



Using Lexical Analysis Software to Understand Student Knowledge Transfer between Chemistry and Biology

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Chemistry of Biology

- Evaluate students' understanding of basic chemistry that may be related to conceptual problems students have in cellular and molecular biology
- Introductory Cell and Molecules Biology
- Focus on two topics: Free energy and acid/base chemistry



Research Questions

- Do students use chemistry concepts in response to biological chemistry problems?
- Can lexical analysis extract and categorize chemistry terms from student responses?
- Are these categories meaningful?

Strong and weak acids: constructed response



- Students were asked to:
 - Give an explanation of a strong acid.
 - Give an explanation of a weak acid.
- 382 student responses collected from Fall 2008

Lexical analysis



Strong acid explanation

Categories Statistics

All Records (382)

- Uncategorized (0)
- ionization (262)
- solution (242)
- hydrogen (69)
- donate (35)
 - donate (16)
 - give up (5)
 - release proton (5)
 - donor (3)
 - give off (2)
 - willingness to donate proton (2)
 - give away (2)
 - give (1)
 - donates (0)
- ph (29)
- ions (21)
- concentration (17)
- hydronium (15)

Unused Extractions All Extractions

Extract Term

Extraction needed

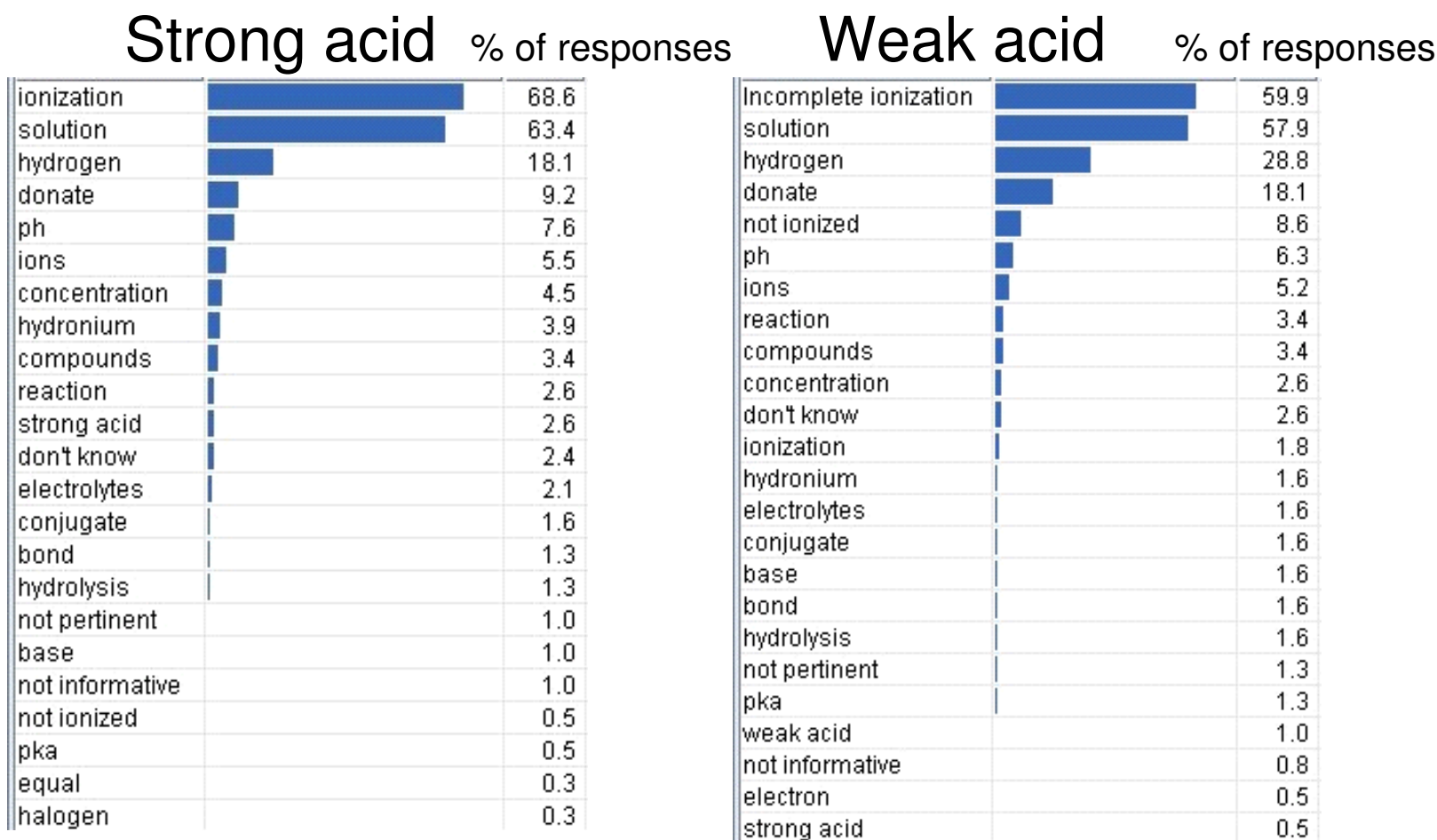
- solute (1)
- solution (40)
- solution to form (1)
- strong (23)
- strong acid (129)

	Id	Response	Categories
			compounds
13	1184	An acid that when dissolved in water will have a high concentration of Hydrogen ions	concentration hydrogen solution
14	1148	Has a large concentration of H+ ions when dissolved into water.	concentration hydrogen solution
15	1079	when put in water increases the concentration of H+ ions in solution	concentration hydrogen solution
16	1075	Increases the concentration of H+ ions in a solution and completely ionizes	concentration ionization hydrogen
17	1372	Has a low level of hydrogen concentration: pH is closer to 1.	concentration ph
18	1324	concentrations of ions are = to concentration of strong acid in solution	concentration solution
19	1306	Produces a large number of hydrogen ions in a solution	concentration solution
20	1129	Has a conjugate base that is weak.	conjugate
21	1201	Has a weak conjugate base.	conjugate
22	1188	An acid with a high pH level that has a weak conjugate base.	conjugate ph
23	1166	Something that can easily give away a H+ ion and that has a low pH which means that there will be a weak conjugate base.	conjugate ph donate hydrogen

Strong and weak acids: lexical analysis



- Each explanation was analyzed separately
- Lexical analysis resulted in a total of 27 categories

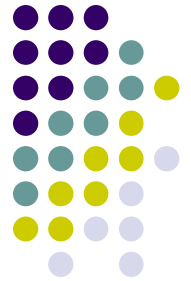


Strong and weak acids: expert scoring



- Two experts rated 150 responses using 4-bin rubric
 - Level 1: Correct explanations both strong and weak acids
 - Level 2: Mostly correct explanations with minor errors
 - Level 3: Mostly incorrect explanations OR one explanation completely correct; one explanation completely incorrect
 - Level 4: Totally incorrect explanations / irrelevant information

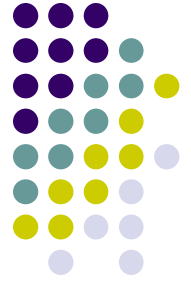
Strong and weak acids: expert rating



- Inter-rater reliability of two expert raters
 - Cronbach alpha > .97

LEVEL				TOTAL
1	2	3	4	
91	9	18	18	136

Predicting expert score: Discriminant analysis



- Step-wise discriminant analysis
- Dependent Variable: Expert rating
- Independent Variables: Lexical categories

Strong and weak acids: discriminant analysis



Categories		Function		
		1	2	3
Strong acid explanation:	hydrolysis	-.111	.288	-.248
	donate	-.071	.143	.257
	solution	.323	.237	-.178
	hydrogen	-.062	.101	.187
	not pertinent	-.086	-.132	-.016
	conjugate	-.060	.156	-.134
	ionization	.446	.056	-.417
Weak acid explanation:	incomplete ionized	.624	-.094	.222
	not ionized	-.157	.354	.279
Function 1 accounts for 69.9% of variance Function 2: 25.2% Function 3: 5%	solution	.294	.056	-.269
	not pertinent	-.122	.106	-.168
	electrolytes	.011	.031	.424

Strong and weak acids: cross-validation



- Cross-validation classification functions results in 83.8% of responses being correctly scored

	Computer Predicted Rating			
Expert Rating	1	2	3	4
1	93.4	3.3	3.3	0.0
2	33.3	33.3	22.2	11.1
3	5.6	5.6	66.7	33.3
4	11.1	0.0	11.1	77.8

Functional groups: multiple choice



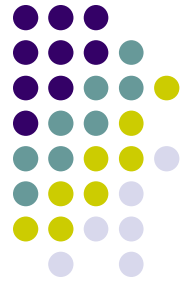
- Consider two small organic molecules in the cytoplasm of a cell, one with a hydroxyl group (-OH) and the other with an amino group (-NH₂). Which of these small molecules (either or both) is most likely to have an impact on the cytoplasmic pH?
- **A. Compound with amino group (35%)**
- B. Compound with hydroxyl group (45%)
- C. Both (7%)
- D. Neither (13%)

Functional groups: constructed response



- “Explain your answer.”
- 2 Experts rated 131 correct answers using 3-bin rubric
 - Level 1: Correct explanations of functional group chemistry (may include correct supporting reasoning)
 - Level 2: Partly correct explanations with errors in facts or reasoning
 - Level 3: Totally incorrect/irrelevant response

Expert rating: Functional group open response

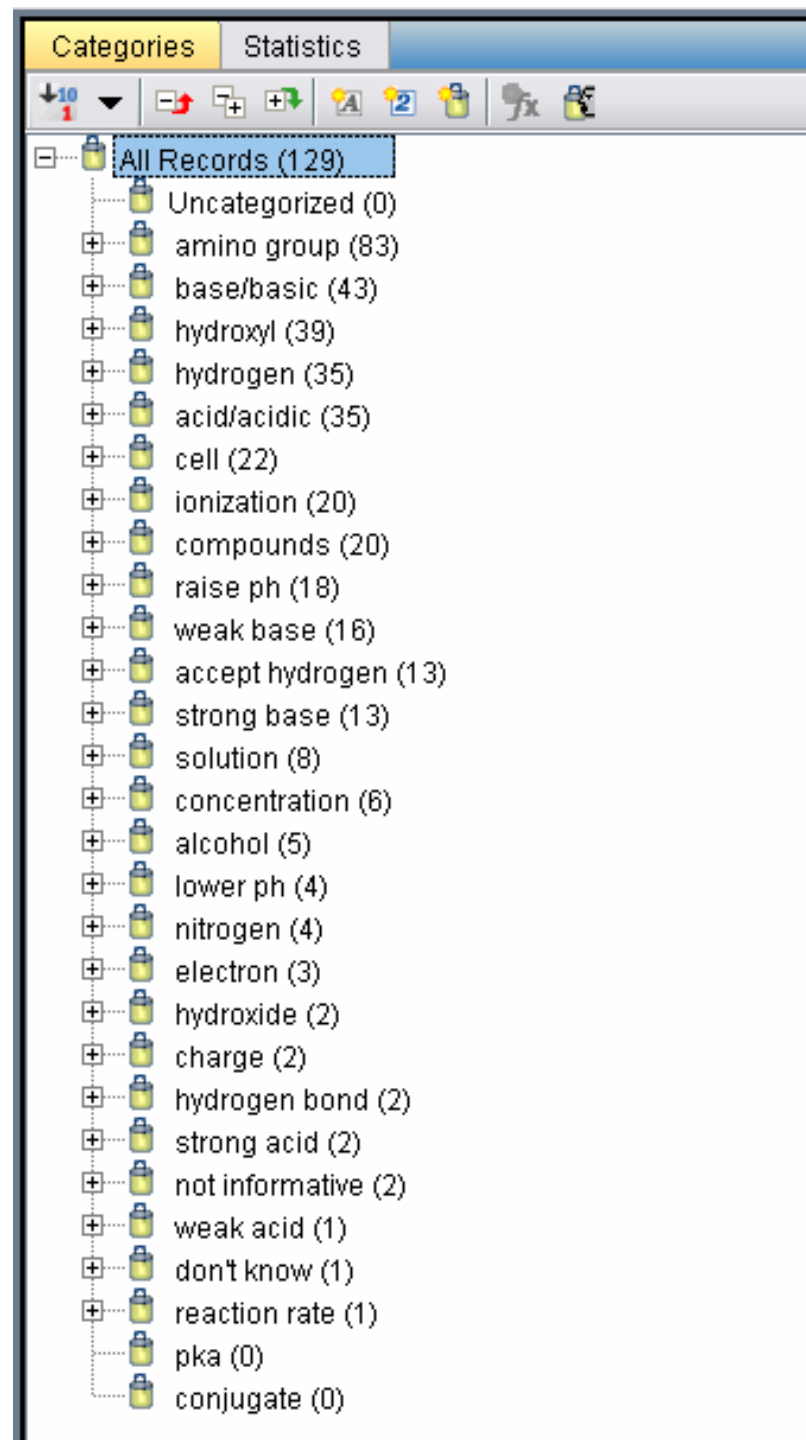


- Cronbach Alpha > .92

LEVEL			TOTAL
1	2	3	
41	14	58	113
36%	12%	51%	

Lexical analysis: Functional groups

- Each foil analyzed separately
- For correct response: 28 categories
 - Ionization – 20
 - Accept hydrogen - 13



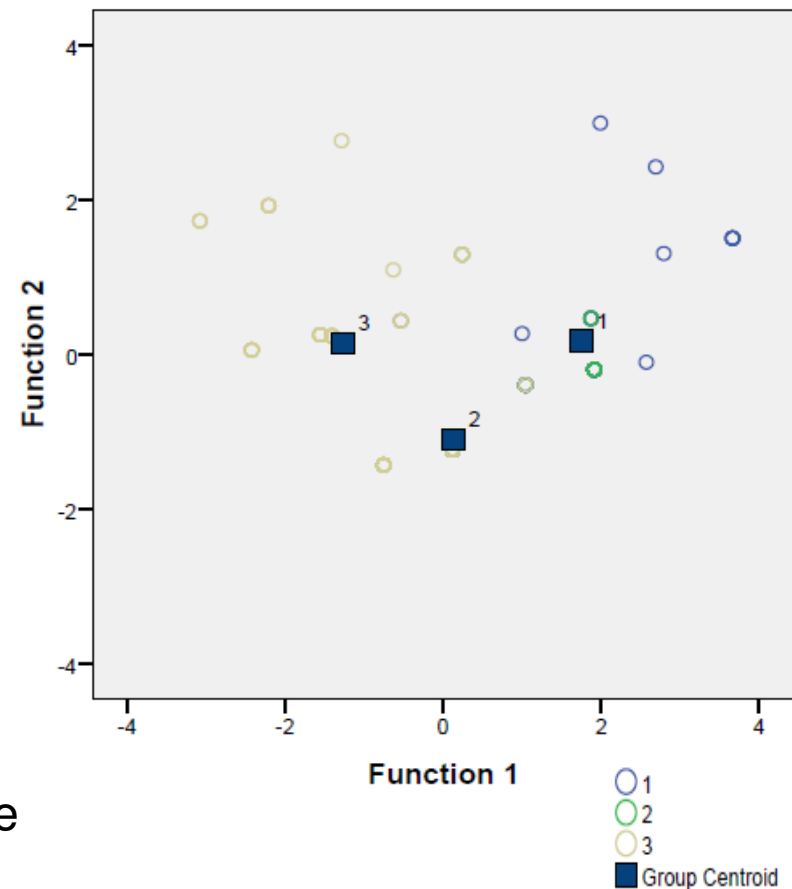
Functional group: Discriminant analysis



Category	Function	
	1	2
base/basic	.517	.349
acid/acidic	-.378	.629
amino group	.280	.128
hydrogen	-.039	.605
charge	.121	.137
accept hydrogen	.345	.388

- Function 1 accounts for 91% of variance

Canonical Discriminant Functions



Functional Groups: Computer predicted



- Cross-validation results in 77% of the cases scored correctly

	Computer Predicted Rating		
Expert Rating	1	2	3
1	82.9	12.2	4.9
2	21.4	42.9	35.7
3	6.9	12.1	81.0

- Expert/computer inter-rater reliability
 - Intraclass correlation = 0.835

Conclusions:

Acid/base chemistry



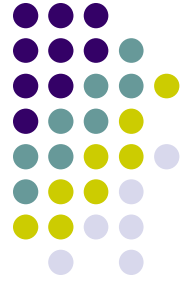
- Students have difficulty describing activity of acid/base behavior of functional groups, but not in a general format.
- Important terms in student responses change between explanations.
- Inappropriate concept application revealed by lexical analysis.

Conclusions:

Lexical and discriminant analyses

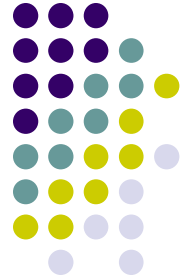


- Lexical analysis provides a whole-class picture of term / concept usage.
- Discriminant analysis can help identify categories of importance.
- Classification functions can be developed to accurately predict human scoring.
- Analysis can be done per rubric.



Future Directions

- Refinement of current analysis
 - Collection of more student responses
- Sequences and related constructed response items
- Expansion into other disciplines
 - Evolution and natural selection, geology, genetics
- Just in Time Teaching



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