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4] ECTEINASCIDINS 736 AND 722 Nakagawa et al., J. Amer. Chem. Soc., 111: 2721-2722 (1989).

both of Urbana, Ill.

The Merck Index, 11th Edition, page 1540, Monograph No. 9705 (1989).

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[57] ABSTRACT

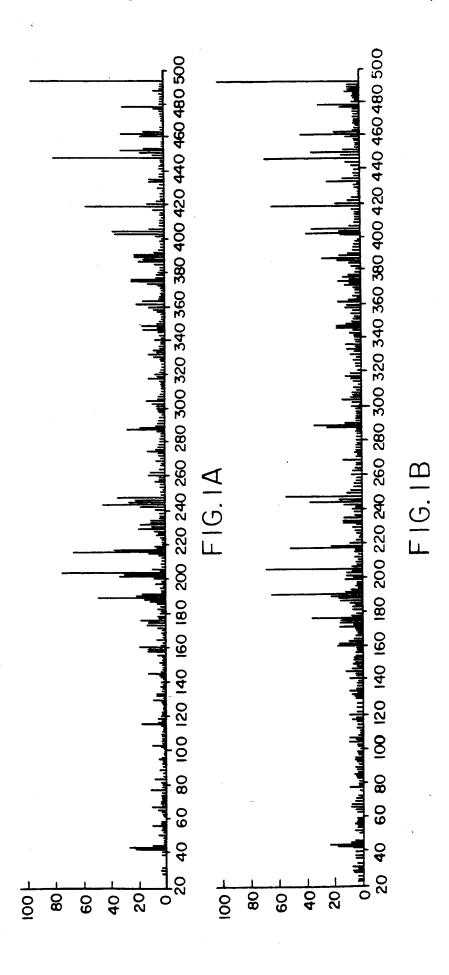
Ecteinascidins 722 and 736 (Et's 722 and 736) have been isolated from the Caribbean tunicate *Ecteinascidia turbinata* and their structures have been assigned as tetrahydro-β-carboline-substituted bis(tetrahydro-isoquinolines) related to the previously reported Et's 729 and 743. Et's 722 and 736 protect mice in vivo at very low concentrations against P388 lymphoma, B16 melanoma, and Lewis lung carcinoma.

3 Claims, 3 Drawing Sheets

[54] ECTEINASCIDINS 736 AND 722 [75] Inventors: Kenneth Rinehart; Ryuichi Sakai, [73] Assignee: The Board of Trustees of the University of Illinois, Urbana, Ill. [21] Appl. No.: 620,427 [22] Filed: Nov. 30, 1990 [51] Int. Cl.⁵ C07D 515/22 U.S. Cl. 540/466 [58] Field of Search 540/466 References Cited [56] **PUBLICATIONS** Rinehart et al., Journal of Natural Products, vol. 53,

No. 4, pp. 771-792 Jul. Aug. 1990. Rinehart et al., Pure & Appl. Chem., vol. 62, No. 7, pp. 1277-1280 (1990). Wright et al., J. Org. Chem., 55, 4508-4512 (1990). Rinehart et al., J. Org. Chem., 55, 4512-4515, (1990). Rinehart et al., Chem. Abstr., 109: 811j (1988).

Ito, CRC Crit. Rev. Anal. Chem., 17: 65-143 (1986).



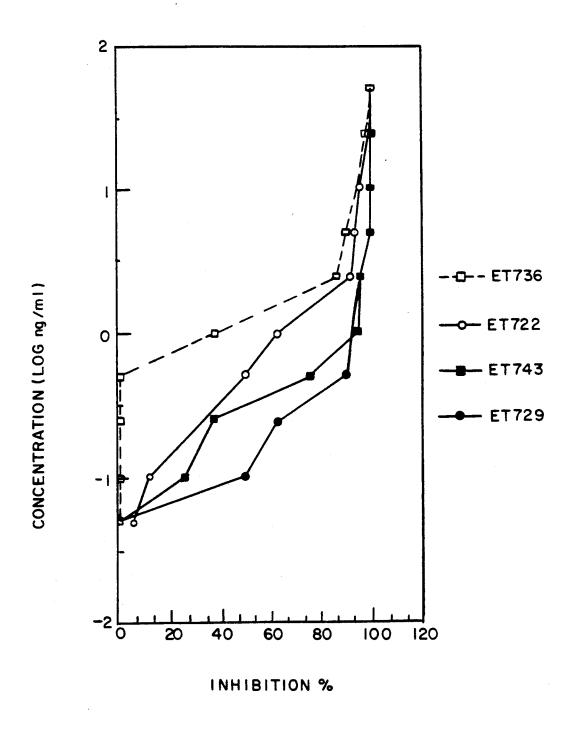
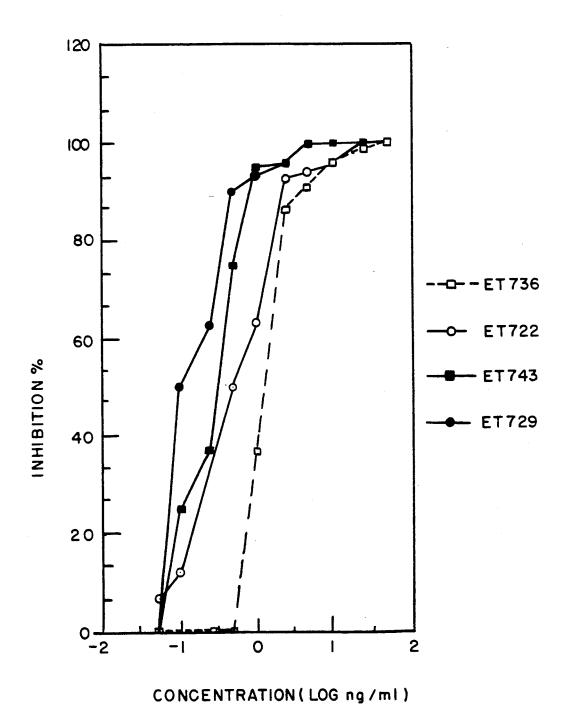


FIG. 2



F I G. 3

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ECTEINASCIDINS 736 AND 722

STATEMENT OF GOVERNMENT SUPPORT

This invention was supported in part by a grant from the National Institute of Allergy and Infectious Diseases (No. AI04769). Mass spectra were obtained in the Mass Spectrometry Laboratory, School of Chemical Sciences, University of Illinois, and supported in part by a grant from the National Institute of General Medical Sciences (No. GM27029).

CROSS REFERENCE TO RELATED APPLICATION

This application describes compounds related to 15 those described in copending U.S. Pat. Application Ser. No. 07/548.060. filed Jul. 5, 1990, the disclosure of which is incorporated herein by reference.

BACKGROUND OF THE INVENTION

Rinehart et al. have recently reported on ecteinascidins (Et's) 729 (hydrated molecular weight, 747), 743 (761), 745. 759A,B (777). 770, and their derivatives O-methyl-Et 729 and O-methyl-Et 743. See for exam- 25 CH ple, J. Orz. Chem., 1990, 55, 4512-4515; Topics in Pharmaceutical Sciences 1989, Amsterdam Medical Press, 1989, pp. 613-626; J. Nat. Prod., 1990, 53, 771-792; Biological Mass. Spectrometry, Elsevier 1990, pp. 233-258; and Pure Appl. Chem., 1990, 62, 1277-1280. 30 Two of those compounds (Et 729, 743) have also been described by others. (See for example, Wright et al., J. Org. Chem., 1990, 55, 4508-4512).

The major component, ecteinascidin 743 (Et 743, (Rinehart et al.. J. Orz. Chem., 1990, 55, 4512-4515), and the others were assigned tris(tetrahydroisoquinoline) structures by correlation NMR techniques, as well as by fast atom bombardment (FAB)MS and tandem MS (FABMS/MS). Among these potent antitumor agents, 40 Et 729 showed especially promising activities vs. tumor cells, but only minute quantities of pure sample were obtained. See for example, Rinehart et al., Topics in Pharmaceutical Sciences 1989 pp. 613-626, Amsterdam Medical Press B.V., The Netherlands, (1989), Holt et 45 on a fragment ion at m/z 493 of Et 743. al., Diss. Abstr. Int. B. 47, 3771-3772 (1987) and Rinehart et al., U.S. Pat. Appln. Ser. No. 872,189, filed Jun. 9, 1986; PCT Intl. Appln. W087 07,610, filed Dec. 17, 1987; Chem. Abstr., 109, 811j, (1988).

The need for further biological evaluation promoted 50 the development of a more efficient large-scale isolation procedure. During that process, two new biologically active ecteinascidins; Et 736 (754) and Et 722 (740), were isolated from Ecteinascidia turbinata samples collected at various locations in the Caribbean.

SUMMARY OF THE INVENTION

The present invention is directed to the isolation of two new compounds, Et 736 and 722 from E. turbinata, 60 together with assignment of their structures and biological activities. The data reported herein support our previously proposed biogiogenetic pathway. See, Rinehart et al., J. Org. Chem., supra.

Thus, the present invention is directed to the follow- 65 ing new compounds 3, 4 and 5. Compound 1 and 2, Et 743 and 729, respectively, are shown for comparison purposes.

*stereochemistry not determined

HN

BRIEF DESCRIPTION OF THE DRAWINGS

FIG. 1A (top) is the mass spectrum MS/MS (FAB),

FIG. 1B (bottom) is the mass spectrum MS/MS (FAB), on a fragment ion at m/z 493 of Et 736.

FIGS. 2 and 3 are graphs illustrating the percentage of inhibition of the growth of L1210 cells by Et 736 (---) and Et 722 (-). Also shown on the graphs for comparison are the inhibitory effects of Et 743 (...) and Et 729 (—).

DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENTS

Samples of E. turbinata were collected in the Florida Keys and Belize, and their cytotoxic extracts were separated by using solvent partition, countercurrent chromatography (CCC), (see for example, Y. Ito, CRC Crit. Rev. Anal. Chem., 17, 65-143, (1986)) and normal and reversed-phase (RP) gravity columns. Final purifications were carried out by (C-18) RP-HPLC.

All samples examined contained Et's 743 (761), 729 (747), 736 (754), and 722 (740) in various proportions. Et 736 (754), $[\alpha]_D = 76^{\circ}$ (c 0.53, CHCl₃), showed "molecular" ions at m/z 753.2588 (C₄₀H₄₁N₄O₉S, Δ0.6 mmu, M - H, negative ion HRFABMS). Et 722 (740), $[\alpha]_D$ =-40° (c 1.64, CHCl₃), showed "molecular" ions at

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m/z 739.2433 $C_{39}H_{39}N_4O_8S$, $\Delta-0.7$ mmu, $M_-+H_ H_2O$, positive ion HRFABMS). 1H and ^{13}C NMR spectra for Et 722 (740) vis-a-vis Et 736 (754) lacked an N^{12} — CH_3 signal and showed an upfield shift for the adjacent carbons C-11 and C-13 (Table 1), indicating 5 that Et 722 (740) was the N^{12} -demethyl derivative of Et 736 (754).

Comparison of NMR data (Table 1) for these new compounds with those for Et 743 (761) and Et 729 (747) indicated that the bis(tetrahydroisoquinoline) units A and B are the same in Et 736 (754) and Et 722 (740) as in the earlier Et's. This was also supported by NMR correlation spectroscopy data, including COSY, phasesensitive COSY, CSCM, and COLOC sequences, although some of the expected correlations were missing to the broad peaks observed.

Et 743
744.2648
495.2126
493.1980
477.1978
463.1837
218.1174

fragmentation patterns. Important fragmentation ions for the A-B bis(tetrayhydroisoquinoline) unit observed for Et 743 were also seen for Et 736 (Table II).

TABLE II

Comparison of HRFABMS Fragmentation Data for

Et 743 and Et 736 Observed ions Et 743 Et 736 Formula Fragment $(M + H - H_2O)$ $(M + H - H_2O)$ 744.2648 C39H42N3O10S 737.2655 C40H40N4O8S 523.2011 523.1960 C28H31N2O8 a + 2H495.2126 495.2126 C27H31N2O7 b + 2H493.1980 493.1980 C27H29N2O7 ь 477.1978 477.2024 C27H29N2O6 463.1837 463.1862 C₂₆H₂₇N₂O₆ d 218.1180 $C_{13}H_{16}NO_2$ g

TABLE 1

	¹ H and ¹³ C NME Data for Et's 743, 729, 736, and 722 (1-4) in CD ₃ OD—CDCl ₃ (3:1) Chemical shift (δ), multiplicity ^a (j in Hz)									
Carbon			Cnemica		olicity" (j. i					
or Proton ^b	13 _C	1 1 _H	13C	2 1H	13Cc	3 1H	13C	4 1H		
1	56.3, d	4.78, br s	56.8, d	4.69, br s	54.8, d	4.71, br s	56.7, d	4.72, br s		
3	58.8, d	3.72 ^c	57.1, d	3.72 br d (5.5)	57.8, d	3.76, br s	58.5, d	3.53, d (4.5)		
4	42.7, d	4.58, br s	42.5, d	4.58, br s	42.3, d	4.58, br s	43.1, d	4.50, br s		
5	142.2, s	,	142.3, s	,	140.8, s	,	141.9, s ^d	,		
6	113.9, s		114.0, s		112.6, s		113.4, s			
7	146.5, s ^d		146.6, s ^d		145.4, s		146.8, s			
8	141.9, s		141.5, s		140.5, s		142.1, s ^d			
9	116.0, s	•	115.5, s		115.4, s		115.9, s			
10	122.0, s		121.4, s		120.9, s		121.8, s			
11	55.6, d	4.40, br d (3.5)	47.8, d	4.73 (5.0)	54.5, d	4.73, br s	48.0, d	4.44, d (4.8)		
13	54.0, d	3.12, br s	47.2, d	3.94, d (10.0)	52.6, d	3.90, br s	46.9, d	3.57, br d (9.0)		
14	24.5, t	2.51, 2H, br d (4.5).	25.1, t	3.22. d (18.0)	23.2, t	3.30 ^c	27.9, t	3.15, d (17.7)		
	2, .	2.01, 2.11, 0.1 0 (1.0)	20,	3.12, dd (9.8, 18.0)	25.2, 1	3.08, dd (10.0, 19.0)	27.5, 1	3.01, dd (17.7, 9.3		
15	120.9, d	6.15, s	121.2. d	6.62, s	120.3, d	6.70, s	121.0, d	6.59, s		
16	131.2. s		130.6, s		130.3, s	• •	131.4, s	,		
17	145.1, s		144.9, s	•	143.0, s		144.4, s			
18	149.8, s		148.6, s		148.2, s		148.2, s			
19	119.2, s		120.2, s		118.5, s		124.3, s			
20	131.5, s		132.7, s		130.9, s		132.2, s			
21	92.1, d	4.26, d (3.0)	90.1, d	4.33, d (3.0)	91.5, d	4.46, d (2.4)	91.1, d	4.12, s		
22	61.2, t	5.14, d (11.0)	61.5, t	5.15, d (11.0)	62.0, t	5.20, dd (12.5, 0.5)	61.6, t	5.17, d (11.1)		
		4.09, dd (11.0, 2.0)	01.5, (4.11, dd (2.5, 11.0)	02.0, 1	4.18, dd (12.5, 1.5)	01.0, 1	4.14, dd (11.4, 1.2		
OCH ₂ O	103.1, t	6.07, d (1.0) 5.98, d (1.0)	103.1, t	6.09, d (0.5) 6.00, d (0.5)	101.7, t	6.26, d (1.0) 6.07, d (1.0)	103.1, t	6.21, d (1.0) 6.04, d (1.0)		
1'	65.3, s		65.2, s	. , ,	61.9, s	, , ,	63.1, s	, , , , ,		
3'	40.3, t	3.13, dt (11.0, 4.0)	40.4, t	3.12, m	39.6, t	3.30 ^c	40.6, t	3.30, m		
	,	2.77, ddd (3.5, 5.5, 11.0)		2.77, m	,	2.90, dt (11.5, 4.5)	, .	2.86, m		
4'	28.6, t	2.60, ddd (5.5, 10.5, 16.0) 2.42, ddd (3.5, 3.5, 16.0)	28.6, t	2.60, ddd (5.5, 10.5, 16.0) 2.42, ddd (3.5, 3.5, 16.0)	20.9, t	2.63, m (2H)	21.6, t	2.61, m (2H)		
5'	115.6, d	6.38, s	115.7, d	6.39, s	116.9, d	7.33, d (8.0	118.8, d	7.31, d (7.8)		
6'	146.4, s ^d	<i>,</i> -	146.4, s ^d		117.7, d	6.92, dt (8.0)	119.6, d	6.91, dt (0.9, 7.8)		
7'	146.4, s ^d		146.4, s ^d		120.7, d	7.12, dt (0.6, 8.0)	122.5, d	7.00, dt (0.9, 7.8)		
8′	111.3, d	6.42, br s	111.3, d	6.41, br s	111.7, d	7.29, d (0.6, 8.0)	111.9, d	7.21, d (7.8)		
9'	125.4, s	,	125.2, s	0.11, 0.10	126.6, s	1125, 2 (0.0, 0.0)	127.3, s	, a (7.0)		
10'	128.8, s		129.0, s		135.6, s		137.4, s			
11'	173.1, s		173.2, s		171.2, s		172.5, s			
12'	43.1, t	2.33, br d (15.5)	42.7, t	2.40	38.9, t	2.78, d (15.6)	39.9, t	2.74, d (15.0)		
13'	.2.2, .	2.05	, .	2.07/	129.3, s	2.15, br d (15.3)	130.8, s	2.12, br d (15.0)		
14'					129.5, s		109.8, s			
C = 0	1698 6		169.8, s		169.5, s		109.8, S 170.7, s			
5 OAc	20.5, q	2.29, s	20.5, q	2.30, s		228 6		2 27 .		
6 CH ₃	20.3, q 9.9, q	2.29, s 2.01, s			20.3, q	2.28, s	20.6, q	2.27, s		
16 CH ₃	16.1, q	2.01, s 2.28, s	9.8, q	2.02, s	9.4, q	2.02, s	9.7, q	2.01, s		
			16.1, q	2.29, s	15.7, q	2.37, s	16.2, q	2.32, s		
7 OCH ₃	60.2, q	3.73, s 3.58, s	60.3, q	3.71, s	60.0, q	3.76, s	60.3, q	3.72, s		
	55.7, q	J.J0, 3	55.6, q	3.58, s		2.49, br s				

as = singlet, d = doublet, t = triplet, q = quartet, br = broad. Proton assignments are based on COSY and homonuclear decoupling experiments; carbon multiplicities were determined by APT and DEPT spectra. Carbons for 4 were assigned by analogy to those of 3. CD₃OD—CDCl₃, 7:1, Assignments are interchangeable. Signals overlap the solvent peak.

Further support for A-B units' identity was provided by HRFABMS (Table II) and FABMS/MS (Scheme I)

TABLE II-continued

	omparison of	Et 743 and Et 73	
Obse	rved ions		
Et 743	Et 736	Formula	Fragment
204.1027	204.1025	C ₁₂ H ₁₄ NO ₂	h – H

In addition, the tandem FAB mass spectra of the key fragment ion m/z 493 for Et 736 and Et 743 were essentially identical (see FIGS. 1A and 1B), arguing that

these Et's contain the same bis(tetrahydro-isoquinoline) subunit.

Addition of 5 μL (ca. 10 equiv.) of methanol-d₄ to a CDCl₃ solution of Et 736 gave a drastic downfield 5 change in chemical shift for C-21 (δ81 →90) due to the chemical exchange of OH at C-21 by OCD₃, just as in the case of Et 743 (67 82 →92), see, Rinehart et al., J. Org. Chem.. 1990, 55, 4512-4515. Similarly, treatment of Et 736 with methanol at room temperature and evaporation of the solvent gave O-methyl-Et 736 (754) (5, M - H at 767.2761 for C₄₁H₄₃N₄O₉S, Δ0.1 mmu, negative ion HRFABMS), which sh showed a new methoxyl signal (δ53.8, CDCl₃) in its hu 13C NMR spectrum.

Subtraction of the bis(tetrahydroisoguinoline) unit 15 (A-B) from the molecular formula for Et 736 (754) gives the formula C₁₃H₁₂N_{2O2}S for the rest of the molecule (unit C). The ¹³C NMR signals for this subunit include one carbonyl carbon and eight aromatic/olefinic carbons, leaving three rings for the structure. The UV 20 spectrum (MeOH) λ_{max} 292 (ξ 11 900), 283 (12 500), 221 (sh 44 800), 207 (11 900) nm along with ¹³C NMR resonances at 8109.5 and 129.3 suggest this structural unit to be a tetrahydro- β-carboline; (see, Shamma et al., Carbon-13 NMR Shift Assignments of Amines and Alka-25 loids: Plenum Press, New York. (1979) and Nakagawa et al., J. Am. Chem. Soc., 111. 2721-2722 (1989) and Rinehart et al., J. Am. Chem. Soc., 23, 3290-3297, (1984)), this was confirmed by COSY spectra showing the aromatic spin system of an ortho-disubstituted ben-30 zene ring with signals from $\delta 7.32$ to 6.91, as well as an aliphatic -CH2-CH2-X spin system (Scheme II). NMR data for unit C closely resemble those of the dihydro-β-carboline debromoeudistomin L. (See, Nakagawa et al., J. Am. Chem. Soc., 111, 2721-2722 35 (1989). The remaining atoms in unit C —a carbonyl, a CH₂ and a sulfur atom - can be assembled as shown in Scheme II to be consistent with the chemical shifts for C-11' and C-12' in Et 743 (see, Table I).

HRMS data on fragmentation ions at m/z 216 and 243, which were seen both in FAB and tandem FAB mass spectra, also supported this assignment (Scheme I). A COLOC spectrum showing a long-range correlation between C-11' and a proton on C-22, along with an IR (CCl₄) absorption at 1753 cm⁻¹, agreed with an 45 ester linkage between C-11'(carbonyl) and C-22.

The molecular formula C₄₀H₄₂N₄O₉S for Et 736 (754) requires 22 degrees of unsaturation, one more than assigned thus far. The additional ring required is consistent with the ¹³C NMR chemical shifts only if it is formed between the sulfur and C-4 of the isoquinoline B ring, as seen in the Et 743 series. Consequently, the structures of Et 736 and 722 were assigned as 3 and 4. These compounds are closely related biogenetically to those of the Et 743 series, except for their tetrahydro-β-55 carboline portion, which presumably comes from tryptamine instead of dopamine, (see, Rinehart et al., J. Org. Chem.. 55, 4512–4512, (1990)). Indeed, the water-soluble portion of the same tunicate extract yielded tryptamine itself, also supporting this biogenetic proposal.

The bioactivities of Et's 722 and 736 appear to be comparable to those of Et's 729 and 743. Et's 722 and 736 inhibit L1210 leukemia cells to the extent of 90% in plate assays at 2.5 and 5.0 ng/mL, respectively (see, FIGS. 2 and 3). More importantly, Et 722 is highly active in vivo, giving T/C 230 (4/6 survivors) at 25 µg/kg day vs. P388 murine leukemia, T/C 200 at 50 µg/kg day vs. B16 melanoma, and T/C 0.27 at 50 µg/kg day vs. Lewis lung carcinoma (see, Table III).

TABLE III

		Activity in vivo of	Ectein	ascidins 729 and 7	22		
	P388	lymphocyclic			Lewi	s lung carcinoma	
Dose		leukemia	В	16 melanoma	_	Mean tumor	
μg/Kg/mi	T/C ^a	Survivors (day)	T/C ^a	Survivors (day)	T/C^b	volume (ram ³)	
Control Et 729	100	0(12)	100	0	1.00	1512	
25.0	130	0(13)	76	0(42)	0.00	2	
12.5 6.25 Et 722	190 N T	2(21) NT	253 197	5/10(42) 0(37)	0.04 0.14	57 216	
50.0	150	1(23)	200 185	0(36) 0(35)	0.27 0.62	412 934	
25.0 12.0	>230 205	4(23) 0(23)	156	0(33)	0.87	1319	

Significant activity: ${}^oT/C \ge 125$; ${}^bT/C \le 40$.

Additional data supporting the activity of Et 722 are shown in the following tables:

The present invention will be further illustrated with reference to the following examples which aid in the

			TUMOR GROV	VTH II	NHIBITION -	Day 14				
	Tumo Gener Tissue Level Site: (MOR	Species: Mouse Strain: 8DF1 Male Source: Charles River Kingston DOB:							
	Mean Body Wt			Median Mean			Mean			
Compound	Dose ug/kg/inJ	Schedule & Route	Change (grams) Day 1-5	N.P. D-14	Tumor Vol. (mm 3)	T/C	Tumor Vol. (mm 3)	ST DEV	T/C	т
ET 722	50.00 25.00 12.50	QD1-9, 1P	-0.5 1.0 1.9	0 0 0	395 908 1204	0.29 0.66 0.88	412* 934 1319	157 158 509	0.27 0.82 0.87	14,862 7,588 1,420

N.P. = # of Non-palpable Tumors on Day 14 *Significant Activity: T/C < 0.40 and p = <0.01 By t Test

		ANTI-TUM		rim Results: Day 23 7 VS. P388 LYMPH	OCYTIC L	.EUKEM	IA_		
	Tissue Level	r: 816 ration: 77801 r: SOLID TU: r: 1:10 BREI richton, Sc.	MOR	Species: Mouse Strain: 8DF1 Male Source: Charles River K DOB:			gston		
Compound	Dose ug/kg/inJ	Schedule & Route	Body Wt. Change (gm) Day 5	Day of Death	Mean Survival Time	% T/C	Median Survival Time	% T/C	Alive Day 23
ET 722	50.00 25.00 12.50	QD1-9, 1P	-0.3 0.2 0.4	10 13 15 15 21 13 21 14 19 20 21 22 22	14.8 19.7	145* 199*	15.0 >23 20.5	150 >230 205	1 4 0

N.P. = # of Non-palpable Tumors on Day 14 *Significant Activity: T/C < 0.40 and p = 0.01 By t Test

			ANTI-TUM	Interim Results: Day 42 OR ACTIVITY VS. 816 MELA	NOMA				
		Tumor: 816 Generation: Tissue: BREI Level: 1:10: 0.5 cc		Species: Mot Strain: 8DF1 Sex: Male Source: Chai		Lingston			
Compound	Dose ug/kg/inJ	Schedule & Route	Change (gm) Day 5	Day of Death	Survival Time	% T/C	Survival Time	% T/C	Day 24
ET 722	50.00 25.00 12 50	QD1-9,1P	-1.1 0.9 0.8	32 32 33 34 34 34 34 35 35 36 28 30 30 30 31 32 34 34 35 35 17 18 23 23 26 27 27 29 30 30	33.9 31.9 25.0	185* 183* 144*	34.0 31.5 28.5	200 185 155	0 0 0

816 (0.5 ml, 1:10 brie) implanted ip into male BDF1 mice on day 0, compounds dissolved or suspended in sterile 0.9% NaCl solution (plus minimal amounts of ethanol and Tween-80 as needed) and administered ip days 1-9 in a volume of 0.5 ml/mouse. Mice were weighed days 1 and 5 and deaths were recorded daily.

*Significant activity: T/C> = 125%

It seems especially promising that some in vivo selectivity is demonstrated by the ecteinascidins; Et 722 is 65 not to be construed as limitations thereof. All percentmore active than Et 729 vs. P388 (T/C 190 at 12.5 μ /kg day for 729) but less active against B16 (T/C 253 for 729).

understanding of the present invention, but which are ages reported herein, unless otherwise specified, are percent by weight. All temperatures are expressed in degrees Celsius.

GENERAL

IR spectra were recorded on an IBM IR/32 FTIR spectrophotometer. Optical rotations were measured with a DIP 370 digital polarimeter with a sodium lamp 5 (589 nm) and 5 cm (1 mL) cell. Melting points were measured with a melting point apparatus and were not corrected. NMR spectra were obtained with QE 300 and GN 500 spectrometers. High- and low-resolution FAB mass spectra and FABMS/MS data were mea- 10 sured on a 70-SE-4F spectrometer. Gravity columns were prepared with silica gel (70-230 mesh) or RP C-18 silica gel (Martex 20-40 μ or Fuji-Division 100 -200 μ). An Ito multi-layer coil separator-extractor was used for CCC, (Y. Ito, CRC Crit. Rev. Anal. Chem., 17, 15 65-143, (1986).

EXAMPLE 1

Collection and Extraction. — A sample (19 kg), collected in the Florida Keys in Aug., 1989, and immedi- 20 ately frozen on site, was stored at -20° C. until use. The defrosted sample was squeezed gently by hand, and the solid material was soaked in 2-propanol (4 L \times 3). The alcoholic extract was separated by decantation 25 from the solid and concentrated to an aqueous emulsion, which was then extracted with $CH_2Cl_2(0.5 L \times 8)$. The CH₂Cl₂ extract was concentrated to a crude oil (20.2 g).

EXAMPLE 2

Separation and Purification. — All separations were monitored by bioassays against L1210 murine leukemia cells and Micrococcus luteus. The crude extract was partitioned between the lower and upper layers of the solvent system heptane - CH₂Cl₂ -CH₃CN (50:15:35). The lower layer was concentrated to an oil (5.76 g), which was partitioned again between the upper and lower layers of the solvent system EtOAc-heptane-MeOH-water (7:4:4:3). The lower layer, showing strong activity, yielded a solid (800 mg), which was 40 rial of E. turbinata, and having the structural formula: partitioned again between the upper and lower layers of the solvent system EtOAc-heptane-MeOH-water (7:4:4:3). The lower layer, showing strong activity, yielded a solid (800 mg), which was then chromatographed to give four fractions over an RP silica gel 45 gravity column with MeOH-aqueous NaCl (0.4 M) (7:1).

The first and most active fraction (333 mg) was separated by CCC into ten fractions with EtOAc-benzene-MeOH-cyclohexane-water (3:4:4:4:3) by using the 50 upper layer as a mobile phase. Fraction 7, containing Et 736 as the major component, was separated by silica gel (treated with NH₃) column chromatography with CHCl₃-MeOH (12:1). The first fraction (30.4 mg) was purified by C-18 HPLC with CH₃CN-MeOH-aqueous 55 NaCl (0.25 M) (5:7:3) to give colorless needles (from CH₃CN-H₂O) of 3 (25 mg, $1.3 \times 10^{-4}\%$): m.p. 140-150° C. dec., IR (CCl₄) 3530, 3480 (NH, OH) 2934, 1768 (C=O), 1753 (C=O), 1196, 1153, 1089 cm⁻¹; IR (film) 3350, 3200 (NH, OH), 2928, 1753 (C=O), 1440, 60 1250, 1200, 1088 cm $^{-1}$; [α], see above; NMR, see Table I; HRFABMS, see Table II.

Fraction 9 of the CCC separation was chromatographed on a silica gel (NH3 treated) column with CHCl₃-MeOH (8:1). The first fraction of this chro- 65 matogram was purified by the HPLC system described above to give light-brown solid 4 (4 mg, $2.1 \times 10^{-5}\%$); m.p. 160-164° C., IR (film) 3291 (NH, OH), 2930, 1753

(C=), 1440, 1238, 1200, 1086 cm⁻¹; $[\alpha]$, see above; NMR, see Table I.

Fraction 10 (200 mg), most polar of the CCC separation, was further separated into five fractions by CCC with CHCL₃—MeOH—H₂O (4:4:3), using the lower phase as the mobile phase. Fraction 5 (51 5 mg) of this CCC run was separated on a silica gel (50 g) column with CHCl₃—MeOH—H₂O (30:20:4) into 11 fractions. Of these, fraction 7 gave crystalline tryptamine hydrochloride (7 mg); m.p. 230° C. dec. (lit, 248° C., see, Merck Index, 1989 1540); TLC behavior and spectral data identical with those of an authentic sample (Aldrich).

The present invention has been described in detail, including the preferred embodiments thereof. However, it will be appreciated that those skilled in the art, upon consideration of the present disclosure, may make modifications and/or improvements on this invention and still be within the scope and spirit of this invention 35 as set forth in the following claims.

What is claimed is:

1. Ecteinascidin 736, essentially free of cellular mate-

wherein X = OH and $R = CH_3$.

2. Ecteinascidin 722, essentially free of celluar material of E. turbinata, and having the structural formula:

3. O-Methyl-ecteinascidin 736, essentially free of cellular material of *E. turbinata*, and having the structural

wherein
$$X = OH$$
 and $R = H$.

wherein
$$X = OCH_3$$
 and $R = CH_3$.