

UNIT-II CARBOHYDRATES

(Saccharides / Saccharose)

→ Biomolecules which are made up of carbon, Hydrogen & Oxygen.

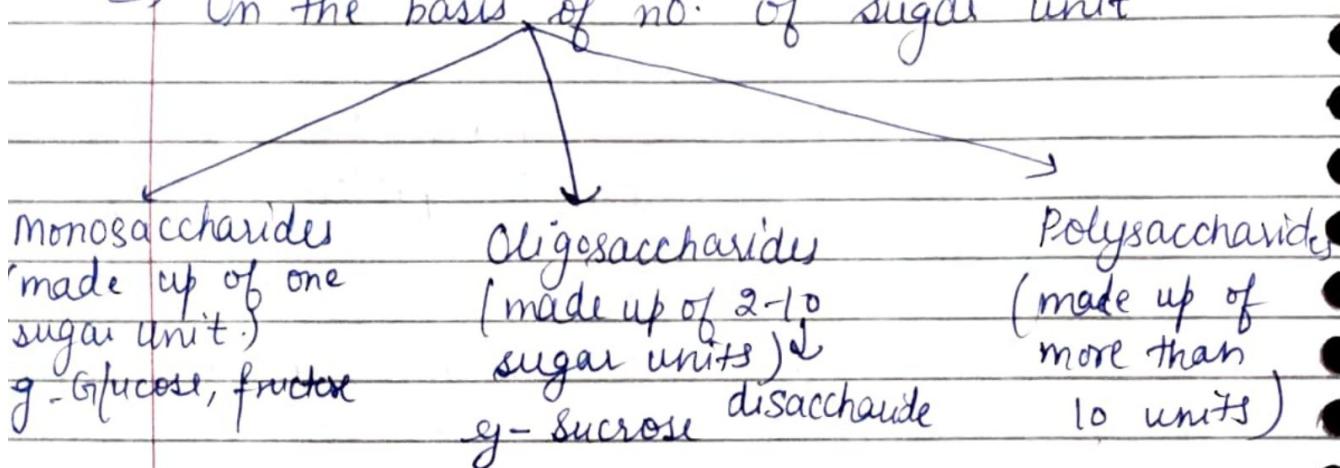
→ most abundant organic molecules in nature.

→ Hydrates of carbon.

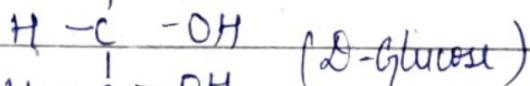
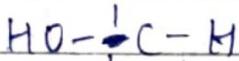
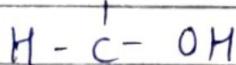
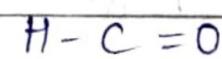
→ Polyhydroxy, aldehyde or ketone compounds

↓ ↓ ↓
multiple -OH group -CHO -C=O

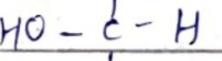
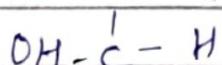
→ On the basis of no. of sugar unit



* [GLUCOSE], [α . & β isomers].



CH_2OH (written right side)



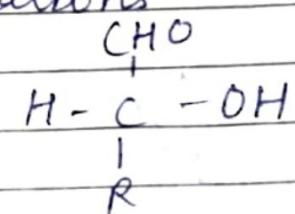
CH_2OH
(β -Glucose)

* Stereoisomerism :-

- Same structural formula but different spatial configuration.
- Due to the presence of asymmetric or chiral carbon.

Example : D & L Glucose.

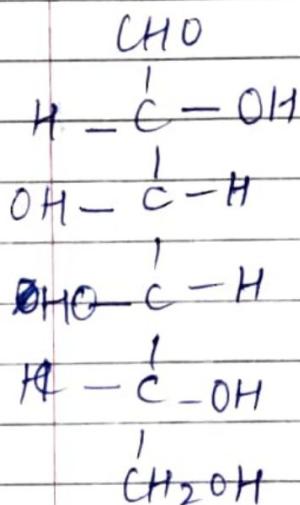
- Chiral carbon → carbon which is attached with 4 carbons



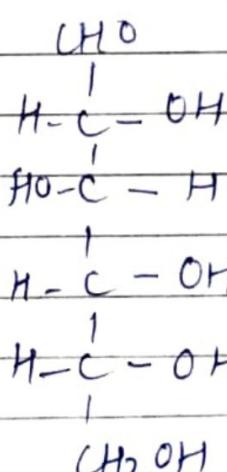
* Epimers :-

- Carbohydrates which are different at a single specific carbon other than C₁ atom.
- e.g. D-Glucose & D-Galactose (C₄ epimers)
- D-Glucose & D-Mannose (C₂ epimers)

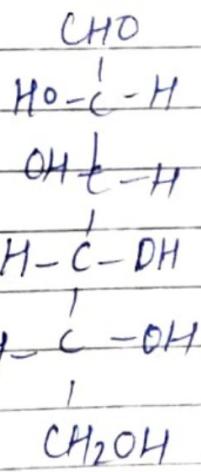
D-Galactose



D-Glucose



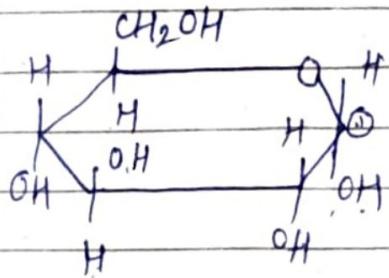
D-Mannose



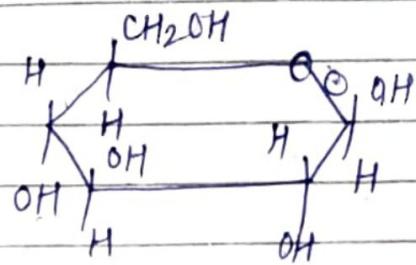
Epimers :-

- If two monosaccharides diff from each other in their configuration around a single specific carbon other than C₁ carbon.
ex glucose & galactose (C₄)
Glucose & Mannose (C₂)

- Anomers :- If two monosaccharides differ from each other around also anomeric carbon. ex. α - β Cyclic forms of D-Glucose.



α -D Glucose



β -D Glucose.

Carbohydrates

Reducing sugar

Non-Reducing sugar

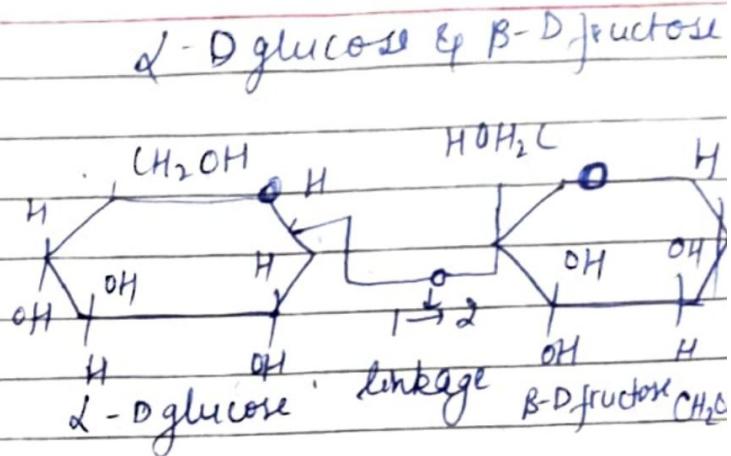
free aldehyde (-CHO)
or ketone gp (-C=O)
is present eg-
lactose, mannose,
maltose.

No free aldehyde
or ketone gp
eg- Sucrose.

→ Oligosaccharides
↓

Dissacharides
↓

Made up of 2
sugar units
↓



two units linked
together by glyco-
sidic bonds
eg. sucrose made up of.

* functions of Carbohydrates :-

- (i) Carbohydrates are the most abundant source of energy.
- (ii) Carbohydrates are the structural component of many organisms. eg cellulose forms the cell wall of the plants.
- (iii) Carbohydrates are precursors for many organisms components. eg. lipids & Amino acids.
- (iv) Carbohydrates participate in the structure of cell Membrane.
- (v) Carbohydrates also serve as the storage form of energy to meet the immediate energy demands of the body.

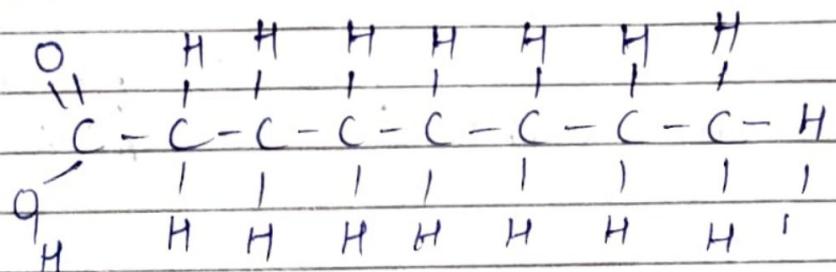
* LIPIDS :-

- organic compounds which contain carbon, hydrogen & oxygen.
- made up of hydrocarbons soluble in organic solvent (alcohol, ether) soluble in water.
- lipids are found in oil, butter, whole milk cheese & fried foods.

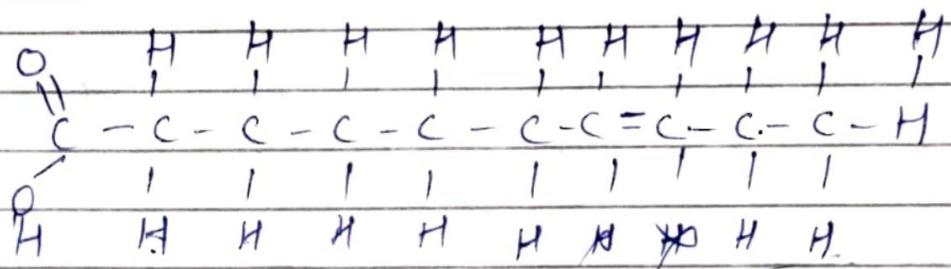
* LIPID STRUCTURE :-

Polymer of fatty acid that contain along non polar hydrocarbon chain with small polar region containing oxygen.

Saturated :-



Unsaturated :-



* Classifications of Lipids :-

(i) Non Saponifiable → cannot break down into smaller molecules through hydrolysis. eg. cholesterol.

(ii) Saponifiable :- break down into ester gbs through hydrolysis. eg. wax.

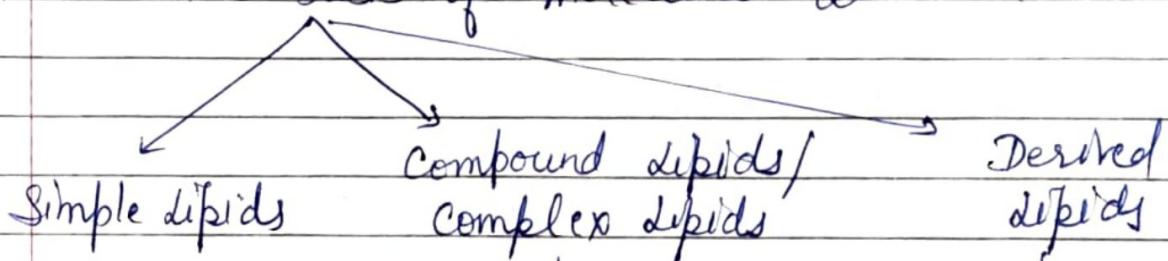
Polar lipids

Non-polar lipids.

→ Polar lipids :- form bimolecules with an external in water environment.

→ Non-polar lipids :- are utilized as fuel and to store energy.

* On the basis of molecules attached



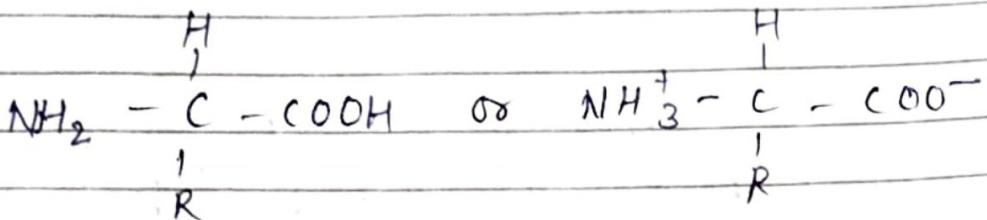
If it is a fatty acid ester of different alcohols & carries no other substance.
eg. phospholipids.

If consist of a fatty acid, an alcohol, & one or more other gbs such as phosphorus & Nitrogen
eg. steroids

These are the hydrolyzed compounds of simple & complex lipids.
eg fatty acids.

* AMINO ACIDS :-

→ organic compounds containing amino (-NH₂) group & carboxyl group (-COOH)



Basic Structure of Amino acids.

* Glycine is the simplest amino acids.

* Classification of Amino Acids:-

On the basis of synthesis by body

Essential

Non-Essential

Essential :- Those amino acids which are not synthesised by body. These amino acids include in our diet. for ex. 9 amino acids are essential amino acids. Lysine, Valine, Threonine, Histidine.

Non-Essential :- Those amino acids are synthesised by body. for ex. glycine, Serine. Total 11 amino acids in Non-essential.

* On the basis of charge :-

net Positive charge (Basic)
e.g. Lysine

Neutral (No net) charge
e.g. Valine, glycine

Negative net charg.
(Acidic)
e.g. glutamic acid

* On the basis of aromaticity :-

Aliphatic (linear chain)
e.g. Valine, glycine

Aromatic (ring like structure)
e.g. Tryptophan

* Functions of Amino Acids :-

- Amino acids are monomers of protein that form building blocks of the body.
- Amino acids acts as fuel in glycolysis & fatty acids synthesis.
- Deficiency of amino acids causes various diseases.

* NUCLEIC ACID *

- Nucleic acids are polymers of Nucleotides.
- two type of Nucleic acid presents
- DNA ('Deoxyribonucleic acid') & RNA (Ribonucleic acid)

(Deoxyribonucleic acid)

* DNA :- Adenine
valine & Cytosine
thymine + sugar
Nucleoside
+ phosphate Nucleotides

(Ribonucleic acid)

RNA :- Guanine, Cytosine
Adenine, Uracil.

* Adenine { Purine
&
Guanine } (double ring)

* Cytosine { Pyrimidine
Thymine } (single ring)
Uracil structure

* PROTEINS :-

→ 6 polypeptides linear chain of aa linked together by peptide bond.

→ complex structure larger than carbohydrates of lipids e.g. Insulin, Collagen

Denaturation :-

→ means breaking of bonds within protein molecule.

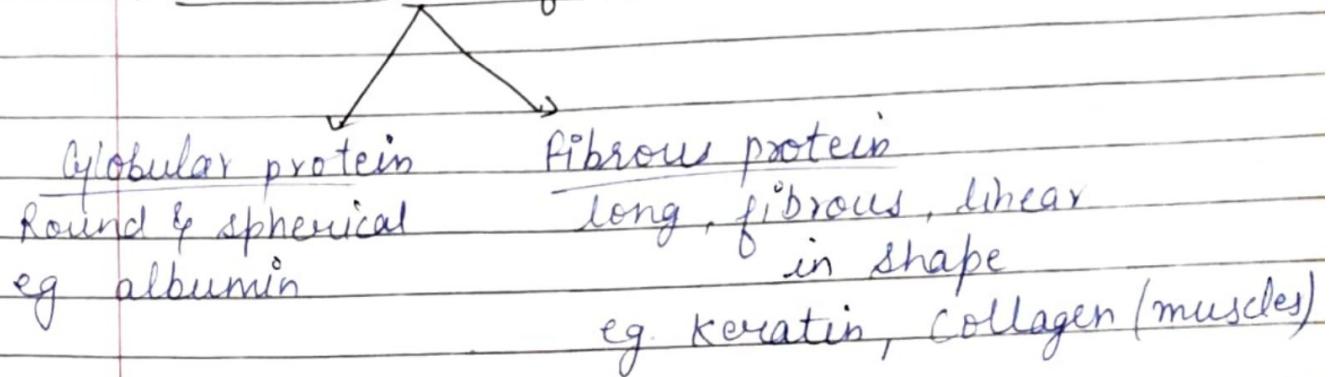
→ property of protein to change its form due to change in temperature or pH in KJ as denaturation.

e.g. Egg → albumin protein → liquid state in initial

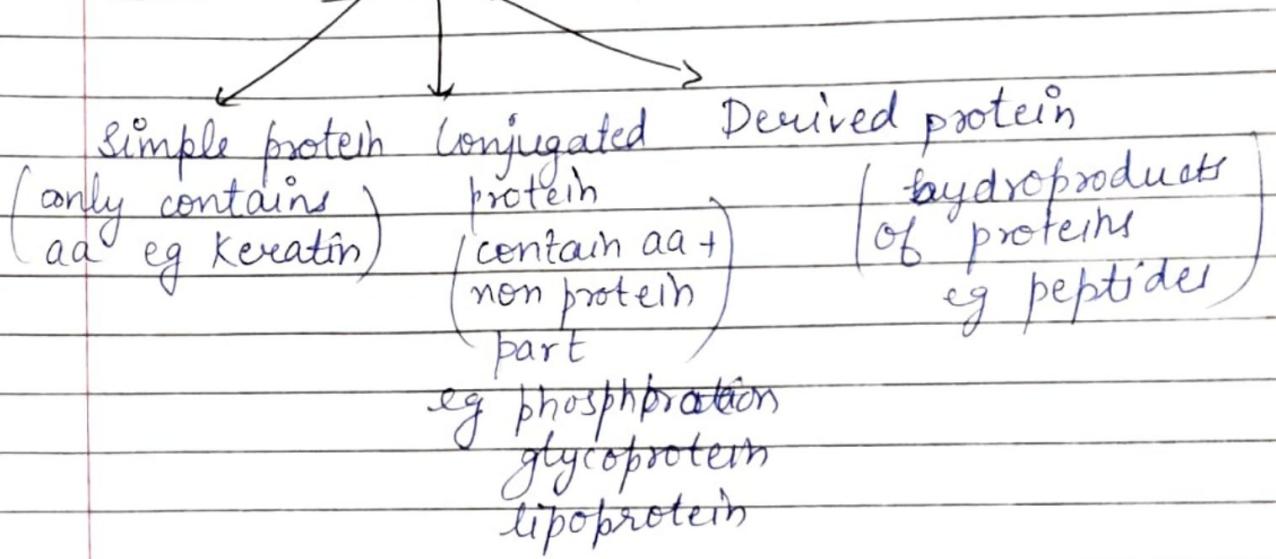
heating; comes in solid state

* Classification of proteins :-

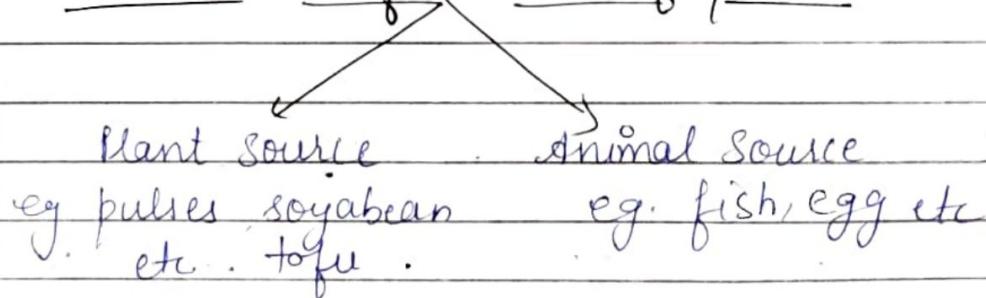
On the basis of structure :-



On the basis of composition :-



On the basis of source of protein :-



On the basis of structural configuration of proteins

Primary structure (1°)

linear chain of α linked together by peptide bond

Secondary structure (2°)

α -Helix

- interhydrogen bond present
- helical structure
- covalent bond is present
- e.g. keratin

Tertiary structure

→ 3 bonds present

→ covalent bond

→ H. bonds

→ Disulphide bounded hydrophobic bond.

→ proper folding of protein e.g. Insulin

β -pleated sheet

→ inter H bond present
→ sheet like structure
e.g. fibroin & silk.

- 4 bonds present
- H-bond
- covalent bond
- Disulphide bond
- Vanderwaal force
- e.g. Haemoglobin

On the basis of function:

Structural

Enzymatic

Hormones

Respiratory

pigments

Transport

proteins

Storage

proteins

→ structural

forms the structure of body
eg keratin (Hair) collagen/muscle)

Enzymatic

→ Biological catalyst

eg Lipase, Amylase, protease.

Hormones

→ Maintain balance in body

eg. Insulin

→ Respiratory

coloured protein eg. Haemoglobin

Transport

→ material from one part to another part

eg. Haemoglobin

→ Storage

store metal ions of amino acids in cells.

TISSUES :-

→ Tissues are cells with similar functions are called Tissues.

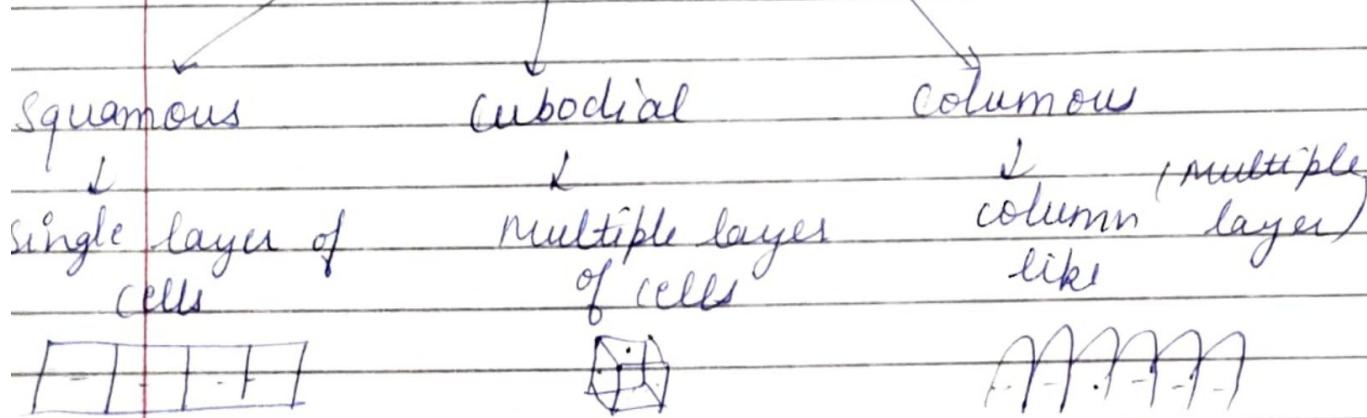
They may be classified as :-

1) Animal Tissue 2) Plant Tissue

Animal Tissue are of four types :-

- (i) Epithelial
- (ii) Connective
- (iii) Muscular
- (iv) Nervous

(1) Epithelial → They are the outermost layer of the body.



Example :-

Skin, stomach

function → To protect stomach (protection)

= Absorption-

(2) Connective Tissue :- They are present inside the body and they generally connect or separate a body part

Types of cells :- Blood Cells (Rbc's, Wbc's)

example = Bone, Blood

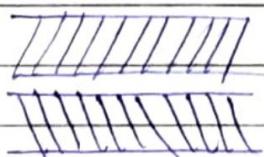
function = Distribute blood nutrients in the whole body.

Bones = Skeleton system

(3) Muscular Tissue :- It is present in each and every organ of the body and body parts.

Types of cells => Satisfied Unsatisfied
 | |

Layers in a pattern Layers in Random



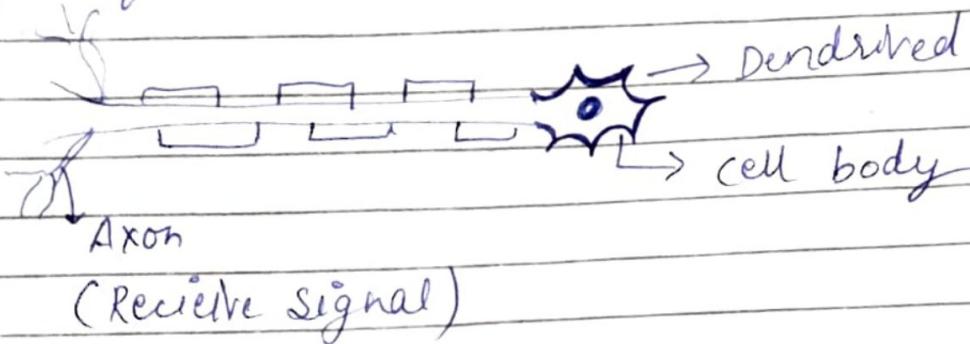
Example = Cardiac muscle, Hand muscles

function = Muscles = Movement of body

Voluntary Involuntary

(4) Nervous Tissue: → It is present in brain from spinal cord to the whole body.

Types of cells: Neurons



Example: Brain, spinal cord, sense organs.
functions: Send and receive signals.

Plant Tissue

Meristematic Tissue Permanent Tissue

(1) Meristematic Tissue: →

- ★ It is present in tips of stem & roots.
- ★ They are generally square & cubical.

function

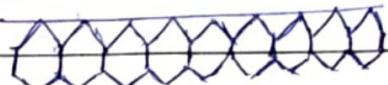
→ growth of plants.

(2) Permanent Tissue: →

Epidermis Parenchyma sclerenchyma Xylem phloem

① Epidermis :-

- * It is present in outermost layer of the plant. It is present on stem, leaves and roots.
- * Hexagonal type of cell is present in it.

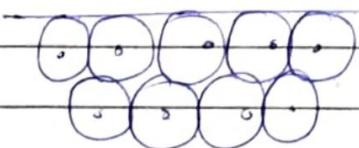


function :-

- * It acts as a barrier b/w plants and the surrounding.

② Parenchyma :-

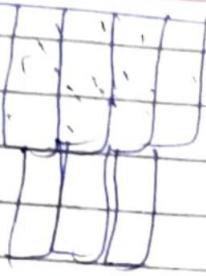
- * It is a basal soft tissue that forms majority of stem, root, leaves and fruits.
- * They are generally oval or round.



* function :- Storage of Nutrients

③ Sclerenchyma :-

- * It is hard, stiff tissue. present in branches, bark and seeds.
- * They are generally long, column type cells.



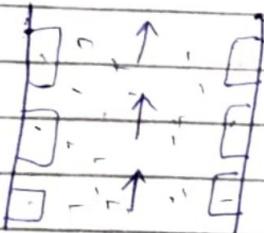
* It gives stiffness and support to the plant.

4. Xylem :-

* It is present inner to parenchyma.

* It forms cambium.

* They are generally dead cells that regulates one way flow of water from water to higher.

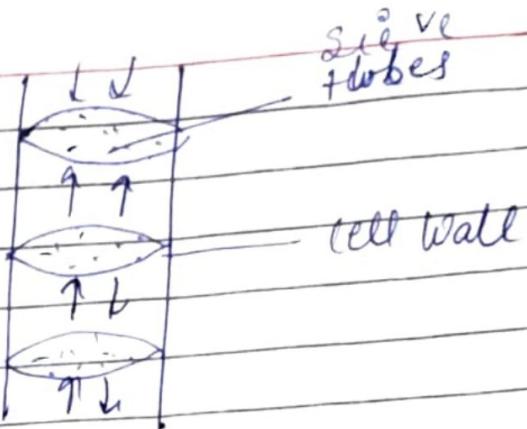


* Transportation of water and Minerals.

5. Phloem :-

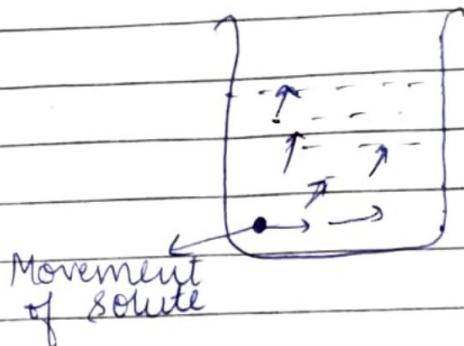
* It is present in the innermost part of the plant.

* They are live cells, multidirectional cells, long columnar cells that are joint with cell wall.



* Transportation of food and Nutrients.

Diffusion :→ Movement of solute particles from higher concentration to lower concentration.
ex → mixture of salts & sugar with water.



facilitated diffusion :→ To speed up the process of diffusion when an aid is used. ex → mixing salt & water with spoon.

Osmosis :→ Movement of solvent molecules through a semi-permeable membrane.
ex → raisins kept in water.

Absorption :→ It is a condition at which molecules of solvent enter

into bulk phase.

ex. Sponge kept in water.

Adsorption :→ Adsorption at surface is called adsorption.

ex → Adsorption of water by plant root.

Osmoregulation :→ It is process in which organism regulates the ionic balance and water to maintain its osmotic pressure.

There are two types :→

1.) Osmoconformers :→ They try to maintain water and ionic balance of their body with surroundings. for ex → Starfish, jelly fish.

2.) Osmoregulators :→ They do not have any concern with surrounding and they can maintain their own water & ionic balance.

for ex. Human, frogs, Tortoise.