

$$\textcircled{5} \quad Q = 4e^{7t} \rightarrow Q = ab^t \\ = 4(1096.6)^t$$

$$\textcircled{9} \quad Q = 12(0.9)^t \rightarrow Q = ae^{kt} \\ = 12(e^k)^t$$

$$e^k = 0.9 \rightarrow k = \ln 0.9 \approx -1.05$$

$$Q = 12e^{-1.05t}$$

$$Q = 230(1.182)^t$$

$$a = 230 \\ r = 18.2\%$$

$$e^k = 1.182 \\ k = \ln 1.182$$

$$\textcircled{21} \quad P = P_0 e^{0.2t}$$

$$2P_0 = P_0 e^{0.2t}$$

$$2 = e^{0.2t}$$

$$\ln 2 = 0.2t$$

$$3.74 \approx \frac{\ln 2}{0.2} = t$$

30 11,000 to 13,000 in 3 years.

$$f(t) = P_0(1+r)^t$$

$$f(t) = 11,000(1+r)^t$$

$$13,000 = 11,000(1+r)^3$$

$$\frac{13}{11} = (1+r)^3$$

$$\sqrt[3]{\frac{13}{11}} = 1+r$$

$$\sqrt[3]{\frac{13}{11}} - 1 = r$$

$$r \approx 0.05726$$

$$5.726\%$$

$$f(t) = 11,000(1.05726)^t = 11,000 e^{0.05726t}$$

$$e^k = 1.05726$$

$$k = \ln 1.05726$$

$$k = 0.0557$$

Decay

Geiger Counter.

Tritium 5.471% per year decay  
 $t$

$$A(t) = A_0 (1 - 0.05471)$$

$$A(t) = A_0 e^{-0.05626t}$$

$$\frac{1}{2} A_0 = A_0 e^{-0.05626t}$$

$$\frac{1}{2} = e^{-0.05626t}$$

$$\ln \frac{1}{2} = -0.05626t$$

$$12.32 \text{ year.} = t$$

Newton's Law of Cooling

$$T(t) - T_A = a e^{kt}$$

$k$  is constant that  
depends on substance!

Let  $t=0$

$$T_0 - T_A = a(1)$$

$$T(t) - T_A = (T_0 - T_A) e^{kt}$$

