Effect of surface modifications to single and multilayer graphene temperature coefficient of resistance

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ABSTRACT: The effect of the interface on the temperature coeffienct of resistance (TCR) on single-layer and multilayer graphene due to composite core-shell nanoparticles and various substrates was experimentally investigated. Hall Effect measurements using the van der Pauw method demonstrate that these modifications strongly influence the electrical response of single-layer graphene, thus increasing TCR up to a 0.456% per K resistance change on SiO2 with red nanoparticles (RNPs) deposited on the top surface of graphene. However, these interactions have muted influence on multilayer graphene due to the screening effect of non-superficial layers, only achieving a -0.0998% per K resistance change on Si3N4 with RNP. We also show contrary thermal sensitivity responses between single-layer and multilayer graphene after the addition of RNP to the surface.

Graphene is a two-dimensional carbon-based material. Due to its hexagonal lattice configuration, graphene has many outstanding thermal, electrical, and optical characteristics, granting promising applications in advanced electronics. Significant efforts have been spent on studying these intrinsic properties, including multiple low temperature measurements of suspended graphene. While these works have greatly advanced understanding of the material itself, there is still much to learn about material interactions. Many of graphene's properties are due to its immense surface-area to volume ratio, however this also implies that external interfacing is a key factor in determining material behavior. Different substrates and surface modifications can drastically alter graphene's most important properties, such as electrical conduction pathing.

Graphene has a theoretical electrical conductivity limit of 200,000 cm²·V⁻¹·s⁻¹, and practical experiments have observed conductivity of ~185,000 cm² V⁻¹ s⁻¹ in optimized conditions.¹ In the lattice geometry, each carbon atom covalently bonds to 3 others, leaving its 4th valence electron to form the pi and pi-star bands responsible for both dense carrier concentration and high carrier mobility. In the case of single-layer graphene (SLG), there are just two channels for carrier flow, corresponding to the top and bottom surfaces of the layer. Ideally, these channels have an equal carrier distribution, allowing for uniform motion while under the influence of an external electric field. These electrons almost never interact with carbon, making electron-electron scattering the dominant source of carrier interactions in these channels.³

The introduction of additional layers forms multilayer graphene (MLG). The top and bottom surfaces act similarly to SLG, but every extra layer adds an interlayer channel. Even while under the influence of an external

electric field, the electrons in the interlayers are affected primarily by phonon and electronic scattering. This is because inter channels operate under zero electric field intensity due to equal interference generated by surrounding carbon ions, leaving carriers to undergo entropic thermal motion.³ This is significant as it creates a distinguishing property between SLG and MLG forms.

Carrier mobility's temperature dependence differs depending on whether it is flowing in a surface or inter layer channel. In SLG, the dominant electron-electron scattering is negligibly dependent on temperature and therefore suspended SLG has a very small Temperature Coefficient of Resistance (TCR). However, in MLG, the effect of temperature is much more significant as there is no electric field intensity in the interlayers. Changes in temperature strongly influence the thermal motion and can actually reduce the resistivity as temperature increases. This results in a negative TCR value for MLG. Additionally, since the interlayer channel has more influence in thicker graphene, the TCR is larger for isolated graphene with more layers. Thus, graphene thickness can rapidly begin to influence electrical conductivity.³

Importantly, this mobility temperature dependence can be affected by different material interfaces such as substrate and surface modification as mentioned before. The effect of adhering graphene to a substrate has been well observed to alter its conductive properties when compared to suspended graphene; however, there has been little research into specific substrate dependence. This work investigates the interface effects between three different substrates with SLG or MLG. Specifically, the TCR of graphene forms are measured when on $\mathrm{Si}_3\mathrm{N}_4$, amorphous SiO_2 (glass), and crystalline SiO_2 substrates to explore bottom

channel interactions. Additionally, the effects on top channel interactions are studied by the incorporation of silver shell, silica-core red nanoparticles (RNP). This work aims to advance our understanding of how external influences can affect graphene conduction paths thus enabling the production of higher quality graphene devices.

Experimental Procedure

Samples were created via chemical vapor deposition (CVD) which was then wet transferred onto a target substrate, before further modification. Figure 1 depicts a broad overview of the sample preparation process. RNPs were analyzed after full preparation, concluding with multiple TCR characterization tests.

Single layer graphene was acquired commercially (Cheap Tubes Inc.), and MLG was grown onto precleaned copper foil (Alfa Aesar) using a standard high temperature CVD process. After growth, the MLG/copper was spin coated with PMMA495 A2, and backside graphene was removed by soft abrasion. The copper pieces were then cut to fit each sample and immersed in aqueous ammonium persulfate. After complete etching, MLG/PMMA was left behind. The resulting structure was transferred to a water bath three times for 10 minutes each time to remove contaminants before transferring onto the target substrate. The samples were dried via hot plate. The PMMA layer was then dissolved via acetone bath for 30 minutes, and the final structure was rinsed with isopropyl alcohol (IPA) and dried via hotplate.

Wet transfer of SLG and MLG samples was completed onto each of three acetone-IPA cleaned target substrates: glass, SiO₂, Si₃N₄. Once mounted, Raman spectra analysis was used to confirm and characterize quality of the MLG. The inset in Figure 1b shows the Raman spectra data (532nm inlaser wavelength), confirming the presence and proper relation of D, G, G', and 2D peaks. Atomic force microscopy (AFM) found a thickness of ~8nm, consistent with Raman results.

The silver shell, silica-core nanoparticles were created manually. First, 300nm silicon oxide nanoparticle cores were modified in an APTES solution to prepare for shell deposition. In parallel, an HAuCl₄ solution was reduced in order to form gold seeds, which were then mixed with the modified silica cores to create silica/gold seed nanoparticles. The resulting solution was mixed with AgNO₃ to reduce the seed particles and grow a silver nano-shell. The shell was allowed to grow 15nm before the process was stopped in order to ensure an absorption peak of 800nm.

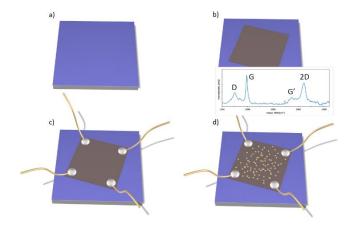
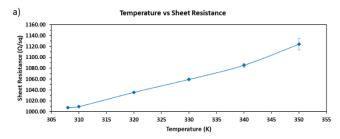


Figure 1. Overview of the sample preparation process. a) target substrate is cleaned in preparation for transfer; b) graphene transfer using a wet process. Inset: Raman spectra confirms the presence of MLG; c) contacts are soldered or wire bonded onto the graphene; d) nanoparticles are deposited onto the graphene

Sample contacts for SLG samples were made by soldering indium-tin directly onto the graphene, and MLG samples received gold contacts through wire bonding. A Hall Effect machine was used obtain resistivity, charge concentration, and mobility results from 310-350K with a 10K temperature step for SLG samples. MLG samples were tested in a vacuum cryostat in van der Pauw configuration from 200-320K with a 10K temperature step. After baseline characteristics were observed, samples were coated with the RNP and then retested to determine how the nanoparticles affected electrical properties of both graphene forms.

Results and Discussion

Figure 2 shows the data for SLG on glass. Figure 2a shows the sheet resistance in units of Ω/sq . Figure 2b shows the mobility in units of cm²/Vs on the left axis and the sheet concentration in units of cm⁻² on the right axis. The blue line with the diamond markers is the mobility while the orange line with the circle markers is the sheet concentration. All figures are shown as a function of temperature. Table 1 shows the linear fit and % change of each of these graphs. SLG on SiO₂ experiences a linear increase in resistance as temperature increases. The mobility decreases while the sheet concentration increases over the same range. Since resistivity is inversely proportional to the mobility and the sheet concentration, this shows that the reduction in mobility is the dominant mechanism by which the resistivity increases as expected by theoretical results.3 A simple check of the % change shows that the combined change is similar to the change in resistivity. The reduction in mobility is caused by interfacial electron-phonon scattering between the graphene and the glass substrate.3 The slight increase in sheet concentration is due to increased absorption of ambient molecules.4



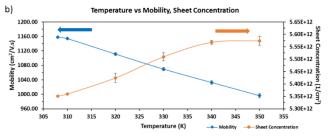


Figure 2. SLG results on glass without nanoparticles for a) Sheet Resistance, b) Mobility (blue line with diamond markers, left side axis), and Sheet Concentration (orange line with circle markers, right side axis) over the temperature range 300-350K

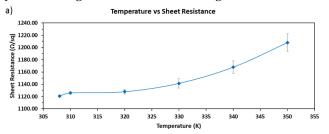
Table 1. SLG on glass data in Figure 2

	Slope	310K Value	% Change per K (310-350K)
Sheet	2.712	1009.304	0.284%
Resistance			
Mobility	-3.914	1150.810	-0.341%
Sheet	5.847E+09	5.369E+12	0.100%
Concentration			

Linear fit slope, 310K value, and % change of data shown in Figure 2 for SLG on glass

Figure 3 shows the analysis of the same SLG sample now modified with RNP. Figure 3a shows the sheet resistance, Figure 3b shows the mobility as the blue line with the diamond markers oriented on the left axis and the sheet concentration as the orange line with circle markers oriented on the right axis. All figures are shown as a function of temperature. As before, Table 2 shows the linear fit data. Note that a linear fit for Figure 3a still shows a fit of about 90%. The RNP can be seen to cause a large change in observed properties of graphene. The sheet resistance increases by about $100\Omega/\text{sq}$ as a result of the new scattering centers offered by the presence of the nanoparticles. This also causes a reduction in the magnitude of the temperature coefficient. However, this change is largely due to the swap in dominant mechanism for resistance change. Here, the mobility is increased as a result of increasing temperature and the sheet concentration decreases. Since the nanoparticles are made of a silica core surrounded by a silver shell, it is likely that the silver of the nanoparticles are causing a charge transfer to take place. The nanoparticles act as carrier absorbers and interrupt the surface channel such that increased thermal motion influences the ability of the charge carriers to flow inside the graphene.^{5,6}

The data in Figure 4 shows the results of SLG on crystalline SiO₂. Figure 4a shows the sheet resistance in ohms/sq. As before, a rise in the sheet resistance as temperature increases can be seen. Figure 4b shows the measured mobility and the sheet concentration. As before, the blue line with the diamond markers is the mobility, oriented on the left axis, and the orange line with the circle markers is the sheet concentration, oriented on the right axis. Table 3 shows the linear fit data. In comparison with the glass substrate, the mobility is much lower. This indicates that the ordered lattice of crystalline SiO₂ is likely providing a greater scattering source than the amorphous lattice of glass. It is known that Si does not affect graphene as strongly as O does. The ordered O atoms of the SiO₂ can thus be considered the scattering centers. Oxygen is also known to induce a p-type doping effect on graphene and this is seen by the increased sheet concentration measured.7 However, the mobility increases with increasing temperature while the sheet concentration decreases with increasing temperature. This is the opposite situation from the glass substrate. Through this combination, the overall percent change remains similar to the glass case.



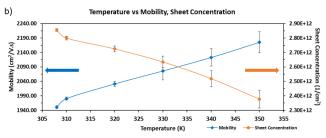


Figure 3. SLG results on glass with RNP for a) Sheet Resistance, b) Mobility (blue line with diamond markers, left side axis), and Sheet Concentration (orange line with circle markers, right side axis) over the temperature range 300-350K

Table 2. SLG on glass with RNP data in Figure 3

	Slope	310K Value	% Change per K (310-350K)
Sheet	1.893	1117.715	0.182%
Resistance			
Mobility	5.082	1972.971	0.246%
Sheet	-1.068E+10	2.826E+12	-0.377%
Concentration			

Linear fit slope, 310K value, and % change of data shown in Figure 3 for SLG on glass with nanoparticles

Figure 5a shows the sheet resistance data for SLG on crystalline SiO₂ with RNP. As expected, the resistance

value has increased due to the increased scattering centers. Additionally, there is a larger percent change than without nanoparticles. Figure 5b shows the mobility and sheet concentration for SLG on SiO₂ with RNP. The format follows the previous graphs. Table 4 shows the linear fit data. The mobility has seen an increase compared to without nanoparticles while the sheet concentration has decreased. The data shows similar characteristics to SLG on glass with RNP. The decrease in concentration allows for less likelihood of electron-electron scattering. This allows the mobility to increase enough to offset the decrease in carrier concentration.

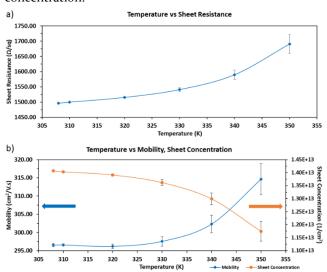


Figure 4. SLG results on SiO_2 without nanoparticles for a) Sheet Resistance, b) Mobility (blue line with diamond markers, left side axis), and Sheet Concentration (orange line with circle markers, right side axis) over the temperature range 300-350K

Table 3. SLG on SiO2 data in Figure 4

	Slope	310K Value	% Change per K (310-350K)
Sheet	4.183	1487.182	0.281%
Resistance			
Mobility	0.365	294.671	0.124%
Sheet	-4.975E+10	1.421E+13	-0.350%
Concentration			

Linear fit slope, 310K value, and % change of data shown in Figure 4 for SLG on glass

The final substrate tested was Si_3N_4 . Figure 6a shows the sheet resistance of SLG on Si_3N_4 without nanoparticles. As in the other cases, the sheet resistance is increasing. Figure 6b shows the mobility and the sheet concentration, formatted as previously discussed. Table 5 shows the linear fit data. As in SiO_2 , they are increasing and decreasing respectively. Of particular interest is that SLG on Si_3N_4 has the highest resistance and most sensitivity of any of the tested substrates without nanoparticles. While some studies have indicated that Si_3N_4 has weak interactions with graphene, this was the case for β - Si_3N_4 (0001) where graphene has the

potential to lie completely flat.⁸ It is unlikely that our graphene lies as flat as possible due to the wet transfer procedure. This causes an increase in resistance. Other studies indicate that our value for sheet resistance of graphene on SiN is acceptable.⁹ It is likely that the low thermal mass of the Si_3N_4 substrate helps to achieve a higher thermal response under the testing procedure. This increase in heat transfer allows the graphene to respond accordingly.

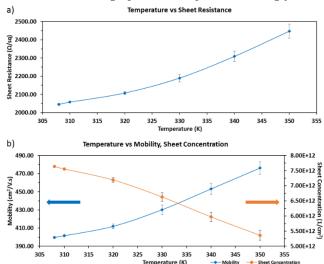


Figure 5. SLG results on SiO2 with RNP for a) Sheet Resistance, b) Mobility (blue line with diamond markers, left side axis), and Sheet Concentration (orange line with circle markers, right side axis) over the temperature range 300-350K

Table 4. SLG on SiO2 with RNP data in Figure 5

	Slope	310K Value	% Change per K (310-350K)
Sheet	9.297	2040.913	0.456%
Resistance			
Mobility	1.812	399.203	0.454%
Sheet	-5.426E+10	7.609E+12	-0.713%
Concentration			

Linear fit slope, 310K value, and % change of data shown in Figure 5 for MLG on glass

The final SLG test was Si₃N₄ with RNP. Figure 7a shows the sheet resistance data gathered for this test. Figure 7b shows the mobility on the left axis using the blue line with diamond markers. Additionally, Figure 7b shows the sheet concentration on the right axis using the orange line with circle markers. Table 6 shows the linear fit data. As expected, the RNP increases the resistance as compared to without nanoparticles. However, in this case, the sensitivity decreases. While SiN has a strong effect on the electrical channels, the temperature coupling is less significant when nanoparticles are introduced. The percent change is strongly reduced in this scenario. While the mobility slightly increases, the decrease in concentration offsets this gain and results in a weaker temperature connection. In this case, the charge transfer results in less carriers to be affected by thermal motion.5,6

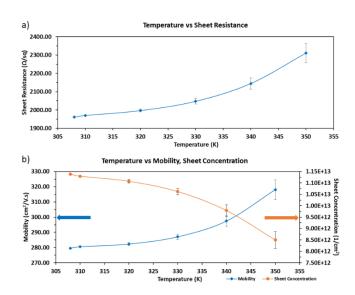


Figure 6. SLG results on SiN without nanoparticles for a) Sheet Resistance, b) Mobility (blue line with diamond markers, left side axis), and Sheet Concentration (orange line with circle markers, right side axis) over the temperature range 300-350K

Table 5. SLG on SiN data in Figure 6

	Slope	310K Value	% Change per K (310-350K)
Sheet	7.659	1946.703	0.393%
Resistance			
Mobility	0.820	277.450	0.296%
Sheet	-6.373E+10	1.149E+13	-0.555%
Concentration			

Linear fit slope, 310K value, and % change of data shown in Figure 6 for MLG on glass

Testing was repeated with MLG equivalents of the previous samples in order to observe the effects of different substrates on the resistivity and TCR of multilayer graphene. As a direct comparison to the SLG samples, Figure 8 shows the results for MLG on glass. The blue line with diamond markers on the left axis is MLG on glass without nanoparticles. As expected from theory, the resistance of the MLG samples goes down as temperature increases.3 This is because the increase in temperature amplifies the thermally driven carrier activity. This is opposed to the SLG behavior shown above where the resistance increases with increasing temperature. Figure 8 also shows MLG with red nanoparticles on glass using the orange line with circle markers on the right axis. Here the resistance is increased a significant amount, showing that the nanoparticles had a pronounced presence as a scattering center. Table 7 shows the linear fit data for Figure 8. The % change is smaller than in the SLG case, as the substrate effect is lessened with increasing thickness of graphene. While the resistance is lower than the SLG case, this is likely due to a difference in testing protocol between SLG and MLG samples. SLG samples were tested in ambient and thus were contaminated

by atmospheric particles. MLG samples were tested in vacuum and not subject to such external scattering sources.

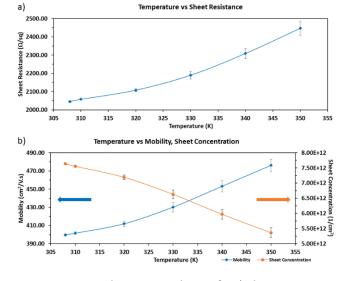


Figure 7. SLG results on SiN with RNP for a) Sheet Resistance, b) Mobility (blue line with diamond markers, left side axis), and Sheet Concentration (orange line with circle markers, right side axis) over the temperature range 300-350K

Table 6. SLG on SiN with RNP data in Figure 7

	Slope	310K Value	% Change per K (310-350K)
Sheet	2.238	2349.740	0.095%
Resistance			
Mobility	0.826	325.342	0.254%
Sheet	-2.533E+10	8.152E+12	-0.311%
Concentration			

Linear fit slope, 310K value, and % change of data shown in Figure 7 for MLG on glass

Figure 9a shows MLG on a crystalline SiO2 substrate using the blue line with diamond markers oriented along the left axis. The orange line with circle markers oriented along the right axis in Figure 9 shows MLG with red nanoparticles on a crystalline SiO₂ substrate. Unlike the amorphous glass case, the presence of ordered SiO₂ increases the interfacial scattering between the graphene and the substrate layers. This interfacial scattering decreases the magnitude of the % change as the electron-phonon scattering present reduces the mobility in the bottom most layers of graphene.3 The linear fit data in Table 8 show this change. As it is known that the O atoms in SiO₂ are mainly responsible for the scattering effect on graphene, it is likely that the ordered SiO₂ results in more opportunities for the O atom to interact with graphene than in the unordered case of glass.7

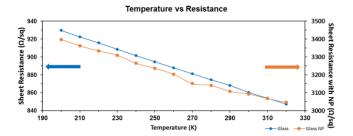


Figure 8. Sheet resistance of MLG on glass with (a) no nanoparticles (b) RNP

Table 7. MLG on glass data in Figure 8

	Slope	300K Value	% Change per K (200-310K)
Glass	-0.685	860.833	-0.0737%
Glass (RNP)	-3.010	3088.748	-0.0861%

Linear fit slope, 310K value, and % change of data shown in Figure 8 for MLG on glass

The last case observed was MLG on Si₃N₄. Figure 10 shows the results as a function of temperature. Table 9 shows the linear fit data. As is easily viewed from the graph, graphene on silicon nitride results in the lowest resistance of the three interfaces. This corresponds with experimental data from Benyamin Davaji et al.10 and is due to the reduced influence of N atoms on interfacial scattering.8 In this case, the addition of red nanoparticles did not increase the measured resistance by a significant margin. Whereas the resistance increased by at least $100\Omega/\text{sq}$ in all previous cases, the resistance only shifted upwards by 30-40 Ω /sq in this case. However, this sample yielded the strongest dependence on temperature as shown in Table 9. It is likely that the reduced effect on the electric field from the graphene/substrate interface within the interlayers of the MLG allow for the thermal motion of the carriers to be more prominent.

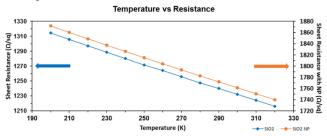


Figure 9. Sheet resistance of multilayer graphene on SiO₂ with (a) no nanoparticles (b) RNP.

Table 8. MLG on SiO₂ data in Figure 9

	Slope	300K Value	% Change per K (200-310K)
SiO ₂	-0.815	1231.879	-0.0622%
SiO ₂ with RNP	-1.096	1761.078	-0.0587%

Linear fit slope, 310K value, and % change of data shown in Figure 9 for MLG on SiO_2

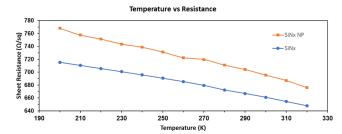


Figure 10. Sheet resistance of multilayer graphene on Si₃N₄ with RNP (square markers) and without red nanoparticles (circle markers).

Table 9. MLG on Si₃N₄ data in Figure 10

	Slope	300K Value	% Change per K (200-310K)
Si ₃ N ₄	-0.561	661.163	-0.0785%
Si ₃ N ₄ (RNP)	-0.721	694.652	-0.0998%

Linear fit slope, 310K value, and % change of data shown in Figure 10 for MLG on $\mathrm{Si_3N_4}$

Conclusion

Substrate-induced scattering changes the TCR of SLG by breaking the charge symmetry of top and bottom surfaces, as well as by introducing an additional scattering source. The oxide layer of the SiO2 acted as such a scattering source when interacting with graphene, similar to Si₃N₄ substrate. In these cases, increasing temperature increased mobility while decreasing the sheet concentration. Without the introduction of red nanoparticles, Si₃N₄ was found to affect graphene the most, achieving a 0.393% change per K. With RNP, crystalline SiO₂ to provide the largest overall change of 0.456% per K. In the case of MLG, the effect of the substrate is screened by the bottom layer and is largely influenced by graphene thickness." This is due to the bottom surface carriers providing a counter electric field, and the force still diminishes with distance at this scale. Therefore, substrates should have a minor interfacial scattering effect on MLG compared to SLG, which was confirmed in the data.

Top surface channel modulation was also explored in this work with the introduction of RNP. In all cases, RNP samples had higher resistance for two reasons: (1) strong surface binding caused structural deformations that are known to increase graphene's resistivity; (2) the RNPs create charge transfer wells, reducing graphene carrier efficiency. This consequently affects the temperature response of the resistivity, differentiating SLG and MLG. The former exhibits a decrease in the magnitude of TCR due to decreases in efficiency. The latter displays an increase in the magnitude TCR as the effect is screened and unable to outweigh the increased thermal motion. The situation is reversed in crystalline SiO₂, likely due to the ordered influence of oxygen.

These interfacial effects play a significant, complex role in determining graphene behavior. Continued characteri-

zation of material interactions is crucial, as a better understanding will lead to improvements in quality and reliability of future graphene-based devices.

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Author Contributions

J.T. conducted sample preparation, measurements for SLG data, analysis, and prepared the manuscript. Y.L. prepared the nanoparticles. H.Y. performed electrical measurements for MLG data. S.S. reviewed data and edited the manuscript. All authors have given approval to the final version of the manuscript.

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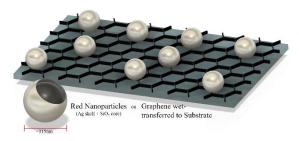
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ABBREVIATIONS

SLG, single layer graphene; MLG, multilayer graphene; TCR, Temperature Coefficient of Resistance; RNP, red nanoparticle; CVD, chemical vapor deposition; IPA, isopropyl alcohol; AFM, atomic force microscopy.

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