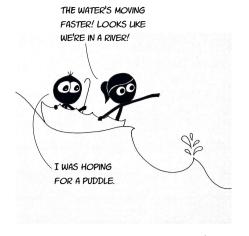
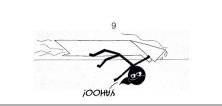
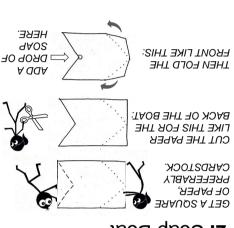
# SCIENCE MOM'S Guide to WATER Part 2



More FREE Science Mom Guides available at www.jennyballif.com

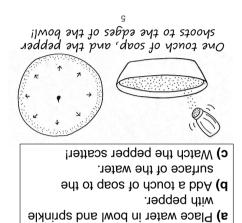


THEN SET THE BOAT IN WATER AND WATCH IT GO!



2. Soap Boat

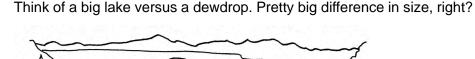


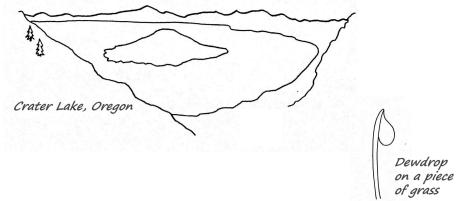


- :pou¡əɪ/i
- Water · Concentrated dish soap Ground black pepper
  - Bowl or plate

Materials:

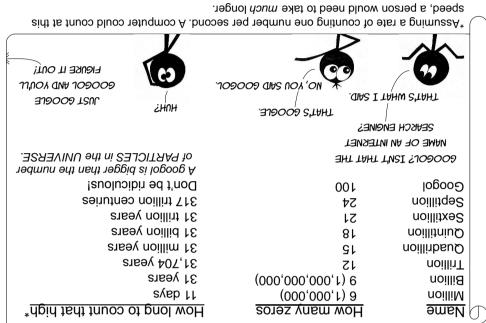
### 1. Pepper Scatter





The dewdrop is SUPER small compared to the lake. But a water molecule (the smallest bit of water you can have) is MUCH smaller than a dewdrop.

A single drop of water has more than 1,000,000,000,000,000,000,000 water molecules! That huge number with 21 zeros is called a sextillion, and it is a TRILLION TIMES BIGGER than one billion.



## LET'S TALK ABOUT BIG NUMBERS

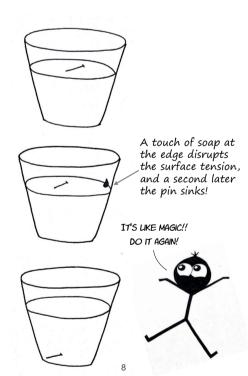
## 3. Floating Pin

#### Materials:

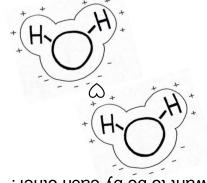
- A small pin or needle
- Bowl or cup
- Concentrated dish soap
- Water

#### Method:

- a) Fill bowl or cup with water and carefully place pin on surface. Hint: tweezers may help. The pin must be flat with the surface of the water. It will sink if it comes in at an angle.
- b) Add a touch of soap.
- c) Watch the pin sink!



negative sides. form between the positive and negative ( - ). Hydrogen bonds (♥) molecule is part positive (+) and part Positive loves negative. Each water Because opposites attract!



want to be by each other? But  $\mathcal{M}\mathcal{D}\mathcal{Y}$  do water molecules

## 4. Floating Paperclip

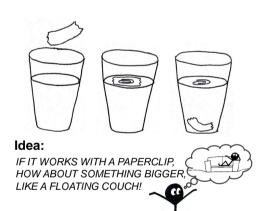
### Materials:

- Paper clip
- Tissue paper or paper towel
- Cup or bowl
- Water

#### Method:

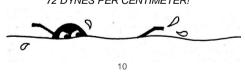
- a) Fill the cup with water and gently place a piece of tissue paper on the surface.
- b) Carefully place a dry paperclip on the tissue.
- c) The tissue should sink. If it doesn't, give it a gentle push downward.

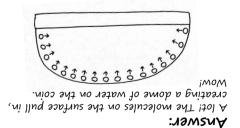
Tip: be sure that the cup and water are not soapy.

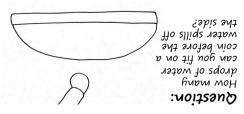


### Reality:

THE SURFACE TENSION OF WATER IS ONLY 72 DYNES PER CENTIMETER!

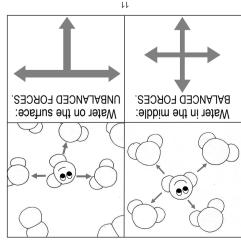






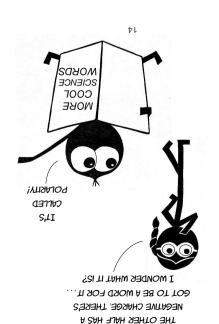
d allows us to fill cups above the brim, eighbors. This creates surface tension,

than they like air, so the molecules on . Moisnel easing



or make a dome of water on a coin. which helps raindrops stay together an the surface bond more tightly to their n Water molecules like each other more

HOM DOES IL MOKKS



OF WATER IS POSITIVE AND THAT'S SO COOL THAT PART

$\mathbf{B}$	A		
B			
F	E	E	b
E	G		