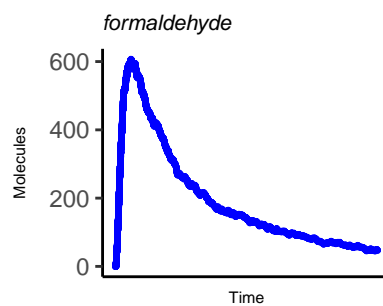
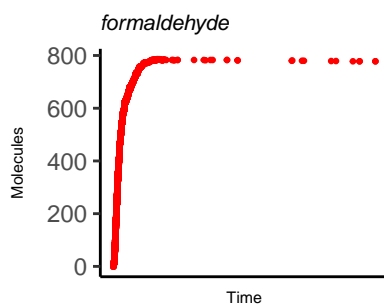
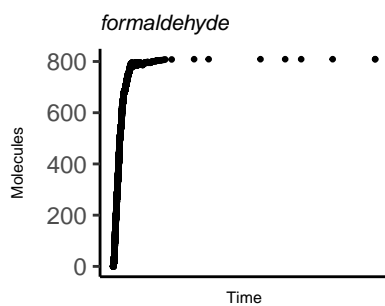
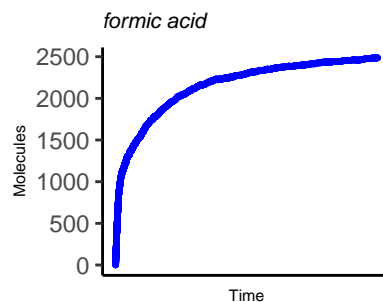
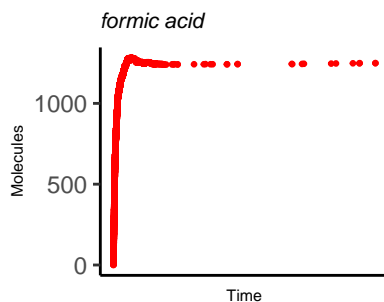
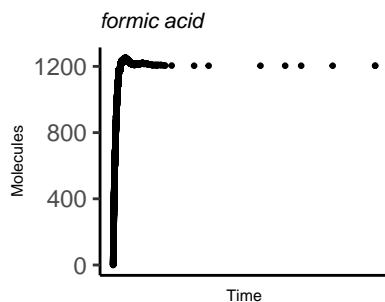
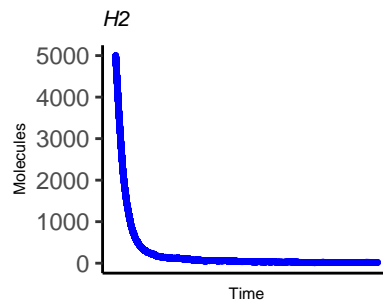
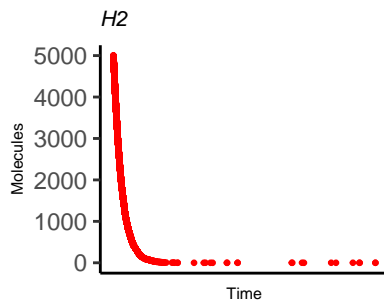
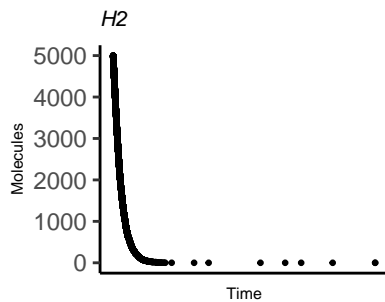
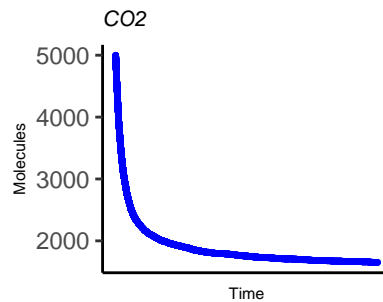
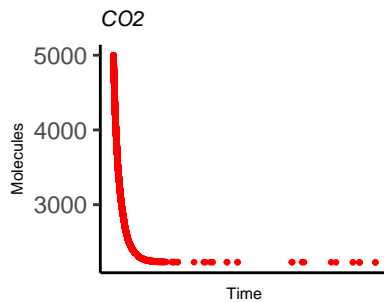
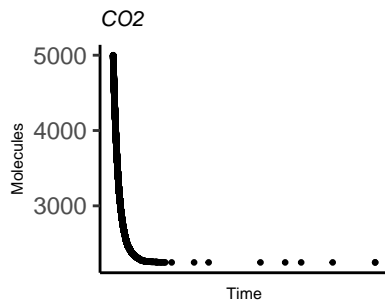
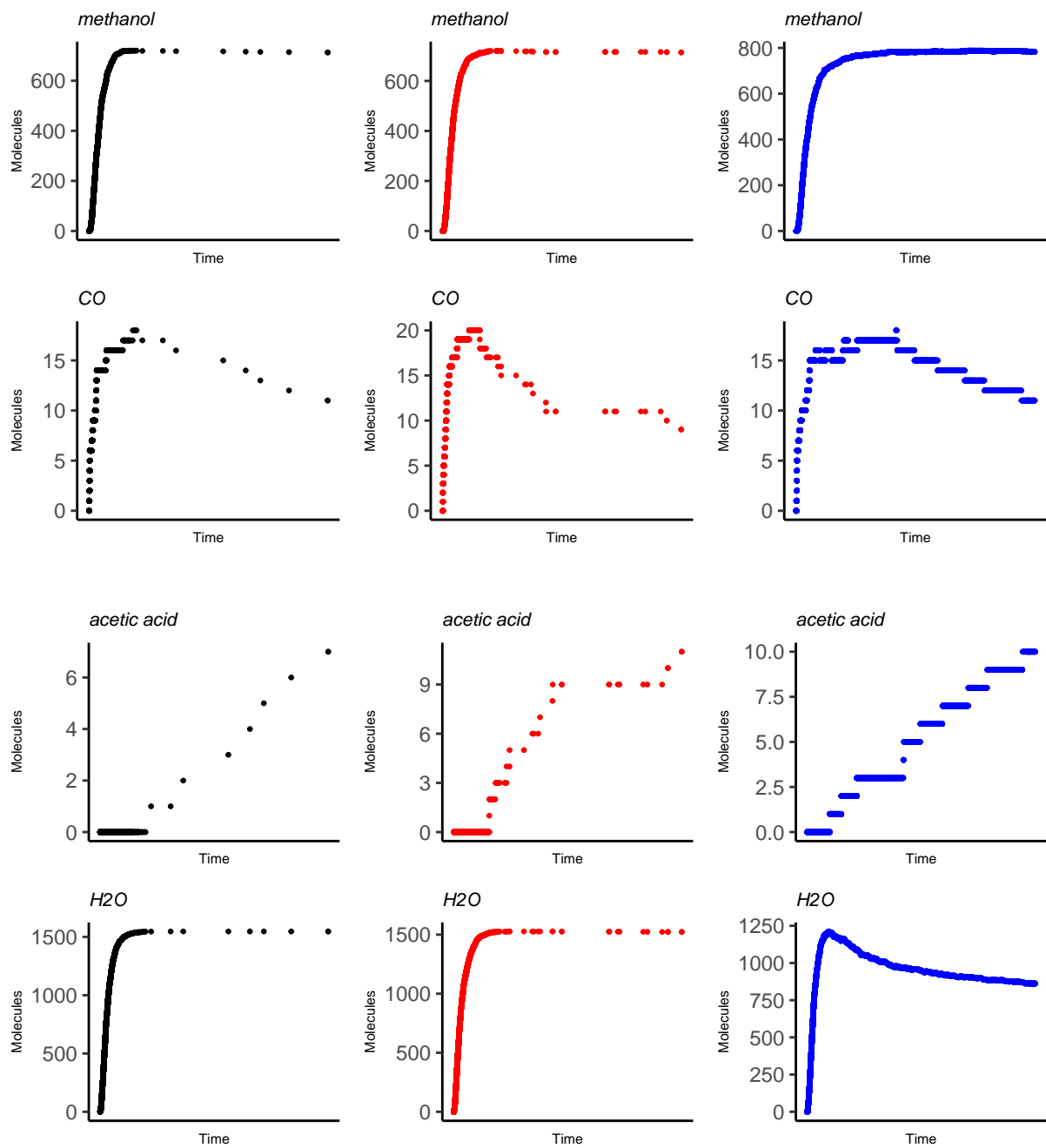


Comparing simulations of irreversible and reversible CO_2 fixation reactions (simplified CO_2 fixation)

The output on the next page is generated after manually tweaking Lia's CO_2 fixation simulation from the summer. The original simulation is the irreversible CO_2 fixation network (output in **black**). In **red**, reversible reactions with negligible backward rates yield simulations which are very similar to the irreversible reaction simulations. In **blue**, the reversible reactions with equal forward and backward reaction rates result in a distinct trajectory.

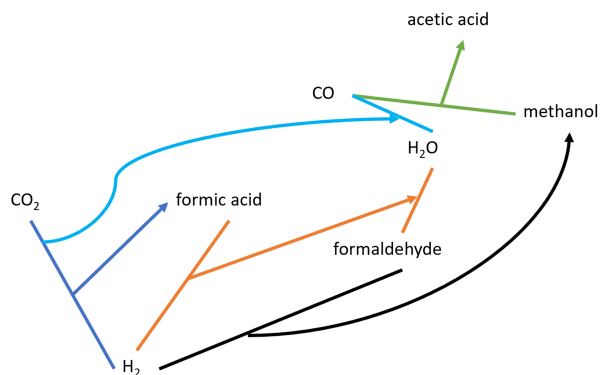
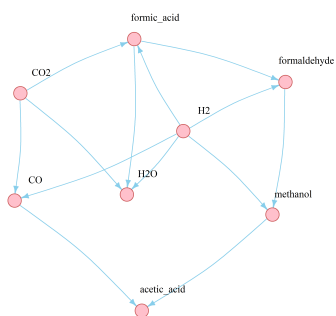
| Forward reactants | Forward products | Propensity function | Parameters |
|-----------------------------|------------------------------|-------------------------------------|---------------|
| CO_2, H_2 | <i>formic acid</i> | $k_1 * CO_2 * H_2$ | $k_1 = 10$ |
| <i>formic acid</i> , H_2 | <i>formaldehyde</i> , H_2O | $k_2 * \textit{formic acid} * H_2$ | $k_2 = 19.9$ |
| <i>formaldehyde</i> , H_2 | <i>methanol</i> | $k_3 * \textit{formaldehyde} * H_2$ | $k_3 = 23.5$ |
| <i>methanol</i> , CO | <i>acetic acid</i> | $k_4 * \textit{methanol} * CO$ | $k_4 = 0.869$ |
| CO_2, H_2 | CO, H_2 | $k_5 * CO_2 * H_2$ | $k_5 = 0.061$ |



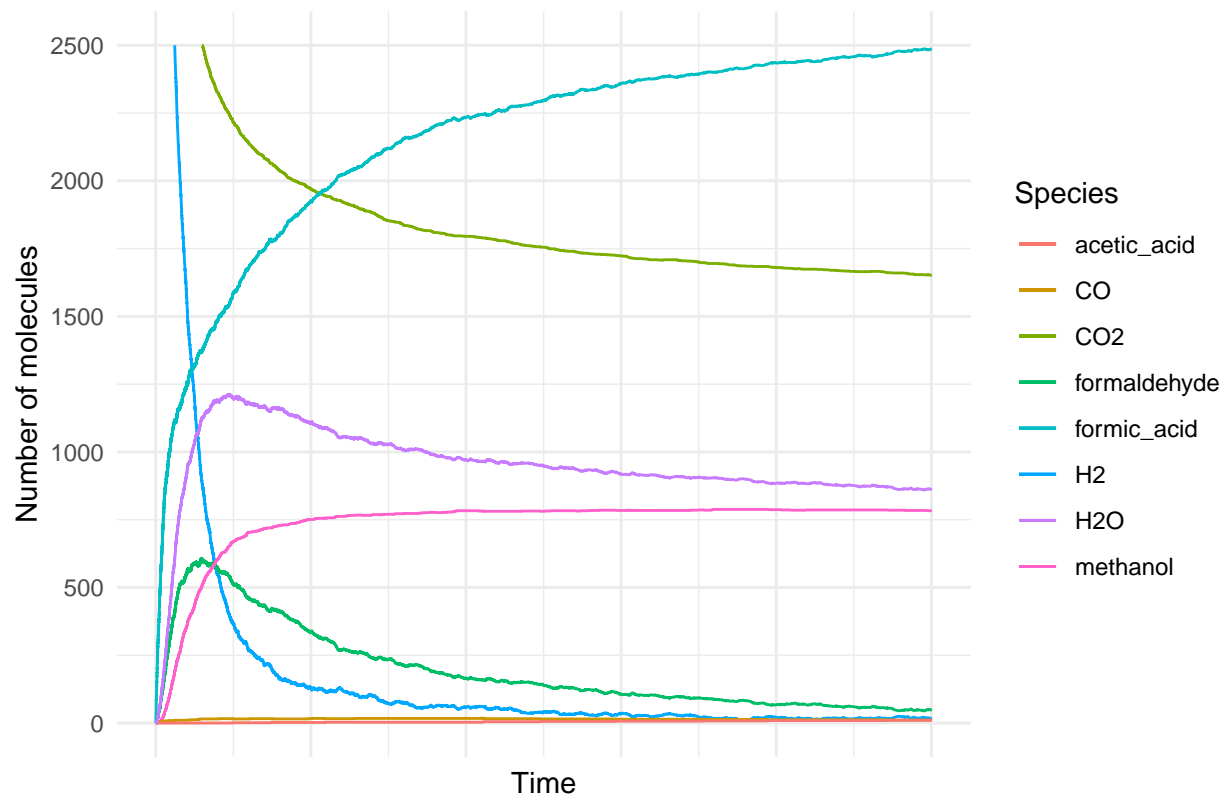


Inspecting the reversible network

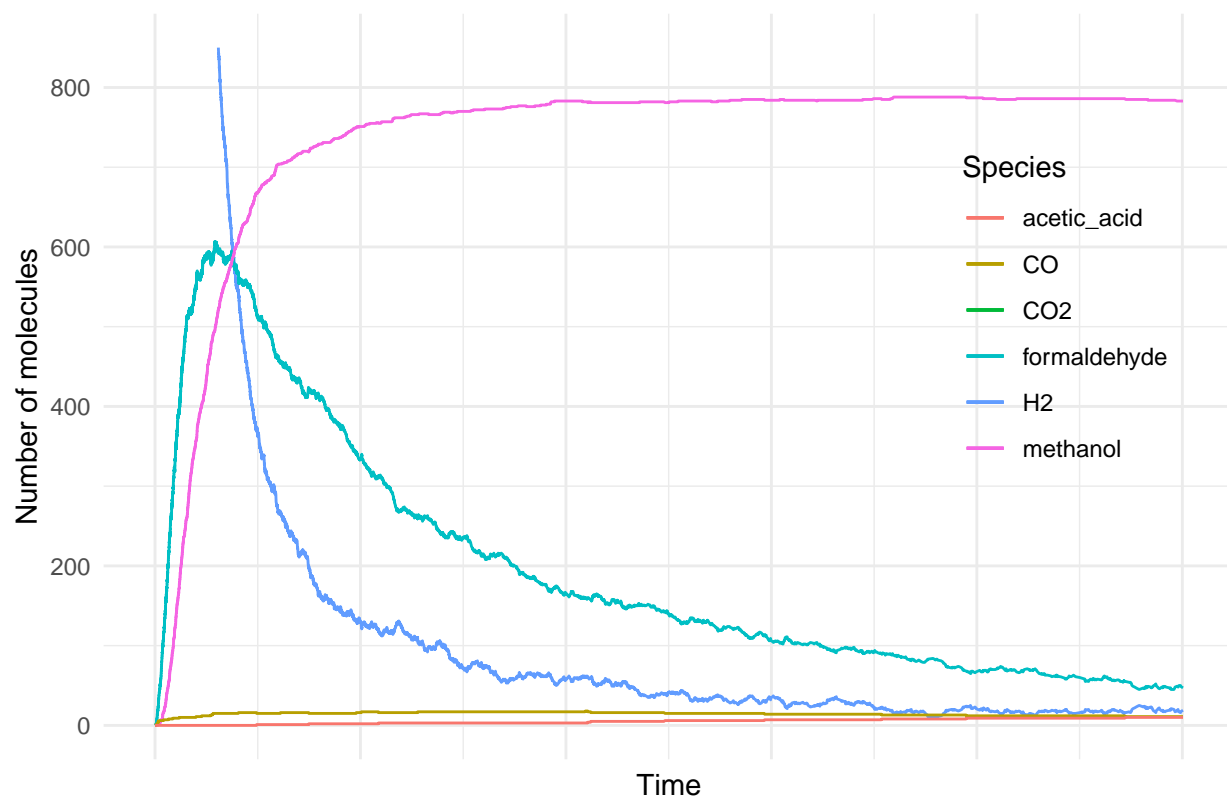
The network diagram below shows the relationship between the different reactants and products.



Reversible C fixation



Reversible C fixation



Compare to Lia's simulations

