# Predefined internal function variables

Defintion	Description	Defintion	Description	Defintion	Description
$n_1$	number of carbon atoms in $FA_1$	$P = \min(n_1, n_3)$	shortest outer chain length	$n_{le}$	number of linolenic chains
$n_2$	number of carbon atoms in $FA_2$	$Q=n_2$	middle chain length	$n_{OO}$	number of oleic-oleic pairs
$n_3$	number of carbon atoms in FA <sub>3</sub>	$R = \max(n_1, n_3)$	longest outer chain length	$n_{\text{EE}}$	number of elaidic-elaidic pairs
$\boldsymbol{n} = \textstyle\sum n_i$	total carbon number	x = Q - P	chain length difference	$n_{ll}$	number of linoleic-linoleic pairs
$\mathbf{u}_1$	number of double bonds in $FA_1$	y = R - P	chain length difference	$n_{lele}$	number of linolenic-linolenic pairs
$\mathbf{u}_2$	number of double bonds in $FA_2$	$n_{O}$	number of oleic chains	$n_{Ol}$	number of oleic-linoleic pairs
$\mathbf{u}_3$	number of double bonds in $FA_3$	$n_{\rm E}$	number of elaidic chains	n <sub>Ole</sub>	number of oleic-linolenic pairs
$\boldsymbol{u} = \textstyle\sum \! u_i$	total number of double bonds	$n_l$	number of linoleic chains	$n_{lle}$	number of linoleic-linolenic pairs

# Saturated triglycerides

Predicting the enthalpy and entropy of fusion

$$\Delta H^{\text{sat}} = h \, \text{n} + h_0 + h_{xy} \, f_{xy} + h_{odd} \, f_{\text{odd}} \, f_{\beta}$$
 [1]

$$\Delta S^{\text{sat}} = s \, \text{n} + s_0 + s_{xy} \, f_{xy} + s_{odd} \, f_{odd} \, f_{\beta} + R \ln 2 \, f_{\text{asym}} \, f_{\beta}$$
 [2]

$$f_{xy} = 2 - \exp\left(-\left(\frac{x - x_0}{k}\right)^2\right) - \exp\left(-\left(\frac{y}{k}\right)^2\right)$$
 [3]

$$f_{odd} = \begin{cases} 1, \text{if } n_1 \text{ or } n_2 \text{ or } n_3 \text{ is odd numbered} \\ 0, \text{if } n_1 \text{ or } n_2 \text{ or } n_3 \text{ is even numbered} \end{cases}$$
 [4]

$$f_{asym} = \begin{cases} 1, & \text{if } y \neq 0 \\ 0 & \text{if } y = 0 \end{cases}$$
 [5]

$$f_{\beta} = \begin{cases} 1, & \text{if TAG is in } \beta \text{ polymorph} \\ 0, & \text{if TAG is not in } \beta \text{ polymorph} \end{cases}$$
 [6]

$$T_{\rm m}^{\rm sat} = \frac{\Delta H^{\rm sat}}{\Delta S^{\rm sat}} = \frac{h \, n + h_0'}{s \, n + s_0'} \tag{7}$$

$$h'_0 = h_0 + h_{xy} f_{xy} + h_{odd} f_{odd} f_{\beta}$$
 [8]

$$s'_0 = s_0 + s_{xy} f_{xy} + s_{odd} f_{odd} f_{\beta} + R \ln 2 f_{asym} f_{\beta}$$
 [9]

Re-writing eq. [7] as Taylor series expansion around  $\left(\frac{1}{n}\right)$ :

$$T_{\rm m}^{\rm sat} = \frac{h}{s} \left( 1 + \left( \frac{h_0'}{h} - \frac{s_0'}{s} \right) \frac{1}{n} - \frac{s_0'}{s} \left( \frac{h_0'}{h} - \frac{s_0'}{s} \right) \frac{1}{n^2} + \dots \right)$$
[10]

Truncating the series and substituting for convenience results in eq. [11]

### Parameter Description

h, s hydrocarbon chain contribution

 $h_0$ ,  $s_0$  end group contribution

h<sub>xy</sub>, s<sub>xy</sub> contribution of differences in chain length

k, x<sub>0</sub> constant

h<sub>odd</sub>, s<sub>odd</sub> odd chain contribution

A<sub>0</sub>, B<sub>0</sub> end group contribution

A<sub>odd</sub>, B<sub>odd</sub> odd chain contribution

 $A_x$ ,  $A_{x^2}$ ,  $A_{xy}$ ,  $A_y$ ,  $A_{y^2}$ , fitting parameter

 $B_x$ ,  $B_{x^2}$ ,  $B_{xy}$ ,  $B_y$ ,  $B_{y^2}$ 

### Predicting the melting point

$$T_{m}^{sat} = T_{inf} \left( 1 + \frac{A^{sat}}{n} - \frac{A^{sat} B^{sat}}{n^2} \right)$$
 [11]

1st approach

$$A^{sat} = A_0 + A_{odd} f_{odd} + A_x x + A_{x^2} x^2 + A_{xy} xy + A_y y + A_{y^2} y^2$$
 [12]

$$\begin{split} B^{sat} &= B_0 + B_{odd} f_{odd} \\ &+ B_{x} x + B_{x^2} x^2 + B_{xy} xy + B_{y} y + B_{y^2} y^2 \end{split} \tag{13}$$

2<sup>nd</sup> approach

$$A^{sat} = \frac{\widehat{h_0}}{h} - \frac{\widehat{s_0}}{s}$$
 [14] 
$$B^{sat} = \frac{\widehat{s_0}}{s}$$
 [15]

$$\widehat{\mathbf{h}_0} = h_0 + h_{xy} \, \mathbf{f}_{xy} + h_{odd} \, \mathbf{f}_{odd} \, \mathbf{f}_{\beta} \tag{16}$$

$$\hat{s_0} = s_0 + s_{xy} f_{xy} + s_{odd} f_{odd} f_{\beta} + R \ln 2 f_{\beta} f_{asym}$$
 [17]

# Unsaturated triglycerides

Predicting the enthalpy of fusion

$$\Delta H^{\text{unsat}} = H^{\text{sat}} + h_O \, n_O + h_E \, n_E + h_l \, n_l$$
 [18]

Predicting the melting point

$$A^{\text{unsat}} = A^{\text{sat}} + A_O \, n_O + A_E \, n_E + A_l \, n_l + A_{le} \, n_{le} + A_{OO} \, n_{OO} + A_{EE} \, n_{EE} + A_{ll} \, n_{ll} + A_{Ol} \, n_{Ol} + A_{Ole} \, n_{Ole} + A_{lle} \, n_{lle}$$
[19]

$$B^{unsat} = B^{sat} + B_O n_O + B_l n_l + B_{le} n_{le}$$
 [20]

# $\begin{array}{lll} \textbf{Parameter} & \textbf{Description} \\ h_O & \text{oleic chain contribution} \\ h_E & \text{elaidic chain contribution} \\ h_l & \text{linoleic chain contribution} \\ A_O, A_E, A_l, A_{le}, & \text{correction parameter accounting for oleic,} \\ B_O, B_l, B_{le} & \text{elaidic, linoleic and linolenic acid} \\ A_{OO}, A_{EE}, A_{ll}, A_{lele} & \text{interaction of the same unsaturated FA} \\ A_{Ol}, A_{Ole}, A_{lle} & \text{interaction of different unsaturated FA} \\ \end{array}$