SBML Model Report

Model name: "Lee2003 - Roles of APC and Axin in Wnt Pathway (without regulatory loop)"



May 17, 2018

1. General Overview

This is a document in SBML Level 2 Version 4 format. This model was created by the following two authors: Varun Kothamachu Kothamachu¹ and Matthew Grant Roberts² at March 20th 2018 at 3:52 p. m. and last time modified at March 20th 2018 at 3:52 p. m. Table 1 gives an overview of the quantities of all components of this model.

Table 1: Number of components in this model, which are described in the following sections.

Element	Quantity	Element	Quantity
compartment types	0	compartments	2
species types	0	species	16
events	0	constraints	0
reactions	17	function definitions	3
global parameters	31	unit definitions	3
rules	7	initial assignments	1

Model Notes

Lee2003 - Roles of APC and Axin in WntPathway (without regulatory loop)

¹Babraham Institute, kothamav@babraham.ac.uk

 $^{^2} EMBL\text{-}EBI, \verb|mroberts@ebi.ac.uk|$

This model is described in the article: The roles of APC and Axin derived from experimental and theoretical analysis of the Wnt pathway. Lee E, Salic A, Krger R, Heinrich R, Kirschner MW. PLoS Biol. 2003 Oct; 1(1): E10

Abstract:

Wnt signaling plays an important role in both oncogenesis and development. Activation of the Wnt pathway results in stabilization of the transcriptional coactivator beta-catenin. Recent studies have demonstrated that axin, which coordinates beta-catenin degradation, is itself degraded. Although the key molecules required for transducing a Wnt signal have been identified, a quantitative understanding of this pathway has been lacking. We have developed a mathematical model for the canonical Wnt pathway that describes the interactions among the core components: Wnt, Frizzled, Dishevelled, GSK3beta, APC, axin, beta-catenin, and TCF. Using a system of differential equations, the model incorporates the kinetics of protein-protein interactions, protein synthesis/degradation, and phosphorylation/dephosphorylation. We initially defined a reference state of kinetic, thermodynamic, and flux data from experiments using Xenopus extracts. Predictions based on the analysis of the reference state were used iteratively to develop a more refined model from which we analyzed the effects of prolonged and transient Wnt stimulation on beta-catenin and axin turnover. We predict several unusual features of the Wnt pathway, some of which we tested experimentally. An insight from our model, which we confirmed experimentally, is that the two scaffold proteins axin and APC promote the formation of degradation complexes in very different ways. We can also explain the importance of axin degradation in amplifying and sharpening the Wnt signal, and we show that the dependence of axin degradation on APC is an essential part of an unappreciated regulatory loop that prevents the accumulation of beta-catenin at decreased APC concentrations. By applying control analysis to our mathematical model, we demonstrate the modular design, sensitivity, and robustness of the Wnt pathway and derive an explicit expression for tumor suppression and oncogenicity.

This model is hosted on BioModels Database and identified by: BIOMD0000000658.

To cite BioModels Database, please use: Chelliah V et al. BioModels: ten-year anniversary. Nucl. Acids Res. 2015, 43(Database issue):D542-8.

To the extent possible under law, all copyright and related or neighbouring rights to this encoded model have been dedicated to the public domain worldwide. Please refer to CCO Public Domain Dedication for more information.

2. Unit Definitions

This is an overview of five unit definitions of which two are predefined by SBML and not mentioned in the model.

2.1. Unit volume

Name volume

Definition ml

2.2. Unit time

Name time

Definition 60 s

2.3. Unit substance

Name substance

Definition µmol

2.4. Unit area

Notes Square metre is the predefined SBML unit for area since SBML Level 2 Version 1.

Definition m^2

2.5. Unit length

Notes Metre is the predefined SBML unit for length since SBML Level 2 Version 1.

Definition m

3. Compartments

This model contains two compartments.

Table 2: Properties of all compartments.

			_				
Id	Name	SBO	Spatial Dimensions	Size	Unit	Constant	Outside
Cytoplasm Nucleus	Cytoplasm Nucleus		3 3	1 1	litre litre	1	

3.1. Compartment Cytoplasm

This is a three dimensional compartment with a constant size of one ml.

Name Cytoplasm

3.2. Compartment Nucleus

This is a three dimensional compartment with a constant size of one ml.

Name Nucleus

4. Species

This model contains 16 species. The boundary condition of one of these species is set to true so that this species' amount cannot be changed by any reaction. Section 10 provides further details and the derived rates of change of each species.

Table 3: Properties of each species.

Id	Name	Compartment	Derived Unit	Constant	Boundary Condi- tion
Dsh_i	Dsh_i	Cytoplasm	$\mu \text{mol} \cdot \text{ml}^{-1}$		
Dsh_a	Dsh_a	Cytoplasm	$\mu mol \cdot ml^{-1}$		
APC_axin_GSK3	APC*/axin*/GSK3	Cytoplasm	$\mu mol \cdot ml^{-1}$		
APC_axin_GSK3	APC/axin/GSK3	Cytoplasm	$\mu mol \cdot ml^{-1}$		
GSK3	GSK3	Cytoplasm	$\mu mol \cdot ml^{-1}$		
APC_axin	APC/axin	Cytoplasm	$\mu mol \cdot ml^{-1}$		\Box
APC	APC	Cytoplasm	μ mol·ml ⁻¹		\Box
B_catenin_APC _axinGSK3	B_catenin/APC*/axin*/GSK3	Cytoplasm	$\mu \text{mol} \cdot \text{ml}^{-1}$		
B_cateninAPC _axinGSK3	B_catenin*/APC*/axin*/GSK3	Cytoplasm	$\mu mol \cdot ml^{-1}$		
B_catenin	B_catenin*	Cytoplasm	$\mu mol \cdot ml^{-1}$		
B_catenin_0	B_catenin	Nucleus	$\mu \text{mol} \cdot \text{ml}^{-1}$		
Axin	Axin	Cytoplasm	$\mu \text{mol} \cdot \text{ml}^{-1}$		
TCF	TCF	Nucleus	$\mu \text{mol} \cdot \text{ml}^{-1}$		
B_catenin_TCF	B_catenin/TCF	Nucleus	$\mu \text{mol} \cdot \text{ml}^{-1}$		
B_catenin_APC	B_catenin/APC	Cytoplasm	μ mol·ml ⁻¹		
W	W	Cytoplasm	$\mu \text{mol} \cdot \text{ml}^{-1}$		

5. Parameters

This model contains 31 global parameters.

Table 4: Properties of each parameter.

Id	Name	SBO	Value	Unit	Constant
k1	k1		0.182		
k2	k2		0.018		$\overline{\mathbf{Z}}$
k3	k3		0.050		$\overline{\mathbf{Z}}$
k4	k4		0.267		$ \overline{\mathscr{L}} $
k5	k5		0.133		$ \overline{\mathscr{L}} $
k6	k6		0.091		\square
k_6	k_6		0.909		\checkmark
k7	k7		500.000		\checkmark
$k_{-}7$	k_7		25000.000		
k8	k8		500.000		\checkmark
k8	k_8		60000.000		
k9	k9		206.000		
k10	k10		206.000		\checkmark
k11	k11		0.417		\square
k12	k12		0.423		\square
k13	k13		$2.57 \cdot 10^{-4}$		\square
k14	k14		$8.22 \cdot 10^{-5}$		\checkmark
k15	k15		0.167		\square
k16	k16		500.000		
$k_{-}16$	k_16		15000.000		
k17	k17		500.000		\square
$k_{-}17$	$k_{-}17$		600000.000		
K_7	$K_{-}7$	0000282	50.000		\square
K_8	$K_{-}8$	0000282	120.000		\square
$K_{-}16$	$K_{-}16$	0000282	30.000		\square
$K_{-}17$	$K_{-}17$	0000282	1200.000		\checkmark
lambda	lambda	0000356	0.050		\checkmark
t0	t0		40.000		\checkmark
Dsh0	Dsh0		100.000		\checkmark
Total_B-	Total_B_Catenin		34.984		
$_{ extsf{L}}$ Catenin					
$Total_Axin$	Total_Axin		0.020		

6. Initialassignment

This is an overview of one initial assignment.

6.1. Initialassignment Dsh_i

Derived unit contains undeclared units

Math $Dsh0 - [Dsh_a]$

7. Function definitions

This is an overview of three function definitions.

7.1. Function definition Constant_flux__irreversible

Name Constant flux (irreversible)

Argument v

Mathematical Expression

I (1)

7.2. Function definition function_for_v1

Name function for v1

Arguments k1, x1, [W]

Mathematical Expression

$$k1 \cdot x1 \cdot [W] \tag{2}$$

7.3. Function definition function_for_v3

Name function for v3

Arguments k3, x2, x4

Mathematical Expression

$$k3 \cdot x2 \cdot x4 \tag{3}$$

8. Rules

This is an overview of seven rules.

8.1. Rule Total_B_Catenin

Rule Total_B_Catenin is an assignment rule for parameter Total_B_Catenin:

$$Total_B_Catenin = [B_catenin_APC_axin_GSK3] + [B_catenin_APC_axin_GSK3] + [B_catenin] + [B_catenin_0] + [B_catenin_TCF] + [B_catenin_APC]$$

$$(4)$$

Derived unit $\mu mol \cdot ml^{-1}$

8.2. Rule Total_Axin

Rule Total_Axin is an assignment rule for parameter Total_Axin:

$$Total_Axin = [APC_axin_GSK3] + [APC_axin_GSK3] + [APC_axin] + [B_catenin_APC_axin_GSK3] + [B_catenin_APC_axin_GSK3] + [Axin]$$

$$(5)$$

Derived unit $\mu mol \cdot ml^{-1}$

8.3. Rule k_16

Rule k_16 is an assignment rule for parameter k_16:

$$k_{-}16 = k_{1}6 \cdot K_{-}16$$
 (6)

8.4. Rule k_7

Rule k_7 is an assignment rule for parameter k_7:

$$k_{-}7 = k7 \cdot K_{-}7 \tag{7}$$

8.5. Rule k_17

Rule k_17 is an assignment rule for parameter k_17:

$$k_{-}17 = k_{1}7 \cdot K_{-}17$$
 (8)

8.6. Rule k_8

Rule k_8 is an assignment rule for parameter k_8:

$$k_{-}8 = k8 \cdot K_{-}8 \tag{9}$$

8.7. Rule W

Rule W is an assignment rule for species W:

$$W = \begin{cases} 0 & \text{if time} < t0 \\ \exp(1 \cdot \text{lambda} \cdot (\text{time} - t0)) & \text{otherwise} \end{cases}$$
 (10)

9. Reactions

This model contains 17 reactions. All reactions are listed in the following table and are subsequently described in detail. If a reaction is affected by a modifier, the identifier of this species is written above the reaction arrow.

Table 5: Overview of all reactions

N⁰	Id	Name	Reaction Equation SBO		
1	v1	v1	$Dsh_i \xrightarrow{W} Dsh_a$		
2	v2	v2	$Dsh_a \longrightarrow Dsh_i$		
3	v3	v3	$APC_axin_GSK3 \xrightarrow{Dsh_a} GSK3 + APC_axin$		
4	v4	v4	$APC_axin_GSK3 \longrightarrow APC_axin_GSK3$		
5	v5	v5	APC_axin_GSK3 → APC_axin_GSK3		
6	v6	v6	$GSK3 + APC_axin \rightleftharpoons APC_axin_GSK3$		
7	v7	v7	$APC + Axin \Longrightarrow APC_axin$		
8	v8	v8	APC_axin_GSK3 +		
			B_catenin_0 === B_catenin_APCaxinGSK3		
9	v9	v9	B_catenin_APC_axin_GSK3 → B_catenin_APC_axin_GSK3		
10	v10	v10	B_cateninAPCaxinGSK3 → B_catenin +		
			APC_axin_GSK3		
11	v11	v11	B_{-} catenin $\longrightarrow \emptyset$		
12	v12	v12	$\emptyset \longrightarrow B_{catenin}_{0}$		
13	v13	v13	$B_{\text{catenin}} = 0 \longrightarrow \emptyset$		
14	v14	v14	$\emptyset \longrightarrow Axin$		
15	v15	v15	$Axin \longrightarrow \emptyset$		
16	v16	v16	B_catenin_0+TCF ⇒ B_catenin_TCF		
17	v17	v17	APC + B_catenin_0 \ightharpoonup B_catenin_APC		

9.1. Reaction v1

This is an irreversible reaction of one reactant forming one product influenced by one modifier.

Name v1

Reaction equation

$$Dsh_{-}i \xrightarrow{W} Dsh_{-}a \tag{11}$$

Reactant

Table 6: Properties of each reactant.

Id	Name	SBO
Dsh_i	Dsh_i	

Modifier

Table 7: Properties of each modifier.

Id	Name	SBO
W	W	

Product

Table 8: Properties of each product.

Id	Name	SBO
Dsh_a	Dsh_a	

Kinetic Law

Derived unit contains undeclared units

$$v_1 = \text{vol}(\text{Cytoplasm}) \cdot \text{function_for_v1}(k1, [\text{Dsh_i}], [\text{W}])$$
 (12)

$$function_for_v1(k1,x1,[W]) = k1 \cdot x1 \cdot [W]$$

$$(13)$$

$$function_for_v1(k1,x1,[W]) = k1 \cdot x1 \cdot [W]$$
 (14)

9.2. Reaction v2

This is an irreversible reaction of one reactant forming one product.

Name v2

Reaction equation

$$Dsh_a \longrightarrow Dsh_i$$
 (15)

Reactant

Table 9: Properties of each reactant.

Id	Name	SBO
Dsh_a	Dsh_a	

Product

Table 10: Properties of each product.

Id	Name	SBO
Dsh_i	Dsh_i	

Kinetic Law

Derived unit contains undeclared units

$$v_2 = \text{vol}\left(\text{Cytoplasm}\right) \cdot \text{k2} \cdot [\text{Dsh_a}] \tag{16}$$

9.3. Reaction v3

This is an irreversible reaction of one reactant forming two products influenced by one modifier.

Name v3

Reaction equation

$$APC_axin_GSK3 \xrightarrow{Dsh_a} GSK3 + APC_axin$$
 (17)

Reactant

Table 11: Properties of each reactant.

Id	Name	SBO
APC_axin_GSK3	APC/axin/GSK3	

Modifier

Table 12: Properties of each modifier.

Id	Name	SBO
Dsh_a	Dsh_a	

Products

Table 13: Properties of each product.

Id	Name	SBO
GSK3	GSK3	
${\tt APC_axin}$	APC/axin	

Kinetic Law

Derived unit contains undeclared units

$$v_3 = \text{vol}(\text{Cytoplasm}) \cdot \text{function_for_v3}(\text{k3}, [\text{Dsh_a}], [\text{APC_axin_GSK3}])$$
 (18)

9.4. Reaction v4

This is an irreversible reaction of one reactant forming one product.

Name v4

Reaction equation

$$APC_axin_GSK3 \longrightarrow APC_axin_GSK3$$
 (21)

Reactant

Table 14: Properties of each reactant.

Id	Name	SBO
APC_axin_GSK3	APC/axin/GSK3	

Product

Table 15: Properties of each product.

Id	Name	SBO
APC_axin_GSK3	APC*/axin*/GSK3	

Kinetic Law

Derived unit contains undeclared units

$$v_4 = \text{vol}\left(\text{Cytoplasm}\right) \cdot \text{k4} \cdot \left[\text{APC_axin_GSK3}\right]$$
 (22)

9.5. Reaction v5

This is an irreversible reaction of one reactant forming one product.

Name v5

Reaction equation

$$APC_axin_GSK3 \longrightarrow APC_axin_GSK3$$
 (23)

Reactant

Table 16: Properties of each reactant.

Id	Name	SBO
APCaxinGSK3	APC*/axin*/GSK3	

Product

Table 17: Properties of each product.

Id	Name	SBO
APC_axin_GSK3	APC/axin/GSK3	

Kinetic Law

Derived unit contains undeclared units

$$v_5 = \text{vol}(\text{Cytoplasm}) \cdot \text{k5} \cdot [\text{APC}_\text{axin}_\text{GSK3}]$$
 (24)

9.6. Reaction v6

This is a reversible reaction of two reactants forming one product.

Name v6

Reaction equation

$$GSK3 + APC_axin \Longrightarrow APC_axin_GSK3$$
 (25)

Reactants

Table 18: Properties of each reactant.

Id	Name	SBO
GSK3	GSK3	
APC_axin	APC/axin	

Product

Table 19: Properties of each product.

Id	Name	SBO
APC_axin_GSK3	APC/axin/GSK3	

Kinetic Law

Derived unit contains undeclared units

$$v_6 = vol\left(Cytoplasm\right) \cdot \left(k6 \cdot [GSK3] \cdot [APC_axin] - k_6 \cdot [APC_axin_GSK3]\right) \tag{26}$$

9.7. Reaction v7

This is a reversible reaction of two reactants forming one product.

Name v7

Reaction equation

$$APC + Axin \Longrightarrow APC - axin$$
 (27)

Reactants

Table 20: Properties of each reactant.

Id	Name	SBO
APC	APC	
Axin	Axin	

Product

Table 21: Properties of each product.

Id	Name	SBO
APC_axin	APC/axin	

Kinetic Law

Derived unit contains undeclared units

$$v_7 = \text{vol}\left(\text{Cytoplasm}\right) \cdot \left(\text{k7} \cdot [\text{APC}] \cdot [\text{Axin}] - \text{k_7} \cdot [\text{APC_axin}]\right) \tag{28}$$

9.8. Reaction v8

This is a reversible reaction of two reactants forming one product.

Name v8

Reaction equation

$$APC_axin_GSK3 + B_catenin_0 \Longrightarrow B_catenin_APC_axin_GSK3$$
 (29)

Reactants

Table 22: Properties of each reactant.

Id	Name	SBO
APCaxinGSK3	APC*/axin*/GSK3	
$B_{catenin_0}$	B_catenin	

Product

Table 23: Properties of each product.

Id	Name	SBO
B_catenin_APCaxinGSK3	B_catenin/APC*/axin*/GSK3	

Kinetic Law

Derived unit contains undeclared units

$$v_8 = k8 \cdot [APC_axin_GSK3] \cdot [B_catenin_0] - k_8 \cdot [B_catenin_APC_axin_GSK3]$$
 (30)

9.9. Reaction v9

This is an irreversible reaction of one reactant forming one product.

Name v9

Reaction equation

$$B_catenin_APC_axin_GSK3 \longrightarrow B_catenin_APC_axin_GSK3$$
 (31)

Reactant

Table 24: Properties of each reactant.

Id	Name	SBO
B_catenin_APCaxinGSK3	B_catenin/APC*/axin*/GSK3	

Product

Table 25: Properties of each product.

Id	Name	SBO
B_catenin_APC_axin_GSK3	B_catenin*/APC*/axin*/GSK3	

Kinetic Law

Derived unit contains undeclared units

$$v_9 = \text{vol}(\text{Cytoplasm}) \cdot \text{k9} \cdot [\text{B_catenin_APC_axin_GSK3}]$$
 (32)

9.10. Reaction v10

This is an irreversible reaction of one reactant forming two products.

Name v10

Reaction equation

$$B_{\text{catenin}} - APC_{\text{-}axin} - GSK3 \longrightarrow B_{\text{catenin}} + APC_{\text{-}axin} - GSK3$$
 (33)

Reactant

Table 26: Properties of each reactant.

Id	Name	SBO
B_catenin_APC_axin_GSK3	B_catenin*/APC*/axin*/GSK3	

Products

Table 27: Properties of each product.

Id	Name	SBO
B_catenin APCaxinGSK3	B_catenin* APC*/axin*/GSK3	

Kinetic Law

Derived unit contains undeclared units

$$v_{10} = \text{vol}\left(\text{Cytoplasm}\right) \cdot \text{k10} \cdot \left[\text{B_catenin_APC_axin_GSK3}\right]$$
 (34)

9.11. Reaction v11

This is an irreversible reaction of one reactant forming no product.

Name v11

Reaction equation

$$B_{\text{-}catenin} \longrightarrow \emptyset$$
 (35)

Reactant

Table 28: Properties of each reactant.

Id	Name	SBO
B_{-} catenin	B_catenin*	

Kinetic Law

Derived unit contains undeclared units

$$v_{11} = \text{vol}(\text{Cytoplasm}) \cdot \text{k11} \cdot [\text{B_catenin}]$$
 (36)

9.12. Reaction v12

This is an irreversible reaction of no reactant forming one product.

Name v12

Reaction equation

$$\emptyset \longrightarrow B_catenin_0$$
 (37)

Product

Table 29: Properties of each product.

Id	Name	SBO
B_catenin_0	B_catenin	

Kinetic Law

Derived unit contains undeclared units

$$v_{12} = \text{vol}(\text{Nucleus}) \cdot \text{Constant_flux_irreversible}(\text{k}12)$$
 (38)

$$Constant_flux_irreversible(v) = v$$
 (39)

Constant_flux__irreversible
$$(v) = v$$
 (40)

9.13. Reaction v13

This is an irreversible reaction of one reactant forming no product.

Name v13

Reaction equation

$$B_{\text{catenin}} = 0 \longrightarrow \emptyset$$
 (41)

Reactant

Table 30: Properties of each reactant.

Id	Name	SBO
B_catenin_0	B_catenin	

Kinetic Law

Derived unit contains undeclared units

$$v_{13} = \text{vol}(\text{Nucleus}) \cdot \text{k13} \cdot [\text{B_catenin_0}]$$
 (42)

9.14. Reaction v14

This is an irreversible reaction of no reactant forming one product.

Name v14

Reaction equation

$$\emptyset \longrightarrow Axin$$
 (43)

Product

Table 31: Properties of each product.

Id	Name	SBO
Axin	Axin	

Kinetic Law

Derived unit contains undeclared units

$$v_{14} = \text{vol}\left(\text{Cytoplasm}\right) \cdot \text{Constant_flux_irreversible}\left(\text{k14}\right)$$
 (44)

Constant_flux_irreversible
$$(v) = v$$
 (45)

Constant_flux_irreversible
$$(v) = v$$
 (46)

9.15. Reaction v15

This is an irreversible reaction of one reactant forming no product.

Name v15

Reaction equation

$$Axin \longrightarrow \emptyset \tag{47}$$

Reactant

Table 32: Properties of each reactant.

Id	Name	SBO
Axin	Axin	

Kinetic Law

Derived unit contains undeclared units

$$v_{15} = \text{vol}\left(\text{Cytoplasm}\right) \cdot \text{k15} \cdot [\text{Axin}]$$
 (48)

9.16. Reaction v16

This is a reversible reaction of two reactants forming one product.

Name v16

Reaction equation

$$B_{\text{catenin}} - 0 + \text{TCF} \Longrightarrow B_{\text{catenin}} - \text{TCF}$$
 (49)

Reactants

Table 33: Properties of each reactant.

Id	Name	SBO
B_catenin_0 TCF	B_catenin TCF	

Product

Table 34: Properties of each product.

Id	Name	SBO
B_catenin_TCF	B_catenin/TCF	

Kinetic Law

Derived unit contains undeclared units

$$v_{16} = \text{vol}\left(\text{Nucleus}\right) \cdot \left(\text{k16} \cdot \left[\text{B_catenin_0}\right] \cdot \left[\text{TCF}\right] - \text{k_16} \cdot \left[\text{B_catenin_TCF}\right]\right)$$
 (50)

9.17. Reaction v17

This is a reversible reaction of two reactants forming one product.

Name v17

Reaction equation

$$APC + B_catenin_0 \Longrightarrow B_catenin_APC$$
 (51)

Reactants

Table 35: Properties of each reactant.

Id	Name	SBO
APC	APC	
$B_{-} catenin_{-} 0$	B_catenin	

Product

Table 36: Properties of each product.

Id	Name	SBO
B_catenin_APC	B_catenin/APC	

Kinetic Law

Derived unit contains undeclared units

$$v_{17} = k17 \cdot [APC] \cdot [B_catenin_0] - k_17 \cdot [B_catenin_APC]$$
 (52)

10. Derived Rate Equations

When interpreted as an ordinary differential equation framework, this model implies the following set of equations for the rates of change of each species.

Identifiers for kinetic laws highlighted in gray cannot be verified to evaluate to units of SBML substance per time. As a result, some SBML interpreters may not be able to verify the consistency of the units on quantities in the model. Please check if

- parameters without an unit definition are involved or
- volume correction is necessary because the hasOnlySubstanceUnits flag may be set to false and spacialDimensions > 0 for certain species.

10.1. Species Dsh_i

Name Dsh_i

Initial concentration $100 \ \mu mol \cdot ml^{-1}$

Initial assignment Dsh_i

This species takes part in two reactions (as a reactant in v1 and as a product in v2).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{Dsh.i} = |v_2| - |v_1| \tag{53}$$

10.2. Species Dsh_a

Name Dsh_a

Initial concentration $0 \, \mu mol \cdot ml^{-1}$

This species takes part in three reactions (as a reactant in v2 and as a product in v1 and as a modifier in v3).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{Dsh}_{-a} = |v_1| - |v_2| \tag{54}$$

10.3. Species APC_axin_GSK3

Name APC*/axin*/GSK3

Initial concentration $0.00966~\mu mol \cdot ml^{-1}$

This species takes part in four reactions (as a reactant in v5, v8 and as a product in v4, v10).

$$\frac{d}{dt}APC_{-axin_{-}GSK3} = |v_4| + |v_{10}| - |v_5| - |v_8|$$
 (55)

10.4. Species APC_axin_GSK3

Name APC/axin/GSK3

Initial concentration $0.00483 \ \mu mol \cdot ml^{-1}$

This species takes part in four reactions (as a reactant in v3, v4 and as a product in v5, v6).

$$\frac{\mathrm{d}}{\mathrm{d}t} APC_{-}axin_{-}GSK3 = v_5 + v_6 - v_3 - v_4$$
 (56)

10.5. Species GSK3

Name GSK3

Initial concentration 50 µmol·ml⁻¹

This species takes part in two reactions (as a reactant in v6 and as a product in v3).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{GSK3} = |v_3| - |v_6| \tag{57}$$

10.6. Species APC_axin

Name APC/axin

Initial concentration $9.8065 \cdot 10^{-4} \ \mu mol \cdot ml^{-1}$

This species takes part in three reactions (as a reactant in v6 and as a product in v3, v7).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{APC}_{-}\mathrm{axin} = |v_3| + |v_7| - |v_6| \tag{58}$$

10.7. Species APC

Name APC

Initial concentration $98 \mu mol \cdot ml^{-1}$

This species takes part in two reactions (as a reactant in v7, v17).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{APC} = -|v_7| - |v_{17}| \tag{59}$$

10.8. Species B_catenin_APC_axin_GSK3

Name B_catenin/APC*/axin*/GSK3

Initial concentration $0.00202 \ \mu mol \cdot ml^{-1}$

This species takes part in two reactions (as a reactant in v9 and as a product in v8).

$$\frac{\mathrm{d}}{\mathrm{d}t} \mathbf{B}_{-} \operatorname{catenin}_{-} \mathbf{APC}_{-} \operatorname{axin}_{-} \mathbf{GSK3} = |v_8| - |v_9| \tag{60}$$

10.9. Species B_catenin_APC_axin_GSK3

Name B_catenin*/APC*/axin*/GSK3

Initial concentration $0.00202 \ \mu mol \cdot ml^{-1}$

This species takes part in two reactions (as a reactant in v10 and as a product in v9).

$$\frac{d}{dt}B_{\text{catenin}} - APC_{\text{-}axin} - GSK3 = v_9 - v_{10}$$
 (61)

10.10. Species B_catenin

Name B_catenin*

Initial concentration $1 \ \mu mol \cdot ml^{-1}$

This species takes part in two reactions (as a reactant in v11 and as a product in v10).

$$\frac{\mathrm{d}}{\mathrm{d}t} \mathbf{B}_{-} \operatorname{catenin} = |v_{10}| - |v_{11}| \tag{62}$$

10.11. Species B_catenin_0

Name B_catenin

Initial concentration 25.1 μmol·ml⁻¹

This species takes part in five reactions (as a reactant in v8, v13, v16, v17 and as a product in v12).

$$\frac{d}{dt}B_{\text{catenin}} = |v_{12} - v_8| - |v_{13}| - |v_{16}| - |v_{17}|$$
(63)

10.12. Species Axin

Name Axin

Initial concentration $4.93 \cdot 10^{-4} \ \mu mol \cdot ml^{-1}$

This species takes part in three reactions (as a reactant in v7, v15 and as a product in v14).

$$\frac{d}{dt}Axin = |v_{14}| - |v_7| - |v_{15}| \tag{64}$$

10.13. Species TCF

Name TCF

Initial concentration $8.17 \, \mu mol \cdot ml^{-1}$

This species takes part in one reaction (as a reactant in v16).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathrm{TCF} = -v_{16} \tag{65}$$

10.14. Species B_catenin_TCF

Name B_catenin/TCF

Initial concentration $6.83 \ \mu mol \cdot ml^{-1}$

This species takes part in one reaction (as a product in v16).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathbf{B}_{-}\mathrm{catenin}_{-}\mathrm{TCF} = v_{16} \tag{66}$$

10.15. Species B_catenin_APC

Name B_catenin/APC

Initial concentration 2.05 µmol⋅ml⁻¹

This species takes part in one reaction (as a product in v17).

$$\frac{\mathrm{d}}{\mathrm{d}t}\mathbf{B}_{-}\mathrm{catenin}_{-}\mathrm{APC} = v_{17} \tag{67}$$

10.16. Species W

Name W

Initial concentration $0 \, \mu \text{mol} \cdot \text{ml}^{-1}$

Involved in rule W

This species takes part in one reaction (as a modifier in v1). Not this but one rule determines the species' quantity because this species is on the boundary of the reaction system.

A. Glossary of Systems Biology Ontology Terms

SBO:0000282 dissociation constant: Equilibrium constant that measures the propensity of a larger object to separate (dissociate) reversibly into smaller components, as when a complex falls apart into its component molecules, or when a salt splits up into its component ions. The dissociation constant is usually denoted Kd and is the inverse of the affinity constant.

SBO:0000356 decay constant: Kinetic constant characterising a mono-exponential decay. It is the inverse of the mean lifetime of the continuant being decayed. Its unit is "per tim".

SBML2LATEX was developed by Andreas Dräger^a, Hannes Planatscher^a, Dieudonné M Wouamba^a, Adrian Schröder^a, Michael Hucka^b, Lukas Endler^c, Martin Golebiewski^d and Andreas Zell^a. Please see http://www.ra.cs.uni-tuebingen.de/software/SBML2LaTeX for more information.

^aCenter for Bioinformatics Tübingen (ZBIT), Germany

^bCalifornia Institute of Technology, Beckman Institute BNMC, Pasadena, United States

^cEuropean Bioinformatics Institute, Wellcome Trust Genome Campus, Hinxton, United Kingdom

^dEML Research gGmbH, Heidelberg, Germany