## 1 Percent

## Example:

What is 28% of 286?

Step 1. Change the 28% to a decimal equivalent:

$$28\% \div 100 = 0.28$$

Step 2. Multiply  $286 \times 0.28 = 80$ Thus 28% of 286 is 80.

**Example:** A filter bed will expand 25% during backwash. If the filter bed is 36 inches deep, how deep will it be during backwash?

Step 1. Change the percent to a decimal.

$$25\% \div 100 = 0.25$$

Step 2. Add the whole number 1 to this value.

$$1 + 0.25 = 1.25$$

Step 3. Multiply times the value.

$$36 \text{ in } \times 1.25 = 45 \text{ inches}$$

1. What is 20% of 250?

Solution:

$$20\% = \frac{20}{100} = 0.2 \implies 20\% \text{ of } 250 = 0.2 * 250 = 50$$

2. What percent is 0.4 of 4?

Solution:

$$x\% = \frac{x}{100} \implies 0.4 = \frac{x}{100} * 4 \implies x = \frac{0.4 * 100}{4} = \boxed{10\%}$$

## 1.1 Percentage Concentrations

**Example 1:** A chlorine solution was made to have a 4% concentration. It is often desirable to determine this concentration in mg/L. This is relatively simple: the 4% is four percent of a million.

To find the concentration in mg/L when it is expressed in percent, do the following:

1. Change the percent to a decimal.

$$4\% \div 100 = 0.04$$

3. Multiply times a million.

$$0.04 \times 1,000,000 = 40,000 \text{mg/L}$$

We get the million because a liter of water weighs 1,000,000mg.1mg in 1 liter is 1 part in a million parts (ppm).1% = 10,000mg/L.

**Example 2:** How much 65% calcium hypochlorite is required to obtain 7 pounds of pure chlorine? 65% implies that in every lb of calcium hypochlorite has 65% lbs of available chlorine.

Therefore,  $\frac{0.65 \text{ lbs available chlorine}}{\text{lb of calcium hypochlorite}}$  or conversely  $\frac{\text{lb of calcium hypochlorite}}{0.65 \text{ lbs available chlorine}}$ 

- $\implies$  lbs calcium hypchlorite required =  $\frac{\text{lb of calcium hypochlorite}}{0.65 \text{ lbs available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}{7 \text{ lb of available chlorine}} * \frac{7 \text{ lb of available chlorine}}$
- = 10.8 lbs of calcium hypochlorite with 65% available chlorine is required
  - 1. 25% of the chlorine in a 30-gallon vat has been used. How many gallons are remaining in the vat? Solution:

Amount of chlorine remaining in the vat is 100%-25%=75%Gallons of chlorine remaining in the vat: 30\*0.75 = 22.5 gallons

2. The annual public works budget is \$147,450. If 75% of the budget should be spent by the end of September, how many dollars are to be spent? How many dollars will be remaining?

Solution:

Amount to be spent = 
$$$147,450*0.75 = $110,812.50$$

Amount remaining = 
$$147,450 - 110,812.50 = 536,367.50$$

3. A 75 pound container of calcium hypochlorite has a purity of 67%. What is the actual weight of the calcium hypochlorite in the container?

Solution:

Note: Calcium Hypochlorite can be written as 
$$Ca(OCl)_2$$
  
75 lbs  $Ca(OCl)_2$  - product in container\*  $\frac{0.67 \ lbs \ Ca(OCl)_2}{lb \ Ca(OCl)_2 \ - \ product \ in \ container} = \boxed{50.25 \ lbs \ Ca(OCl)_2}$ 

4. 3/4 is the same as what percentage?

$$\frac{3}{4} = 0.75$$
 which is  $\frac{75}{100} = \boxed{75\%}$ 

5. An operator mixes 40 lb of lime in a 100-gal tank containing 80 gal of water. What is the percent of lime in the slurry? Solution: