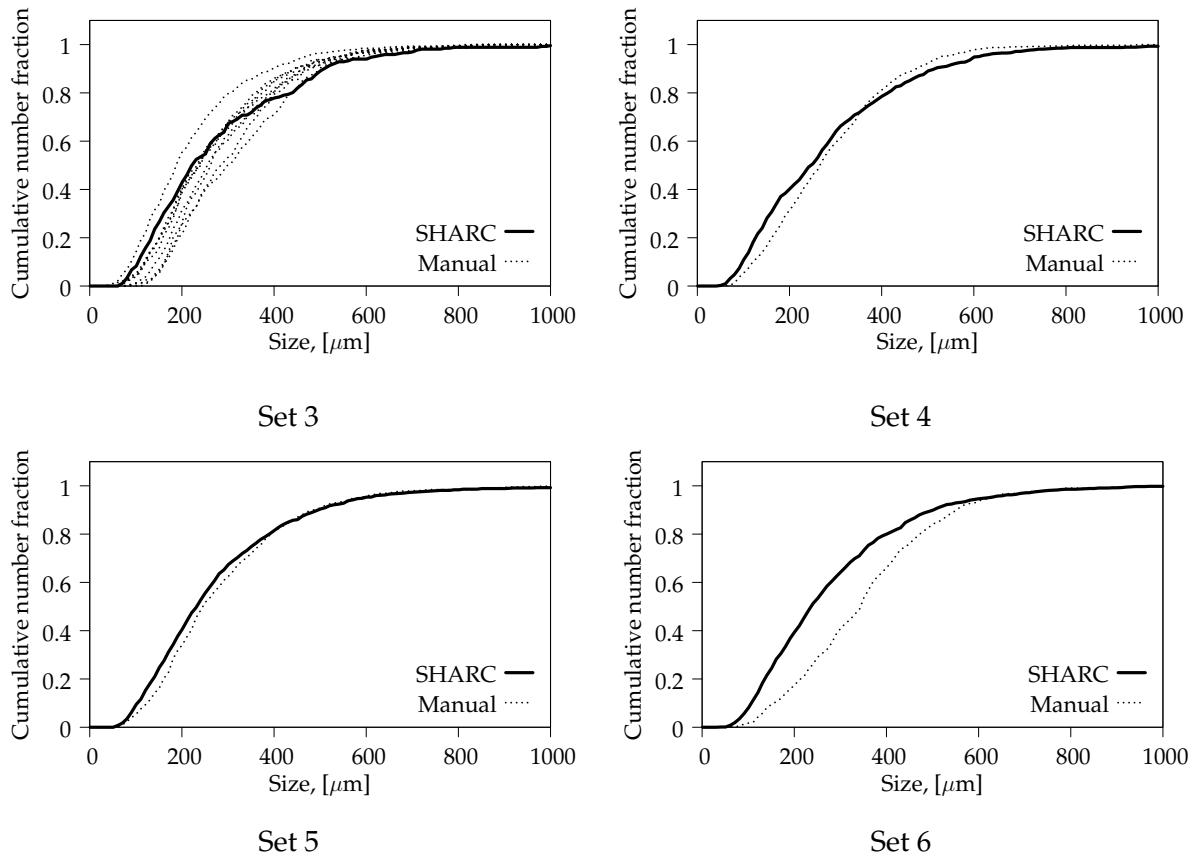


Figure 5.12: Comparison of cumulative number fractions obtained from manual and automatic sizing of crystals for video image sets 3, 4, 5, and 6. Set 3 was manually sized by nine different operators.

ity criteria, but this would likely lead to further false positives. Investigating the performance of the many available collinear identification methods noted in [58] may be the best way to improve algorithm performance for high solids concentrations.

## 5.2.2 Algorithm speed

This section assesses whether SHARC is sufficiently fast to be useful for on-line monitoring and control of crystallization processes. Table 5.4 shows the amount of time required to process an image for each of the image sets and shows how this time is partitioned amongst the different steps of SHARC. Table 5.4 indicates that a significant part of the processing time for most sets is spent

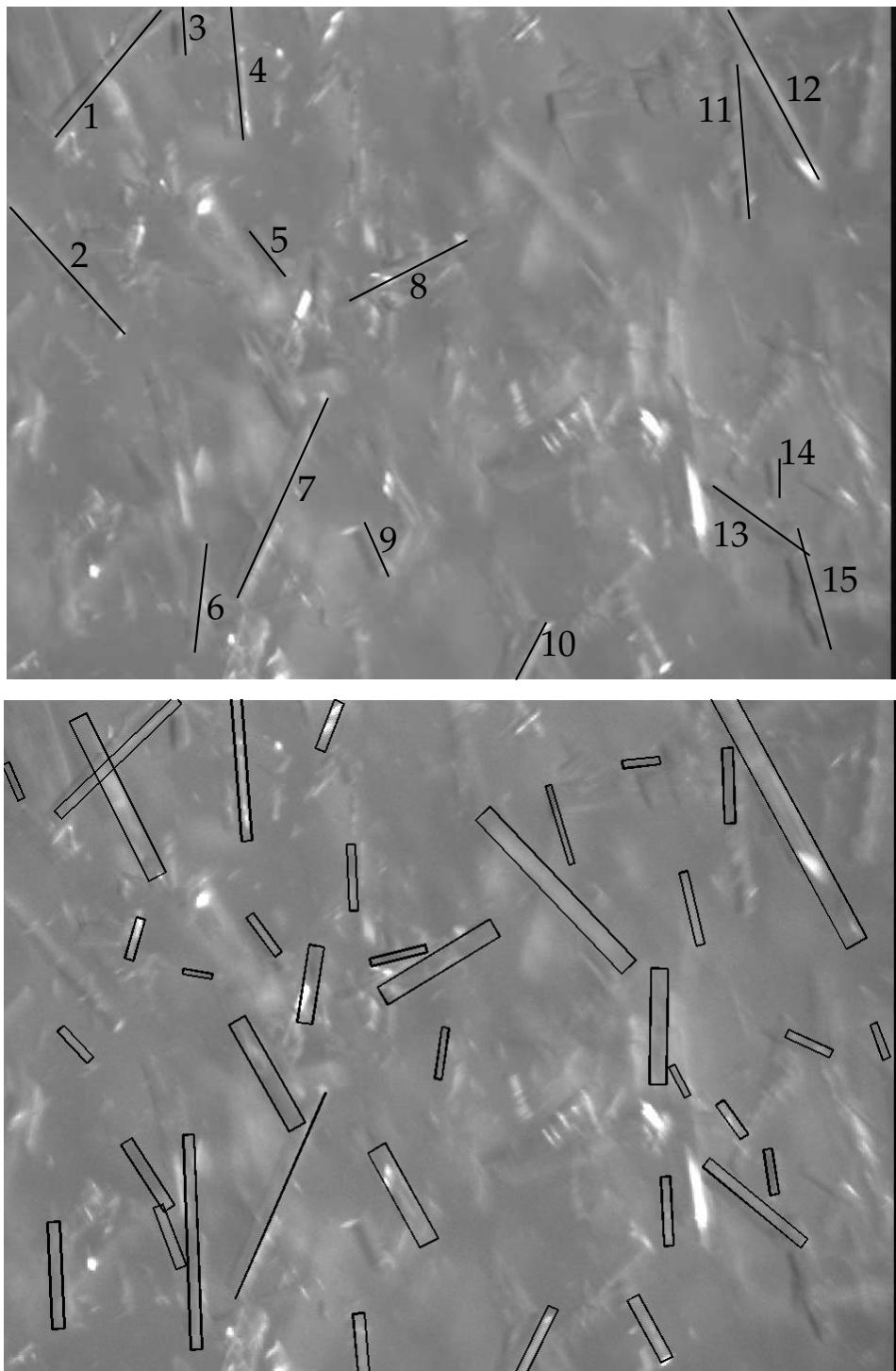


Figure 5.13: Comparison of crystals sized manually (top) and using SHARC (bottom).

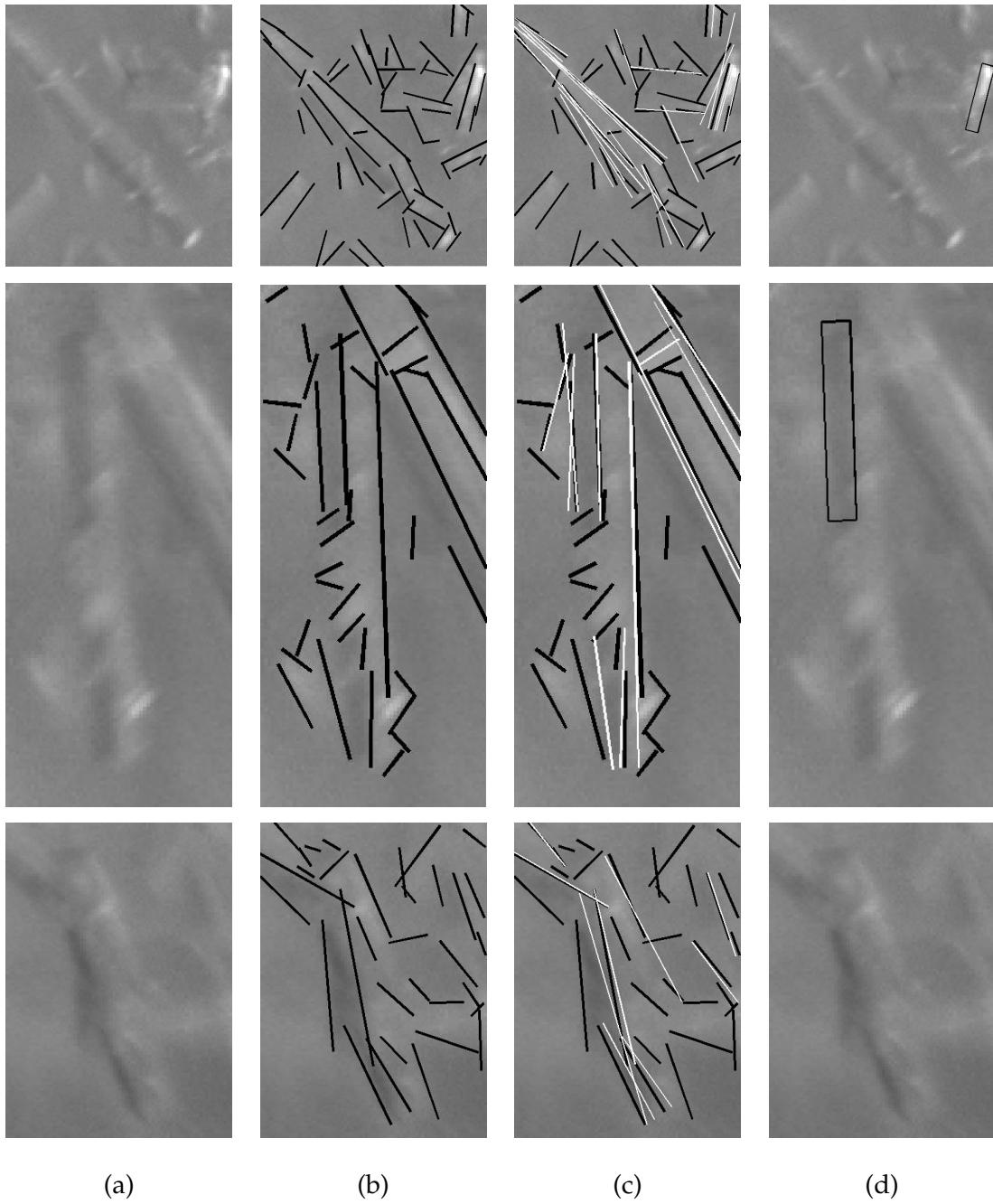


Figure 5.14: Zoomed-in view of crystals that SHARC failed to identify correctly. From top to bottom, the crystal labels are two, eleven, and fifteen (according to labels in Figure 5.13). Column (a): Original image. Column (b): ELS data. Column (c): ELSs and base lines. Column (d): Result of clustering.

Set	average total cputime per image [s]	average cputime per image [s]			
		linear feature detection	collinearity identification	parallelism identification	clustering
2	1.9	1.8 (97)	0.0 (1)	0.0 (0)	0.0 (0)
3	3.1	2.0 (65)	0.4 (14)	0.6 (19)	0.0 (0)
4	4.3	2.1 (47)	0.8 (17)	1.5 (33)	0.0 (0)
5	7.1	2.5 (35)	1.4 (19)	3.2 (44)	0.0 (0)
6	10.8	2.9 (26)	2.2 (20)	5.6 (52)	0.1 (0)

Table 5.4: Computational requirements for analyzing different image sets (averaged over 10 images). The numbers in parentheses give the percentages of total cputime. Images are analyzed using a 2.2 GHz AMD Athlon 64 processor.

on the collinear and parallel grouping operations. These operations can be made more efficient by limiting the computations to line pairs that are spatially proximate using a data structure in which the lines are sorted by their endpoint locations, as suggested in [75].

To determine how much time is required to obtain a sufficiently accurate estimate of the mean particle length, we find the number of samples  $n$  such that the size of the 95% confidence interval for the population mean particle length is less than 10% of the sample mean. That is, we find the smallest  $n$  for which

$$t(\alpha, n - 1) \frac{s_n}{\sqrt{n}} \leq 0.1 \bar{l}_n \quad (5.7)$$

in which  $t$  is the t-distribution,  $\alpha$  is the confidence level,  $\bar{l}_n$  is the sample mean, and  $s_n$  is the sample standard deviation, defined as

$$\bar{l}_n = \frac{1}{n} \sum_{i=1}^n l_i \quad (5.8)$$

$$s_n = \sqrt{\frac{1}{n-1} \sum_{i=1}^n (l_i - \bar{l}_n)^2} \quad (5.9)$$

in which  $l_i$  is the length of particle  $i$ . Similarly, to determine how much time is required to obtain

Set	average # of crystals	# of crystals to converge	cputime to converge to	cputime to converge to
	per image	to mean	mean [min.]	variance [min.]
2	1	93	1.9	18.5
3	25	163	0.3	1.8
4	28	129	0.3	2.2
5	30	106	0.4	3.4
6	38	143	0.7	4.2

Table 5.5: Computational requirements for SHARC to achieve convergence of particle size distribution mean and variance.

a sufficiently accurate estimate of the variance in particle length, we find  $n$  such that the size of the 95% confidence interval for the population variance is less than 10% of the sample variance  $s_n^2$ , satisfying

$$\frac{n - 1}{\chi^2(\alpha, n - 1)} - 1 \leq 0.1 \quad (5.10)$$

in which  $\chi^2$  is the chi-squared distribution (see [115, p.75]). Equation 5.10 is satisfied for  $n = 889$  samples.

Given SHARC's speed, the number of crystals per image, and the number of samples necessary to obtain sufficient measurement accuracy, we can calculate the rate at which SHARC provides accurate mean and variance measurements. Table 5.5 indicates that SHARC requires approximately two minutes to measure the PSD mean and fifteen minutes to measure the PSD variance. Given the time scales of most crystallization processes, SHARC is sufficiently fast to provide measurements for a feedback control system based on measurements of the PSD mean and variance. If necessary, the measurement rate can be increased by implementing SHARC in a compiled language.

### 5.3 Conclusion

The SHARC algorithm can robustly and efficiently extract crystal size information from in situ images of suspended, high-aspect-ratio crystals for moderate solids concentrations, giving results consistent with measurements obtained through manual image analysis by human operators. SHARC's performance declines for high solids concentrations and high levels of particle attrition because the degree of particle overlap and the noise arising from attrited particulate matter hinder the identification of the suspended crystals' edges. Implementing improved methods for identifying instances of collinearity may enable suitable performance for these conditions.

The speed with which SHARC analyzes the images is suitable for real-time monitoring and control of PSD mean and variance.

## Chapter 6

# Three-dimensional Object Recognition for Complex Crystal Shapes <sup>1</sup>

The SHARC algorithm described in Chapter 5 handles only one of the many possible shapes that can result from suspension crystallization processes. In this chapter, an image analysis algorithm called M-SHARC (Model-based SHApe Recognition for Crystals) is developed that can be applied to images of crystals of any shape, provided the shape can be represented as a wire-frame model. The wire-frame models used by the algorithm are parameterized. Thus, a single model can be used to identify crystal objects exhibiting a wide range of sizes and shapes within a given shape class. The algorithm therefore enables the measurement of shape factor distributions. Furthermore, the algorithm can be applied using multiple wire-frame models representing different shape classes to measure the distribution of particles between different shape classes.

The algorithm described in this chapter is classified as a model-based object recognition algorithm. Model-based object recognition is a widely-used approach to computer vision that has been developed to enable automatic recognition of complex objects with unknown pose (i.e. orientation with respect to the camera) in the presence of missing or occluded data [54, 75, 39, chapter 18]. The model-based object recognition approach is based on matching raw image features (such as arcs or lines) with one of several pre-defined models. The model-based approach does not involve exhaustive searches of the model parameter space and is therefore more efficient than purely top-down approaches (such as Hough-transform-based methods). Model-based object recogni-

---

<sup>1</sup>Portions of this chapter appear in Larsen, Rawlings, and Ferrier [70]

tion is more robust to noise than purely bottom-up approaches because it can be applied even if part of the object to be identified is occluded or missing. Furthermore, the model-based approach leads to algorithms that can be implemented in a parallel fashion to enable real-time analysis. Algorithms based on this approach have been developed and applied to systems of circular particles [113] and elliptical particles [48]. The SHARC algorithm is essentially a two-dimensional model-based recognition algorithm.

The chapter is organized as follows. Section 6.1 describes the algorithm, and Section 6.2 discusses the algorithm's accuracy and speed by comparing the algorithm results with those obtained by manual, human analysis of *in situ* video images acquired at different solids concentrations during an  $\alpha$ -glycine cooling crystallization experiment.

## 6.1 Model-based recognition algorithm

This section describes the model-based recognition algorithm designed to extract crystal size and shape information from *in situ* crystallization images. The algorithm is called M-SHARC (**M**odel-based **SH**Ape **R**ecognition for **C**rystals) and has been implemented in MATLAB 7.0.

### 6.1.1 Preliminaries

The model-based object recognition framework involves matching a set of primitive features extracted from an image (such as points, corners, or lines) to a pre-defined set of models. The primitive image features used by M-SHARC are lines, and the models are parameterized, wire-frame models. Wire-frame models consist of a set of  $q$  vertices  $V = \{\mathbf{X}_K[\mathbf{p}_m]\}_{K=1\dots q}$  and a set of  $r$  lines or edges  $E = \{E_J\}_{J=1\dots r}$ .  $\mathbf{X}_K$  is a three-dimensional vector defined in a model-centered coordinate system as a function of the model internal parameters  $\mathbf{p}_m$ .  $E_J$  is a set of two labels pointing to the vertices in  $V$  that are connected by edge  $J$ . The model used in this study was designed to capture the range of shapes exhibited by crystals of glycine, an amino acid of importance in the pharmaceutical industry. The model, shown in Figure 6.1, has three internal parameters

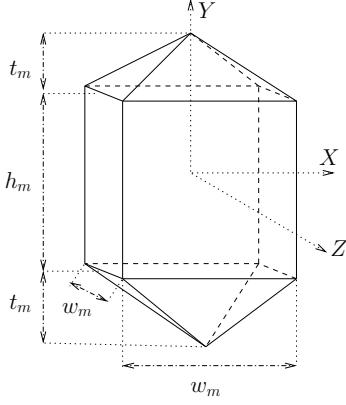


Figure 6.1: Wire-frame glycine crystal model. The parameters for the model are the crystal body height,  $h_m$ , the width,  $w_m$ , and the pyramid height,  $t_m$ .

$(\mathbf{p}_m = (h_m, w_m, t_m))$ , 20 edges, and 10 vertices.

To fit the wire-frame model to the linear features in the image, the model must be projected onto the image plane. This projection is computed by first applying rigid-body rotations and translations to change each model point  $\mathbf{X}$  from the model-centered coordinate frame to the camera-centered coordinate frame:

$$\mathbf{X}_c = \mathbf{R}_z \mathbf{R}_y \mathbf{R}_x \mathbf{X} + \mathbf{T} \quad (6.1)$$

in which  $\mathbf{R}_z$ ,  $\mathbf{R}_y$ , and  $\mathbf{R}_x$  are rigid-body rotation matrices, which are functions of the in-plane orientation  $\theta_z$  and the orientations in depth  $\theta_y$  and  $\theta_x$ , respectively.  $\mathbf{T} = (t_x, t_y, t_z)$  is a translation vector. Next, each model point is projected onto the image plane according to some imaging model. Under perspective projection with a pinhole camera, the transformation from a 3-D model point  $\mathbf{X}_c = (X_c, Y_c, Z_c)$  to an image point  $\mathbf{x} = (x, y)$  is given by

$$x = \frac{f_c}{Z_c} X_c, \quad y = \frac{f_c}{Z_c} Y_c \quad (6.2)$$

in which  $f_c$  is the focal length of the camera. Figure 6.2 depicts the perspective projection of the glycine model onto the image plane. M-SHARC uses the “weak perspective” imaging model, which accurately approximates perspective projection provided the depth of the imaged objects

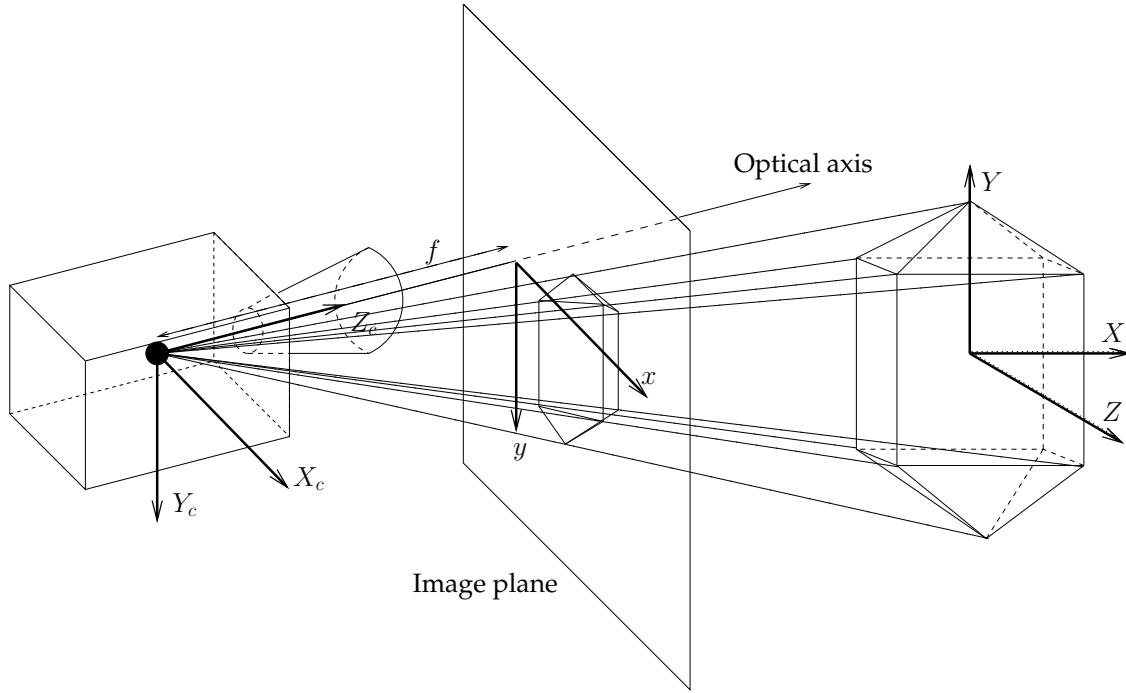


Figure 6.2: Depiction of the perspective projection of the glycine model onto the image plane. For simplicity, the image plane is displayed in front of the camera.

is small relative to the distance of the objects from the camera. Under such imaging conditions,  $f_c/Z_c$  and  $t_z$  can be assumed constant for all objects. In this chapter, we let  $f_c/Z_c = 1$  and  $t_z = 0$ .

The projection of the model onto the image plane is completed by determining the lines that are visible for the given pose. Given a convex model, the visible model lines can be determined by computing an outward normal vector (in camera-centered coordinates) for each surface of the 3-D model. The sign of the dot product of this normal vector with the camera's optical axis determines whether or not the surface is visible. The visible model lines are the lines that bound the visible surfaces.

The projection of the wire-frame model onto the image plane results in a set of projected model lines  $E^P = \{(\mathbf{M}_J, \hat{\mathbf{T}}_J, L_J)\}_{J=1\dots m}$  in which  $\mathbf{M}_J$  is a vector pointing from the origin of the image coordinate system to the midpoint of the  $J$ th model line,  $\hat{\mathbf{T}}_J$  is the unit tangent of the line,  $L_J$  is the length of the line, and  $m$  is the number of visible model lines. The set of data lines are

defined similarly as  $S = \{(\mathbf{m}_j, \hat{\mathbf{t}}_j, l_j)\}_{j=1\dots n}$ , in which  $n$  is the number of lines detected by the line finder.

M-SHARC follows the approach developed by Lowe [75, 76], consisting of three main steps: First, M-SHARC identifies linear features in the image. Second, M-SHARC identifies linear feature clusters that appear significant on the basis of viewpoint-independent relationships such as collinearity, parallelism, and end-point proximity. Third, M-SHARC fits a three-dimensional, wire-frame model to each significant linear feature cluster. The following sections describe each of these steps.

### 6.1.2 Linear feature detection

The M-SHARC algorithm uses the line finder proposed by Burns et al. [18], incorporating some of the speed-up suggestions given by Kahn et al. [61] (see Chapter 5 for details). The Burns line finder detects lines by identifying regions of pixels having similar image intensity gradient orientation. By detecting lines on the basis of gradient orientation (as opposed to gradient magnitude), the Burns line finder's performance is relatively insensitive to variations in contrast and brightness. This property is important for crystallization imaging because, as crystallization occurs, the increasing solids concentration causes more reflected light to reach the camera CCD, which increases the image brightness while decreasing the image contrast.

During linear feature extraction, single physical edges are commonly broken up into multiple collinear lines due to particle overlap, noise, or poor lighting. A key component of the algorithm, therefore, is the grouping of these collinear lines prior to searching for viewpoint-invariant line groups. The M-SHARC algorithm uses the method developed by Etemadi et al. [33] because it is straightforward to implement and relatively inexpensive computationally. The method uses simple thresholding on the angle, perpendicular distance, and endpoint distance between two lines to determine whether or not the lines are collinear. If collinearity requirements are met, M-SHARC creates a new line based on the collinear pair but also retains the two original lines in case the instance of collinearity is accidental. Therefore, subsequent grouping operations operate on

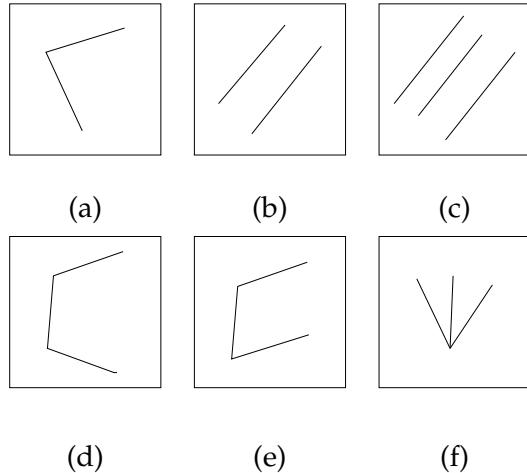


Figure 6.3: Depiction of different viewpoint-invariant line groups (VIGs) used by M-SHARC. (a) Junction. (b) Parallel pair. (c) Parallel triple. (d) C-triple. (e) C-square. (f) Arrow.

both the lines created from collinear grouping and all lines determined by the Burns line finder (whether or not they are involved in a collinear group). Retaining the original lines involved in collinear groups makes M-SHARC's performance less sensitive to the collinear grouping thresholds.

### 6.1.3 Perceptual grouping

Perceptual grouping refers to the task of organizing primitive objects, such as points or lines, into higher-level, meaningful structures. These structures, or groups, are useful as visual cues for the location, size, and orientation of a given object in the image. Viewpoint-invariant groups (VIGs), or groups that maintain certain properties regardless of the camera viewpoint, are necessary because the orientation of the object with respect to the camera is generally unknown. M-SHARC identifies line groups that can be assigned to one of the classifications depicted in Figure 6.3. The groups are identified based on orientation differences, spatial differences, and connectivities between the lines in the image. For example, junctions are line pairs that satisfy angle dissimilarity and endpoint proximity thresholds. Parallel line groups satisfy angle similarity and perpendicular distance thresholds. C-triples consist of three lines connected at two junctions where the angle

between the lines at each junction is greater than 90 degrees. C-squares consist of three lines connected at two junctions with two of the lines being approximately parallel. Arrows are three lines connected at a single junction where the angles between lines are less than 90 degrees.

M-SHARC calculates a significance measure for each VIG based on the line lengths, end-point distances, and (for groups of parallel lines) orientation differences, as described by Lowe [75]. This significance measure is used to ensure that the most visually salient VIGs are considered first in the model-fitting stage.

#### **6.1.4 Model-fitting**

The objective of the model-fitting process is to determine the model parameters and viewpoint parameters such that the two-dimensional projection of the geometric model matches the low-level features extracted from the image. The VIGs identified during the perceptual grouping stage provide the starting point for the tasks associated with the model-fitting process. The first task, called the correspondence problem, involves deciding which data lines correspond to which model lines. In general, multiple correspondences are possible, so M-SHARC may generate multiple hypotheses for a given VIG. Next, M-SHARC uses the data lines' positions and lengths to estimate the model and viewpoint parameters for each correspondence hypothesis. Given these parameters, M-SHARC projects the model into the image, identifies additional correspondences between model lines and data lines, and calculates a verification score for the set of correspondences. In the case of multiple hypotheses, M-SHARC chooses the correspondence hypothesis that results in the highest verification score and performs a least-squares minimization to achieve a better fit between the model and data lines. Finally, M-SHARC invalidates any VIGs that contain lines that are completely enclosed within the bounding box of the projection of the optimized model. These tasks are described in the following subsections.

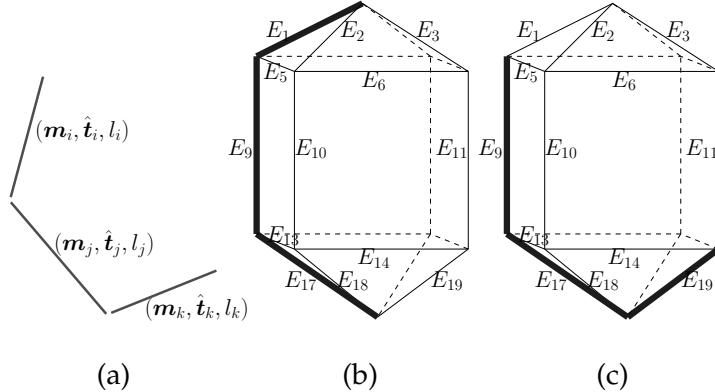


Figure 6.4: Depiction of two correspondence hypotheses. (a) Data line segments. (b) Hypothesis 1: Data lines  $i$ ,  $j$ , and  $k$  correspond to model edges  $E_1$ ,  $E_9$ , and  $E_{17}$ , respectively. (c) Hypothesis 2: Data lines  $i$ ,  $j$ , and  $k$  correspond to model edges  $E_9$ ,  $E_{17}$ , and  $E_{19}$ , respectively.

### Determining correspondences

The first step in the model-fitting process is to determine which of the model lines from the set  $E$  correspond to the data lines in a given VIG. M-SHARC solves the correspondence problem by rotating the data lines in the VIG into a standard position such that unambiguous descriptors can be assigned to each line in the group. These descriptors are used to hypothesize one-to-one correspondences between each data line in the VIG and a model line. M-SHARC currently has methods for hypothesizing correspondences for parallel pairs, parallel triples, and C-triples. For example, a triple of parallel lines is rotated such that all lines are basically vertical, and each line is labeled as left-most, center, or right-most. These labels are used to assign each of the parallel data lines to their corresponding model lines. In the case of a C-triple, the lines are rotated such that the center line is vertical, and the lines are labeled as top, center, or bottom. The correspondence remains ambiguous, however, so multiple hypotheses must be tested, as depicted in Figure 6.4. In this figure, data lines  $i$ ,  $j$ , and  $k$  could correspond, respectively, to either  $E_1$ ,  $E_9$ , and  $E_{17}$ , or  $E_9$ ,  $E_{17}$ , and  $E_{19}$ .

### Estimating model and viewpoint parameters

The problem of estimating viewpoint parameters, also referred to as pose estimation or alignment, has been studied extensively in the literature. The research has focused on estimating pose from both point correspondences [49, 55] and line correspondences [28, 21, 132]. A major drawback of these methods is that the internal model parameters are assumed known. The utility of these methods is therefore limited for the parameterized models of interest in our study.

M-SHARC estimates the internal model parameters using the properties of the data lines and an assumed orientation in depth. For example, given the first hypothesis in Figure 6.4, the model height  $h_m$  is estimated as  $l_j \cos \theta_x$  and the width  $w_m$  is estimated as  $2 \cos \theta_y (\max(|\hat{t}_i \cdot \hat{t}_j^\perp|, |\hat{t}_k \cdot \hat{t}_j^\perp|))$ , in which  $\theta_y$  and  $\theta_x$  are assumed orientations in depth and  $\hat{t}_j^\perp$  is a unit vector perpendicular to  $\hat{t}_j$ . The pyramid height  $t_m$  is estimated as  $w_m \tan \alpha_m / 2$ , in which  $\alpha_m$  is assumed to be 45 degrees. M-SHARC currently has methods for estimating model parameters on the basis of correspondences for parallel pairs, parallel triples, and C-triples.

Given the correspondences, assumed orientations in depth, and model parameters, the remaining viewpoint parameters are estimated in a straightforward manner. First, M-SHARC projects the 3-D model onto the image plane using Equations (6.1) and (6.2), assuming  $\theta_z = t_x = t_y = 0$ . Next, the in-plane orientation  $\theta_z$  is estimated using a weighted average of the orientation differences between the data lines and their corresponding, projected model lines. Given the estimated  $\theta_z$ , the 3-D model is then projected once again to the image plane and the translations  $t_x$  and  $t_y$  are estimated from a weighted average of the spatial differences between the model line midpoints and their corresponding data line midpoints.

### Identifying additional correspondences

As described in section 6.1.1, the projection of the wire-frame model onto the image plane results in a set of projected model lines  $E^P$  that can be compared directly with the set of data lines  $S$ . M-SHARC identifies additional model-data line correspondences by identifying instances of

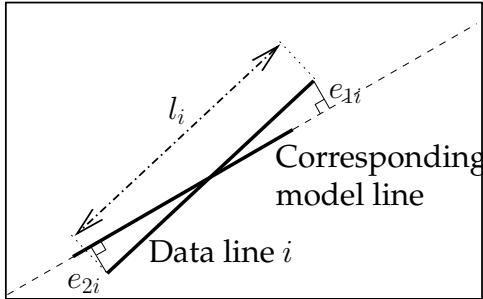


Figure 6.5: Depiction of variables used in mismatch calculation for a single line correspondence.

parallelism between the model lines and data lines.

### Calculating the verification score

The purpose of the verification score is to quantify the amount of evidence in the image supporting a given correspondence hypothesis. M-SHARC calculates the verification score as the percentage of visible model line length that is supported or overlapped by data lines. The verification score threshold is set based on experimentation. A more advanced method for setting this threshold is described by Grimson and Huttenlocher [44], but their method is not generalized to parameterized models.

### Minimizing model-data mismatch

Once a suitable hypothesis is identified, M-SHARC minimizes the mismatch between corresponding model and data lines by solving the optimization problem suggested by Lowe [76]

$$\min_{\mathbf{p}} \Phi = \sum_{i \in D_c} l_i (e_{1i} + e_{2i})^2 \quad (6.3)$$

subject to the model and imaging constraints (Equations (6.1) and (6.2)). In Equation (6.3),  $D_c$  is the set of data lines for which a correspondence has been identified,  $\mathbf{p}$  is the parameter vector,  $l_i$  is the length of data line  $i$ , and  $e_{1i}$  and  $e_{2i}$  are the perpendicular distances from data line  $i$ 's endpoints to the corresponding model line, as depicted in Figure 6.5. The parameter vector for the

glycine model in Figure 6.1 is

$$\mathbf{p} = \begin{bmatrix} h_m & t_m & w_m & \theta_x & \theta_y & \theta_z & t_x & t_y \end{bmatrix} \quad (6.4)$$

### **Invalidating overlapping VIGs**

The VIGs identified during M-SHARC's perceptual grouping stage can overlap, sharing lines. We assume, however, that each line in the image can be attributed to only one crystal. Thus, once a model is successfully fit to a set of lines, any VIG that contains one or more of those lines is considered invalid. Furthermore, we assume that any line completely contained within the bounding box of an identified crystal has arisen due to unmodeled features of that crystal. Thus, any VIG that contains one or more of these lines is also considered invalid. M-SHARC does not attempt to fit a model to invalid VIGs.

Applying this constraint on overlapping VIGs is advantageous in that it significantly reduces the number of VIGs investigated, thus increasing M-SHARC's efficiency. Furthermore, initial studies indicated that a large number of false positives are identified if VIG invalidation is not employed. Thus, the overlapping crystal constraint serves to increase the algorithm's accuracy as well as reduce the computational burden such that the algorithm can run in real-time on a single processor. However, this feature requires M-SHARC to investigate each VIG serially, starting with the most significant VIG and proceeding to the least significant, such that M-SHARC currently cannot be implemented in parallel.

#### **6.1.5 Summary and example**

The M-SHARC algorithm can be summarized using the following pseudo-code:

1. Detect linear features in image
2. Identify and sort viewpoint-invariant line groups (VIG)

For each VIG

If VIG is valid

```

3. Generate correspondence hypotheses

For each hypothesis

    4. Estimate model and viewpoint parameters

    5. Project model into image

    6. Search for additional correspondences

    7. Compute verification score

Endfor

8. Select hypothesis with highest score

If (score > Verification threshold)

    9. Minimize model-data mismatch

    10. Save optimized model information

    11. Invalidate overlapped VIGs

Endif

Endif

Endfor

```

Figure 6.6 shows the step-by-step results of applying the M-SHARC algorithm to a region of interest from an image of  $\alpha$ -glycine crystals. This region of interest contains a large, well-defined crystal that is partially overlapped by a small, poorly-defined crystal. Applying the line finder to the image in Figure 6.6(a) produces the dark lines shown in Figure 6.6(b). Because of overlap by the smaller crystal, a few of the edges of the larger crystal are broken up into two or more lines. Using collinear line grouping, these broken-up lines are combined into the single, light lines displayed in Figure 6.6(c). In the perceptual grouping stage, the line group that is identified as being most significant is the triple of parallel lines shown in Figure 6.6(d). The lengths, spatial positions, and relative distances between these lines provide the information necessary to estimate the internal model parameters and pose, resulting in the model projection shown in Figure 6.6(e). In this figure, the gray, solid lines are the visible model lines and the hidden model lines are not shown.

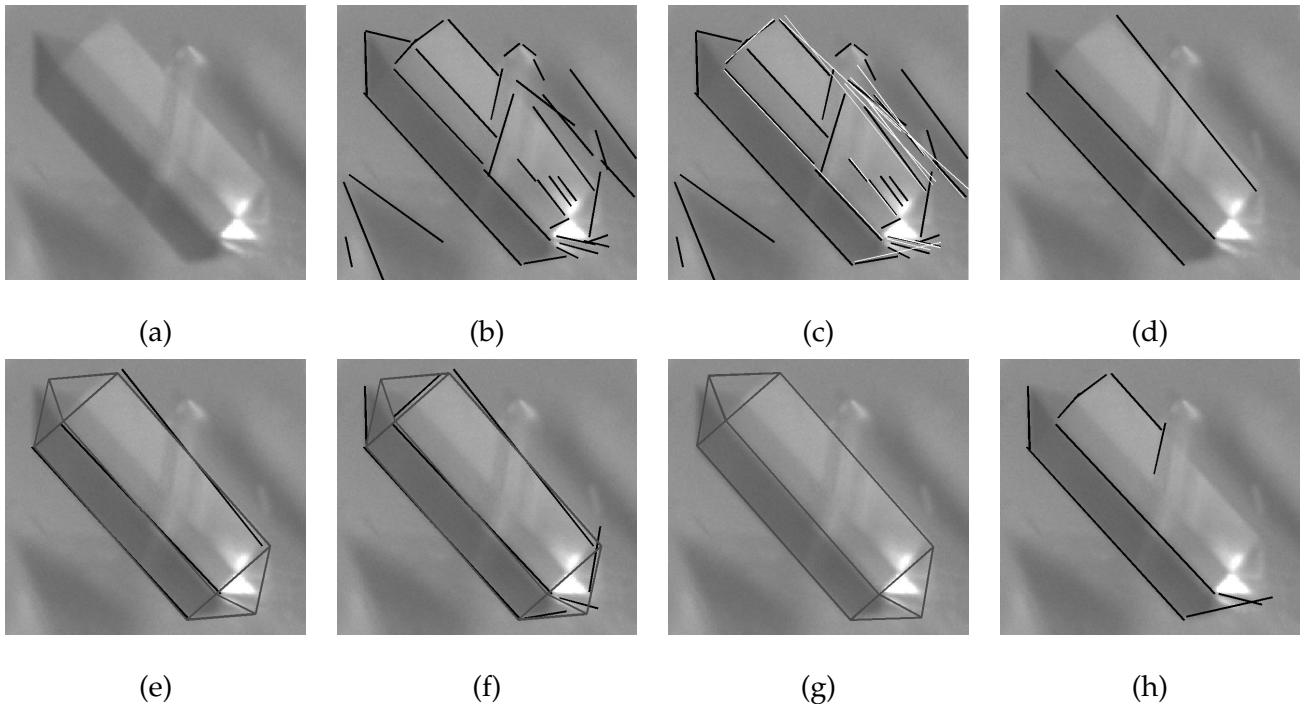


Figure 6.6: Result of applying M-SHARC to image of  $\alpha$ -glycine crystal. (a) Original region of interest. (b) Linear features extracted using Burns line finder (dark lines). (c) Linear features extracted using collinear grouping (light lines). (d) Most salient line group. (e) Model initialization. (f) Identification of additional correspondences. (g) Optimized model fit. (h) Invalidated VIGs.

Figure 6.6(f) shows the additional correspondences between model and data lines found by identifying instances of parallelism between the visible model lines in Figure 6.6(e) and the data lines in Figure 6.6(c). Figure 6.6(g) shows the optimized model obtained by minimizing the perpendicular distances between each of the model-data line correspondences. Finally Figure 6.6(h) indicates three of the VIGs that are invalidated due to overlap with the identified model.

## 6.2 Results

To assess M-SHARC’s performance, we carried out an unseeded,  $\alpha$ -glycine cooling crystallization by dissolving 180 g of glycine (p.a., Acros Organics) in 600 mL of deionized water at 55 °C. The solution was cooled to 25 °C at 5 °C/hr. Spontaneous nucleation was observed around 29 °C.

Three sets of video images were acquired during the course of the experiment, each set consisting of 100 images acquired at a rate of 30 frames per second. Each set was acquired once a noticeable increase in the solids concentration had occurred. The first set of video images was acquired approximately 13 minutes after nucleation at low solids concentration with the draft tube in the middle of the crystallizer clearly visible. The second set of images was acquired approximately 24 minutes after nucleation at medium solids concentration with the draft tube just barely visible. The final set of images was acquired 43 minutes after nucleation at high solids concentration with the draft tube completely obscured by the slurry. The polymorphic form of the observed crystals was confirmed using X-ray powder diffraction. Given the rate of acquisition (30 frames per second), each set of images essentially represents a snapshot in time. All images were analyzed by M-SHARC using the same set of parameters. The VIGs used to initialize model-fitting were parallel triples and C-triples.

### 6.2.1 Visual evaluation

Figures 6.7–6.9 show the results of applying M-SHARC to images acquired at low, medium, and high solids concentrations. These figures demonstrate M-SHARC’s ability to identify crystals covering a wide range of sizes and orientations. These figures also demonstrate M-SHARC’s ability to handle poor image quality. Although several options exist for improving the images, such as lowering the mixing rate to decrease motion blur, successful image analysis of these low-quality images demonstrates M-SHARC’s robustness and helps ensure success in the rugged industrial environment.

### 6.2.2 Comparison with human analysis

To quantify M-SHARC’s performance, we compare its results with the results obtained by manual analysis of the images by human operators. All 300 images acquired during the  $\alpha$ -glycine crystallization were analyzed both by M-SHARC and by a human operator. The human operator annotated the images using LabelMe, a database and web-based image annotation tool developed

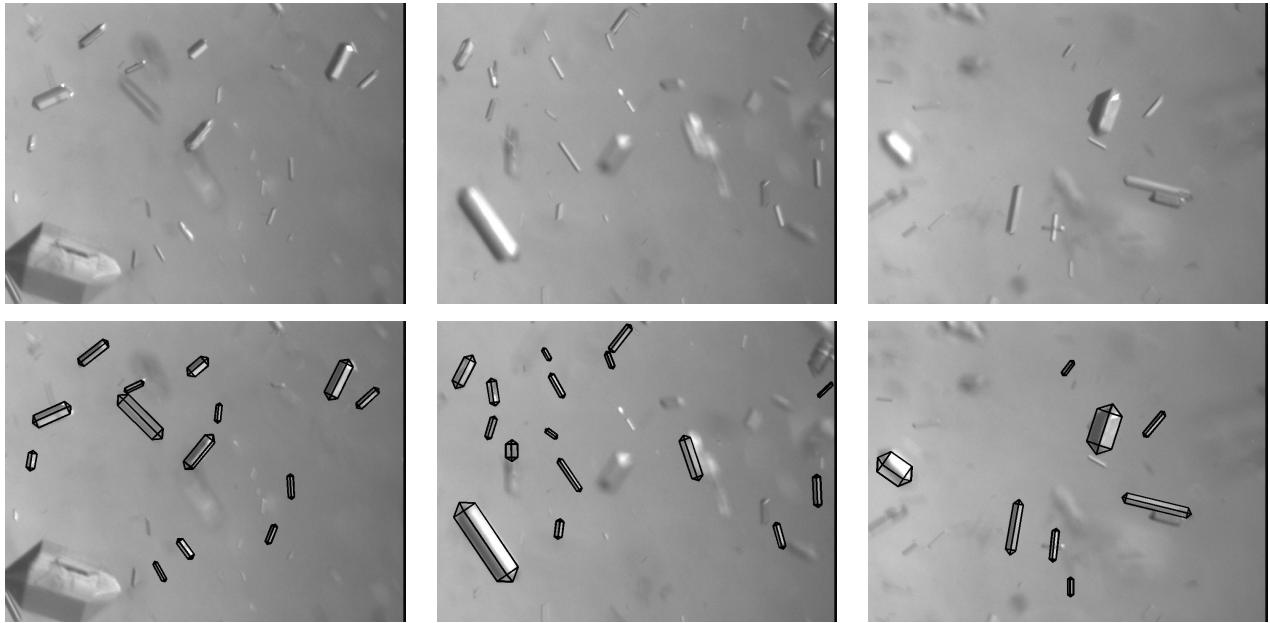


Figure 6.7: M-SHARC segmentation results for selected images acquired at low solids concentration (13 min. after appearance of crystals).

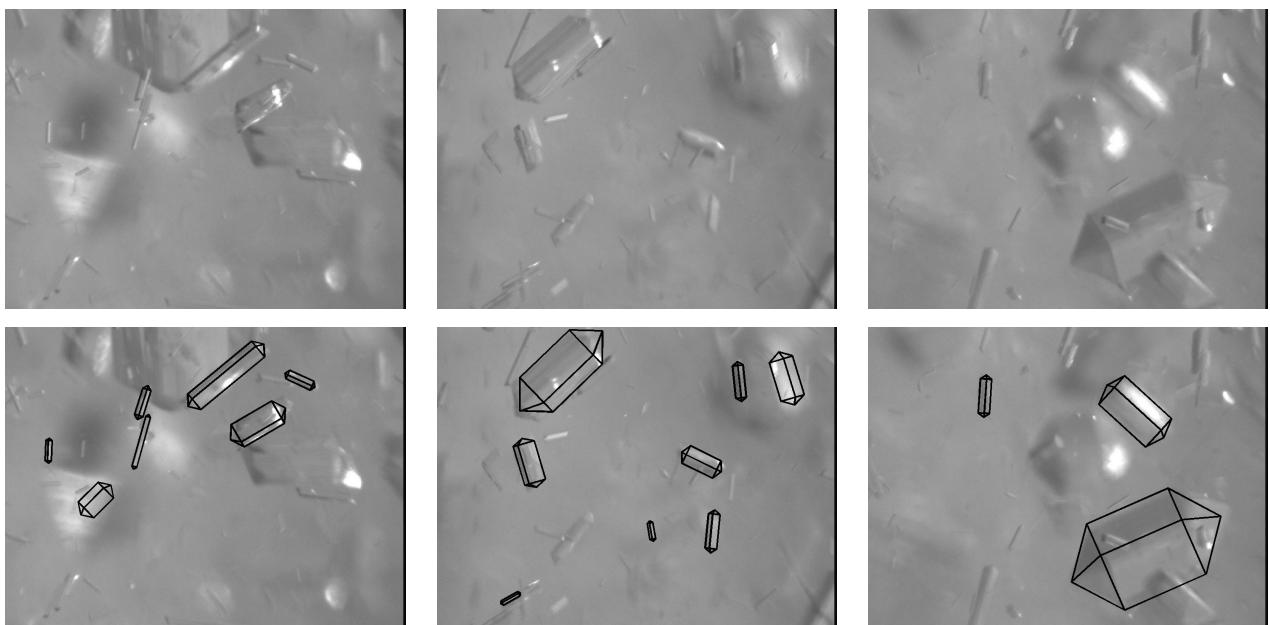


Figure 6.8: M-SHARC segmentation results for selected images acquired at medium solids concentration (24 min. after appearance of crystals).

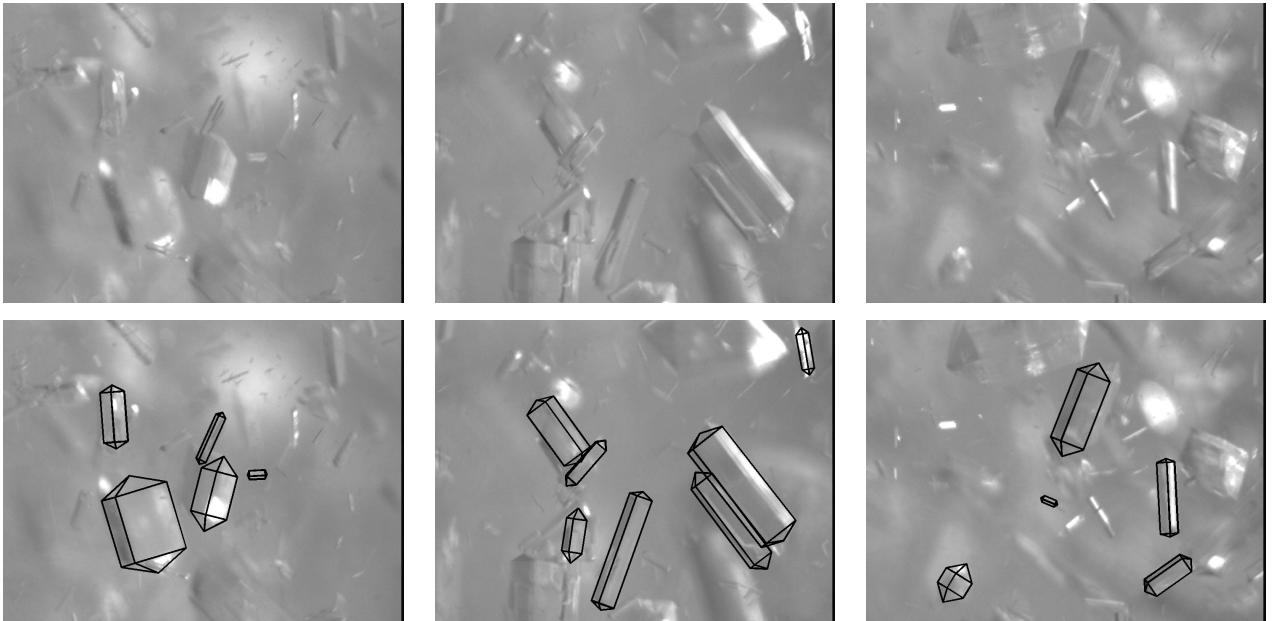


Figure 6.9: M-SHARC segmentation results for selected images acquired at high solids concentration (43 min. after appearance of crystals).

by Russel et al. [108]. The images and annotations are available to the general scientific community through the LabelMe website.

Figure 6.10 illustrates how M-SHARC's results are compared with the human operator's results. Figure 6.10(b) shows the crystal outlines determined by the human operator while Figure 6.10(c) shows the outlines determined by M-SHARC. Each outline is simply a set of straight lines. If a sufficiently large set of correspondences can be found between the set of lines in an M-SHARC outline and the set of lines in a human outline, the crystal corresponding to those outlines is classified as a *hit*. If a sufficiently large set of correspondences is *not* found between a given M-SHARC outline and any other human outline, the crystal corresponding to the M-SHARC outline is classified as a *false positive*. Similarly, a crystal identified by a human operator for which no corresponding M-SHARC outline can be found is classified as a *miss*. In Figure 6.10(d), the false positives are displayed in white and the misses in black.

Table 6.1 shows the number of hits  $N_H$ , misses  $N_M$ , and false positives  $N_{FP}$  identified

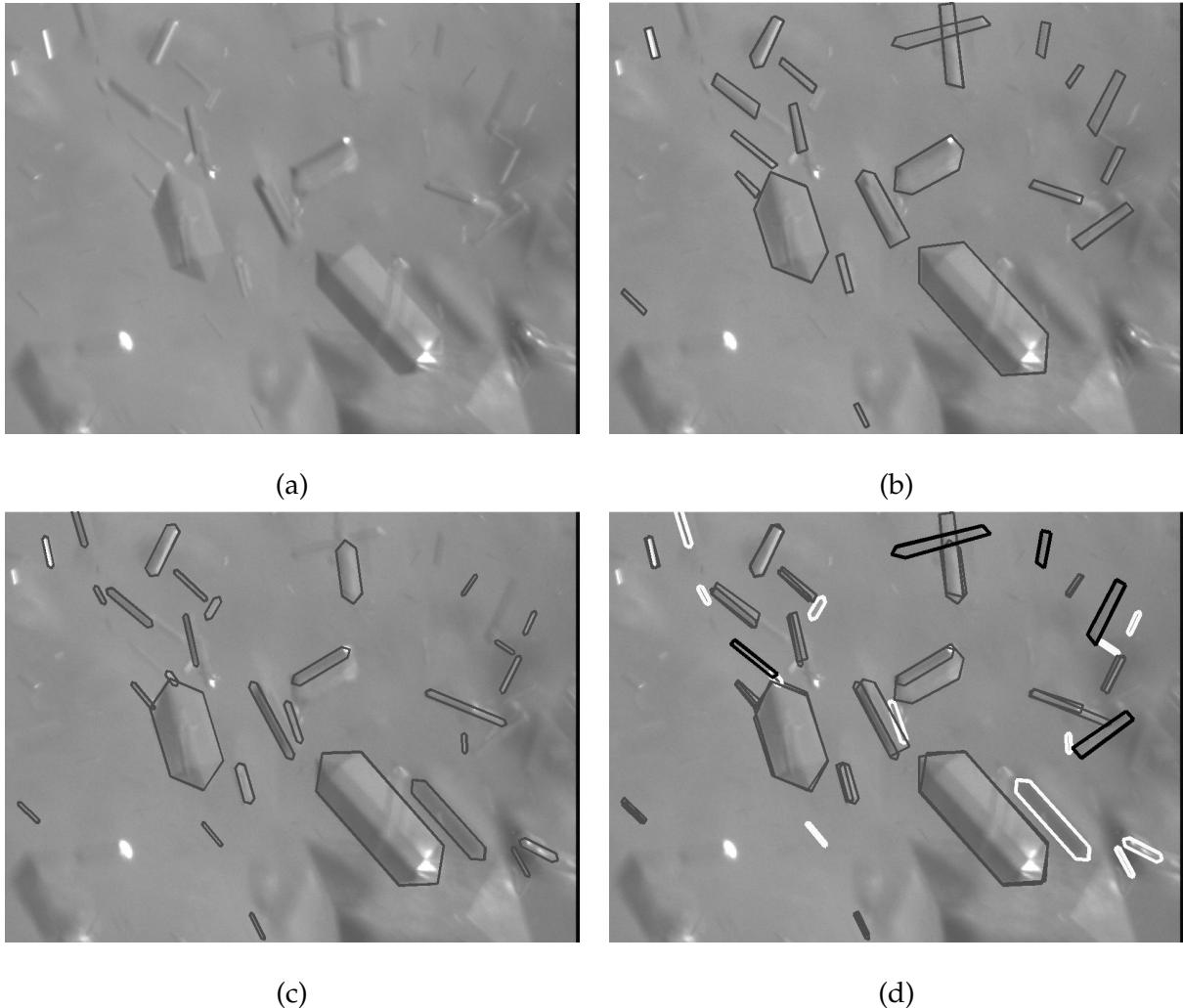


Figure 6.10: Illustration of comparison between human operator results and M-SHARC results. (a) Original image. (b) Outlines of crystals identified by human operator. (c) Outlines of crystals identified by M-SHARC. (d) Result of comparison between human outlines and M-SHARC outlines. Crystals identified as false positives are outlined in white while those identified as misses are outlined in black.

by comparing M-SHARC's results with the human operator's results for each of the three sets of video images, along with number and area fractions for the hits and false positives. The hit, miss, and false positive areas ( $A_H$ ,  $A_M$ , and  $A_{FP}$ , respectively) are calculated based on the areas of the polygons defined by each crystal outline. For each level of solids concentration, the hit number

	Low	Med.	High
Hits ( $N_H$ )	500	279	220
Misses ( $N_M$ )	514	657	445
False Positives ( $N_{FP}$ )	130	191	352
Hit number fraction ( $N_H/(N_H + N_M)$ )	0.49	0.30	0.33
False Pos. number fraction ( $N_{FP}/(N_H + N_{FP})$ )	0.21	0.41	0.62
Hit area fraction ( $A_H/(A_H + A_M)$ )	0.63	0.38	0.31
False Pos. area fraction ( $A_{FP}/(A_H + A_{FP})$ )	0.23	0.36	0.53

Table 6.1: Summary of comparison between M-SHARC results and human operator results for in situ video images obtained at low, medium, and high solids concentrations (100 images at each concentration).

fraction is significantly less than the hit area fraction while the false positive number fraction is comparable to the false positive area fraction. Thus, we would expect the number-based area distribution to be biased towards larger particles. This expectation is verified by Figure 6.11, which compares the number-based cumulative distribution functions (CDFs) for particle area. As a standard of comparison, Figure 6.11 also shows the CDFs constructed using only particles classified as hits. The confidence interval displayed in these figures is calculated using the Kolmogorov-Smirnov statistic [80, 24].

To identify possible improvements to the M-SHARC algorithm, we examined the results of M-SHARC’s subroutines (line finding, perceptual grouping, and model fitting) for the 100 largest misses and the 100 largest false positives. Examination of the large misses reveals that many of the misses are particles that are somewhat blurry with low contrast edges that are difficult for M-SHARC’s line finder to identify (see row 1 in Figure 6.12). Despite the low contrast, the visual cues are sufficient that the outline of these particles can be discerned by a human operator. Particles that are clearly in-focus also sometimes exhibit low contrast edges and cause difficulties for M-SHARC’s line finder (see row 2 in Figure 6.12). Particle agglomeration is another major source

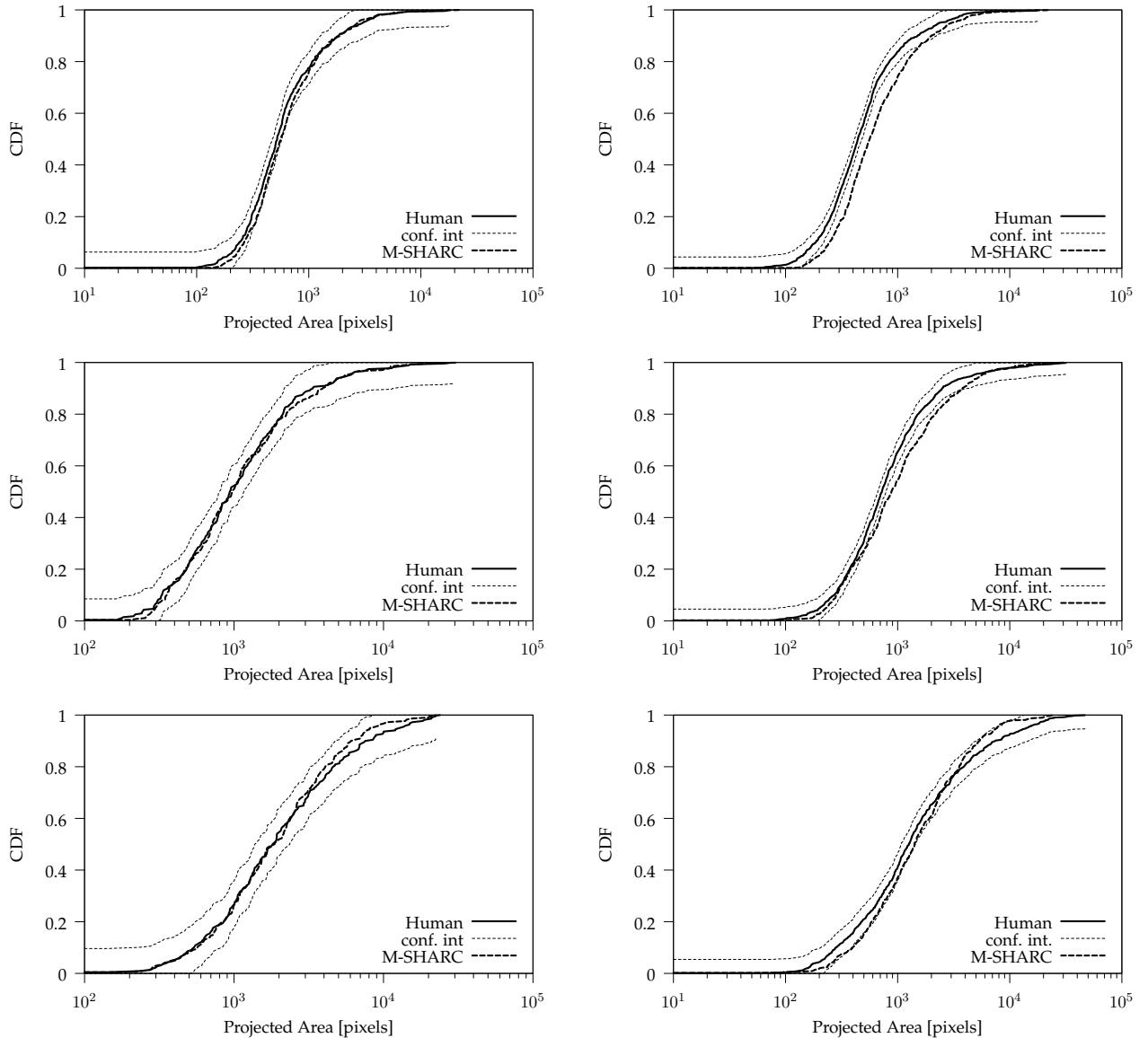


Figure 6.11: Comparison of Human and M-SHARC cumulative distribution functions for projected area. Rows 1, 2, and 3 show results for the  $\alpha$ -glycine experiment at low, medium, and high solids concentrations, respectively. Column 1: CDFs constructed using only crystals classified as hits. Column 2: CDFs constructed using all crystals (i.e. the Human CDF is based on crystals classified as either hit or miss, while the M-SHARC CDF is based on crystals classified as either hit or false positive).

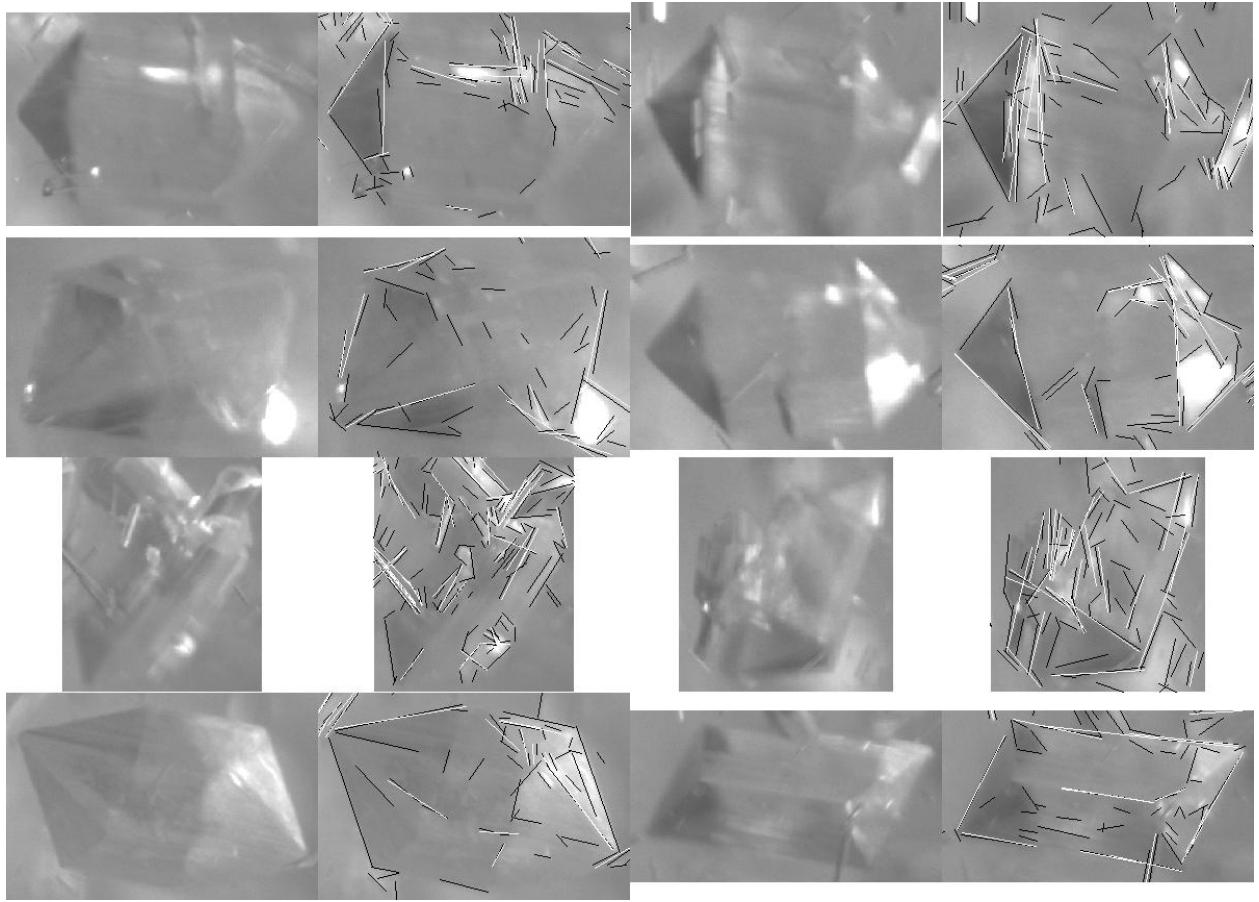


Figure 6.12: Results of linear feature detection for selected crystals missed by M-SHARC. The poor contrast for the crystals in row 1 is due to out-of-focus blur. The crystals in row 2 also exhibit poor contrast despite being seemingly in-focus. The crystals in row 3 show examples of agglomeration. The crystals in row 4 may be identifiable given further development of M-SHARC’s correspondence and model parameter estimation routines described in Sections 6.1.4 and 6.1.4.

of difficulty for M-SHARC, causing the particle edges to be broken up at each point of agglomeration. For agglomerated particles, the line-finding routine results in a concentrated group of small linear features, complicating the perceptual grouping stage (see row 3 in Figure 6.12). The degree of failure at the line-finding stage due to blur and agglomeration usually makes successful perceptual grouping an unreasonable goal. In some cases, salient VIGs can be found but have not been

utilized because routines for solving the correspondence and model parameter estimation problems have not yet been developed. Row 4 in Figure 6.12 gives two examples of crystals resulting in significant VIGs for which initialization routines have yet to be developed. The development of such routines may be the best approach to minimize the number of misses.

Examination of the 100 largest false positives showed that a large fraction (more than 1/3) of the false positives are agglomerated or blurry crystals. The ambiguity associated with determining an outline for blurry and agglomerated crystals caused the human operators to pass over these crystals. Whether or not these false positives should be counted as false positives is unclear. The other major source of false positives arises due to shortcomings in M-SHARC's method of verification. For many of the false positives, the model and data lines do not align well, indicating that the orientational and spatial offset thresholds used in the verification process are not sufficiently stringent. Another indication that more stringent thresholds are necessary in the verification stage is that many of the false positives have only 3 or 4 data lines that correspond to the model lines. Unfortunately, this is also true for many of the hits, partly because the wire-frame model is only a rough approximation of the  $\alpha$ -glycine shape. Perhaps using wire-frame model that more accurately represents the  $\alpha$ -glycine shape would result in better fits for the hits, enabling the use of more stringent verification thresholds to eliminate false positives.

### 6.2.3 Algorithm speed

The average cputimes required to analyze images from each of the three video sets from the  $\alpha$ -glycine experiment are shown in Table 6.2. The amount of time between each video acquisition during the crystallization experiment was approximately 12 minutes. Based on the cputimes in Table 6.2, M-SHARC can analyze approximately 10 images per minute, or 100 images every 10 minutes. This analysis speed is sufficiently fast for real-time implementation on chemical systems with crystallization dynamics similar to glycine.

The results presented in this chapter are based on using a single wire-frame model. M-SHARC can be applied, however, using multiple models to represent different shape classes. Us-

Set	Line Finding	Group Finding	Initialization	Optimization	Total
1	2.6 (55)	1.2 (25)	0.2 (5)	0.7 (14)	4.8
2	3.3 (43)	3.3 (43)	0.3 (4)	0.7 (9)	7.6
3	3.3 (16)	15.4 (76)	0.4 (2)	1.0 (5)	20.2

Table 6.2: Average cputime required to analyze single image for three different image sets of increasing solids concentration. The first number in each number pair is the cputime in seconds. The second, parenthesized number gives the percentage of total cputime.

ing multiple models would affect the computational requirements in two ways. First, each additional model may require the identification of additional VIGs to serve as visual cues for that model. However, the increase in computation time for the VIG identification stage may be small because many VIGs are identified using the same set of logical decisions. For instance, to identify a C-triple, M-SHARC must determine whether a given triple of lines is connected at one or two junctions. This same determination must be made to identify a C-square and an arrow. Second, the number of hypotheses would likely increase linearly with the number of models, resulting in a linear increase of computational requirements for the model initialization stage. In Table 6.2, the requirements for the initialization stage represent a small fraction of the total computational requirements, indicating that M-SHARC could accommodate several more models and remain suitably fast for real-time implementation.

### 6.3 Conclusions

We have developed a model-based object recognition algorithm that is effective for extracting crystal size and shape information from noisy, *in situ* crystallization images. The algorithm's accuracy has been assessed by comparing its results with those obtained by manual, human analysis of the images. At low solids concentrations, the algorithm identifies approximately half of the crystals identified by humans, while at medium to high solids, the algorithm identifies approxi-

mately one-third of the crystals. At low solids, false positives constitute approximately one-fifth of all identified crystals, while at medium to high concentrations the false positives constitute approximately half of the identified crystals. Despite the misses and false positives, the algorithm's cumulative size distribution measurements compare favorably with measurements obtained by humans but are biased towards larger particles. To improve the algorithm's accuracy, further development should focus on the algorithm's verification stage and on creating initialization routines for additional viewpoint-invariant line groups. The algorithm is sufficiently fast to provide on-line measurements for typical cooling crystallization processes.



## Chapter 7

# Statistical Estimation of PSD from Imaging Data <sup>1</sup>

The methods described in Chapters 5 and 6 address the image segmentation problem, which is the first of two main challenges that must be overcome to use *in situ* microscopy for PSD measurement. The second challenge is to estimate the PSD given the size and shape information obtained through successful image segmentation. Each segmented particle provides a single observation, which can be either *censored* or *uncensored*. A censored observation refers to an observation in which only partial information is obtained. For example, an observation of a particle touching the image border is censored because only a portion of the particle is visible. An observation of a particle with one end partially occluded by another particle is also censored. An observation is uncensored only if the particle is enclosed entirely within the image frame, is not occluded by other particles, and is oriented in a plane perpendicular to the optical axis of the camera.

A natural approach to estimate the PSD is to count only those particles appearing entirely within the field of view, not touching the image boundary. This approach, called minus-sampling, introduces sampling bias. Sampling bias occurs when the probability of observing an object depends on its size or shape. For example, a small particle randomly located in the image has a high probability of appearing entirely within the field of view, while a sufficiently large particle randomly located in the image may have zero probability of appearing entirely within the field of view. Miles [83] presented the first treatment of spatial sampling bias, developing a minus-

---

<sup>1</sup>Portions of this chapter are to appear in Larsen and Rawlings [68]

sampling estimator that corrects spatial sampling bias by weighting each observation by  $M^{-1}$ , with  $M$  being related to the sampling probability of the observed particle. Miles derived formulas for  $M$  assuming a circular sampling region. Lantuéjoul[64] extended Miles' results by showing how to calculate  $M$  for a rectangular sampling region. Baddeley [5] provides an excellent review of various methods for correcting edge effects under a variety of situations.

The primary drawback of the Miles-Lantuéjoul approach is that it uses only uncensored observations. If the size of the particles is large relative to the size of the image window, using censored observations (i.e. particles touching the image border) would be expected to result in improved PSD estimation. The primary goal of this chapter is to develop a PSD estimator using both censored and uncensored observations and to evaluate the benefits and drawbacks of this estimator relative to the Miles-Lantuéjoul approach. We assume the censoring is due only to particles touching the image border and not due to orientation or occlusion effects. A secondary goal is to develop practical methods for determining confidence intervals for the estimated PSD. The methods developed in this study are intended for systems of high-aspect-ratio particles, which are commonplace in the pharmaceutical and specialty chemical industries.

This chapter is organized as follows. Section 7.1 describes previous work related to PSD estimation of high-aspect-ratio particles and describes the application of the Miles-Lantuéjoul estimator. Section 7.2 presents the formulation of the maximum likelihood PSD estimator. Section 7.3 presents the results of applying the estimator to artificial images generated as described in Chapter 4. Section 7.4 summarizes our findings. The full derivation of the maximum likelihood estimator can be found in Appendix A.

## 7.1 Previous work

PSD estimation for high-aspect-ratio particles using imaging-based measurements is related to the problem of estimating the cumulative length distribution function  $H$  of line segments observed through a window, which has been investigated by several researchers. Laslett [71] was the first

to derive the log likelihood for this problem. Wijer [130] derived the non-parametric maximum likelihood estimator (NPMLE) of  $H$  for a circular sampling region and an unknown orientation distribution function  $F$ . For arbitrary convex sampling regions, Wijer shows how to estimate  $H$  assuming  $F$  is known. Van Der Laan[121] studies the NPMLE of  $H$  for the one-dimensional line segment problem (i.e. all line segments have same orientation) for a non-convex sampling window, and Van Zwet[122] derives the NPMLE of  $H$  for the two-dimensional problem with a non-convex, highly irregular sampling region and known  $F$ . Svensson et al.[118] derive an estimator for a parametric length density function  $h$  using one-dimensional line segment data from a circular sampling region. Hall [46] derived an estimator for the intensity of a planar Poisson line segment process that is unbiased for any convex sampling region and any length distribution function. All of the above studies utilize both censored and uncensored observations. Baddeley [5] provides an excellent review of various spatial sampling estimation studies.

The goal of the current study is to estimate the particle size distribution  $f$ , which is related to but different than the cumulative distribution function  $H$  or corresponding density function  $h$  for a line segment process. The PSD  $f(L)$  is the number of particles of length  $L$  per unit volume and is related to  $H$  via the relation  $H(l) = \int_0^l f(L)dL / \int_0^\infty f(L)dL$ . The approach commonly used in practice to estimate the PSD from imaging-based measurements is the Miles-Lantuéjoul method [83]. As there is some confusion amongst practitioners regarding the implementation of the Miles-Lantuéjoul method, we describe the method here.

Let  $E^2$  be the Euclidean plane, and let  $J \subset E^2$  be a domain parameterized by  $(z, n, \theta_n)$ , in which  $z$  gives the center point of the domain,  $n$  gives the class, and  $\theta_n$  is a vector giving the parameters necessary to completely specify a domain of class  $n$ . Let  $Q(x) \subset E^2$  be a sampling region centered at  $x$ . For each domain  $J$ , define the set

$$J^\alpha = \{x \in E^2 : J \subset Q(x)\} \quad (7.1)$$

Thus,  $J^\alpha$  is a domain comprising all points at which the sampling region can be placed and enclose entirely the domain  $J$ .

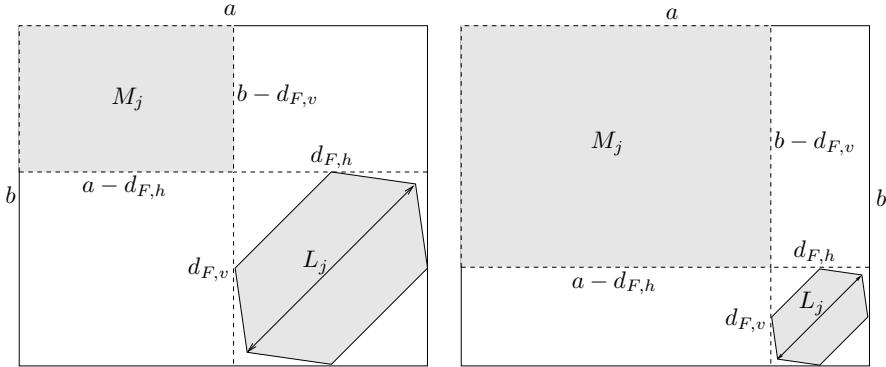


Figure 7.1: Depiction of methodology for calculating Miles-Lantuéjoul M-values for particles of different lengths observed in an image of dimension  $b \times a$ .

Define  $M\{J\} = A\{J^\alpha\}$ , where  $A\{\cdot\}$  denotes area. Let  $M\{J_j\}, j = 1 \dots n$  be the M-values calculated for  $n$  observations of particles with lengths corresponding to size class  $i$ . Miles showed that  $\hat{\rho}_{ML_i} = \sum_j M\{J_j\}$  is an unbiased estimator of  $\rho_i$ , the density of particles in size class  $i$  per area, provided the minimum possible M-value for a particle in size class  $i$  is greater than zero.

In Miles' original paper [83], he derived formulas for calculating M-values for arbitrary domains assuming a circular sampling region. Later, Lantuéjoul[64] extended Miles' results by showing how to calculate  $M$  for the rectangular sampling region typical of microscopy applications. For an image of size  $a \times b$ , the M-value of a particle is calculated by subtracting the vertical and horizontal Feret diameters of the particle ( $d_{F,v}$  and  $d_{F,h}$ ) from, respectively, the vertical and horizontal image dimensions  $b$  and  $a$ , as depicted in Figure 7.1.

## 7.2 Theory

### 7.2.1 PSD Definition

Consider a population of cylindrical or rod-like particles. The geometry of each particle is specified in terms of the cylinder height and radius. Let the characteristic length  $L$  for the population of particles be the cylinder height.

Consider a slurry of volume  $V$  in which a solid phase of discrete particles is dispersed

in a continuous fluid phase. Let  $f(L)$  denote the continuous PSD, or the number of particles of characteristic length  $L$  per unit volume slurry. Number-based PSDs are typically measured by discretizing the characteristic length scale into  $T$  non-overlapping bins or size classes. We therefore define the discrete PSD as

$$\rho_i = \int_{S_i}^{S_{i+1}} f(l) dl, \quad i = 1, \dots, T \quad (7.2)$$

in which  $S = (S_1, \dots, S_{T+1})$  is the vector of breaks between size classes.

The relative PSD  $\mathbf{q}$  is a vector with elements

$$q_i = \frac{\rho_i}{\sum_j^T \rho_j}, \quad i = 1, \dots, T \quad (7.3)$$

In this chapter, the term PSD is assumed to refer to the discrete, absolute PSD  $\rho$  unless specifically noted otherwise.

### 7.2.2 Sampling model

The particle population is sampled using in situ imaging. Let  $V_I \in V$  denote the imaging volume, and assume  $V_I$  is a rectangular region of dimensions  $a \times b \times d$ , in which  $a$  is the horizontal image dimension,  $b$  is the vertical image dimension, and  $d_f$  is the depth of field.  $a$  and  $b$  determine the field of view, and we assume  $a \geq b$ . A single random sample of the population consists of an image containing the two-dimensional projection of the portions of particles inside  $V_I$ . We assume the system is well-mixed such that the centroids of the particles are randomly and uniformly distributed in space.

We assume the camera is positioned a fixed distance  $z_0$  from the imaging volume, and that  $d_f \ll z_0$ . This assumption means the particles in the imaging volume are projected onto the image plane according to the weak perspective projection model. In other words, the projected particle lengths measured in the image coordinate system can be related to the true projected particle lengths by applying a constant magnification factor  $m$ .

We assume all particles are oriented in a plane orthogonal to the camera's optical axis. This assumption, together with the weak perspective assumption, essentially reduces the 3-D process

to a 2-D process, thereby simplifying the analysis considerably. These assumptions are not used only for convenience, however, but rather to reflect the actual conditions under which in situ imaging measurements are made in practice. To obtain useful in situ images in high solids concentrations, the camera must have a small depth of field and be focused only a small depth into the particulate slurry. It seems reasonable, therefore, to expect the shear flow at the slurry-sensor interface to cause the particles to align orthogonal to the this interface, and thus orthogonal to the camera's optical axis.

### 7.2.3 Maximum likelihood estimation of PSD

Let  $\mathbf{X}_k = (X_{1k}, \dots, X_{Tk})$  be a  $T$ -dimensional random vector in which  $X_{ik}$  gives the number of *non-border* particles of size class  $i$  observed in image  $k$ . A non-border particle is a particle that is completely enclosed within the imaging volume. A *border* particle, on the other hand, is only partially enclosed within the imaging volume such that only a portion of the particle is observable. For border particles, only the observed length (i.e. the length of the portion of the particle that is inside the imaging volume) can be measured. Accordingly, we let  $\mathbf{Y}_k = (Y_{1k}, \dots, Y_{Tk})$  be a  $T$ -dimensional random vector in which  $Y_{jk}$  gives the number of *border* particles with observed lengths in size class  $j$  that are observed in image  $k$ . We denote the observed data, or the realizations of the random vectors  $\mathbf{X}_k$  and  $\mathbf{Y}_k$ , as  $\mathbf{x}_k$  and  $\mathbf{y}_k$ , respectively.

The particle population is represented completely by the vectors  $\boldsymbol{\rho} = (\rho_1, \dots, \rho_T)$  and  $\mathbf{S} = (S_1, \dots, S_{T+1})$  in which  $\rho_i$  represents the number of particles of size class  $i$  per unit volume and  $S_i$  is the lower bound of size class  $i$ . Given the data  $\mathbf{x}$  and  $\mathbf{y}$  (the subscript  $k$  denoting the image index is removed for simplicity), the maximum likelihood estimator of  $\boldsymbol{\rho}$  is defined as

$$\hat{\boldsymbol{\rho}}_b = \arg \max_{\boldsymbol{\rho}} p_{XY}(x_1, y_1, x_2, y_2, \dots, x_T, y_T | \boldsymbol{\rho}) \quad (7.4)$$

in which the subscript  $b$  indicates the use of border particle measurements and  $p_{XY}$  is the joint probability density for  $\mathbf{X}$  and  $\mathbf{Y}$ . In other words, we want to determine the value of  $\boldsymbol{\rho}$  that maximizes the probability of observing exactly  $x_1$  non-border particles of size class 1,  $y_1$  border particles

of size class 1,  $x_2$  non-border particles of size class 2,  $y_2$  border particles of size class 2, and so on.

A simplified expression for  $p_{XY}$  can be obtained by noting that, at least at low solids concentrations, the observations  $X_1, Y_1, \dots, X_T, Y_T$  can be assumed to be independent. This assumption means that the observed number of particles of a given size class depends only on the density of particles in that same size class. At high solids concentrations, this assumption seems unreasonable because the number of particle observations in a given size class is reduced due to occlusions by particles in other size classes. At low concentrations, however, the likelihood of occlusion is low. The independence assumption does *not* imply that the observations are not correlated. Rather, the assumption implies that any correlation between observations is due to their dependence on a common set of parameters. As an example, if we observe a large number of non-border particles, we would expect to also observe a large number of border particles. This correlation can be explained by noting that the probability densities for both border and non-border observations depend on a common parameter, namely, the density of particles. Given the independence assumption, we express the likelihood function  $L(\boldsymbol{\rho})$  as

$$L(\boldsymbol{\rho}) = p_{XY} = \prod_{i=1}^T p_{X_i}(x_i|\boldsymbol{\rho}) \prod_{j=1}^T p_{Y_j}(y_j|\boldsymbol{\rho}) \quad (7.5)$$

in which  $p_{X_i}$  and  $p_{Y_j}$  are the probability densities for the random variables  $X_i$  and  $Y_j$ . The log likelihood is defined as  $l(\boldsymbol{\rho}) = \log L(\boldsymbol{\rho})$ . Maximizing the likelihood function is equivalent to minimizing the log likelihood. Using Equation (7.5), the estimator in Equation (7.4) can therefore be reformulated as

$$\hat{\boldsymbol{\rho}}_b = \arg \min_{\boldsymbol{\rho}} \sum_{i=1}^T -\log p_{X_i}(x_i|\boldsymbol{\rho}) - \sum_{j=1}^T \log p_{Y_j}(y_j|\boldsymbol{\rho}) \quad (7.6)$$

The probability densities  $p_{X_i}$  and  $p_{Y_j}$  can be derived given the particle geometry and the spatial and orientational probability distributions. In Appendix A,  $p_{X_i}$  and  $p_{Y_j}$  are derived assuming the particles have needle-like geometry, are uniformly distributed in space, and are uniformly distributed in orientation. These derivations show that  $X_i \sim \text{Poisson}(m_{X_i})$ , or that  $X_i$  has a Poisson distribution with parameter  $m_{X_i} = \rho_i \alpha_i$ , in which  $\alpha_i$  is a function of the field of view, depth of

field, and the lower and upper bounds of size class  $i$ . Furthermore,  $Y_j \sim \text{Poisson}(m_{Y_j})$ , in which  $m_{Y_j} = \sum_{i=1}^T \rho_i \beta_{ij}$

To extend the analysis to data collected from  $N$  images, we define two new random vectors  $\mathbf{X}_\Sigma$  and  $\mathbf{Y}_\Sigma$  for which  $X_{\Sigma_i} = \sum_{k=1}^N X_{ik}$  and  $Y_{\Sigma_j} = \sum_{k=1}^N Y_{jk}$ . Here, the subscript  $k$  denotes the image index. Given that  $X_{ik} \sim \text{Poisson}(m_{X_i})$ , it can be shown that  $X_{\Sigma_i} \sim \text{Poisson}(Nm_{X_i})$  [12, p. 440]. Likewise,  $Y_{\Sigma_j} \sim \text{Poisson}(Nm_{Y_j})$ .

Differentiating Equation (7.6) with respect to  $\boldsymbol{\rho}$  and equating with zero results in a set of coupled, nonlinear equations for which an analytical solution is not apparent. Equation (7.6) is solved using MATLAB's nonlinear optimization solver FMINCON with initial values obtained from Equation (7.7).

If the border particles are ignored, the estimator reduces to

$$\hat{\boldsymbol{\rho}} = \arg \min_{\boldsymbol{\rho}} \sum_{i=1}^T (-\log p_{X_i}(x_i | \boldsymbol{\rho}))$$

In this case, we can solve for  $\hat{\boldsymbol{\rho}}$  analytically:

$$\hat{\rho}_i = \frac{X_i}{\alpha_i}, \quad i = 1, \dots, T \quad (7.7)$$

The probability density for this estimator can be computed analytically as

$$p_{\hat{\rho}_i}(\tilde{\rho}_i) = p_{\hat{\rho}_i}(X_i / \alpha_i) = p_{X_i}(x_i) \quad (7.8)$$

with  $x_i$  being a non-negative integer. It is straightforward to show that this estimator has the following properties:

$$E[\hat{\rho}_i] = \rho_i$$

$$Var[\hat{\rho}_i] = \rho_i / \alpha_i$$

For the case of multiple images, the maximum likelihood estimate is given by

$$\hat{\rho}_i = \frac{X_{\Sigma_i}}{N\alpha_i}, \quad i = 1, \dots, T \quad (7.9)$$

which has the following properties:

$$\begin{aligned} E[\hat{\rho}_i] &= \rho_i \\ Var[\hat{\rho}_i] &= \rho_i/(N\alpha_i) \end{aligned}$$

#### 7.2.4 Confidence Intervals

Let  $\chi = \{\mathbf{Z}_1, \mathbf{Z}_2, \dots, \mathbf{Z}_N\}$  denote a dataset of  $N$  images, with  $\mathbf{Z}_k = (\mathbf{X}_k, \mathbf{Y}_k)$  containing the data for both border and non-border measurements for image  $k$ . Let  $\mathbf{Z}_1, \mathbf{Z}_2, \dots, \mathbf{Z}_N$  be independent and identically distributed (i.i.d.) with distribution function  $F$ . Let  $\hat{F}$  be the empirical distribution function of the observed data. Let  $R(\chi, F)$  be a random vector giving the PSD estimated using either Miles-Lantuéjoul or maximum likelihood. To construct confidence intervals for the estimated PSD, we require the distribution of  $R(\chi, F)$ . This distribution, called the sampling distribution, is unknown because  $F$  is unknown, being a function of the unknown PSD  $\rho$ . As  $N \rightarrow \infty$ , the limiting distribution of the maximum likelihood estimates is a multivariate normal with mean  $\rho$  and covariance  $\mathbf{I}(\rho)^{-1}$ , where  $\mathbf{I}(\rho)$  is the Fisher information matrix, defined as

$$\mathbf{I}(\rho) = -E[\mathbf{l}''(\rho)]$$

in which  $\mathbf{l}''(\rho)$  is a  $T \times T$  matrix with the  $(i, j)$ th element given by  $\frac{\partial^2 l(\rho)}{\partial \rho_i \partial \rho_j}$ . Approximate confidence intervals for individual parameter estimates can be calculated as  $\rho_i = \hat{\rho}_i \pm \hat{\sigma}_i z_\alpha$  in which  $\hat{\sigma}_i$  is the  $i$ th diagonal element of the observed Fisher information matrix  $-\mathbf{l}''(\hat{\rho})$  and  $z_\alpha$  gives the appropriate quantile from the standard normal distribution for the confidence level  $\alpha$ .

Given that the underlying distributions for  $\mathbf{X}_k$  and  $\mathbf{Y}_k$  are Poisson, we expect the sampling distributions to be non-normal in general. We therefore use bootstrapping [41, p.253] to approximate the distribution of  $R(\chi, F)$  and construct confidence intervals. Let  $\chi^* = \{\mathbf{Z}_1^*, \dots, \mathbf{Z}_N^*\}$  denote a bootstrap sample of the dataset  $\chi$ . The elements of  $\chi^*$  are i.i.d. with distribution function  $\hat{F}$ . In other words,  $\chi^*$  is obtained by sampling  $\chi$   $N$  times where, for each of the  $N$  samples, the probability of selecting the data  $\mathbf{Z}_k$  is  $1/N$ . We denote a set of  $B$  bootstrap samples as  $\chi_l^* = \{\mathbf{Z}_{l1}^*, \dots, \mathbf{Z}_{lN}^*\}$

for  $l = 1, \dots, B$ . The empirical distribution function of  $R(\chi_l^*, \hat{F})$  for  $l = 1, \dots, B$  approximates the distribution function of  $R(\chi, F)$ , enabling confidence intervals to be constructed.

For  $R(\chi, F) = \hat{\rho}$ , the distribution function of  $R(\chi^*, \hat{F})$  is derived analytically using Equation (7.8). For  $R(\chi, F) = \hat{\rho}_b$  and  $R(\chi, F) = \hat{\rho}_{ML}$ , the distribution of  $R(\chi^*, \hat{F})$  is estimated using  $B = 1000$  bootstrap samples. The confidence intervals are obtained using the percentile method, which consists of reading the appropriate quantiles from the cumulative distribution of  $R(\chi^*, \hat{F})$ . To calculate confidence intervals using the normal approximation, the observed Fisher information matrix  $-l''(\rho)$ , also called the Hessian, must be calculated. The  $(i, j)$ th element of this matrix is given by

$$-\frac{\partial l(\rho)}{\partial \rho_i \partial \rho_j} = \frac{X_{\Sigma_i}}{\rho_i^2} \delta_{ij} + \sum_k^T \beta_{jk} \beta_{ik} \frac{Y_{\Sigma_k}}{m_{Y_k}^2} \quad (7.10)$$

### 7.3 Results

To investigate the performance of the maximum likelihood estimator relative to the standard Miles-Lantuéjoul approach, these estimators were applied in several case studies. In each case study, 1000 simulations were carried out. Each simulation consists of generating a set of artificial images and applying the estimators to the particle length measurements obtained from these images. The images were generated using the methods described in Section 4.3. Figure 7.2 shows example images generated for each of these case studies. Each of these images has a horizontal image dimension of  $a=480$  pixels and a vertical dimension of  $b=480$  pixels. The first row displays four simulated images for mono-disperse particles of length  $0.5a$  with  $N_c=25$  crystals per image. The second row shows images of particles uniformly distributed on  $[0.1a \ 0.9a]$  with  $N_c=25$ . The third row shows images of particles normally-distributed with  $\mu = 0.5a$  and  $\sigma = 0.4a/3$  with  $N_c=25$ , and the fourth row shows example images for simulations of particles uniformly-distributed on  $[0.1a \ 2.0a]$  with  $N_c=15$ .

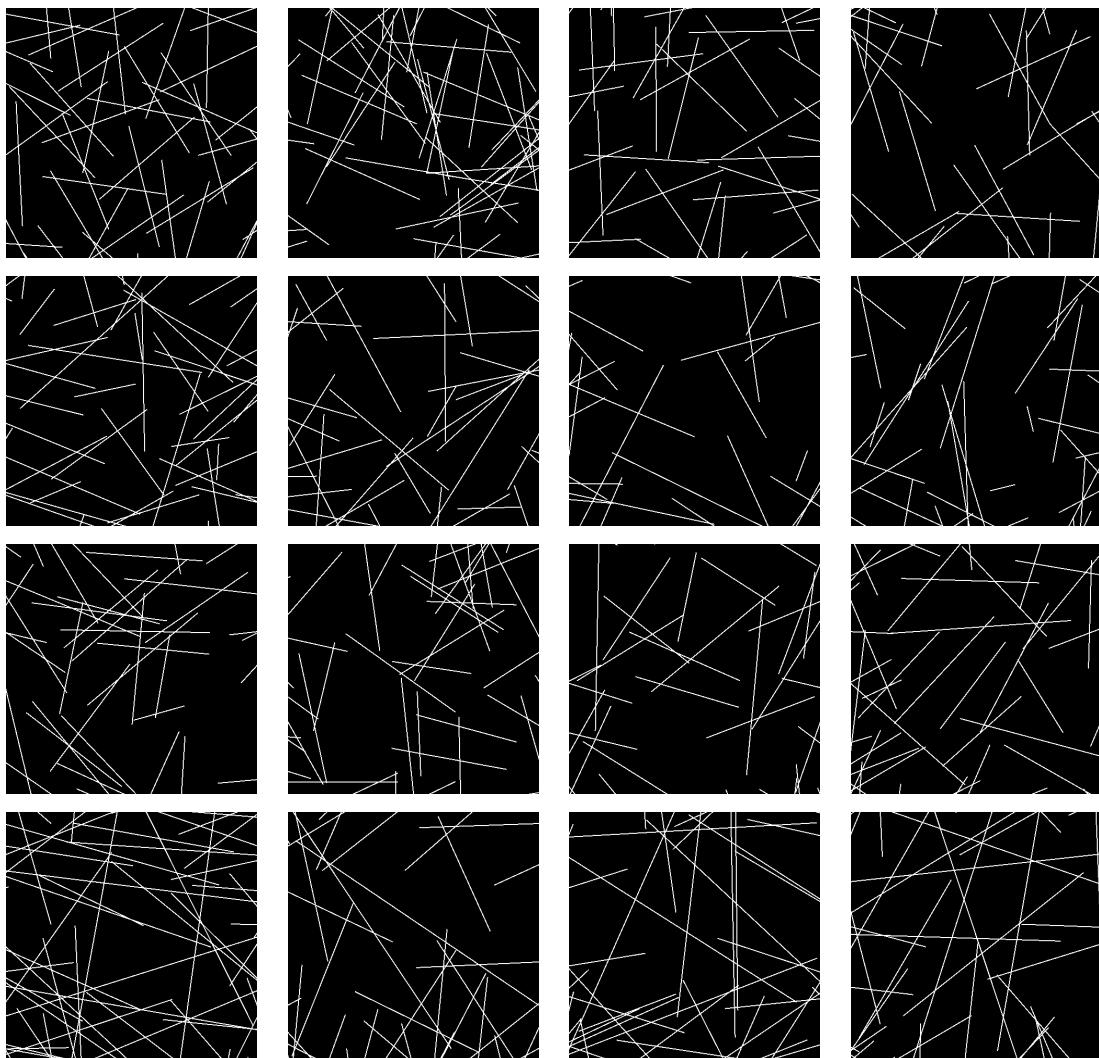


Figure 7.2: Example images for simulations of various particle populations. Row 1: mono-disperse particles of length  $0.5a$ ,  $N_c=25$ . Row 2: particles uniformly distributed on  $[0.1a \ 0.9a]$ . Row 3: particles normally-distributed with  $\mu = 0.5a$  and  $\sigma = 0.4a/3$ ,  $N_c=25$ . Row 4: particles uniformly-distributed on  $[0.1a \ 2.0a]$ ,  $N_c=15$ .

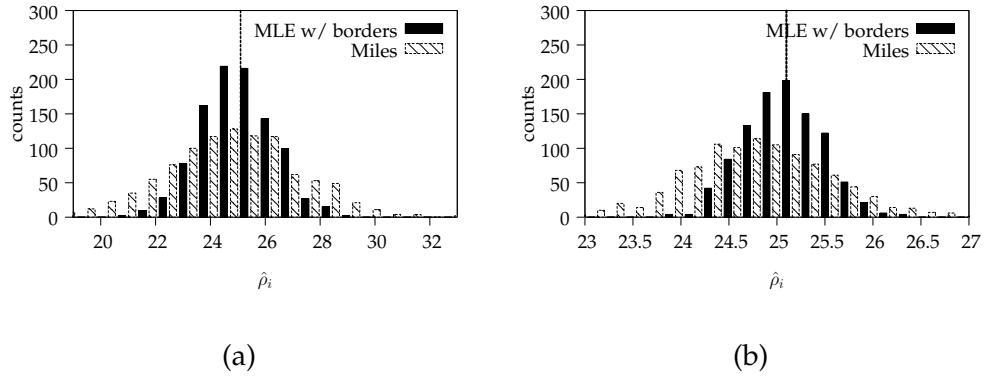


Figure 7.3: Comparison of estimated sampling distributions for absolute PSD for mono-disperse particles. Results based on 1000 simulations, 10 size classes,  $N_c=25$ . (a) Results for 10 images/simulation. (b) Results for 100 images/simulation.

### 7.3.1 Case study 1: mono-disperse particles of length $0.5a$

In the first case study, the particle population consists of mono-disperse particles of length  $0.5a$ . The first row in Figure 7.2 shows example images from these simulations. The length scale is discretized on  $[0.1a \sqrt{2}a]$  into  $T=10$  size classes with the fourth size class centered at  $0.5a$ . The sampling distributions for the various estimators are shown in Figure 7.3 for 1000 simulations using 10 images/simulation and 100 images/simulation. Including the border particle measurements provides better estimates, as evidenced by the lower variance in the sampling distribution for  $\hat{\rho}_b$  relative to the other estimators. As a measure of the improvement gained by including the border particle measurements, we calculate the relative efficiency of  $\hat{\rho}_{b_i}$  versus  $\hat{\rho}_{ML_i}$  for a given size class  $i$  as

$$\text{eff}(\hat{\rho}_{b_i}, \hat{\rho}_{ML_i}) = \frac{\text{MSE}(\hat{\rho}_{ML_i})}{\text{MSE}(\hat{\rho}_{b_i})} \quad (7.11)$$

in which  $\text{MSE}(T) = \text{var}(T) + [\text{bias}(T)]^2 = E[(T - \rho)^2]$  is the mean-squared error for estimator  $T$ .

The MSE is estimated for size class  $i$  as

$$\text{MSE}(T_i) = \frac{1}{n} \sum_{j=1}^n (T_j - \rho_i)^2 \quad (7.12)$$

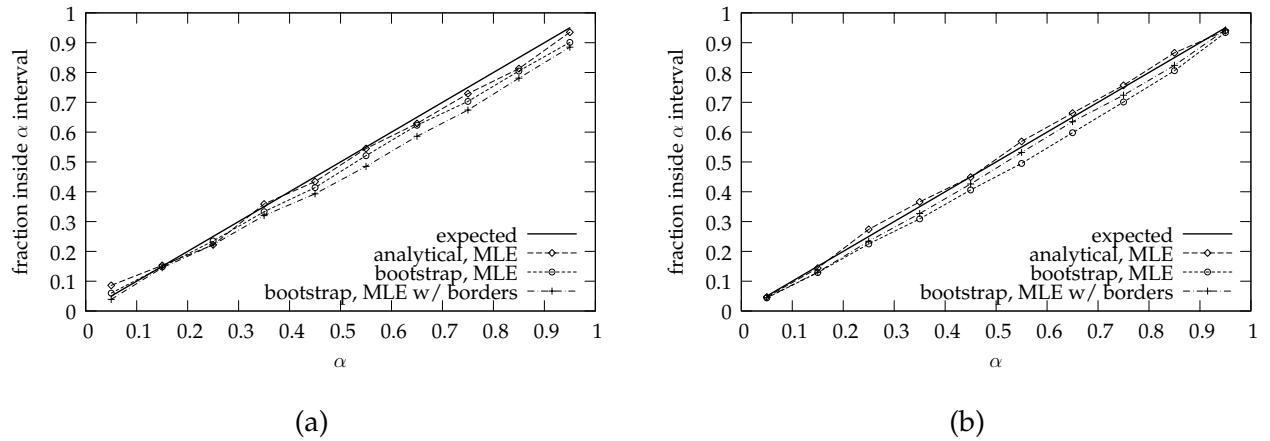


Figure 7.4: Fraction of confidence intervals containing the true parameter value versus confidence level. Results based on 1000 simulations, 10 size classes (results shown only for size class corresponding to mono-disperse particle size),  $N_c = 25$ . (a) Results for 10 images/simulation. (b) Results for 100 images/simulation.

in which  $n$  is the number of simulations. The relative efficiency of the estimators appears relatively independent of the number of images per simulation, with values ranging between 3.5 and 4.0 as the number of images per simulation is varied between 10 and 100. Thus, for this particular case, including the border particle measurements decreases the number of images required to obtain a given accuracy by a factor of about four. For mono-disperse systems in general, we would expect the efficiency to be a monotonically increasing function of particle size.

Figure 7.4 demonstrates the effectiveness of the bootstrap approach for determining confidence intervals. Figure 7.4 is constructed by calculating bootstrap confidence intervals for 1000 different simulations and determining what fraction of these confidence intervals contain the true parameters for a given level of confidence, or  $\alpha$ . The figure shows results for confidence intervals based on the analytical sampling distribution of  $\hat{\rho}$  (Equation (7.8)) as well as sampling distributions estimated using bootstrapping for both  $\hat{\rho}_b$  and  $\hat{\rho}$ . The fraction of confidence intervals containing the true parameter corresponds closely to the expected value (i.e. the confidence level), even for the case of only 10 images/simulation.

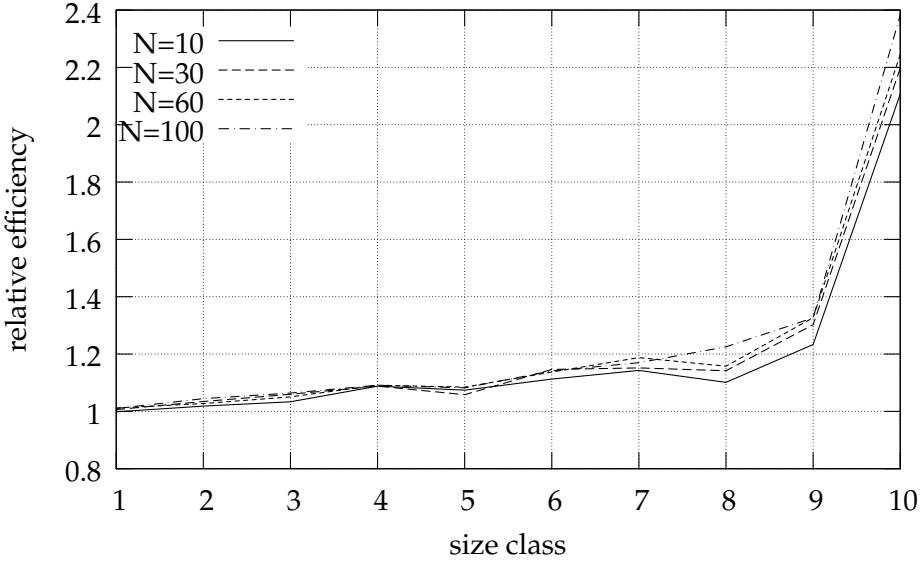


Figure 7.5: Relative efficiencies ( $\text{eff}(\hat{\rho}_b, \hat{\rho}_{ML})$ ) plotted versus size class for various numbers of images per simulation: case study 2.

### 7.3.2 Case study 2: uniform distribution on $[0.1a \ 0.9a]$

In the second case study, the particle population consists of particles uniformly distributed on  $[0.1a \ 0.9a]$ . The second row in Figure 7.2 shows example images from these simulations. The length scale is discretized on  $[0.1a \ 0.9a]$  into  $T=10$  size classes of equal size.

The efficiency of  $\hat{\rho}_b$  relative to  $\hat{\rho}_{ML}$ , calculated using Equation (7.11), is plotted versus size class in Figure 7.5. This plot indicates that including the border particle measurements does not appear to improve the estimation for the lower size classes but results in a significant increase in efficiency for the largest size class.

Figures 7.6 and 7.7 plot the fraction of bootstrap confidence intervals containing the true value of  $\rho_i$  for various size classes based on 100 images/simulation and 10 images/simulation. The bootstrap approach is effective for 100 images/simulation but underestimates the size of the confidence interval for 10 images/simulation. Including the border particle measurements enables better determination of confidence intervals, particularly for the largest size class.

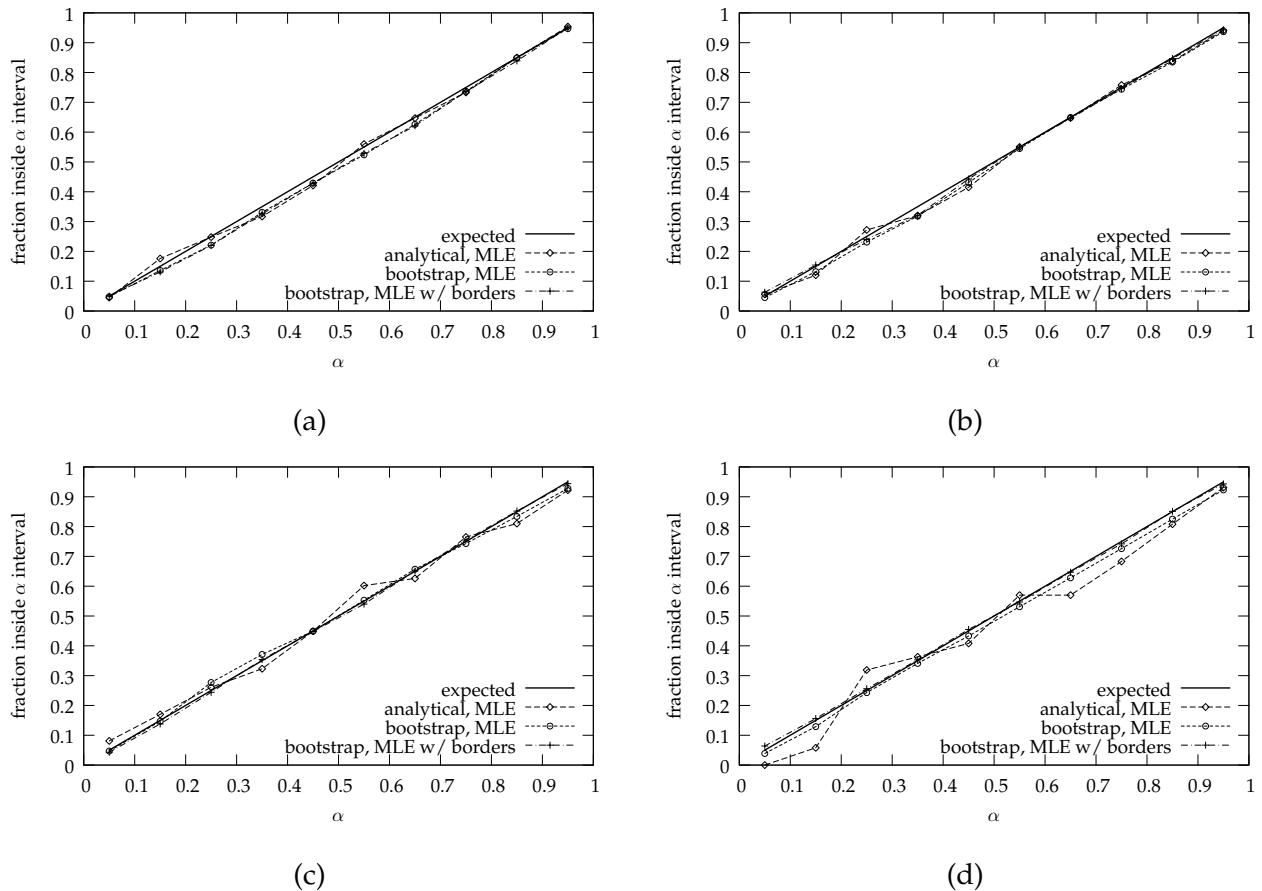


Figure 7.6: Fraction of confidence intervals containing true parameter values for different confidence levels,  $N=100$ . (a) Size class 1 (smallest size class). (b) Size class 4. (c) Size class 7. (d) Size class 10 (largest size class).

### 7.3.3 Case study 3: normal distribution

For the third case study, the particle population consists of particles with lengths distributed as a normal with  $\mu = 0.5a$  and  $\sigma = 0.4a/3$ . The third row in Figure 7.2 shows example images from these simulations. The length scale is discretized on  $[\mu - 3\sigma, \mu + 3\sigma] = [0.1a, 0.9a]$  into  $T=10$  equispaced size classes. Figure 7.8 illustrates the sampling distributions at the various size classes for  $\hat{\rho}_i$ . The x-y plane of Figure 7.8 shows the histogram generated for a normal distribution. The discrete sampling distributions, calculated using Equation (7.8), are plotted for each size class. This figure indicates that the sampling distribution for a given size class can be adequately represented

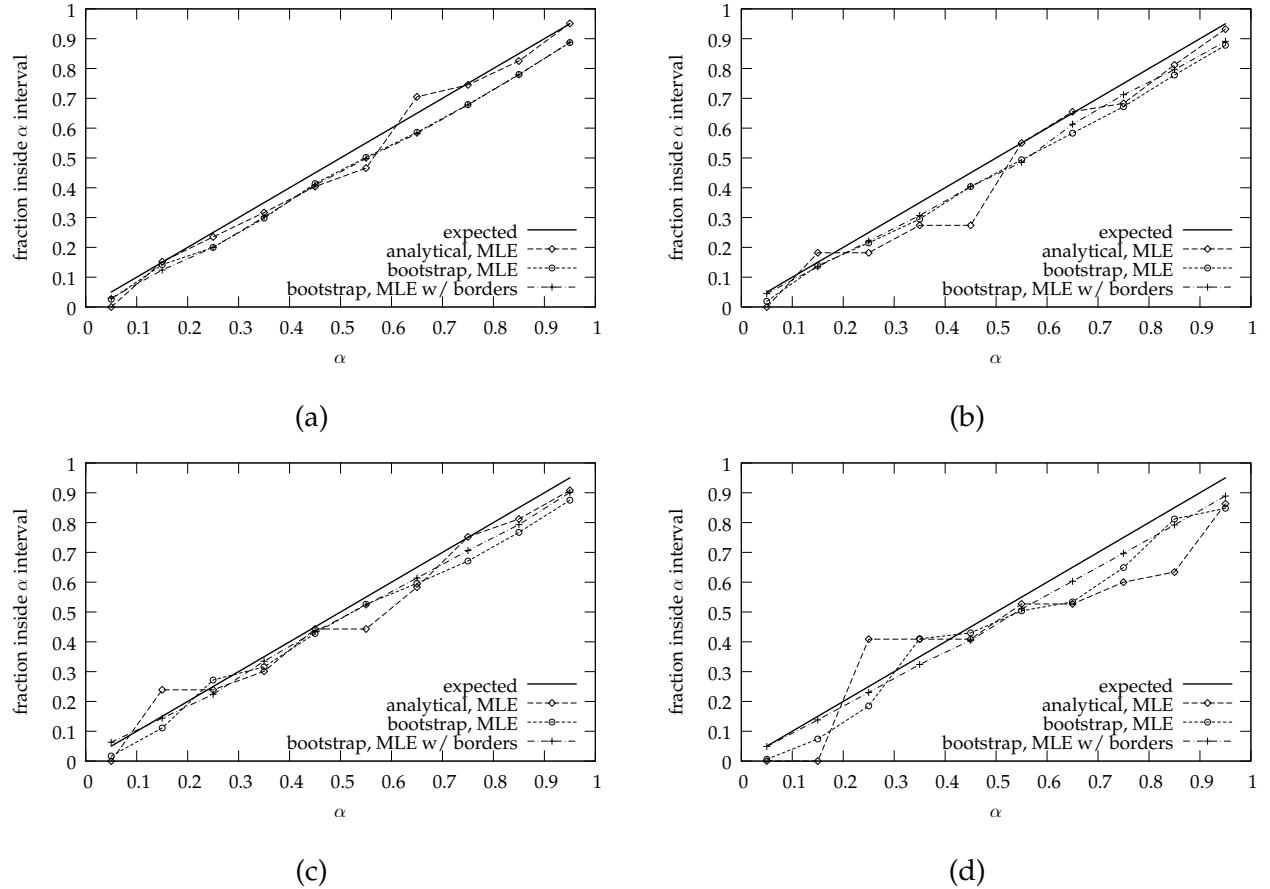


Figure 7.7: Fraction of confidence intervals containing true parameter values for different confidence levels,  $N=10$ . (a) Size class 1 (smallest size class). (b) Size class 4. (c) Size class 7. (d) Size class 10 (largest size class).

by a normal distribution provided the density of particles in that size class is sufficiently high. However, for the larger size classes, approximating the sampling distribution as a normal would lead to inaccurate confidence intervals.

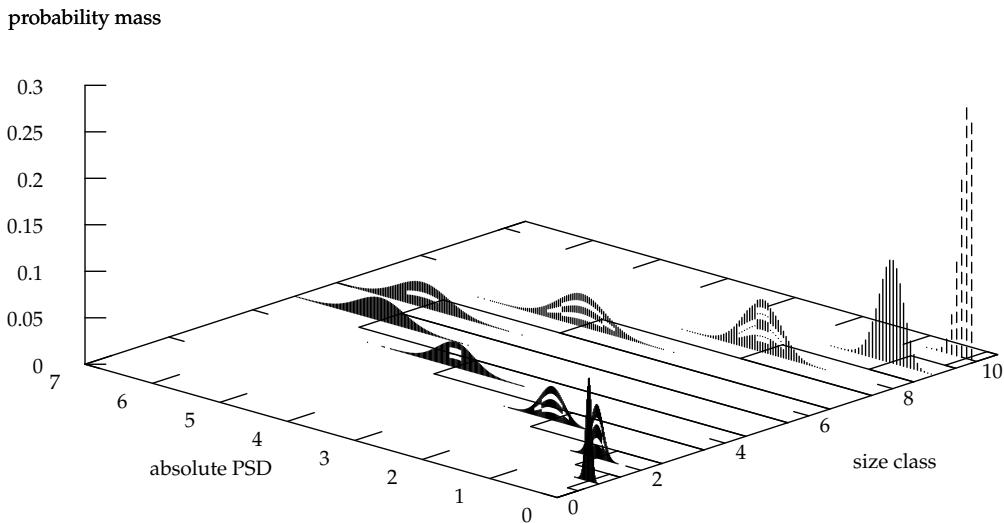


Figure 7.8: Sampling distributions for the various size classes of a discrete normal distribution.  
 $N = 100$ .

Figure 7.9 plots  $\text{eff}(\hat{\rho}_{bi}, \hat{\rho}_{MLi})$  for various numbers of images per simulation. Comparing Figure 7.9 with Figure 7.5 indicates that the relative efficiency for a given size class is a function of both the size and the density of the particles in that size class.

### 7.3.4 Case study 4: uniform distribution on $[0.4a \ 2.0a]$

In the fourth case study, the particle population consists of particles uniformly distributed on  $[0.4a \ 2.0a]$ . The fourth row in Figure 7.2 shows example images from these simulations. The length scale was discretized on  $[0.4a \ a]$  into  $T - 1 = 9$  bins with the  $T$ th bin extending from  $a$  to  $L_{\max}$ .  $L_{\max}$  was assumed unknown and estimated with initial value  $\sqrt{2}a$ . That is, Equation (7.6) was solved as before with the exception that the parameters  $m_{X_i}$  and  $m_{Y_j}$  were updated at each iteration based on the current estimate of  $L_{\max}$ . Figure 7.10 shows the sampling distributions for  $\rho_i$  for

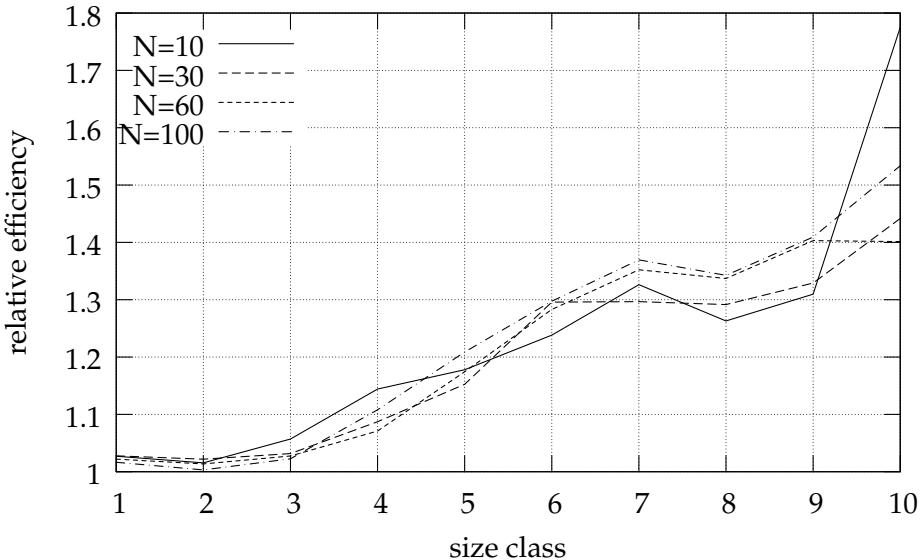


Figure 7.9: Relative efficiencies ( $\text{eff}(\hat{\rho}_b, \hat{\rho}_{ML})$ ) plotted versus size class for various numbers of images per simulation: case study 3.

various size classes, as well as the sampling distribution for  $L_{\max}$ . The MLE w/ borders approach is effective in estimating  $L_{\max}$  and the density of particles in the largest size class. It should be remembered, however, that the estimation is based on the assumption that the particles are uniformly distributed across each size class. This assumption is suitable for finely discretized size classes but is probably not suitable for the single, large, oversized size class. Thus, the estimated value of  $L_{\max}$  should be used only as a rough estimate of the true value and as an indication for the appropriateness of the camera magnification.

## 7.4 Conclusion

The maximum likelihood estimator for imaging-based PSD measurement of zero-width, needle-like particles has been derived using both censored and uncensored observations (i.e. border and non-border particles). The performance of the estimator has been compared with the standard Miles-Lantuéjoul approach using four case studies that highlight several advantages of the MLE

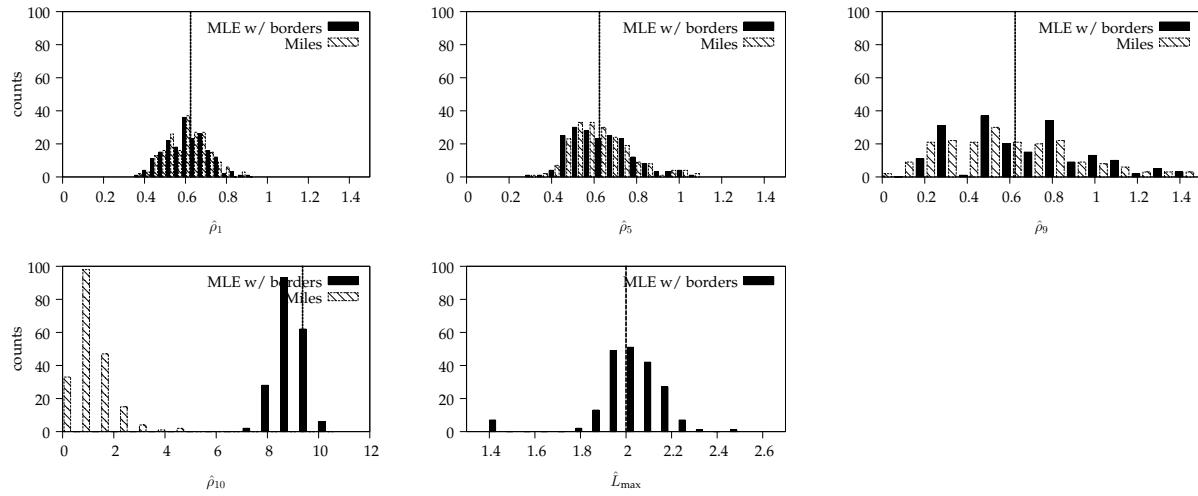


Figure 7.10: Comparison of sampling distributions for absolute PSD for particles distributed uniformly on  $[0.4a \ 2.0a]$ . Results based on 200 simulations, 100 images/simulation, 10 size classes,  $N_c=15$ . (a) Size class 1. (b) Size class 5. (c) Size class 9. (d) Size class 10. (e)  $L_{\max}$ .

approach. The case studies indicate that MLE is more efficient than Miles-Lantuéjoul, particularly if the particle population is mono-disperse or contains particles that are large relative to the size of the image. Furthermore, MLE can estimate the number density of over-sized particles (particles bigger than the image dimension) along with the size  $L_{\max}$  of the largest particle while the Miles-Lantuéjoul approach can be applied only for particles smaller than the image dimension.

The limitations of the MLE approach should also be discussed. The primary limitation of the MLE derived in this chapter is due to the assumption that the particles have needle-like geometry. The Miles-Lantuéjoul approach, on the other hand, can be applied to a much wider class of geometries. Secondly, the MLE approach requires the solution of a nonlinear optimization problem. Thus, confidence interval determination by bootstrapping can be computationally-intensive. Finally, it should be noted that the MLE estimates related to over-sized particles are obtained by making the rather unrealistic assumption that over-sized particles are uniformly distributed in length on  $[a \ L_{\max}]$ . The estimates related to over-sized particles are therefore biased in general but may be useful for identifying whether or not the camera magnification is suitable for the given system.

Several areas for future work are evident. Choosing the optimal number, location, and size of bins for constructing histograms should be addressed. Integrating measurements taken at multiple scales or magnifications is also important. For systems of high-aspect-ratio particles, incorporating the width of border particles into the estimation could lead to increased efficiency by narrowing down the number of size classes to which a border particle may correspond.

## Chapter 8

# Assessing the Reliability of Imaging-based, Number Density Measurement<sup>1</sup>

The methods for PSD estimation in Chapter 7 are based on the assumption that image segmentation is perfect, or that every single particle appearing in the image is identified correctly. Achieving perfect image segmentation is a realistic assumption, however, only at low solids concentrations. At higher solids concentrations, a significant fraction of the particles may be overlapping or occluded, which often results in these particles being missed by the automated image analysis algorithm. The density of particles is thus underestimated. The objective of this chapter is to address this third challenge by developing a semi-empirical, probabilistic model that enables PSD estimation for imperfect image analysis. This paper also presents a descriptor that can be used to quantify the expected amount of particle overlap and thereby assess the reliability of the PSD estimate.

The chapter is organized as follows. Section 8.1 presents previous work relevant to the scope of this chapter. Section 8.2 presents theory related to particle overlap probabilities, proposes a descriptor for measurement reliability, and presents an estimator for particle number density that accounts for particle overlap. Section 8.3 describes the simulation studies used to determine

---

<sup>1</sup>Portions of this chapter are to appear in Larsen and Rawlings [66]

the conditions for which SHARC gives an accurate PSD measurement. Section 8.4 presents the results of these simulation studies, and Section 8.5 summarizes our findings.

## 8.1 Previous work

Armitage [4] was one of the first to investigate the effects of particle overlap on particle number density estimation. Armitage derived formulas for the expected numbers of isolated clumps (groups of one or more overlapping particles) for circular and rectangular particles, and used these formulas to estimate the mean clump size and number density of particles. Mack [77, 78] extended Armitage's results to three-dimensional particles of any convex shape, deriving formulas for the expected numbers of clumps and isolated particles based on the perimeters and areas of the two-dimensional projections of the particles. Roach [104] summarized the work of Mack and Armitage, described several applications, and developed theory for specialized applications. Kellerer [63] was the first to account for edge effects and overlap simultaneously and derived a formula for the expected number of clumps minus the number of enclosed voids (i.e. regions of the image completely enclosed by particles). To model the complicated, random shapes that form due to the random placement and overlap of simple geometric objects, the Boolean model [1, 6] can be used.

The methods cited above assume that a clear distinction can be made between the image background and the objects of interest, or that the image can be segmented by simple thresholding. In some applications, however, more advanced image analysis is required, and modeling the output of a complicated image analysis algorithm to enable statistical inference is non-trivial. For example, the SHARC algorithm described in Chapter 5 is a model-based vision algorithm that can successfully identify overlapped particles provided the degree of overlap is minor. Initial studies have shown that the methods cited above do not adequately model the output of the SHARC algorithm. In this work, we extend these methods to enable statistical inference for SHARC's output. We expect the methodology developed here to be applicable to other situations in which advanced

image analysis is necessary to extract information from noisy images.

## 8.2 Theory

### 8.2.1 Particulate system definition

Consider a slurry  $S$  of volume  $V$  in which a solid phase of discrete particles is dispersed in a continuous fluid phase. Let  $\rho_c$  be the solid phase density (mass solid/volume solid) and  $M_T$  the slurry density (mass solid/volume slurry), or solids concentration. The volume solids concentration (volume solid/volume slurry) is given by  $M_T/\rho_c$ . Let  $L$  be the characteristic length of a particle and define a shape factor  $k_v$  such that the volume of a single particle is given by  $k_v L^3$ . Let  $f(L)$  denote the PSD, or the number of particles of characteristic length  $L$  per unit volume slurry. The  $i$ th moment of the PSD is denoted  $\mu_i = \int_0^\infty f(L)L^i dL$ . Given the PSD, the slurry density can be calculated as  $M_T = \rho_c k_v \mu_3$  assuming  $k_v$  is independent of length. The zeroth moment,  $\mu_0$ , equals the number of particles per unit volume slurry. For a mono-disperse system in which all particles have the same length  $l$ ,  $\mu_0 = M_T / \rho_c k_v l^3$ .

### 8.2.2 Sampling and measurement definitions

Let  $V_I \in S$  denote an imaging volume, and let  $I$  denote an image created by perspective projection of  $V_I$  onto a two-dimensional image plane. Let  $a$  and  $b$  denote the horizontal and vertical dimensions of  $V_I$ , or the field of view, and let  $d_f$  denote the depth dimension of  $V_I$ , or the depth of field. Thus, the volume of  $V_I$  is  $abd_f$ , and the average number of particles per image is  $\lambda = \mu_0 abd_f$ .

Number-based PSDs are typically measured by discretizing the characteristic length scale into  $T$  non-overlapping bins or size classes. We therefore define the discrete PSD as

$$\rho_i = \int_{S_i}^{S_{i+1}} f(l) dl, \quad i = 1, \dots, T \quad (8.1)$$

in which  $S = (S_1, \dots, S_{T+1})$  is the vector of breaks between size classes. In this work, we consider only mono-disperse populations, so we represent the PSD for a given system using a single scalar

value  $\rho$ . For a mono-disperse system,  $\rho$  equals the number density.

### 8.2.3 Descriptor for number density reliability

Our goal is to estimate the number density of particles, and our hypothesis is that the quality or reliability of the estimate correlates with the amount of particle overlap. We therefore want to quantify the amount of particle overlap observed in the acquired images. We can calculate the probability that a given particle is overlapped by other particles as follows. Consider a population of  $n$  identical particles randomly located within the slurry system  $S$  of volume  $V$ . The number density of particles is given by  $n/V$ . Let the particles be projected orthogonally onto a plane of area  $A_S$ , giving a particle density per unit area of  $\rho_A = n/A_S$ . Let  $K$  be a random variable giving the number of times a given particle's projection is overlapped by other particles' projections. Assuming  $A_S$  is sufficiently large that edge effects are negligible, the probability that the projection of a given particle is overlapped by the projection of a second, given particle is  $p_{\text{ovp}} = \Omega/A_S$ , in which  $\Omega$  denotes the admissible area, or the area of the region inside which the second particle's projection overlaps the first particle's projection. Thus, the probability that a given particle is overlapped by  $k$  particles is given by the binomial distribution with  $n$  Bernoulli trials with the probability of success in each trial given by  $p_{\text{ovp}}$ ,

$$p_K(k) = \binom{n}{k} p_{\text{ovp}}^k (1 - p_{\text{ovp}})^{n-k} \quad (8.2)$$

As  $n \rightarrow \infty$  with  $\rho_A = n/A_S$  constant, the binomial distribution converges to the Poisson distribution with parameter  $\rho_A \Omega$ .

$$p_K(k) = \frac{e^{-\rho_A \Omega} (\rho_A \Omega)^k}{k!} \quad (8.3)$$

Thus, the probability that a given particle is completely isolated is given by  $p_{\text{iso}} = p_K(0) = \exp(-\rho_A \Omega)$ .

As already mentioned, we expect the reliability of the PSD estimate to correlate with particle overlap. Therefore, as an indicator of the reliability of the PSD estimate, we define the param-

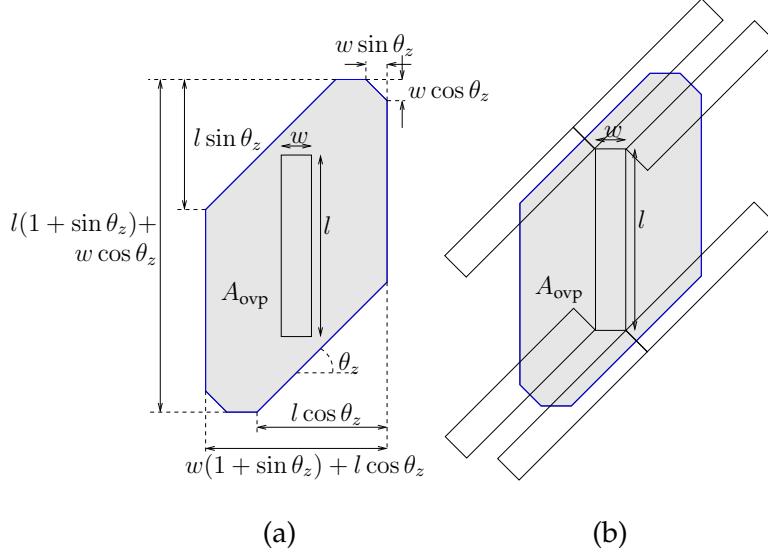


Figure 8.1: Geometric representation of admissible area, or region in which a particle is overlapped by another particle.

eter  $D$  as

$$D = -\log(p_{\text{iso}}) = \rho_A \Omega \quad (8.4)$$

The area number density  $\rho_A$  of particles in an image can be calculated as  $\rho_A = \mu_0 d_f$ . The admissible area  $\Omega$  depends on the particle geometry. To illustrate how the admissible area can be calculated, consider the rectangular particle of length  $l$  and width  $w$  in the center of Figure 8.1. Next, consider a second particle of identical dimensions placed in the image with orientation  $\theta_z$ . If the midpoint of the second particle is placed anywhere inside the shaded region in Figure 8.1, the first and second particles overlap. This fact is illustrated in Figure 8.1(b) using the various particles placed around the border of the shaded area. Using straightforward geometric relationships (see Figure 8.1(a)), the area  $A_{\text{ovp}}$  of this shaded region can be shown to be

$$A_{\text{ovp}} = (w(1 + \sin \theta_z) + l \cos \theta_z)(l(1 + \sin \theta_z) + w \cos \theta_z) - l^2 \cos \theta_z \sin \theta_z - w^2 \cos \theta_z \sin \theta_z \quad (8.5)$$

and  $\Omega$  is obtained by integrating over  $\theta_z$

$$\Omega = E[A_{\text{ovp}}] = \frac{1}{\pi} \int_{-\pi/2}^{\pi/2} A_{\text{ovp}} d\theta_z = \frac{2}{\pi} (l^2 + w^2 + lw(2 + \pi)) \quad (8.6)$$

in which a uniform distribution in orientation has been assumed.

Mack [77] (see also Roach [104, p.44]) derived a more general result that enables the calculation of the admissible area  $\Omega$  for any set of convex bodies of identical size and shape and having random orientation. Mack's surprisingly simple result gives  $\Omega$  based on the area  $a_p$  and perimeter  $s_p$  of a two-dimensional domain:

$$\Omega = 2a_p + \frac{s_p^2}{2\pi} \quad (8.7)$$

#### 8.2.4 Estimation of number density

Given our hypothesis that particle overlap is the primary cause of failure for image analysis-based measurement, it seems reasonable to estimate the number density of particles based on the number of completely isolated particles observed in the image. Letting  $X$  be a random variable giving the number of observations of completely isolated, non-border particles, it can be shown that the probability density for  $X$  is Poisson with parameter  $m_X = \rho d_f N A_{NB} \exp(-\rho d_f \Omega)$ , in which  $N$  is the number of images,  $A_{NB}$  is the area of the region inside which a particle does not touch the image border (see Chapter 7 and Appendix A),  $\Omega$  is the admissible area defined above, and  $\rho$  is the number density as defined above. The maximum likelihood estimate of  $\rho$  is therefore

$$\hat{\rho} = \arg \max_{\rho} \frac{e^{-m_X} m_X^x}{x!} \quad (8.8)$$

in which  $x$  is the realization of  $X$ , or the measured number of isolated, non-border particles.

The likelihood function in Equation (8.8) gives the probability of observing the data as a function of  $\rho$  and has unique properties that merit consideration. Consider a population of particles having lengths 1/10 the size of the image dimension and a number density corresponding to about 10 particles per image. Under these conditions, the amount of particle overlap is small, so the actual number of particles is essentially equal to the number of isolated, non-overlapping particles. However, it is also likely that only 10 isolated particles would be observed if the number density of particles is so high that nearly all particles are overlapping. Thus, in Figure 8.2, two spikes are observed in the likelihood function: one corresponding to the low concentration case

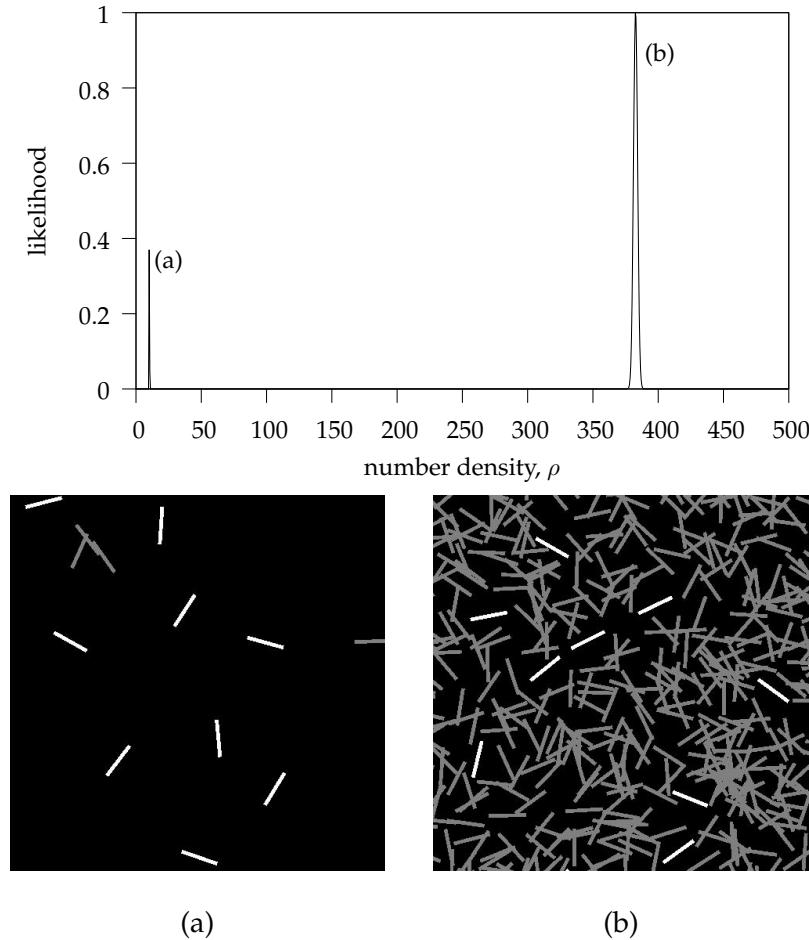


Figure 8.2: Likelihood of observing  $n$  non-overlapping particles with example images at low and high number densities giving the same number of non-overlapping particles. Overlapping particles appear gray while non-overlapping particles appear white.

when the total number of particles is close to the number of isolated particles (corresponding to the example image in Figure 8.2(a)), and one corresponding to the high concentration case (see example image in Figure 8.2(b)). To determine which maxima corresponds to the true value of  $\rho$ , additional information is required that makes it clear whether the system is in the low number density regime or high number density regime. Kellerer [63], for example, incorporates the number of voids (i.e. regions completely surrounded by particles) to calculate the expected number of clumps of particles. The additional information available depends on the type of image analysis

algorithm used. The algorithm used in this study is based on the identification of lines in the image, so we use the number of lines to determine which of the two peaks to choose. The correct peak is defined as the peak corresponding to a value of  $\rho$  that predicts most nearly the number of lines identified by SHARC. A rough prediction of the number of lines is given by  $\hat{n}_l(\rho) = 4\rho d_f A_I$ , in which  $A_I = ab$  is the image area.

The estimator given by Equation (8.8) is correct if the image analysis algorithm identifies only isolated particles. Model-based image analysis algorithms, however, are designed to identify the objects of interest even in the presence of overlap or occlusion. The number of particles observed by a model-based image analysis algorithm is therefore greater than the number of isolated particles, and Equation (8.8) can be expected to give bad estimates given such data. Assuming the number of particles identified by any given image analysis algorithm depends primarily on the amount of overlap, a reasonable model can be formulated based on the overlap model given above. Letting  $\tilde{X}$  denote the number of particles identified by an image analysis algorithm, a reasonable form for the probability density of  $\tilde{X}$  would be Poisson with parameter  $m_{\tilde{X}} = \rho d_f N A_{NB} \exp(-\rho d_f \Omega \theta)$ , in which  $\theta$  is an empirical parameter and the other variables are as defined previously.

### 8.3 Image analysis methods summary

To determine the conditions under which reliable measurements can be obtained using image analysis, we applied the SHARC algorithm described in Chapter 5 to artificial images generated at various solids concentrations  $M_T$  and levels of overlap  $D$ . The images were generated using the methods described in Section 4.3 with the parameters given in Table 8.1. For simulations at a given solids concentration  $M_T$ , the expected number of particles per image  $\lambda$  is calculated as

$$\lambda = \frac{M_T ab d_f}{\rho_c \int_0^\infty p(L) k_v L^3 dL}$$

Description	Symbol	Value
Horizontal dimension of imaging volume	$a$	2 mm
Vertical dimension of imaging volume	$b$	2 mm
Depth of field	$d_f$	0.25 mm
Solid phase density	$\rho_c$	2 mg/mm <sup>3</sup>
Number of horizontal CCD pixels	$u_{\max}$	480
Number of vertical CCD pixels	$v_{\max}$	480

Table 8.1: Parameters used to simulate imaging of particle population at a given solids concentration.

For a mono-disperse population with  $p(L) = \delta(L - l)$ ,  $\lambda$  is given by

$$\lambda = \frac{M_T ab d_f}{\rho_c k_v l^3}$$

For simulations at a given value of  $D$ ,  $\lambda$  is calculated using Equation (8.4):

$$\lambda = \frac{Dab}{\Omega}$$

The SHARC algorithm consists of three main steps. In the first step, SHARC identifies lines in the image corresponding to the crystals' edges. Because crystal edges are often broken due to noise or overlap, SHARC attempts to complete these edges by identifying instances of collinearity between the identified lines. Finally, SHARC identifies clusters of parallel lines having similar lengths and fits a two-dimensional rectangle model to each cluster, giving the length and width of the crystal. SHARC was applied to the artificial images using the parameters shown in Table 8.2.

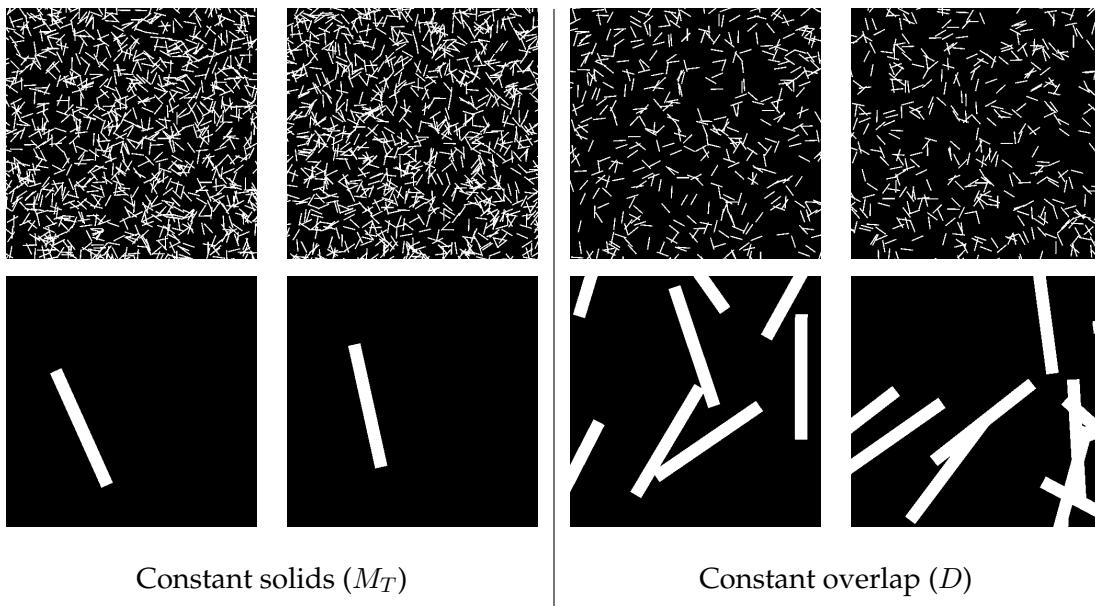
## 8.4 Results

### 8.4.1 Descriptor comparison: $M_T$ versus $D$

Figure 8.3 compares images generated for two different mono-disperse particle populations generated at the same solids concentration  $M_T$  and at the same  $D$ . The images generated at constant

Line finder		Collinearity		Parallelism	
Parameters		Thresholds		Thresholds	
$n_\nabla$	5	$\epsilon_{\theta_C}$	20 degrees	$\epsilon_{\theta_P}$	5 degrees
$\epsilon_{ \nabla }$	1	$\epsilon_{EP}$	0.5	$\epsilon_Q$	0.85
$n_b$	6 buckets	$\epsilon_{PD}$	0.5	$\epsilon_{AR}$	6.0
$\epsilon_A$	20 pixels				

Table 8.2: Parameter values used to analyze artificial images of overlapping particles.

Figure 8.3: Comparison of images generated for two different mono-disperse particle populations at the same solids concentration  $M_T$  and at the same  $D$ . The top row shows the images for particles with length 1/20th of the image dimension and the bottom row shows images for particles with length one-half the image dimension. The aspect ratio of the particles is 10.

$D$  appear to be more similar with respect to the amount of particle overlap than the images generated at constant solids concentration. This qualitative assessment is confirmed by Figure 8.4, which plots the average number of overlaps per crystal for images simulated at constant  $D$  and at constant solids concentration  $M_T$  for mono-disperse populations of various crystal sizes and aspect ratios. The figure shows that, at a given solids concentration, the average number of over-

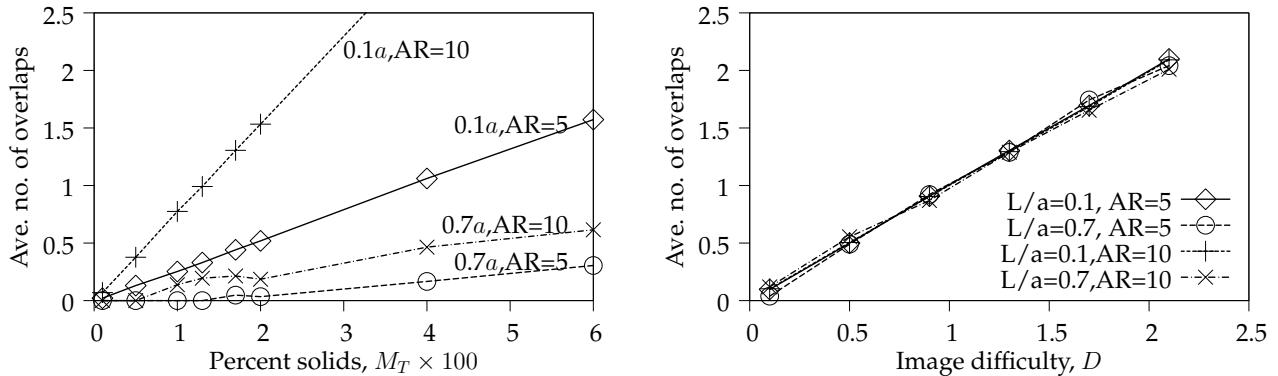


Figure 8.4: Comparison of average number of overlaps per crystal for images simulated at constant  $D$  and at constant solids concentration for mono-disperse populations of various crystal sizes.

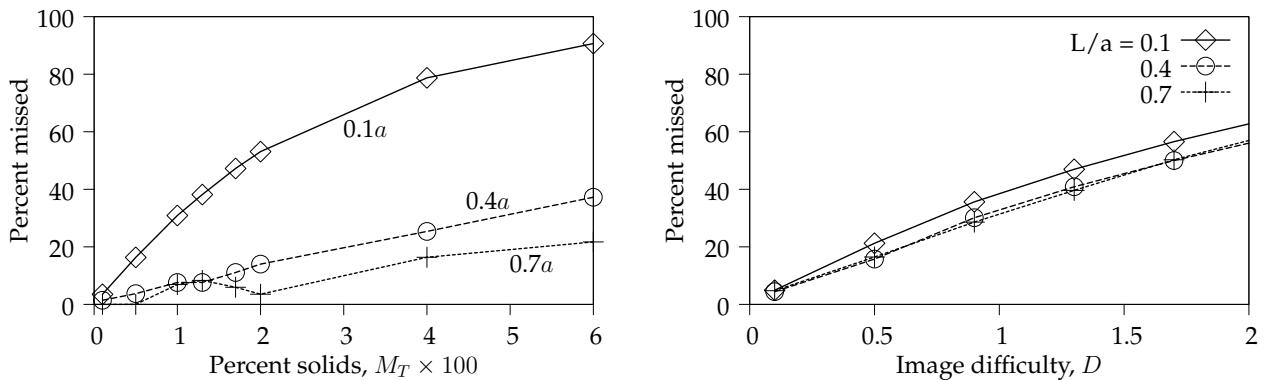


Figure 8.5: Comparison of percentage of particles missed by automated image analysis for images simulated at constant  $D$  and at constant solids concentration for mono-disperse populations of various crystal sizes.

laps per crystal is a strong function of the particle size and shape while, at a given level of  $D$ , the number of overlaps is independent of particle size and shape. If measurement failure is caused by particle overlap, we would expect the results of automated image analysis at a given  $D$  to be relatively independent of the size and shape of the particles. Figure 8.5 shows the percentage of particles missed by automated image analysis at different levels of solids concentrations and  $D$ . As expected, the percent missed is similar for various particle sizes when considered in terms of  $D$ , but is vastly different when considered in terms of solids concentration.

### 8.4.2 Estimation of number density

In this section, we examine different methods for estimating the number density  $\rho$ . We first examine the behavior of the Miles-Lantuéjoul estimator, which corrects for edge effects but not for particle overlap. Next, we examine the performance of the maximum likelihood estimator presented in Section 8.2.4. The estimators are applied to data acquired by analyzing images using SHARC. Example images for various levels of  $D$  and three different particle sizes ( $L/a = 0.1, 0.3,$  and  $0.5$ ) are shown in Figure 8.6.

#### Miles-Lantuéjoul method

Figure 8.7 shows the ratio of the estimated and true number density as a function of  $D$  for various particle sizes. In this figure, the number density of particles is estimated using the Miles-Lantuéjoul method [83, 64], which does not account for overlap. The particle size and shape measurements are obtained using the SHARC algorithm. The estimator's bias increases with  $D$ , or as the amount of overlap increases. Given the well-behaved and relatively size-independent correlation between  $\hat{\rho}/\rho$  and  $D$ , one may attempt to correct  $\hat{\rho}$  based on an estimated value of  $D$ . The result of such an approach is shown in Figure 8.7(b). The number density estimates for this figure are given by

$$\hat{\rho} = \frac{\rho_{ML}}{g(\hat{D})}$$

in which  $\rho_{ML}$  denotes the Miles-Lantuéjoul estimate,  $\hat{D}$  is an estimate of  $D$  given by  $\hat{D} = \rho_{ML}\Omega$ , and  $g$  is an empirical function generated by fitting  $g(D) = \exp(-\theta D)$  to the data in Figure 8.7(a). The parameter  $\theta$  is determined using nonlinear least-squares minimization. This approach is ineffective because  $D$  cannot be estimated independently of  $\rho$ , so both  $D$  and  $\rho$  are underestimated.

#### Maximum likelihood method

To implement the maximum likelihood estimator given by Equation (8.8), we first need to complete the probabilistic model by determining the empirical parameter  $\theta$ . We determine  $\theta$  by solving

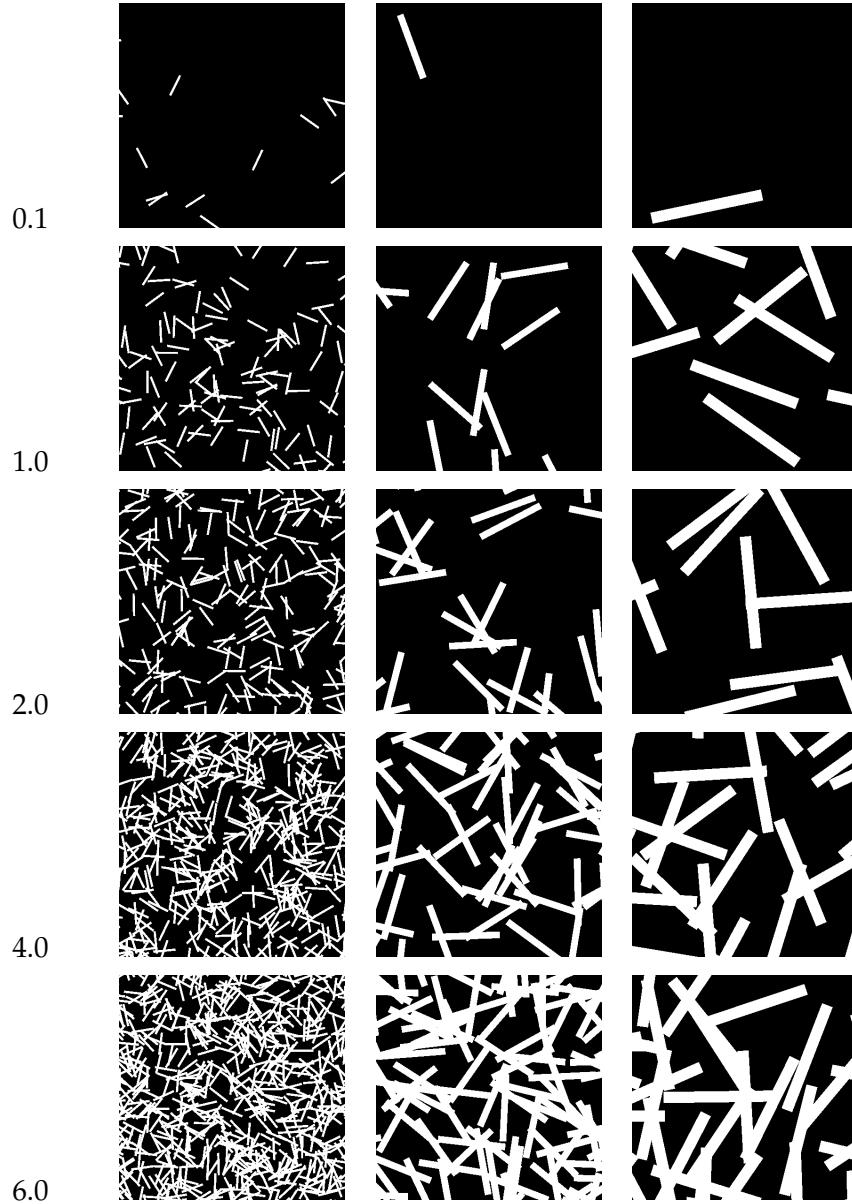


Figure 8.6: Examples of synthetic images generated at various  $D$ . The first, second, and third column correspond to particle sizes  $L/a = 0.1, 0.3$ , and  $0.5$ , respectively.

the nonlinear least squares minimization problem given by

$$\hat{\theta} = \arg \min_{\theta} \sum_{i=1}^{n_d} \left( \frac{\tilde{x}_i}{N} - h(\rho_i, \theta) \right)^2$$

subject to  $0 \leq \theta \leq 1$

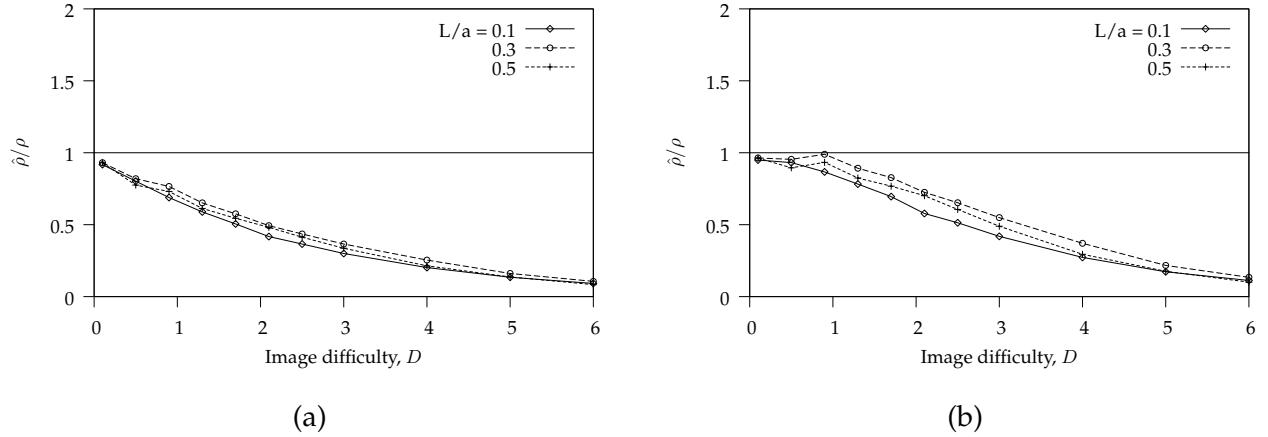


Figure 8.7: Results of number density estimation using Miles-Lantuéjoul method for various particle sizes and various levels of image difficulty. (a) Results for applying Miles-Lantuéjoul method directly to image analysis output. (b) Results for empirical correction of the Miles-Lantuéjoul estimate based on estimated value of  $D$ .

in which  $\tilde{x}_i$  is the number of particles identified by analysis of  $N$  images acquired at a given number density  $\rho_i$ , and  $h(\rho, \theta)$  is the model prediction of the average number of particles identified per image, given by

$$h(\rho, \theta) = \rho d_f A_{NB} \exp(-\rho d_f \Omega \theta) \quad (8.9)$$

A value of  $\theta = 1$  indicates that the image analysis algorithm finds only particles that are completely isolated while a value of  $\theta = 0$  indicates the image analysis algorithm identifies every particle in the image, regardless of overlap.

Figure 8.8 shows the optimal fit of the model in Equation (8.9) to the data observed by applying SHARC to artificial images generated at various number densities with particles of size  $L/a = 0.1$ . The values of  $D$  corresponding to the number densities range from 0.1 to 6.0. Figure 8.8 indicates that the empirical overlap model with  $\theta = 0.33$  gives an excellent prediction of SHARC's behavior for a wide range of densities. Also shown in Figure 8.8 is the model prediction for  $\theta = 0$ , corresponding to perfect image analysis (i.e. the image analysis algorithm identifies every single particle in the image). The slope of this line equals  $A_{NB}$ , which is related to the magnitude of the

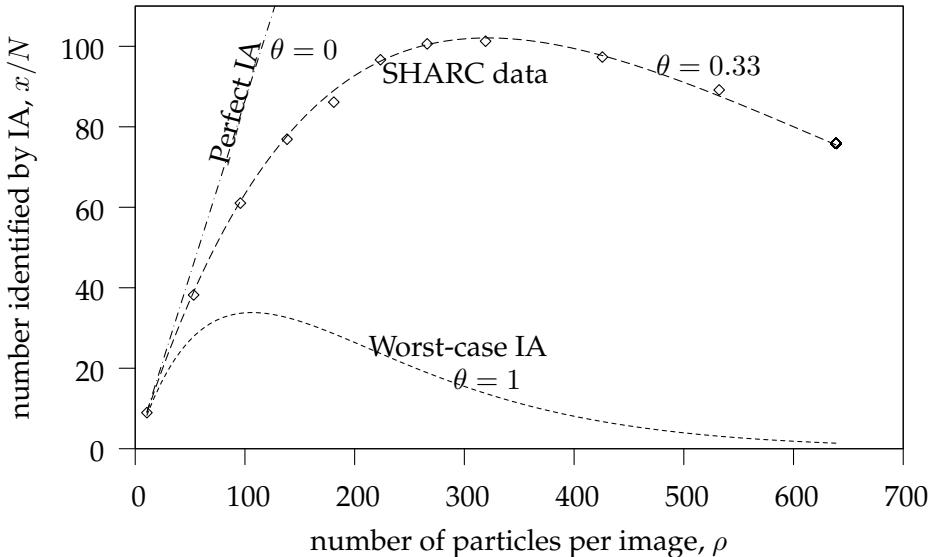


Figure 8.8: Data and model prediction for number of particles with length  $\leq 0.1a$  identified per image by automated image analysis. Also shown is the expected number of completely isolated particles (Equation (8.9) with  $\theta = 1$ ) and the expected number of non-border particles (Equation (8.9) with  $\theta = 0$ ).

edge effects as discussed in Section 8.2.4. The worst-case prediction in Figure 8.8 corresponds to  $\theta = 1$ , indicating that the image analysis algorithm identifies only particles that are completely isolated. This figure indicates that  $\theta$  can be used to evaluate and compare image analysis algorithms in terms of their ability to identify particles in the presence of overlap. The most effective algorithms will have a value of  $\theta$  approaching zero.

Unfortunately, a single value of  $\theta$  cannot be used to predict the image analysis behavior for all particle sizes, at least for the SHARC algorithm. For particles of sizes  $L/a = 0.3$  and  $L/a = 0.5$ , the optimal fits for the image analysis data correspond to  $\theta$  values of 0.21 and 0.14, respectively, as shown in Figure 8.9. The decreasing  $\theta$  values with increasing particle size indicate that SHARC's ability to detect overlapping particles improves as the size of the particles (relative to the image size) increases. Thus, SHARC is not an entirely scale-independent algorithm. The ability to detect a target object in an image regardless of the object's scale is a desirable feature

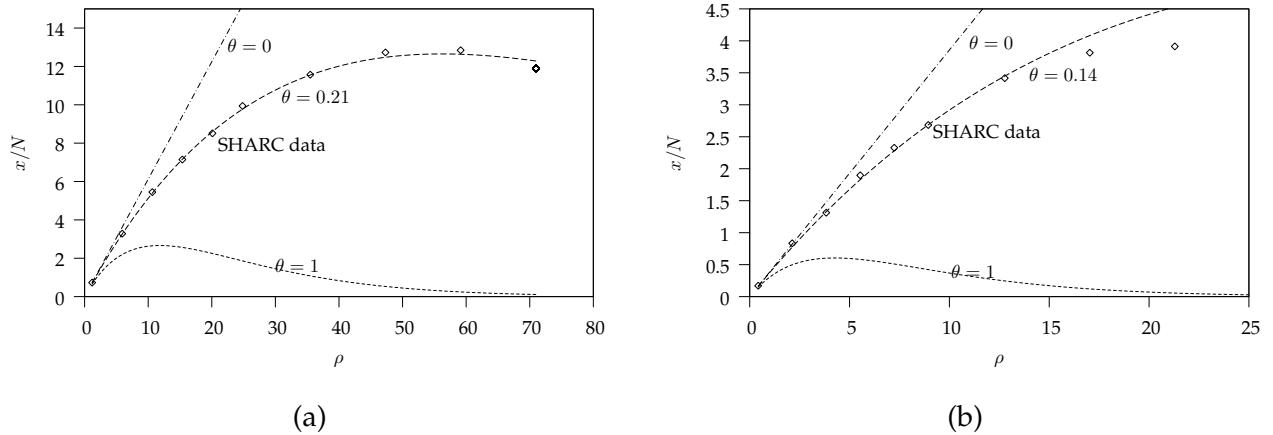


Figure 8.9: Data and model prediction for number of particles with lengths  $\leq 0.3a$  or  $\leq 0.5a$  identified per image by automated image analysis. Also shown is the expected number of completely isolated particles (Equation (8.9) with  $\theta = 1$ ) and the expected number of non-border particles (Equation (8.9) with  $\theta = 0$ ). (a)  $L/a = 0.3$ ,  $\hat{\theta} = 0.21$ ; (b)  $L/a = 0.5$ ,  $\hat{\theta} = 0.14$ .

of any object recognition algorithm. It should be noted also that  $\theta$  depends on both the image analysis algorithm parameters and on how the user filters the image analysis algorithm results. The results shown in Figures 8.8 and 8.9 correspond to the image analysis parameters given in Table 5.2 with any particle having a length less than or equal to the target length (e.g.  $L/a \leq 0.1$  for Figure 8.8) being counted.

Given the empirical parameter  $\theta$ , the maximum likelihood estimator in Equation (8.8) enables effective estimation of  $\rho$  for a variety of conditions, as shown in Figure 8.10. The confidence intervals shown in this figure are obtained using the percentile bootstrapping method [41]. The estimates of  $\rho$  for  $L/a = 0.5$  degrade for large  $D$ .

## 8.5 Conclusion

A practical approach has been developed for assessing the reliability of number density estimates obtained using imaging measurements in the presence of particle overlap. The single dimensionless parameter  $D$  correlates with the measurement reliability based on the amount of particle

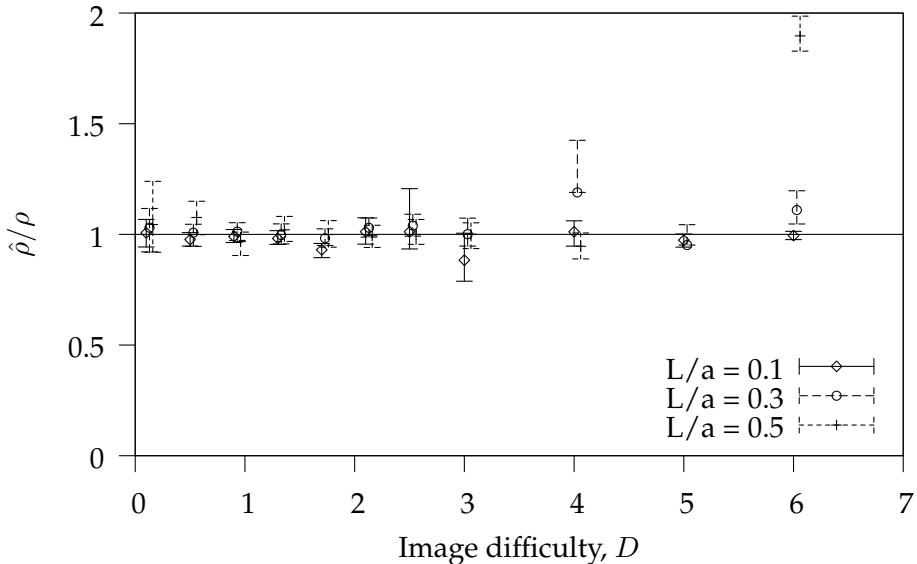


Figure 8.10: Ratio of estimated number density and true number density versus image difficulty using SHARC data and empirical correction factors calculated for each different particle size.

overlap. Thus, the parameter  $D$  can be used to estimate the errors in the measurements and to aid practitioners in determining the sampling conditions necessary to obtain reliable measurement of particle number density.

It has been shown that the Miles-Lantuéjoul estimator, which accounts for edge effects but not particle overlap, underestimates the number density. A maximum likelihood estimator that accounts for both edge effects and particle overlap has been presented. The estimator is based on a semi-empirical model of the probability that a given particle is correctly identified by automatic image analysis. For a given particle size, a single empirical parameter is sufficient to enable effective number density estimation for a wide range of conditions, particularly for systems in which the particles' dimensions are significantly smaller than the image dimensions. The model also provides a convenient tool for comparing different image analysis algorithms in terms of their ability to handle particle overlap.

Various issues should be addressed in future work. Most importantly, an extension of the methods developed in this chapter should be developed for polydisperse systems. The discus-

sions in Armitage [4] and Roach [104, p.46] may provide a good starting point for such investigations. The incorporation of these methods into a state estimator for feedback control of number density should also be investigated.

## **Chapter 9**

# **High-resolution PSD Measurement for Industrial Crystallization <sup>1</sup>**

As discussed in Section 2.3.1, the imaging-based, particulate population dynamics studies in the literature have focused on monitoring either the normalized PSD (i.e. number fraction), moments of the PSD (e.g. mean size and variance), the particle length cumulative distribution function (CDF), or the polymorphic fraction. Given that the dynamics of particle populations are modeled in terms of the absolute PSD, it is desirable to monitor the absolute PSD to enable better model identification and control. The objective of this chapter is to demonstrate the feasibility of imaging-based PSD measurement for realistic and changing process and imaging conditions using the methods developed in previous chapters. We achieve this objective by simulating a well-studied industrial crystallization process and generating images of the process based on the specifications of a commercially-available *in situ* imaging probe. Furthermore, this chapter shows the value of imaging-based measurement by demonstrating the measurement of process properties that cannot be measured by conventional technologies. Section 9.1 describes the crystallizer model used in this study and the method for generating images corresponding to the current state of the crystallizer. Section 9.2 presents the simulation results, demonstrating the measurement of important product-quality metrics and multi-modal PSDs. Section 9.2 also offers general guidelines for designing imaging-based PSD monitoring systems. Section 9.3 summarizes our conclusions.

---

<sup>1</sup>Portions of this chapter are to appear in Larsen and Rawlings [67]

## 9.1 Crystallizer model and imaging summary

Batch crystallization can be modeled using a system of partial integro-differential equations that couples mass and energy balances with a population balance describing the evolution of the crystal population's PSD. The population balance used in this study is

$$\frac{\partial f}{\partial t} + G \frac{\partial f}{\partial L} = 0 \quad (9.1)$$

in which  $f(L, t)$  is the PSD,  $L$  is the characteristic length, and  $G$  is the crystal growth rate. The boundary condition is

$$f(0, t) = \frac{B^\circ}{G(L=0)} \quad (9.2)$$

in which  $B^\circ$  is the nucleation rate density.  $G$  and  $B^\circ$  depend on the relative supersaturation  $\hat{S}$  and are assumed to follow standard semi-empirical power laws:

$$\hat{S} = \frac{\hat{C} - \hat{C}_{\text{sat}}}{\hat{C}_{\text{sat}}} \quad G = k_g \hat{S}^g \quad B^\circ = k_b \hat{S}^b \mu_i^j \quad (9.3)$$

in which  $\hat{C}$  is the liquid-phase solute concentration,  $\hat{C}_{\text{sat}}$  is the saturation concentration,  $k_g$  and  $g$  are growth rate constants, and  $k_b$ ,  $b$ , and  $j$  are nucleation rate constants. The  $i$ th moment of the PSD,  $\mu_i$ , is defined as

$$\mu_i = \int_0^\infty f L^i dL \quad (9.4)$$

In this study, we also are concerned with moments corresponding to subsets of the total crystal population. Moments corresponding to crystals grown from nuclei are denoted with a subscript  $N$  (e.g.  $\mu_{N_3}$ ) while moments corresponding to crystals grown from seeds are denoted with a subscript  $S$ .

Assuming, as above, that growth rate is size-independent and the system is closed (no input or output streams), the mass balance for the solute concentration is

$$\frac{d\hat{C}}{dt} = -3\rho_c k_v \hat{h} G \int_0^\infty f L^2 dL \quad (9.5)$$

in which  $\hat{C}$  is the solute concentration,  $\rho_c$  is the crystal density,  $k_v$  is a shape factor defined such that  $k_v L^3$  gives the volume of a crystal of characteristic length  $L$ ,  $\hat{h}$  converts solvent mass to slurry

Description	Symbol	Value	Units
kinetic growth rate constant	$k_g$	0.3e-4	cm/min
power-law exponent	$g$	2.0	dimensionless
kinetic nucleation rate constant	$k_b$	2.6e10	g solvent/cm <sup>3</sup> min
power-law exponent	$b$	3.0	dimensionless
moment exponent	$j$	2.0	dimensionless
crystal density	$\rho_c$	1.183	g / cm <sup>3</sup>
volumetric shape factor	$k_v$	0.93	dimensionless
area shape factor	$k_a$	9.8	dimensionless
Crystallizer volume	$V$	2.3	L

Table 9.1: Parameters used to simulate industrial batch crystallization process of photochemical [81]. The characteristic length is the particle width and an aspect ratio of 10 is assumed.

volume.  $\hat{C}$  is given in terms of mass of solute per total mass of solution or liquid phase. The initial condition is given by  $\hat{C} = \hat{C}_0$ .

For the simulations used in this study, the energy balance is unnecessary as we assume perfect temperature control. The temperature follows a fixed temperature trajectory without deviation.

The model parameters used in this study correspond to a well-studied, industrial photochemical crystallization process [81, 82]. The parameters for this process are given in Table 9.1. The particles have a needle-like morphology and the characteristic length corresponds to the particle width. An aspect ratio of 10 is assumed. The saturation concentration  $\hat{C}_{\text{sat}}$  for this system is given by

$$\hat{C}_{\text{sat}}(T) = 0.185 - 2.11 \times 10^{-2}T + 7.46 \times 10^{-4}T^2 \quad (9.6)$$

The crystallizer model is solved using orthogonal collocation on moving finite elements. The method is described in detail in Chapter 3.

Description	Symbol	Value
Horizontal dimension of imaging volume	$a$	$1075 \mu\text{m}$
Vertical dimension of imaging volume	$b$	$850 \mu\text{m}$
Depth of field	$d_f$	$10 \mu\text{m}$
Number of horizontal CCD pixels	$u_{\max}$	1360
Number of vertical CCD pixels	$v_{\max}$	1024
Micron to pixel ratio	$m$	0.8

Table 9.2: Parameters used to simulate imaging of particle population using industrial video imaging probe.

The images corresponding to a given crystallizer state are generated as described in Chapter 4. The imaging parameters used in this study correspond to the specifications of a commercially-available *in situ* video probe and are given in Table 9.2.

## 9.2 Results

This section presents simulation results illustrating the capabilities and limitations of imaging-based PSD measurement for industrial crystallization processes. First, we present the process and imaging simulation results for the batch crystallization process discussed in Section 9.1. Next, we analyze the data from these simulations to demonstrate the measurements obtainable by imaging. Specifically, we demonstrate both number and weight-based PSD monitoring as well as product quality measurements that are difficult to obtain using alternative monitoring technologies. We also discuss some of the limitations of current image analysis technology. Finally, we give general considerations for the design of imaging-based PSD monitoring systems.

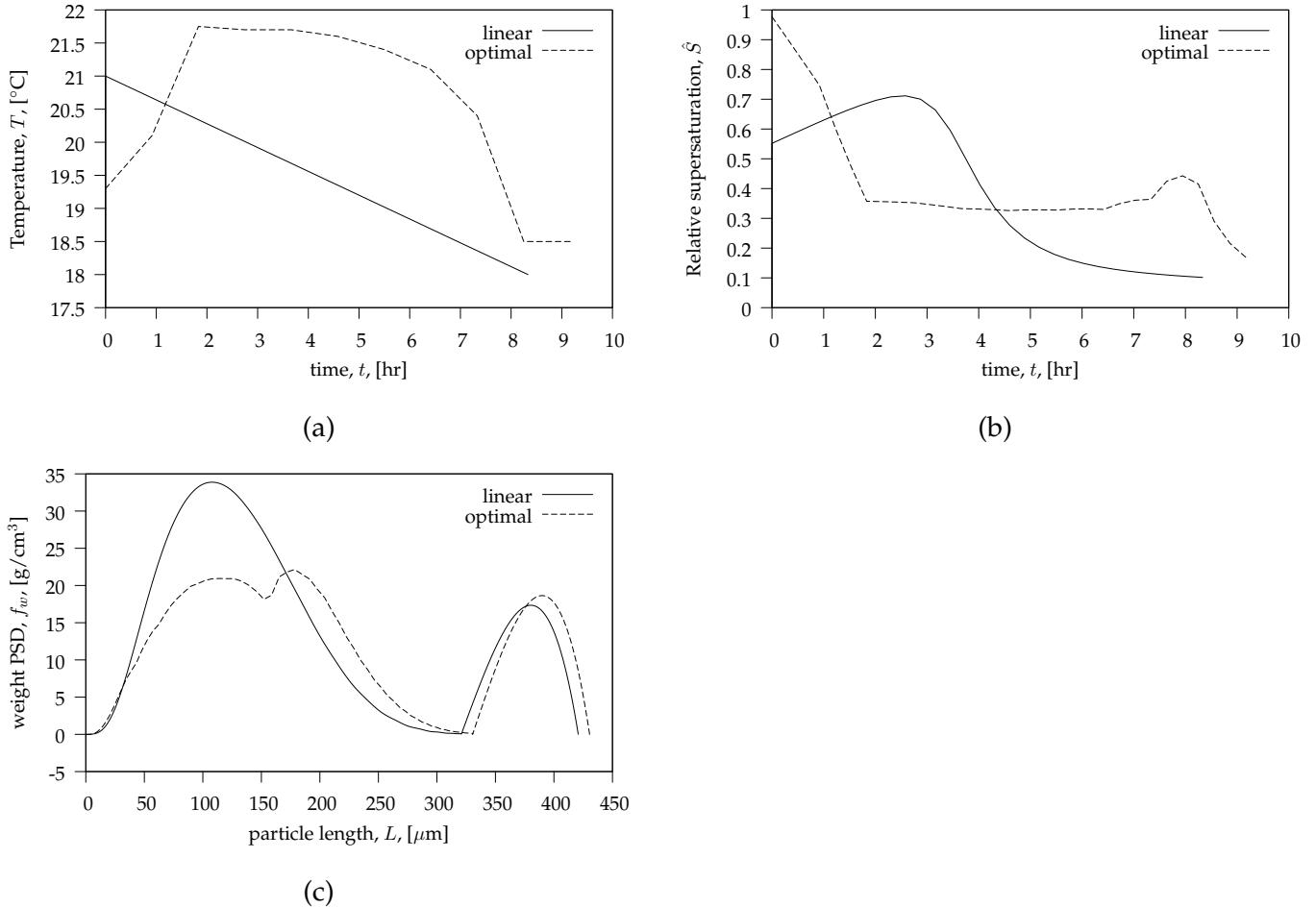


Figure 9.1: Comparison of simulation results for optimal and linear temperature trajectories. (a) Temperature profiles. (b) Relative supersaturation profiles. (c) End-point PSDs.

### 9.2.1 Process and imaging simulations

In previous studies of the photochemical system considered here, an optimal cooling profile was determined that minimized the ratio of the nucleus-grown and seed-grown crystal mass. The temperature and relative supersaturation profiles associated with the optimal cooling profile are compared against a linear cooling profile in Figure 9.1. This figure also shows the final weight PSDs resulting from the linear and optimal cooling profiles. The optimal profile uses high supersaturation at the beginning to maximize crystal growth. Despite the high supersaturation, secondary nucleation is negligible because it depends not only on supersaturation but also on the amount of

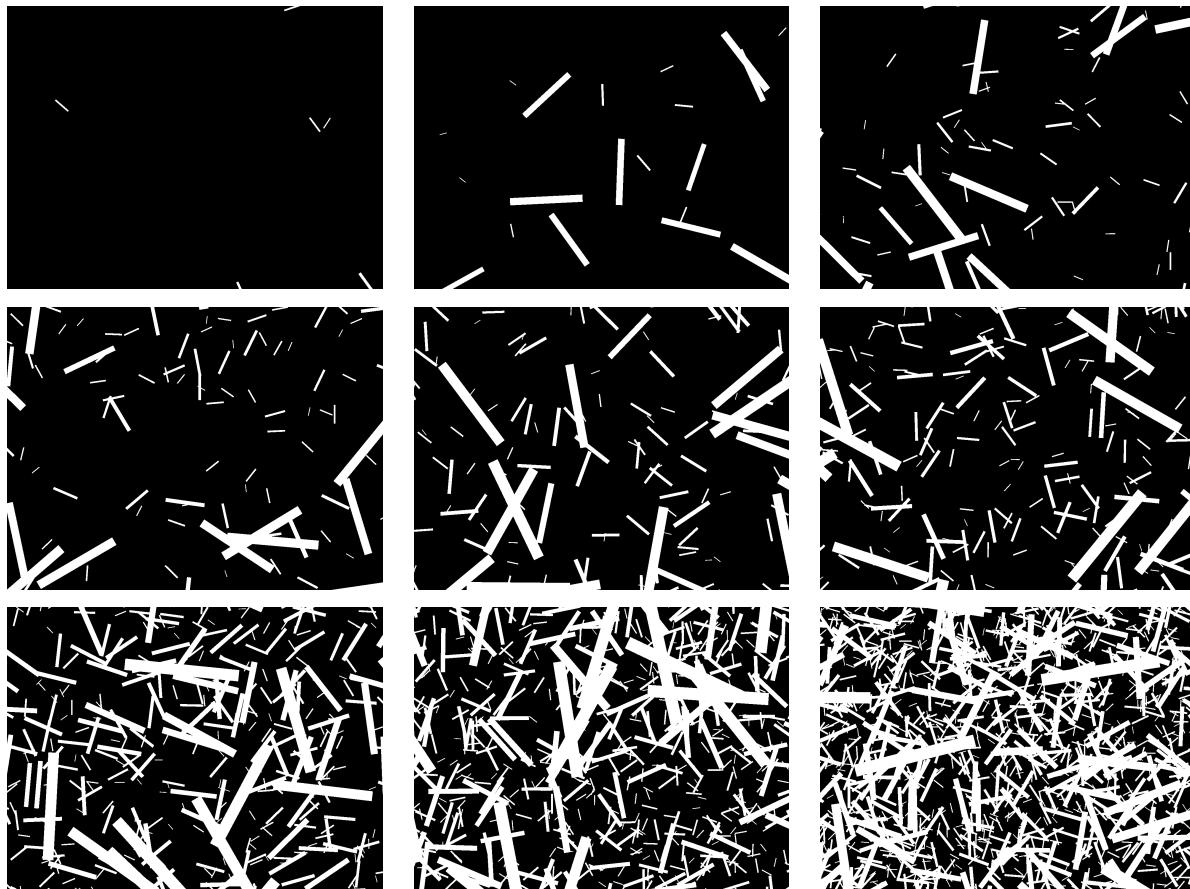


Figure 9.2: Examples of images generated at various times during optimal cooling simulation. The images correspond to 60-minute intervals from 0 hours (upper left) to 8 hours (lower right).

crystal mass present in the crystallizer. As the amount of crystal mass increases, the crystallizer is heated to reduce supersaturation and minimize secondary nucleation. The rapid cooling at the end of the batch results in both crystal growth and significant generation of nuclei, but the batch is terminated before these nuclei acquire appreciable mass.

Figures 9.2 and 9.3 show examples of artificial images corresponding to different times during the optimal and linear cooling processes. These artificial images simulate the images that would be obtained by a commercially available *in situ* video microscopy probe.

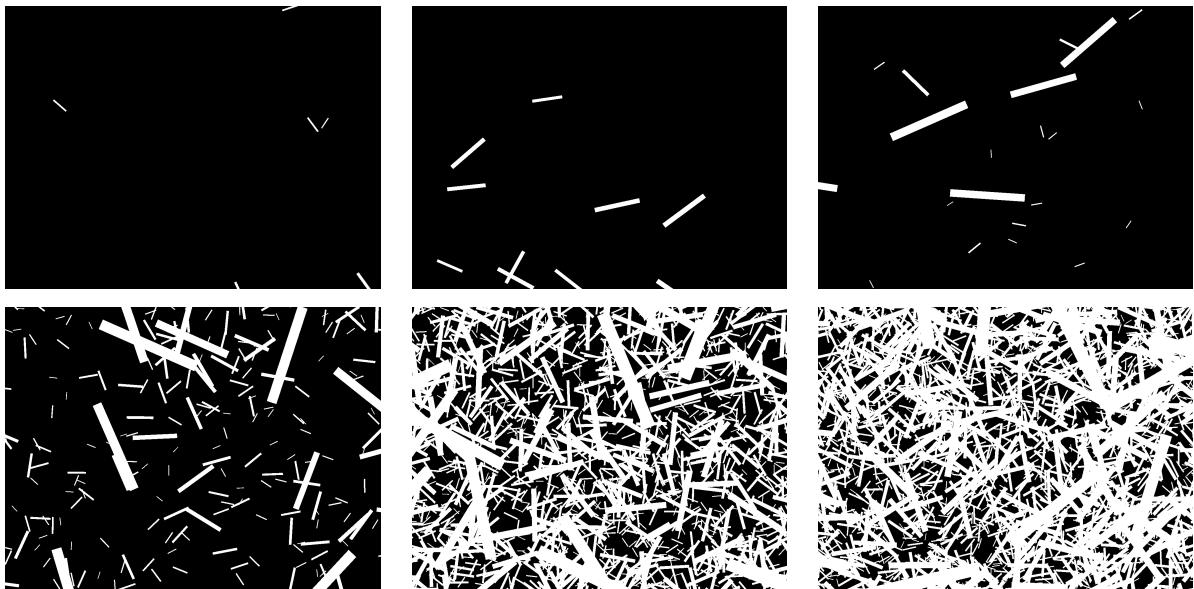


Figure 9.3: Examples of images generated at various times during linear cooling simulation. The images correspond to 60-minute intervals from 0 hours (upper left) to 5 hours (lower right).

### 9.2.2 Absolute PSD measurement

Figures 9.4 and 9.5 show, respectively, the number-based and weight-based PSDs corresponding to the optimal cooling profile. The continuous curve shows the simulated PSD and the histograms show the estimated PSD based on imaging measurements. These figures show that high-resolution ( $10 \mu\text{m}$ ) PSD measurement is achievable using only 100 images.

Accurate PSD estimation requires not only sufficient sampling, but also effective image analysis. For example, Figure 9.5 indicates that, assuming perfect image analysis, 100 images provides sufficient samples to estimate the weight PSD for the given process conditions. Figure 9.6, on the other hand, shows the PSD estimated from image analysis data generated using the SHARC algorithm described in Chapter 5. This figure indicates that about half-way through the experiment, the image analysis method becomes ineffective, causing measurement failure. The failure of the image analysis algorithm is due to particle overlap. Different image analysis algorithms have different tolerances for overlap, with model-based algorithms typically tolerating more overlap than other methods. Based on the images in Figure 9.2, one could reasonably expect improve-

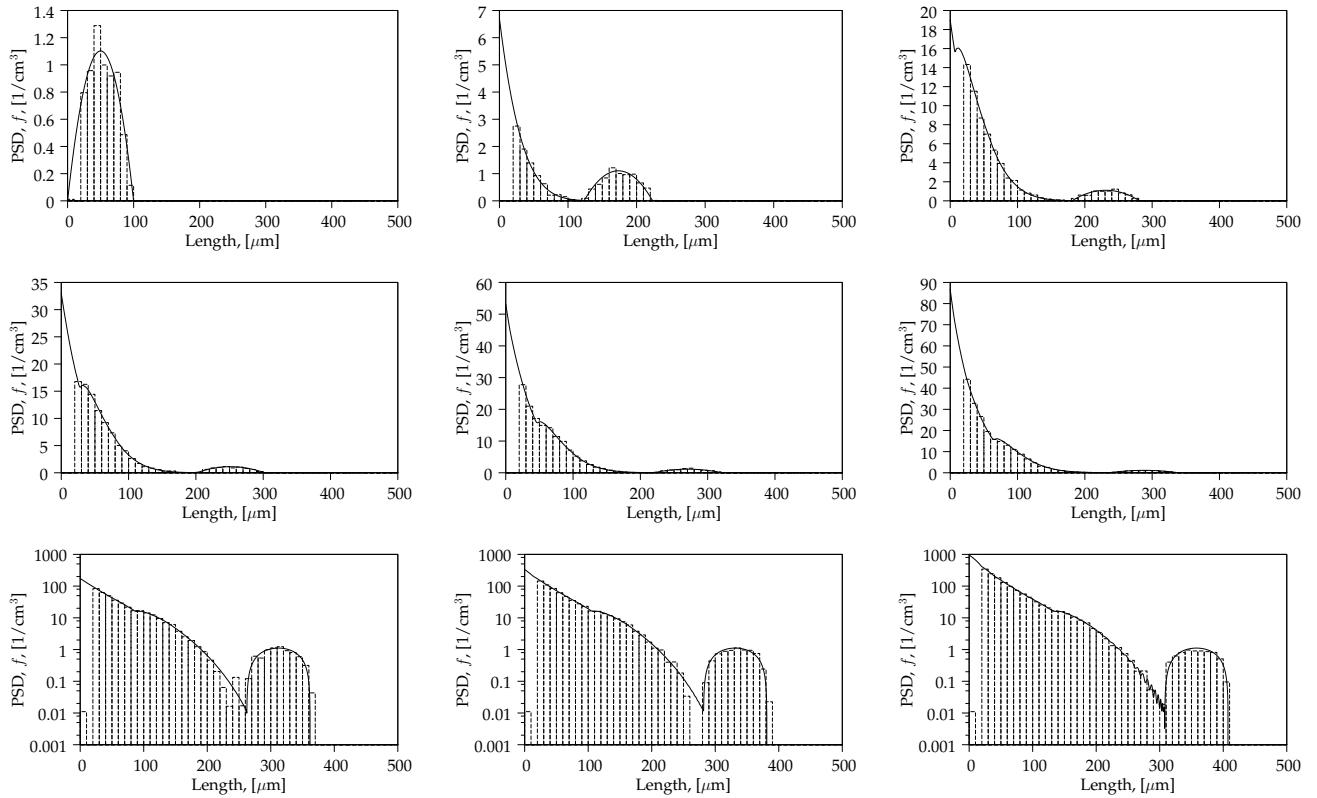


Figure 9.4: Evolution of measured and estimated number-based PSD for optimal cooling and perfect image analysis. Snapshots shown from  $t = 0$  min. to  $t = 500$  min. at 60 minute intervals. Bin size =  $10 \mu\text{m}$  and  $N = 100$ .

ments in model-based image analysis algorithms to enable effective measurement for almost the entire duration of the process. On the other hand, as exemplified by the final image in Figure 9.3, particle overlap can be so excessive that effective image analysis is unachievable.

### 9.2.3 Measurements for product quality

Product quality is often assessed based on the PSD. A variety of product quality metrics are used in the crystallization optimal control literature (see Ward et al. [126] for a summary of this literature). The metrics involve both lower (number-based) and higher (weight-based) moments of the PSD, as well as moments specific to the seed-grown crystals and nucleus-grown crystals. In the

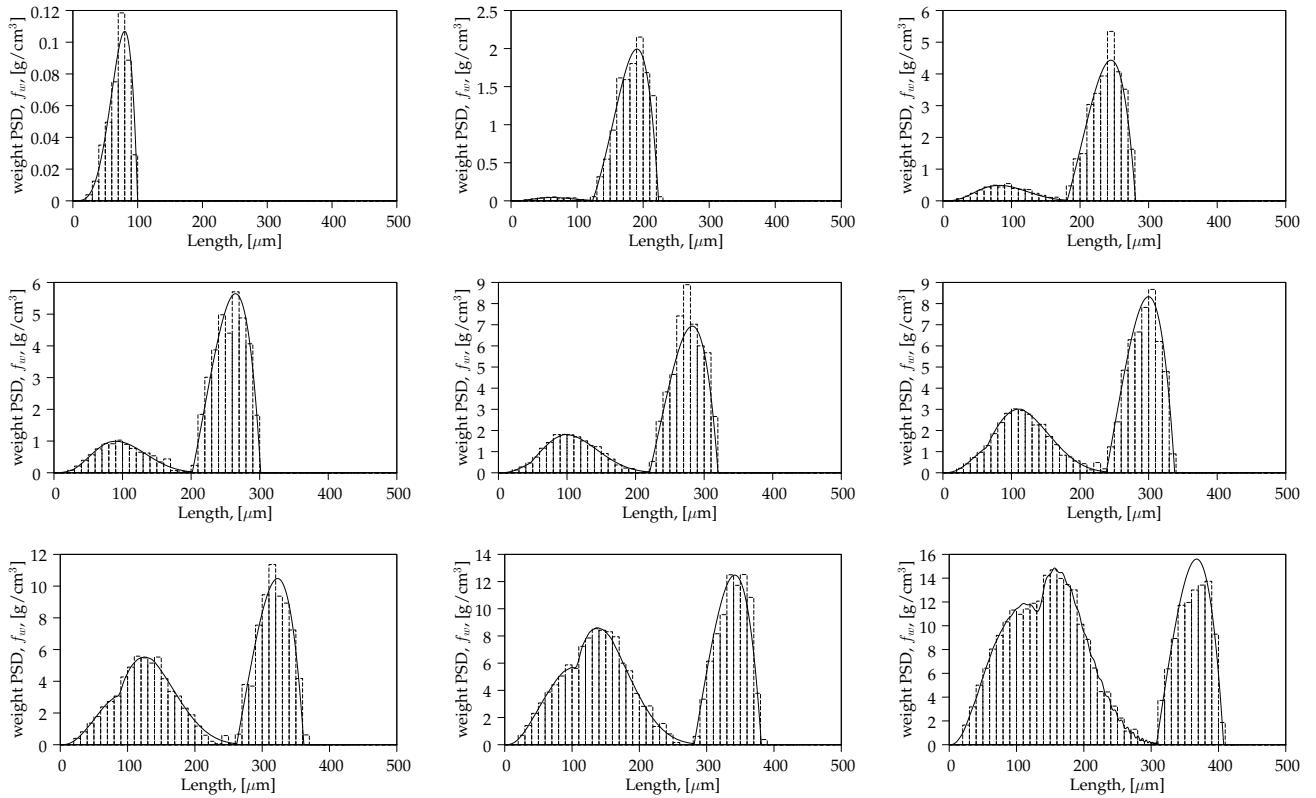


Figure 9.5: Evolution of measured and estimated weight PSD for optimal cooling and perfect image analysis. Snapshots shown from  $t = 0$  min. to  $t = 500$  min. at 60 minute intervals. Bin size =  $10 \mu\text{m}$  and  $N = 100$ .

following, we demonstrate the feasibility of monitoring a variety of these metrics using imaging.

### Ratio of nuclei mass relative to seed mass

The ratio of nucleus-grown crystal mass  $m_N$  to seed-grown crystal mass  $m_S$  affects the efficiency of downstream filtration processes [59, 82]. For the photochemical system considered here,  $m_N/m_S = \mu_{N_3}/\mu_{S_3}$  cannot be measured using conventional technologies. The needle-like habit violates the sphericity assumption necessary for effective PSD measurement by light scattering. The needle-like morphology also complicates mechanical sieving. In previous studies of this system, these measurement difficulties motivated the use of scanning electron and optical microscopy to characterize the PSD qualitatively in terms of habit and maximum size [81, p. 7]. Quantitative PSD

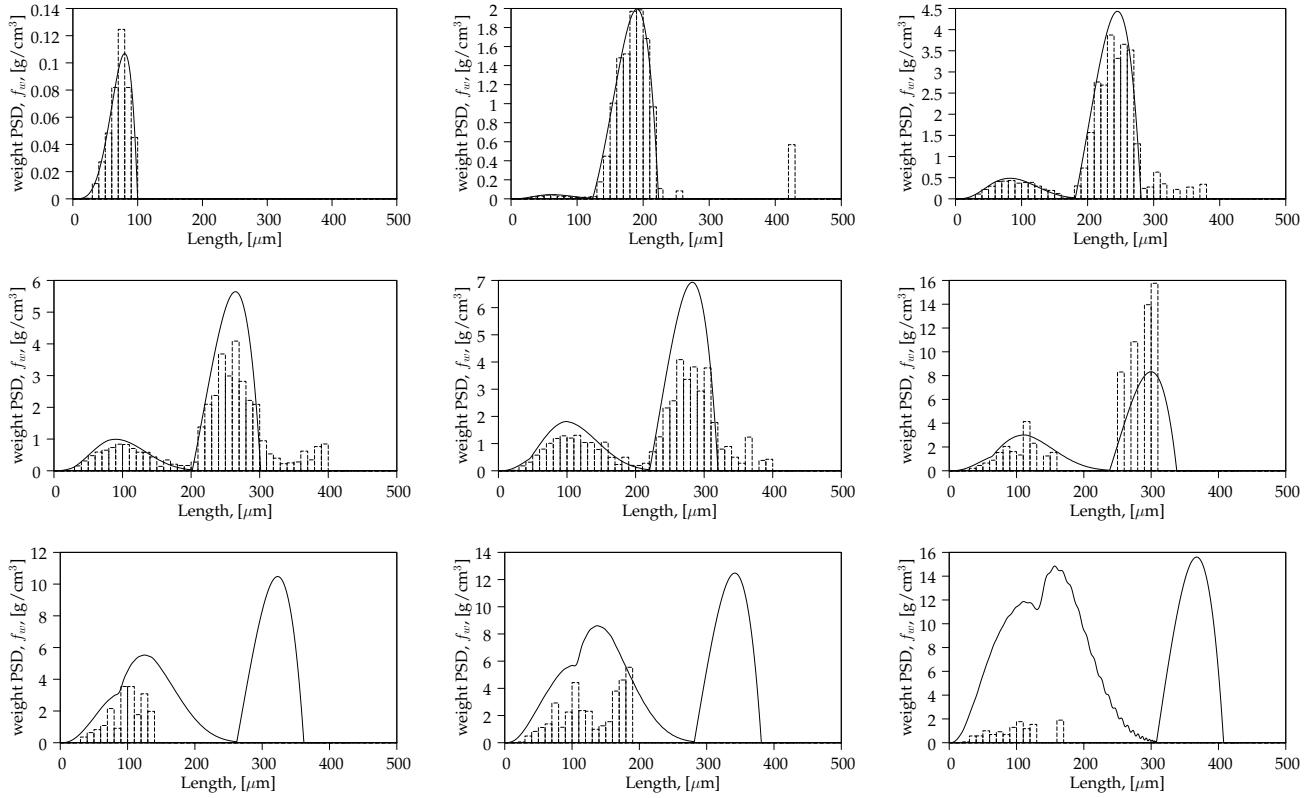


Figure 9.6: Evolution of measured and estimated weight PSD for optimal cooling and image analysis using SHARC. Snapshots shown from  $t = 0$  min. to  $t = 500$  min. at 60 minute intervals. Bin size =  $10 \mu\text{m}$  and  $N = 100$ .

measurement by microscopy could not be achieved due to sampling limitations. Using high-speed, in situ video microscopy, however, a sufficient number of samples can be obtained to enable measurement of  $m_N/m_S$ .

In both cases, the images contain sufficient particles to enable estimation of  $m_N/m_S$ , as shown in Figure 9.7. The estimates of  $m_N/m_S$  shown in Figure 9.7 are based on a sample size of 100 images, corresponding to a sample time of approximately 3 seconds (assuming the standard 30 frames/second acquisition rate). The  $m_N/m_S$  estimates are also based on the assumption that every particle appearing completely inside the imaging volume (not touching any borders) is identified perfectly.

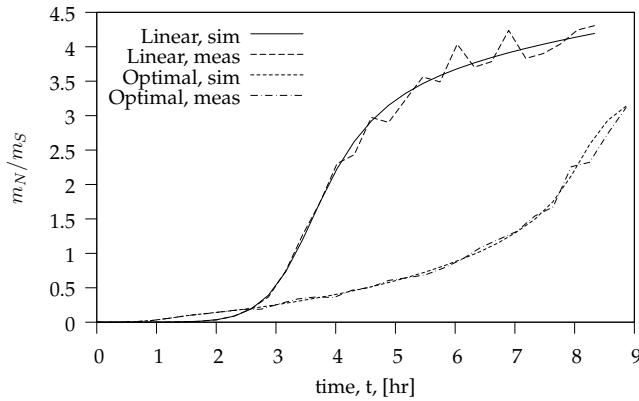


Figure 9.7: Estimated ratios of nuclei mass to seed crystal mass for optimal and linear cooling for  $N = 100$ .

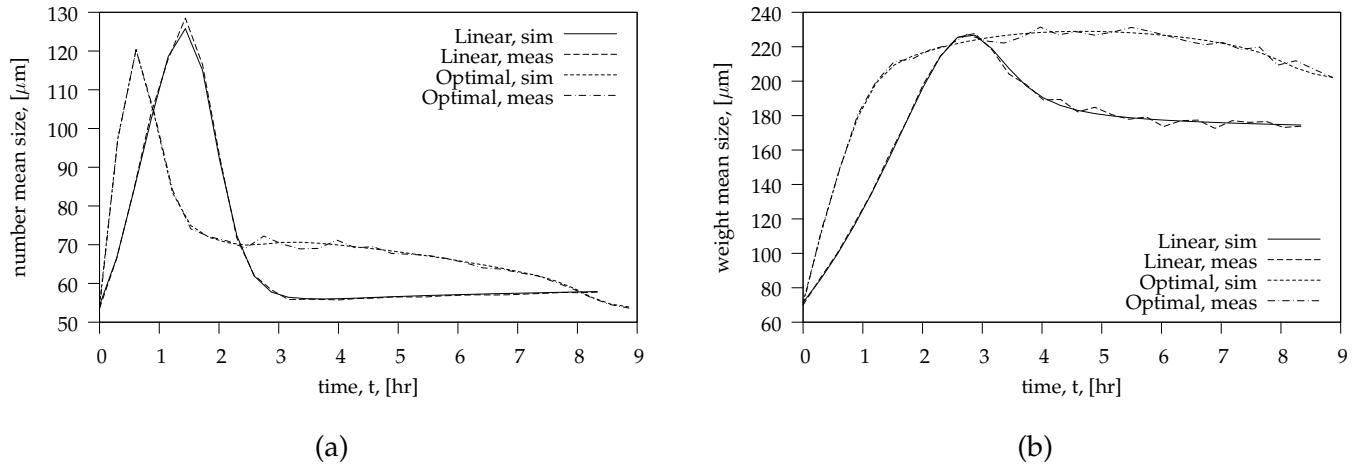


Figure 9.8: Estimated mean crystal sizes for optimal and linear cooling,  $N = 100$ . (a) Number-based. (b) Weight-based.

### Mean crystal size and coefficient of variation

The mean crystal size is commonly reported as an indication of product quality. Both the number-based mean size ( $\mu_1/\mu_0$ ) and weight-based mean size ( $\mu_4/\mu_3$ ) can be estimated effectively using imaging-based measurement, as shown in Figure 9.8. The coefficient of variation  $c_v$  quantifies the distribution spread. Typically a small  $c_v$  is desired to improve the efficiency of downstream manufacturing processes. Figure 9.9 demonstrates that both the number-based  $c_v$  ( $\mu_2\mu_0/\mu_1^2$ ) and

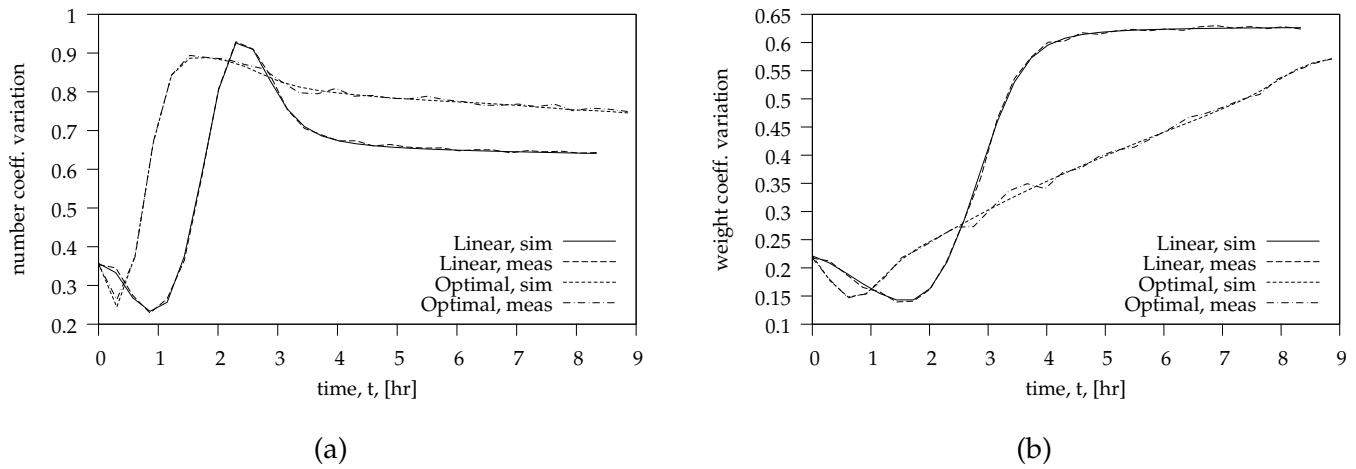


Figure 9.9: Estimated coefficients of variation for optimal and linear cooling,  $N = 100$ . (a) Number-based. (b) Weight-based.

weight-based  $c_{v_w}$  ( $\mu_5\mu_3/\mu_4^2$ ) can be measured effectively by imaging.

## 9.2.4 Discussion

The previous section demonstrated the feasibility of high-resolution PSD measurement for a specific, industrial crystallization process and realistic imaging conditions. This section discusses general sampling and image analysis considerations for imaging-based PSD measurement.

### Sampling

A fundamental problem for imaging-based PSD monitoring is determining the number of images to acquire. The problem is non-trivial because the answer depends not only on the desired accuracy, but also on the way in which the histogram is binned, the imaging conditions, and on the PSD itself. Many of the imaging-related papers cited previously discuss the number of samples required for their specific system. Here, we propose some general guidelines applicable to imaging of particulate processes in general.

Chapter 7 proposes a methodology for constructing confidence intervals for an imaging-based PSD measurement. The methodology provides a framework for determining the number

of images necessary to achieve a desired accuracy. Number-based PSDs are typically measured by discretizing the characteristic length scale into  $T$  non-overlapping bins or size classes. We therefore define the discrete PSD as

$$\rho_i = \int_{S_i}^{S_{i+1}} f(l) dl, \quad i = 1, \dots, T \quad (9.7)$$

in which  $S = (S_1, \dots, S_{T+1})$  is the vector of breaks between size classes. The maximum likelihood estimate of  $\rho_i$  is calculated from imaging data using

$$\hat{\rho}_i = \frac{X_{\Sigma_i}}{N\alpha_i} \quad (9.8)$$

in which  $X_{\Sigma_i}$  is the total number of particles observed in  $N$  images with lengths corresponding to bin  $i$ , and  $\alpha_i$  is a scalar that corrects for edge effects due to the finite size of the imaging frame ( $\alpha_i$  approaches 1 for infinitely-small particles and approaches 0 for particles the size of the image field of view). The variance of this estimator, which is directly related to the confidence intervals, is given by

$$Var(\hat{\rho}_i) = \frac{\rho_i}{N\alpha_i} \quad (9.9)$$

Equation (9.9) contains all the process, imaging, and sampling information necessary to assess the accuracy of an imaging-based PSD measurement. As mentioned, the accuracy depends not only on the number of images  $N$  and the desired resolution, but also on the imaging conditions, particle size, and particle geometry (all of which are encapsulated in the correction factor  $\alpha_i$ ) and on the number density of particles  $\rho_i$ . As a result, the number of images required to obtain a given accuracy changes as the particle population evolves.

To use Equation (9.9), one needs to know  $\rho_i$ . Batch crystallizations are typically seeded, and usually some knowledge of the seed population is available. For example, if the seed crystal population is prepared by sieving, the initial minimum crystal size and maximum size are known. The mass of seed crystals injected into the crystallizer is also known. This information is sufficient to obtain a rough estimate of  $\rho_i$  for the bins corresponding to the seed population, enabling calculation of the  $N$  that gives a sufficiently small variance (i.e. confidence interval). It is, of course,

desirable to monitor the PSD of the nucleus-grown crystals also, but the number of the nucleus-grown crystals typically is much greater than the number of the seed-grown crystals, as illustrated in Figure 9.4. Thus, a value of  $N$  that enables accurate tracking of the seed PSD will, in most cases, also enable accurate tracking of the nucleated crystals' PSD.

## Image analysis

As discussed in Section 9.2.2, imaging-based PSD measurement requires not only sufficient sampling but also effective image analysis. The effectiveness of any image analysis algorithm depends strongly on the amount of particle overlap. Chapter 8 gives an approach for quantifying the amount of particle overlap for mono-disperse populations. With this approach, the average number of overlaps per particle is predicted to be

$$D = \mu_0 \Omega d_f \quad (9.10)$$

in which  $D$  is the average number of overlaps,  $\mu_0$  is the zeroth moment of the PSD,  $d_f$  is the depth of field, and  $\Omega$  is the “admissible area,” which is based on the particle geometry. Assuming random orientation in the plane perpendicular to the camera’s optical axis,  $\Omega$  is calculated using Mack’s formula [77]:

$$\Omega = 2a_p + \frac{s_p^2}{2\pi} \quad (9.11)$$

in which  $a_p$  and  $s_p$  are, respectively, the projected area and perimeter of a single particle in its preferred (resting) orientation. Equation (8.4) can be used to determine appropriate sampling conditions for imaging-based measurement. For example, one could assume the seed population to be approximately mono-disperse and use Equation (8.4) to predict the number of overlaps per particle for a given mass of seed. If the number of overlaps is too high for current image analysis methods, one may choose to use dilution or an optical system with a smaller depth of field.

### 9.3 Conclusion

Video imaging is a viable technology for on-line PSD monitoring and control. The feasibility of high-resolution, imaging-based PSD measurement has been demonstrated using a realistic industrial crystallization process. Imaging-based measurement enables both number and weight-based PSD monitoring as well as product quality measurements that are difficult to obtain using alternative monitoring technologies, such as the ratio of nuclei mass to seed mass. General recommendations for determining appropriate sampling conditions for imaging-based PSD measurement have also been given. The proposed tools facilitate the design of effective, imaging-based PSD monitoring technology.

This chapter has also highlighted directions for future research to improve imaging-based PSD measurement. Most importantly, advances in image analysis algorithms are needed to improve image segmentation when particle overlap is significant. These advances will require further statistical estimation studies to address sampling biases due to particle overlap or occlusion. Current statistical methods adequately address edge effects (i.e. sampling bias associated with the finite size of the imaging window), but further consideration is required for errors associated with depth of field effects.



## Chapter 10

### Conclusion

This thesis has developed tools that enable effective use of video imaging technology for on-line monitoring of particulate populations. These tools include (1) image analysis algorithms that enable segmentation of *in situ* video images of crystallization processes, (2) an easy-to-implement, computationally inexpensive method for estimating the PSD from imaging data, and (3) a dimensionless parameter that is useful for quantifying particle overlap and assessing the reliability of imaging-based PSD measurement. In this chapter, we briefly review the contributions made by these tools and discuss areas that merit further research.

The SHARC algorithm discussed in Chapter 5 can robustly and efficiently extract crystal length and width information from *in situ* images of suspended, high-aspect-ratio crystals for moderate solids concentrations, giving results consistent with measurements obtained through manual image analysis by human operators. The speed with which SHARC analyzes the images is suitable for real-time monitoring and control of PSD mean and variance. Implementation using compiled code is expected to enable real-time, high-resolution PSD monitoring. SHARC has been patented by the Wisconsin Alumni Research Foundation (WARF) and licensed by Mettler-Toledo (MT). MT expects to include the SHARC algorithm in their next generation Particle and Vision Measurement (PVM) *in situ* video probe.

SHARC's performance declines for high solids concentrations and high levels of particle attrition because the degree of particle overlap and the noise arising from attrited particulate matter hinder the identification of the suspended crystals' edges. Implementing improved methods for identifying instances of collinearity may enable suitable performance for these conditions.

SHARC currently performs collinear identification only once, but we expect iterative collinear identification to improve SHARC's performance in the presence of significant particle overlap.

The M-SHARC algorithm discussed in Chapter 6 is useful for extracting crystal size and shape information from *in situ* crystallization images of particles with complex shapes. The algorithm's accuracy has been assessed by comparing its results with those obtained by manual, human analysis of the images. Despite a large number of misses and false positives, M-SHARC's cumulative size distribution (CDF) measurements compare favorably with measurements obtained by humans. In general, the CDFs are biased towards larger particles. The algorithm is sufficiently fast to provide on-line measurements for typical cooling crystallization processes. M-SHARC has been patented by WARF.

To improve the algorithm's accuracy, further development should focus on the algorithm's verification stage and on creating initialization routines for additional viewpoint-invariant line groups. The images and human annotations that have been used to evaluate the M-SHARC algorithm are available online via MIT's LabelMe project ([labelme.csail.mit.edu](http://labelme.csail.mit.edu)). We hope this data set will be useful for benchmarking and comparing the performance of image analysis algorithms developed in the future.

Chapter 7 developed the maximum likelihood estimator (MLE) for imaging-based PSD measurement of needle-like particles. The estimator uses both censored and uncensored observations (i.e. border and non-border particles). MLE is more efficient than the standard Miles-Lantu joul approach, particularly if the particle population is mono-disperse or contains particles that are large relative to the size of the image. Furthermore, MLE can estimate the number density of over-sized particles (particles bigger than the image dimension) along with the size  $L_{\max}$  of the largest particle while the Miles-Lantu joul approach can be applied only for particles smaller than the image dimension.

The primary limitation of the MLE derived in Chapter 7 is due to the assumption that the particles have needle-like geometry. The Miles-Lantu joul approach, on the other hand, can be applied to a much wider class of geometries. Secondly, the MLE approach requires the solution

of a nonlinear optimization problem. To avoid the necessity of solving an optimization problem, a simpler MLE was proposed in Chapter 7 that uses only non-border particle measurements. The efficiency of the simple MLE is comparable to that of the Miles-Lantu joul estimator.

Several areas for future work in the area of statistical estimation of PSD are evident. Choosing the optimal number, location, and size of bins for constructing histograms should be addressed. Integrating measurements taken at multiple scales or magnifications is also important. For systems of high-aspect-ratio particles, incorporating the width of border particles into the estimation could lead to increased efficiency by narrowing down the number of size classes to which a border particle may correspond.

In Chapter 8, a practical approach was developed for assessing the reliability of number density estimates obtained using imaging measurements in the presence of particle overlap. The single dimensionless parameter  $D$  correlates with the measurement reliability based on the amount of particle overlap. Thus, the parameter  $D$  can be used to estimate the errors in the measurements and to aid practitioners in determining the sampling conditions necessary to obtain reliable measurement of particle number density.

It was shown in Chapter 8 that the Miles-Lantu joul estimator, which accounts for edge effects but not particle overlap, underestimates the number density. Thus, a maximum likelihood estimator that accounts for both edge effects and particle overlap has been presented. The estimator is based on a semi-empirical model of the probability that a given particle will be correctly identified by automatic image analysis. For a given particle size, a single empirical parameter is sufficient to enable effective number density estimation for a wide range of conditions, particularly for systems in which the particles' dimensions are significantly smaller than the image dimensions.

Various issues related to statistical estimation in the presence of particle overlap should be addressed in future work. Most importantly, an extension of the methods developed in this paper should be developed for polydisperse systems. Armitage [4] and Roach [104, p.46] give discussions that may provide a good starting point for such investigations. The incorporation

of these methods into a state estimator for feedback control of number density should also be investigated.

Chapter 9 demonstrated the feasibility of high-resolution, imaging-based PSD measurement using a realistic industrial crystallization process. Imaging-based measurement enables both number and weight-based PSD monitoring as well as product quality measurements that are difficult to obtain using alternative monitoring technologies, such as the ratio of nuclei mass to seed mass. Chapter 9 also discussed how the tools developed in this thesis can be used to determine appropriate sampling conditions for imaging-based PSD measurement.

The results in Chapter 9 highlight directions for future research to improve imaging-based PSD measurement. Most importantly, and as noted previously, advances in image analysis algorithms are needed to improve image segmentation when particle overlap is significant. These advances will require further statistical estimation studies to address sampling biases due to particle overlap or occlusion. Current statistical methods adequately address edge effects (i.e. sampling bias associated with the finite size of the imaging window), but further consideration is required for errors associated with depth of field effects.

In summary, the tools developed in this thesis facilitate the design of effective, imaging-based PSD monitoring technology. As the advances discussed in this thesis are incorporated into commercial sensor technology, we expect improved understanding and control of industrial crystallization processes.

## Appendix A

# Derivations for Maximum Likelihood Estimation of PSD

## A.1 Maximum likelihood estimation of PSD

Let  $\mathbf{X}_k = (X_{1k}, \dots, X_{Tk})$  be a  $T$ -dimensional random vector in which  $X_{ik}$  gives the number of *non-border* particles of size class  $i$  observed in image  $k$ . A non-border particle is a particle that is completely enclosed within the imaging volume. A *border* particle, on the other hand, is only partially enclosed within the imaging volume such that only a portion of the particle is observable. For border particles, only the observed length (i.e. the length of the portion of the particle that is inside the imaging volume) can be measured. Accordingly, we let  $\mathbf{Y}_k = (Y_{1k}, \dots, Y_{Tk})$  be a  $T$ -dimensional random vector in which  $Y_{jk}$  gives the number of *border* particles with observed lengths in size class  $j$  that are observed in image  $k$ . We denote the observed data, or the realizations of the random vectors  $\mathbf{X}_k$  and  $\mathbf{Y}_k$ , as  $\mathbf{x}_k$  and  $\mathbf{y}_k$ , respectively.

The particle population is represented completely by the vectors  $\boldsymbol{\rho} = (\rho_1, \dots, \rho_T)$  and  $\mathbf{S} = (S_1, \dots, S_{T+1})$  in which  $\rho_i$  represents the number of particles of size class  $i$  per unit volume and  $S_i$  is the lower bound of size class  $i$ . Given the data  $\mathbf{x}$  and  $\mathbf{y}$  (the subscript  $k$  denoting the image index is removed here for simplicity), the maximum likelihood estimator of  $\boldsymbol{\rho}$  is defined as

$$\hat{\boldsymbol{\rho}}_b = \arg \max_{\boldsymbol{\rho}} p_{XY}(x_1, y_1, x_2, y_2, \dots, x_T, y_T | \boldsymbol{\rho}) \quad (\text{A.1})$$

in which in which the subscript  $b$  indicates the use of border particle measurements and  $p_{XY}$  is the

joint probability density for  $\mathbf{X}$  and  $\mathbf{Y}$ . In other words, we want to determine the value of  $\boldsymbol{\rho}$  that maximizes the probability of observing exactly  $x_1$  non-border particles of size class 1,  $y_1$  border particles of size class 1,  $x_2$  non-border particles of size class 2,  $y_2$  border particles of size class 2, and so on.

A simplified expression for  $p_{XY}$  can be obtained by noting that, at least at low solids concentrations, the observations  $X_1, Y_1, \dots, X_T, Y_T$  can be assumed to be independent. This assumption means that the observed number of particles of a given size class depends only on the density of particles in that same size class. At high solids concentrations, this assumption seems unreasonable because the number of particle observations in a given size class is reduced due to occlusions by particles in other size classes. At low concentrations, however, the likelihood of occlusion is low. The independence assumption does *not* imply that the observations are not correlated. Rather, the assumption implies that any correlation between observations is due to their dependence on a common set of parameters. As an example, if we observe a large number of non-border particles, we would expect to also observe a large number of border particles. This correlation can be explained by noting that the probability densities for both border and non-border observations depend on a common parameter, namely, the density of particles. Given the independence assumption, we express  $p_{XY}$  as

$$p_{XY} = \prod_{i=1}^T p_{X_i}(x_i|\boldsymbol{\rho}) \prod_{j=1}^T p_{Y_j}(y_j|\boldsymbol{\rho}) \quad (\text{A.2})$$

in which  $p_{X_i}$  and  $p_{Y_j}$  are the probability densities for the random variables  $X_i$  and  $Y_j$ . Using Equation (A.2), the estimator in Equation (A.1) can be reformulated as

$$\hat{\boldsymbol{\rho}}_b = \arg \min_{\boldsymbol{\rho}} \sum_{i=1}^T -\log p_{X_i}(x_i|\boldsymbol{\rho}) - \sum_{j=1}^T \log p_{Y_j}(y_j|\boldsymbol{\rho}) \quad (\text{A.3})$$

The probability densities  $p_{X_i}$  and  $p_{Y_j}$  are derived in the following sections. These derivations show that  $X_i \sim \text{Poisson}(m_{X_i})$ , or that  $X_i$  has a Poisson distribution with parameter  $m_{X_i} = \rho_i \alpha_i$ , in which  $\alpha_i$  is a function of the field of view, depth of field, and the lower and upper bounds of size class  $i$ . Furthermore,  $Y_j \sim \text{Poisson}(m_{Y_j})$ , in which  $m_{Y_j} = \sum_{i=1}^T \rho_i \beta_{ij}$

## A.2 Derivation of probability densities

The probability densities  $p_{X_i}$  and  $p_{Y_i}$  in Equation (A.3) can be derived given the particle geometry and the spatial and orientational probability distributions. Here, we derive  $p_{X_i}$  and  $p_{Y_i}$  for needle-like particles assuming the particles are randomly, uniformly distributed, both in their 3-dimensional spatial location and in their orientation in the plane perpendicular to the optical axis. To simplify the discussion, we initially present the derivation assuming a 2-dimensional system with monodisperse, vertically-oriented particles. Later, we relax these assumptions and present the derivation for randomly-oriented, polydisperse particles in 3-dimensional space.

### A.2.1 Non-border particles

Let  $S$  be a square domain in  $\mathcal{R}^2$  with dimension  $B$  and area  $A_S = B^2$ . Let  $I$  be a rectangular domain in  $\mathcal{R}^2$  with horizontal and vertical dimensions  $a$  and  $b$ , respectively, and area  $A_I = ab$ . Assume  $A_I \ll A_S$  and  $I \in S$ . Let  $n_{\text{tot}}$  be the total number of vertically-oriented particles with midpoints randomly and uniformly distributed in  $S$ , and define  $\rho = n_{\text{tot}}/A_S$  as the density of particles per unit area. Let the length of all particles be  $l$ , with  $l < \min(a, b)$ , and define  $A_{nb}$  as the area of the domain in which particles are inside  $I$  but do not touch the border of  $I$ . Because the particles are oriented vertically, it is easy to show that  $A_{nb} = a(b - l)$ , as depicted in Figure A.1(a). Finally, let  $X$  be a random variable denoting the number of non-border particles appearing in  $I$ . Assuming the location of each particle in  $S$  is independent of the remaining particles' locations, the probability that a *specific* particle will appear entirely within  $I$  is given by  $p = A_{nb}/A_S$ . Given the above assumptions, this probability is constant for all particles. The probability of observing  $x$  non-border particles in  $I$  is analogous to the probability of observing  $x$  successes in  $n_{\text{tot}}$  Bernoulli trials in which the probability of success in each trial is  $p$ . Thus,  $X$  is a binomial random variable with probability distribution

$$p_X(x) = \binom{n_{\text{tot}}}{x} p^x (1-p)^{n_{\text{tot}}-x}$$

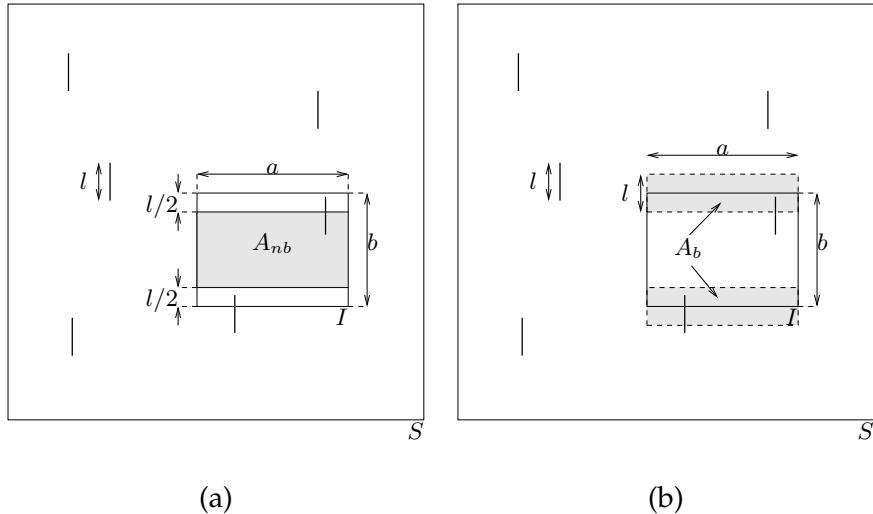


Figure A.1: Depiction of hypothetical system of vertically-oriented particles randomly and uniformly distributed in space.

Now, assume  $B \rightarrow \infty$  while keeping  $\rho$  constant. Then  $n_{\text{tot}} \rightarrow \infty$  and  $p = A_{nb}/A_S = A_{nb}\rho/n_{\text{tot}} \rightarrow 0$  while  $Np = \rho A_{nb}$  remains constant. The limiting distribution of  $X$  is therefore Poisson

$$p_X(x) = \frac{e^{-m_X} m_X^x}{x!}, \quad m_X = \rho A_{nb}$$

To extend the analysis to polydisperse, randomly-oriented needles, we discretize the length scale into  $T$  size classes and let  $\mathbf{X} = (X_1, \dots, X_T)$  be a  $T$ -dimensional random vector in which  $X_i$  gives the number of *non-border* particles of size class  $i$  observed in a single image. An orientation  $\Theta$  and length  $L$  are assigned to each particle, where  $\Theta_1, \Theta_2, \dots, \Theta_{n_{\text{tot}}}$  are i.i.d. with density function  $p_\Theta(\theta), \theta \in [-\pi/2, \pi/2]$  and  $L_1, L_2, \dots, L_{n_{\text{tot}}}$  are i.i.d. with density function  $p_L(l), l \in (0, \infty)$ .  $\Theta$  and  $L$  are independent of each other and independent of the particle's spatial location. We define  $\mathbf{S}$  as the  $T+1$ -dimensional vector of breaks between size classes. A particle of length  $l$  belongs to size class  $i$  if  $S_i \leq l < S_{i+1}$ . Let  $\Delta_i = S_{i+1} - S_i$ . Our goal is to determine the probability that a particle of size class  $i$  will appear entirely inside the image  $I$ , given its density  $\rho_i$ . Following the approach used to solve the Buffon-Laplace needle problem [116, p. 4], Figure A.2 shows geometrically that

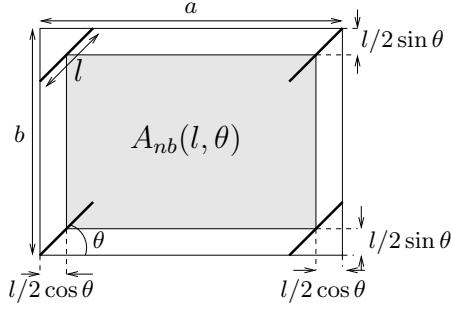


Figure A.2: Depiction of geometrical properties used to derive the non-border area function  $A_{nb}(l, \theta)$ .

for a given orientation  $\theta$  and length  $l$ ,  $A_{nb}(l, \theta)$  can be calculated as

$$A_{nb}(l, \theta) = \begin{cases} (a - l \cos \theta)(b - l \sin \theta) & 0 \leq \theta \leq \pi/2 \\ (a - l \cos \theta)(b + l \sin \theta) & -\pi/2 \leq \theta \leq 0 \end{cases} \quad (\text{A.4})$$

The probability that a given particle in size class  $i$  will appear entirely within  $I$  is given by

$$p_i = \frac{\int_{S_i}^{S_{i+1}} \int_{-\frac{\pi}{2}}^{\frac{\pi}{2}} A_{nb}(l, \theta) p_\Theta(\theta) p_L(l) d\theta dl}{\int_{S_i}^{S_{i+1}} \int_{-\frac{\pi}{2}}^{\frac{\pi}{2}} A_S p_\Theta(\theta) p_L(l) d\theta dl} \quad (\text{A.5})$$

Thus, the probability that a specific particle of size class  $i$  will appear entirely within the image is given by  $p_i = \tilde{\alpha}_i / A_S$ , where  $\tilde{\alpha}_i$  is the numerator in Equation (A.5). Following the same arguments as above, we can show that for an infinitely large system,  $X_i$  is a Poisson random variable with parameter  $m_{X_i} = \rho_i \tilde{\alpha}_i$ .

Extending the analysis to three-dimensional space is trivial because we assume the particles are oriented in the plane perpendicular to the camera's optical axis and assume no interaction between particles. Thus, for a three-dimensional system,  $X_i$  is a Poisson random variable with parameter  $m_{X_i} = \rho_i \alpha_i$ , with  $\alpha_i = \tilde{\alpha}_i d$ , in which  $d$  is the depth of field.

Assuming  $\Theta$  is distributed uniformly and  $L$  is distributed uniformly across each size class,  $\alpha_i$  can be calculated as follows. Let  $\Delta S_i = S_{i+1} - S_i$ ,  $S_{\max} = \sqrt{a^2 + b^2}$ ,  $\Delta S_{i,\max} = S_{\max} - S_i$ , and

assume  $a > b$ . For  $S_{i+1} \leq b$ ,

$$\alpha_i = \frac{d}{\pi \Delta S_i} \left[ \frac{1}{3} (S_{i+1}^3 - S_i^3) - (a+b)(S_{i+1}^2 - S_i^2) + ab\pi\Delta S_i \right]$$

For  $b < S_i, S_{i+1} \leq a$ ,

$$\begin{aligned} \alpha_i = & \frac{d}{\pi \Delta S_i} \left[ S_{i+1} \left( a\sqrt{S_{i+1}^2 - b^2} + 2ab \sin^{-1} \left( \frac{b}{S_{i+1}} \right) - b \right) \dots \right. \\ & - S_i \left( a\sqrt{S_i^2 - b^2} + 2ab \sin^{-1} \left( \frac{b}{S_i} \right) + b \right) \dots \\ & \left. + ab^2 \log \left( \frac{S_{i+1} + \sqrt{S_{i+1}^2 - b^2}}{S_i + \sqrt{S_i^2 - b^2}} \right) - a(S_{i+1}^2 - S_i^2) \right] \end{aligned}$$

For  $a \leq S_i, S_{i+1} \leq S_{\max}$ ,

$$\begin{aligned} \alpha_i = & \frac{d}{\pi \Delta S_i} \left\{ S_{i+1} \left[ a\sqrt{S_{i+1}^2 - b^2} + b\sqrt{S_{i+1}^2 - a^2} + 2ab \left( \sin^{-1} \left( \frac{b}{S_{i+1}} \right) - \cos^{-1} \left( \frac{a}{S_{i+1}} \right) \right) \right] \dots \right. \\ & - S_i \left[ a\sqrt{S_i^2 - b^2} + b\sqrt{S_i^2 - a^2} + 2ab \left( \sin^{-1} \left( \frac{b}{S_i} \right) - \cos^{-1} \left( \frac{a}{S_i} \right) \right) \right] \dots \\ & + ab^2 \log \left( \frac{S_{i+1} + \sqrt{S_{i+1}^2 - b^2}}{S_i + \sqrt{S_i^2 - b^2}} \right) + a^2 b \log \left( \frac{S_{i+1} + \sqrt{S_{i+1}^2 - a^2}}{S_i + \sqrt{S_i^2 - a^2}} \right) \dots \\ & \left. -(a^2 + b^2)\Delta S_i - \frac{1}{3}(S_{i+1}^3 - S_i^3) \right\} \end{aligned}$$

For  $a \leq S_i \leq S_{\max}$  and  $S_{i+1} > S_{\max}$ ,

$$\begin{aligned} \alpha_i = & \frac{d}{\pi \Delta S_{i,\max}} \left\{ S_{\max} \left[ a\sqrt{S_{\max}^2 - b^2} + b\sqrt{S_{\max}^2 - a^2} + 2ab \left( \sin^{-1} \left( \frac{b}{S_{\max}} \right) - \cos^{-1} \left( \frac{a}{S_{\max}} \right) \right) \right] \dots \right. \\ & - S_i \left[ a\sqrt{S_i^2 - b^2} + b\sqrt{S_i^2 - a^2} + 2ab \left( \sin^{-1} \left( \frac{b}{S_i} \right) - \cos^{-1} \left( \frac{a}{S_i} \right) \right) \right] \dots \\ & + ab^2 \log \left( \frac{S_{\max} + \sqrt{S_{\max}^2 - b^2}}{S_i + \sqrt{S_i^2 - b^2}} \right) + a^2 b \log \left( \frac{S_{\max} + \sqrt{S_{\max}^2 - a^2}}{S_i + \sqrt{S_i^2 - a^2}} \right) \dots \\ & \left. -(a^2 + b^2)\Delta S_{i,\max} - \frac{1}{3}(S_{\max}^3 - S_i^3) \right\} \end{aligned}$$

### A.2.2 Border particles

As before, we simplify the discussion by first presenting the derivation of  $p_{Y_i}$  for monodisperse, vertically-oriented particles. Let  $Y$  be a random variable denoting the total number of border

particles appearing in  $I$ . Define  $A_b$  as the area of the domain in which particles touch the border of  $I$ , as depicted in Figure A.3(a). For the present system,  $A_b = 2al$ . The probability that a *specific* particle will touch the border of  $I$  is given by  $p = A_b/A_S$ . Following the same arguments as above, we can show that for an infinitely large system,  $Y$  is a Poisson random variable with parameter  $m_Y = \rho A_b$ .

Now, assume we would like to incorporate additional information into our estimation by taking into account not only the *number* of border particles, but also their observed lengths. For a monodisperse population, these observed lengths can take on values anywhere between 0 and  $l$ . We therefore discretize the length scale on  $[0 l]$  and let  $j$  denote the size class corresponding to the observed length. We define  $Y_j$  as a random variable denoting the number of border particles appearing in  $I$  with observed length in size class  $j$ . Figure A.3(b) illustrates this approach for two size classes. In this figure,  $A_{b_1}$  is the area of the region in which particles produce observed lengths from 0 to  $l/2$ , corresponding to size class 1, while  $A_{b_2}$  is the area of the region in which particles produce observed lengths from  $l/2$  to  $l$ , corresponding to size class 2. The probability that a specific particle will touch the border of  $I$  and produce an observed length in size class  $j$  is  $p = A_{b_j}/A_S$ . Thus,  $Y_j$  is a Poisson random variable with parameter  $m_{Y_j} = \rho A_{b_j}$ .

In Figure A.3(b),  $A_{b_1} = A_{b_2}$ . This equality between the areas of different observed length size classes does not hold in general, however, as illustrated in Figure A.3(c). In this figure, we assume all particles are oriented diagonally, at 45 degrees from the horizontal, and the figure illustrates that  $A_{b_1} > A_{b_2}$ . Hence, in general, border particles are more likely to result in observed lengths in the lower size classes.

To extend the analysis to polydisperse systems with random orientation, we define a new random variable  $Y_{ij}$  that gives the number of particles in size class  $i$  that intersect the image border, producing an observed length in size class  $j$ . Given that the size class of each border particle is unknown, we define the random variable  $Y_j$  as the total number of border particles producing observed lengths in size class  $j$ , noting that  $Y_j = \sum_i Y_{ij}$ . Our approach is to determine the probability density for  $Y_{ij}$  for all  $i$  and to use these densities to derive the probability density for

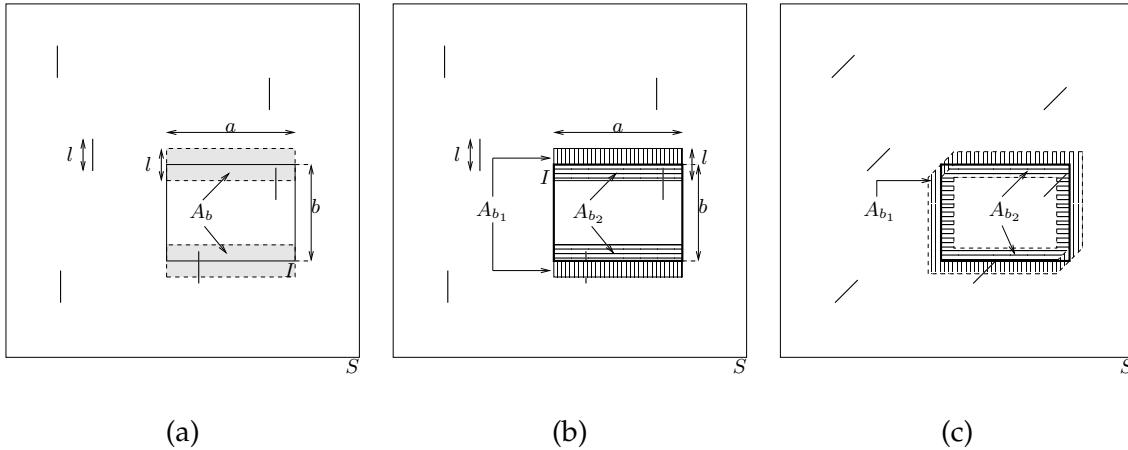


Figure A.3: Depiction of hypothetical system of vertically-oriented particles randomly and uniformly distributed in space.

$Y_j$ .

We define the function  $A_{b_j}(l, \theta)$  as the area of the region in which a particle of length  $l$  and orientation  $\theta$  produces an observed length corresponding to size class  $j$ . To calculate  $A_{b_j}(l, \theta)$ , it is convenient to define an area function  $\tilde{A}(l, \theta, \tilde{l})$  as the area of the region in which particles of length  $l$  and orientation  $\theta$  either intersect or are enclosed within the image boundary and produce an observed length greater than or equal to  $\tilde{l}$ .  $\tilde{A}(l, \theta, \tilde{l})$  can be calculated using the geometric relationships shown in Figure A.4: In this figure, the thick-lined, outer rectangle is the image region, and the inner rectangle is the region inside which a particle with length  $l$  and orientation  $\theta$  will be entirely enclosed within the image boundary, thus producing an observed length of exactly  $l$ . A particle with its midpoint along the perimeter of the outermost hexagon would touch the image boundary but give an observed length of 0. A particle with its midpoint anywhere inside the innermost hexagon will produce an observed length greater than or equal to  $\tilde{l}$ . Using the relationships indicated in this figure, and assuming  $\tilde{l} \leq l$ ,  $\tilde{A}(l, \theta, \tilde{l})$  can be calculated as

$$\tilde{A}(l, \theta, \tilde{l}) = \begin{cases} (a + (l - 2\tilde{l}) \cos \theta)(b + (l - 2\tilde{l}) \sin \theta) - (l - \tilde{l})^2 \sin \theta \cos \theta & 0 \leq \theta \leq \pi/2 \\ (a + (l - 2\tilde{l}) \cos \theta)(b - (l - 2\tilde{l}) \sin \theta) + (l - \tilde{l})^2 \sin \theta \cos \theta & -\pi/2 \leq \theta \leq 0 \end{cases} \quad (\text{A.6})$$

If  $b \leq \tilde{l} < a$ , Equation (A.6) is valid only for  $\theta$  on  $(-\sin^{-1}(b/\tilde{l}), \sin^{-1}(b/\tilde{l}))$ . If  $a, b \leq \tilde{l} < (a^2 + b^2)^{1/2}$ ,

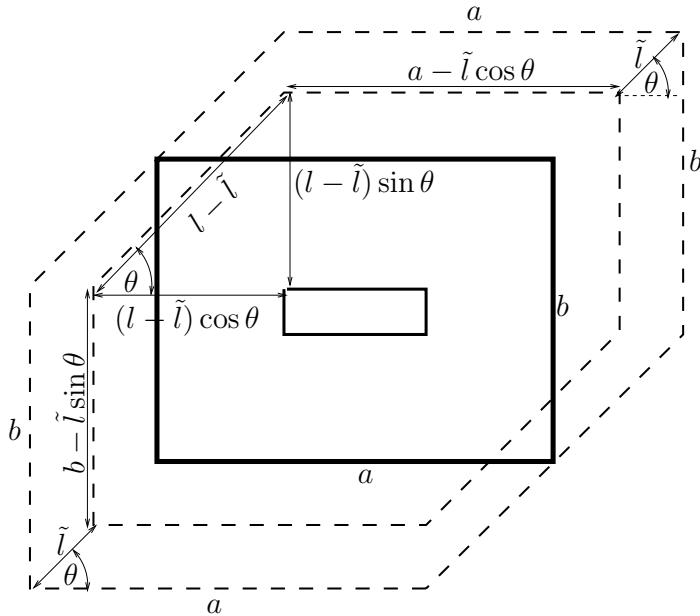


Figure A.4: Depiction of non-border area for arbitrary length and orientation.

Equation (A.6) is valid only for  $\theta$  on  $(-\sin^{-1}(b/\tilde{l}), -\cos^{-1}(a/\tilde{l}))$  and  $(\cos^{-1}(a/\tilde{l}), \sin^{-1}(b/\tilde{l}))$ .  $A_{b_j}(l, \theta)$  is given by

$$A_{b_j}(l, \theta) = \begin{cases} \tilde{A}(l, \theta, S_j) - \tilde{A}(l, \theta, S_{j+1}) & l \geq S_{j+1} \\ \tilde{A}(l, \theta, S_j) - A_{nb}(l, \theta) & S_j \leq l < S_{j+1} \\ 0 & l < S_j \end{cases} \quad (\text{A.7})$$

The probability that a given particle in size class  $i$  will appear within  $I$  and produce an observed length in size class  $j$  is given by

$$p_{ij} = \frac{\int_{S_i}^{S_{i+1}} \int_{-\frac{\pi}{2}}^{\frac{\pi}{2}} A_{b_j}(l, \theta) p_\Theta(\theta) p_L(l) d\theta dl}{\int_{S_i}^{S_{i+1}} \int_{-\frac{\pi}{2}}^{\frac{\pi}{2}} A_S p_\Theta(\theta) p_L(l) d\theta dl} \quad (\text{A.8})$$

The probability that a specific particle in size class  $i$  will touch the border of  $I$  and produce an observed length in size class  $j$  is  $p_{ij} = \tilde{\beta}_{ij}/A_S$ , with  $\tilde{\beta}_{ij}$  being the numerator in Equation (A.8). Thus, for an infinitely large system,  $Y_{ij}$  is a Poisson random variable with parameter  $m_{Y_{ij}} = \rho_i \tilde{\beta}_{ij}$ . Assuming  $Y_{1j}, Y_{2j}, \dots, Y_{Tj}$  are independent, then  $Y_j = \sum_i Y_{ij}$  is also a Poisson random variable

with parameter  $m_{Y_j} = \sum_i \rho_i \tilde{\beta}_{ij}$  [12, p.440]. As in the non-border case, the analysis is extended to three-dimensional space assuming the particles are oriented in the plane perpendicular to the camera's optical axis and that the particles do not interact. Thus, for a three-dimensional system,  $Y_j$  is a Poisson random variable with parameter  $m_{Y_j} = \sum_i \rho_i \beta_{ij}$ , with  $\beta_{ij} = \tilde{\beta}_{ij} d$ .

Assuming  $\Theta$  is distributed uniformly and  $L$  is distributed uniformly across each size class,  $\beta_{ij}$  is calculated as follows. Let the length scale discretization be the same for both border and non-border particles. As before, let  $\Delta S_i = S_{i+1} - S_i$ ,  $S_{\max} = \sqrt{a^2 + b^2}$ ,  $\Delta S_{i,\max} = S_{\max} - S_i$ , and assume  $a > b$ . Then  $\beta_{ij}$  is given by

$$\beta_{ij} = \begin{cases} \bar{A}(i, S_j) - \bar{A}(i, S_{j+1}) & i > j \\ \bar{A}(i, S_j) - \alpha_i & i = j \\ 0 & i < j \end{cases} \quad (\text{A.9})$$

in which  $\bar{A}(i, S)$  is calculated as

$$\bar{A}(i, S) = \frac{d}{\pi \Delta S_i} [\Delta S_i (2ab\gamma_1 - 4bS\gamma_2 - 4aS\gamma_3 + 3S^2\gamma_4) + (S_{i+1}^2 - S_i^2)(b\gamma_2 + a\gamma_3 - S\gamma_4)]$$

$$\gamma_1 = \begin{cases} \pi/2 & S < b \\ \sin^{-1}(b/S) & b < S < a \\ \sin^{-1}(b/S) - \cos^{-1}(a/S) & a < S < S_{\max} \\ \sin^{-1}(b/S_{\max}) - \cos^{-1}(a/S_{\max}) & S > S_{\max} \end{cases}$$

$$\gamma_2 = \begin{cases} 1 & S < b \\ b/S & b < S < a \\ (b - \sqrt{S^2 - a^2})/S & a < S < S_{\max} \\ (b - \sqrt{S_{\max}^2 - a^2})/S_{\max} & S > S_{\max} \end{cases}$$

$$\gamma_3 = \begin{cases} 1 & S < b \\ 1 - \sqrt{S^2 - b^2}/S & b < S < a \\ (a - \sqrt{S^2 - b^2})/S & a < S < S_{\max} \\ (a - \sqrt{S_{\max}^2 - b^2})/S_{\max} & S > S_{\max} \end{cases}$$

$$\gamma_4 = \begin{cases} 1 & S < b \\ b^2/S^2 & b < S < a \\ (a^2 + b^2 - S^2)/S^2 & a < S < S_{\max} \\ (a^2 + b^2 - S_{\max}^2)/S_{\max}^2 & S > S_{\max} \end{cases}$$

### A.3 Validation of marginal densities

To ensure the correctness of the probability densities derived in the previous section, four different Monte Carlo simulations were carried out in which artificial images of particulate populations were generated. Figure 7.2 shows example images generated for each simulation. Each of these images has a horizontal image dimension of  $a=480$  pixels and a vertical dimension of  $b=480$  pixels. The first row displays four simulated images for monodisperse particles of length  $0.5a$  with  $N_c=25$  crystals per image. The second row shows images of particles uniformly distributed on  $[0.1a \ 0.9a]$  with  $N_c=25$ . The third row shows images of particles normally-distributed with  $\mu = 0.5a$  and  $\sigma = 0.4a/3$  with  $N_c=25$ , and the fourth row shows example images for simulations of particles uniformly-distributed on  $[0.1a \ 2.0a]$  with  $N_c=15$ . For each simulation, 20,000 artificial images were generated. Based on the observations in these 20,000 images, a histogram was generated for each size class giving the frequency of observations for both border and non-border particles. These histograms are compared with the theoretical marginal densities in Figures A.5– A.12.

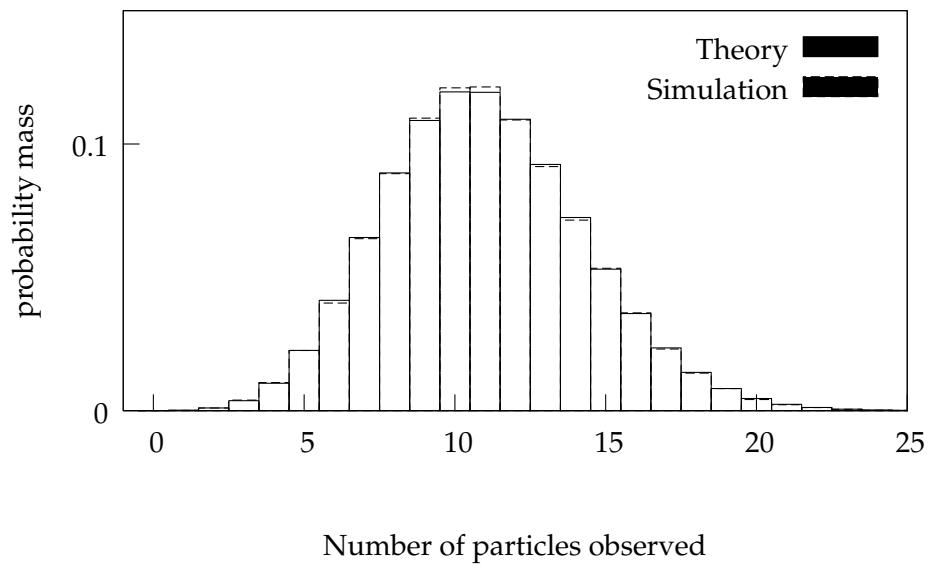


Figure A.5: Comparison of theoretical and simulated marginal densities for randomly-oriented, monodisperse particles of length 0.5 and measured by partitioning [0.1 0.9] into ten bins. Results are for non-border particles.

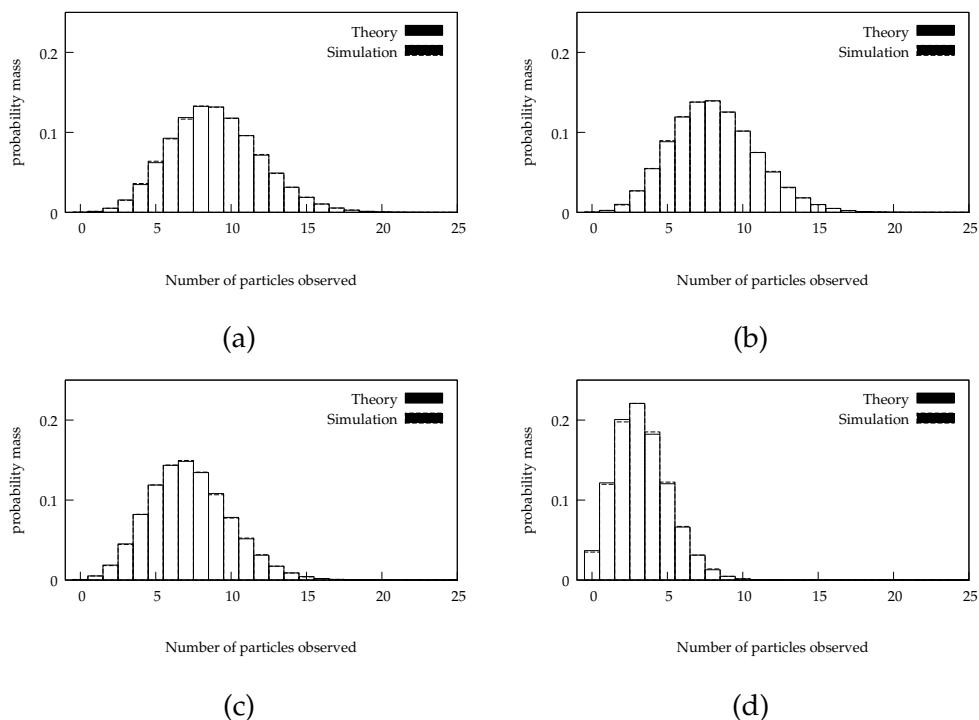


Figure A.6: Comparison of theoretical and simulated marginal densities for randomly-oriented, monodisperse particles of length 0.5 and measured by partitioning [0.1 0.9] into ten bins (results are shown only for bins 1–4 because the probability of observing a border length in size class 5 or above is zero). Results are for border particles.

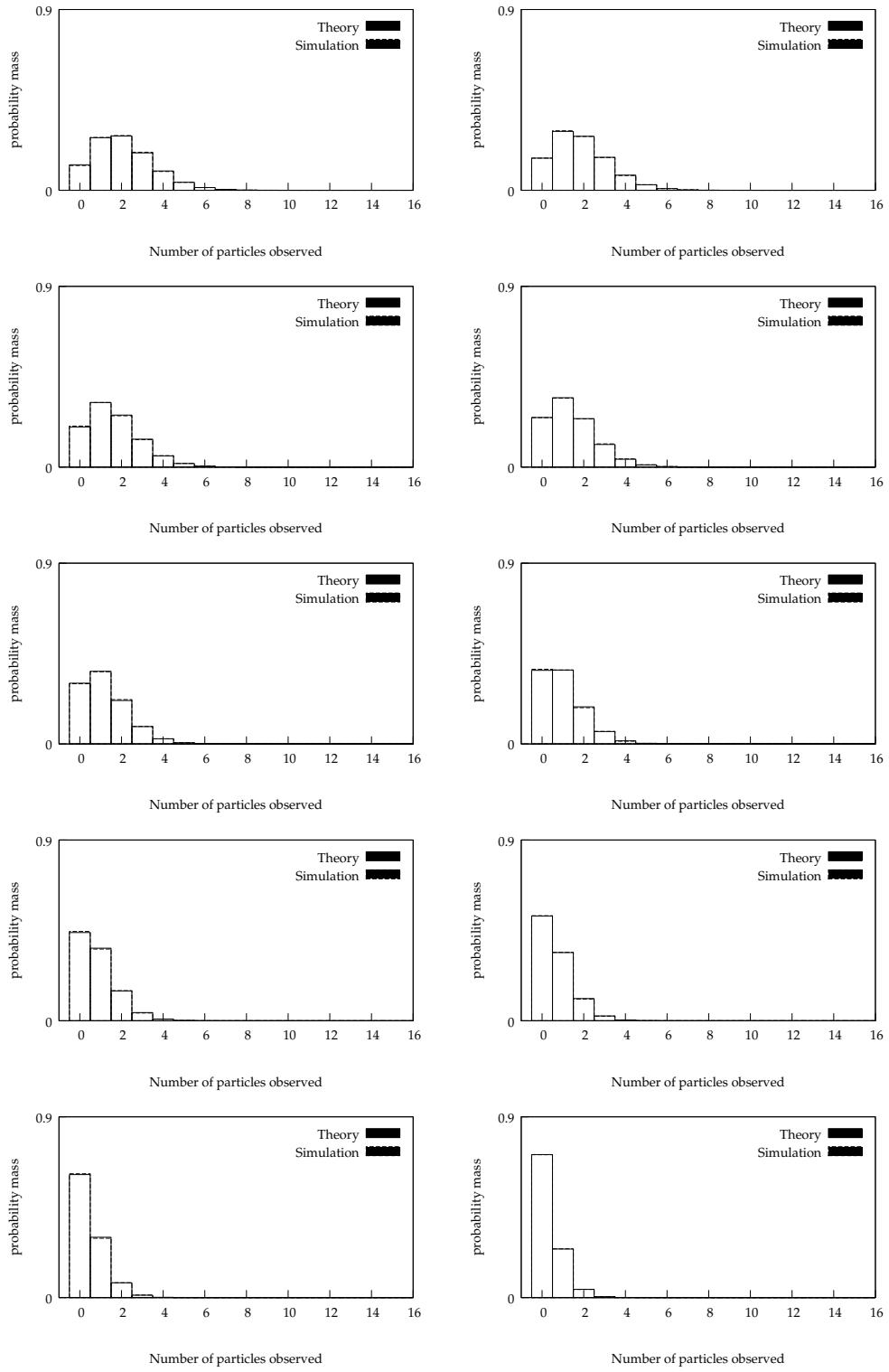


Figure A.7: Comparison of theoretical and simulated marginal densities for randomly-oriented particles distributed uniformly on  $[0.1 \ 0.9]$  and measured by partitioning  $[0.1 \ 0.9]$  into ten bins. Results are for non-border particles.

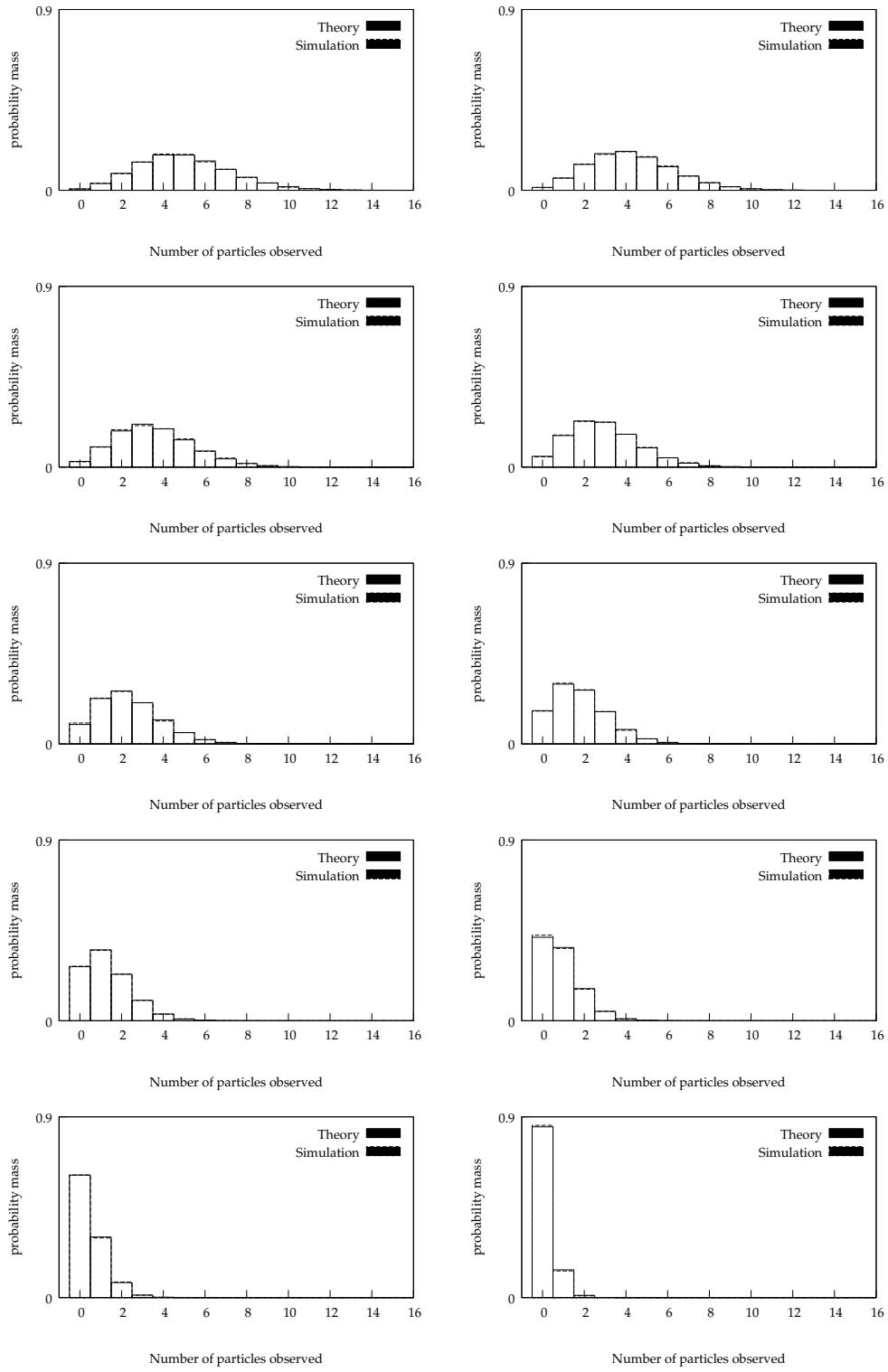


Figure A.8: Comparison of theoretical and simulated marginal densities for randomly-oriented particles distributed uniformly on  $[0.1 \ 0.9]$  and measured by partitioning  $[0.1 \ 0.9]$  into ten bins. Results are for border particles.

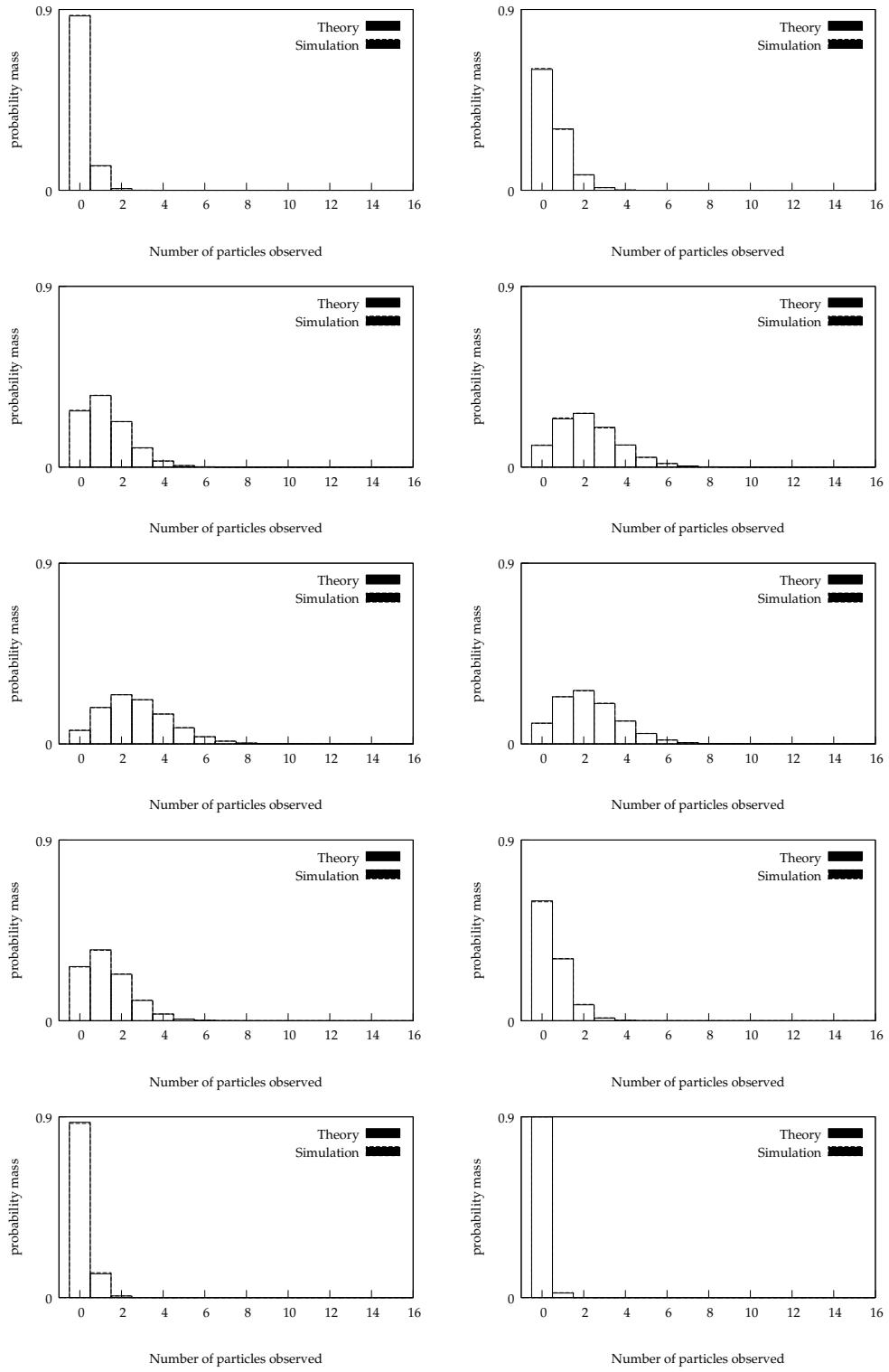


Figure A.9: Comparison of theoretical and simulated marginal densities for randomly-oriented particles distributed normally and measured by partitioning [0.1 0.9] into 10 bins. Results are for non-border particles.

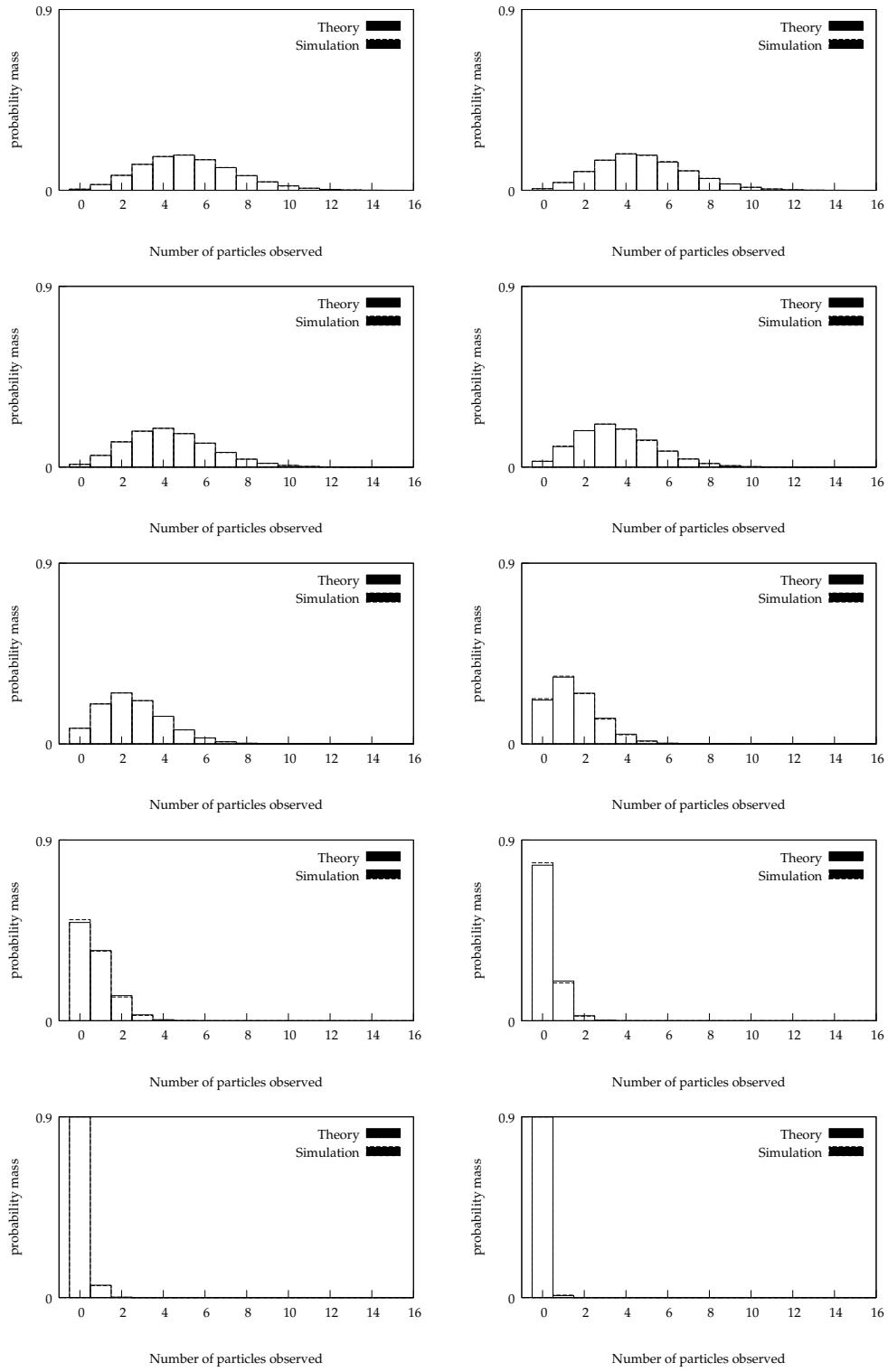


Figure A.10: Comparison of theoretical and simulated marginal densities for randomly-oriented particles distributed normally and measured by partitioning [0.1 0.9] into 10 bins. Results are for border particles.

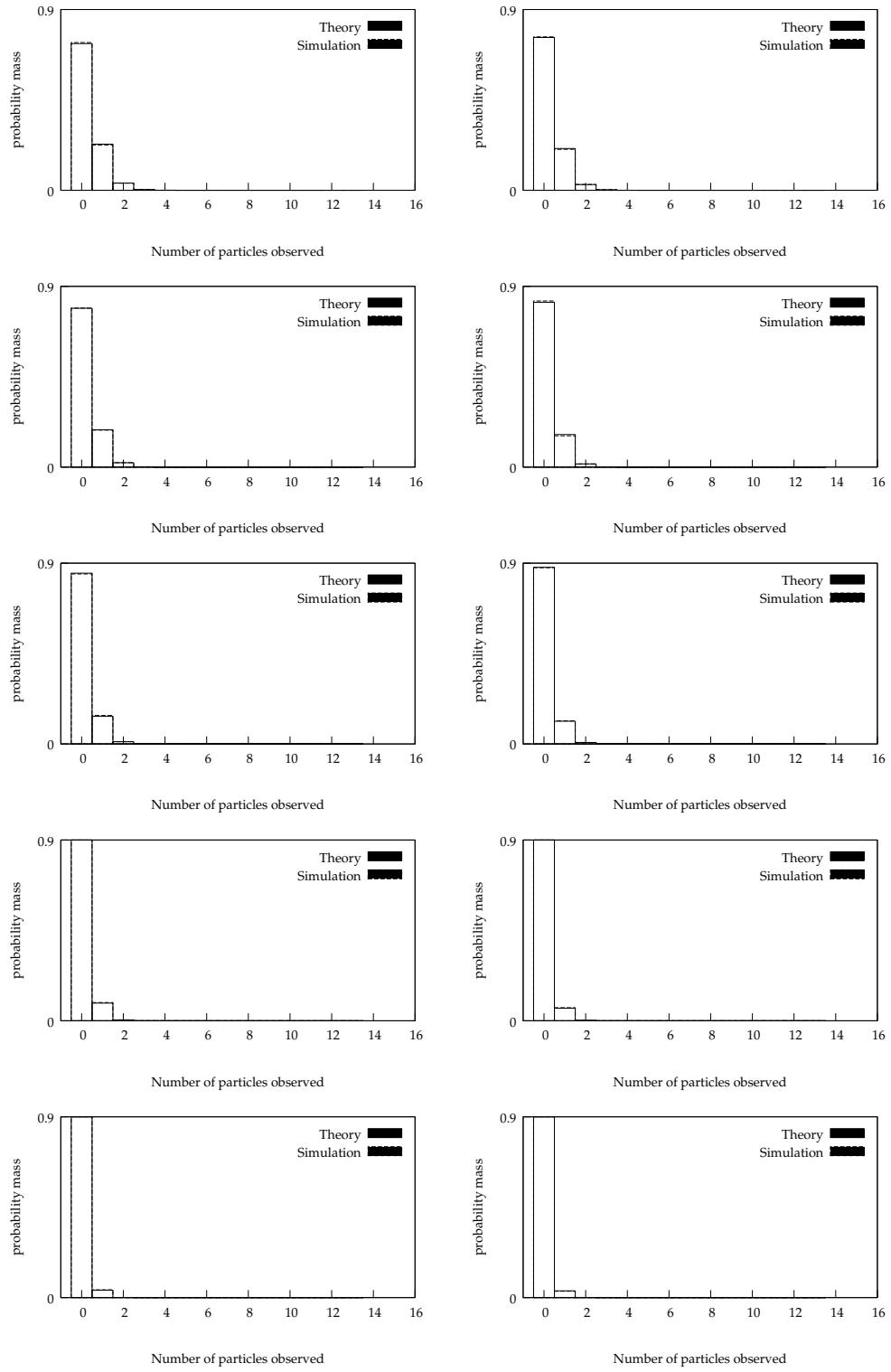


Figure A.11: Comparison of theoretical and simulated marginal densities for randomly-oriented particles distributed uniformly on  $[0.4 \ 2.0]$  and measured by partitioning  $[0.4 \ 1.0]$  into 9 bins with a 10th bin spanning  $[1.0 \ \sqrt{2}]$ . Results are for non-border particles.

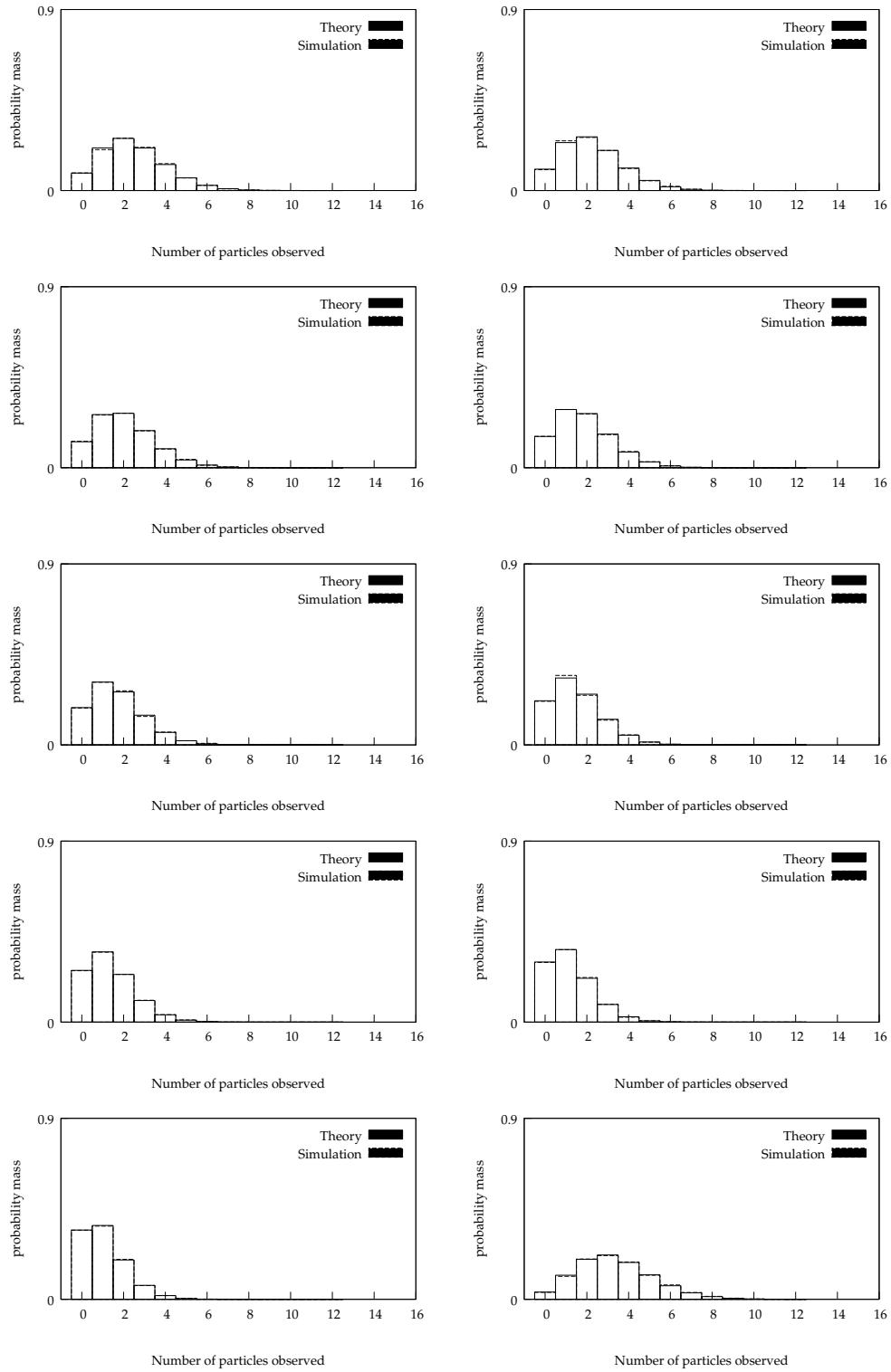


Figure A.12: Comparison of theoretical and simulated marginal densities for randomly-oriented particles distributed uniformly on  $[0.4 \ 2.0]$  and measured by partitioning  $[0.4 \ 1.0]$  into 9 bins with a 10th bin spanning  $[1.0 \ \sqrt{2}]$ . Results are for border particles.



# Bibliography

- [1] *Stochastic Geometry and Its Applications*. John Wiley & Sons, Chichester, 1987.
- [2] P. Agarwal and K. A. Berglund. In situ monitoring of calcium carbonate polymorphs during batch crystallization in the presence of polymeric additives using Raman spectroscopy. *Crystal Growth and Design*, 3(6):941–946, 2003.
- [3] T. Allen. *Powder Sampling and Particle Size Determination*. Elsevier, 2003.
- [4] P. Armitage. An overlap problem arising in particle counting. *Biometrika*, 36(3/4):257–266, December 1949.
- [5] A. J. Baddeley. *Stochastic Geometry Likelihood and Computation*, chapter Spatial sampling and censoring, pages 37–78. Chapman and Hall/CRC, Boca Raton, FL, 1999.
- [6] A. J. Baddeley. *Stochastic Geometry Likelihood and Computation*, chapter A crash course in stochastic geometry, pages 1–35. Chapman and Hall/CRC, Boca Raton, FL, 1999.
- [7] P. Barrett. Selecting in-process particle-size analyzers. *Chemical Engineering Progress*, 99(8):26–32, August 2003.
- [8] P. Barrett and B. Glennon. In-line FBRM monitoring of particle size in dilute agitated suspensions. *Particle and Particle Systems Characterization*, 16(5):207–211, 1999.
- [9] P. Barrett and B. Glennon. Characterizing the metastable zone width and solubility curve using Lasentec FBRM and PVM. *Chemical Engineering Research and Design*, 80(A7):799–805, 2002.
- [10] J. Bauer, S. Spanton, R. Henry, J. Quick, W. Dziki, W. Porter, and J. Morris. Ritonavir: An extraordinary example of conformational polymorphism. *Pharmaceutical Research*, 18(6):859–866, 2001.
- [11] M. Birch, S. J. Fussell, P. D. Higginson, N. McDowall, and I. Marziano. Towards a PAT-based strategy for crystallization development. *Organic Process Research & Development*, 9(3):360–364, 2005.
- [12] Y. M. M. Bishop, S. E. Fienberg, and P. W. Holland. *Discrete Multivariate Analysis: Theory and Practice*. The MIT Press, Cambridge, Massachusetts, 1975.

- [13] N. Blagden, R. Davey, R. Rowe, and R. Roberts. Disappearing polymorphs and the role of reaction by-products: The case of sulphathiazole. *International Journal of Pharmaceutics*, 172(1–2):169–177, 1998.
- [14] A.-F. Blandin, A. Rivoire, D. Mangin, J.-P. Klein, and J.-M. Bossoutrot. Using in situ image analysis to study the kinetics of agglomeration in suspension. *Particle and Particle Systems Characterization*, 17:16–20, 2000.
- [15] S. Boerrigter, G. Josten, J. van de Streek, F. Hollander, J. Los, H. Cuppen, P. Bennema, and H. Meekes. MONTY: Monte Carlo crystal growth on any crystal structure in any crystallographic orientation; application to fats. *Journal of Physical Chemistry A*, 108(27):5894–5902, 2004.
- [16] R. D. Braatz. Advanced control of crystallization processes. *Annual Reviews in Control*, 26:87–99, 2002.
- [17] R. D. Braatz and S. Hasebe. Particle size and shape control in crystallization processes. In *Chemical Process Control—CPC 6*, pages 307–327, Tucson, Arizona, January 2001.
- [18] J. B. Burns, A. R. Hanson, and E. M. Riseman. Extracting straight lines. *IEEE Transactions on Pattern Analysis and Machine Intelligence*, 8(4):425–455, July 1986.
- [19] J. Calderon De Anda, X. Wang, X. Lai, and K. Roberts. Classifying organic crystals via in-process image analysis and the use of monitoring charts to follow polymorphic and morphological changes. *Journal of Process Control*, 15(7):785–797, 2005.
- [20] J. Calderon De Anda, X. Wang, and K. Roberts. Multi-scale segmentation image analysis for the in-process monitoring of particle shape with batch crystallizers. *Chemical Engineering Science*, 60:1053–1065, 2005.
- [21] S. Christy and R. Horau. Iterative pose computation from line correspondences. *Computer Vision and Image Understanding*, 73(1):137–144, January 1999.
- [22] S. H. Chung, D. L. Ma, and R. D. Braatz. Optimal seeding in batch crystallization. *The Canadian Journal of Chemical Engineering*, 77(3):590–596, 1999.
- [23] G. Clydesdale, R. Docherty, and K. J. Roberts. HABIT—a program for predicting the morphology of molecular crystals. *Computer Physics Communications*, 64(2):311–328, 1991.
- [24] W. W. Daniel. *Applied Nonparametric Statistics*. Houghton Mifflin Company, Boston, MA, 1978.
- [25] S. Datta and D. J. W. Grant. Crystal structure of drugs: Advances in determination, prediction and engineering. *Nature Reviews Drug Discovery*, 3(1):42–57, January 2004.

- [26] C. Deeley, R. Spragg, and T. Threlfall. A comparison of Fourier-Transform Infrared and Near-Infrared Fourier-Transform Raman-spectroscopy for quantitative measurements — An application in polymorphism. *Spectrochimica Acta A*, 47(9–10):1217–1223, 1991.
- [27] S. Dharmayat, J. Calderon De Anda, R. B. Hammond, X. Lai, K. J. Roberts, and X. Z. Wang. Polymorphic transformation of L-glutamic acid monitored using combined on-line video microscopy and X-ray diffraction. *Journal of Crystal Growth*, 294:35–40, 2006.
- [28] M. Dhome, M. Richetin, J.-T. Lapresté, and G. Rives. Determination of the attitude of 3-D objects from a single perspective view. *IEEE Transactions on Pattern Analysis and Machine Intelligence*, 11(12):1265–1278, December 1989.
- [29] N. Doki, H. Seki, K. Takano, H. Asatani, M. Yokota, and N. Kubota. Process control of seeded batch cooling crystallization of the metastable  $\alpha$ -form glycine using an in-situ ATR-FTIR spectrometer and an in-situ FBRM particle counter. *Crystal Growth and Design*, 4(5):949–953, 2004.
- [30] W. Doyle. Technical note AN-923. A user's guide to: spectroscopic analysis for industrial applications. Technical report, Axiom Analytical, Inc., Irvine, CA, January 2005.
- [31] D. D. Dunuwila and K. A. Berglund. ATR FTIR spectroscopy for in situ measurement of supersaturation. *Journal of Crystal Growth*, 179(1–2):185–193, 1997.
- [32] D. D. Dunuwila, L. B. Carroll II, and K. A. Berglund. An investigation of the applicability of attenuated total reflection infrared spectroscopy for measurement of solubility and supersaturation of aqueous citric acid solutions. *Journal of Crystal Growth*, 137(3–4):561–568, 1994.
- [33] A. Etemadi, J.-P. Schmidt, G. Matas, J. Illingworth, and J. Kittler. Low-level grouping of straight line segments. In *Proceedings of the British Machine Vision Conference*, 1991.
- [34] J. A. Falcon and K. A. Berglund. Monitoring of antisolvent addition crystallization with Raman spectroscopy. *Crystal Growth and Design*, 3(6):947–952, 2003.
- [35] L. Feng and K. A. Berglund. ATR-FTIR for determining optimal cooling curves for batch crystallization of succinic acid. *Crystal Growth and Design*, 2(5):449–452, 2002.
- [36] E. S. Ferrari, R. J. Davey, W. I. Cross, A. L. Gillon, and C. S. Towler. Crystallization in polymorphic systems: The solution-mediated transformation of  $\beta$  to  $\alpha$  glycine. *Crystal Growth and Design*, 3(1):53–60, 2003.
- [37] G. Févotte. New perspectives for the on-line monitoring of pharmaceutical crystallization processes using in situ infrared spectroscopy. *International Journal of Pharmaceutics*, 241(2):263–278, 2002.

- [38] G. Févotte, J. Calas, F. Puel, and C. Hoff. Applications of NIR spectroscopy to monitoring and analyzing the solid state during industrial crystallization processes. *International Journal of Pharmaceutics*, 273(1–2):159–169, 2004.
- [39] D. A. Forsyth and J. Ponce. *Computer Vision: A Modern Approach*. Prentice Hall series in artificial intelligence. Prentice Hall, New Jersey, 2003.
- [40] M. Fujiwara, P. S. Chow, D. L. Ma, and R. D. Braatz. Paracetamol crystallization using laser backscattering and ATR-FTIR spectroscopy: Metastability, agglomeration, and control. *Crystal Growth and Design*, 2(5):363–370, 2002.
- [41] G. H. Givens and J. A. Hoeting. *Computational Statistics*. Wiley Series in Probability and Statistics. John Wiley & Sons, New Jersey, 2005.
- [42] R. C. Gonzalez and R. E. Woods. *Digital Image Processing*. Prentice Hall, second edition, 2002.
- [43] D. A. Green. Particle formation and morphology control. *DuPont Magazine*, College Report supplement, November/December 1988.
- [44] W. Grimson and D. Huttenlocher. On the verification of hypothesized matches in model-based recognition. *IEEE Transactions on Pattern Analysis and Machine Intelligence*, 13(12):1201–1213, 1989.
- [45] H. Gron, A. Borissova, and K. Roberts. In-process ATR-FTIR spectroscopy for closed-loop supersaturation control of a batch crystallizer producing monosodium glutamate crystals of defined size. *Industrial and Engineering Chemistry Research*, 42(1):198–206, 2003.
- [46] P. Hall. Correcting segment counts for edge effects when estimating intensity. *Biometrika*, 72(2):459–63, 1985.
- [47] R. W. Hartel. Crystallization in foods. In A. S. Myerson, editor, *Handbook of industrial crystallization*, pages 287–304. Butterworth-Heinemann, USA, 2nd edition, 2002.
- [48] M. Honkanen, P. Saarenrinne, T. Stoor, and J. Niinimaki. Recognition of highly overlapping ellipse-like bubble images. *Measurement Science and Technology*, 16(9):1760–1770, 2005.
- [49] R. Horaud, B. Conio, O. Leboulleux, and B. Lacolle. An analytic solution for the perspective 4-point problem. *Computer Vision, Graphics, and Image Processing*, 47:33–44, 1989.
- [50] Y. Hu, J. K. Liang, A. S. Myerson, and L. S. Taylor. Crystallization monitoring by Raman spectroscopy: Simultaneous measurement of desupersaturation profile and polymorphic form in flufenamic acid systems. *Industrial and Engineering Chemistry Research*, 44(5):1233–1240, 2005.
- [51] E. J. Hukkanen and R. D. Braatz. Measurement of particle size distribution in suspension polymerization using in situ laser backscattering. *Sensors and Actuators B*, 96:451–459, 2003.

- [52] E. J. Hukkanen, J. G. VanAntwerp, and R. D. Braatz. Determination of breakage and coalescence kinetics in suspension polymerization reactors using in-situ laser backscattering and process video microscopy. Submitted to *Chemical Engineering Science*, 2007.
- [53] H. M. Hulbert and S. Katz. Some problems in particle technology: A statistical mechanical formulation. *Chemical Engineering Science*, 19:555–574, 1964.
- [54] D. P. Huttenlocher. Recognition by alignment. In A. K. Jain and P. J. Flynn, editors, *Three-dimensional object recognition systems*, Advances in image communication, pages 311–326. Elsevier, Amsterdam, 1993.
- [55] D. P. Huttenlocher and S. Ullman. Recognizing solid objects by alignment with an image. *International Journal of Computer Vision*, 5(2):195–212, 1990.
- [56] Y. Iitaka. The crystal structure of  $\beta$ -glycine. *Acta Crystallographica*, 13:35–45, 1960.
- [57] Y. Iitaka. The crystal structure of  $\gamma$ -glycine. *Acta Crystallographica*, 14:1–10, 1961.
- [58] J.-H. Jang and K.-S. Hong. Fast line segment grouping method for finding globally more favorable line segments. *Pattern Recognition*, 35:2235–2247, 2002.
- [59] A. G. Jones, J. Budz, and J. W. Mullin. Batch crystallization and solid-liquid separation of potassium sulphate. *Chemical Engineering Science*, 42(4):619–629, 1987.
- [60] P.-G. Jonsson and A. Kvick. Precision neutron diffraction structure determination of protein and nucleic acid components. iii. the crystal and molecular structure of the amino acid  $\alpha$ -glycine. *Acta Crystallographica*, B28:1827–1833, 1972.
- [61] P. Kahn, L. Kitchen, and E. Riseman. A fast line finder for vision-guided robot navigation. *IEEE Transactions on Pattern Analysis and Machine Intelligence*, 12(11):1098–1102, November 1990.
- [62] E. N. Kaufman and T. C. Scott. In situ visualization of coal particle distribution in a liquid fluidized bed using fluorescence microscopy. *Powder Technology*, 78:239–246, 1994.
- [63] A. Kellerer. On the number of clumps resulting from the overlap of randomly placed figures in a plane. *Journal of Applied Probability*, 20(1):126–135, 1983.
- [64] C. Lantuéjoul. Computation of the histograms of the number of edges and neighbours of cells in a tessellation. In R. Miles and J. Serra, editors, *Geometrical Probability and Biological Structures: Buffon's 200th Anniversary*, number 23 in Lecture Notes in Biomathematics, pages 323–329, Berlin-Heidelberg-New York, 1978. Springer-Verlag.
- [65] P. A. Larsen, D. B. Patience, and J. B. Rawlings. Industrial crystallization process control. *IEEE Control Systems Magazine*, 26(4):70–80, August 2006.

- [66] P. A. Larsen and J. B. Rawlings. Assessing the reliability of particle size distribution measurements obtained by image analysis. Submitted to *Particle and Particle Systems Characterization*, June 2007.
- [67] P. A. Larsen and J. B. Rawlings. High-resolution imaging-based PSD measurement for industrial crystallization. Submitted for publication in *AICHE J.*, June 2007.
- [68] P. A. Larsen and J. B. Rawlings. Maximum likelihood estimation of particle size distribution for high-aspect-ratio particles using in situ video imaging. Submitted to *Technometrics*, April 2007.
- [69] P. A. Larsen, J. B. Rawlings, and N. J. Ferrier. An algorithm for analyzing noisy, in situ images of high-aspect-ratio crystals to monitor particle size distribution. *Chemical Engineering Science*, 61(16):5236–5248, 2006.
- [70] P. A. Larsen, J. B. Rawlings, and N. J. Ferrier. Model-based object recognition to measure crystal size and shape distributions from in situ video images. *Chemical Engineering Science*, 62:1430–1441, 2007.
- [71] G. Laslett. The survival curve under monotone density constraints with application to two-dimensional line segment processes. *Biometrika*, 69(1):153–160, 1982.
- [72] F. Lewiner, J. P. Klein, F. Puel, and G. Févotte. On-line ATR FTIR measurement of supersaturation during solution crystallization processes. Calibration and applications on three solute/solvent systems. *Chemical Engineering Science*, 56(6):2069–2084, 2001.
- [73] M. Li and D. Wilkinson. Determination of non-spherical particle size distribution from chord length measurements. Part 1: Theoretical analysis. *Chemical Engineering Science*, 60(12):3251–3265, 2005.
- [74] V. Liotta and V. Sabesan. Monitoring and feedback control of supersaturation using ATR-FTIR to produce an active pharmaceutical ingredient of a desired crystal size. *Organic Process Research & Development*, 8(3):488–494, 2004.
- [75] D. G. Lowe. Three-dimensional object recognition from single two-dimensional images. *Artificial Intelligence*, 31(3):355–395, 1987.
- [76] D. G. Lowe. Fitting parameterized three-dimensional models to images. *IEEE Transactions on Pattern Analysis and Machine Intelligence*, 13(5):441–450, 1991.
- [77] C. Mack. The expected number of clumps when convex laminae are placed at random and with random orientation on a plane area. *Proceedings of the Cambridge Philosophical Society*, 50:581–585, 1954.
- [78] C. Mack. On clumps formed when convex laminae or bodies are placed at random in two or three dimensions. *Proceedings of the Cambridge Philosophical Society*, 52:246–256, 1956.

- [79] D. L. Marchisio, R. D. Vigil, and R. O. Fox. Quadrature method of moments for aggregation-breakage processes. *Journal of Colloid and Interface Science*, 258(2):322–334, 2003.
- [80] F. J. Massey. The Kolmogorov-Smirnov test for goodness of fit. *Journal of the American Statistical Association*, 46(253):68–78, March 1951.
- [81] H. B. Matthews. *Model Identification and Control of Batch Crystallization for an Industrial Chemical System*. PhD thesis, University of Wisconsin-Madison, 1997.
- [82] H. B. Matthews and J. B. Rawlings. Batch crystallization of a photochemical: Modeling, control and filtration. *AIChE Journal*, 44:1119–1127, 1998.
- [83] R. Miles. *Stochastic Geometry*, chapter On the elimination of edge effects in planar sampling, pages 228–247. John Wiley & Sons, 1974.
- [84] S. M. Miller and J. B. Rawlings. Model identification and control strategies for batch cooling crystallizers. *AIChE Journal*, 40(8):1312–1327, August 1994.
- [85] C. A. Mitchell, L. Yu, and M. D. Ward. Selective nucleation and discovery of organic polymorphs through epitaxy with single crystal substrates. *Journal of the American Chemical Society*, 123(44):10830–10839, 2001.
- [86] O. Monnier, G. Fevotte, C. Hoff, and J. P. Klein. Model identification of batch cooling crystallizations through calorimetry and image analysis. *Chemical Engineering Science*, 52(7):1125–1139, 1997.
- [87] M. Moscosa-Santillán, O. Bals, H. Fauduet, C. Porte, and A. Delacroix. Study of batch crystallization and determination of an alternative temperature-time profile by on-line turbidity analysis—application to glycine crystallization. *Chemical Engineering Science*, 55(18):3759–3770, 2000.
- [88] S. Motz, S. Mannal, and E.-D. Gilles. Integral approximation—an approach to reduced models for particulate processes. *Chemical Engineering Science*, 59(5):987–1000, 2004.
- [89] M. Naito, O. Hayakawa, K. Nakahira, H. Mori, and J. Tsubaki. Effect of particle shape on the particle size distribution measured with commercial equipment. *Powder Technology*, 100:52–60, 1998.
- [90] T. Norris, P. K. Aldridge, and S. S. Sekulic. Determination of end-points for polymorph conversions of crystalline organic compounds using on-line near-infrared spectroscopy. *The Analyst*, 122(6):549–552, 1997.
- [91] L. E. O'Brien, P. Timmins, A. C. Williams, and P. York. Use of in situ FT-Raman spectroscopy to study the kinetics of the transformation of carbamazepine polymorphs. *Journal of Pharmaceutical and Biomedical Analysis*, 36(2):335–340, 2004.

- [92] T. Ono, J. ter Horst, and P. Jansens. Quantitative measurement of the polymorphic transformation of L-glutamic acid using in-situ Raman spectroscopy. *Crystal Growth and Design*, 4(3):465–469, 2004.
- [93] B. O’Sullivan, P. Barrett, G. Hsiao, A. Carr, and B. Glennon. In situ monitoring of polymorphic transitions. *Organic Process Research & Development*, 7(6):977–982, 2003.
- [94] D. B. Patience. *Crystal Engineering Through Particle Size and Shape Monitoring, Modeling, and Control*. PhD thesis, University of Wisconsin–Madison, 2002.
- [95] D. B. Patience, P. C. Dell’Orco, and J. B. Rawlings. Optimal operation of a seeded pharmaceutical crystallization with growth-dependent dispersion. *Organic Process Research & Development*, 8(4):609–615, 2004.
- [96] D. B. Patience and J. B. Rawlings. Particle-shape monitoring and control in crystallization processes. *AIChE Journal*, 47(9):2125–2130, 2001.
- [97] K. Pollanen, A. Hakkinen, S.-P. Reinikainen, M. Louhi-Kultanen, and L. Nystrom. A study on batch cooling crystallization of sulphathiazole: process monitoring using ATR-FTIR and product characterization by automated image analysis. *Chemical Engineering Research and Design*, 84(A1):47–59, 2006.
- [98] M. Pons, H. Vivier, K. Belaroui, B. Bernard-Michel, F. Cordier, D. Oulhana, and J. Dodds. Particle morphology: from visualization to measurement. *Powder Technology*, 103:44–57, 1999.
- [99] S. L. Price. The computational prediction of pharmaceutical crystal structures and polymorphism. *Advanced Drug Delivery Reviews*, 56(3):301–319, February 2004.
- [100] F. Puel, P. Marchal, and J. Klein. Habit transient analysis in industrial crystallization using two dimensional crystal sizing technique. *Chemical Engineering Research and Design*, 75(A2):193–205, 1997.
- [101] H. Qu, M. Louhi-Kultanen, and J. Kallas. In-line image analysis on the effects of additives in batch cooling crystallization. *Journal of Crystal Growth*, 289:286–294, 2006.
- [102] A. D. Randolph and M. A. Larson. *Theory of Particulate Processes*. Academic Press, San Diego, second edition, 1988.
- [103] J. B. Rawlings, S. M. Miller, and W. R. Witkowski. Model identification and control of solution crystallization processes: A review. *Industrial and Engineering Chemistry Research*, 32(7):1275–1296, July 1993.
- [104] S. Roach. *The Theory of Random Clumping*. Methuen’s monographs on applied probability and statistics. Methuen & Company, London, 1968.

- [105] S. Rohani and G. Zhang. On-line optimal control of a seeded batch cooling crystallizer. *Chemical Engineering Science*, 58(9):1887–1896, 2003.
- [106] A. L. Rohl. Computer prediction of crystal morphology. *Current Opinion in Solid State & Material Science*, 7(1):21–26, 2003.
- [107] A. Ruf, J. Worlitschek, and M. Mazzotti. Modeling and experimental analysis of PSD measurements through FBRM. *Particle and Particle Systems Characterization*, 17:167–179, 2000.
- [108] B. C. Russell, A. Torralba, K. P. Murphy, and W. T. Freeman. LabelMe: a database and web-based tool for image annotation. MIT AI Lab Memo, AIM-2005-025, September 2005.
- [109] H. Sakai, H. Hosogai, T. Kawakita, K. Onuma, and K. Tsukamoto. Transformation of  $\alpha$ -glycine to  $\gamma$ -glycine. *Journal of Crystal Growth*, 116:421–426, 1992.
- [110] K. Sakamoto and R. W. Rousseau. Sizing elongated crystals using a width distribution function: Application to aspartame. *Industrial and Engineering Chemistry Research*, 39:3949–3952, 2000.
- [111] J. Sanyal, D. L. Marchisio, R. O. Fox, and K. Dhanasekharan. On the comparison between population balance models for CFD simulation of bubble columns. *Industrial and Engineering Chemistry Research*, 44(14):5063–5072, 2005.
- [112] J. Scholl, D. Bonalumi, L. Vicum, M. Mazzotti, and M. Muller. In situ monitoring and modeling of the solvent-mediated polymorphic transformation of L-glutamic acid. *Crystal Growth and Design*, 6(4):881–891, 2006.
- [113] L. Shen, X. Song, M. Iguchi, and F. Yamamoto. A method for recognizing particles in overlapped particle images. *Pattern Recognition Letters*, 21(1):21–30, January 2000.
- [114] D. Shi, N. H. El-Farra, M. Li, P. Mhaskar, and P. D. Christofides. Predictive control of particle size distribution in particulate processes. *Chemical Engineering Science*, 61(1):268–281, 2006.
- [115] G. W. Snedecor and W. G. Cochran. *Statistical Methods*. Iowa State University Press, Ames, Iowa, 8 edition, 1989.
- [116] H. Solomon. *Geometric Probability*. SIAM Publications, Philadelphia, PA, 1978.
- [117] C. Starbuck, A. Spartalis, L. Wai, J. Wang, P. Fernandez, C. M. Lindemann, G. X. Zhou, and Z. Ge. Process optimization of a complex pharmaceutical polymorphic system via in situ Raman spectroscopy. *Crystal Growth and Design*, 2(6):515–522, 2002.
- [118] I. Svensson, S. Sjostedt-De Luna, and L. Bondesson. Estimation of wood fibre length distributions from censored data through an EM algorithm. *Scandinavian Journal of Statistics*, 33:503–522, 2006.

- [119] T. Togkalidou, M. Fujiwara, S. Patel, and R. D. Braatz. Solute concentration prediction using chemometrics and ATR-FTIR spectroscopy. *Journal of Crystal Growth*, 231(4):534–543, 2001.
- [120] C. S. Towler, R. J. Davey, R. W. Lancaster, and C. J. Price. Impact of molecular speciation on crystal nucleation in polymorphic systems: the conundrum of  $\gamma$  glycine and molecular “self poisoning.”. *Journal of the American Chemical Society*, 126:13347–13353, 2004.
- [121] M. J. Van Der Laan. The two-interval line-segment problem. *Scandinavian Journal of Statistics*, 25:163–186, 1998.
- [122] E. W. Van Zwet. Laslett’s line segment problem. *Bernoulli*, 10(3):377–396, 2004.
- [123] J. Villadsen and M. L. Michelsen. *Solution of Differential Equation Models by Polynomial Approximation*. Prentice-Hall, Englewood Cliffs New Jersey, 1978.
- [124] U. Vollmer and J. Raisch. Population balance modelling and H-infinity—controller design for a crystallization process. *Chemical Engineering Science*, 57(20):4401–4414, 2002.
- [125] F. Wang, J. A. Wachter, F. J. Antosz, and K. A. Berglund. An investigation of solvent-mediated polymorphic transformation of progesterone using in situ Raman spectroscopy. *Organic Process Research & Development*, 4(5):391–395, 2000.
- [126] J. D. Ward, D. A. Mellichamp, and M. F. Doherty. Choosing an operating policy for seeded batch crystallization. *AIChE Journal*, 52(6):2046–2054, June 2006.
- [127] S. Watano and K. Miyamoto. Image processing for on-line monitoring of granule size distribution and shape in fluidized bed granulation. *Powder Technology*, 83:55–60, 1995.
- [128] I. Weissbuch, L. Addadi, M. Lahav, and L. Leiserowitz. Molecular recognition at crystal interfaces. *Science*, 253(5020):637–645, 1991.
- [129] I. Weissbuch, M. Lahav, and L. Leiserowitz. Toward stereochemical control, monitoring, and understanding of crystal nucleation. *Crystal Growth and Design*, 3(2):125–150, 2003.
- [130] B. Wijers. *Nonparametric estimation for a windowed line-segment process*. Stichting Mathematisch Centrum, Amsterdam, Netherlands, 1997.
- [131] D. Winn and M. Doherty. A new technique for predicting the shape of solution-grown organic crystals. *AIChE Journal*, 44(11):2501–2514, 1998.
- [132] K. C. Wong and J. Kittler. Recognition of polyhedral objects using triplets of projected spatial edges based on a single perspective image. *Pattern Recognition*, 34:561–586, 2001.
- [133] J. Worlitschek, T. Hocker, and M. Mazzotti. Restoration of PSD from chord length distribution data using the method of projections onto convex sets. *Particle and Particle Systems Characterization*, 22(2):81–98, August 2005.

- [134] J. Worlitschek and M. Mazzotti. Model-based optimization of particle size distribution in batch-cooling crystallization of paracetamol. *Crystal Growth and Design*, 4(5):891–903, 2004.
- [135] M. Wulkow, A. Gerstlauer, and U. Nieken. Modeling and simulation of crystallization processes using PARSIVAL. *Chemical Engineering Science*, 56(7):2575–2588, 2001.
- [136] R. Xu and O. A. D. Guida. Comparison of sizing small particles using different technologies. *Powder Technology*, 132:145–153, 2003.
- [137] L. Yu and K. Ng. Glycine crystallization during spray drying: The pH effect on salt and polymorphic forms. *Journal of Pharmaceutical Sciences*, 91(11):2367–2375, 2002.
- [138] L. X. Yu, R. A. Lionberger, A. S. Raw, R. D'Costa, H. Wu, and A. S. Hussain. Applications of process analytical technology to crystallization processes. *Advanced Drug Delivery Reviews*, 56(3):349–369, 2004.



# Vita

Paul A. Larsen was born in Blackfoot, Idaho to Stephen and Susan Larsen. In June 1996, he graduated as valedictorian of his class from Snake River High School in Blackfoot. After attending Ricks College for one year, Paul spent two years living in El Salvador, volunteering as a missionary for the Church of Jesus Christ of Latter-Day Saints. Upon returning from El Salvador, he completed his Associate's Degree in Chemical Engineering at Ricks College, receiving the Spori Scholar award. Paul transferred to Brigham Young University (BYU) in Provo, UT and graduated *cum laude* in 2002 with a Bachelor of Science degree in Chemical Engineering. During his undergraduate studies, Paul worked part-time and during two summers at Ceramatec in Salt Lake City, UT. In the fall of 2002, he began graduate studies in the Department of Chemical Engineering at the University of Wisconsin-Madison under the direction of James B. Rawlings. Paul will begin work in Separations R&D at Dow Chemical in Midland, Michigan this fall. Paul is married to Jenny Cutler and has three children—Beth, Benjamin, and Sophia.

Permanent Address: 80 N 740 W

Blackfoot, ID 83221

This dissertation was prepared with  $\text{\LaTeX} 2\epsilon$ <sup>1</sup> by the author.

---

<sup>1</sup>This particular University of Wisconsin compliant style was carved from The University of Texas at Austin styles as written by Dinesh Das ( $\text{\LaTeX} 2\epsilon$ ), Khe-Sing The ( $\text{\LaTeX}$ ), and John Eaton ( $\text{\LaTeX}$ ). Knives and chisels wielded by John Campbell and Rock Matthews.