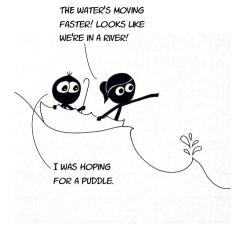
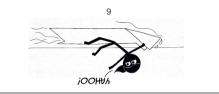
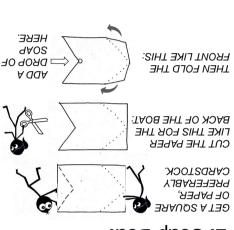
SCIENCE MOM'S Guide to WATER Part 2



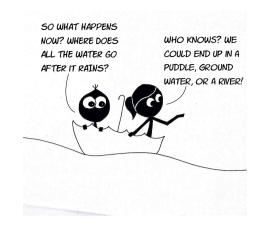
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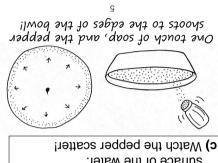


THEN SET THE BOAT IN WATER AND WATCH IT GO!



2. Soap Boat





- surface of the water. b) Add a touch of soap to the with pepper. s) Place water in bowl and sprinkle
 - Water

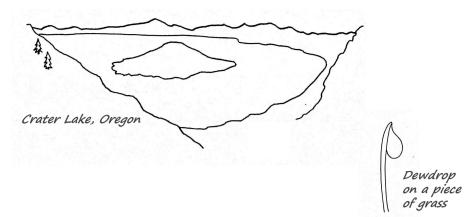
:pou¡əɪ/i

- · Concentrated dish soap
- Ground black pepper Bowl or plate

Materials:

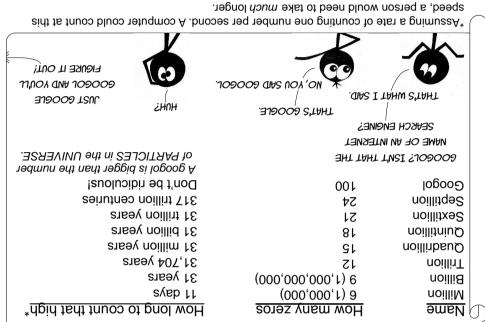
1. Pepper Scatter

Think of a big lake versus a dewdrop. Pretty big difference in size, right?



The dewdrop is SUPER small compared to the lake. But a water molecule (the smallest bit of water you can have) is MUCH smaller than a dewdrop.

A single drop of water has more than 1,000,000,000,000,000,000,000 water molecules! That huge number with 21 zeros is called a sextillion, and it is a TRILLION TIMES BIGGER than one billion.



LET'S TALK ABOUT BIG NUMBERS

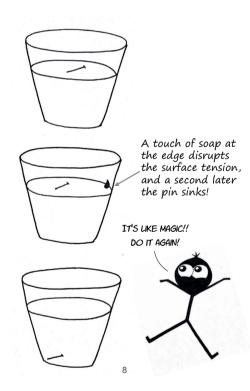
3. Floating Pin

Materials:

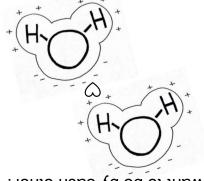
- A small pin or needle
- Bowl or cup
- Concentrated dish soap
- Water

Method:

- a) Fill bowl or cup with water and carefully place pin on surface. Hint: tweezers may help. The pin must be flat with the surface of the water. It will sink if it comes in at an angle.
- **b)** Add a touch of soap.
- c) Watch the pin sink!



negative sides. form between the positive and negative (-). Hydrogen bonds (♥) molecule is part positive (+) and part Positive loves negative. Each water Because opposites attract!



want to be by each other? But $\mathcal{M}\mathcal{D}\mathcal{Y}$ do water molecules

4. Floating Paperclip

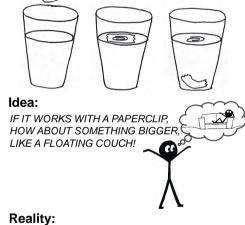
Materials:

- Paper clip
- Tissue paper or paper towel
- Cup or bowl
- Water

Method:

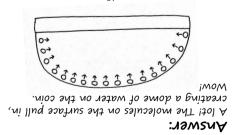
- a) Fill the cup with water and gently place a piece of tissue paper on the surface.
- b) Carefully place a dry paperclip on the tissue.
- c) The tissue should sink. If it doesn't, give it a gentle push downward.

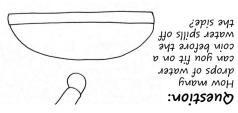
Tip: be sure that the cup and water are not soapy.



THE SURFACE TENSION OF WATER IS ONLY 72 DYNES PER CENTIMETER!

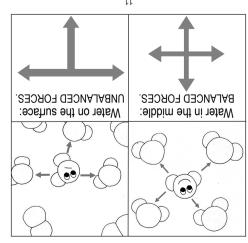






d allows us to fill cups above the brim, eighbors. This creates surface tension, than they like air, so the molecules on

Surface Tension.



or make a dome of water on a coin. which helps raindrops stay together an the surface bond more tightly to their n Water molecules like each other more

HOM DOES IL MOKKS



GOT TO BE A WORD FOR IT... NEGATIVE CHARGE, THERE'S THE OTHER HALF HAS A OF WATER IS POSITIVE AND THAT'S SO COOL THAT PART

\mathbf{B}	A		
B			
F	E	E	b
E	G		